

Supporting Information for

Asymmetric Synthesis of 3,3,5,5-Tetrasubstituted 1,2-Dioxolanes:

Total Synthesis of Epiplakinic Acid F

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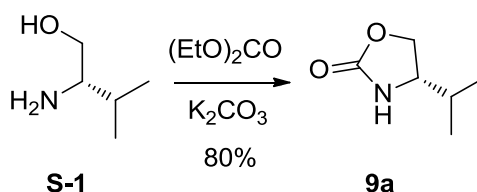
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General Information

All non-aqueous reactions were carried out using oven-dried glassware under a positive pressure of dry nitrogen unless otherwise noted. Solvents were predried over activated 4Å molecular sieves and were refluxed over magnesium (methanol), sodium (toluene, THF, Et₂O, benzene, dioxane, cyclohexane), or calcium hydride (DCM, DCE, EA, CH₃CN) under an argon atmosphere and collected by distillation. All evaporation of organic solvents was carried out with a rotary evaporator. Column chromatography was performed on silica gel 60 (Huanghai, 300-400mesh). Yields refer to chromatographically and spectroscopically pure compounds, unless otherwise stated. ¹H and ¹³C NMR spectra were recorded on a Varian Mercury 300 (300 MHz) or Bruker AM 400 (400 MHz) spectrometer. ¹H and ¹³C NMR spectra were referenced internally to residual protio-solvent (¹H) or solvent (¹³C) resonances and are reported relative to tetramethylsilane. Data are reported as follows: brs = broad singlet, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet; coupling constants in Hz. HPLC analyses on a Agilent 1100 Series chromatograph. Infrared spectra were prepared as KBr pellets and were recorded on a Bio-Rad FTS-185 FT-IR spectrometer. Optical rotations were measured with a Perkin Elmer 241 polarimeter in a 1 dm cuvette. Mass spectra were recorded by the mass spectrometry service of Shanghai Institute of Organic Chemistry.

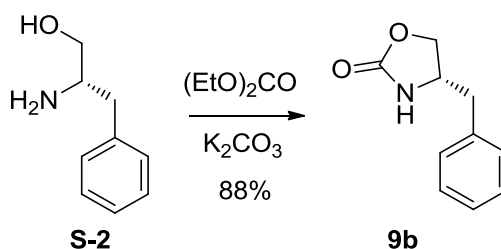
Experimental Details and Characteristic Data for Compounds

Synthesis of (S)-4-Isopropylloxazolidin-2-one (9a):¹



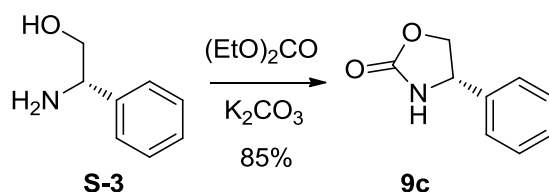
Into a 25-mL flask equipped with a 10 cm Vigreux column was added **S-1** (2 g, 19.5 mmol), diethyl carbonate (4.7 ml, 38.8 mmol), and anhydrous potassium carbonate (2.68 g, 19.5 mmol). The mixture was heated at 130-135 °C (oil temperature) and ethanol (1.8 g, 38.8 mmol) was distilled (ca. 1 hour). The resultant mixture was then cooled to room temperature and dissolved in CH₂Cl₂ (20 mL), and the mixture was washed with water (10 mL × 2) and brine (10 mL), dried over Na₂SO₄, and concentrated. The residual was crystallized by EtOAc to give **9a** (white needles, 2.0g, 80%). **M.p.**: 73 °C (Lit¹: 69-70 °C); $[\alpha]_D^{25} = +11.3$ (c, 0.5, CHCl₃), [Lit²: $[\alpha]_D^{25} = +13.0$ (c, 2.6, CHCl₃)]; **IR** (Film): 3271, 2976, 2961, 2915, 2875, 1751, 1725, 1472, 1406, 1247, 1091, 1010, 936, 769, 710 cm⁻¹; **MS** (EI): *m/z* (relative intensity), 129 (M⁺, 6); **¹H NMR** (300 MHz, CDCl₃): δ (ppm) 7.00 (s, 1H), 4.42-4.48 (t, *J* = 8.4 Hz, 1H), 4.11 (dd, *J* = 8.4, 6.3 Hz, 1H), 3.61 (q, *J* = 6.9 Hz, 1H), 1.70-1.76 (m, 1H), 0.97 (d, *J* = 6.9 Hz, 3H), 0.90 (d, *J* = 6.9 Hz, 3H).

Synthesis of (S)-4-Benzoyloxazolidin-2-one (**9b**):¹



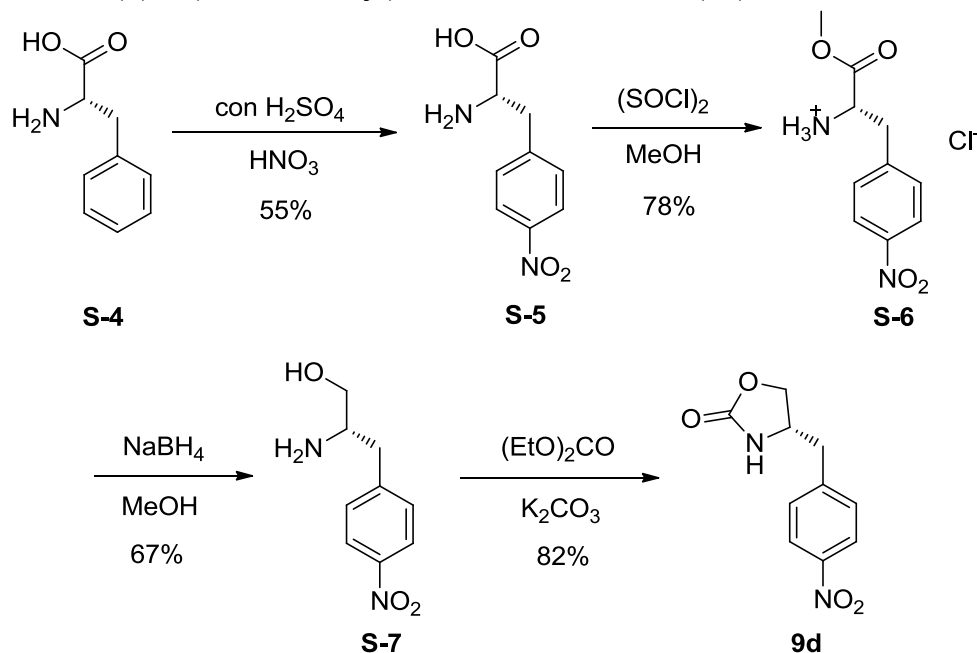
Into a 25-mL flask equipped with a 10 cm Vigreux column was added **S-2** (1 g, 6.6 mmol), diethyl carbonate (1.6 ml, 13.2 mmol), and anhydrous potassium carbonate (0.91 g, 6.6 mmol). The mixture was heated at 130-135 °C and ethanol (0.60 g, 13.2 mmol) was distilled (ca. 1 hour). The resultant mixture was then cooled to room temperature and dissolved in CH₂Cl₂ (20 mL), and the mixture was washed with water (10 mL × 2) and brine (10 mL), dried over Na₂SO₄, and concentrated. The residual was purified by flash chromatography (hexane/EtOAc, 2/1) to obtain **9b** (white solid, 1.03 g, 88 %). **M.p.**: 89 °C (Lit³: 89-90 °C); $[\alpha]_D^{25} = -62.7$ (c, 1.0, CHCl₃), [Lit⁴: $[\alpha]_D^{22} = -63.3$ (c, 1.0, CHCl₃)]; **IR** (Film): 3281, 2924, 1751, 1709, 1404, 1245, 1096, 1063, 1021, 943, 758, 708, 618, 528 cm⁻¹; **MS** (EI): *m/z*, 177 (M⁺); **¹H NMR** (300 MHz, CDCl₃): δ (ppm) 7.13-7.42 (m, 5H), 5.82 (br, 1H), 4.45 (t, *J* = 7.5 Hz, 1H), 4.03-4.20 (m, 2H), 2.88 (d, *J* = 6.3 Hz, 2H).

Synthesis of (S)-4-phenyloxazolidin-2-one (**9c**):¹



Into a 25-mL flask equipped with a 10 cm Vigreux column was added **S-3** (1 g, 7.3 mmol), diethyl carbonate (1.75 ml, 14.6 mmol), and anhydrous potassium carbonate (1.0 g, 7.3 mmol). The mixture was heated at 130-135 °C and ethanol (0.68 g, 14.6 mmol) was distilled (ca. 1 hour). The resultant mixture was then cooled to room temperature and dissolved in CH₂Cl₂ (20 mL), and the mixture was washed with water (10 mL × 2) and brine (10 mL), dried over Na₂SO₄, and concentrated. The residual was purified by flash chromatography (hexane/EtOAc, 1/1) to obtain **9b** (white solid, 1.01 g, 85 %). **M.p.**: 127-128 °C (Lit⁵: 124-126 °C); **[α]_D²⁵** = +47.6 (c, 1.2, CHCl₃), [Lit⁵: **[α]_D²³** = +48.1 (c, 1.0, CHCl₃); **IR** (Film): 3250, 3031, 1743, 1705, 1488, 1402, 1236, 1098, 1077, 1002, 967, 924, 768, 697 cm⁻¹; **MS** (ESI): *m/z*, 164.1 (M+H⁺); **¹H NMR** (300 MHz, CDCl₃): δ (ppm) 7.34-7.45 (m, 5H), 6.30 (s, 1H), 4.97 (t, *J* = 7.8 Hz, 1H), 4.74 (t, *J* = 8.7 Hz, 1H), 4.19 (t, *J* = 6.9 Hz, 1H).

Synthesis of (S)-4-(4-Nitrobenzyl)-1,3-oxazolidin-2-one (**9d**):⁶



Synthesis of (S)-2-amino-3-(4-nitrophenyl)propanoic acid (**S-5**):

To a stirring concentrated sulfuric acid (40 mL) at 0 °C was added **S-4** (30 g, 182 mmol). After **S-4** was dissolved, the concentrated nitric acid (21 mL, 364 mmol) was gradually added over 45 min. The reaction mixture was then stirred for another 30 min at 0 °C and was allowed to warm to room temperature and stirred for 1 hour. The reaction mixture was poured onto ice/water (200 L), and ammonia (26~28% in water) was added until adjust pH to 7~8. Then the solution was filtrated and the precipitate was washed by cooled water (20 mL 3), dried out to give the crude product. The crude

product was recrystallized from water (250 mL) to give **S-5** (white crystals, 21 g, 55%). **M.p.**: 236 °C (Lit⁶: 239-241 °C); $[\alpha]_D^{25} = +5.6$ (c, 1.0, 3M HCl), [Lit⁷: $[\alpha]_D^{20} = +6.8$ (c, 1.0, 3M HCl)]; **IR** (Film): 3293, 2895, 2651, 1700, 1647, 1616, 1571, 1538, 1516, 1419, 1350, 1332, 1312, 878, 863 cm⁻¹; **MS** (EI): *m/z* (relative intensity) 43 (100), 41(75), 94 (50), 120 (9), 105(8), 129(7), 144 (7); **¹H NMR** (300 MHz, MeOH): δ (ppm) 8.05 (d, *J* = 8.1 Hz, 2H), 7.33 (d, *J* = 8.1 Hz, 2H), 3.86 (t, *J* = 6.9 Hz, 1H), 3.05-3.25 (m, 2H).

Synthesis of (S)-Methyl 4-Nitrophenylalanate Hydrochloride (S-6):

To a solution of **S-5** (21 g, 100 mmol) in methanol (200 mL) was added thionyl chloride (10.6 mL, 150 mmol). And the reaction mixture was stirred for overnight at room temperature. Then the solvent was removed under reduced pressure and the residue was dried out to give **S-7** (white solid, 26 g, 100 %). **M.p.**: 219 °C (Lit⁶: 227-228 °C); **MS** (ESI): *m/z* (relative intensity) 235([M-Cl]⁺); **¹H NMR** (300MHz, MeOH): δ (ppm) 8.07 (d, *J* = 8.1Hz, 2H), 7.35 (d, *J* = 8.1Hz, 2H), 4.35 (t, *J* = 6.9Hz, 1H), 3.65 (s, 3H), 3.17-3.34 (m, 2H); **IR** (Film): 2800-3400(br), 2984, 2917, 2677, 2631, 1744, 1604, 1544, 1519, 1508, 1492, 1454, 1349, 1241, 1146, 752cm⁻¹; $[\alpha]_D^{25} = +43.9$ (c, 1.0, H₂O), [Lit⁸: $[\alpha]_D^{20} = +34.2$ (c, 1.0, EtOH)];

Synthesis of (S)-2-Amino-3-(4-nitrophenyl)propanol (S-7):

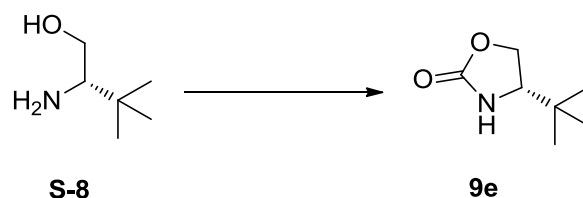
To a stirring solution of sodium borohydride (0.53 g, 14 mmol) in EtOH/water (1:1) (30 mL) at 0 °C was added the amino ester **S-7** (0.9 g, 3.5 mmol). The mixture was refluxed for 4 hours and then allowed to cool to room temperature. The solvent was filtrated and the precipitate was washed by EtOH. The filtrate was combined and the ethanol was then removed under vacuum. The remaining water was extracted with EtOAc (20 mL × 6) and the combined organic extracts was washed with brine (50 mL), dried over Na₂SO₄, and concentrated. The yellow solid was recrystallized from EtOH to give **S-7** (yellow crystals, 230 mg, 67 %). **M.p.**: 145 °C (Lit⁶: 141-142 °C); $[\alpha]_D^{25} = -26.5$ (c, 1.0, MeOH), [Lit⁹: $[\alpha]_D^{20} = -26.6$ (c, 2.3, MeOH)]; **IR** (Film): 3352, 3297, 3106, 3079, 2944, 2905, 2837, 1607, 1596, 1585, 1515, 1344, 1109, 1058, 858, 703 cm⁻¹; **MS** (ESI): *m/z* (relative intensity) 197(M+H⁺); **¹H NMR** (300 MHz, CDCl₃): δ (ppm) 8.17 (d, *J* = 8.7 Hz, 2H), 7.37 (d, *J* = 8.7 Hz, 2H), 3.65 (dd, *J* = 3.9, 10.8 Hz, 1H), 3.41 (dd, *J* = 7.2, 10.8 Hz, 1H), 3.18 (m, 1H), 2.92 (dd, *J* = 3.9, 13.5 Hz, 1H), 2.67 (dd, *J* = 8.4, 13.5 Hz, 1H), 1.6 (br, 2H).

