

# Major mechanistic differences between the reactions of hydroxylamine with phosphate di- and tri-esters

*Michelle Medeiros<sup>†</sup>, Eduardo H. Wanderlind<sup>†</sup>, José R. Mora<sup>†</sup>, Raphaell Moreira<sup>†</sup>,  
Anthony J. Kirby<sup>\*,§</sup> and Faruk Nome<sup>\*,†</sup>.*

<sup>†</sup> Instituto Nacional de Ciência e Tecnologia (INCT) de Catálise, Departamento de Química, Universidade Federal de Santa Catarina (UFSC), CEP 88040-900, Florianópolis, Santa Catarina (SC), Brazil <sup>§</sup> University Chemical Laboratory, Cambridge CB2 1EW, UK.

\* Corresponding authors' email addresses: [faruk.nome@ufsc.br](mailto:faruk.nome@ufsc.br); [ajk1@cam.ac.uk](mailto:ajk1@cam.ac.uk)

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## 1. Reaction of TPP with hydroxylamine

### 1.1. Kinetic data

**Table S.1.** Values of  $k_{\text{obs}}$  as a function of pH in the reaction of **TPP** with hydroxylamine (0.5M),  $I=1.0$  (KCl), 25°C.

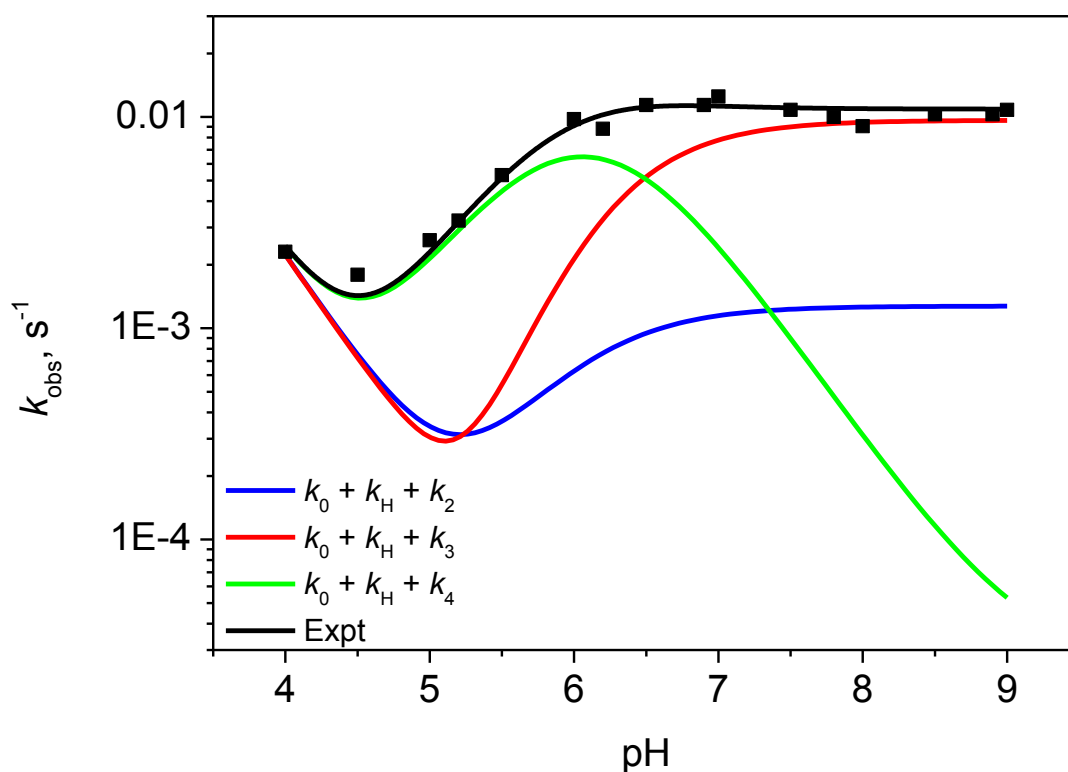
pH	$k_{\text{obs}}, \text{s}^{-1}$
4.0	$2.30 \times 10^{-3}$
4.5	$1.79 \times 10^{-3}$
5.0	$2.61 \times 10^{-3}$
5.2	$3.23 \times 10^{-3}$
5.5	$5.30 \times 10^{-3}$
6.0	$9.77 \times 10^{-3}$
6.2	$8.79 \times 10^{-3}$
6.5	$1.14 \times 10^{-2}$
6.9	$1.14 \times 10^{-2}$
7.0	$1.25 \times 10^{-2}$
7.5	$1.08 \times 10^{-2}$
7.8	$1.00 \times 10^{-2}$
8.0	$9.04 \times 10^{-3}$
8.5	$1.03 \times 10^{-2}$
8.9	$1.03 \times 10^{-2}$
9.0	$1.08 \times 10^{-2}$

#### S.1a. Curve fitting for the data of Table S.1

These data were fit to equation 2 of the full paper, using values of  $k_0$ ,  $k_2$  and  $k_3$  and  $pK_{\text{a}}$ s for  $\text{TPPH}^+$  and  $\text{NH}_3^+\text{OH}$  measured in earlier experiments. (Table 1).

$$k_{\text{obs}} = k_{\text{H}} \cdot \chi_{\text{TPPH}^+} + [k_0 + (k_2 [\text{NH}_2\text{OH}] \cdot \chi_{\text{NH}_2\text{OH}} + k_3 ([\text{NH}_2\text{OH}] \cdot (\chi_{\text{NH}_2\text{OH}}))^2 + k_4 ([\text{NH}_2\text{OH}]^2 \cdot (\chi_{\text{NH}_2\text{OH}}) \cdot (\chi_{\text{NH}_3\text{OH}^+}))^2] \cdot \chi_{\text{TPP}} \quad (2)$$

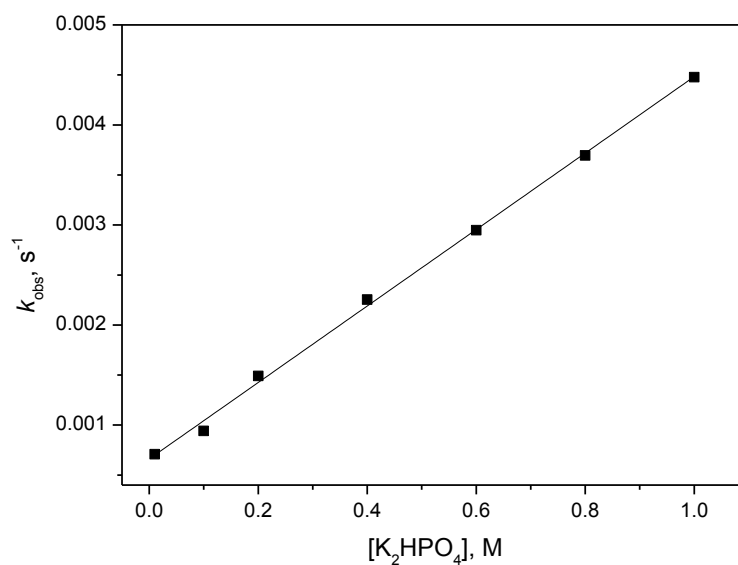
**Figure S.1a** shows how each of the various reactions contributes to the observed overall rate of release of 2-pyridone as the pH is varied. The coloured curves are calculated using the constants shown in Table 1 of the full paper, for the three separate reactions ( $k_2$ ,  $k_3$  and  $k_4$ ) catalyzed by one and two molecules of hydroxylamine reacting with **TPP** and **TPPH<sup>+</sup>** in the pH-range studied. The background hydrolysis rate is included in each case, showing how the reactions second order in hydroxylamine in particular dominate above pH 5, where significant amounts of the free base are present.



**Figure S.1a.** Breakdown of  $k_{\text{obs}}$  (black curve with data points) as a function of pH for the reaction of **TPP** with 0.5M  $\text{NH}_2\text{OH}$ , at 25°C and  $I = 1\text{M}$  (KCl). Values of  $k_{\text{H}}$  and  $k_4$  (full paper, Table 1) were obtained by iterative fit and the errors are presented in the Table. For the 16-point (black) curve Reduced Chisq and  $R^2$  are  $7.09645 \times 10^{-7}$  for 14 degrees of freedom, and 0.949, respectively.

**Table S.2.** Values of  $k_{\text{obs}}$  as a function of hydroxylamine concentration in the reaction with **TPP**,  $I=1.0$  (KCl), at pH 8.5 and 25°C.

