

Migration of methylethynyl group in a long-lived carbocation

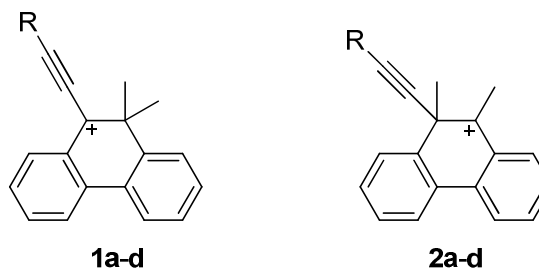
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Electronic Supplementary Information

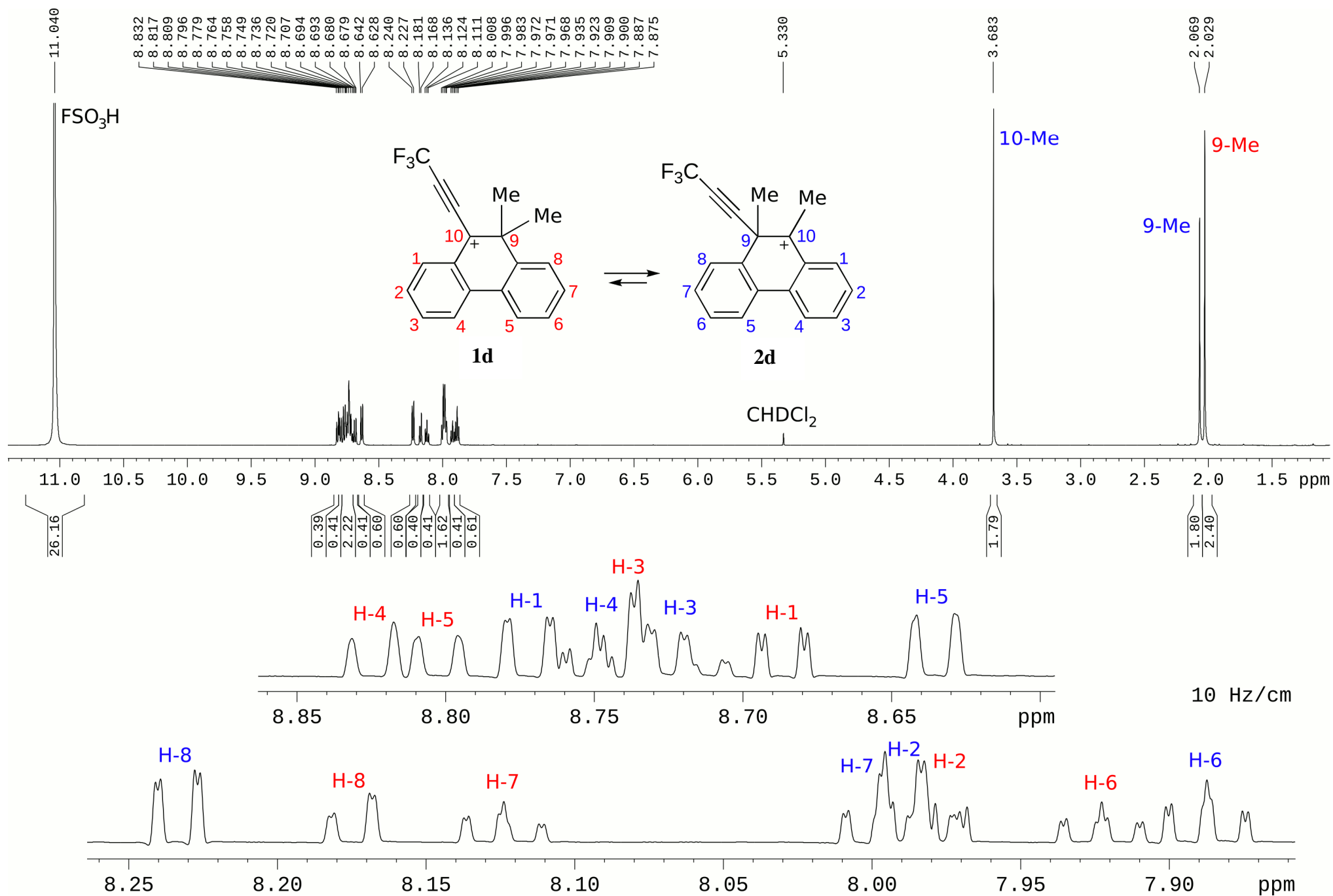


R = H (**a**), Me (**b**), Ph (**c**), CF₃ (**d**)

