Supporting Information

Copper-catalyzed decarboxylative C3-acylation of free (N–H) indoles with

α-oxocarboxylic acids

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1. General Experimental Method

1.1 General Considerations

All reactions were carried out in dried glassware. Pd and Cu catalysts, Ag (I) salts, SeO₂, indoles and solvents were purchased from commercial sources and used without further purification. Phenylglyoxylic acid was purchased from Alfa Aesar and other α -oxocarboxylic acids were prepared from oxidation of corresponding methyl ketones with SeO₂ according to the reported procedure¹.

Thin layer chromatography (TLC) employed glass 0.25 mm silica gel plates. Flash chromatography columns were packed with 300-400 mesh silica gel in petroleum ether. All melting points are determined on a WRS-2A micro processor melting-point apparatus and uncorrected. IR spectra were recorded on a WQF-200 FTIR spectrometer and only major peaks were recorded in cm⁻¹. ¹H NMR and ¹³C NMR data were recorded on Bruker Advance NMR spectrometer (500 MHz) in C₂D₆O as solvent. Chemical shifts (δ) were reported in parts per million (ppm) using tetramethylsilane (TMS) as the internal standard. The peak patterns are indicated as follows: s, singlet; d, doublet; t, triplet; m, multiplet; q, quartet. The coupling constants, *J*, are reported in Hertz (Hz). HRMS (ESI) analysis was obtained from Micromass UPLC/Q-TOF Micro Mass Spectrometer. XRD powder diffraction was carried out on Bruker's D8 ADVANCE X-Ray Diffractometer with Guinier imaging plate detector (Cu-K_{a1} radiation, λ = 154.056 pm).

1.2 General procedure for Copper-catalyzed decarboxylative C3-acylation of free (N–H) indoles with α-oxocarboxylic acids

A dried two-neck round bottom flask was charged with indole (0.8 mmol), α -oxocarboxylic acid (1.6 mmol), Cu(OAc)₂·H₂O (39.9 mg, 20 mol%), Ag₂CO₃ (441.2 mg, 1.6 mmol) and DMSO or DMF (4 mL) under air atmosphere. The reaction mixture was heated at 90 °C and stirred for 1.5-8 h. After completion of the reaction (monitored by TLC or HPLC), the reaction mixture was cooled to room temperature and diluted with ethyl acetate (5 mL) and water (5 mL), then the filtrate collected by centrifugation and filtration was extracted with ethyl acetate (3×10 mL). The organic layers were combined, washed with saturated NaHCO₃, dried with anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. The crude products were purified by flash column chromatography on silica gel (petroleum ether/ethyl acetate, 5/1, v/v) to afford the desired products. The structures of the products were identified by IR, NMR and HRMS analyses.

2.	Optimization	of	silver	oxidants,	its	loading	and	the	reaction	temperature'
(T	able S1)									

	O O O O O O O O O O	$\frac{(DAc)_2 \cdot H_2O, Oxidant}{DMSO, T, 3 h}$	O Ph N H			
	1a 2a		3a			
Entry	Catalyst	Oxidant (equiv)	Yield $(\%)^b$			
1	$Cu(OAc)_2 \cdot H_2O$	$Ag_2CO_3(2)$	71			
2	$Cu(OAc)_2 \cdot H_2O$	Ag ₂ O (2)	63			
3	$Cu(OAc)_2 \cdot H_2O$	AgOAc (2)	37			
4	$Cu(OAc)_2 \cdot H_2O$	$AgNO_3(2)$	0			
5	$Cu(OAc)_2 \cdot H_2O$	$Ag_2CO_3(1)$	49			
6	$Cu(OAc)_2 \cdot H_2O$	$Ag_2CO_3(1.5)$	60			
7	$Cu(OAc)_2 \cdot H_2O$	$Ag_2CO_3(1.7)$	64			
8	$Cu(OAc)_2 \cdot H_2O$	$Ag_2CO_3(2.5)$	61			
9 ^c	$Cu(OAc)_2 \cdot H_2O$	$Ag_2CO_3(2)$	68			
10^d	$Cu(OAc)_2 \cdot H_2O$	$Ag_2CO_3(2)$	58			
^{<i>a</i>} Conditions: 1a (0.8 mmol), 2a (1.6 mmol), Cu(OAc) ₂ ·H ₂ O (20 mol%), Oxidant, DMSO (4 mL),						
90 °C, 3 h, under air. ^{<i>b</i>} Isolated yields. ^{<i>c</i>} The reaction was performed at 80 °C. ^{<i>d</i>} The reaction was						