$[\text{NH}_2\text{OH}], \text{M}$	$k_{\text{obs}}, \text{s}^{-1}$
0.05	$2.60 \times 10^{-4}$
0.1	$3.83 \times 10^{-4}$
0.15	$1.17 \times 10^{-3}$
0.2	$1.73 \times 10^{-3}$
0.3	$3.89 \times 10^{-3}$
0.4	$6.97 \times 10^{-3}$
0.5	$1.06 \times 10^{-2}$
0.6	$1.53 \times 10^{-2}$
0.7	$2.08 \times 10^{-2}$
0.8	$2.68 \times 10^{-2}$

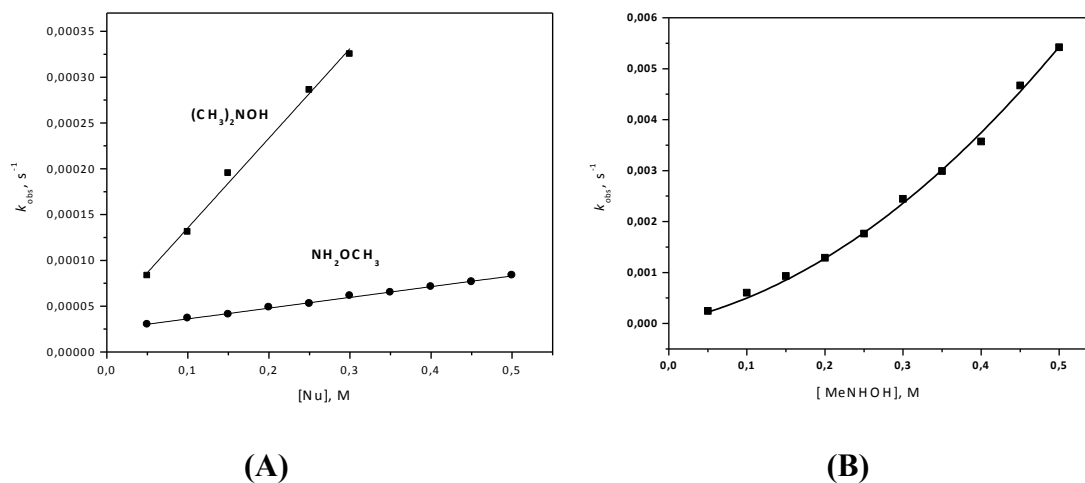


**Figure S.1.** Values of  $k_{\text{obs}}$  as a function of buffer [dipotassium hydrogen phosphate] concentration in the reaction of **TPP** with hydroxylamine 0.1M, at pH 8.5 and 25°C.

The linear fit gives a  $k_2$  value of  $3.82 \times 10^{-3} \text{ M}^{-1} \text{ s}^{-1}$ .

**Table S.3.** Values of  $k_{\text{obs}}$  as a function of nucleophile concentration for the reactions of TPP with hydroxylamine derivatives: all at  $I=1.0$  (KCl), pH 8.5 and 25°C.

[Nucleophile], M	$k_{\text{obs}}, \text{s}^{-1}$		
	MeNHOH	NMe <sub>2</sub> OH	NH <sub>2</sub> OMe
0.05	$2.47 \times 10^{-4}$	$8.32 \times 10^{-5}$	$3.00 \times 10^{-5}$
0.1	$6.03 \times 10^{-4}$	$1.31 \times 10^{-4}$	$3.68 \times 10^{-5}$
0.15	$9.32 \times 10^{-4}$	$1.95 \times 10^{-4}$	$4.09 \times 10^{-5}$
0.2	$1.29 \times 10^{-3}$	-	$4.87 \times 10^{-5}$
0.25	$1.76 \times 10^{-3}$	$2.86 \times 10^{-4}$	$5.28 \times 10^{-5}$
0.3	$2.45 \times 10^{-3}$	$3.25 \times 10^{-4}$	$6.14 \times 10^{-5}$
0.35	$2.99 \times 10^{-3}$	-	$6.41 \times 10^{-5}$
0.4	$3.57 \times 10^{-3}$	-	$7.11 \times 10^{-5}$
0.45	$4.67 \times 10^{-3}$	-	$7.65 \times 10^{-5}$
0.5	$5.42 \times 10^{-3}$	-	$8.35 \times 10^{-5}$



**Figure S.2.** Plots of  $k_{\text{obs}}$  vs. nucleophile concentration for the reactions of TPP with the hydroxylamine derivatives: (A)  $(\text{Me})_2\text{NOH}$  and  $\text{NH}_2\text{OMe}$  and (B)  $\text{MeNHOH}$ : all at  $I=1.0$  (KCl), pH 8.5 and 25°C. Lines in (A) represent linear fits, curve fit in (B) is to eq. 1 (in main article). The derived data appear in **Table 2** in the main article.

**Table S.4** Solvent deuterium isotope effect measurements

**Conditions:** T=25°C, Ionic strength = 1 M (KCl + NH<sub>2</sub>OH). [NH<sub>2</sub>OH]=0.5 M  
pH = 8.5 and pD = 8.5 (based on pD = pH<sub>rdg</sub> + 0.4); 0.01M TRIS buffer.

$$k_{\text{obs}} (\text{H}_2\text{O}) = 1.03\text{e-}2 \text{ s}^{-1}$$

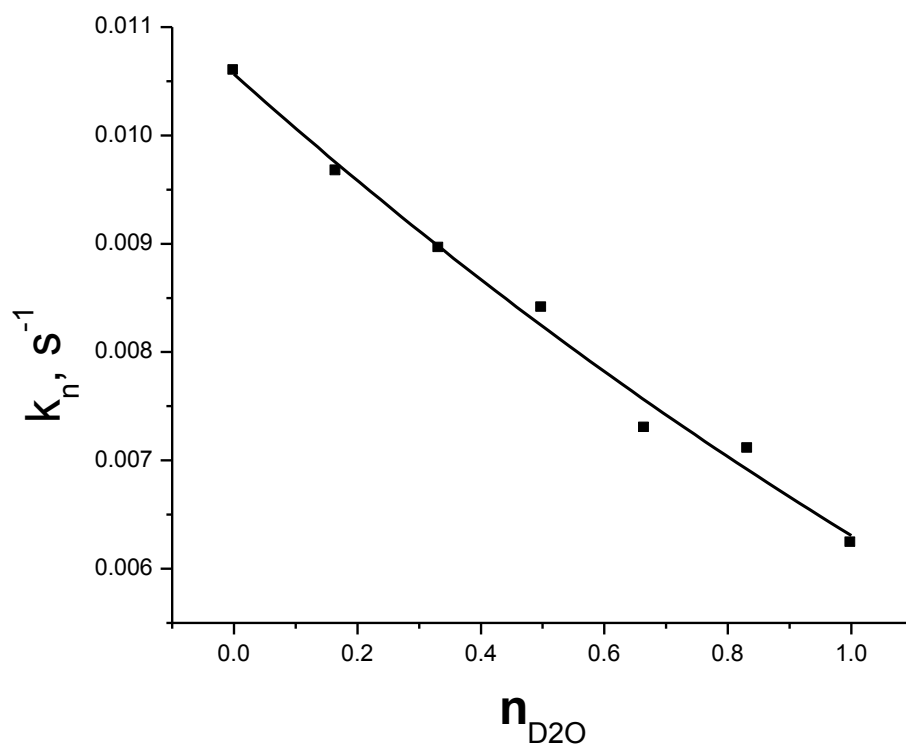
$$k_{\text{obs}} (\text{D}_2\text{O}) = 6.13\text{e-}3 \text{ s}^{-1} \text{ (mean of 3 measurements)}$$

$$\text{gives } k_{\text{H}_2\text{O}}/k_{\text{D}_2\text{O}} = 1.7$$

**Table S.4** Proton inventory for the reaction between hydroxylamine and TPP

Conditions: as above at T=25°C, 0.01M TRIS buffer, [NH<sub>2</sub>OH]=0.5 M  
Ionic strength = 1 M (KCl + NH<sub>2</sub>OH), pH=8.5 and pD=8.5 (based on pD = pH<sub>rdg</sub> + 0.4);.

V, H <sub>2</sub> O (mL)	V, D <sub>2</sub> O (mL)	Mols of H <sub>2</sub> O	Mols of D <sub>2</sub> O	n D <sub>2</sub> O	k <sub>n</sub> , s <sup>-1</sup>
-	2.0	0.000	0.110	1.000	6.24x10 <sup>-3</sup>
0.5	2.5	0.028	0.138	0.833	7.11x10 <sup>-3</sup>
1.0	2.0	0.055	0.110	0.666	7.30x10 <sup>-3</sup>
1.5	1.5	0.083	0.083	0.499	8.41x10 <sup>-3</sup>
2.0	1.0	0.111	0.055	0.333	8.96x10 <sup>-3</sup>
2.5	0.5	0.138	0.028	0.166	9.67x10 <sup>-3</sup>
2.0	-	0.111	0.000	0.000	1.06x10 <sup>-2</sup>



**Figure S.3a.** Proton inventory plot for the reaction between hydroxylamine and **TPP** at 25°C, pH/ pD= 8.5 and  $\mu = 1.0$  M.