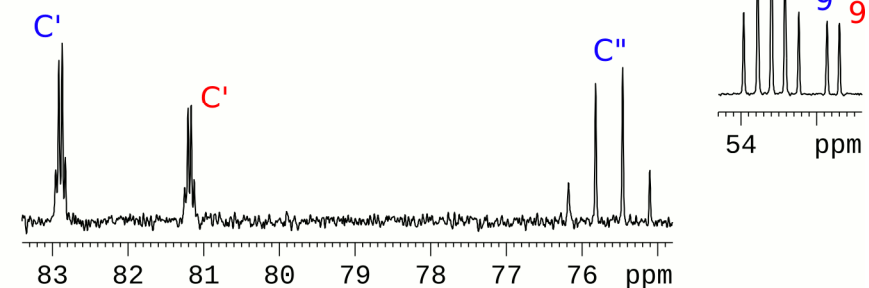
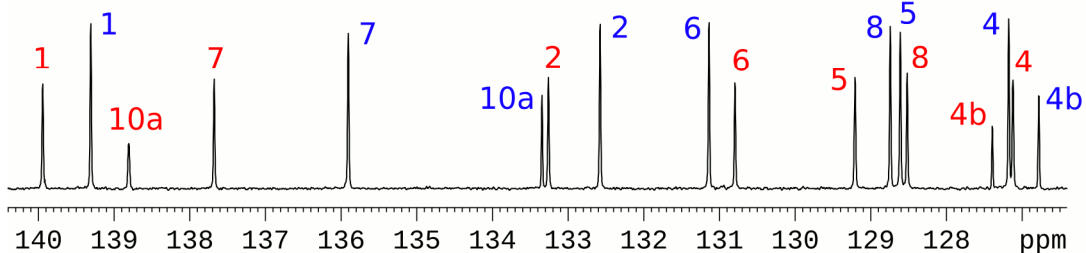
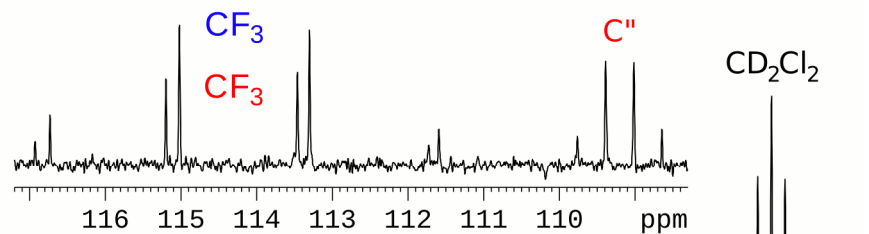
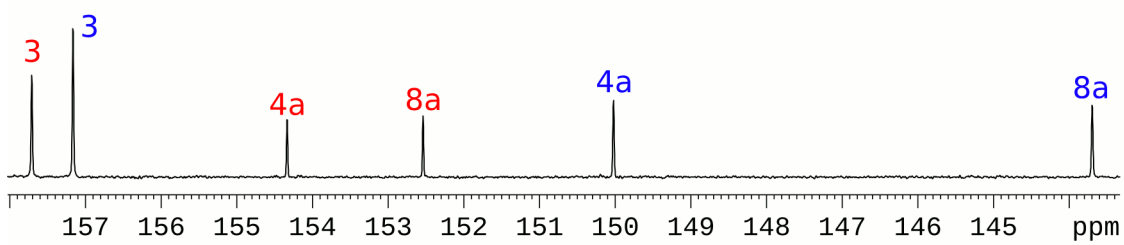
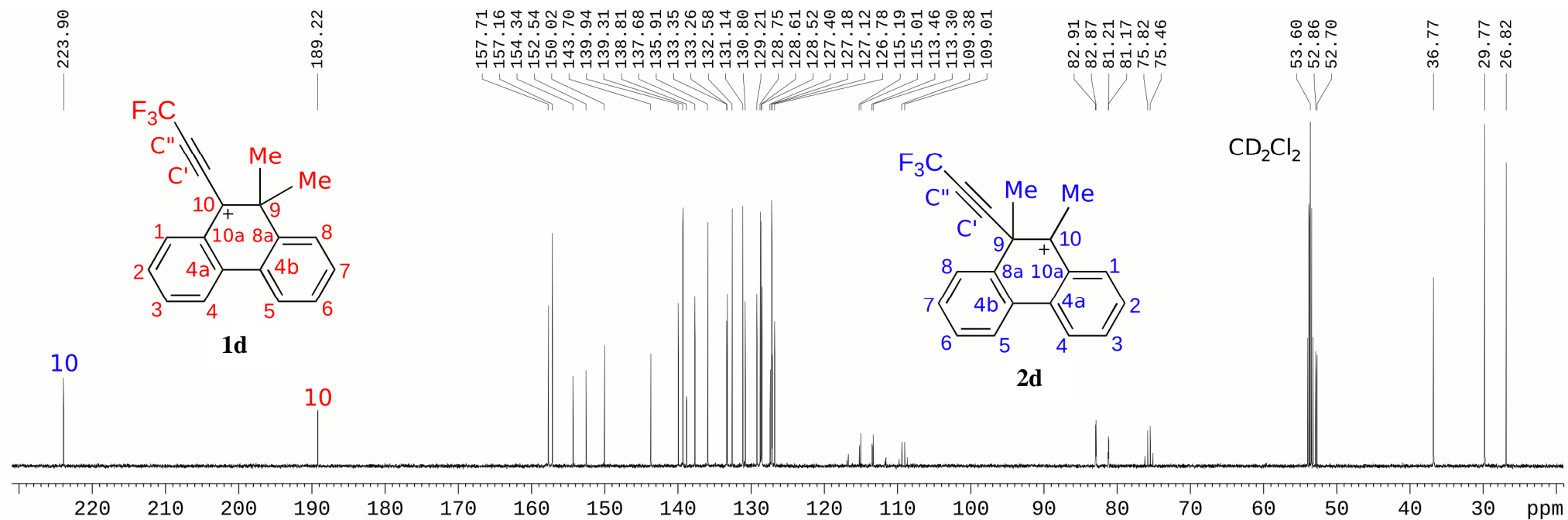
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¹H NMR spectrum of cations **1d** and **2d** in FSO₃H-SO₂ClF-CD₂Cl₂ at -50 °C

