SUPPORTING INFORMATION

Synthesis and photophysical properties of novel butterfly-shaped blue emitters based on pyrene

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Figure S1-1. ¹H NMR spectrum of 4a (300 MHz, CDCl₃, 293 K).



Figure S1-2. ¹³C NMR spectrum of 4a (75 MHz, CDCl₃, 293 K).

NMR spectrum



Figure S1-3. ¹H NMR spectrum of 4b (400 MHz, CDCl₃, 293 K).



Figure S1-4. ¹³C NMR spectrum of 4b (100 MHz, CDCl₃, 293 K).



Figure S1-5. ¹H NMR spectrum of 4c (300 MHz, CDCl₃, 293 K).



Figure S1-6. ¹³C NMR spectrum of 4c (75 MHz, CDCl₃, 293 K).

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Figure S1-7. ¹H NMR spectrum of 4d (300 MHz, CDCl₃, 293 K).



Figure S1-8. ¹³C NMR spectrum of 4d (100 MHz, CDCl₃, 293 K).



Figure S1-9. ¹H NMR spectrum of 4e (300 MHz, CDCl₃, 293 K).



Figure S1-10. ¹³C NMR spectrum of 4e (100 MHz, CDCl₃, 293 K).



Figure S1-11. ¹H NMR spectrum of 4f (300 MHz, CDCl₃, 293 K).



Figure S1-12. ¹³C NMR spectrum of 4f (100 MHz, CDCl₃, 293 K).



Figure S1-13. ¹H NMR spectrum of **5** (400 MHz, CDCl₃, 293 K).



Figure S1-14. ¹³CNMR spectrum of **5** (100 MHz, CDCl₃, 293 K).



Figure S1-15. ¹H NMR spectrum of 6 (400 MHz, CDCl₃, 293 K).



Figure **S1-16**. ¹³CNMR spectrum of **6** (100 MHz, CDCl₃, 293 K).



PhotophyscalAnalysis

Figure S2-1. UV/Vis absorption (left) and fluorescence spectra (right) of the compound 4a recorded in different solvents at $\sim 10^{-5}$ M and 25 °C.



Figure S2-2. UV/Vis absorption (left) and fluorescence spectra (right) of the compound 4b recorded in different solvents at $\sim 10^{-5}$ M and 25 °C.



Figure S2-3. UV/Vis absorption (left) and fluorescence spectra (right) of the compound 4c recorded in different solvents at $\sim 10^{-5}$ M and 25 °C.



Figure S2-4.UV/Vis absorption (left) and fluorescence spectra (right) of the compound 4d recorded in different solvents at $\sim 10^{-5}$ M and 25 °C.



Figure S2-5. UV/Vis absorption (left) and fluorescence spectra (right) of the compound 4e recorded in different solvents at $\sim 10^{-5}$ M and 25 °C.



Figure S2-6. UV/Vis absorption (left) and fluorescence spectra (right) of the compound 4f recorded in different solvents at $\sim 10^{-5}$ M and 25 °C.



Figure S2-7. UV/Vis absorption (left) and fluorescence spectra (right) of the compound 5 recorded in different solvents at $\sim 10^{-5}$ M and 25 °C.



Figure S2-8Effect of concentration on the UV/Vis (left) and fluorescence emission (right) of 4ain

CH₂Cl₂ at 25 °C.



Figure S2-9Effect of concentration on the UV/Vis (left) and fluorescence emission (right) of 4bin

 CH_2Cl_2 at 25 °C.



Figure S2-10Effect of concentration on the UV/Vis (left) and fluorescence emission (right) of 4cin CH_2Cl_2 at 25 °C.



Figure S2-11Effect of concentration on the UV/Vis (left) and fluorescence emission (right) of 4din CH_2Cl_2 at 25 °C.



Figure S2-12Effect of concentration on the UV/Vis (left) and fluorescence emission (right) of 4ein CH_2Cl_2 at 25 °C.



Figure S2-13Effect of concentration on the UV/Vis (left) and fluorescence emission (right) of 4fin CH_2Cl_2 at 25 °C.



Figure S2-14Effect of concentration on the UV/Vis (left) and fluorescence emission (right) of 5in CH₂Cl₂ at 25 °C.







Electrochemistry Analysis

Figure S4-1 Cyclic voltammogramscu for compound5.

Quantum Chemistry Computation



Figure S4-1. Computed molecular orbital plots (B3LYP/6–31G*) of the compound 4a. The left plots represent the HOMOs, and the right plots represent the LUMOs.



Figure S4-2. Computed molecular orbital plots (B3LYP/6–31G*) of the compound 4c. The left plots represent the HOMOs, and the right plots represent the LUMOs.



Figure S4-3. Computed molecular orbital plots (B3LYP/6–31G*) of the compound 4f. The left plots represent the HOMOs, and the right plots represent the LUMOs.



Figure S4-4. Computed molecular orbital plots (B3LYP/6–31G*) of the compound 6. The left plots represent the HOMOs, and the right plots represent the LUMOs.