Supporting Information

Fluorene-Based Material Containing Triple Azacrown Ether Groups: Synthesis, Characterization and Application in Chemosensor and Electroluminescent Device

Chia-Shing Wu, Ying-Ju Lin and Yun Chen*

Department of Chemical Engineering, National Cheng Kung University

E-mail address: yunchen@mail.ncku.edu.tw*



Electronic Supplementary Material (ESI) for Organic & Biomolecular Chemistry This journal is The Royal Society of Chemistry 2013



Figure S1 ¹H NMR and ¹³C NMR spectra of compound FTC.



Electronic Supplementary Material (ESI) for Organic & Biomolecular Chemistry This journal is © The Royal Society of Chemistry 2013



Figure S2 COSY spectrum of FTC [(a) 0.4~2.4 ppm and (b) 6.2~8.2 ppm)].



Figure S3 NOESY spectrum of FTC (6.4~8.2 ppm).



Figure S4 (a) Differential scanning calorimetric curves of **FTC** obtained from the first and second scan with a heating rate of 10 °C/min. (b) Thermogravimetric analysis of **FTC** with heating rate of 10 °C/min under nitrogen atmosphere.



Figure S5 Absorption spectra of FTC with various metal ions in solution (methanol/H₂O = 9/1, v/v). The concentration of FTC and metal ions were fixed at 10^{-5} M and 10^{-4} M, respectively.



Figure S6 Photoluminescence spectra of FTC with various metal ions in solution (methanol/H₂O = 9/1, v/v). The concentration of FTC and metal ions were fixed at 10^{-5} M and 10^{-4} M, respectively. Excitation wavelength was 412 nm.



Figure S7 Fluorescence emission response profile of **FTC** upon addition of various metal ions in solution (methanol/H₂O = 9/1, v/v). The concentration of **FTC** and metal ions were fixed at 10^{-5} M and 10^{-4} M, respectively. Excitation wavelength was 412 nm.



Figure S8 Absorption spectrum of Fe^{3+} (solid line) and fluorescence spectrum of **FTC** (dashed line) in solution (THF/H₂O = 9/1, v/v). Excitation wavelength was 418 nm.



Figure S9 Absorption spectrum of Fe^{3+} (solid line) and fluorescence spectrum of **FTC** (dashed line) in solution (Methanol/H₂O = 9/1, v/v). Excitation wavelength was 412 nm.



Figure S10 Photoluminescence spectra of **FTC** with varying concentration of Fe^{3+} in solution (ethanol/H₂O = 9/1, v/v). The concentration of **FTC** was fixed at 10⁻⁵ M. Excitation wavelength was 413 nm.



Figure S11 The Stern-Volmer plot of **FTC** with varying concentration of Fe^{3+} in solution (ethanol/H₂O = 9/1, v/v). The concentration of **FTC** was fixed at 10⁻⁵ M. Excitation wavelength was 413 nm.



Figure S12 The Stern-Volmer plot of **FTC** with varying concentration of Fe^{3+} in solution (THF/H₂O = 9/1, v/v). The concentration of **FTC** was fixed at 10⁻⁵ M. Excitation wavelength was 418 nm.



Figure S13 Absorption spectra of **FTC** with varying concentration of HCl in solution (THF/H₂O = 9/1, v/v). The concentration of **FTC** was fixed at 10^{-5} M.



Figure S14 Photoluminescence spectra of **FTC** with various concentration of HCl in solution (THF/H₂O = 9/1, v/v). The concentration of **FTC** was fixed at 10^{-5} M. Excitation wavelength was 419 nm



Figure S15 Cyclic voltammogram of FTC in 0.1 M n-Bu₄NClO₄, scan rate: 100 mV/s.