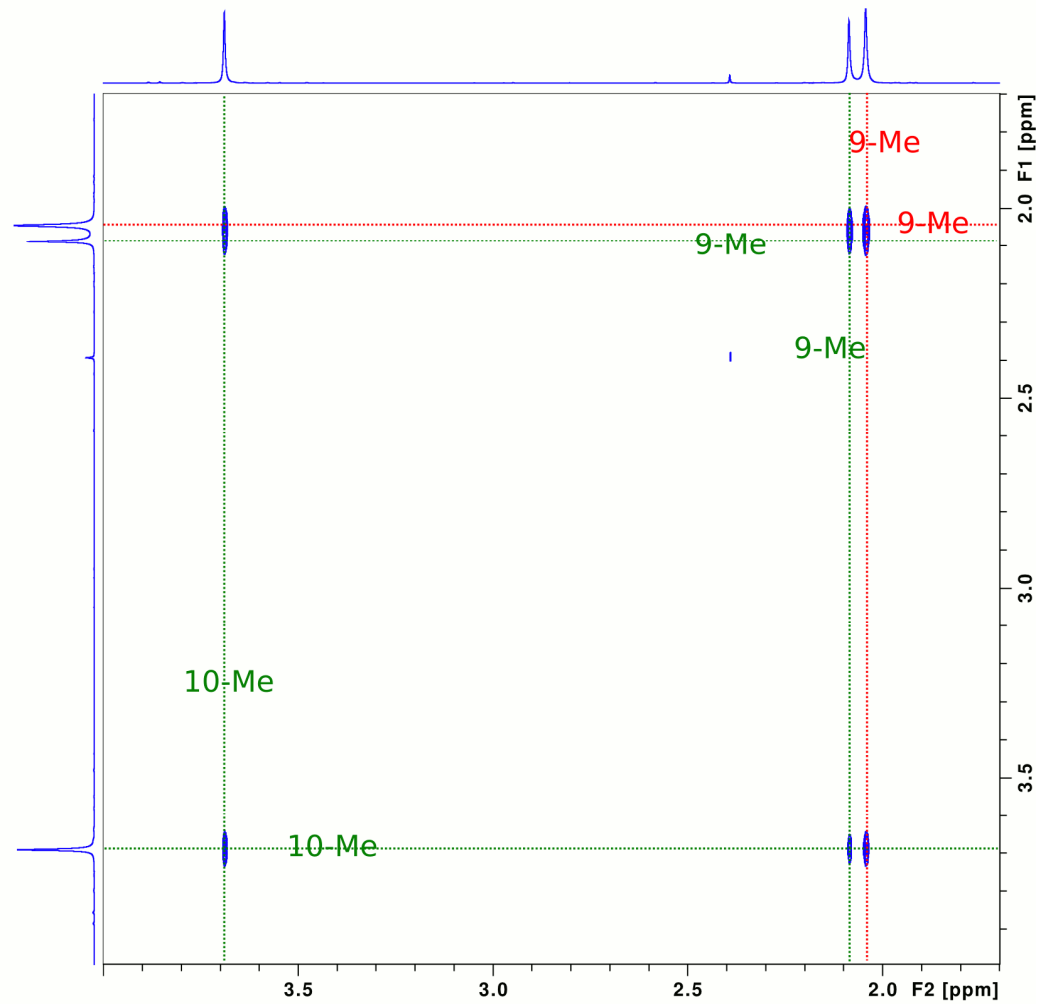
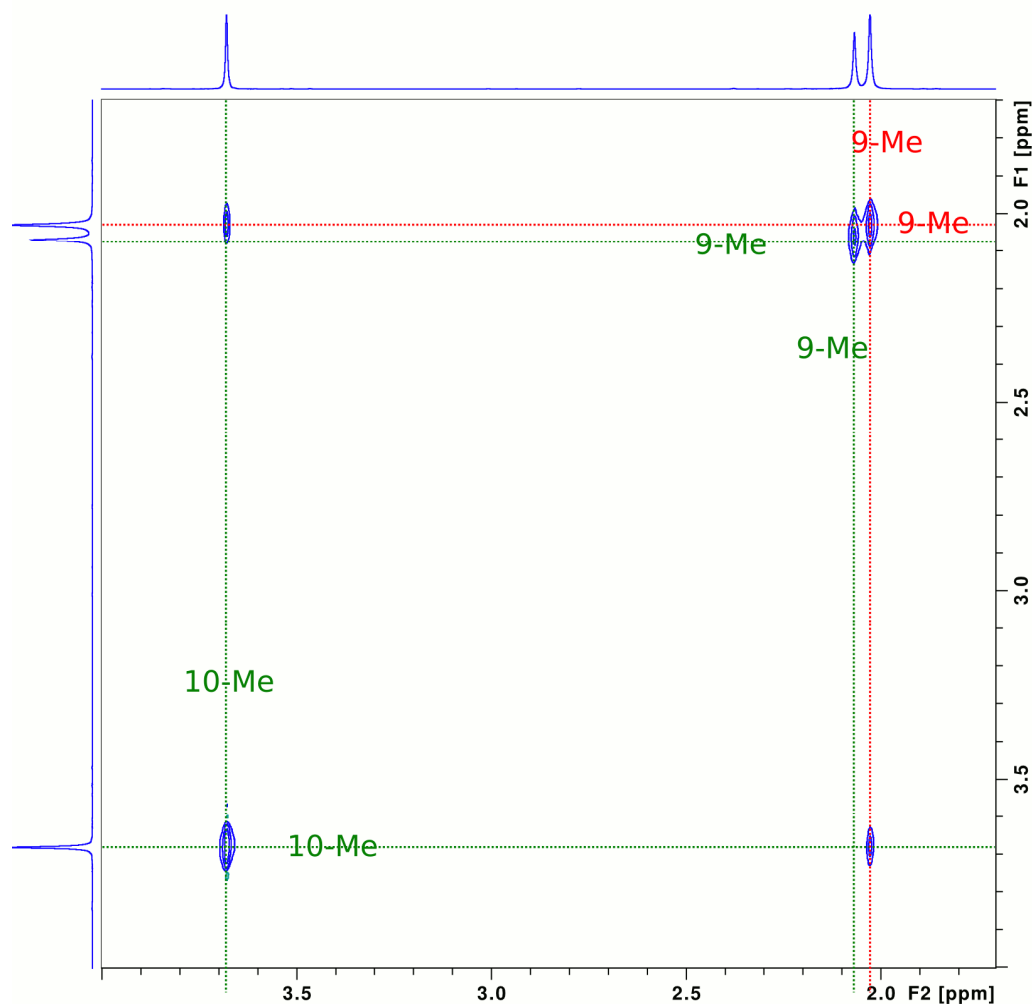
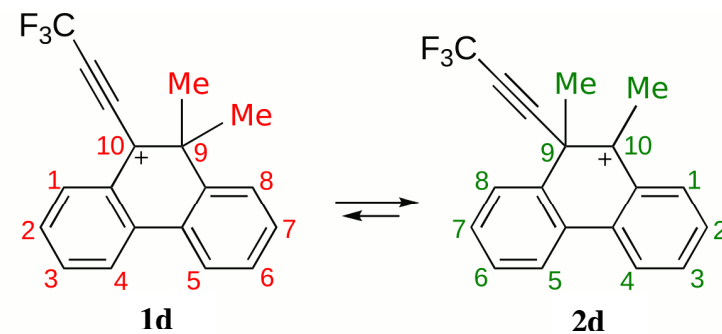


^{13}C NMR spectrum of cations **1d** and **2d** in $\text{FSO}_3\text{H-SO}_2\text{ClF-CD}_2\text{Cl}_2$ at -50°C

