Supplementary Information

Revitalizing the aromatic aza-Claisen rearrangement: implications for the mechanism of 'on-water' catalysis.

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General Experimental Details	 S2
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Claisen Rearrangement of	
Compounds 2b, 2c, 2d, 9a, 9b, 9c	
NMR spectra	 S6

General Experimental Details:

All solvents and reagents were used as received from commercial sources. Melting points were determined using a Stanford Research Systems Optimelt automated melting point system and are uncorrected. Infrared spectra were acquired neat on a Bruker Alpha-E ATR spectrometer. ¹H and ¹³C NMR spectra were recorded on a Bruker ASCEND 500 (¹H frequencies 500 MHz; ¹³C frequencies 125 MHz), a Bruker AVANCE DPX300 (¹H frequencies 300 MHz; ¹³C frequencies 75 MHz) or a Bruker AVANCE DPX200 (¹H frequencies 200 MHz; ¹³C frequencies 50 MHz). ¹H chemical shifts are expressed as parts per million (ppm) with residual chloroform (δ 7.26) as reference and are reported as chemical shift ($\delta_{\rm H}$); relative integral; multiplicity (s singlet, br broad, d doublet, t triplet, dd doublet of doublets, dt doublet of triplets, q quartet, m multiplet); and coupling constants (*J*) reported in Hz. ¹³C NMR chemical shifts are expressed as parts per million (ppm) with residual chloroform (δ 77.1) as internal reference and are reported as chemical shift ($\delta_{\rm C}$); multiplicity (assigned from DEPT experiments). High resolution mass spectra were recorded on a Bruker ApexII Fourier Transform Ion Cyclotron Resonance mass spectrometer with a 7.0 T magnet, fitted with an off-axis Analytical electrospray source.

Comparative Rate Plots for the Aza-Claisen Rearrangement of Compounds 2b, 2c, 2d, 9a, 9b, 9c:





















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LL.₽2 ——

24.58 77.00 77.42

70°.807 -----

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N-(2-methylbut-3-en-2-yl)naphthalen-1-amine

S7

ppr

2

20

30

6

50

09

2

8

6

100

110

120

130

160

150

160

170

180

190

200





S9







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τς8.	52	
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22.17

4-chloro-N-(2-methylbut-3-en-2-yl)aniline