Synthesis of (S)-4-(4-Nitrobenzyl)-1,3-oxazolidin-2-one (9d):

Into a 50-mL flask equipped with a 10-cm Vigreux column was added **S-7** (5 g, 25.5 mmol), diethyl carbonate (6.19 ml, 51 mmol), and anhydrous potassium carbonate (3.52 g, 25.5 mmol). The mixture was heated at 130-135 °C (oil temperature) for 1 hour and ethanol was distilled. The resultant mixture was then cooled to room temperature and dissolved in CH₂Cl₂ (50 mL), and the mixture was washed with water (50 mL×2) and brine (50 mL), dried over Na₂SO₄, and concentrated. The residual was purified by flash chromatography (hexane/EtOAc, 3/1) to obtain **9d** (yellow solid, 4.63 g, 87 %). **M.p.**: 132 °C; $[\alpha]_D^{25} = -65.0$ (c, 0.8, CHCl₃),

[Lit¹⁰: $[\alpha]_D^{20} = -63$ (c, 5.5, CHCl₃)]; **IR** (Film): 3254, 3104, 3077, 3000, 2973, 2192, 2857, 1774, 1610, 1599, 1536, 1420, 1346, 1253, 1224, 1109, 703 cm⁻¹; **MS** (ESI): *m/z* (relative intensity) 223 (M+H⁺); **¹H NMR** (300 MHz, CDCl₃): δ (ppm) 8.21-8.24 (d, *J* = 8.7 Hz, 2H), 8.37-8.40 (d, *J* = 8.7 Hz, 2H), 5.57 (s, 1H), 4.51 (m, 1H), 4.17 (m, 2H), 3.01 (d, *J* = 6.0 Hz, 2H).

Synthesis of (S)-4-(tert-butyl)oxazolidin-2-one (9e):¹



Into a 25-mL flask equipped with a 10-cm Vigreux column was added **S-8** (800 mg, 6.8 mmol), diethyl carbonate (1.6 g, 13.7 mmol), and anhydrous potassium carbonate (1.9 g, 13.7 mmol). The mixture was heated at 130-135 °C for 1 hour and ethanol was distilled. The resultant mixture was then cooled to room temperature and dissolved in CH₂Cl₂ (20 mL), and the mixture was washed with water (10 mL × 2) and brine (10 mL), dried over Na₂SO₄, and concentrated. The residual was purified by flash chromatography (hexane/EtOAc, 5/1) to obtain **9e** (white solid, 640 mg, 65 %). **M.p.**: 116-118 °C (Lit¹¹: 120 °C); $[\alpha]_D^{25} = +13.5$ (c, 0.5, CHCl₃), [Lit⁴: $[\alpha]_D^{21} = +12.8$ (c, 1.0, CHCl₃)]; **IR** (Film): 3299, 3244, 3008, 2970, 1744, 1717, 1480, 1401, 1238, 1101, 1052, 985, 962 cm⁻¹; **MS** (EI): *m/z*, 143 (M⁺); **¹H NMR** (300 MHz, CDCl₃): δ (ppm) 6.98 (br, 1H), 4.37 (t, *J* = 9.0 Hz, 1H), 4.20 (dd, *J* = 9.0, 6.0 Hz, 1H), 3.61 (dd, *J* = 9.0, 6.0 Hz, 1H), 0.93 (s, 9H).

Tables for Feldman Reaction Conditions

Table S1. Solvent effective studies of the Feldman reaction of *trans*-**10a** and *trans*-**11a**

Entry ^a	Solvent	Yield % ^b	<i>trans</i> - 12a / <i>trans</i> - 13a / <i>cis</i> - 14a ^c
1	DCM	quant	25/64/11
2	CH ₃ CN	quant	26/64/10
3	DMF	90%	26/64/10
4	Et ₂ O	quant	23/67/10
5	THF	99%	25/64/11
6	toluene	94%	23/67/10
7	EtOH	No product	complex

^a Reaction conditions: 1:1 mixture of *trans*-**10a** and *trans*-**11a** (10 mg), Ph₂Se₂(0.2 equiv.), AIBN (0.4 equiv.), solvent(1 mL), 300 W sunlamp, room temperature. ^b Isolated yields of all diastereomers. ^c Determined by HPLC analysis.

Table S2. Additive effective studies of the Feldman reaction of *trans*-**10a** and *trans*-**11a**

Entry ^a	Additive (1 eq.)	Yield % ^b	<i>trans</i> - 12a / <i>trans</i> - 13a / <i>cis</i> - 14a ^c
1	None	quant	26/66/9
2	LiCl	85	24/66/10
3	Mg(ClO ₄) ₂	96	22/63/15
4	Ti(ⁱ PrO) ₄	96	25/65/10
5	Yb(OTf) ₃	86	21/67/12
6	La(OTf) ₃	82	23/66/11
7	Sc(OTf) ₃	92	16/76/8

^a Reaction conditions: 1:1 mixture of *trans*-**10a** and *trans*-**11a** (10 mg), Ph₂Se₂(0.2 equiv.), AIBN (0.4 equiv.), ether (1 mL), Additive(1 equiv.), 300 W sunlamp, room temperature. ^b Isolated yields of all diastereomers. ^c Determined by HPLC analysis.

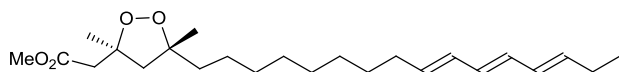
Table S3. Temperature effective studies of the Feldman reaction of *trans*-**10a** and *trans*-**11a**

Entry ^a	Temp.	Yield % ^b	<i>trans</i> - 12a / <i>trans</i> - 13a / <i>cis</i> - 14a ^c
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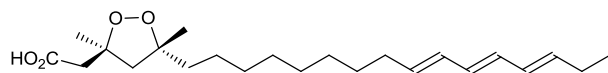
1	r.t.	quant	26/66/9
2	0 °C	88	26/62/13
3	reflux	85	26/66/9

^a Reaction conditions: 1:1 mixture of *trans*-**10a** and *trans*-**11a** (10 mg), Ph₂Se₂(0.2 equiv.), AIBN (0.4 equiv.), ether (1 mL), Sc(OTf)₃(1 equiv.), 300 W sunlamp. ^b Isolated yields of all diastereomers. ^c Determined by HPLC analysis.

Characterization Tables for Compounds 1 and 2



Epiplakinic acid F methyl ester (+)-2



Epiplakinic acid F (1)

Table S4. The data reported for natural Epiplakinic acid F methyl ester and the data for our Synthetic compound 2 (for comparison)

Epiplakinic acid F methyl ester						
Source	Natural Product ¹⁶			synthetic compound		
HRMS	HR-EI-MS <i>m/z</i> 392.2918 (calcd for C ₂₄ H ₄₀ O ₄ [M] ⁺ , 392.2927).			HR-ESI-MS <i>m/z</i> 392.2911 (calcd for C ₂₄ H ₄₀ O ₄ [M] ⁺ , 392.2927).		
[α] _D ²⁵	[α] _D ²⁰ = 34.4 (c, 1.0, CHCl ₃)			[α] _D ²⁵ = 32.39 (c, 0.5, CHCl ₃)		
NMR(CDCl ₃)	¹ H (500 MHz)	¹³ C (125 MHz)		¹ H (500 MHz)	¹³ C (125 MHz)	
equipment	Bruker DRX-500			Bruker DRX-500		
H-1		C-1	171.1		C-1	171.3
H-2	2.76, α(d, 14.5) 2.65, β(d, 14.5)	C-2	44.0	2.76, α(d, 14.5) 2.64, β(d, 14.5)	C-2	44.2

H-3		C-3	83.9		C-3	84.2
H-4	2.22 α (d, 12.5) 2.46 β (d, 12.5)	C-4	55.4	2.46 α (d, 12.5) 2.22 β (d, 12.5)	C-4	55.6
H-5		C-5	86.5		C-5	86.7
H-6	1.69 α (t, 12.5) 1.53 α (t, 12.5)	C-6	39.6	1.69 α (br, m) 1.53 β (br, m)	C-6	39.9
H-7	1.28 α (br, s) 1.37 β (br, m)	C-7	24.5	1.33 α (br, s) 1.37 β (br, m)	C-7	24.4
H-8	1.28 (br, s)	C-8	29.5	1.33 (br, s)	C-8	29.7
H-9	1.28 (br, s)	C-9	29.4	1.33 (br, s)	C-9	29.6
H-10	1.28 (br, s)	C-10	29.3	1.33 (br, s)	C-10	29.6
H-11	1.28 (br, s)	C-11	30.0	1.33 (br, s)	C-11	30.2
H-12	1.37 (br, m)	C-12	29.1	1.37 (br, m)	C-12	29.4
H-13	2.05 (m)	C-13	32.8	2.07 (m)	C-13	33.0
H-14	5.65 (dt, 7.0,14.2)	C-14	134.5	5.68 (dt, 7.5,14.0)	C-14	134.7

H-15	6.03 (br, m)	C-15	130.4	6.02 (m)	C-15	130.7
H-16	6.10 (br, m)	C-16	130.9	6.10 (m)	C-16	131.1
H-17	6.08 (br, m)	C-17	130.8	6.08 (m)	C-17	130.0
H-18	6.06 (br, m)	C-18	129.5	6.05 (m)	C-18	129.7
H-19	5.71 (dt, 7.5,14.5)	C-19	135.9	5.70 (dt, 7.5,14.0)	C-19	136.1
H-20	2.08 (m)	C-20	25.8	2.09 (m)	C-20	26.0
H-21	0.97 (t, 7.4)	C-21	13.6	1.00 (t, 7.5)	C-21	13.8
H-22	1.43 (s)	C-22	24.1	1.44 (s)	C-22	24.8
H-23	1.28 (s)	C-23	23.3	1.28 (s)	C-23	23.5
-OCH ₃	3.69 (s)	C-24	51.7	3.69 (s)	C-24	51.9

Table S5. The data reported for natural Epiplakinic acid F and the data for our Synthetic compound 1 (for comparison)

Epiplakinic acid F						
Source	Natural Product ¹⁷			synthetic compound		
HRMS	HR-FAB-MS <i>m/z</i> 379.2816 (calcd for C ₂₃ H ₃₉ O ₄ [M+H] ⁺ , 379.2850).			HR-ESI-MS <i>m/z</i> 396.3102 (calcd for C ₂₃ H ₄₂ O ₄ N [M+NH ₄] ⁺ , 396.3108).		
[α] _D ²⁵				[a] _D ²⁵ = 31.21 (c, 0.5, CHCl ₃)		
NMR(CDCl ₃)	¹ H (500 MHz)	¹³ C (125 MHz)		¹ H (500 MHz)	¹³ C (125 MHz)	
equipment	Bruker AMX-500			Bruker DRX-500		
H-1		C-1	171.80		C-1	173.3
H-2	2.75 (s)	C-2	43.94	2.79 (dd, 14.5, 23.5)	C-2	43.1
H-3		C-3	83.68		C-3	83.4
H-4	2.40 α(d, 12.5) 2.25 β(d, 12.5)	C-4	55.51	2.46 α(d, 12.5) 2.28 β(d, 12.5)	C-4	55.2
H-5		C-5	86.86		C-5	86.2

H-6	1.65 (m) 1.28 (m)	C-6	38.77	1.72 (m) 1.30 (m)	C-6	39.3
H-7	1.16-1.53 (m)	C-7	24.86	1.39 α (br, s) 1.30 β (br, m)	C-7	24.0
H-8	1.16-1.53 (m)	C-8	29.70	1.30 (br, s)	C-8	29.0
H-9	1.16-1.53 (m)	C-9	29.42	1.30 (br, s)	C-9	28.8
H-10	1.16-1.53 (m)	C-10	29.12	1.30 (br, s)	C-10	28.7
H-11	1.16-1.53 (m)	C-11	30.02	1.30 (br, s)	C-11	29.5
H-12	1.39 (m)	C-12	29.70	1.39 (br, m)	C-12	29.0
H-13	2.05 (q, 7.2)	C-13	32.78	2.13 (m)	C-13	32.3
H-14	5.63 (dt, 7.2,14.1)	C-14	134.41	5.68 (dt, 7.5,14.0)	C-14	133.9
H-15	6.02 (m)	C-15	130.48	6.04 (m)	C-15	130.0
H-16	6.06 (m)	C-16	130.83	6.12 (m)	C-16	130.3
H-17	6.06 (m)	C-17	130.87	6.10 (m)	C-17	130.4
H-18	6.02 (m)	C-18	129.51	6.05 (m)	C-18	129.0

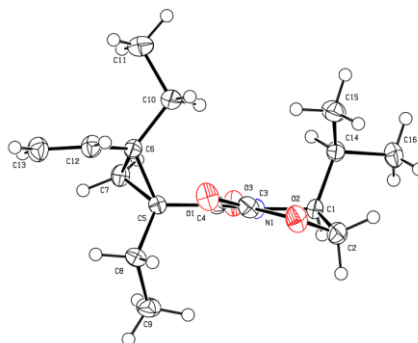
H-19	5.68 (dt, 7.2,14.4)	C-19	135.90	5.72 (dt, 7.5,14.0)	C-19	135.4
H-20	2.08 (q, 7.2)	C-20	25.81	2.12 (m)	C-20	25.3
H-21	0.97 (t, 7.5)	C-21	14.11	1.03 (t, 7.5)	C-21	13.1
H-22	1.45 (s)	C-22	24.75	1.50 (s)	C-22	23.2
H-23	1.28 (s)	C-23	23.69	1.28 (s)	C-23	22.6

Basic Crystal Data for Compounds *trans*-10a, *trans*-12a and *trans*-20a

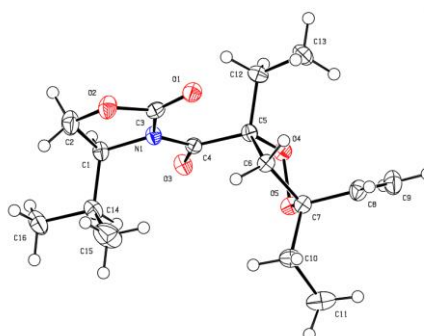
CCDC 978791 (*trans*-10a), CCDC 978789 (*trans*-12a), CCDC 978792 (*trans*-20a) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Deposition Number(s): 978791, 978789, 978792.