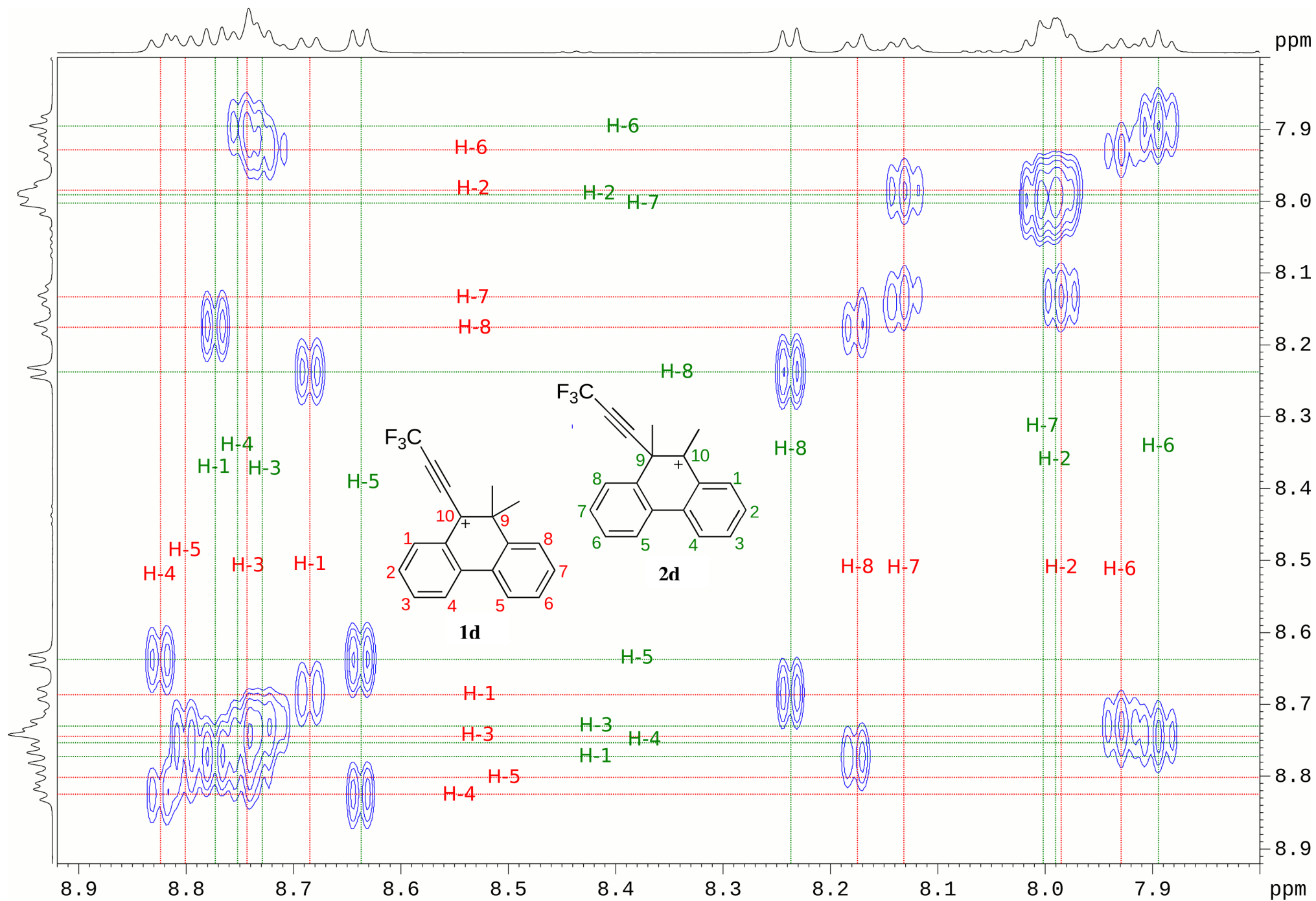


NOESY spectra (methyl region) of cations **1d** and **2d** in FSO₃H-SO₂ClF-CD₂Cl₂

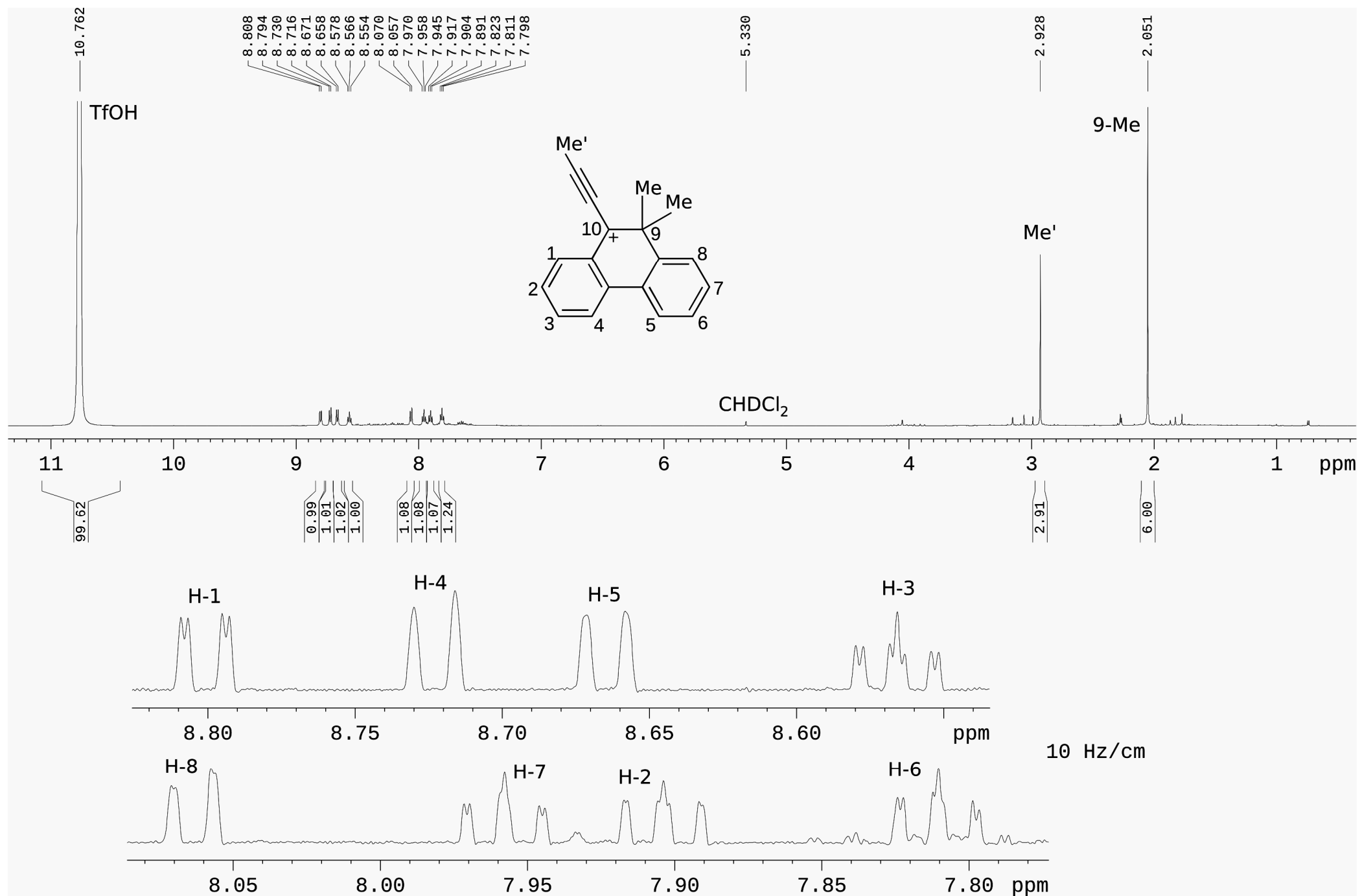
Left: at -47 °C (mixing time 0.4 s)
Right: at -22.5 °C (mixing time 0.5 s)



