

Electronic Supplementary Information

Triptycene-derived macrotricyclic polyether containing anthracene unit as a powerful host for 1,2-bis(pyridium)ethane, diquat and 2,7-diazapyrenium salt

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Contents

1. ¹H NMR titration experiments of the complexes	S2
2. ¹H-¹H COSY NMR spectra of the complexes	S4
3. Mole ratio plots for the host and the guests of the complexes	S7
4. ESI MS spectra of the complexes	S8
5. Crystal packing of the complexes	S10
6. Crystal data of the complexes	S11
7. Ellipsoid plot for crystal structures of the complexes	S14
8. K⁺-Ion-controlled binding and release of guests in the complexes	S17
9. Computational methods and coordinates for the guests	S19
10. References	S22

1. ^1H NMR spectroscopic titrations of the complexes

Binding studies by proton ^1H NMR. Since binding was a fast exchange process, the association constants were determined by titrating a solution (3.0×10^{-3} M) of host **1** in $\text{CD}_3\text{CN}/\text{CDCl}_3$ (1:1, v/v) with the increased amount of a solution (0.3 M in CD_3CN) of guests **2a-2c**. Deuterated acetonitrile was used as the lock, and TMS was employed as the internal standard. Chemical shifts were reported in parts per million (ppm). Fitting of chemical shifts of proton H_1 of **1** was performed a nonlinear regression algorithm using MATLAB.

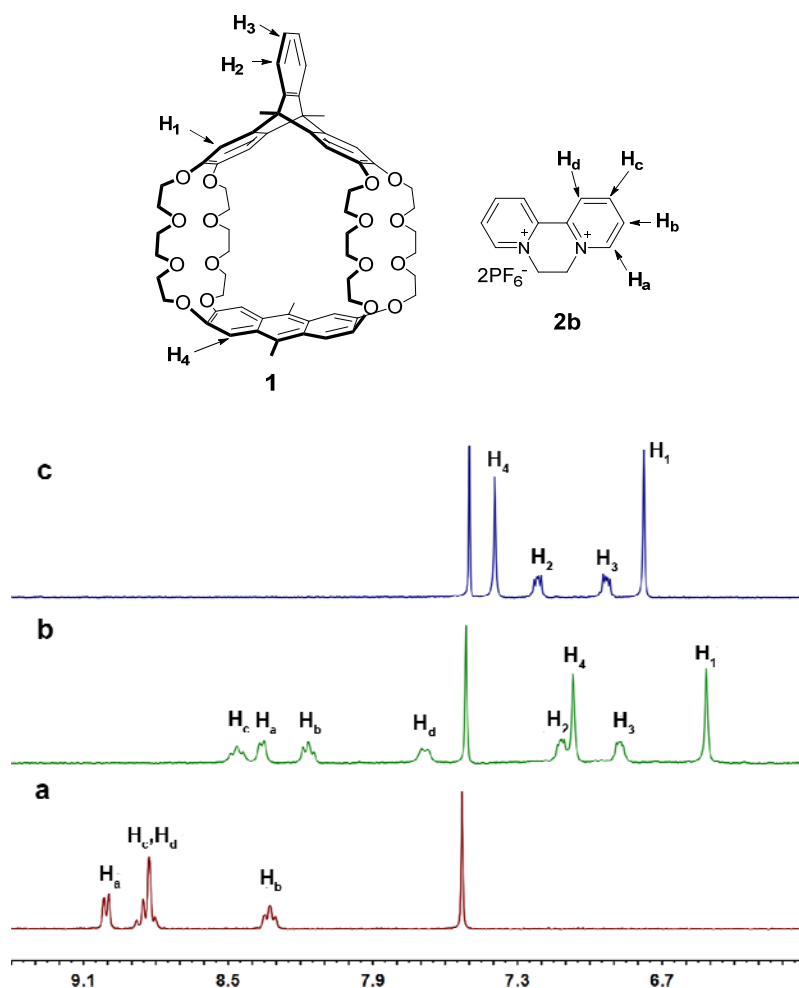


Figure S1. Partial ^1H NMR spectra (300 MHz, $\text{CD}_3\text{CN}/\text{CDCl}_3=1:1$, v/v, 295K) of (a) free guest **2b**, (b) **1** and 1.0 equiv. of **2b**, and (c) free host **1**. $[\text{H}]_0 = 3.0$ mM.

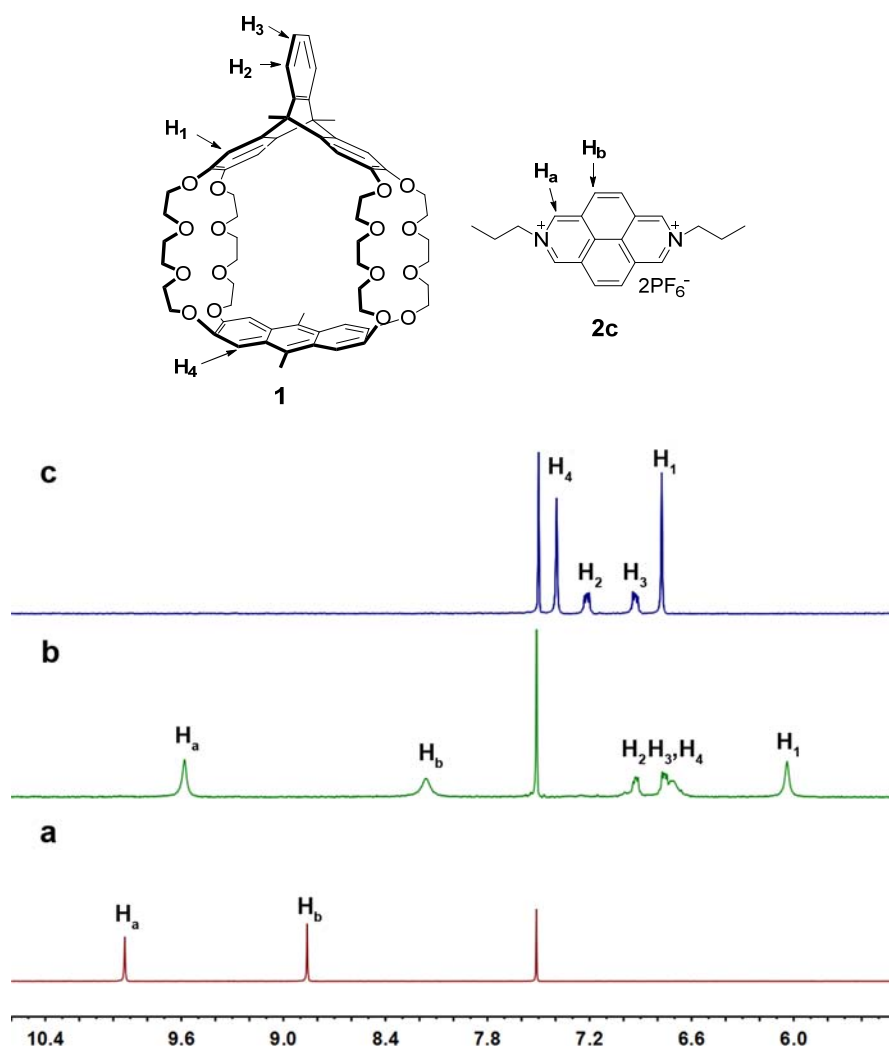


Figure S2. Partial ^1H NMR spectra (300 MHz, $\text{CD}_3\text{CN}/\text{CDCl}_3=1:1$, v/v, 295K) of (a) free guest **2c**, (b) **1** and 1.0 equiv. of **2c**, and (c) free host **1**. $[\mathbf{1}]_0 = 3.0$ mM.