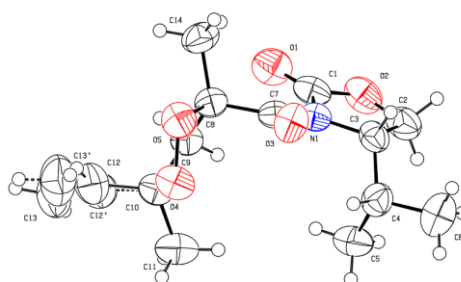
CIF files for these structures are attached to this message.



ORTEP diagram of compound *trans*-10a. Thermal ellipsoids were plotted at 30 % probability level. Hydrogen atoms are omitted for clarity.

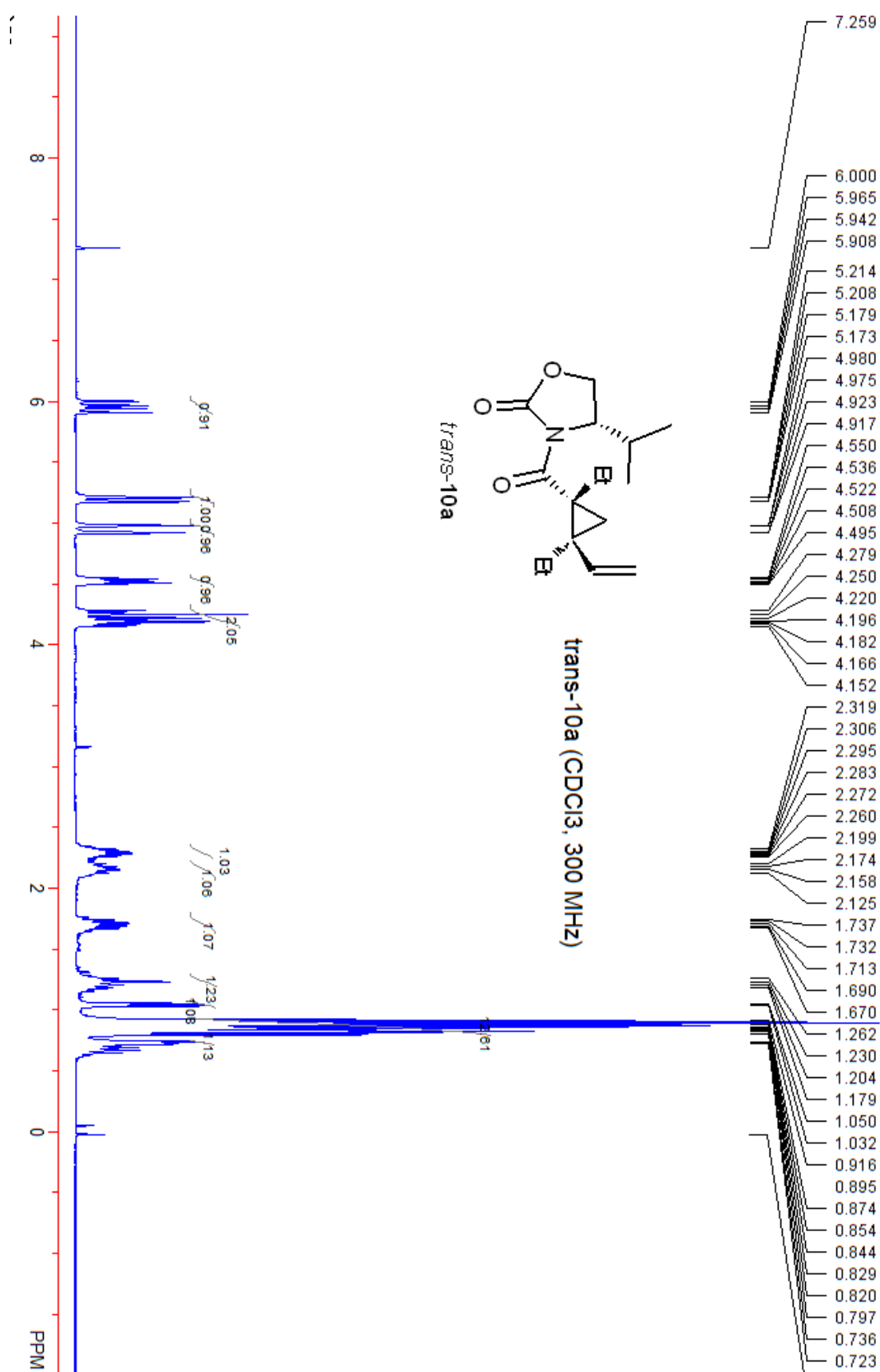


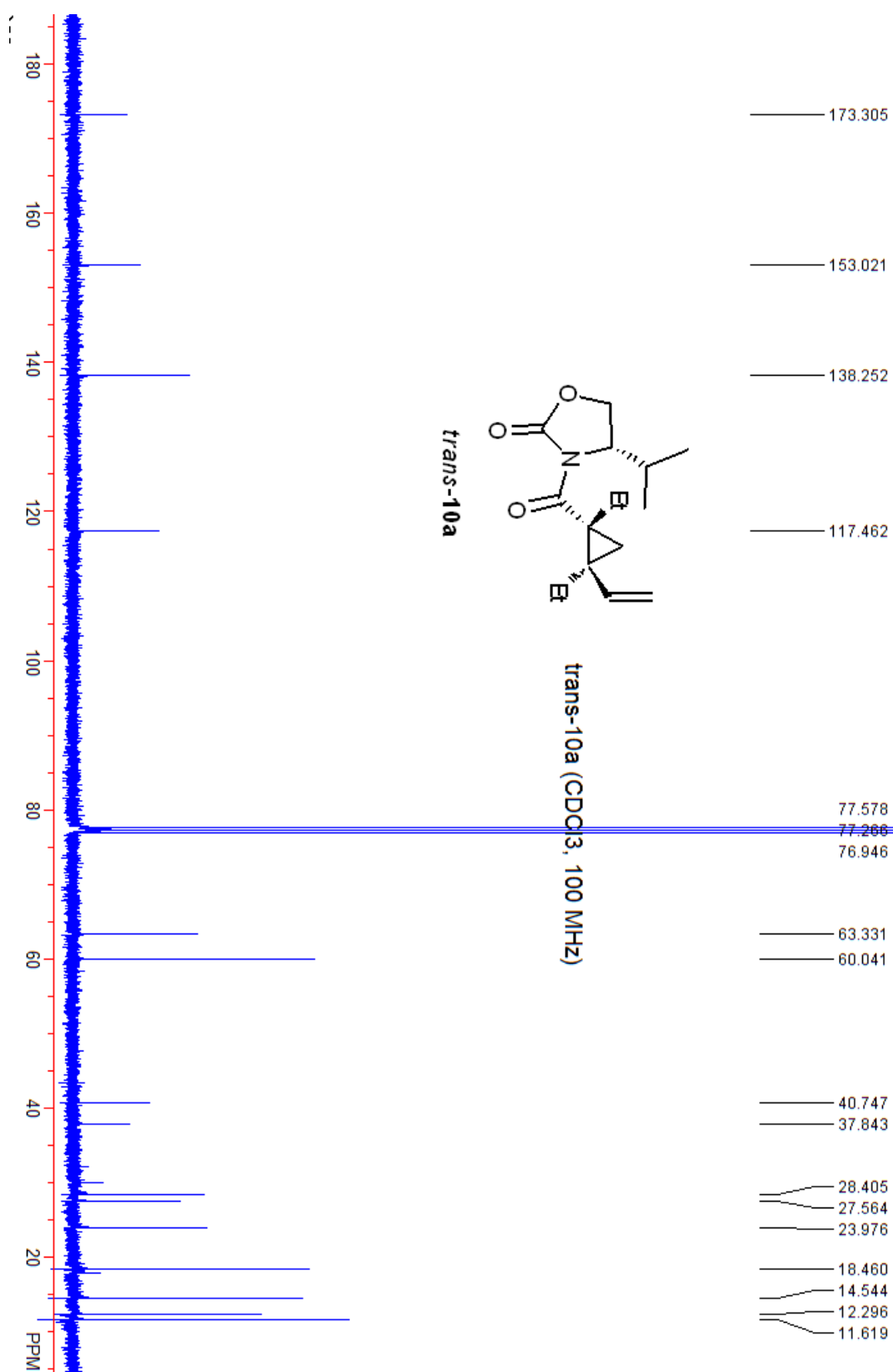
ORTEP diagram of compound *trans*-12a. Thermal ellipsoids were plotted at 30 % probability level. Hydrogen atoms are omitted for clarity.