performed at 100 °C.

3. Characterization data for products 3a-3u

(5-methoxy-1*H*-indol-3-yl)(phenyl)methanone (3a)²



Yield: 71%, pale yellow solid, mp. 223-224 °C. IR (KBr, cm⁻¹): v_{max} 3143, 1601. ¹H NMR (500 MHz, DMSO-*d*₆): δ 11.98 (s, 1H), 7.89 (d, *J*=2.5 Hz, 1H), 7.81 (d, *J*=2.0 Hz, 1H), 7.79 (d, *J*=7.5 Hz, 2H), 7.61 (t, *J*=7.0 Hz, 1H), 7.56 (dd, *J*=7.5, 7.0 Hz, 2H), 7.44 (d, *J*=8.5 Hz, 1H), 6.92 (d, *J*=8.5 Hz, 1H), 3.83 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 190.36, 156.06, 141.16, 136.31, 132.05, 131.37, 128.83, 128.73, 127.55, 115.36, 113.48, 113.43, 103.79, 55.80. HRMS (ESI) calcd for C₁₆H₁₃NO₂Na [M+Na]⁺ 274.0844, found m/z 274.0848.

(5-methyl-1*H*-indol-3-yl)(phenyl)methanone (3b)³



Yield: 77%, yellow solid, mp. 227-228 °C. IR (KBr, cm⁻¹): v_{max} 3154, 1597. ¹H NMR (500 MHz, DMSO-*d*₆): δ 11.97 (s, 1H), 8.08 (s, 1H), 7.87 (d, *J*=3.0 Hz, 1H), 7.77 (d, *J*=7.0 Hz, 2H), 7.61-7.58 (m, 1H), 7.55-7.52 (m, 2H), 7.40 (d, *J*=8.5 Hz, 1H), 7.09 (d, *J*=8.0 Hz, 1H), 2.44 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 190.38, 141.17, 136.16, 135.52, 131.38, 131.19, 128.81, 128.77, 127.01, 125.06, 121.69, 115.16, 112.33, 21.82. HRMS (ESI) calcd for C₁₆H₁₃NONa [M+Na]⁺ 258.0895, found m/z 258.0896.

(6-methoxy-1*H*-indol-3-yl)(phenyl)methanone (3c)⁴



Yield: 37%, brown solid, mp. 226-228 °C. IR (KBr, cm⁻¹): v_{max} 3119, 1596. ¹H NMR (500 MHz, DMSO- d_6): δ 11.87 (s, 1H), 8.11 (d, *J*=9.0 Hz, 1H), 7.80-7.76 (m, 3H), 7.60 (t, *J*=7.0 Hz, 1H), 7.53 (dd, *J*=7.0, 7.0 Hz, 2H), 7.00 (s, 1H), 6.88 (d, *J*=8.0 Hz, 1H), 3.80 (s, 3H); ¹³C NMR (125 MHz, DMSO- d_6) δ 190.28, 157.07, 141.03, 138.11, 135.26, 131.42, 128.81, 128.77, 122.59, 120.71, 115.60, 112.13, 95.77, 55.75. HRMS (ESI) calcd for C₁₆H₁₃NO₂Na [M+Na]⁺ 274.0844, found m/z 274.0848.