2. ^1H - ^1H COSY NMR spectra of the complexes

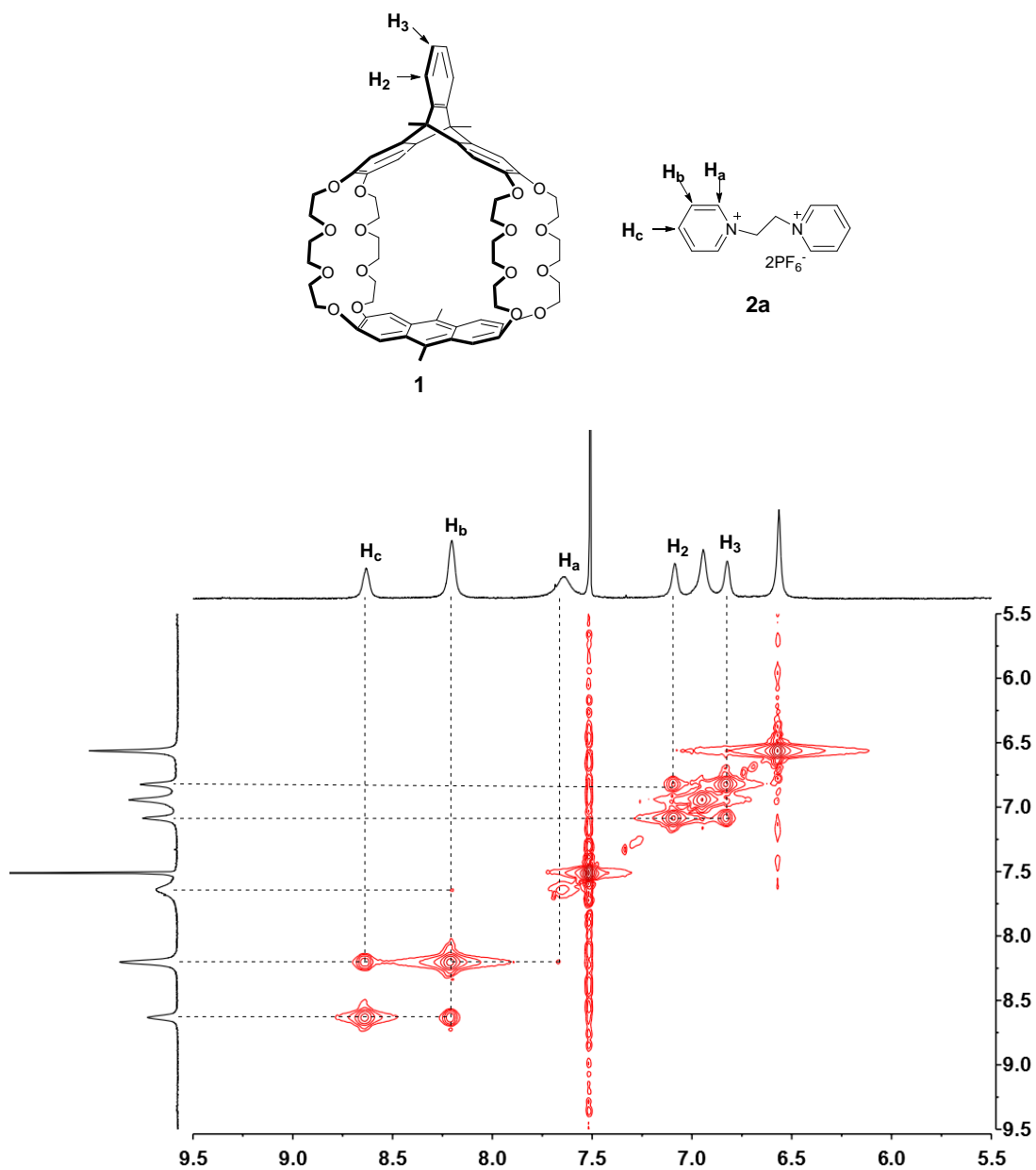


Figure S3. ^1H - ^1H COSY spectrum (300 MHz, $\text{CD}_3\text{CN}/\text{CDCl}_3=1:1$, v/v, 295 K) of host **1** and 1.0 equiv. of **2a**. $[1]_0 = 3.0$ mM.

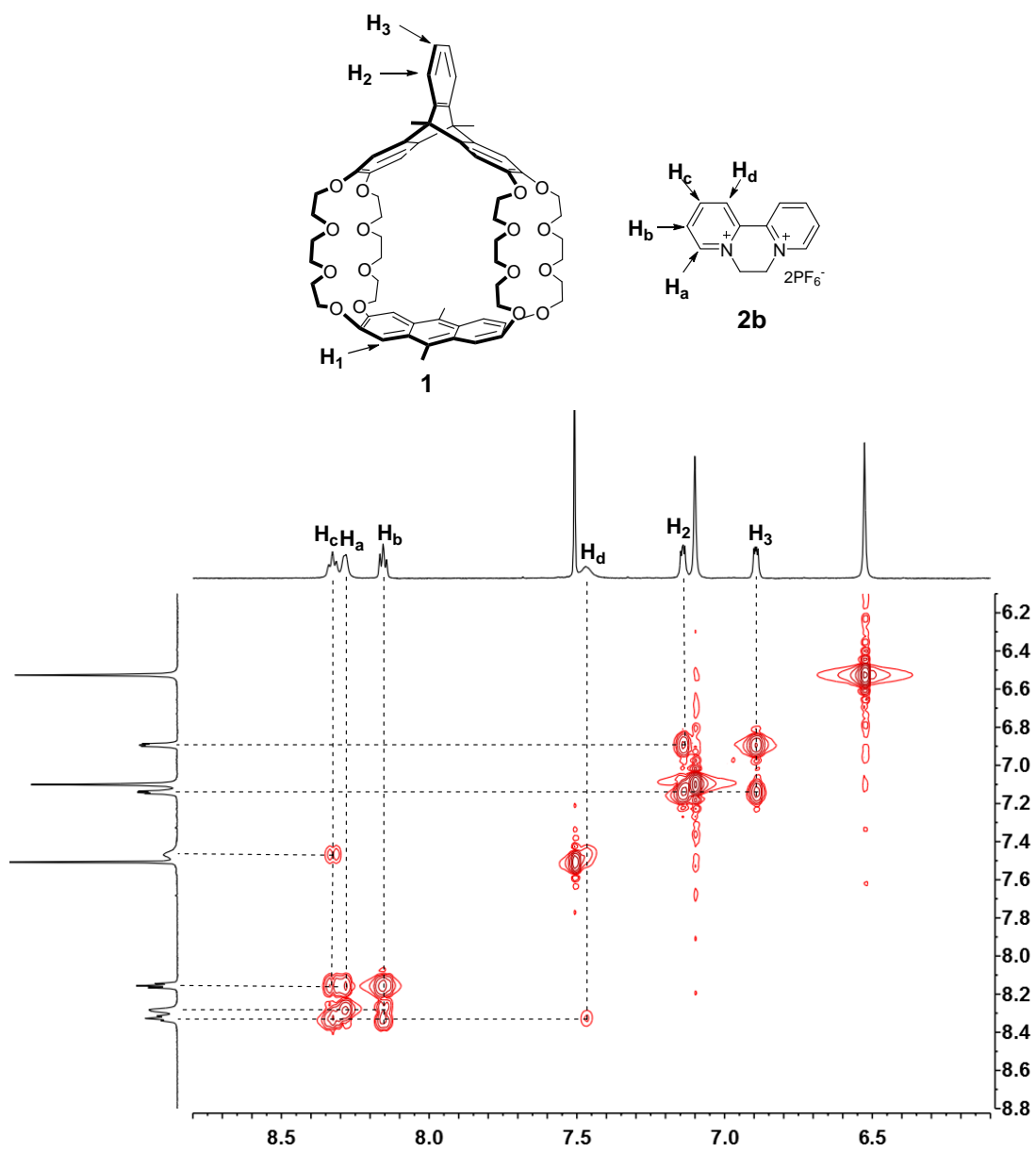


Figure S4. ^1H - ^1H COSY spectrum (300 MHz, $\text{CD}_3\text{CN}/\text{CDCl}_3=1:1$, v/v, 295 K) of host **1** and 1.0 equiv. of **2b**. $[1]_0 = 3.0$ mM.