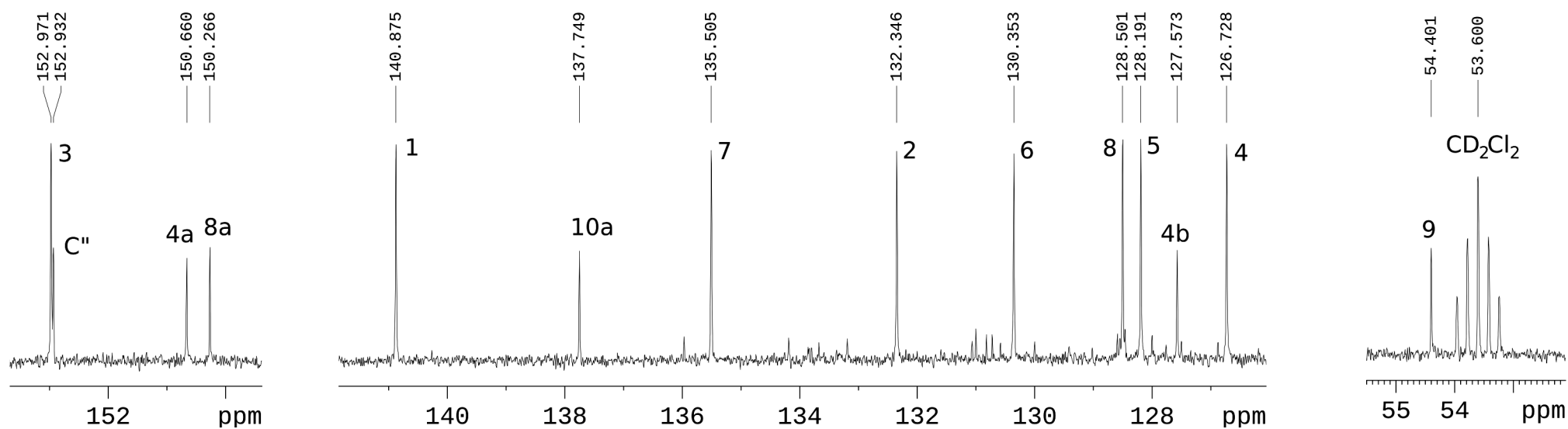
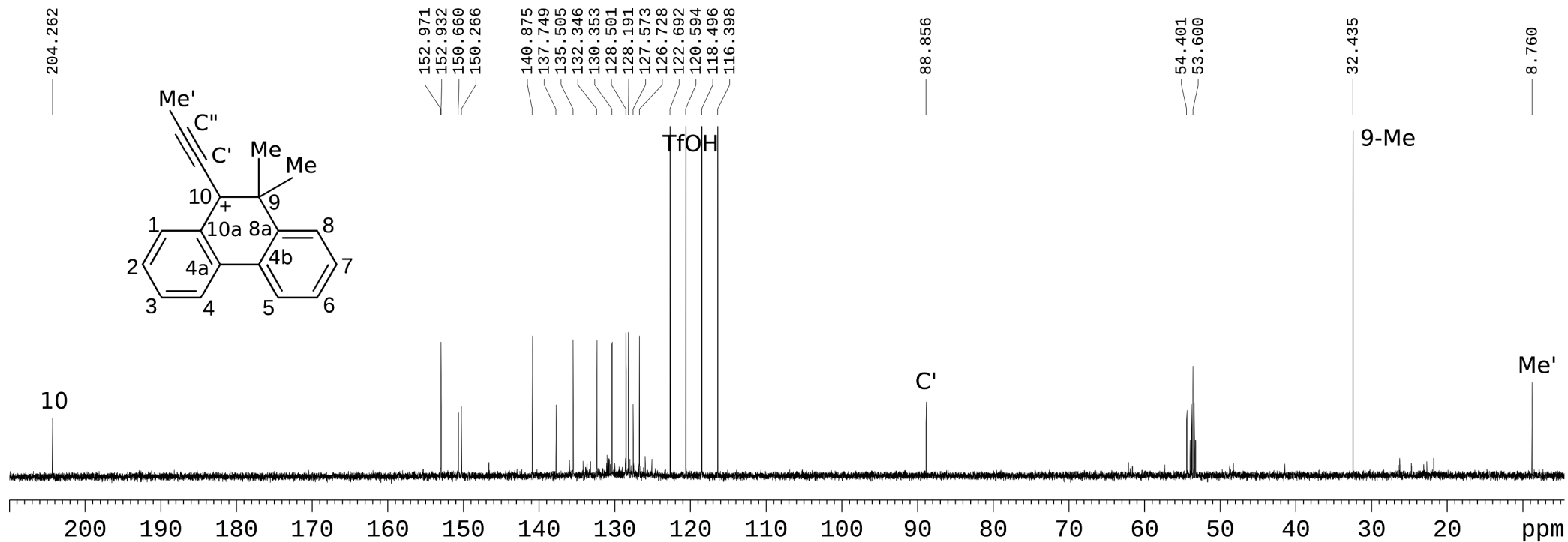
NOESY spectrum (aromatic region) of cations **1d** and **2d** in FSO₃H-SO₂ClF-CD₂Cl₂ at -22.5 °C (mixing time 0.5 s)



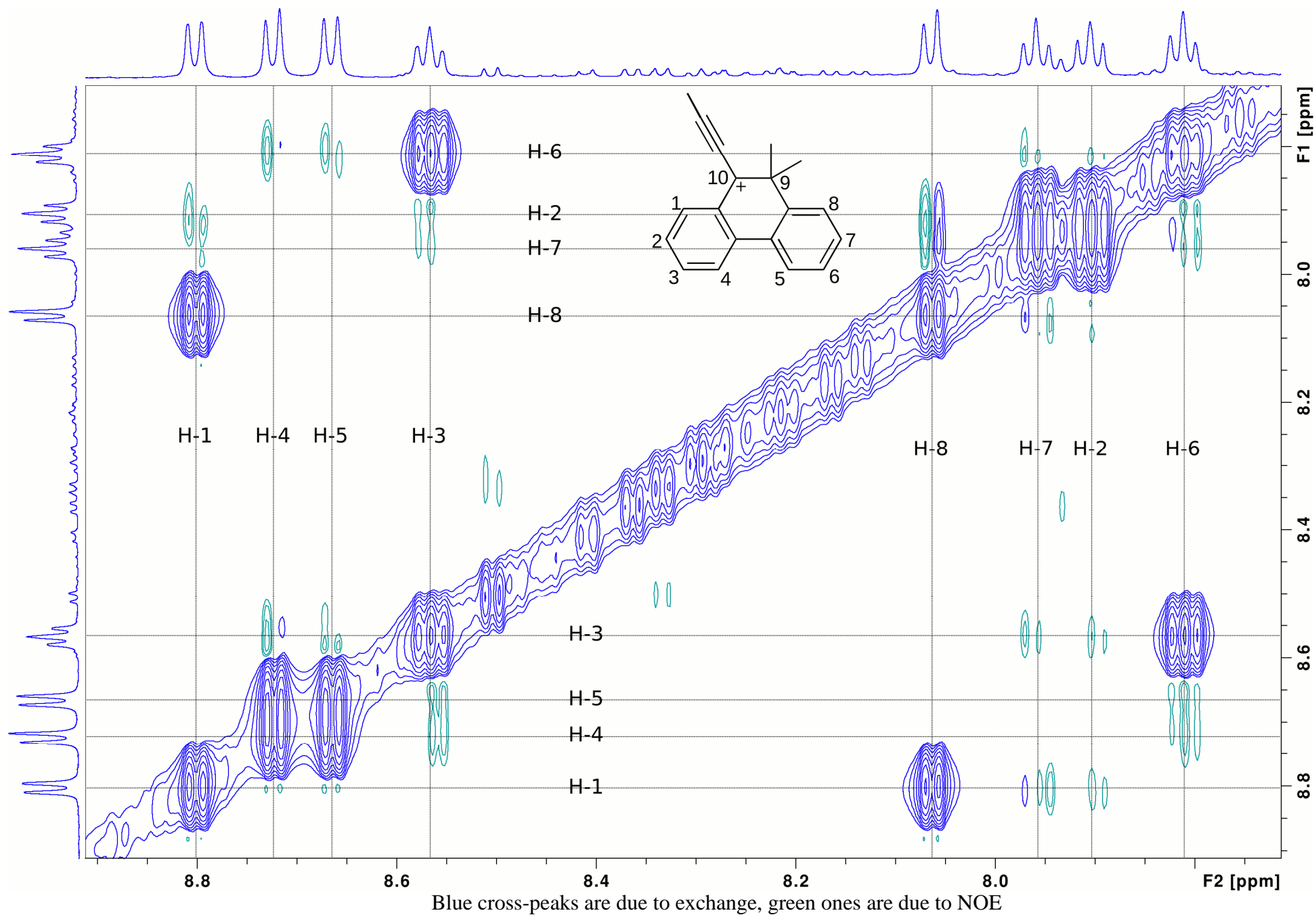
^1H NMR spectrum of cation **1b** in $\text{CF}_3\text{SO}_3\text{H}\text{-CD}_2\text{Cl}_2$ at $30\text{ }^\circ\text{C}$. Unpicked signals correspond to unidentified decomposition products.