**Fitting curve to**

$$k_n = k_H(1 - n + n \cdot \phi_i^T)^m$$

**Gives:**

$$R^2 = 0.9899$$

$$\phi_i = 0.87$$

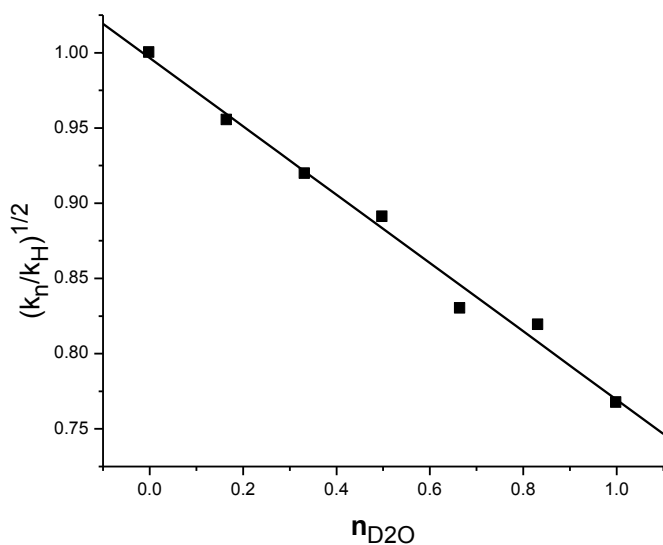
$$m = 3.6 \quad (\text{between 3 and 4 protons})$$

$$k_H = 0.0106 \pm 0.0002 \text{ M}^{-1} \text{ s}^{-1}$$



**A test specifically for two in-flight protons shows an equally reasonable fit.**

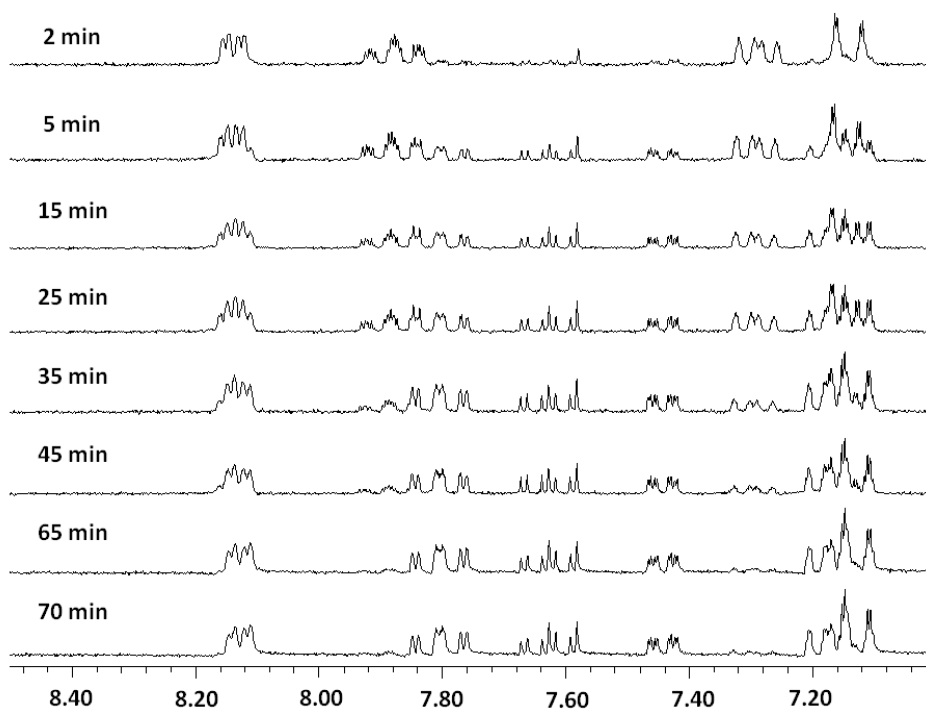
Since a plot of  $(k_n/k_H)^{1/2}$  versus  $n_{D2O}$  is linear:



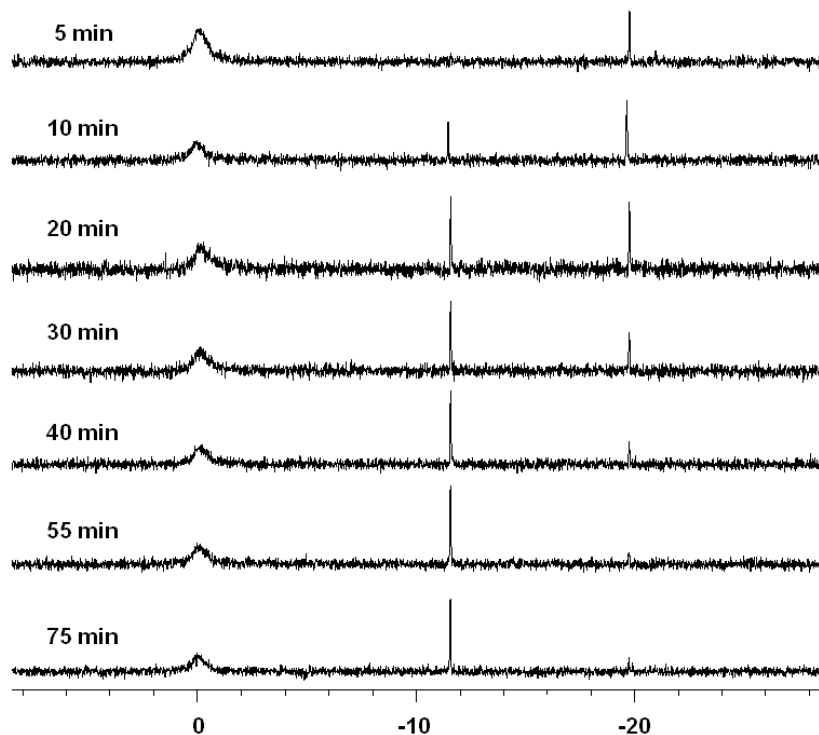
**Figure S.3b.** Proton inventory plot for the reaction between hydroxylamine and **TPP** with two mobile protons in the rate-determining transition state.

**Intercept =  $0.997 \pm 0.007$     Slope =  $-0.227 \pm 0.011$     R = 0.9943**

## 1.2. NMR data

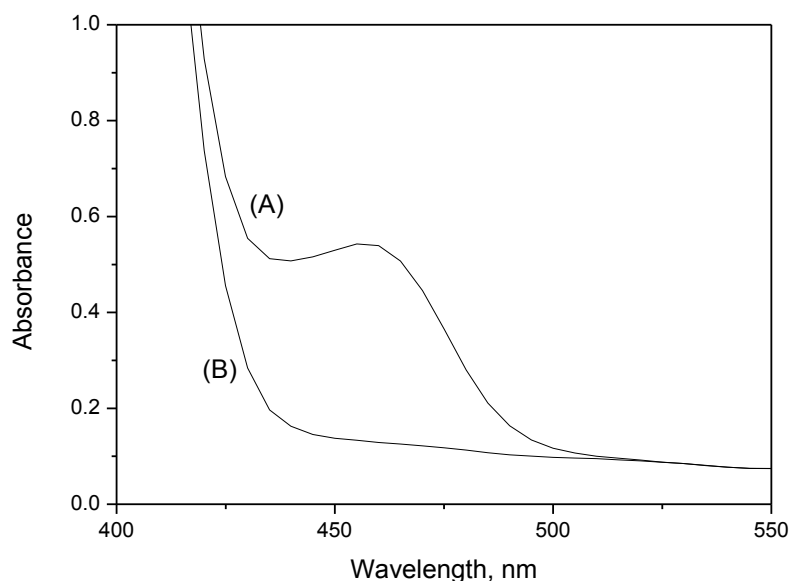


**Figure S.4.** Progressive <sup>1</sup>H NMR spectra for the reaction of **TPP** with hydroxylamine in presence of 0.3M of fumaric acid in D<sub>2</sub>O, at pD=7.2 and 25°C.



**Figure S.5.** Typical progressive <sup>31</sup>P NMR spectra for the reaction of **TPP** (0.02M) with hydroxylamine (0.1M) in D<sub>2</sub>O, at pD=7.2 and 25°C. H<sub>3</sub>PO<sub>4</sub> (85%) as external reference.

### 1.3. Hydrazine and diimine trapping



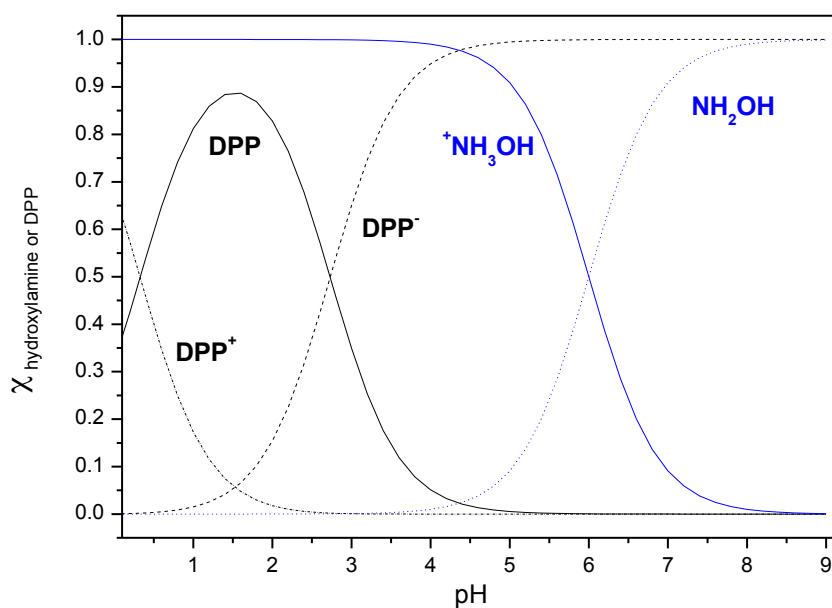
**Figure S.6.** UV-Vis spectra of solutions of 4-(dimethylamino)benzaldehyde and the product mixture of **TPP** reaction with hydroxylamine: **(A)** in the absence and **(B)** in presence of added fumaric acid. The band with maximum at 454nm in **(A)** is consistent with hydrazone formation.

## 2. Reaction of DPP with hydroxylamine

### 2.1. Kinetic data

**Table S.5.** Values of  $k_{\text{obs}}$  as a function of pH in the reaction of **DPP** with hydroxylamine (1M),  $I=1.0$  (KCl), 25°C.