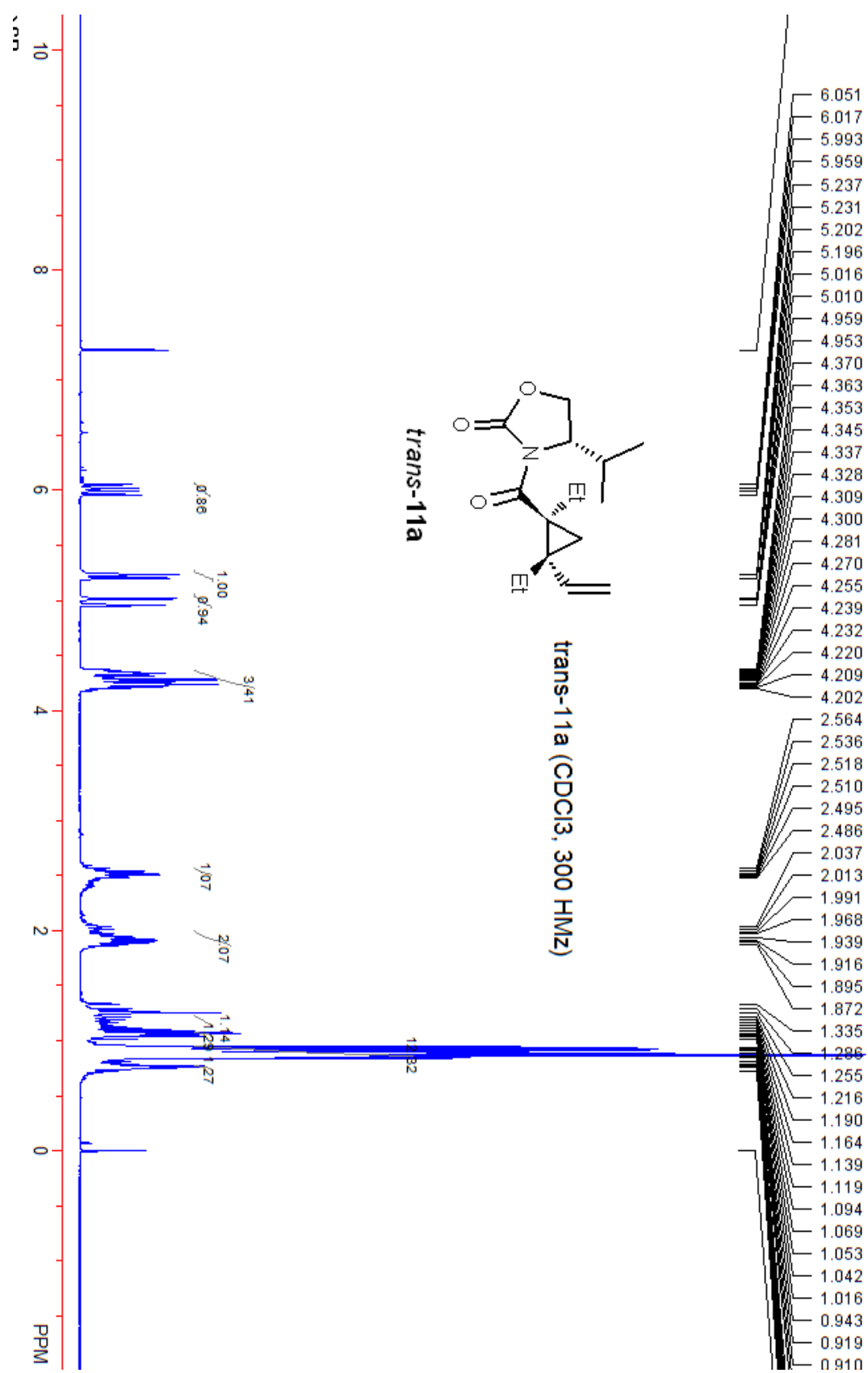


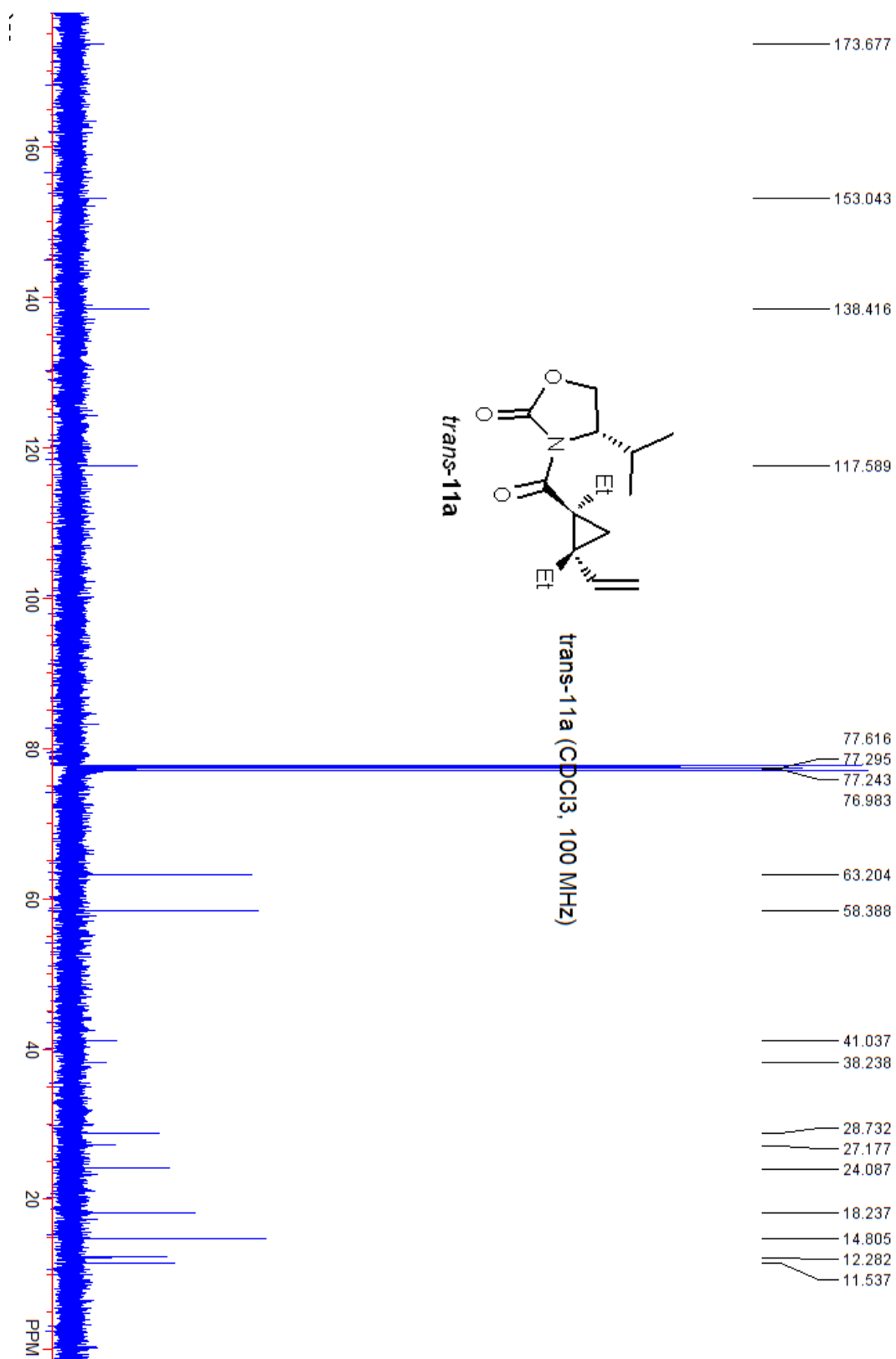
ORTEP diagram of compound *trans*-20a. Thermal ellipsoids were plotted at 30 % probability level. Hydrogen atoms are omitted for clarity.

