

**Tri-substituted hexahomotrioxacalix[3]arene derivatives
bearing imidazole units: synthesis and extraction properties
for cations and chromate anions**

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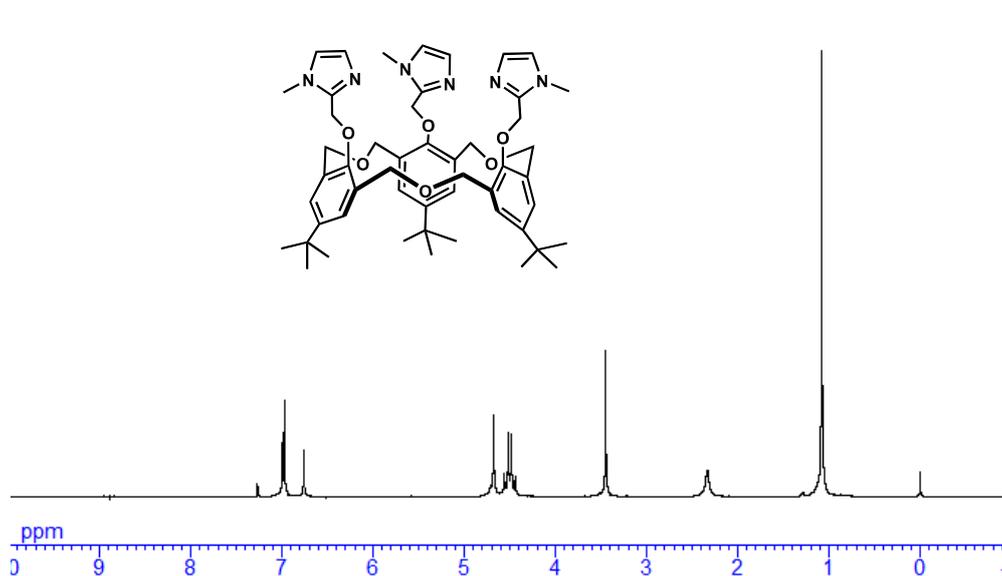


Figure S1. ¹H NMR spectrum of *cone-2* (CDCl₃, 300 MHz).

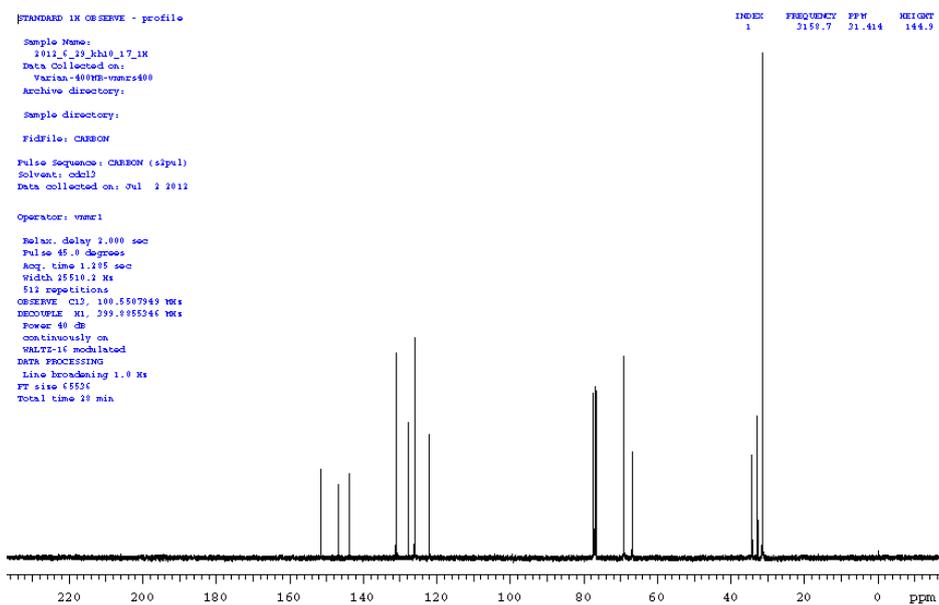


Figure S2. ¹³C NMR spectrum of *cone-2* (CDCl₃, 300 MHz).