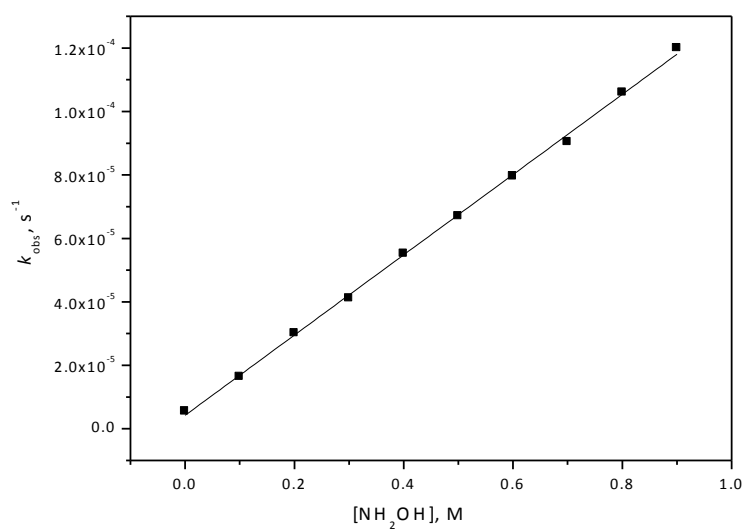
pH	$k_{\text{obs}}, \text{s}^{-1}$
5.0	$1.23 \times 10^{-4}$
5.5	$1.10 \times 10^{-4}$
6.0	$8.27 \times 10^{-5}$
6.3	$8.59 \times 10^{-5}$
6.6	$5.57 \times 10^{-5}$
7.0	$1.92 \times 10^{-5}$
7.6	$5.55 \times 10^{-6}$
8.0	$3.24 \times 10^{-6}$
8.5	$1.02 \times 10^{-6}$
9.0	$4.27 \times 10^{-7}$



**Figure S.7.** Calculated compositions of **DPP** and hydroxylamine as a function of pH. Values of  $pK_a$  used for **DPP** were:  $pK_{a1}=0.33$  and  $pK_{a2}=2.73$ , and for hydroxylamine value was 6.06.

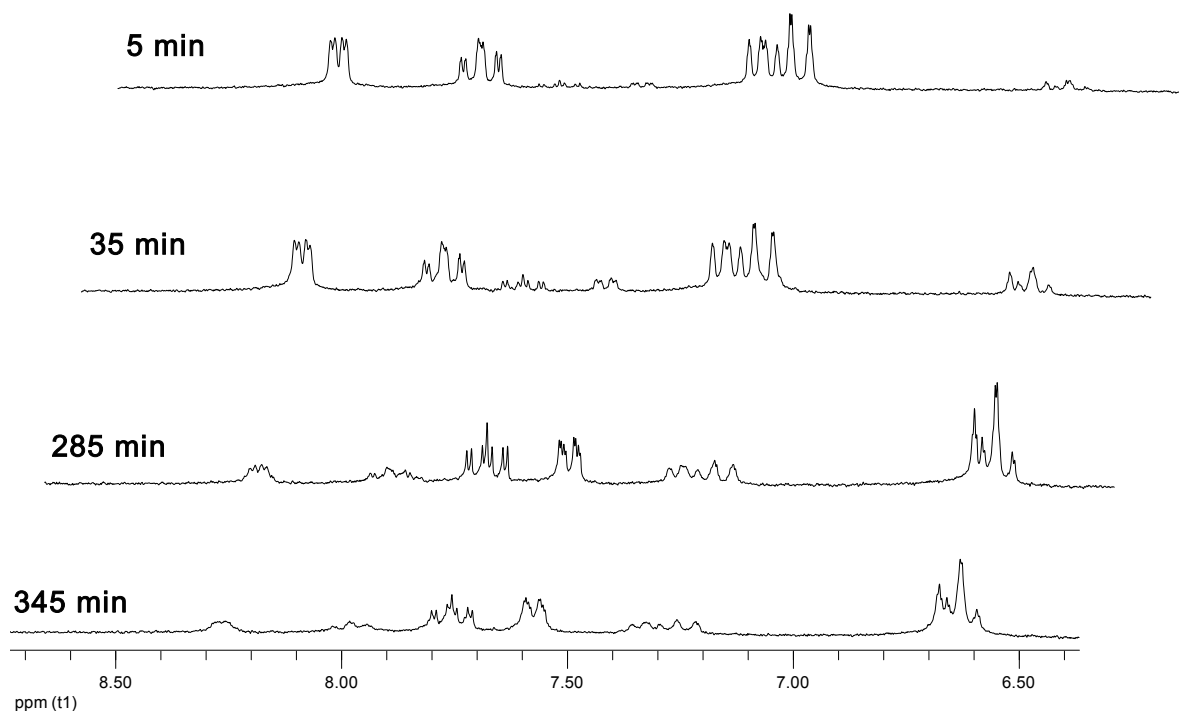
**Table S.6.** Values of  $k_{obs}$  as a function of hydroxylamine concentration in the reaction with **DPP**.

$[\text{NH}_2\text{OH}], \text{M}$	$k_{obs}, \text{s}^{-1}$
0	$5.50 \times 10^{-6}$
0.1	$1.64 \times 10^{-5}$
0.2	$3.01 \times 10^{-5}$
0.3	$4.11 \times 10^{-5}$
0.4	$5.52 \times 10^{-5}$
0.5	$6.70 \times 10^{-5}$
0.6	$7.96 \times 10^{-5}$
0.7	$9.04 \times 10^{-5}$
0.8	$1.06 \times 10^{-4}$
0.9	$1.20 \times 10^{-4}$

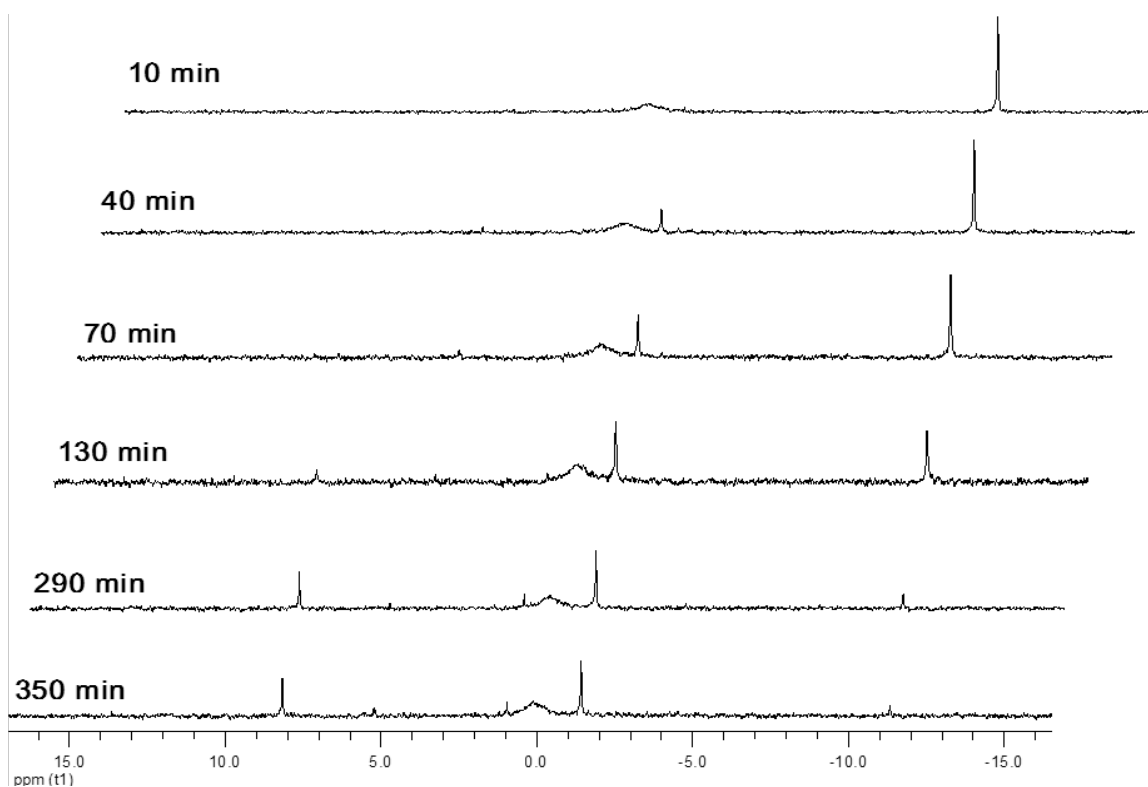


**Figure S.8.** Plot of  $k_{\text{obs}}$  vs.  $[\text{NH}_2\text{OH}]$  in the reaction of **DPP** at 25°C.

## 2.2. NMR data



**Figure S.9.** Progressive  $^1\text{H}$  NMR spectra for the reaction of **DPP** with hydroxylamine (1M) in  $\text{D}_2\text{O}$ , at pD 5.0 and 25°C.



**Figure S.10** Progressive  $^{31}\text{P}$  NMR spectra for the reaction of **DPP** with hydroxylamine (1M) in  $\text{D}_2\text{O}$ , at pD 5.0 and  $25^\circ\text{C}$ . Phosphoric acid (85%) was used as external reference. For assignments see the main text.