(2-methyl-1*H*-indol-3-yl)(phenyl)methanone (3d)⁵



Yield: 35%, brown solid, mp. 178-180 °C. IR (KBr, cm⁻¹): v_{max} 3170, 1606. ¹H NMR (500 MHz, DMSO- d_6): δ 11.96 (s, 1H), 7.61-7.58 (m, 3H), 7.50 (dd, *J*=7.0, 7.5 Hz, 2H), 7.39 (d, *J*=8.0 Hz, 1H), 7.33 (d, *J*=8.0 Hz, 1H), 7.12 (dd, *J*=7.5, 7.5 Hz, 1H), 7.00 (dd, *J*=7.5, 7.5 Hz, 1H), 2.38 (s, 3H); ¹³C NMR (125 MHz, DMSO- d_6) δ 192.18, 144.87, 142.20, 135.46, 131.46, 128.81, 128.46, 127.78,

122.27, 121.39, 120.49, 112.98, 111.71, 14.64. HRMS (ESI) calcd for $C_{16}H_{13}NONa [M+Na]^+$ 258.0895, found m/z 258.0896.

(1*H*-indol-3-yl)(phenyl)methanone (3e)^{2b, 6}



Yield: 55%, pale yellow solid, mp. 244-246 °C. IR (KBr, cm⁻¹): v_{max} 3130, 1593. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.08 (s, 1H), 8.26 (d, *J*=7.0 Hz, 1H), 7.94 (d, *J*=3.0 Hz, 1H), 7.79 (d, *J*=7.0 Hz, 2H), 7.63-7.60 (m, 1H), 7.56-7.52 (m, 3H), 7.29-7.23 (m, 2H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 190.43, 141.05, 137.20, 136.16, 131.48, 128.85, 128.82, 126.74, 123.58, 122.35, 121.94, 115.52, 112.70. HRMS (ESI) calcd for C₁₅H₁₁NONa [M+Na]⁺ 244.0738, found m/z 244.0742.

(5-bromo-1*H*-indol-3-yl)(phenyl)methanone (3f)^{6a}



Yield: 54%, pale yellow solid, mp. 265-266 °C. IR (KBr, cm⁻¹): v_{max} 3136, 1612. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.27 (s, 1H), 8.40 (d, *J*=2.0 Hz, 1H), 8.03 (d, *J*=3.0 Hz, 1H), 7.79 (d, *J*=7.0 Hz, 2H), 7.64-7.61 (m, 1H), 7.55 (dd, *J*=7.5, 7.5 Hz, 2H), 7.51 (d, *J*=8.5 Hz, 1H), 7.41 (dd, *J*=8.5, 2.0 Hz, 1H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 190.28, 140.55, 137.28, 135.95, 131.76, 128.94, 128.86, 128.54, 126.22, 124.10, 115.18, 114.93, 114.83. HRMS (ESI) calcd for C₁₅H₁₀NOBrNa [M+Na]⁺ 321.9843, found m/z 321.9846.

(5-nitro-1*H*-indol-3-yl)(phenyl)methanone (3g)



Yield: 49%, orange yellow solid, mp. 292-294 °C. IR (KBr, cm⁻¹): *v*_{max} 3118, 1606. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.70 (s, 1H), 9.14 (d, *J*=2.5 Hz, 1H), 8.28 (s, 1H), 8.18 (d, *J*=9.0 Hz, 1H),

7.85 (d, *J*=7.0 Hz, 2H), 7.73 (d, *J*=9.0 Hz, 1H), 7.66 (m, 1H), 7.58 (dd, *J*=7.5, 7.5 Hz, 2H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 190.30, 143.36, 140.40, 140.07, 139.40, 132.15, 129.06, 128.98, 126.23, 119.02, 118.41, 116.74, 113.55. HRMS (ESI) calcd for C₁₅H₁₀N₂O₃Na [M+Na]⁺ 289.0589, found m/z 289.0588.