^{13}C NMR spectrum of cation **1b** in $\text{CF}_3\text{SO}_3\text{H}\text{-CD}_2\text{Cl}_2$ at 30 °C. Unpicked signals correspond to unidentified decomposition products.



NOESY spectrum of cation **1b** in $\text{CF}_3\text{SO}_3\text{H}-\text{CD}_2\text{Cl}_2$ at 28 °C (aromatic region; mixing time 0.7 s)



Equilibrium constant of $1d \rightleftharpoons 2d$ ($K = 2d/1d$)

$y = ax+b$, where

$a = \Delta H$, kJ/mole; $b = \Delta S$, J deg.⁻¹ mole⁻¹; $y = R \cdot \ln(K)$; $x = -1000/T$

t, °C	T, K	I ₁	I ₂	$K=I_1/(I_2-I_1)*2$	$x=-1000/T$	$y=R \ln K$
-50.0	223.1	3	7.02	1.493	-4.482	3.329
-50.2	223.0	3	6.98	1.508	-4.484	3.413
-50.2	223.0	3	6.97	1.511	-4.484	3.433
-39.3	233.8	3	7.18	1.435	-4.276	3.005
-28.2	245.0	3	7.33	1.386	-4.082	2.712
-16.8	256.3	3	7.53	1.325	-3.901	2.336
-5.2	268.0	3	7.75	1.263	-3.731	1.942
-22.5	250.6	3	7.49	1.336	-3.990	2.410

I₁ – integral of **2d** 10-Me in ¹H NMR spectrum (see p. S2).

I₂ – integral of **2d** 9-Me and **1d** 9-Me in ¹H NMR spectrum (see p. S2).

R = 1.987*4.184 J/(mole*degree)

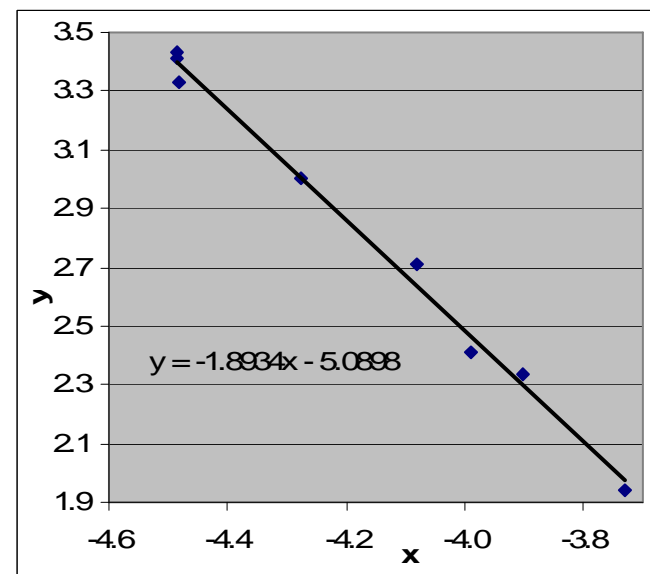
Parameters of the linear regression

ΔH , kJ	-1.893
stdEr ΔH	0.067
r^2	0.9924

ΔS , J deg. ⁻¹ mole ⁻¹	-5.090
stdEr ΔS	0.28
stdEr y	0.053

Extrapolation of equilibrium constant K to T = 177.6 K

T, K	ΔH , J	ΔS	$\Delta G = \Delta H - T \cdot \Delta S$	$K = \exp(-\Delta G/R/T)$
177.6	-1893	-5.090	-989	1.954



Rate constant of the reaction **1d** → **2d** at 177.6 K

$y = ax + b$, where

$a = k, s^{-1}$; $y = -K/(K+1) * \ln((K-\gamma)/(1+\gamma))$; $x = \text{time} - 1340850000$; $\gamma = 2d/1d$

$K = 1.954$ (equilibrium constant at 177.6 K)

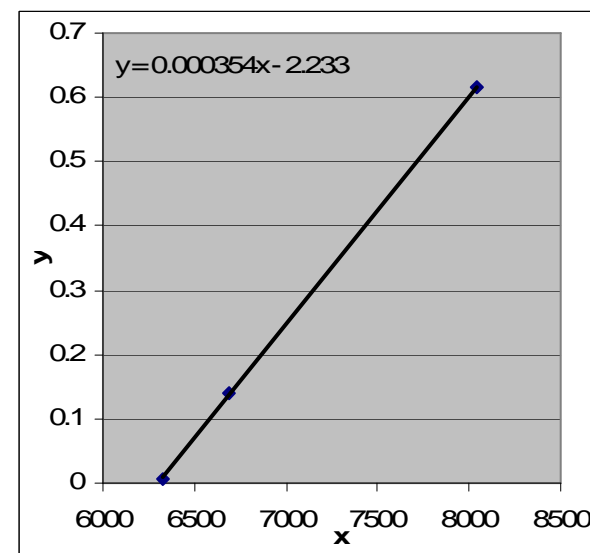
time, s	I ₁	I ₂	$\gamma = I_1/(I_2 - I_1) * 2$	x	y
1340856326	3	15.36	0.485	6326	0.0076
1340856693	3	12.48	0.633	6693	0.1402
1340858042	3	8.36	1.119	8042	0.6164

I₁ – integral of **2d** 10-Me in ¹H NMR spectrum (see p. S2).