### 3. Theoretical calculations

#### 3.1. Reaction of TPP with hydroxylamine

##### 3.1.1 Cartesian coordinates for Mechanism 1 (TPP + NH<sub>2</sub>OH):

##### 3.1.1.1. Reactant

Atom	X	Y	Z
P	0.48017300	0.09257000	-0.18130200
O	-0.07396400	0.10100700	1.20406000
O	-0.18974000	1.14185500	-1.18819200
O	0.41393900	-1.30241700	-0.96511100
O	2.02937900	0.47121500	-0.34188700
C	-0.70875000	-2.15605000	-0.95775900
C	-1.66705600	-1.99831600	-1.95352400
C	-1.72112500	-3.97680100	-0.05706600
C	-2.71336400	-2.92013900	-1.96875200
H	-1.58757700	-1.19706900	-2.67860200
C	-2.74248800	-3.93084200	-1.00337700
H	-1.70695600	-4.74608000	0.70891500
H	-3.48959500	-2.85061300	-2.72401700
H	-3.53766400	-4.66798000	-0.98216500
C	3.13750700	-0.24692000	0.12496900
C	5.39905100	-0.48168900	-0.09569000
C	4.24113300	-1.83372300	1.50368200
C	5.43906200	-1.46568400	0.88668200
H	6.30110700	-0.15812200	-0.60585100
H	4.22606200	-2.59417700	2.27781200
C	-0.98181300	2.23720800	-0.79247300
C	-0.44342900	3.23917600	0.00959400
C	-1.28014800	4.30918600	0.32313300
H	0.57888000	3.18498700	0.36559800
C	-2.99462700	3.26860500	-0.98988800
C	-2.58342400	4.32504400	-0.18125200
H	-0.91531200	5.11809400	0.94793500
H	-3.99519000	3.24680200	-1.41042800
H	-3.26459600	5.13920500	0.04031000
C	3.05191500	-1.21227300	1.12525700
H	2.11111200	-1.46758500	1.59688900
O	-5.46237100	-0.87396600	0.80625000
H	6.38112800	-1.92791500	1.15909200
N	-2.20151000	2.22437100	-1.29993500
N	4.25771700	0.12320900	-0.47762800
N	-0.69908600	-3.09902500	-0.03204800
N	-4.27024900	-1.32911100	1.40898100
H	-3.73979800	-1.93418400	0.77088700
H	-3.66020400	-0.53099100	1.68001800
H	-4.46862100	-1.88006300	2.25160700
O	-2.63415000	0.93258500	2.13101100
H	-2.92101900	1.70440500	1.61970200
H	-1.75566700	0.70244400	1.76679800
O	1.21429700	1.03336100	3.61767500
H	0.82298100	0.74769900	2.77061600
H	1.46141800	1.95837500	3.47678500
O	4.27766700	2.02737500	-2.64815900
H	4.27212000	1.40226000	-1.88669300
H	4.88157800	2.73704300	-2.38766500

##### 3.1.1.2. Transition State (TS1)

Atom	X	Y	Z
P	-0.14898500	-0.12874200	0.49757400
O	-0.65599100	-0.44495800	1.92651800
O	0.80929400	-1.15787700	-0.37544800
O	-0.71200800	1.15919400	-0.37777600
O	-1.43744300	-1.04296900	-0.26933300
C	-0.14465600	2.41789100	-0.52853900
C	0.46697500	2.71705300	-1.74624100
C	0.13485100	4.53872800	0.26966200
C	0.93100000	4.01693300	-1.92907900
H	0.56726300	1.95320100	-2.50878900
C	0.76459600	4.95046700	-0.90035200
H	-0.01582900	5.23016700	1.09363400
H	1.41576000	4.29668600	-2.85919600
H	1.11298200	5.97240700	-1.00228700
C	-2.77345900	-0.90069600	-0.11173400
C	-4.81324400	-1.82025100	-0.69849900
C	-4.79262700	0.11924600	0.68980400
C	-5.52510000	-0.86132000	0.01180800
H	-5.32698800	-2.60786100	-1.24338200
H	-5.30039900	0.89167500	1.25990400
C	2.10150900	-1.55585200	-0.07690400
C	2.40454700	-2.18474800	1.13288600
C	3.70492700	-2.64820500	1.30738500
H	1.64625600	-2.30559800	1.89785000
C	4.22255000	-1.85074200	-0.88961400
C	4.63821100	-2.47944300	0.27818100
H	3.98433500	-3.14255000	2.23261100
H	4.90536100	-1.70277000	-1.72104100
H	5.65901400	-2.83172200	0.37605600
C	-3.40270900	0.11050300	0.63530900
H	-2.82711300	0.86314600	1.15532200
O	1.33498500	0.72424300	0.90868400
H	-6.60898900	-0.88219800	0.03358200
N	2.96772700	-1.39024300	-1.06960500
N	-3.46958300	-1.84639500	-0.76303000
N	-0.32831000	3.28582400	0.45943100
N	1.45333400	1.50556800	2.11591500
H	0.90883100	2.36237200	1.96668700
H	1.11382900	0.68158300	3.31828700
H	2.43906900	1.77390100	2.10221200
O	0.61017000	-0.05339900	3.91937800
H	1.23854300	-0.76502500	4.12663800
H	-0.02706200	-0.35038500	2.99122000
O	-1.83332100	-3.72022000	-2.13912300
H	-0.93364200	-3.38922700	-2.00437800
H	-2.40805800	-3.06903100	-1.66221600
O	2.40401000	-0.36709400	-3.66347800
H	2.11939900	0.55394100	-3.57903000
H	2.54824100	-0.67759600	-2.73609100

### 3.1.1.3. Intermediate

Atom	X	Y	Z
P	-0.21754000	-0.31768900	1.01844800
O	-0.42902700	-0.16484300	2.59497500
O	-1.31330900	0.08215500	-0.13491200
O	1.30639900	-0.75908400	0.55697000
O	0.34419900	1.34271100	0.95898000
C	1.69562300	-1.79805600	-0.27008400
C	2.79779400	-2.55534400	0.13564600
C	1.55999800	-2.85413200	-2.28665000
C	3.28821500	-3.50103300	-0.76054300
H	3.24511200	-2.39426600	1.10970200
C	2.65878200	-3.65777700	-2.00060500
H	1.03821900	-2.94372000	-3.23520300
H	4.14846100	-4.10740100	-0.49426700
C	0.20351100	2.20310700	-0.09163700
C	1.07615300	3.12512600	-2.01125900
C	-1.01280700	3.95458100	-1.18923700
C	-0.00052400	3.99148900	-2.15610800
H	1.89374100	3.12807800	-2.72691100
H	-1.86847300	4.61960200	-1.25709100
C	-2.60984600	-0.35917500	-0.33320200
C	-2.95735500	-0.74129100	-1.62962600
C	-4.29630400	-1.03813400	-1.86933600
H	-2.20557200	-0.78718400	-2.40874500
C	-4.74312600	-0.57000700	0.43687300
C	-5.21256900	-0.95387500	-0.81576800
H	-4.61839000	-1.33181300	-2.86357200
H	-5.42054700	-0.49180600	1.28234000
C	-0.91166900	3.05086300	-0.13823300
H	-1.67154300	2.97816600	0.63142600
O	-0.85140000	-1.88723500	1.00248100
H	-0.23195000	0.74547300	2.97346600
O	0.05031200	2.21081800	3.68531100
H	-0.74349800	2.56764000	4.11336000
H	0.16974200	2.74355200	2.88255200
H	-0.04000500	4.67973800	-2.99326400
N	1.18425600	2.24007800	-0.99815000
H	-6.26401300	-1.17664200	-0.96071700
N	-3.45481100	-0.26136900	0.68445900
H	3.00999300	-4.38271300	-2.72672700
N	1.08254300	-1.92090100	-1.43983000
N	-0.28255430	-2.72712460	1.87803798
H	-0.99344030	-2.83380860	2.60412398
H	-0.27464330	-3.60530460	1.35743798
O	3.77305600	1.15934400	-0.91588000
H	2.81746200	1.43278100	-0.91455200
H	3.76536200	0.19150500	-0.88424500
O	4.85471200	2.13548700	1.44613500
H	4.44680300	1.77006600	0.62616800
H	4.16030500	2.07708900	2.11749500

### 3.1.1.4. Transition State (TS2)

Atom	X	Y	Z
P	-0.30152400	-0.56567000	1.10568200
O	-0.46064500	-0.21926000	2.60972300
O	-1.25982800	0.03015100	-0.02943300
O	1.22134900	-0.82046300	0.66050600
O	0.49156400	1.56948500	0.97800100
C	1.66634400	-1.80696500	-0.23195200
C	2.78843500	-2.53343200	0.16113400
C	1.54401700	-2.81367500	-2.26496300
C	3.29316500	-3.45261800	-0.75587700
H	3.23346200	-2.37966800	1.13710400
C	2.65931400	-3.60019000	-1.99415500
H	1.02105700	-2.89469600	-3.21310100
H	4.16670600	-4.04581500	-0.50464900
C	0.28620700	2.34964800	-0.07083700
C	1.04787400	3.19080000	-2.10490400
C	-1.05530900	3.96736200	-1.27918800
C	-0.07031100	3.99521700	-2.27896900
H	1.84342200	3.18109600	-2.84656000
H	-1.94739900	4.58158200	-1.36711400
C	-2.59092400	-0.34439200	-0.27591400
C	-2.91172500	-0.68323500	-1.58721100
C	-4.25023800	-0.96664600	-1.85084100
H	-2.14844300	-0.71782000	-2.35501000
C	-4.72984600	-0.54620900	0.45977200
C	-5.17987300	-0.90228800	-0.80828000
H	-4.55951000	-1.23700400	-2.85546600
H	-5.41815600	-0.47957000	1.29685900
C	-0.88083400	3.14688000	-0.17502100
H	-1.62134200	3.09626000	0.61647000
O	-0.93302000	-2.07651600	1.06656500
H	-0.27908900	0.74061000	3.00303800
O	0.01384100	2.08469100	3.45045600
H	-0.80493000	2.56025300	3.66080800
H	0.21219500	2.26812600	2.48873300
H	-0.16688700	4.62232400	-3.15853000
N	1.23384000	2.38722800	-1.03745900
H	-6.22957000	-1.11875000	-0.97341300
N	-3.44097800	-0.25824800	0.73135000
H	3.02092000	-4.30656700	-2.73322600
N	1.04877700	-1.91054300	-1.39470700
N	-0.30949800	-2.97966100	2.03709800
H	-1.08605600	-3.20328600	2.66278700
H	-0.14912800	-3.81462200	1.47074100
O	3.71767600	1.21810400	-0.91184300
H	2.77940400	1.57729700	-0.93046300
H	3.62864300	0.25448200	-0.88135600
O	4.83979800	2.12133100	1.44484000
H	4.41655100	1.77873900	0.62104000
H	4.14236800	2.08894400	2.11482900



