Supporting Information

Palladium-Catalyzed Tandem N–H/C–H Arylation: Regioselective Synthesis of N-Heterocycle-Fused Phenanthridines as Versatile Blue-Emitting Luminophores

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I. Fluorescence lifetimes of compounds 3e, 3h, 3l and 4m

Compd	Lifetime in MeCN		Compd	Lifetime in MeC	ĽN
3e	$\tau_1 = 16.38 \text{ ns}$	$\chi^2 = 1.06$	31	$\tau_1 = 3.22 \text{ ns} (17.55\%)$ $\tau_2 = 14.78 \text{ ns} (82.45\%)$	$\chi^2 = 1.04$
3h	$\tau_1 = 2.91 \text{ ns} (3.33\%)$ $\tau_2 = 18.35 \text{ ns} (96.67\%)$	$\chi^2 = 1.05$	4m	$\tau_1 = 4.08 \text{ ns} (41.39\%)$ $\tau_2 = 14.26 \text{ ns} (58.61\%)$	$\chi^2 = 1.01$

 Table S1. Fluorescence lifetimes of compounds 3e, 3h, 3l and 4m.^a

^{*a*} Fluorescence lifetimes were determined on a HORIBA TEMPRO-01 instrument at room temperature.

II. Electrochemical properties of compounds 3e, 3h, 3l and 4m

Cyclic voltammetry (CV) measurements were performed on LK2005A using an Ag/Ag^+ (0.01 M of AgNO₃ in dry acetonitrile) reference electrode, a platinum wire counter electrode, and a platinum plate working electrode. CV measurements were carried out in dry acetonitrile using Fc/Fc⁺ as reference at a scanning rate of 100 mV·s⁻¹ with tetrabutylammonium hexafluorophosphate (NBu₄PF₆, 0.1 M) as supporting electrolyte.



Figure S1 Electrochemical properties of compounds 3e, 3h, 3l and 4m.

III. Thermal properties of compounds 3e, 3h, 3l and 4m

Thermal decomposition temperatures were detected by thermogravimetry/differential thermal analysis (TG/DTA) on a NETZSCH-Leading Thermal Analysis in the temperature range of 30-500 °C at a heating rate of 10 °C/min under a nitrogen atmosphere.



Figure S2 TGA curves of the compounds 3e, 3h, 3l and 4m.



IV. Copies of ¹H, ¹³C and ¹H-¹H NOESY NMR spectra























