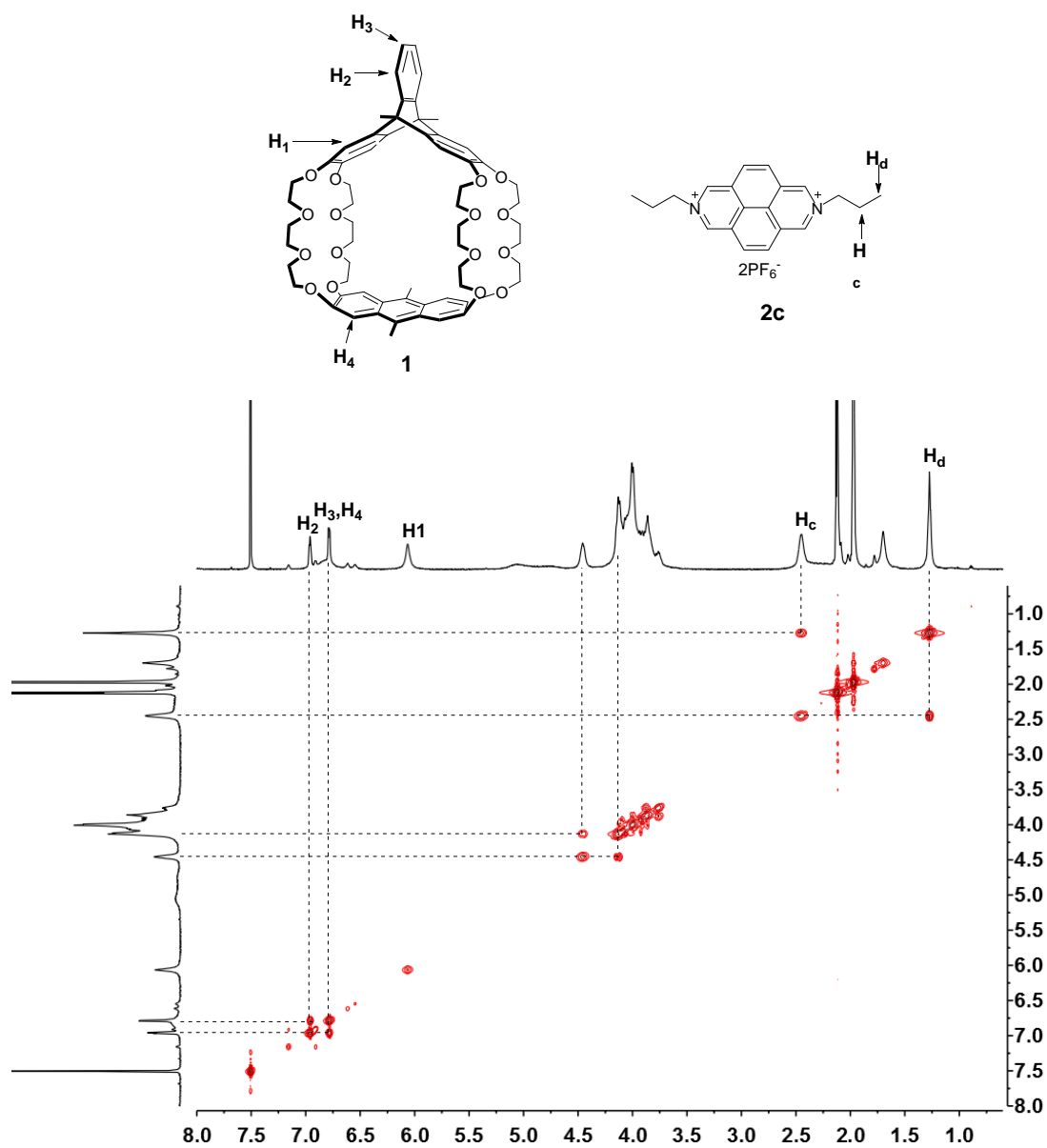


Figure S5. ^1H - ^1H COSY spectrum (300 MHz, $\text{CD}_3\text{CN}/\text{CDCl}_3=1:1$, v/v, 295 K) of host **1** and 1.0 equiv. of **2c**. $[1]_0 = 3.0$ mM.

3. Mole ratio between the host and the guests of the complexes

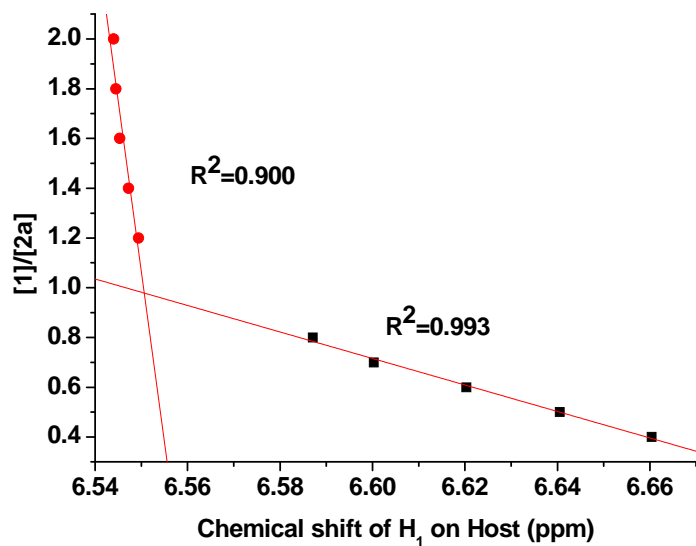


Figure S6. Mole ratio plot for the complexation between **1** and **2a** in $CD_3CN/CDCl_3$ (1:1, v/v) at 295 K. $[1]_0 = 3.0$ mM.

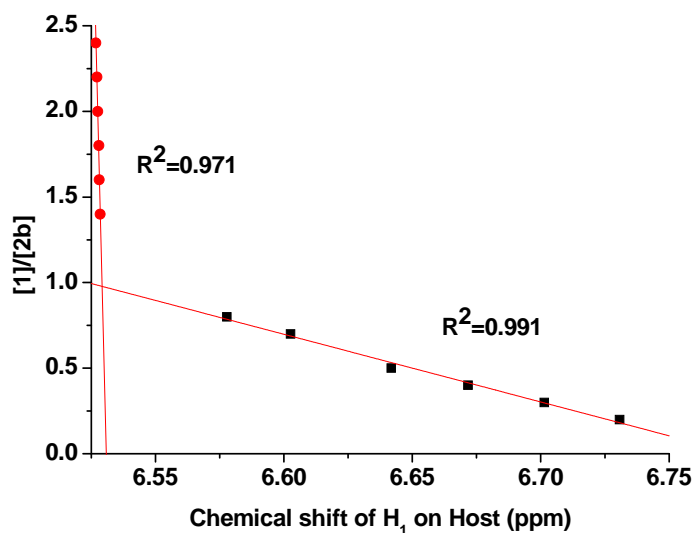


Figure S7. Mole ratio plot for the complexation between **1** and **2b** in $CD_3CN/CDCl_3$ (1:1, v/v) at 295 K. $[1]_0 = 3.0$ mM.