### 3.1.1.5 Product

Atom	X	Y	Z
P	-0.42418600	-0.95222500	1.21881500
O	-0.50933900	-0.34156100	2.57335800
O	-1.21739500	-0.21619300	0.02907200
O	1.10344000	-0.94888000	0.71301800
O	1.08118600	2.28438900	1.05739800
C	1.64592300	-1.83744500	-0.23150600
C	2.80025600	-2.51700000	0.15115200
C	1.60608300	-2.77555600	-2.30087500
C	3.36207300	-3.38125600	-0.78640700
H	3.22199100	-2.37588500	1.13935000
C	2.75244400	-3.51910900	-2.03752600
H	1.10278600	-2.84875800	-3.26029500
H	4.25881200	-3.94100300	-0.54016400
C	0.57205400	2.81606500	-0.07503000
C	0.92270300	3.17239900	-2.32528000
C	-1.06196400	4.04899000	-1.32110500
C	-0.25935300	3.88915100	-2.45929400
H	1.57759000	3.00996300	-3.17704900
H	-1.99841300	4.59583900	-1.37896800
C	-2.57824100	-0.41058700	-0.25243000
C	-2.91236700	-0.63992500	-1.58480800
C	-4.26401100	-0.81521900	-1.87165600
H	-2.14576500	-0.68482900	-2.34881500
C	-4.73443000	-0.50512400	0.45848400
C	-5.19616800	-0.75187800	-0.83084400
H	-4.58239700	-1.00408600	-2.89191900
H	-5.42384400	-0.44291400	1.29501300
C	-0.65287000	3.50462600	-0.10998500
H	-1.24508000	3.60976800	0.79255800
O	-0.99354600	-2.45437800	1.06478200
H	-0.41207500	1.35335900	3.25880700
O	-0.19325400	2.29190700	3.43720700
H	-1.04591300	2.74838300	3.49758400
H	0.51046300	2.44728300	1.85316000
H	-0.54502800	4.30370900	-3.41932400
N	1.33887200	2.64243200	-1.15801100
H	-6.25645800	-0.88780400	-1.01347800
N	-3.43037800	-0.33093200	0.75447200
H	3.15594000	-4.18492500	-2.79248900
N	1.05099300	-1.93202900	-1.40751500
N	-0.45624539	-3.27679103	2.01968768
H	-1.28606739	-3.58786903	2.52851868
H	-0.14878439	-4.07026403	1.45403868
O	3.75687300	1.24962700	-0.96867300
H	2.87570500	1.71255000	-1.02999100
H	3.54782400	0.30657100	-0.89200400
O	4.82755100	2.11974700	1.43717600
H	4.42555700	1.79436400	0.59714000
H	4.09321100	2.14884900	2.06678400
H	3.54782400	0.30657100	-0.89200400
O	4.82755100	2.11974700	1.43717600
H	4.42555700	1.79436400	0.59714000

### 3.1.2. Cartesian coordinates for Mechanism 2 (TPP + NH<sub>2</sub>OH):

#### 3.1.2.1 Reactant

#### 3.1.2.2. Transition State (TS4)

Atom	X	Y	Z	Atom	X	Y	Z
P	-0.40844600	-0.35924600	-0.52060900	P	-0.03672100	-0.57344300	0.18269800
O	0.67273600	0.20798300	-1.36851200	O	-0.54979200	-0.73172900	1.60976600
O	-0.80182500	0.46341100	0.81192500	O	1.05133600	-1.64779200	-0.46061300
O	-0.10785100	-1.80395900	0.11555300	O	-0.54836500	0.54802700	-0.92941900
O	-1.74481500	-0.61182800	-1.36304200	O	-1.19889500	-1.66655000	-0.58270400
C	1.17231600	-2.22121100	0.52766800	C	-0.22526900	1.89367000	-1.04521200
C	2.94193400	-2.24320600	2.10929300	C	0.63375600	2.27532300	-2.07808000
C	2.95701100	-3.56109600	0.10598100	C	-0.62823500	4.05204200	-0.42770700
C	3.59115200	-3.16179800	1.28022600	C	0.85833500	3.63686600	-2.25876100
H	3.39784500	-1.91172900	3.03664400	H	1.09981700	1.52569900	-2.70659500
H	3.42287100	-4.28051500	-0.56039800	C	0.21877900	4.54965700	-1.41277100
H	4.56397100	-3.56579500	1.53818300	H	-1.15270300	4.72585100	0.24379800
C	-3.04362200	-0.91851600	-0.92598000	H	1.52005800	3.98023300	-3.04795800
C	-3.29692300	-1.55350400	0.28824400	H	0.36676300	5.61902800	-1.51761900
C	-4.63065200	-1.82774600	0.58836200	C	-2.55013900	-1.59247500	-0.50915400
H	-2.50117300	-1.83404900	0.96604800	C	-4.51103400	-2.61987800	-1.15136200
C	-5.24324600	-0.84736600	-1.50605900	C	-4.64998300	-0.62313900	0.14406900
C	-5.62753700	-1.47059400	-0.32232700	C	-5.30295200	-1.66337100	-0.52478100
H	-4.88080700	-2.32123300	1.52212900	H	-4.96858200	-3.45023800	-1.68318600
H	-5.98140300	-0.55543100	-2.24662800	H	-5.21734200	0.15105600	0.65275400
C	-1.15753000	1.81381500	0.84596500	C	2.43622600	-1.62200700	-0.37179900
C	-1.75958600	3.72409500	-0.22978100	C	3.05603900	-2.23352700	0.71957700
C	-1.61013200	3.73289300	2.16273900	C	4.44805500	-2.29173800	0.72209700
C	-1.87753500	4.41605200	0.97052200	H	2.46098300	-2.64748700	1.52621100
H	-1.95559800	4.21340300	-1.17911600	C	4.42368300	-1.18522300	-1.40170200
H	-1.68972400	4.23736800	3.12035700	C	5.15142600	-1.75364200	-0.35938400
H	-2.17023800	5.46009600	0.97157000	H	4.97286000	-2.75656400	1.55095200
C	1.69087800	-1.75603700	1.73260700	H	4.93142200	-0.76779900	-2.26653500
H	1.14081200	-1.04589800	2.33855400	H	6.23492400	-1.78008600	-0.40036400
C	-1.23653200	2.39311400	2.11111200	C	-3.25988600	-0.57708400	0.15907600
H	-1.01422100	1.81400700	2.99969800	H	-2.74347300	0.21748700	0.67799800
H	-6.67456400	-1.67083800	-0.12370700	O	1.29518500	0.56727300	0.58485300
N	-1.40295900	2.42481000	-0.29906800	H	-6.38483600	-1.73214400	-0.55788600
N	-3.96204600	-0.56722900	-1.81106800	N	3.07866200	-1.11930000	-1.41990300
N	1.75048900	-3.09840300	-0.27566600	N	-3.16603400	-2.59733700	-1.15117600
O	3.80808000	1.07347600	0.42986200	N	-0.86207000	2.73717300	-0.24345600
N	3.58826900	0.23642600	-0.69674600	N	1.68208500	0.66926900	1.95218800
H	3.86912800	-0.72835300	-0.49025900	H	2.57615200	1.16600900	1.91994600
H	2.58901100	0.22435000	-0.95250700	H	0.85277000	1.28327400	2.66176800
H	4.16246700	0.60441200	-1.49481900	H	1.86546600	-0.26627500	2.32585600
O	5.32887400	1.75775200	-2.22386200	O	0.07492500	1.92824100	3.39396800
N	5.54370700	2.54085600	-1.05963800	N	-1.16707100	1.25601000	3.37937400
H	4.94429700	2.14558900	-0.28236700	H	-1.91237400	1.90611100	3.11062600
H	6.53007900	2.51970500	-0.78106600	H	-1.11876400	0.46789700	2.68760000
H	5.29015300	3.52033900	-1.22470500	H	-1.37928900	0.88629500	4.31152800