(5-methoxy-1*H*-indol-3-yl)(4-nitrophenyl)methanone (3h)



Yield: 90%, yellow solid, mp. 251-253 °C. IR (KBr, cm⁻¹): v_{max} 3163, 1697. ¹H NMR (500 MHz, DMSO- d_6): δ 12.16 (s, 1H), 8.37 (d, *J*=8.5 Hz, 2H), 8.00 (d, *J*=8.5 Hz, 2H), 7.93 (d, *J*=2.5 Hz, 1H), 7.80 (d, *J*=2.0 Hz, 1H), 7.45 (d, *J*=9.0 Hz, 1H), 6.94 (d, *J*=9.0 Hz, 1H), 3.83 (s, 3H); ¹³C NMR (125 MHz, DMSO- d_6) δ 188.53, 156.36, 149.09, 146.58, 137.32, 132.12, 129.92, 127.31, 124.09, 115.17, 113.79, 113.64, 103.75, 55.82. HRMS (ESI) calcd for C₁₆H₁₂N₂O₄Na [M+Na]⁺ 319.0695, found m/z 319.0706.

(4-bromophenyl)(5-methoxy-1*H*-indol-3-yl)methanone (3i)



Yield: 86%, yellow solid, mp. 235-236 °C. IR (KBr, cm⁻¹): v_{max} 3190, 1684. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.01 (s, 1H), 7.89 (s, 1H), 7.77-7.71 (m, 5H), 7.41 (s, 1H), 6.89 (s, 1H), 3.79 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 189.12, 156.15, 140.11, 136.52, 132.07, 131.86, 130.82, 127.49, 125.04, 115.14, 113.57, 113.48, 103.79, 55.81. HRMS (ESI) calcd for C₁₆H₁₂NO₂BrNa [M+Na]⁺ 351.9949, found m/z 351.9952.

(4-chlorophenyl)(5-methoxy-1*H*-indol-3-yl)methanone (3j)⁷



Yield: 89%, yellow solid, mp. 199-201 °C. IR (KBr, cm⁻¹): v_{max} 3176, 1678. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.01 (s, 1H), 7.89 (d, *J*=3.0 Hz, 1H), 7.79-7.77 (m, 3H), 7.58 (d, *J*=8.5 Hz, 2H), 7.40 (d, *J*=8.5 Hz, 1H), 6.89 (dd, *J*=8.5, 2.0 Hz, 1H), 3.79 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 189.01, 156.15, 139.77, 136.49, 136.17, 132.07, 130.63, 128.93, 127.50, 115.16, 113.57, 113.48, 103.78, 55.81. HRMS (ESI) calcd for C₁₆H₁₂NO₂ClNa [M+Na]⁺ 308.0454, found m/z 308.0448.

(2-chlorophenyl)(5-methoxy-1*H*-indol-3-yl)methanone (3k)



Yield: 44%, yellow solid, mp. 207-209 °C. IR (KBr, cm⁻¹): v_{max} 3177, 1676. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.01 (s, 1H), 7.66 (s, 1H), 7.57-7.46 (m, 5H), 7.41 (d, *J*=7.5 Hz, 1H), 6.90 (d, *J*=5.5 Hz, 1H), 3.81 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 188.54, 156.29, 140.83, 137.29, 132.29, 131.09, 130.20, 130.09, 129.13, 127.51, 126.73, 116.36, 113.73, 113.67, 103.43, 55.79. HRMS (ESI) calcd for C₁₆H₁₂NO₂ClNa [M+Na]⁺ 308.0454, found m/z 308.0448.

(5-methoxy-1*H*-indol-3-yl)(p-tolyl)methanone (3l)



Yield: 36%, brown solid, mp. 235-237 °C. IR (KBr, cm⁻¹): v_{max} 3167, 1678. ¹H NMR (500 MHz, DMSO- d_6): δ 11.92 (s, 1H), 7.87 (d, *J*=2.5 Hz, 1H), 7.78 (d, *J*=1.5 Hz, 1H), 7.69 (d, *J*=8.0 Hz, 2H), 7.41 (d, *J*=8.5 Hz, 1H), 7.34 (d, *J*=7.5 Hz, 2H), 6.89 (dd, *J*=8.5, 2.5 Hz, 1H), 3.80 (s, 3H), 2.40 (s, 3H); ¹³C NMR (125 MHz, DMSO- d_6) δ 190.08, 155.98, 141.34, 138.44, 135.94, 132.01, 129.36, 128.91, 127.61, 115.40, 113.41, 113.36, 103.80, 55.79, 21.49. HRMS (ESI) calcd for C₁₇H₁₅NO₂Na [M+Na]⁺ 288.1000, found m/z 288.1006.