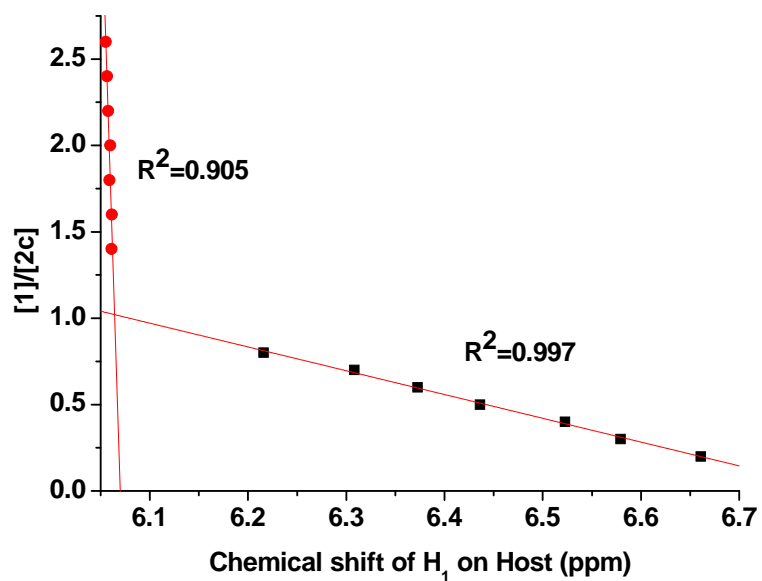


Figure S8. Mole ratio plot for the complexation between **1** and **2c** in $\text{CD}_3\text{CN}/\text{CDCl}_3$ (1:1, v/v) at 295 K. $[\mathbf{1}]_0 = 3.0$ mM.

4. ESI MS spectra of the complexes

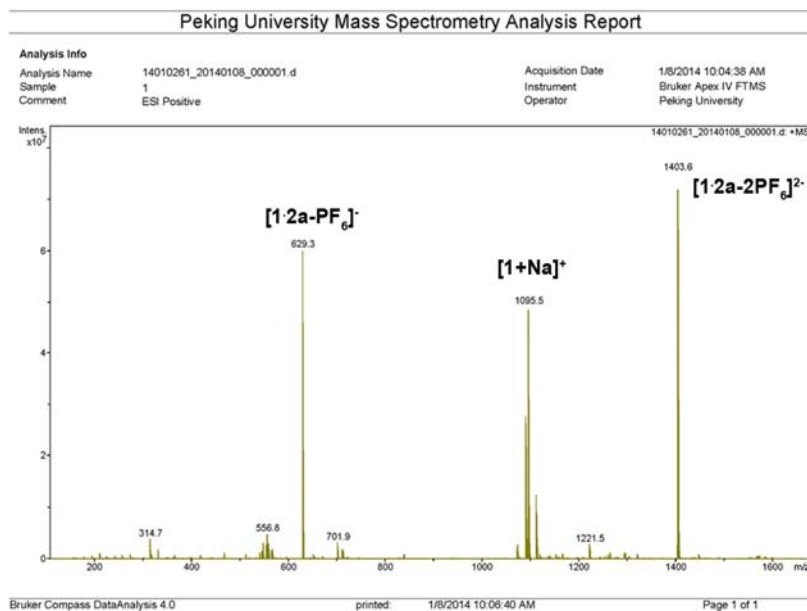


Figure S9. ESI MS of a solution of **1** and **2a** in acetonitrile-chloroform (1:1, v/v).

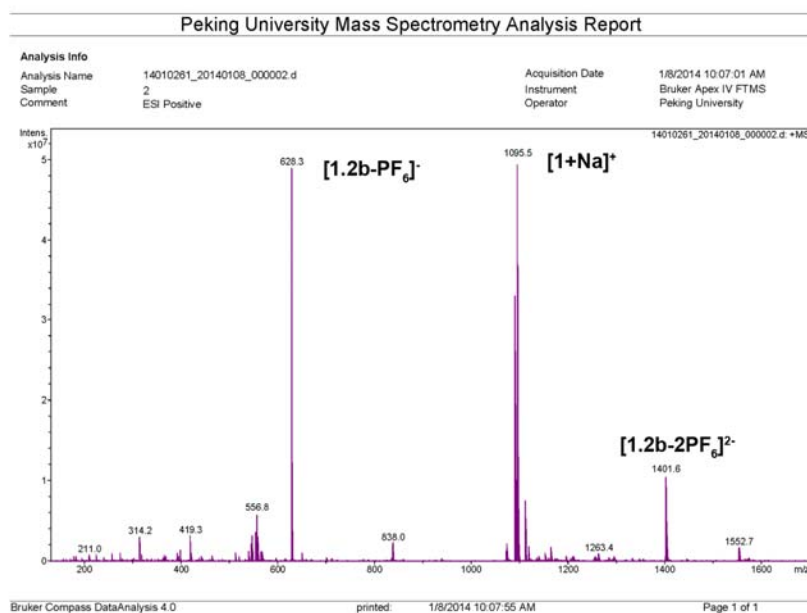


Figure S10. ESI MS of a solution of **1** and **2b** in acetonitrile-chloroform (1:1, v/v).

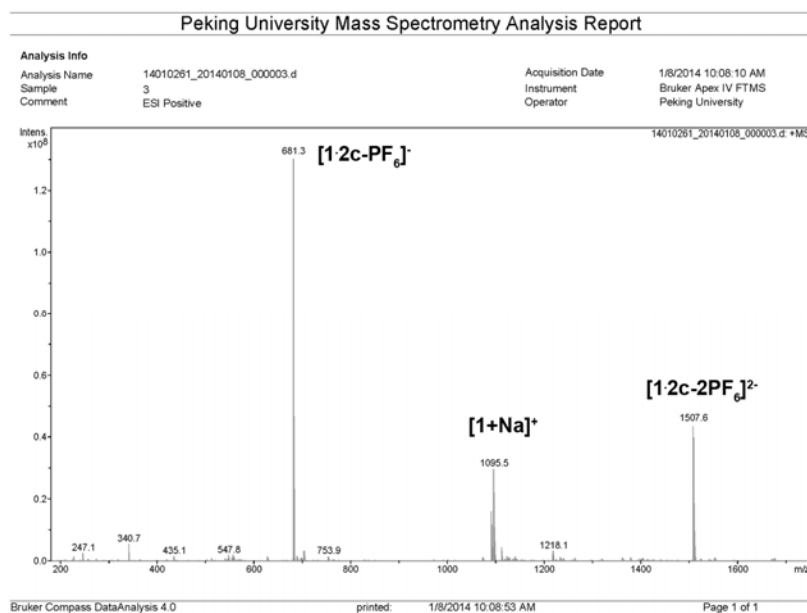


Figure S11. ESI MS of a solution of **1** and **2c** in acetonitrile-chloroform (1:1, v/v).

5. Crystal packing of the complexes

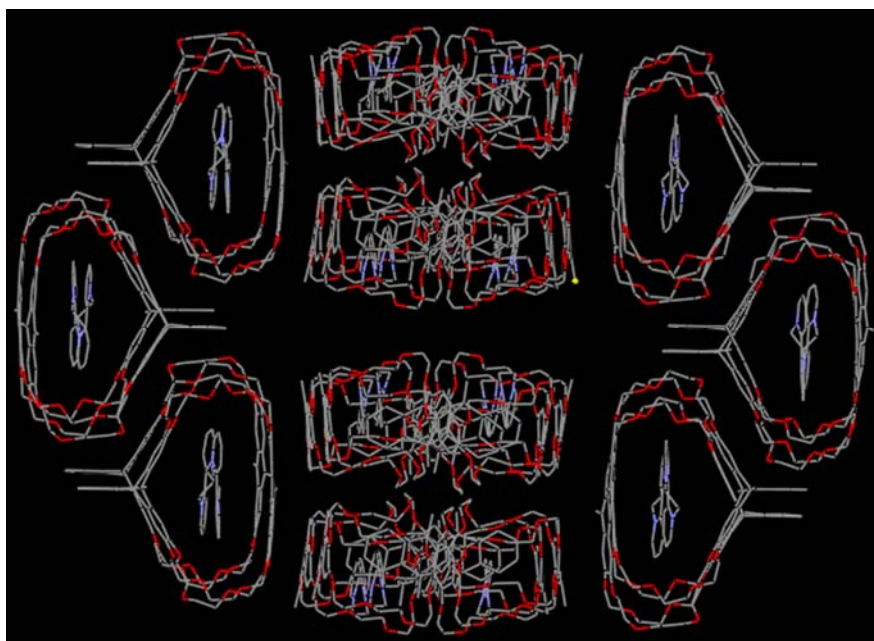


Figure S12. Crystal packing of complex **1·2b**. PF_6^- counterions and hydrogen atoms were omitted for clarity.