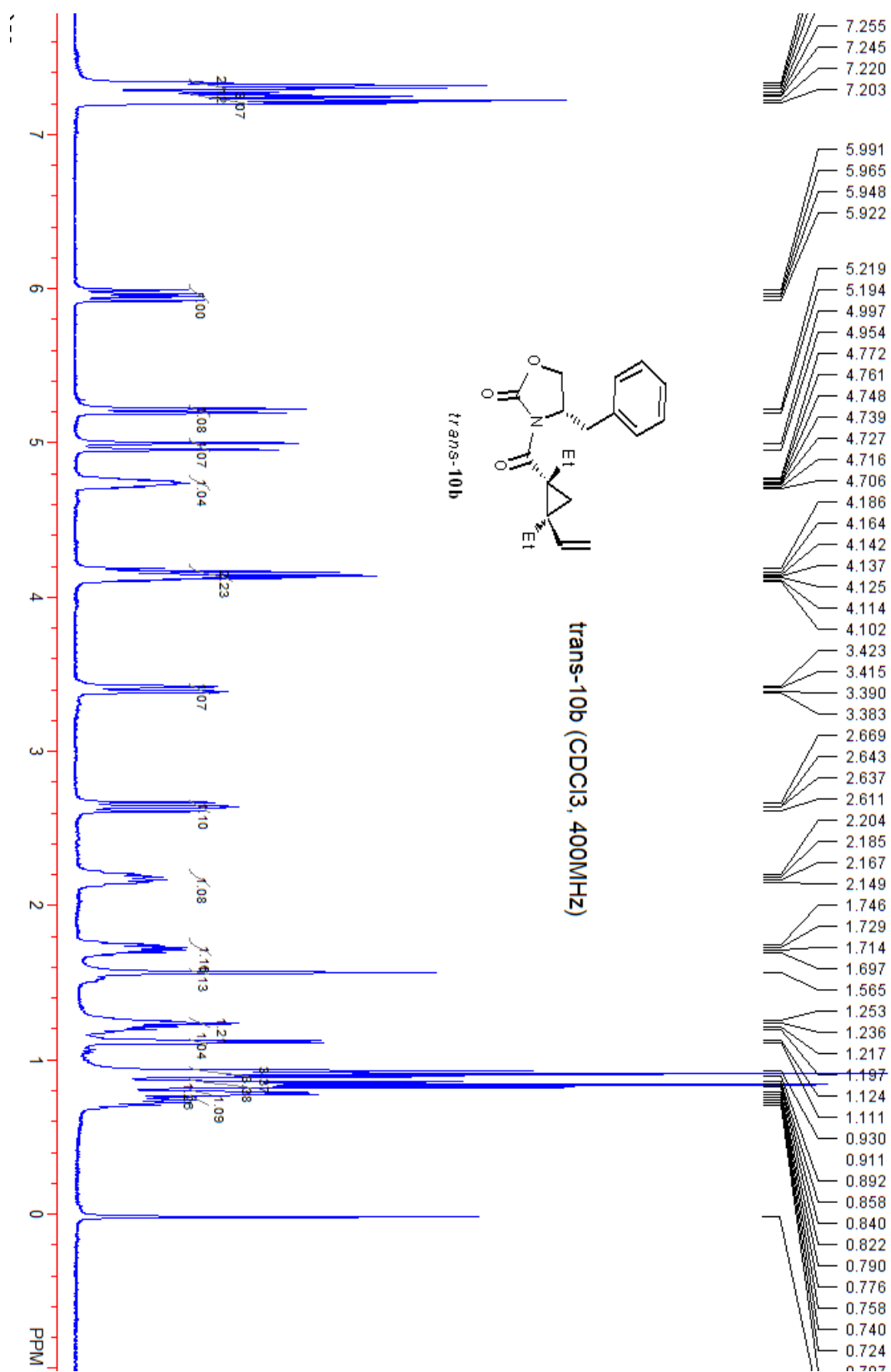
^1H and ^{13}C NMR Spectra

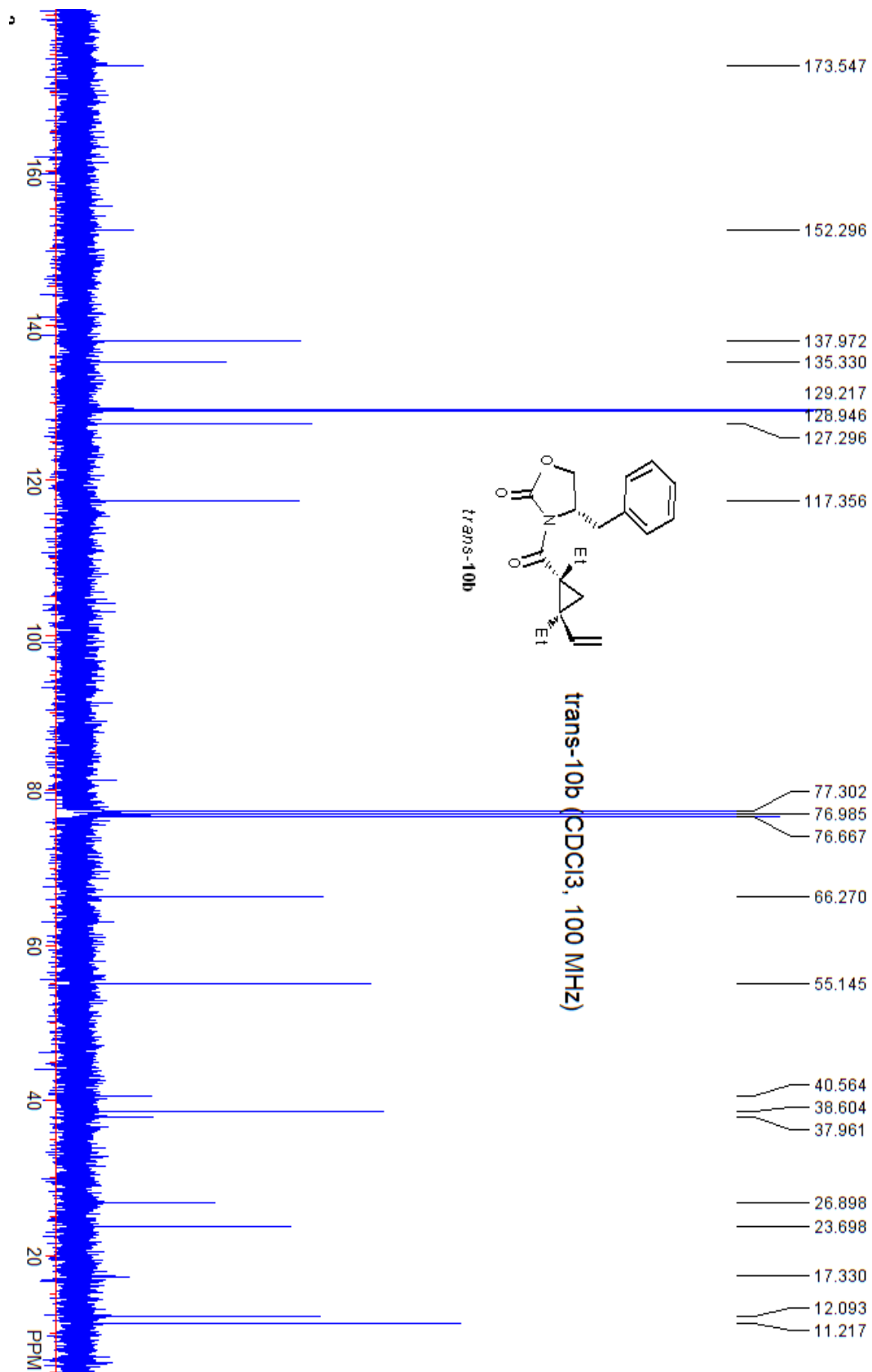


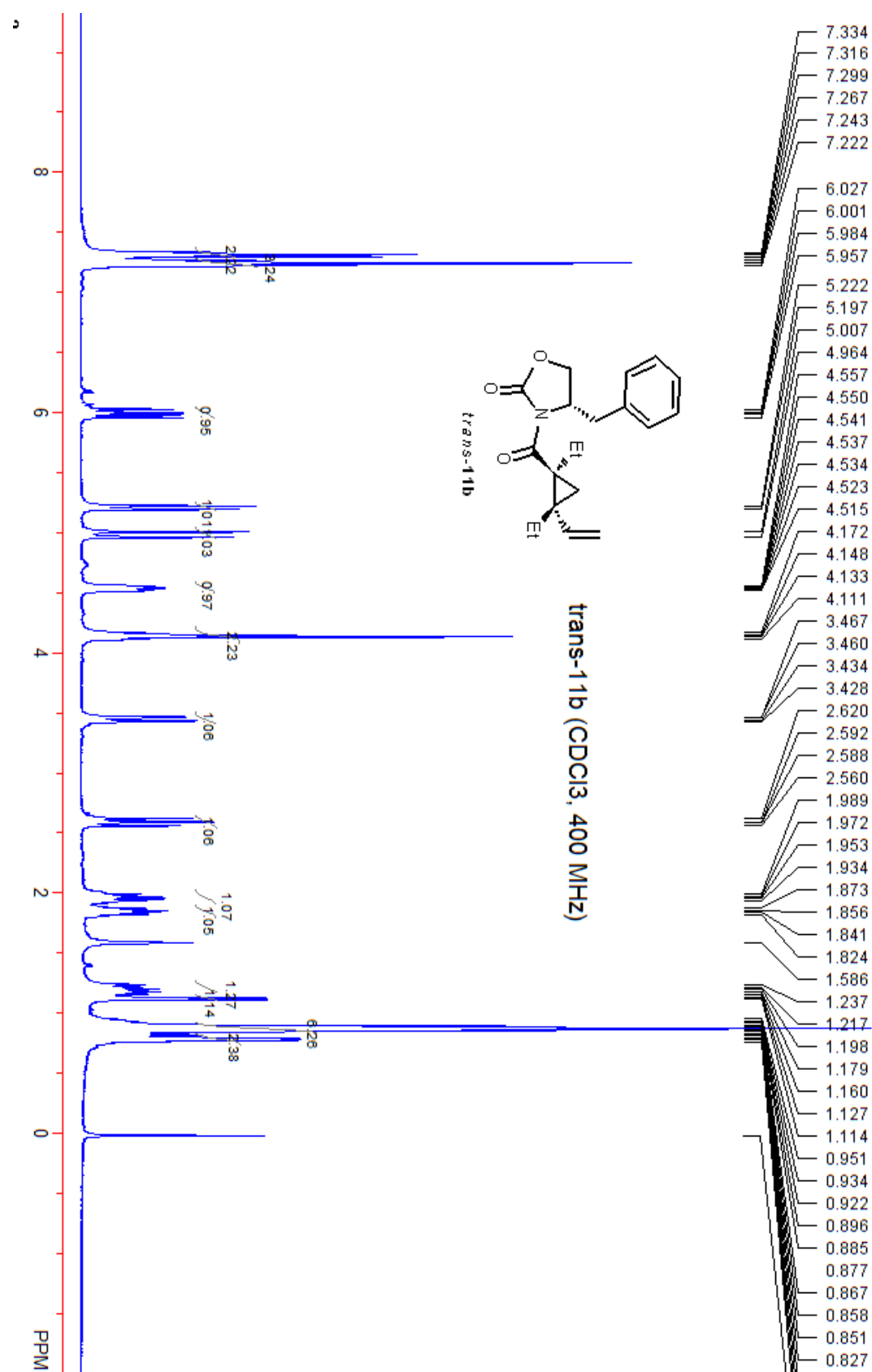


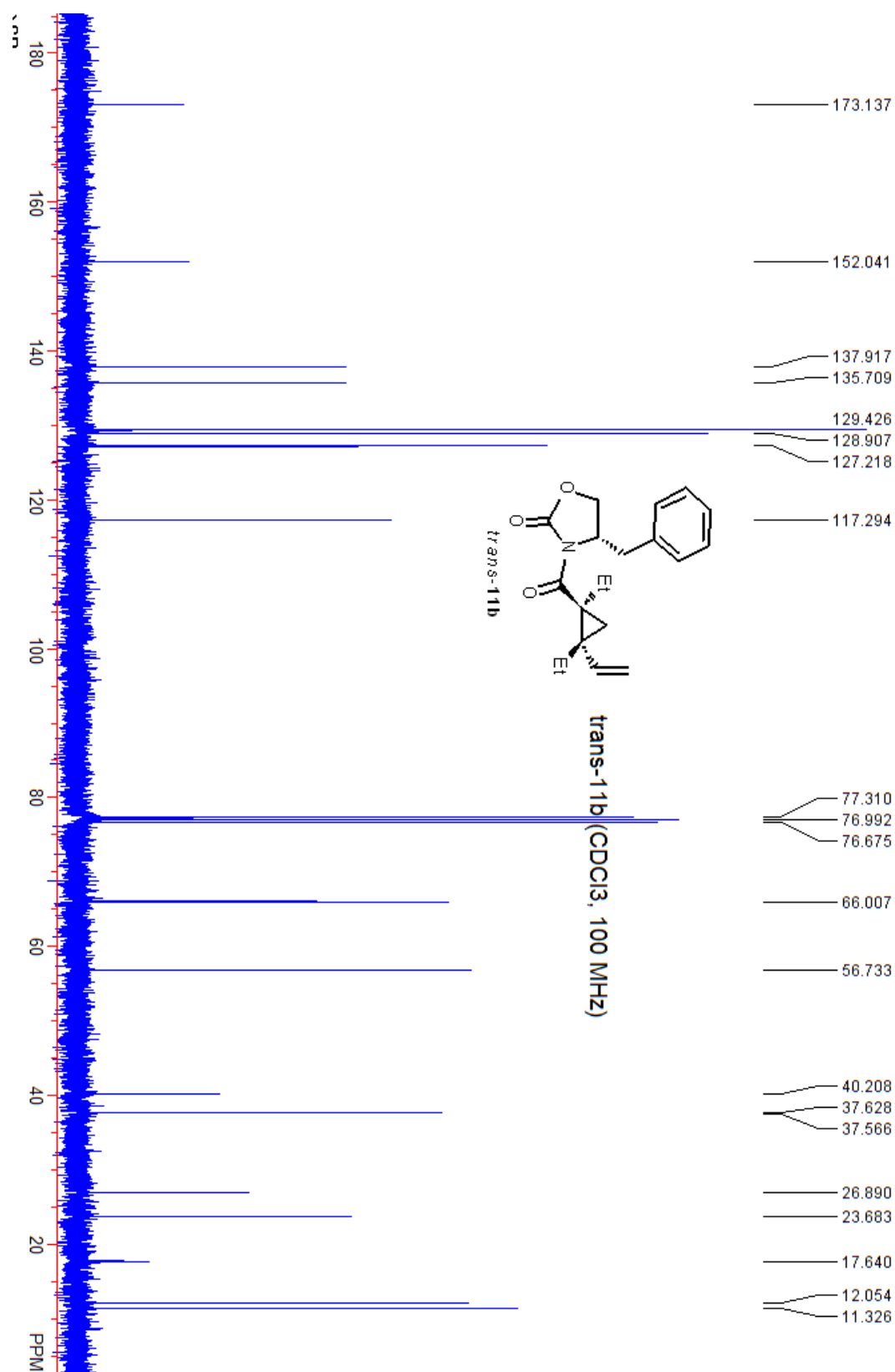