### 3.1.2.3 Intermediate

### 3.1.2.4. Transition State (TS5)

Atom	X	Y	Z	Atom	X	Y	Z
P	-0.16470200	-0.40895500	0.35044200	P	0.47937300	-0.24424700	-0.78386900
O	0.54534700	-0.90406500	1.69885900	O	-0.15709200	-0.93861300	-2.02237200
O	0.23681600	1.00152100	-0.39903000	O	-0.35149000	0.76652300	0.13444400
O	-1.20180700	-1.48140600	-0.35518500	O	1.53334800	-1.18631700	-0.02992300
O	1.00673300	-1.18635600	-0.68169500	O	-0.86088000	-1.70713700	0.44104200
C	-2.57625500	-1.38737900	-0.52091900	C	2.75472300	-0.78984300	0.53604600
C	-3.35087100	-2.43972300	-0.02872600	C	3.83444900	-1.64294800	0.31913600
C	-4.36931700	-0.35348000	-1.47510000	C	3.94391400	0.62633600	1.85374800
C	-4.71545800	-2.41114100	-0.29891300	C	5.03514700	-1.30076400	0.93586500
H	-2.88992500	-3.23786200	0.54123400	H	3.72779600	-2.52393700	-0.30256400
C	-5.24182100	-1.34305600	-1.03438000	C	5.09550200	-0.14205300	1.71762300
H	-4.73403200	0.49037800	-2.05361900	H	3.94521800	1.53243400	2.45203800
H	-5.35848000	-3.20806400	0.06156100	H	5.91068400	-1.92850000	0.80432700
C	2.26257100	-0.75832400	-1.01193500	C	-2.02078200	-1.33592900	0.96771900
C	3.89007600	-0.82319200	-2.63290900	C	-3.32604600	-1.05857200	2.86402200
C	4.36011600	0.32705100	-0.59443200	C	-4.29762000	-0.53093700	0.74773300
C	4.77916000	-0.07829900	-1.86552300	C	-4.42741800	-0.61204200	2.14205900
H	4.17251300	-1.17007600	-3.62330000	H	-3.38199100	-1.13959900	3.94764800
H	5.01829100	0.90220400	0.05008600	H	-5.12745500	-0.19048100	0.13382200
C	-0.04551200	2.30710900	-0.00297200	C	-0.64123000	2.11036800	-0.16819900
C	-0.86235100	3.07336700	-0.83445300	C	-0.31268500	3.05211300	0.80204400
C	-1.04850800	4.41042900	-0.49170300	C	-0.67953600	4.37119300	0.54231400
H	-1.32370200	2.63317000	-1.71031800	H	0.20095000	2.75791000	1.70939500
C	0.38575400	4.05427900	1.39340300	C	-1.61435700	3.63524400	-1.53910700
C	-0.41639000	4.91480900	0.64918900	C	-1.34044700	4.67279000	-0.65236100
H	-1.67656100	5.04741200	-1.10675300	H	-0.44967200	5.15019200	1.26214400
H	0.90243700	4.40725200	2.28134300	H	-2.12838000	3.82431700	-2.47665800
C	3.08795900	-0.01845400	-0.15011500	C	-3.09652800	-0.89365400	0.15561200
H	2.74841600	0.27022000	0.83624600	H	-2.96472900	-0.84607200	-0.92049300
O	-1.35323000	0.43710800	1.20689700	O	1.40763700	0.89708600	-1.48673000
H	1.23315700	-1.62413300	1.59329600	H	-0.51561600	-1.91346300	-1.92802500
O	2.31267600	-2.88464900	1.69370700	O	-0.93324600	-3.29168900	-1.60833300
H	2.66761900	-3.08782700	0.81395400	H	-1.08702900	-2.97427500	-0.67392600
H	5.76341500	0.16945800	-2.24776000	H	-5.34748800	-0.33985700	2.64842000
N	2.65125700	-1.15932400	-2.22683700	N	-2.14773500	-1.41015700	2.31339200
H	-0.53616100	5.94914800	0.95262100	H	-1.64015300	5.68722100	-0.89128500
N	0.58223000	2.76060900	1.07299700	N	-1.27335800	2.35294700	-1.30149600
H	-6.30004200	-1.28069700	-1.26368700	H	6.01289200	0.15855600	2.21159900
N	-3.04225400	-0.37315700	-1.23586900	N	2.76966500	0.30492900	1.27273600
N	-2.03997100	-0.34310700	2.22303100	N	2.35740600	0.34663000	-2.45689400
H	-1.65847100	0.01995500	3.09895500	H	2.03550600	0.75532100	-3.33669100
H	-2.99418300	0.01474200	2.15844800	H	3.21457600	0.84341100	-2.20707000
N	3.49357800	-2.61589800	2.50213100	N	-2.22730700	-3.73114400	-2.11185400
H	3.41445300	-1.62369200	2.73435900	H	-2.51471200	-2.99103300	-2.75603000
H	3.30070700	-3.12232600	3.36701600	H	-1.98041600	-4.52751600	-2.70034800

### 3.1.2.5 Product

Atom	X	Y	Z
P	-1.25696100	0.26620000	0.91391300
O	-0.14613000	-0.21766600	1.78331800
O	-0.85916100	1.19549200	-0.33630700
O	-1.93056100	-1.00786500	0.20387200
O	2.67121300	-2.91325000	-0.46587900
C	-3.07094700	-1.08149300	-0.60310300
C	-3.48178300	-2.37265600	-0.93429300
C	-4.74863300	-0.09300200	-1.77324100
C	-4.60867500	-2.48494700	-1.74237500
H	-2.93747800	-3.23586300	-0.56965300
C	-5.25971300	-1.32255800	-2.17234300
H	-5.22105000	0.83415400	-2.08387300
H	-4.97382800	-3.46522800	-2.03176500
C	3.59228100	-1.92566800	-0.63534500
C	5.31468800	-1.13032400	-1.92395300
C	4.65720600	0.12498600	0.00115900
C	5.49612900	-0.01648600	-1.11180100
H	5.94179400	-1.28207100	-2.79838700
H	4.75816600	0.97815700	0.66555000
C	-0.10511000	2.37125900	-0.27322900
C	0.28749800	2.91093000	-1.49686200
C	1.02531400	4.09029500	-1.45126000
H	0.02298400	2.42319500	-2.42771000
C	0.88111700	4.02136400	0.94044300
C	1.33203200	4.65961200	-0.20965200
H	1.35783500	4.55782200	-2.37255400
H	1.09327900	4.42715200	1.92495500
C	3.68782700	-0.83857600	0.25188400
H	3.01687400	-0.76279600	1.10078800
O	-2.40003700	1.14837400	1.62595800
H	0.57275200	-1.83035600	1.66763800
O	1.03654800	-2.70188800	1.65580000
H	2.11285200	-2.76188000	0.34382100
H	6.26511100	0.71291700	-1.34139200
N	4.38209900	-2.07711900	-1.70176300
H	1.90653700	5.57625300	-0.13551500
N	0.15974400	2.88155600	0.91547000
H	-6.14094000	-1.36837100	-2.80268000
N	-3.65597600	0.03631800	-0.99112400
N	-2.96959700	0.42906700	2.78107700
H	-2.79209000	1.08235100	3.54660400
H	-3.97029400	0.48318600	2.58111900
N	1.73794600	-2.84806800	2.91496300
H	2.28127800	-1.98978600	3.03168800
H	1.01076900	-2.82609300	3.63205700

**Table S.7.** Structural parameters of intermediate (**Int**), transition state (**TS2**) and products (**P**) for Mechanism 1 in the reaction of **TPP** with hydroxylamine at the B3LYP/6-31++G(d,p) level of theory.

Interatomic distances (Å)						
	P <sub>1</sub> -O <sub>2</sub>	O <sub>2</sub> -H <sub>3</sub>	H <sub>3</sub> -O <sub>4</sub>	O <sub>4</sub> -H <sub>5</sub>	H <sub>5</sub> -O <sub>6</sub>	H <sub>6</sub> -O <sub>1</sub>
<b>Int</b>	1.598	1.005	1.653	0.971	2.386	1.754
<b>TS2</b>	1.552	1.053	1.447	0.999	1.688	2.281
<b>P</b>	1.488	1.831	0.980	1.740	0.992	3.573
Dihedral angles						
	P <sub>1</sub> -O <sub>2</sub> -H <sub>3</sub> -O <sub>4</sub>	O <sub>2</sub> -H <sub>3</sub> -O <sub>4</sub> -H <sub>5</sub>	H <sub>3</sub> -O <sub>4</sub> -H <sub>5</sub> -O <sub>6</sub>	O <sub>4</sub> -H <sub>5</sub> -O <sub>6</sub> -P <sub>1</sub>	H <sub>5</sub> -O <sub>6</sub> -P <sub>1</sub> -O <sub>2</sub>	O <sub>6</sub> -P <sub>1</sub> -O <sub>2</sub> -H <sub>3</sub>
<b>TS2</b>	-11.3	11.6	-13.3	18.2	-8.1	5.4
Imaginary frequency (cm <sup>-1</sup> )						
193.10						

**Table S.8.** Structural parameters of intermediate (**Int**), alternative transition states (**TS4** and **TS5**) and products (**P**) for Mechanism 2 in the reaction of **TPP** with hydroxylamine at the B3LYP/6-31++G(d,p) level of theory.