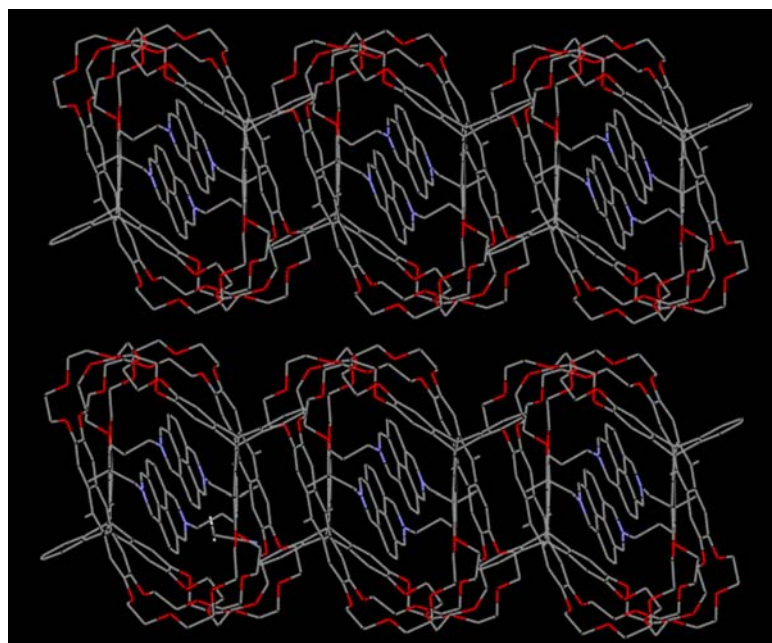


Figure S13. Crystal packing of complex **1·2c**. Solvent molecules, PF_6^- counterions and hydrogen atoms were omitted for clarity.

6. Structural parameters of the complexes

Table S1. Crystal data for **1·2a**

Empirical formula	C ₁₅₈ H ₁₈₇ F ₂₄ N ₉ O ₃₂ P ₄
Formula weight	3304.05
Crystal size (mm)	0.28 × 0.34 × 0.27
Temperature (K)	173(2)
Wavelength (Å)	0.71073
Crystal system	triclinic
Space group	<i>P</i> -1
<i>a</i> (Å)	17.928(4)
<i>b</i> (Å)	21.795(4)
<i>c</i> (Å)	26.597(5)
α (°)	68.93(3)
β (°)	79.43(3)
γ (°)	73.05(3)
<i>V</i> (Å ³)	9240(3)
<i>Z</i>	2
<i>D</i> _{calcd} (g cm ⁻³)	1.188
Absorption coefficient(mm ⁻¹)	0.130
<i>F</i> (000)	3460
θ / (°)	1.38~ 27.48
	-23 \leq h \leq 22
Limiting indices	-27 \leq k \leq 28
	-34 \leq l \leq 30
Reflections collected / unique	86969/41863 [R(int) = 0.0542]
Data/restraints/parameters	41863/0/2056
Goodness of fit on <i>F</i> ²	1.293
<i>R</i> ₁ , <i>wR</i> ₂ [I > 2σ(I)]	0.0980, 0.2526
<i>R</i> ₁ , <i>wR</i> ₂ [all data]	0.1368, 0.2765
Largest diff. peak and hole (e ⁻ Å ⁻³)	1.025 and -0.510

Table S2. Crystal data for **1·2b**

Empirical formula	C ₁₄₈ H ₁₆₆ F ₂₄ N ₄ O ₃₂ P ₄
Formula weight	3092.73
Crystal size (mm)	0.35 × 0.28 × 0.21
Temperature (K)	173(2)
Wavelength (Å)	0.71073
Crystal system	monoclinic
Space group	<i>C2/c</i>
<i>a</i> (Å)	33.928(7)
<i>b</i> (Å)	26.606(5)
<i>c</i> (Å)	41.298(8)
α (°)	90.00
β (°)	112.64(3)
γ (°)	90.00
<i>V</i> (Å ³)	34407(12)
<i>Z</i>	8
<i>D</i> _{calcd} (g cm ⁻³)	1.194
Absorption coefficient(mm ⁻¹)	0.135
<i>F</i> (000)	12912
θ / (°)	1.00~ 26.34 -27 \cong <i>h</i> \cong 40
Limiting indices	-30 \cong <i>k</i> \cong 31 -49 \cong <i>l</i> \cong 48
Reflections collected / unique	68412/30084 [R(int) = 0.0827]
Data/restraints/parameters	30084/85/1916
Goodness of fit on <i>F</i> ²	1.352
<i>R</i> ₁ , <i>wR</i> ₂ [I > 2σ(I)]	0.1200, 0.2932
<i>R</i> ₁ , <i>wR</i> ₂ [all data]	0.1696, 0.3159
Largest diff. peak and hole (e ⁻ Å ⁻³)	1.942 and -0.563

Table S3. Crystal data for **1·2c**

Empirical formula	C ₈₆ H ₁₀₂ F ₁₂ N ₄ O ₁₇ P ₂
Formula weight	1753.66
Crystal size (mm)	0.46 × 0.31 × 0.23
Temperature (K)	173(2)
Wavelength (Å)	0.71073
Crystal system	triclinic
Space group	<i>P</i> -1
<i>a</i> (Å)	12.547(2)
<i>b</i> (Å)	16.146(3)
<i>c</i> (Å)	21.955(4)
α (°)	109.304(4)
β (°)	91.346(3)
γ (°)	91.789(3)
<i>V</i> (Å ³)	4192.7(14)
<i>Z</i>	2
<i>D</i> _{calcd} (g cm ⁻³)	1.389
Absorption coefficient(mm ⁻¹)	0.149
<i>F</i> (000)	1840
θ / (°)	0.98~ 25.00 -14 \cong <i>h</i> \cong 14
Limiting indices	-18 \cong <i>k</i> \cong 19 -26 \cong <i>l</i> \cong 26
Reflections collected / unique	31625/14622 [R(int) = 0.0567]
Params/ restraints	14622/51/1135
Goodness of fit on <i>F</i> ²	1.110
<i>R</i> ₁ , <i>wR</i> ₂ [I>2σ(I)]	0.1017, 0.2583
<i>R</i> ₁ , <i>wR</i> ₂ [all data]	0.1187, 0.2755
Largest diff. peak and hole (e ⁻ Å ⁻³)	0.965 and -0.493