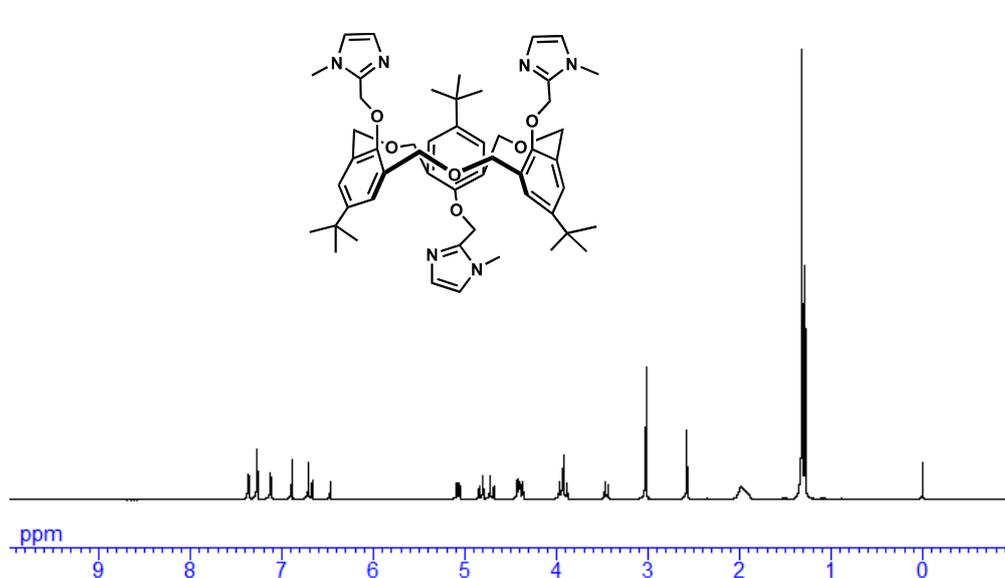


Figure S3. ^1H NMR spectrum of *partial-cone-2* (CDCl₃, 300 MHz).

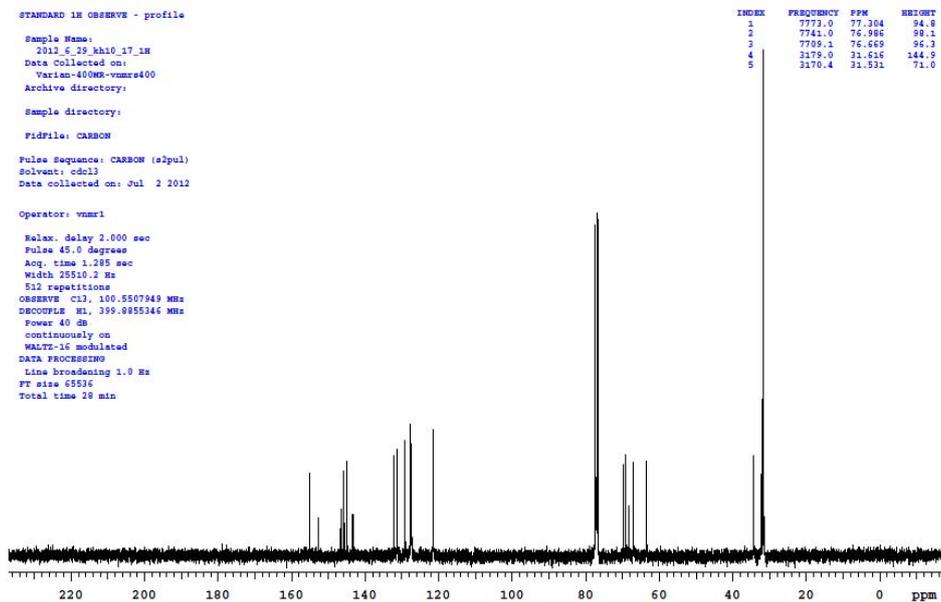


Figure S4. ^{13}C NMR spectrum of *partial-cone-2* (CDCl₃, 300 MHz).