(5-methoxy-1*H*-indol-3-yl)(4-methoxyphenyl)methanone (3m)^{6a}



Yield: 38%, brown solid, mp. 210-212 °C. IR (KBr, cm⁻¹): v_{max} 3169, 1675. ¹H NMR (500 MHz, DMSO- d_6): δ 11.90 (s, 1H), 7.89 (d, *J*=2.5 Hz, 1H), 7.79 (d, *J*=8.5 Hz, 2H), 7.76 (s, 1H), 7.40 (d, *J*=8.5 Hz, 1H), 7.07 (d, *J*=8.0 Hz, 2H), 6.88 (d, *J*=8.5 Hz, 1H), 3.85 (s, 3H), 3.80 (s, 3H); ¹³C NMR (125 MHz, DMSO- d_6) δ 189.17, 162.11, 155.88, 135.46, 133.57, 131.98, 130.93, 127.71, 115.37, 114.12, 113.35, 113.31, 103.77, 55.85, 55.78. HRMS (ESI) calcd for C₁₇H₁₅NO₃Na [M+Na]⁺ 304.0950, found m/z 304.0954.

(1*H*-indol-3-yl)(4-nitrophenyl)methanone (3n)^{7,8}



Yield: 88%, orange yellow solid, mp. 194-195 °C. IR (KBr, cm⁻¹): v_{max} 3151, 1697. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.28 (s, 1H), 8.38 (d, *J*=8.5 Hz, 2H), 8.29 (d, *J*=7.0 Hz, 1H), 8.02 (m, 3H), 7.56 (d, *J*=7.0 Hz, 1H), 7.33-7.28 (m, 2H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 188.63, 149.15, 146.46, 137.33, 137.26, 130.01, 126.45, 124.08, 123.96, 122.79, 121.91, 115.35, 112.88. HRMS (ESI) calcd for C₁₅H₁₀N₂O₃Na [M+Na]⁺ 289.0589, found m/z 289.0588.

(4-bromophenyl)(1*H*-indol-3-yl)methanone (30)



Yield: 72%, yellow solid, mp. 198-200 °C. IR (KBr, cm⁻¹): v_{max} 3143, 1684. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.13 (s, 1H), 8.23 (d, *J*=7.5 Hz, 1H), 7.96 (s, 1H), 7.72 (s, 4H), 7.51 (d, *J*=7.0 Hz, 1H), 7.27-7.22 (m, 2H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 189.21, 140.00, 137.23, 136.42, 131.88, 130.91, 126.65, 125.17, 123.70, 122.49, 121.91, 115.30, 112.75. HRMS (ESI) calcd for C₁₅H₁₀NOBrNa [M+Na]⁺ 321.9843, found m/z 321.9846.

(4-chlorophenyl)(1*H*-indol-3-yl)methanone (3p)^{2b, 6a, 7}



Yield: 67%, yellow solid, mp. 195-197 °C. IR (KBr, cm⁻¹): v_{max} 3154, 1678. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.12 (s, 1H), 8.24 (d, *J*=7.0 Hz, 1H), 7.96 (d, *J*=2.5 Hz, 1H), 7.79 (d, *J*=8.0 Hz, 2H), 7.58 (d, *J*=8.5 Hz, 2H), 7.51 (d, *J*=7.5 Hz, 1H), 7.27-7.22 (m, 2H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 189.10, 139.65, 137.23, 136.40, 136.29, 130.73, 128.95, 126.66, 123.70, 122.48, 121.91, 112.75. HRMS (ESI) calcd for C₁₅H₁₀NOClNa [M+Na]⁺ 278.0349, found m/z 278.0343.