7. Ellipsoid plot for crystal structures of the complexes

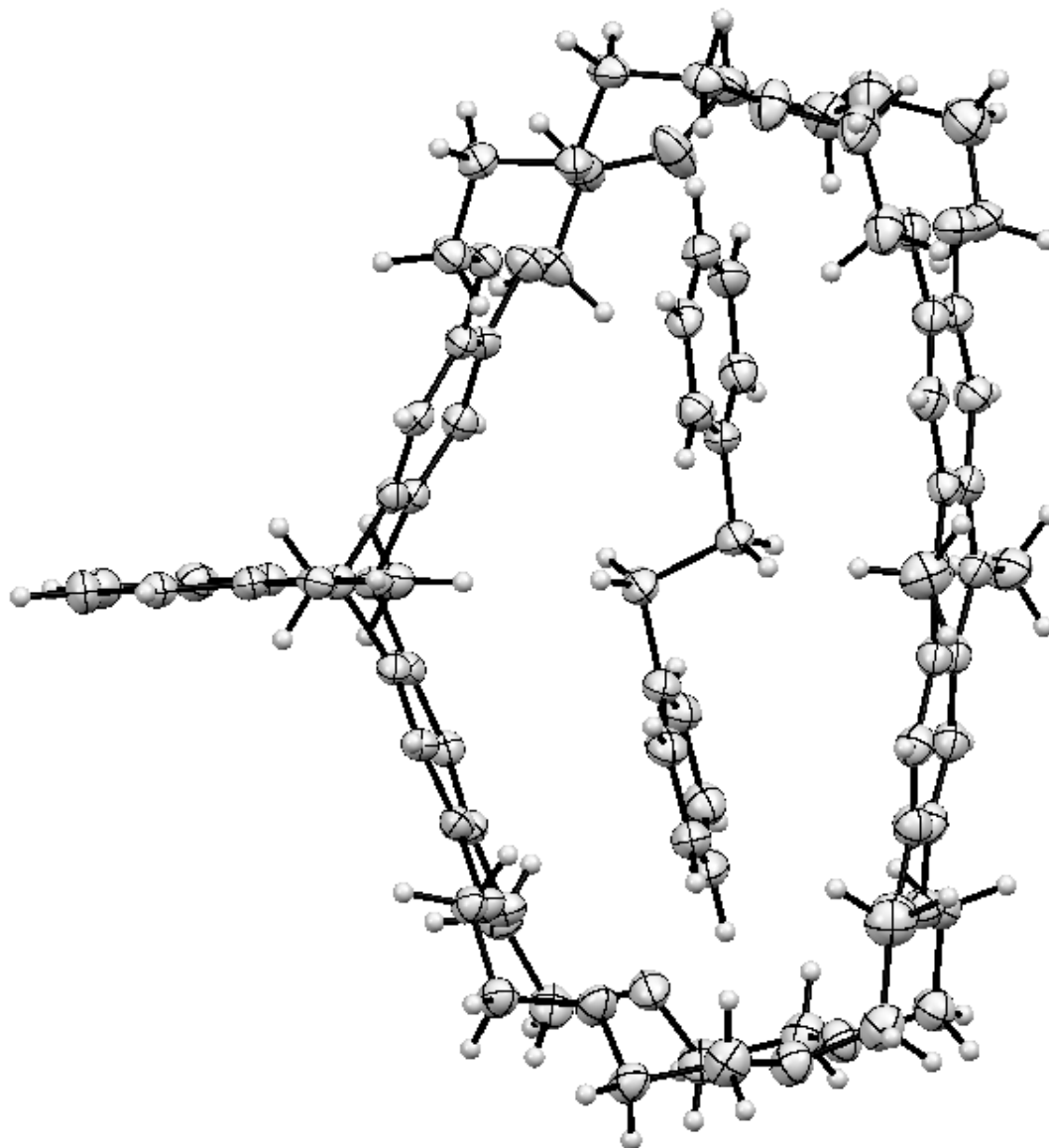


Figure S14. Ellipsoid plot for crystal structure of complex **1·2a** with probability level of 50%. Solvent molecules, and PF_6^- counterions were omitted for clarity.

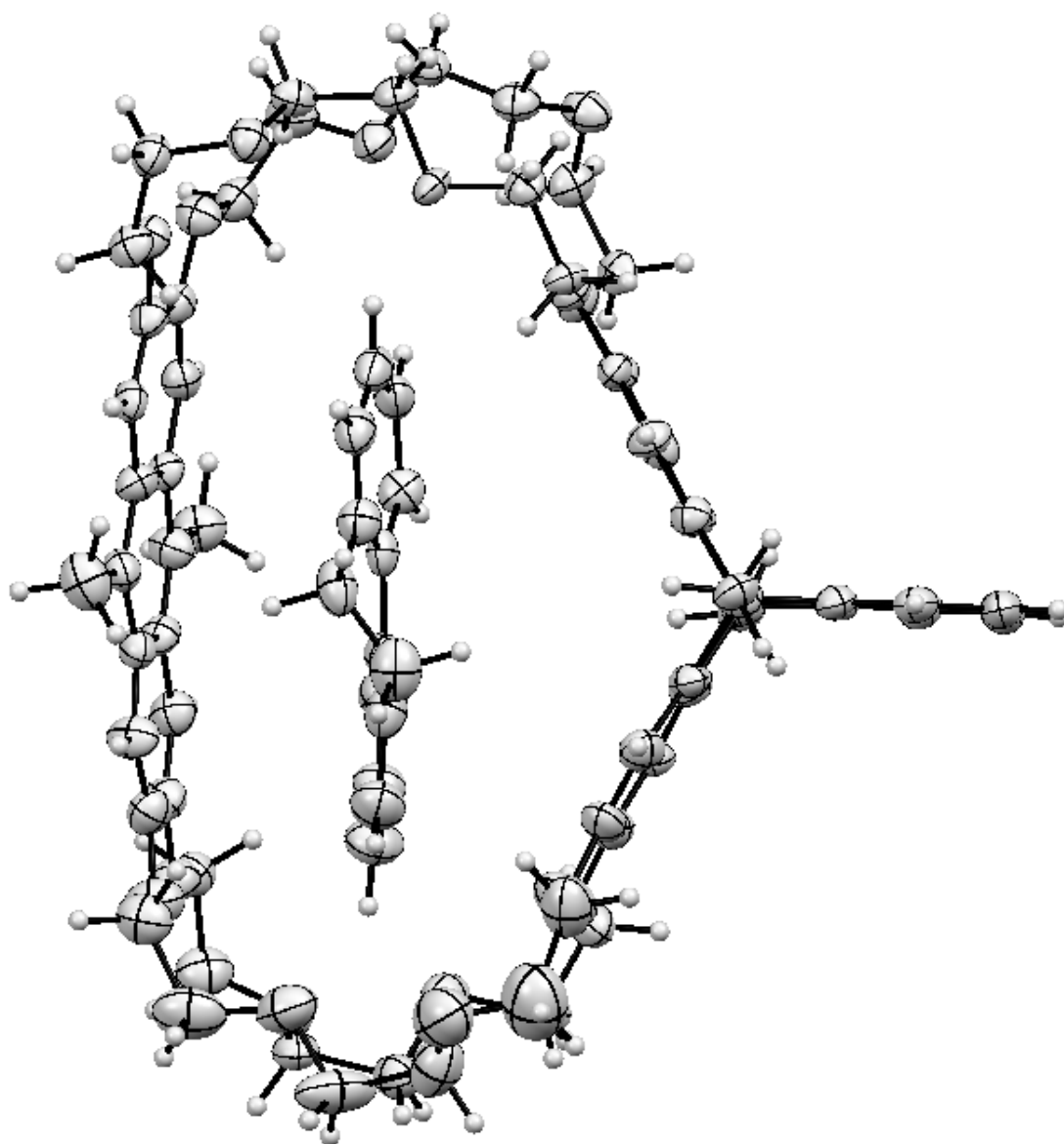


Figure S15. Ellipsoid plot for crystal structure of complex **1·2b** with probability level of 50%. Solvent molecules, and PF_6^- counterions were omitted for clarity.