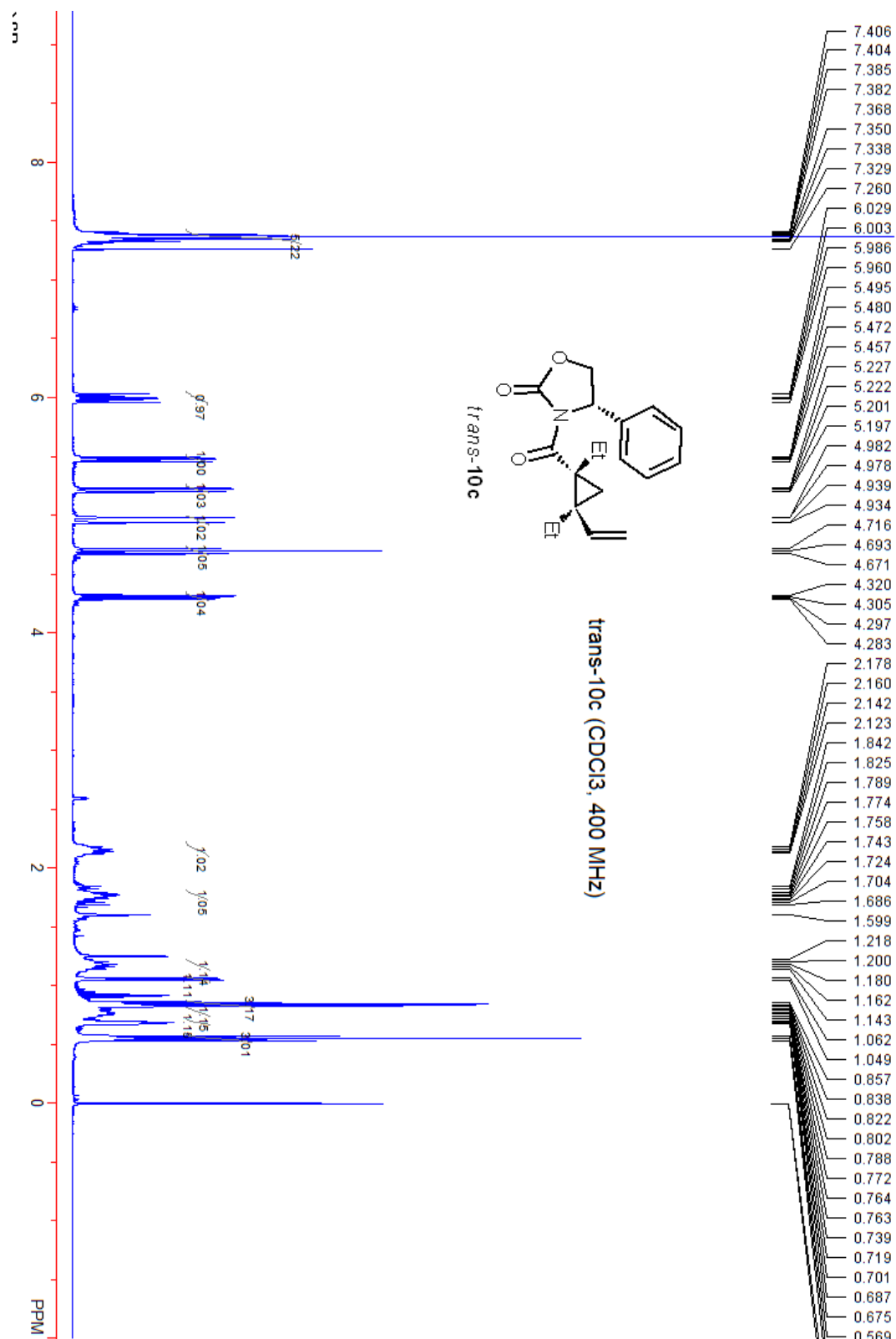


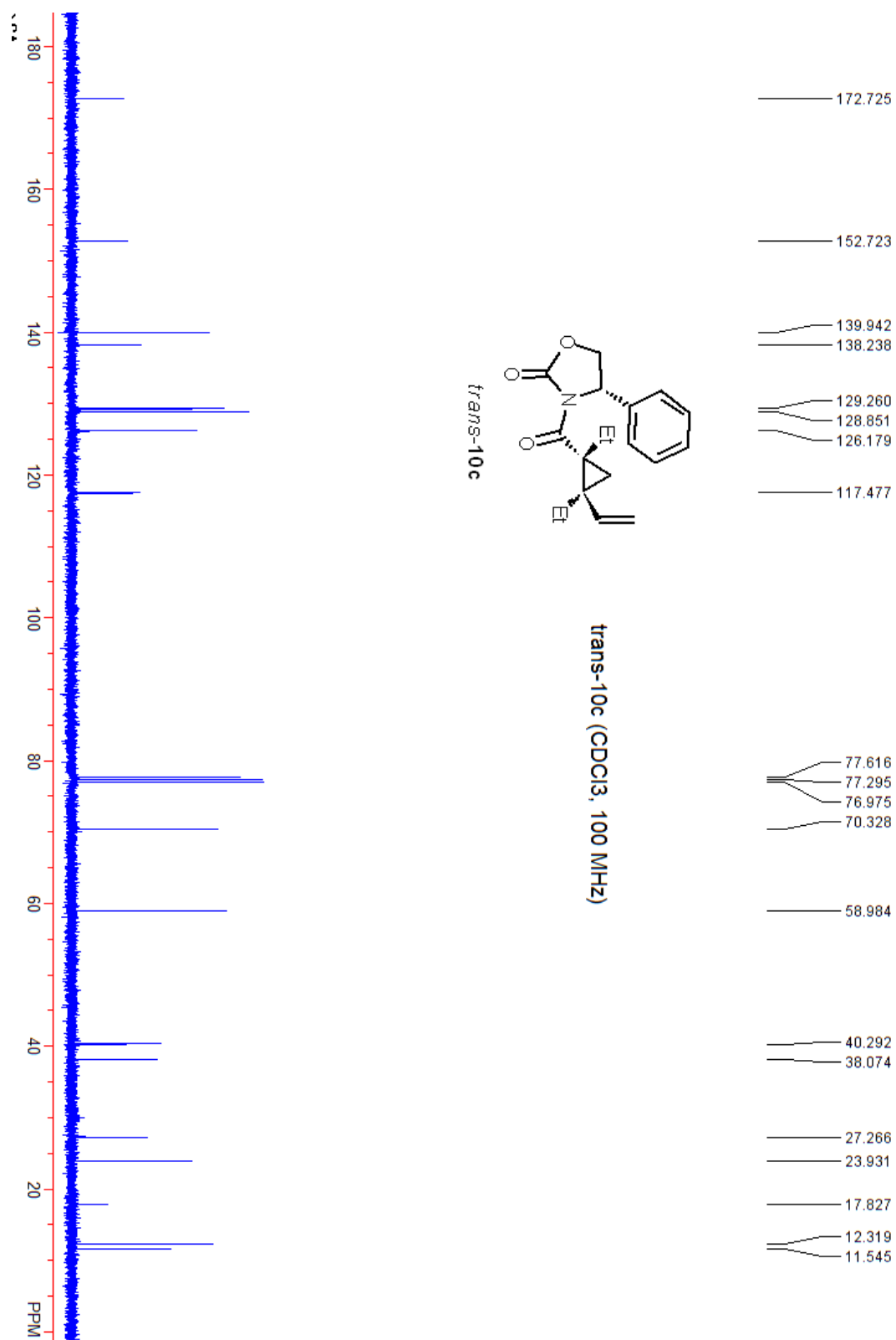


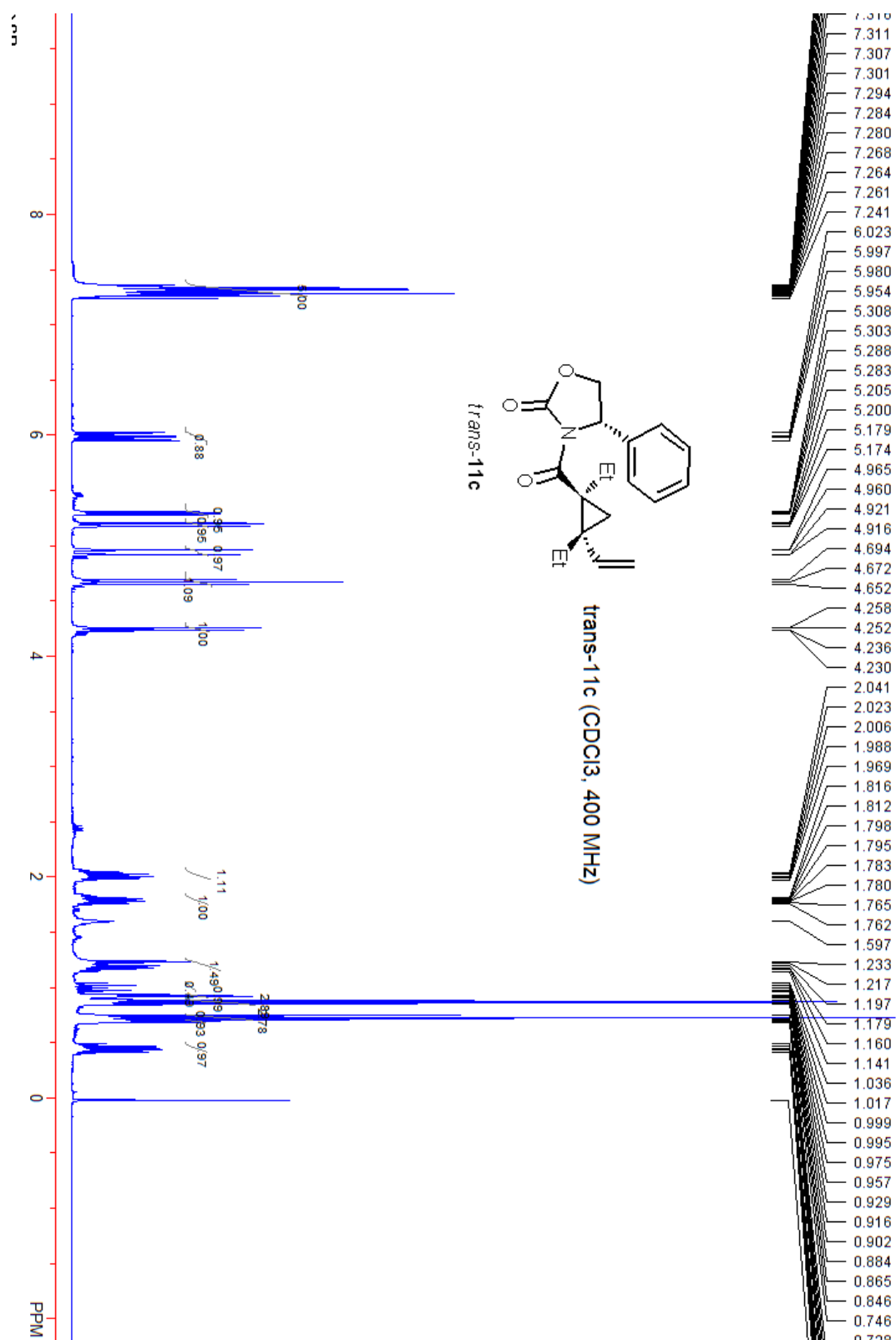


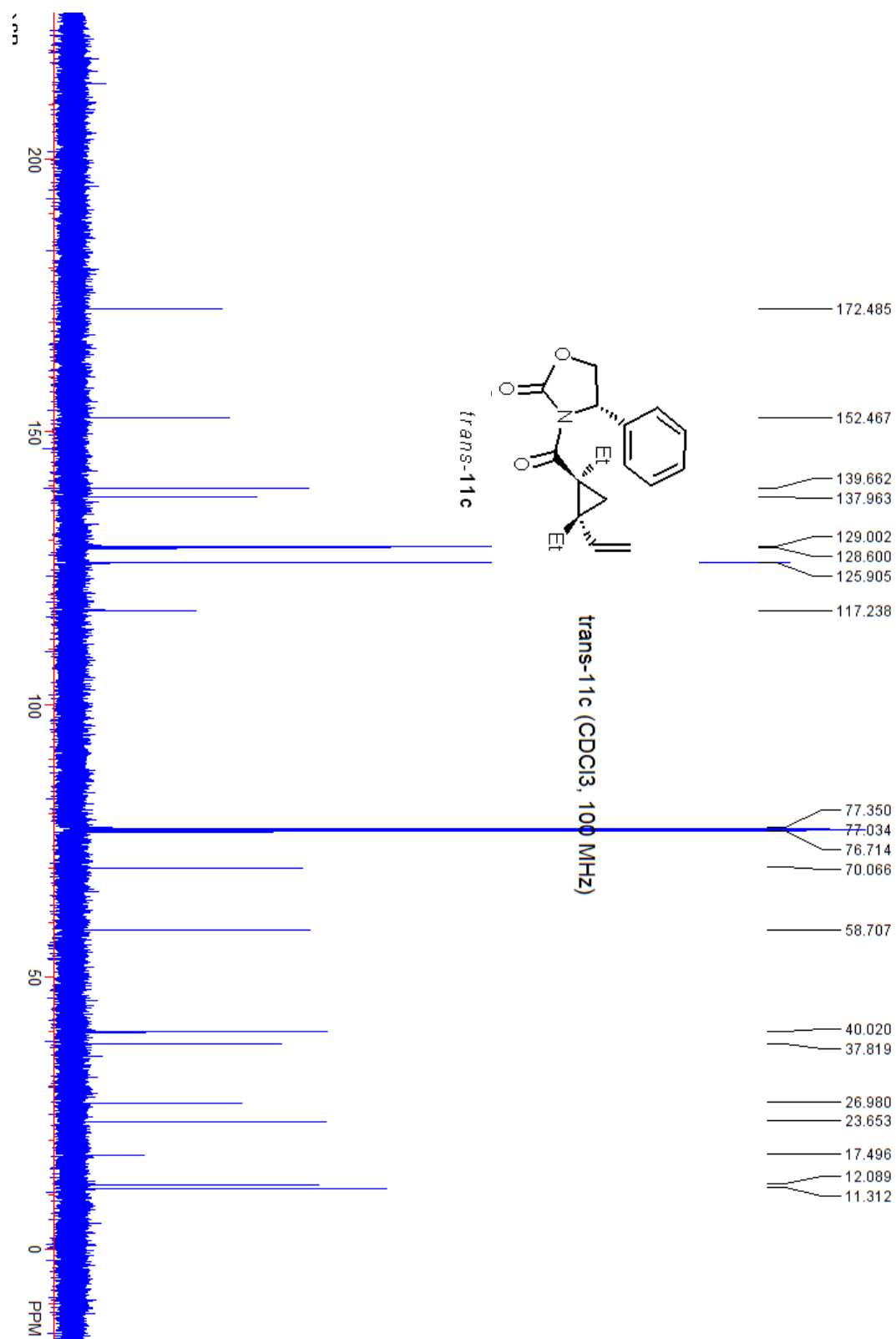