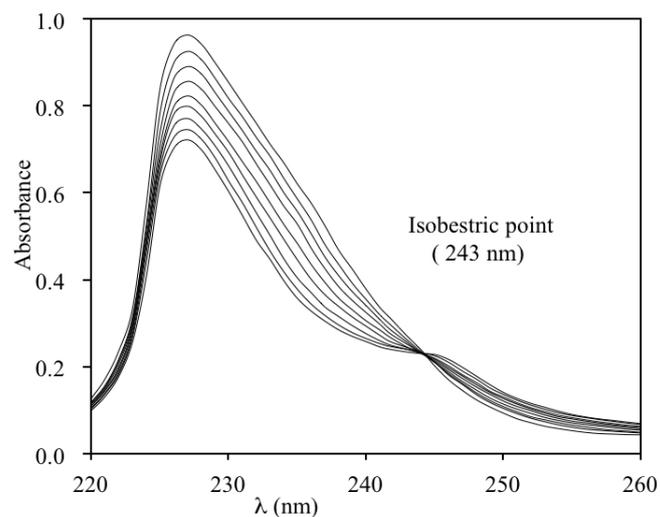


Figure S5. UV-vis spectral changes of *cone-2* on addition of $\text{Zn}(\text{ClO}_4)_2$ ($\text{THF}:\text{CH}_2\text{Cl}_2 = 1:50$), $[\text{cone-2}] = 1 \times 10^{-5} \text{ M}$; $[\text{Zn}(\text{ClO}_4)_2] = 0.1\text{--}2.5$ equiv.

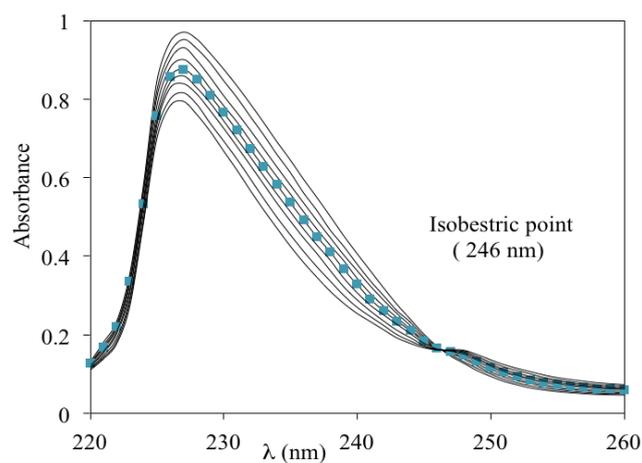


Figure S6. UV-vis spectral changes of *cone-2* on addition of $\text{Cu}(\text{ClO}_4)_2$ ($\text{THF}:\text{CH}_2\text{Cl}_2=1:50$), $[\text{cone-2}] = 1 \times 10^{-5} \text{ M}$; $[\text{Cu}(\text{ClO}_4)_2] = 0.1\text{--}2.5$ equiv.

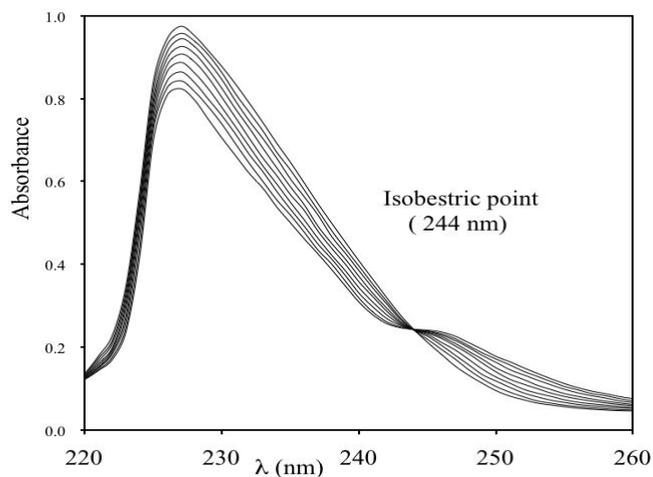


Figure S7. UV-vis spectral changes of *cone-2* on addition of $\text{Hg}(\text{ClO}_4)_2$ ($\text{THF}:\text{CH}_2\text{Cl}_2 = 1:50$), $[\text{cone-2}] = 1 \times 10^{-5} \text{ M}$; $[\text{Hg}(\text{ClO}_4)_2] = 0.1\text{--}2.5$ equiv.

Table S1. Extraction percentages of metal picrates with ligands^a

Ligand	Extracted metal cations										
	Li^+	Na^+	K^+	Cs^+	Co^{2+}	Ni^{2+}	Cu^{2+}	Zn^{2+}	Cd^{2+}	Ag^+	Hg^{2+}
<i>cone-2</i>	2	2	3	4	74	78	98	88	81	93	85
<i>p-cone-2</i>	1	1	2	3	61	63	75	76	69	84	75

^a Aqueous phase: $[\text{metal nitrate}] = 1 \times 10^{-2} \text{ M}$; $[\text{picric acid}] = 2.5 \times 10^{-2} \text{ M}$; organic phase: CH_2Cl_2 , $[\text{ligand}] = 1 \times 10^{-2} \text{ M}$; at 25°C , for 12 h.