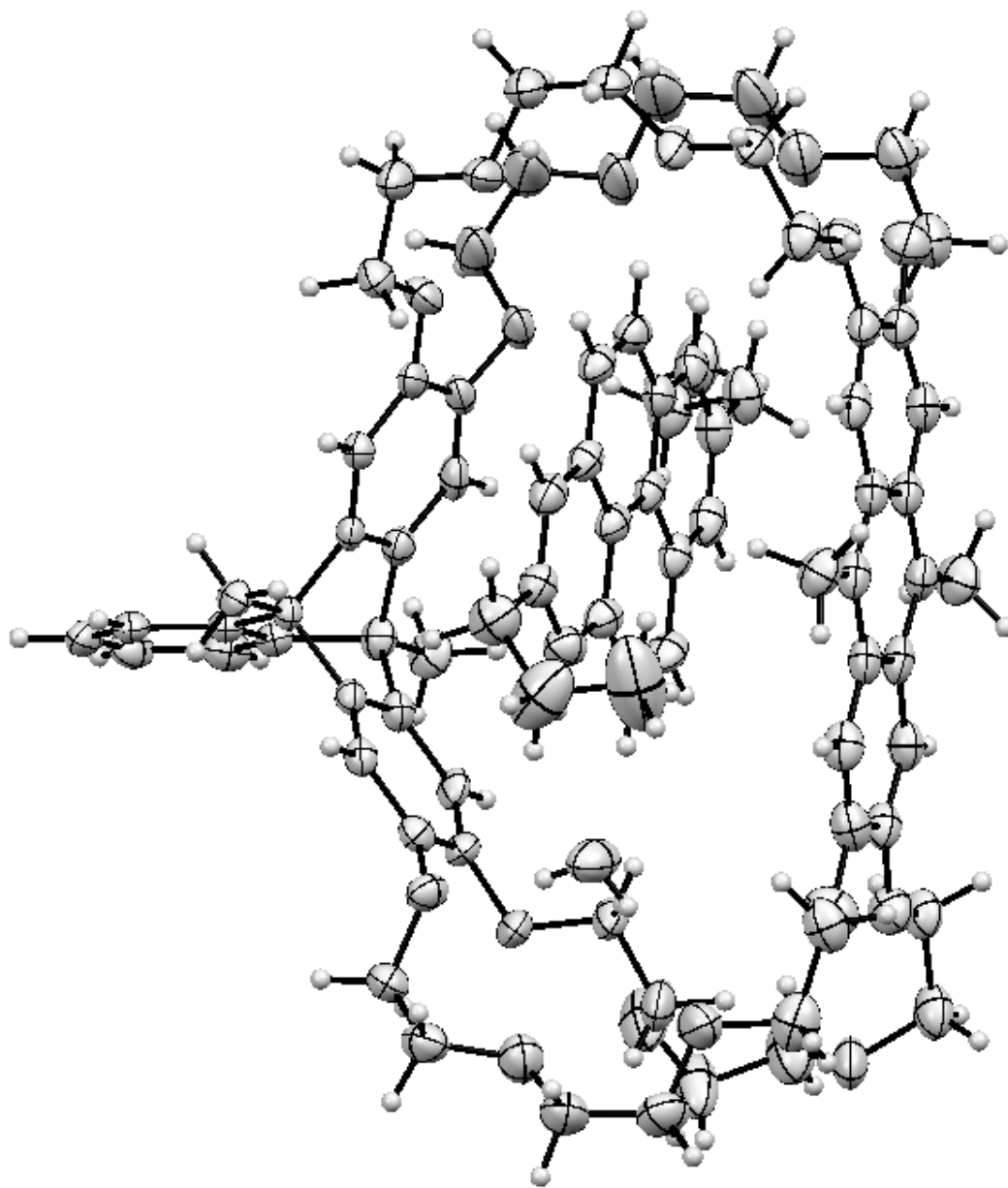


Figure S16. Ellipsoid plot for crystal structure of complex **1·2c** with probability level of 50%. Solvent molecules, and PF_6^- counterions were omitted for clarity.

8. K⁺ Ion-controlled binding and release of guests in the complexes

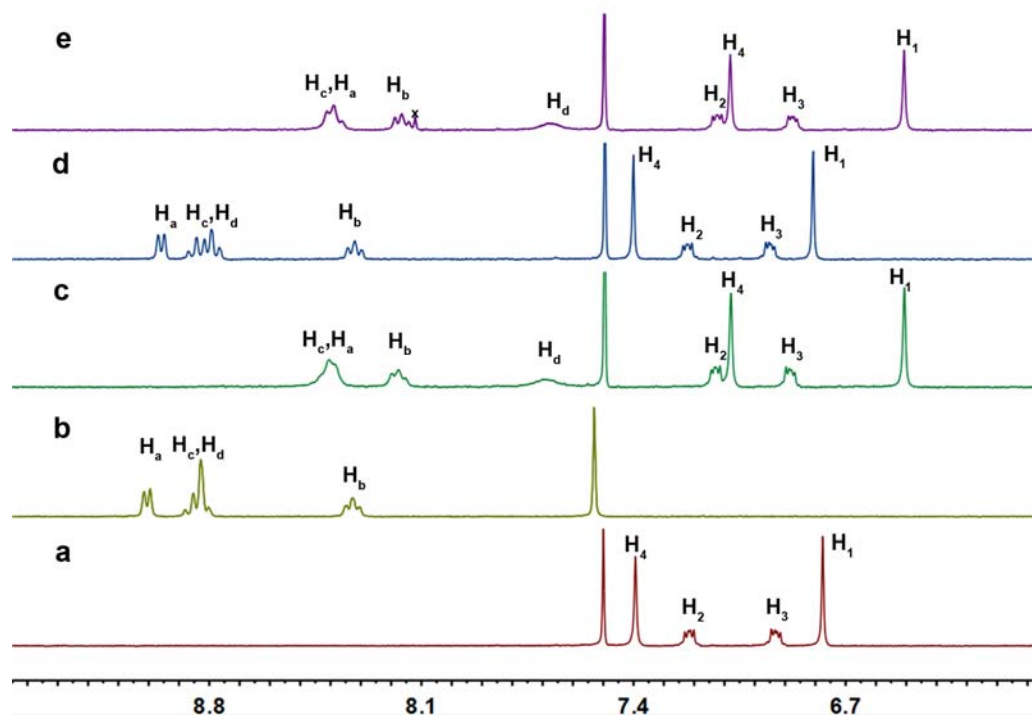


Figure S17. Partial ¹H NMR spectra (300 MHz, CD₃CN/CDCl₃= 1:1, v/v, 295K) of (a) free Host **1**, (b) free guest **2b**, (c) **1** and 1.0 equiv. of **2b**, (d) to the solution of c was added 4.0 equiv. of KPF₆, and (e) to the solution of d was added 6.0 equiv. of [18]-crown-6. [**1**]₀= 3.0 mM.

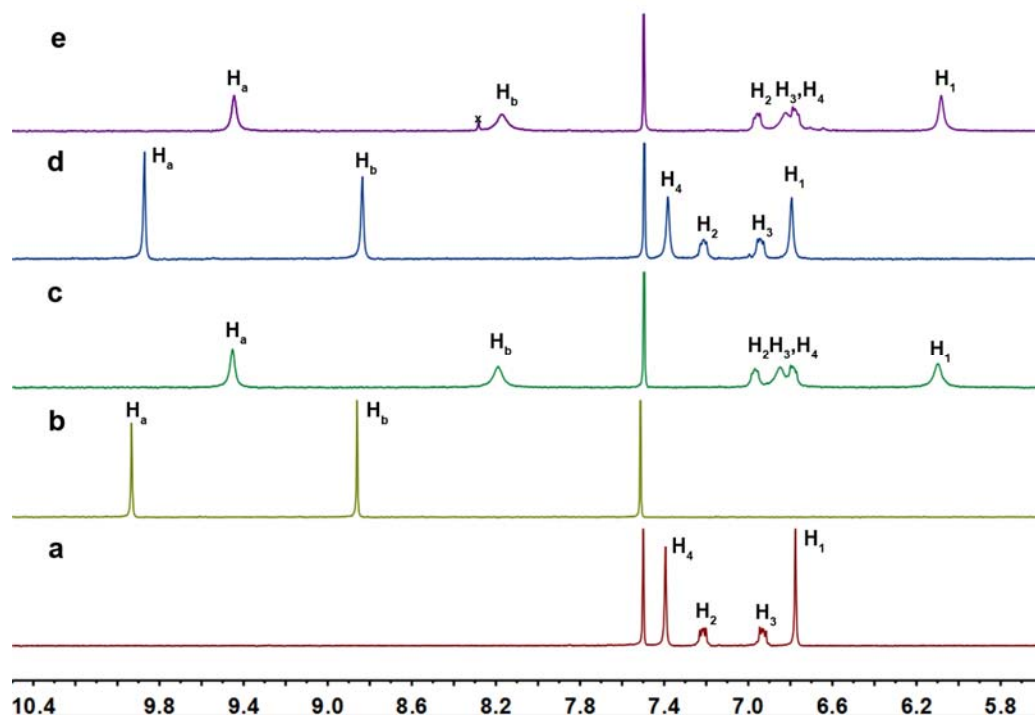


Figure S18. Partial ¹H NMR spectra (300 MHz, CD₃CN/CDCl₃= 1:1, v/v, 295K) of (a) free host **1**, (b) free guest **2c**, (c) **1** and 1.0 equiv. of **2c**, (d) to the solution of c was added 4.0 equiv. of KPF₆, and (e) to the solution of d was added 6.0 equiv. of [18]-crown-6. [1]₀= 3.0 mM.