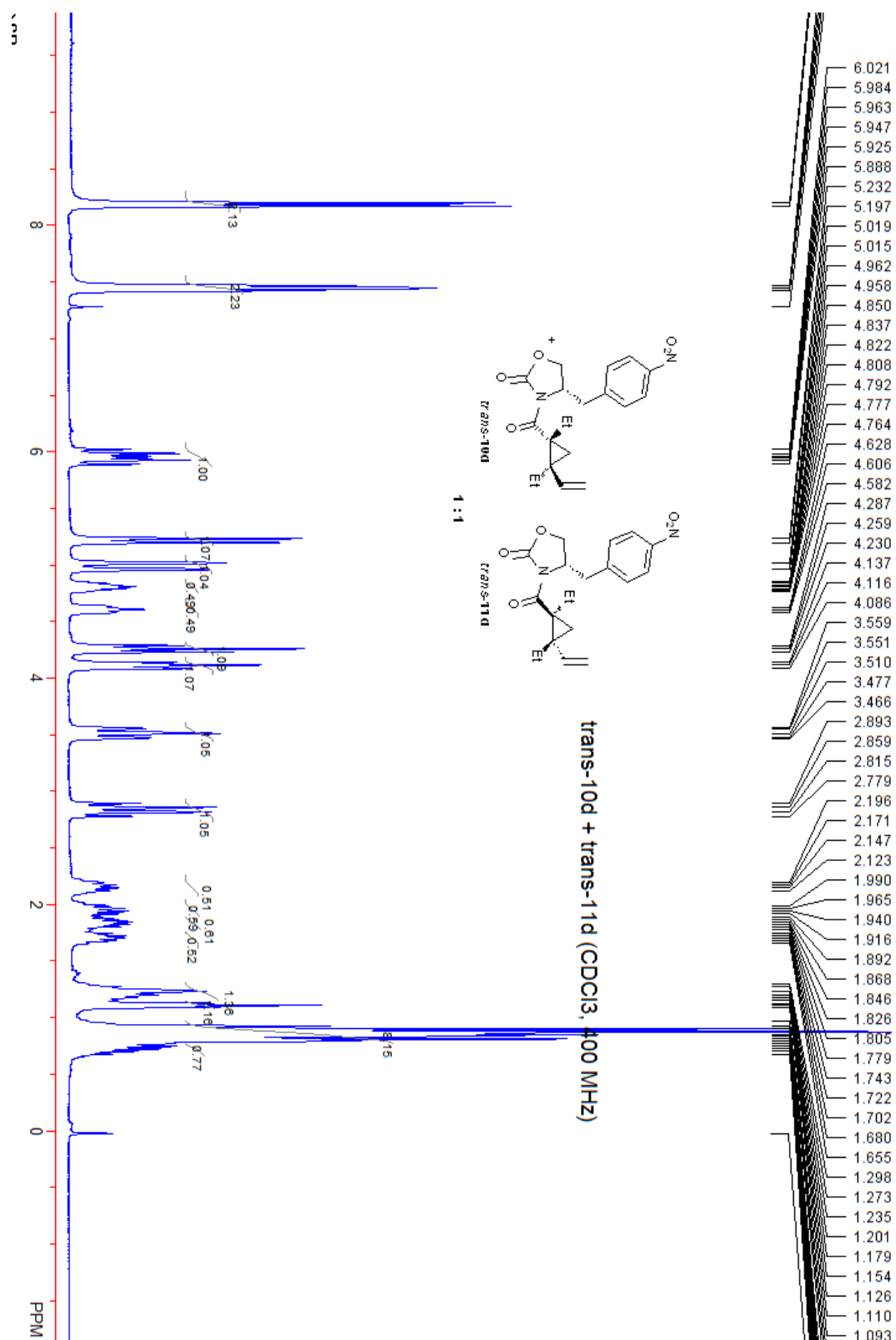


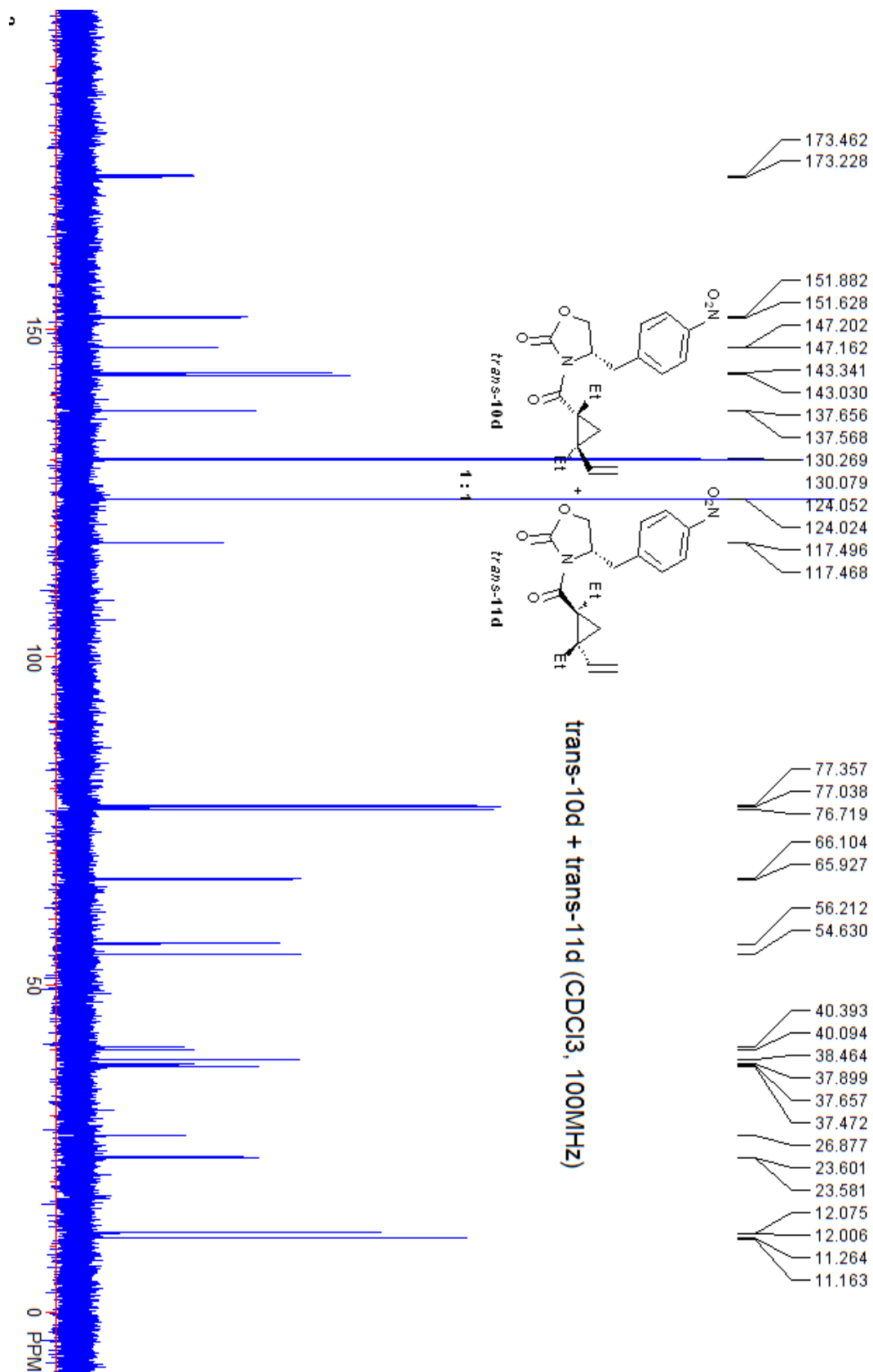


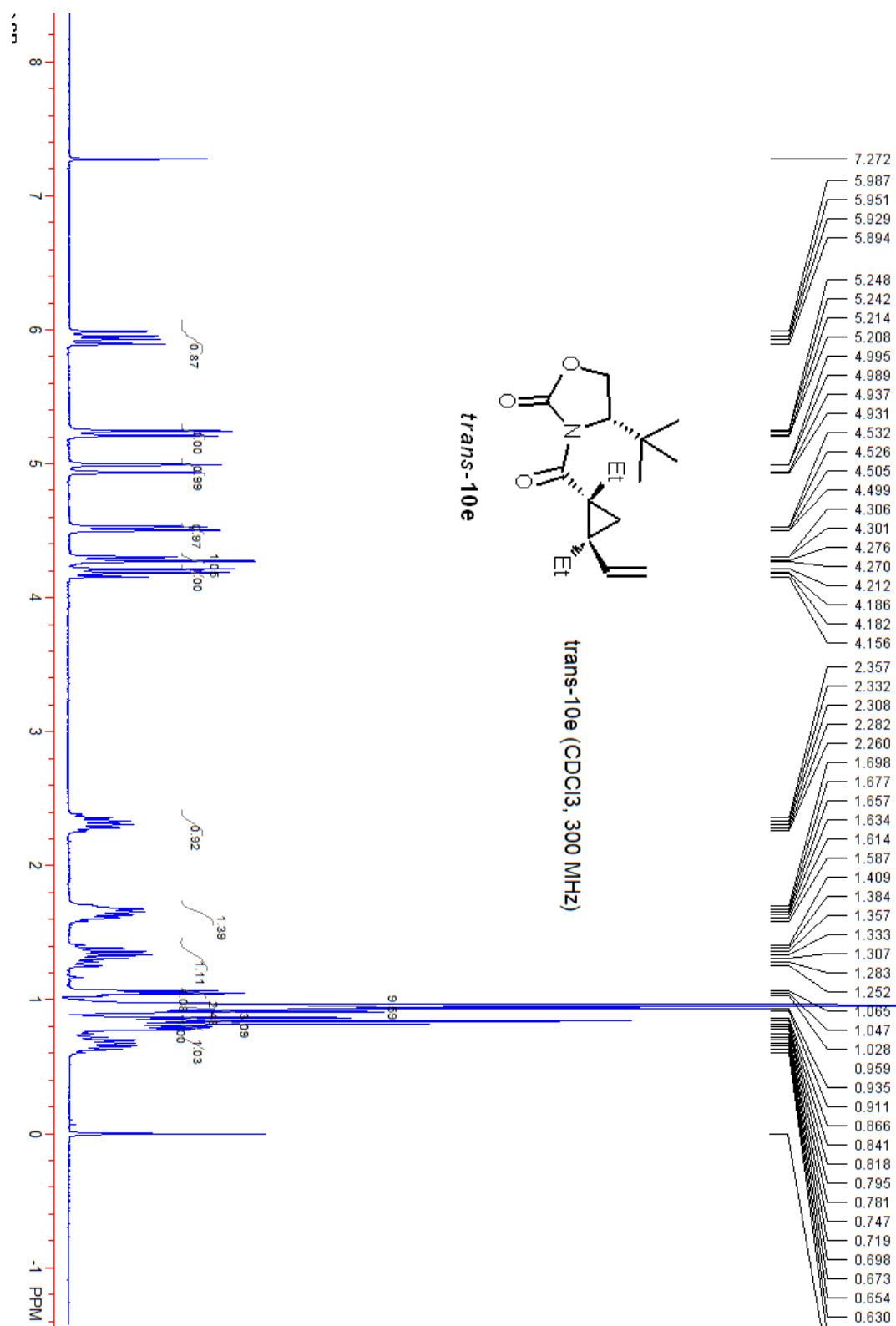


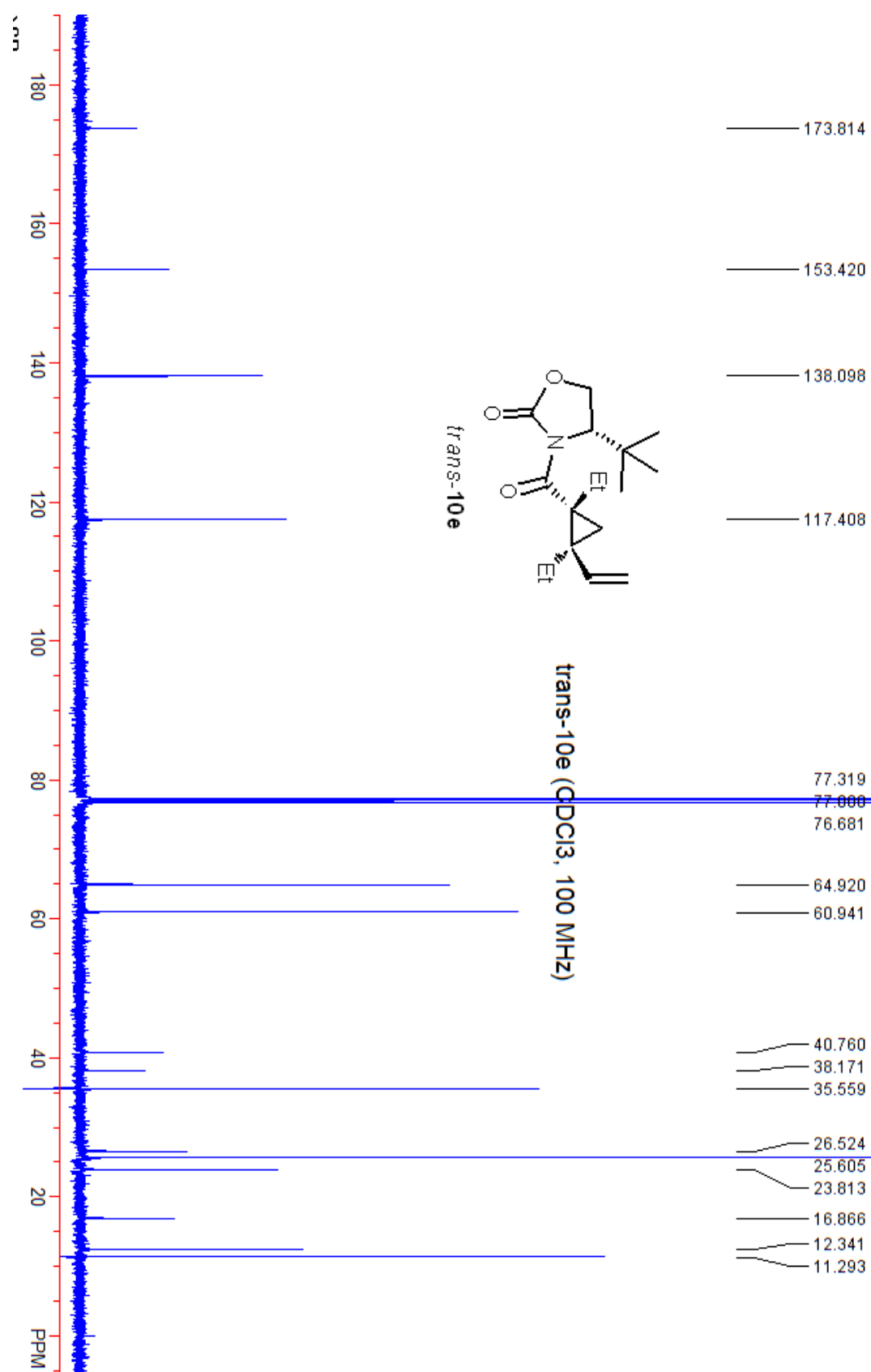


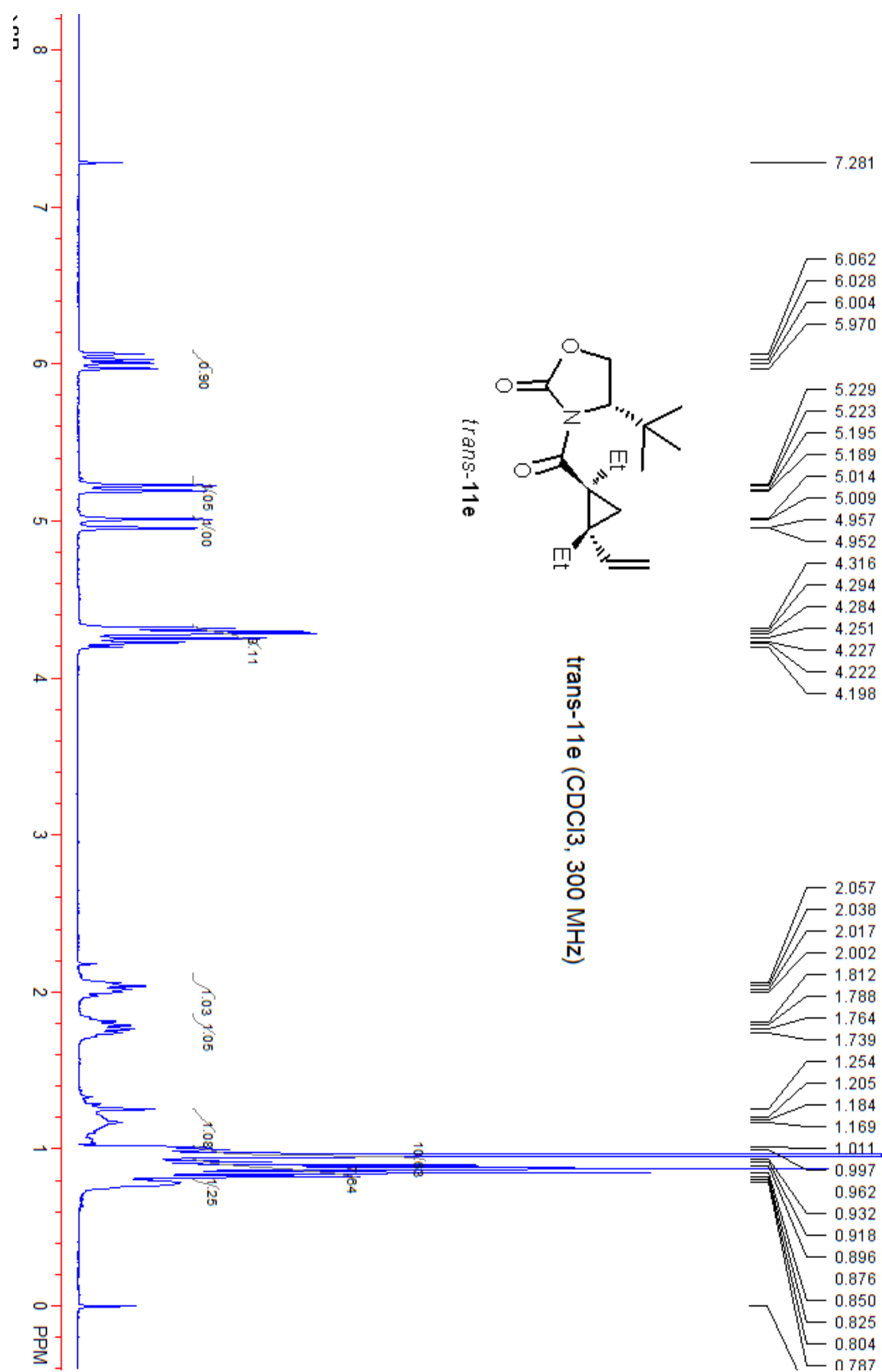