(5-bromo-1*H*-indol-3-yl)(4-chlorophenyl)methanone (3q)



Yield: 62%, yellow solid, mp. 250-251 °C. IR (KBr, cm⁻¹): v_{max} 3163, 1684. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.30 (s, 1H), 8.38 (s, 1H), 8.04 (s, 1H), 7.80 (d, *J*=8.5 Hz, 2H), 7.58 (d, *J*=8.0 Hz, 2H), 7.49 (d, *J*=8.5 Hz, 1H), 7.39 (d, *J*=8.5 Hz, 1H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 188.95, 139.17, 137.44, 136.57, 135.99, 130.75, 129.02, 128.47, 126.33, 124.09, 115.30, 114.85, 114.76. HRMS (ESI) calcd for C₁₅H₉NOBrClNa [M+Na]⁺ 355.9454, found m/z 355.9455.

(4-chlorophenyl)(5-methyl-1*H*-indol-3-yl)methanone (3r)



Yield: 84%, yellow solid, mp. 245-247 °C. IR (KBr, cm⁻¹): v_{max} 3166, 1670. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.03 (s, 1H), 8.08 (s, 1H), 7.91 (d, *J*=3.0 Hz, 1H), 7.80 (d, *J*=8.5 Hz, 2H), 7.60 (d, *J*=8.0 Hz, 2H), 7.41 (d, *J*=8.5 Hz, 1H), 7.10 (d, *J*=8.5 Hz, 1H), 2.44 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 189.02, 139.78, 136.35, 136.18, 135.55, 131.35, 130.67, 128.91, 126.94, 125.17, 121.65, 114.96, 112.36, 21.81. HRMS (ESI) calcd for C₁₆H₁₂NOClNa [M+Na]⁺ 292.0505, found m/z 292.0508.

(5-methyl-1*H*-indol-3-yl)(4-nitrophenyl)methanone (3s)



Yield: 87%, yellow solid, mp. 260-262 °C. IR (KBr, cm⁻¹): v_{max} 3170, 1701. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.15 (s, 1H), 8.37 (d, *J*=8.5 Hz, 2H), 8.10 (s, 1H), 8.00 (d, *J*=8.5 Hz, 2H), 7.93 (s, 1H), 7.43 (d, *J*=8.0 Hz, 1H), 7.13 (d, *J*=8.5 Hz, 1H), 2.45 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 188.56, 149.10, 146.58, 137.24, 135.63, 131.74, 129.96, 126.73, 125.43, 124.07, 121.65, 114.98, 112.50, 21.80. HRMS (ESI) calcd for C₁₆H₁₂N₂O₃Na [M+Na]⁺ 303.0746, found m/z 303.0750.

(4-chlorophenyl)(6-methoxy-1H-indol-3-yl)methanone (3t)



Yield: 53%, yellow solid, mp. 241-243 °C. IR (KBr, cm⁻¹): v_{max} 3122, 1662. ¹H NMR (500 MHz, DMSO-*d*₆): δ 11.95 (s, 1H), 8.10 (d, *J*=8.5 Hz, 1H), 7.85 (d, *J*=3.0 Hz, 1H), 7.80 (d, *J*=8.5 Hz, 2H), 7.60 (d, *J*=8.5 Hz, 2H), 7.01 (d, *J*=2.0 Hz, 1H), 6.90 (dd, *J*=8.5, 2.0 Hz, 1H), 3.81 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 188.94, 157.14, 139.63, 138.14, 136.22, 135.51, 130.68, 128.92, 122.56, 120.60, 115.39, 112.26, 95.78, 55.75. HRMS (ESI) calcd for C₁₆H₁₂NO₂ClNa [M+Na]⁺ 308.0454, found m/z 308.0448.