Interatomic distances (Å)						
	P <sub>1</sub> -O <sub>2</sub>	O <sub>2</sub> -H <sub>3</sub>	H <sub>3</sub> -O <sub>4</sub>	O <sub>4</sub> -H <sub>5</sub>	H <sub>5</sub> -O <sub>6</sub>	H <sub>6</sub> -O <sub>1</sub>
<b>Int</b>	1.602	1.001	1.663	0.970	2.935	1.744
<b>TS4</b>	1.556	1.043	1.475	0.999	1.703	2.332
<b>TS5</b>	1.510	1.601	1.070	1.027	2.938	2.228
<b>P</b>	1.491	1.770	0.987	1.698	0.995	5.239
Dihedral angles						
	P <sub>1</sub> -O <sub>2</sub> -H <sub>3</sub> -O <sub>4</sub>	O <sub>2</sub> -H <sub>3</sub> -O <sub>4</sub> -H <sub>5</sub>	H <sub>3</sub> -O <sub>4</sub> -H <sub>5</sub> -O <sub>6</sub>	O <sub>4</sub> -H <sub>5</sub> -O <sub>6</sub> -P <sub>1</sub>	H <sub>5</sub> -O <sub>6</sub> -P <sub>1</sub> -O <sub>2</sub>	O <sub>6</sub> -P <sub>1</sub> -O <sub>2</sub> -H <sub>3</sub>
<b>TS4</b>	-8.7	38.4	-10.9	-6.0	24.4	-25.6
	P <sub>1</sub> -O <sub>2</sub> -H <sub>3</sub> -N <sub>4</sub>	O <sub>2</sub> -H <sub>3</sub> -N <sub>4</sub> -H <sub>5</sub>	H <sub>3</sub> -N <sub>4</sub> -H <sub>5</sub> -O <sub>6</sub>	N <sub>4</sub> -H <sub>5</sub> -O <sub>6</sub> -P <sub>1</sub>	H <sub>5</sub> -O <sub>6</sub> -P <sub>1</sub> -O <sub>2</sub>	O <sub>6</sub> -P <sub>1</sub> -O <sub>2</sub> -H <sub>3</sub>
<b>TS5</b>	149.5	-151.7	24.0	-39.6	29.6	-28.7
Imaginary frequency (cm <sup>-1</sup> )						
<b>TS4</b>				<b>TS5</b>		
168.9				160.0i		

## 3.2. Reaction of DPP with hydroxylamine

### 3.2.1 Cartesian coordinates for the reaction DPP + NH<sub>2</sub>OH:

#### 3.2.1.1 Reactant

#### 3.2.1.2. Transition State (TS)

Atom	X	Y	Z	Atom	X	Y	Z
P	-0.43922800	-0.71497900	-0.07807800	P	0.76825900	-0.91439900	0.31669500
O	-0.41432600	-1.88861900	0.86083900	O	0.44002900	-2.16876200	-0.46397800
O	1.04840500	-0.65967600	-0.89727300	O	-0.91432100	-0.59833100	1.03472400
O	-0.35637200	0.62157700	0.86716800	O	0.64919700	0.40544800	-0.66634900
O	-1.43739300	-0.55758400	-1.18832500	O	1.44671200	-0.81798100	1.66171900
C	2.26963900	-0.73723900	-0.36267000	C	-2.05927600	-0.59764500	0.40252800
C	2.61713100	-0.72499100	0.98892300	C	-2.28872300	-0.60175200	-0.98786800
C	4.56958400	-0.88254500	-0.98089100	C	-4.43586500	-0.55758100	0.77198700
C	3.96376000	-0.79214300	1.32772400	C	-3.59074000	-0.58408200	-1.45983700
H	1.84981700	-0.66098500	1.74784900	H	-1.44717700	-0.61199700	-1.66528900
C	4.95742600	-0.87008700	0.33985500	C	-4.68866100	-0.56176500	-0.57785700
H	5.24957900	-0.94130200	-1.82086500	H	-5.20046900	-0.54055900	1.53826000
H	4.24382500	-0.78087200	2.37580000	H	-3.76251900	-0.58486700	-2.53161600
H	6.00853900	-0.92061500	0.59394700	H	-5.70902600	-0.54664800	-0.93919500
C	-0.59918100	1.93192100	0.46256700	C	0.77601600	1.74008800	-0.34454000
C	-0.09367700	2.44189500	-0.73494000	C	0.85440600	2.21652500	0.96891000
C	-0.35451500	3.78062500	-1.01952700	C	0.96269200	3.59377100	1.14616700
H	0.48501200	1.82478000	-1.41111700	H	0.83564500	1.54051900	1.81323700
C	-1.52932600	3.92810000	1.05897600	C	0.89599900	3.84766200	-1.22807800
C	-1.08771400	4.54471000	-0.10764200	C	0.98580700	4.43449100	0.03015700
H	0.01716400	4.21971600	-1.94006800	H	1.02722200	4.00189400	2.15025900
H	-2.09704400	4.48351300	1.79969100	H	0.90701200	4.45803100	-2.12667400
H	-1.30850100	5.59040400	-0.29176600	H	1.06861600	5.51114900	0.13029000
N	-1.29503300	2.63364500	1.34839700	N	0.79300600	2.52011200	-1.42337200
N	3.25141700	-0.81890500	-1.29079500	N	-3.15257400	-0.57806000	1.21667100
N	-3.84075800	-2.24209100	-0.98160600	N	3.64798700	-1.63337300	0.21748300
H	-3.06349700	-1.59433500	-1.19518900	H	3.38925700	-1.47250500	1.20263900
H	-3.51994800	-2.83267400	-0.19245800	H	3.53030300	-2.64122800	0.00300200
O	-5.02255300	-1.54205200	-0.64499600	O	2.79248800	-0.86908500	-0.59463300
H	-3.98946800	-2.84454300	-1.79769600	H	4.62360200	-1.35795900	0.07082700
H	2.96974800	-0.82781000	-2.27100800	H	-2.97584800	-0.57723600	2.22009100
O	-2.61819900	-3.54281700	1.35439500	O	2.39438600	-4.12324200	-0.58359700
H	-1.85007900	-2.93805900	1.25479500	H	1.64814900	-3.47915400	-0.61826400
H	-3.09136800	-3.22918800	2.13855800	H	2.58456100	-4.34685900	-1.50596300

### 3.2.1.3 Product

Atom	X	Y	Z
P	1.48141200	-0.95620900	0.12943100
O	0.75734600	-2.17791600	-0.37444400
O	-1.92378300	-0.27212000	1.61721800
O	0.77983000	0.32440400	-0.58887700
O	1.81879500	-0.77187400	1.58102300
C	-2.88289900	-0.39992500	0.80973900
C	-2.78446000	-0.37667300	-0.62278400
C	-5.28554300	-0.72586000	0.52905300
C	-3.90210800	-0.52204200	-1.41006500
H	-1.79895000	-0.23795500	-1.05143600
C	-5.18817500	-0.70072500	-0.83572000
H	-6.21646900	-0.85697800	1.06715600
H	-3.79946300	-0.49947800	-2.49136800
H	-6.07375300	-0.81472800	-1.44811800
C	0.98568000	1.67715700	-0.32293500
C	0.91454600	2.18439800	0.97551000
C	1.05611100	3.56203900	1.12879200
H	0.74839800	1.52855800	1.82030300
C	1.28447900	3.74072000	-1.24744400
C	1.24286700	4.36182300	-0.00250200
H	1.00823900	4.00269300	2.11985500
H	1.42095000	4.32271800	-2.15422700
H	1.34882900	5.43829900	0.07578100
N	1.16157400	2.40973300	-1.41552600
N	-4.16968200	-0.58185400	1.30018600
N	3.99842300	-1.64557500	-0.24031200
H	4.18149700	-1.42636500	0.74875000
H	3.72548500	-2.66278900	-0.36400400
O	2.96384500	-0.81233100	-0.74164900
H	4.83000700	-1.42409800	-0.79885000
H	-4.26679600	-0.60445900	2.31169900
O	2.79500600	-4.11420800	-0.54923400
H	1.92471100	-3.65991200	-0.56721700
H	2.91272100	-4.49090400	-1.43419700

**Table S.9.** Structural parameters of reactant (R2), transition state (TS) and products (P) for the reaction of **DPP** with hydroxylamine at the B3LYP/6-31+G(d,p) level of theory, calculated for mechanism **B** of Scheme 9.

<b>Interatomic distances (Å)</b>								
	O <sub>1</sub> -P <sub>2</sub>	P <sub>2</sub> -O <sub>3</sub>	P <sub>2</sub> -O <sub>6</sub>	P <sub>2</sub> -O <sub>7</sub>	P <sub>2</sub> -O <sub>8</sub>	O <sub>3</sub> -N <sub>4</sub>	N <sub>4</sub> -H <sub>5</sub>	O <sub>9</sub> -H <sub>10</sub>
R2	1.693	5.917	1.508	1.635	1.500	1.412	1.041	0.983
TS	1.840	2.234	1.518	1.648	1.508	1.405	1.037	0.985
P	3.880	1.710	1.505	1.710	1.505	1.360	1.000	0.960

<b>Dihedral angles</b>						
	O <sub>1</sub> -P <sub>2</sub> -O <sub>3</sub> -O <sub>6</sub>	O <sub>1</sub> -P <sub>2</sub> -O <sub>3</sub> -N <sub>4</sub>	O <sub>3</sub> -N <sub>4</sub> -H <sub>5</sub> -O <sub>9</sub>	N <sub>4</sub> -H <sub>5</sub> -O <sub>9</sub> -H <sub>10</sub>	H <sub>5</sub> -O <sub>9</sub> -H <sub>10</sub> -O <sub>6</sub>	O <sub>6</sub> -P <sub>2</sub> -O <sub>3</sub> -N <sub>4</sub>
TS	-140.18	131.96	-30.61	-5.09	-28.16	-87.87

<b>Imaginary frequency (cm<sup>-1</sup>)</b>
215.24