I₂ – integral of **2d** 9-Me and **1d** 9-Me in ¹H NMR spectrum (see p. S2).

Parameters of the linear regression

k, s ⁻¹	0.0003544	r ²	0.999983
stdEr k	0.0000015	stdEr y	0.0019



Kinetics of the reaction **1d** → **2d**

The Eyring equation $\Delta G^\ddagger = -RT(\ln(k/T) - 23.76)$; $\Delta G^\ddagger = \Delta H^\ddagger - T\Delta S^\ddagger$

$y = ax + b$, where

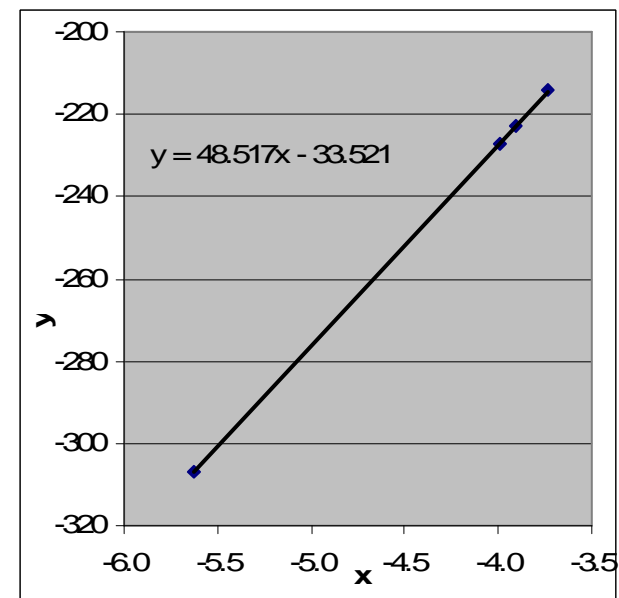
$a = \Delta H^\ddagger, \text{kJ}$; $b = \Delta S^\ddagger, \text{J deg}^{-1} \text{mole}^{-1}$; $y = R * (\ln(k/T) - 23.76)$; $x = -1000/T$

$R = 1.987 * 4.184 \text{ J/(mole*degree)}$; k – rate constant, s⁻¹

t, °C	T, K	k 1d → 2d	x	y	$\Delta G^\ddagger, \text{kJ}$
-95.5	177.6	0.000354	-5.631	-306.7	55.66
-22.5	250.6	7	-3.990	-227.3	58.21
-16.8	256.3	12	-3.901	-223.0	58.42
-5.2	268.0	36	-3.731	-214.2	58.68

Parameters of the linear regression

$\Delta H^\ddagger, \text{kJ}$	48.52	$\Delta S^\ddagger, \text{J deg}^{-1} \text{mole}^{-1}$	-33.52
stdEr ΔH^\ddagger	0.20	stdEr ΔS^\ddagger	0.86
r ²	0.999967	stdEr y	0.301



Kinetics of the reaction **1b** → **1b'**

Eyring equation $\Delta G^\ddagger = -RT(\ln(k/T)-23.76)$; $\Delta G^\ddagger = \Delta H^\ddagger - T\Delta S^\ddagger$

$y = ax+b$, where

$a = \Delta H^\ddagger$, kJ; $b = \Delta S^\ddagger$, J deg.⁻¹ mole⁻¹; $y = R*(\ln(k/T)-23.76)$; $x = -1000/T$

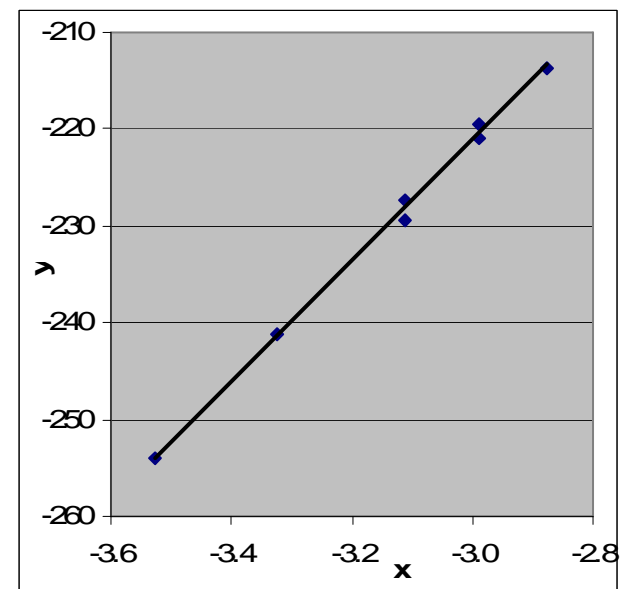
$R = 1.987*4.184$ J/(mole*degree); k – rate constant, s⁻¹

T, °C	T, K	k 1b → 1b'	x	y	ΔG^\ddagger , kJ
48.3	321.4	7	-3.112	-229.3	75.3
61.5	334.7	20	-2.989	-221.0	75.6
74.7	347.9	50	-2.876	-213.7	76.0
48.3	321.4	9	-3.112	-227.3	74.7
61.4	334.5	24	-2.991	-219.4	75.0
27.9	301.1	1.59	-3.323	-241.1	74.2
10.5	283.7	0.32	-3.527	-254.0	73.6