(4-bromophenyl)(2-methyl-1*H*-indol-3-yl)methanone (3u)



Yield: 75%, yellow solid, mp. 190-191 °C. IR (KBr, cm⁻¹): v_{max} 3201, 1679. ¹H NMR (500 MHz, DMSO-*d*₆): δ 12.02 (s, 1H), 7.72 (d, *J*=8.0 Hz, 2H), 7.57 (d, *J*=8.0 Hz, 2H), 7.40 (d, *J*=8.0 Hz, 1H), 7.35 (d, *J*=8.0 Hz, 1H), 7.14 (dd, *J*=7.5, 7.5 Hz, 1H), 7.05 (dd, *J*=7.5, 7.5 Hz, 1H), 2.40 (s, 3H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 190.87, 145.18, 141.10, 135.48, 131.90, 131.74, 130.67, 127.63, 125.13, 122.40, 121.58, 120.41, 112.68, 111.79, 14.72. HRMS (ESI) calcd for C₁₆H₁₂NOBrNa

 $[M+Na]^+$ 336.0000, found m/z 336.0002.

4. References

- K. Wadhwa, C.-X. Yang, P. R. West, K. C. Deming, S. R. Chemburkar and R. E. Reddy, *Synth. Commun.*, 2008, **38**, 4434.
- (2) (a) G. Bartoli, G. Palmieri, M. Petrini, M. Bosco and R. Dalpozzo, *Tetrahedron*, 1990, 46, 1379.
 (b) Y. Ma, J. You and F. Song, *Chem.-Eur. J.*, 2013, 19, 1189.
- (3) Y. Ito, K. Kobayashi and T. Saegusa, J. Org. Chem., 1979, 44, 2030.
- (4) J.-P. Liou, Y.-L. Chang, F.-M. Kuo, C.-W. Chang, H.-Y. Tseng, C.-C. Wang, Y.-N. Yang, J.-Y. Chang, S.-J. Lee and H.-P. Hsieh, *J. Med. Chem.*, 2004, 47, 4247.
- (5) J. G. Rodríguez and A. Urrutia, J. Heterocyclic Chem., 1999, 36, 129.
- (6) (a) S. K. Guchhait, M. Kashyap and H. Kamble, J. Org. Chem., 2011, 76, 4753; (b) J. Bergman and L. Venemalm, *Tetrahedron*, 1990, 46, 6061.
- (7) T.-S. Jiang and G.-W. Wang, Org. Lett., 2013, 15, 788.
- (8) A. R. Katritzky, K. Suzuki, S. K. Singh and H.-Y. He, J. Org. Chem., 2003, 68, 5720.

5. ¹H NMR and ¹³C NMR Spectra of product 3h and its C2-acylated isomer 3h'





6. ¹H NMR and ¹³C NMR Spectra of products 3a-3u





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7. Powder X-Ray Diffraction Analysis (XRD Analysis)

XRD powder diffraction was carried out on Bruker's D8 ADVANCE X-Ray Diffractometer with Guinier imaging plate detector (Cu-K_{α 1} radiation, λ = 154.056 pm).

A dried two-neck round bottom flask was charged with 5-methoxy-1*H*-indole (117.8 mg, 0.8 mmol), phenylglyoxylic acid (240.2 mg, 1.6 mmol), $Cu(OAc)_2 \cdot H_2O$ (39.9 mg, 20 mol%), Ag_2CO_3 (441.2 mg, 1.6 mmol) and DMSO (4 mL) under air atmosphere. The reaction mixture was heated at 90 °C and stirred for 3 h. After completion of the reaction (monitored by TLC or HPLC), the reaction mixture was cooled to room temperature and diluted with EtOAc (5 mL) and water (5 mL), then the precipitate was collected by centrifugation and filtration, washed with EtOAc (2×5 mL) and water (2×5 mL), dried in the vacuum and was analyzed by XRD (Figure S3). Significant peaks labeled were assigned to Ag^0 , implying that Ag_2CO_3 additive can be considered as terminal oxidant in the reaction.

Figure S3. XRD of the precipitate after the reaction