9. Computational methods and coordinates for the guests

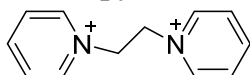
Computational methods

All calculations were performed using the Gaussian 03 suite of programs.¹ Optimized gas-phase structures were obtained using the density functional theory (DFT) method B3LYP, a combination of Becke's three-parameter hybrid exchange functional,²⁻³ as implemented⁴ in Gaussian 03, and the correlation functional of Lee, Yang and Parr,⁵ in conjunction with the 6-31G(d,p) basis set.

Coordinates for guests 2a-2c

B3LYP/6-31G(d,p) optimized geometries (in Cartesian coordinate form) and their B3LYP/6-31G(d,p) electronic energies, enthalpies and free energies for the three guests are listed below for gas phase.

1,2-bis(pyridium) ethane (BPE²⁺) (2a)



gas phase

B3LYP/6-31G(d,p)

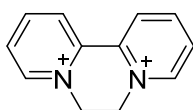
SCF Energy (no zpe)= -574.423860 au; Energy + zpe (0K)= -574.412295 au

Enthalpy (298K)= -574.411351 au; Gibbs free energy (298K)= -574.463826 au

C	-3.889554000	-1.206552000	-0.082394000
C	-2.536838000	-1.182333000	0.204582000
C	-2.536732000	1.182278000	0.204957000
C	-3.889439000	1.206707000	-0.082005000
C	-4.580514000	0.000129000	-0.229684000
H	-4.391792000	-2.162613000	-0.178598000
H	-1.954396000	-2.085098000	0.346328000
H	-1.954200000	2.084936000	0.347000000
H	-4.391606000	2.162835000	-0.177905000
H	-5.643736000	0.000221000	-0.448396000
C	-0.422587000	-0.000193000	0.647714000
C	0.422421000	-0.000044000	-0.647466000
H	-0.209054000	-0.881614000	1.255907000
H	-0.208960000	0.881043000	1.256143000
H	0.209152000	-0.881362000	-1.255892000
H	0.209069000	0.881380000	-1.255708000
C	2.536753000	-1.182305000	-0.204591000
C	2.536775000	1.182329000	-0.204896000
C	3.889522000	-1.206597000	0.082208000

H	1.954215000	-2.085007000	-0.346299000
C	3.889543000	1.206660000	0.081888000
C	1.954248000	2.085002000	-0.346826000
C	4.580563000	0.000043000	0.229459000
H	4.391737000	-2.162683000	0.178319000
H	4.391789000	2.162756000	0.177748000
H	5.643807000	0.000070000	0.448042000
N	-1.876957000	-0.000084000	0.341248000
N	1.876992000	-0.000002000	-0.341033000

Diquat (DQ²⁺) (2b)



gas phase

B3LYP/6-31G(d,p)

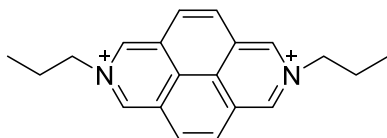
SCF Energy (no zpe)= -573.217833 au; Energy + zpe (0K)= -573.208305 au

Enthalpy (298K)= -573.207360 au; Gibbs free energy (298K)= -573.253400 au

C	-3.539594000	-0.375401000	0.000010000
C	-2.787666000	0.784058000	-0.000127000
C	-0.741012000	-0.426109000	0.000040000
C	-1.476493000	-1.614739000	0.000216000
C	-2.870030000	-1.598966000	0.000204000
C	-0.768274000	2.088344000	-0.000236000
C	0.741012000	-0.426109000	-0.000031000
C	0.768274000	2.088344000	0.000260000
C	2.787666000	0.784058000	0.000111000
H	3.239625000	1.769807000	0.000234000
C	3.539594000	-0.375401000	-0.000030000
C	2.870030000	-1.598966000	-0.000210000
C	1.476493000	-1.614739000	-0.000209000
H	-1.131619000	2.615046000	-0.884775000
H	-4.621976000	-0.310082000	-0.000021000
H	-3.239625000	1.769807000	-0.000256000
H	-0.969841000	-2.569135000	0.000384000
H	-3.421278000	-2.533917000	0.000345000
H	1.131618000	2.615031000	0.884808000
H	4.621976000	-0.310082000	-0.000013000
H	3.421278000	-2.533917000	-0.000351000
H	0.969841000	-2.569135000	-0.000368000

N	-1.429489000	0.752813000	-0.000095000
N	1.429489000	0.752813000	0.000096000
H	1.132134000	2.615485000	-0.883805000
H	-1.132133000	2.615469000	0.883838000

2,7-diazapyrenium (DAP²⁺) (2c)



gas phase

B3LYP/6-31G(d,p)

SCF Energy (no zpe)= -884.052080 au; Energy + zpe (0K)= -884.033822 au

Enthalpy (298K)= -884.032878 au; Gibbs free energy (298K)= -884.099139 au

C	-2.933573000	0.840376000	0.000191000
C	-1.551232000	1.054739000	0.000104000
C	-0.703058000	-0.083886000	0.000015000
C	-1.259206000	-1.393569000	-0.000024000
C	-2.648903000	-1.505161000	0.000041000
C	-0.970715000	2.372601000	0.000112000
C	0.703059000	0.083887000	-0.000040000
C	1.259206000	1.393569000	0.000010000
C	0.382793000	2.536374000	0.000080000
C	2.648904000	1.505161000	-0.000009000
H	3.154333000	2.464107000	0.000041000
C	2.933574000	-0.840376000	-0.000146000
C	1.551233000	-1.054739000	-0.000132000
C	0.970715000	-2.372601000	-0.000197000
C	-0.382792000	-2.536374000	-0.000148000
H	-0.813091000	-3.532536000	-0.000205000
H	1.625379000	-3.238036000	-0.000297000
H	-1.625379000	3.238037000	0.000160000
H	-3.640060000	1.658057000	0.000191000
H	-3.154333000	-2.464106000	0.000048000
H	0.813092000	3.532536000	0.000110000
H	3.640061000	-1.658056000	-0.000219000
C	4.944724000	0.653756000	-0.000065000
C	5.827010000	-0.588437000	0.000002000
H	5.137932000	1.263832000	0.886775000
H	5.137956000	1.263753000	-0.886953000
C	7.307218000	-0.172666000	-0.000010000

H	5.624697000	-1.199219000	0.887803000
H	5.624699000	-1.199305000	-0.887740000
H	7.940307000	-1.061969000	-0.000007000
H	7.558799000	0.415425000	-0.887290000
H	7.558811000	0.415441000	0.887256000
C	-4.944723000	-0.653755000	0.000538000
C	-5.827012000	0.588436000	0.000051000
H	-5.137889000	-1.263472000	0.887646000
H	-5.137997000	-1.264113000	-0.886080000
C	-7.307222000	0.172662000	-0.000452000
H	-5.625157000	1.199385000	0.887838000
H	-5.624243000	1.199138000	-0.887704000
H	-7.940311000	1.061965000	-0.001414000
H	-7.558344000	-0.416054000	-0.887446000
H	-7.559267000	-0.414802000	0.887116000
N	3.445178000	0.408625000	-0.000078000
N	-3.445177000	-0.408623000	0.000198000

10. References

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