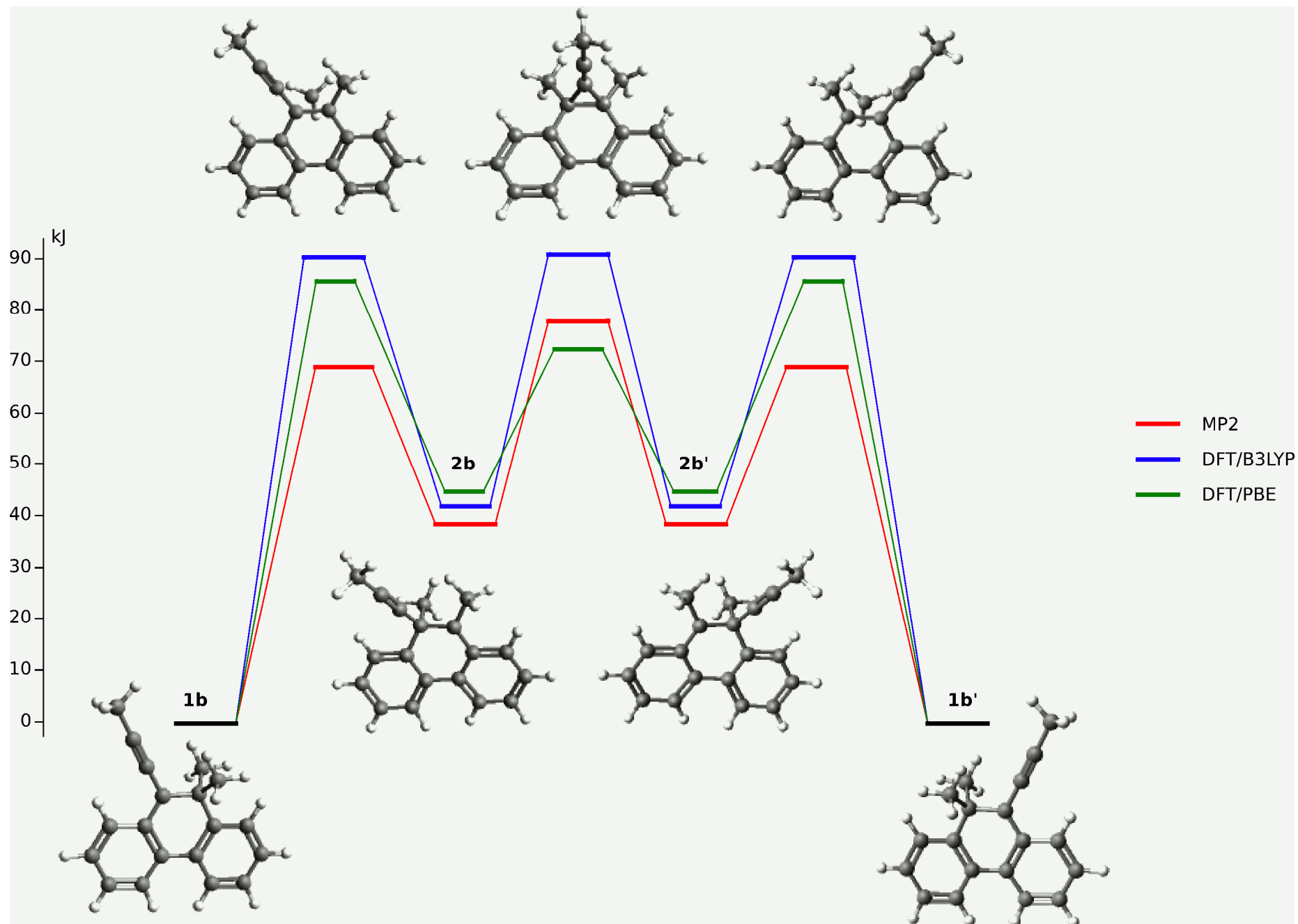
Parameters of the linear regression

ΔH^\ddagger , kJ	62.2
stdEr ΔH^\ddagger	1.6
r^2	0.9967

ΔS^\ddagger , J deg. ⁻¹ mole ⁻¹	-34.4
stdEr ΔS^\ddagger	5.0
stdEr y	0.87



Quantum chemical calculations (basis $\Lambda 1$ (cc-pVDZ), PRIRODA program)



Optimized coordinates of cation **1b** by RI-MP2/Å1

36
Energy -731.85703833
C 1.03529781 -1.50218756 0.01436759
C 1.91976671 -0.35443345 0.00132741
C 1.37449781 0.94771858 -0.01138719
C -0.10941955 1.19743949 -0.01206712
C -0.94117935 -0.04084083 0.00113101
C -0.39103268 -1.33105976 0.01396329
C 3.32270117 -0.51018587 0.00106512
C 4.15895654 0.58721119 -0.01125726
C 3.61728976 1.87513878 -0.02367666
C 2.24263500 2.04460898 -0.02365524
C -1.25835416 -2.46353347 0.02665149
C -0.74189476 -3.73059681 0.03949641
C 0.65573794 -3.90493206 0.04008702
C 1.51444416 -2.82300170 0.02789553
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C -0.50298935 1.99556581 -1.28854811
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H -1.39785889 -4.59385040 0.04921177
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H -1.57406616 2.20756602 -1.28487424
H 0.04253429 2.94054475 -1.30460994
H -0.24791344 1.42937299 -2.18808276
H -5.42077337 0.11461426 0.89636655
H -5.20041390 1.61831111 -0.02582251
H -5.42667806 0.07235632 -0.87424900

Optimized coordinates of cation **2b** by RI-MP2/Å1

36
Energy -731.84237326
C -2.01205654 1.75199969 -0.16886023
C -0.87934608 0.92991098 -0.05422888
C 0.39875373 1.58497028 0.06837211
C 0.46501685 3.01035411 0.14214837
C -0.67271049 3.76703450 0.05749401
C -1.91528557 3.12877559 -0.11577526
C -0.97992097 -0.51418307 -0.02533336
C 0.19619645 -1.28788805 0.00119546
C 1.54460888 -0.63240371 -0.12383956
C 1.57413729 0.83429342 0.10979604
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C -2.29153594 -2.55382971 0.07170348
C -1.11706753 -3.31246263 0.08579628
C 0.11544906 -2.67959264 0.04964641
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H -2.99108171 1.31151478 -0.30280600
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H -3.15084401 -0.61263312 0.03258342
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H 1.96670243 -1.75988855 -1.93121879
H 1.13364853 -0.21348044 -2.26420382
H 3.71094507 0.73865537 0.15074596
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H 5.08402822 -1.97367764 2.56007417
H 3.98128832 -3.34867456 2.71764809
H 5.07980207 -3.24844713 1.33053856

Optimized coordinates of TS **1b** → **2b** by RI-MP2/Λ1

36
Energy -731.83066803
C -0.74037083 1.27905607 0.05725436
C 0.71625954 1.29070859 0.05989948
C 1.46276619 2.48161797 -0.00602868
C 2.84532691 2.46267587 0.00879617
C 3.53941957 1.25077041 0.09265714
C 2.83576268 0.06418816 0.15872578
C 1.43276533 0.07282054 0.14388157
C -1.44839251 0.04704459 0.12634854
C -2.85741608 0.04414078 0.14243904
C -3.56413615 1.22833024 0.08977174
C -2.87555246 2.44325334 0.01412510
C -1.49323565 2.46461714 -0.00052428
C -0.72658858 -1.20085449 0.21061513
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H -0.99479639 3.42395148 -0.05337668
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H -0.73201765 -2.12545677 -1.80056051
H -0.01634415 -0.45168998 -1.99181854
C 2.79176183 -4.51872987 1.36879253
H 2.11954518 -5.27969782 1.77331272
H 3.34115474 -4.95321947 0.52925375
H 3.51103965 -4.24581941 2.14559234

Optimized coordinates of TS **2b** → **2b'** by RI-MP2/Λ1

36
Energy -731.82728662
C -1.92422871 1.71603690 0.52072075
C -0.82456675 0.90706048 0.18264479
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C 1.62916210 0.73968712 -0.28752420
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C -2.36151349 -2.51329491 -0.23160489
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C -0.00567957 -2.68416492 -0.66850030
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