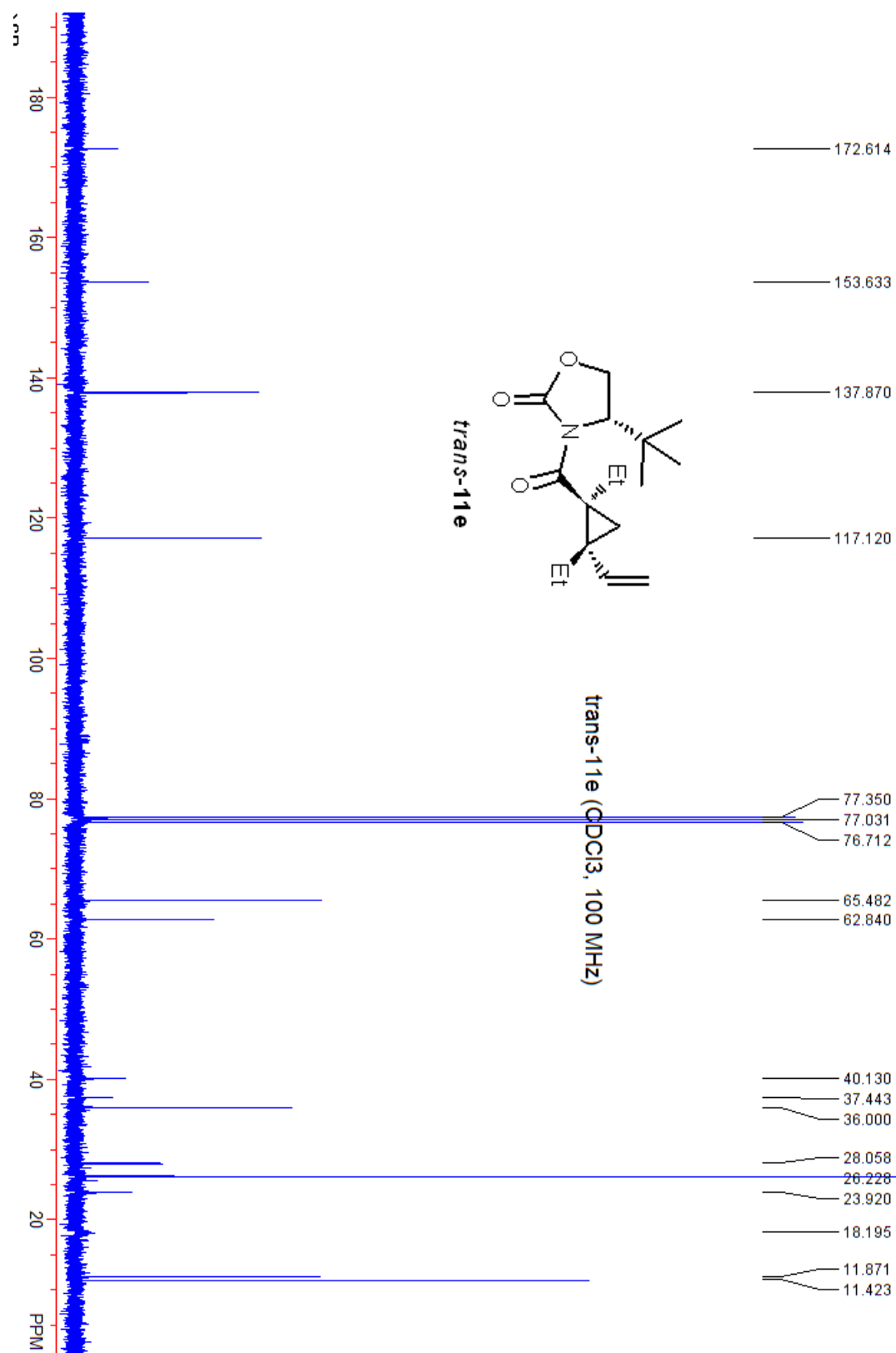


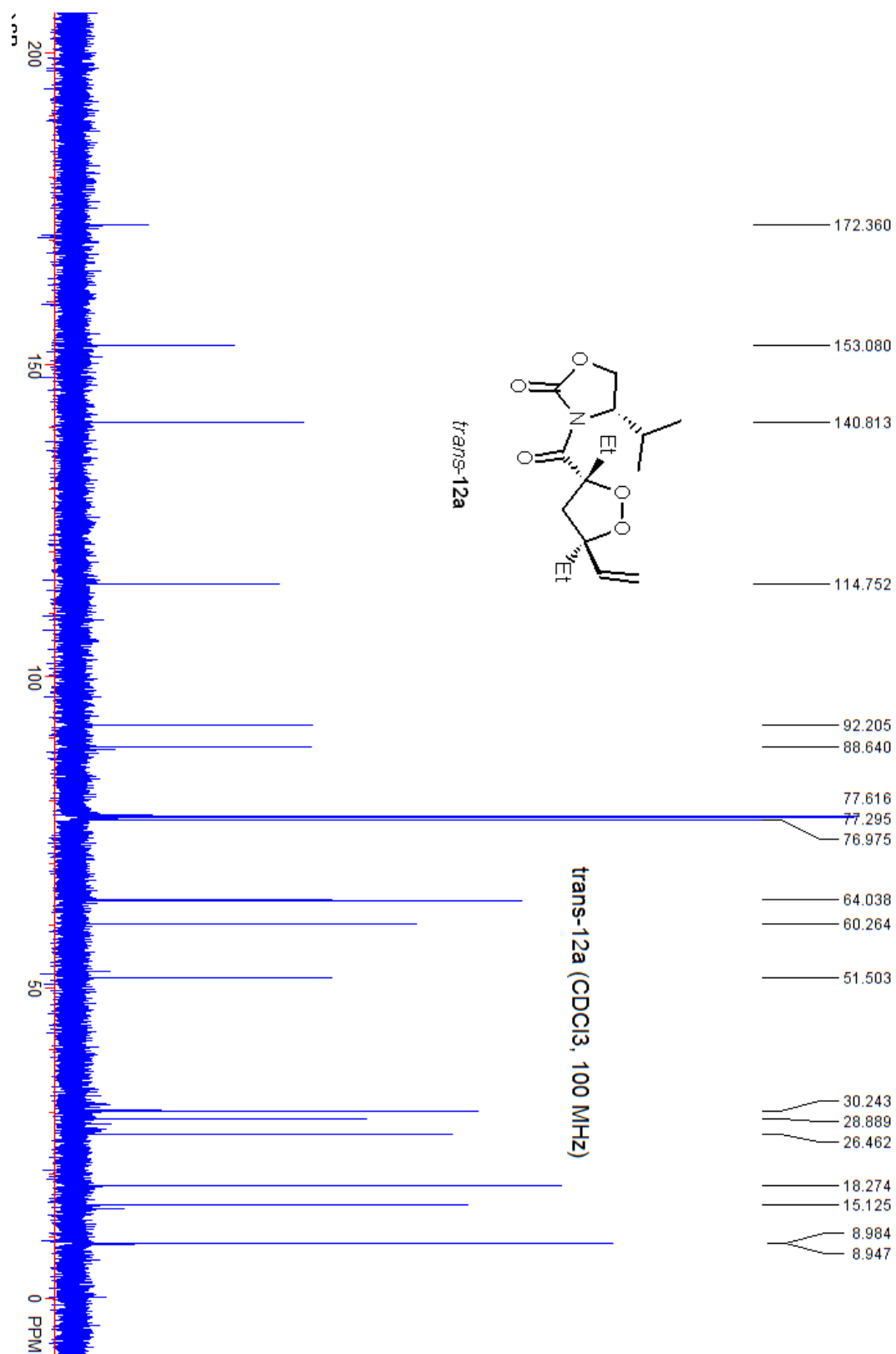


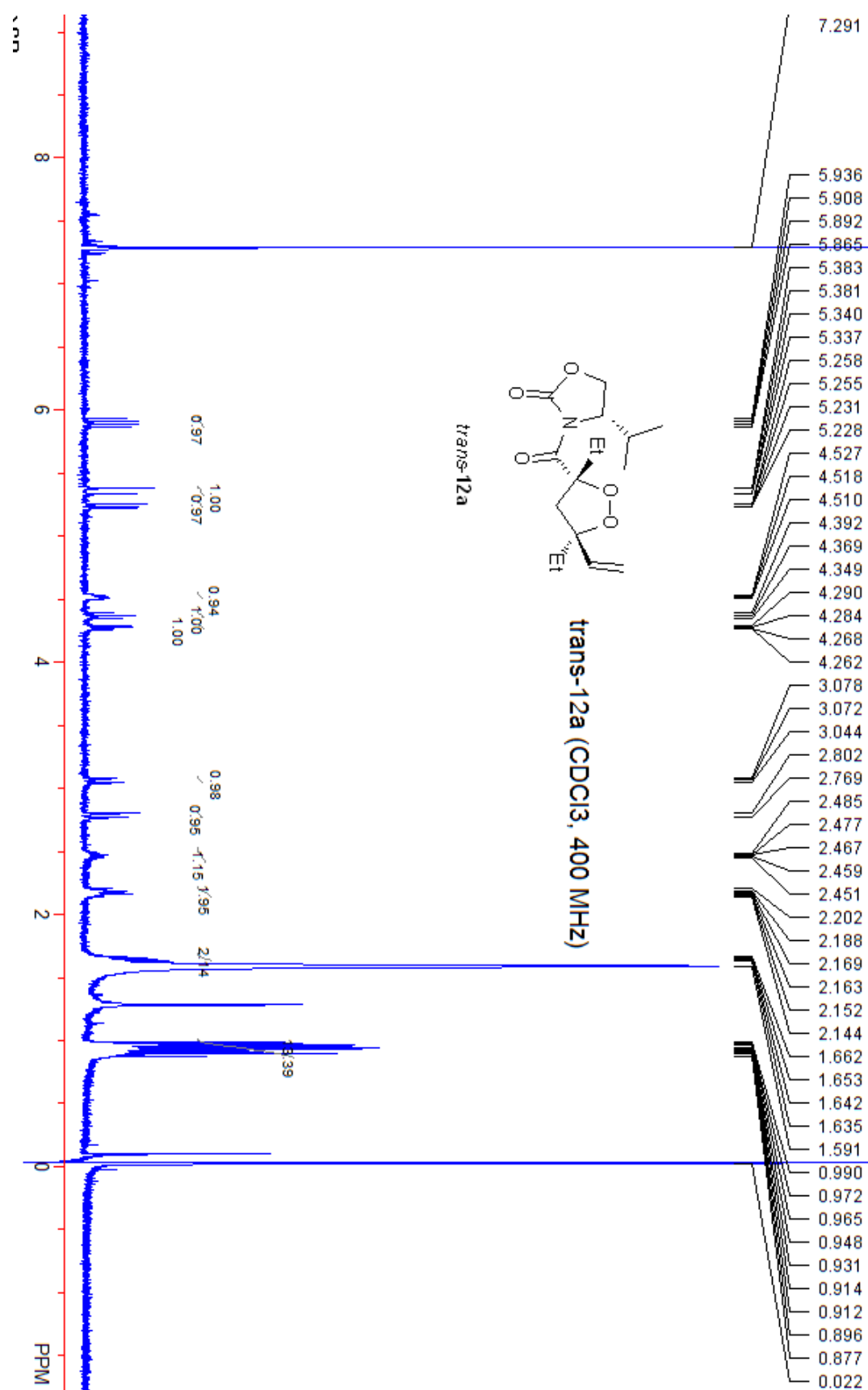


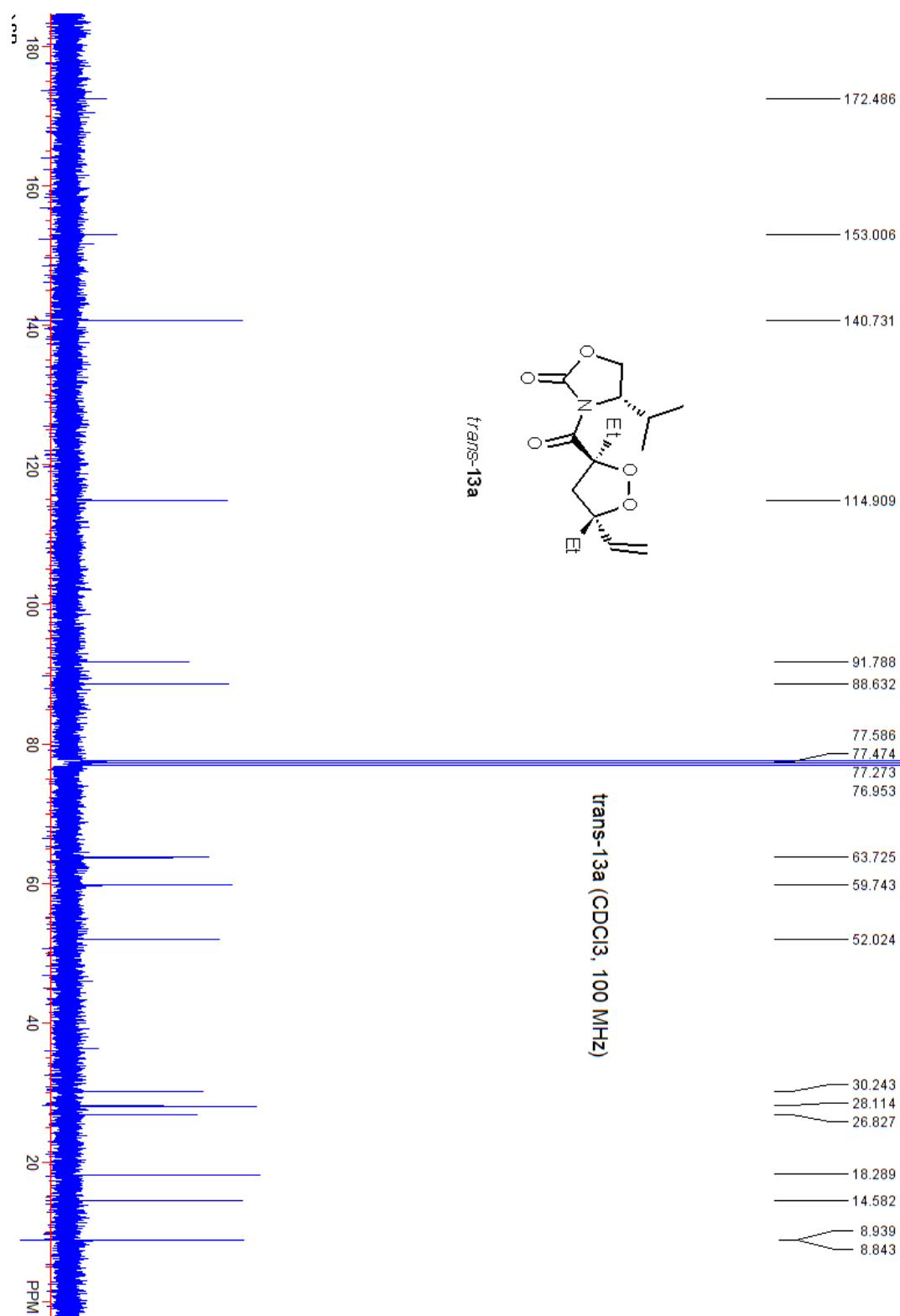


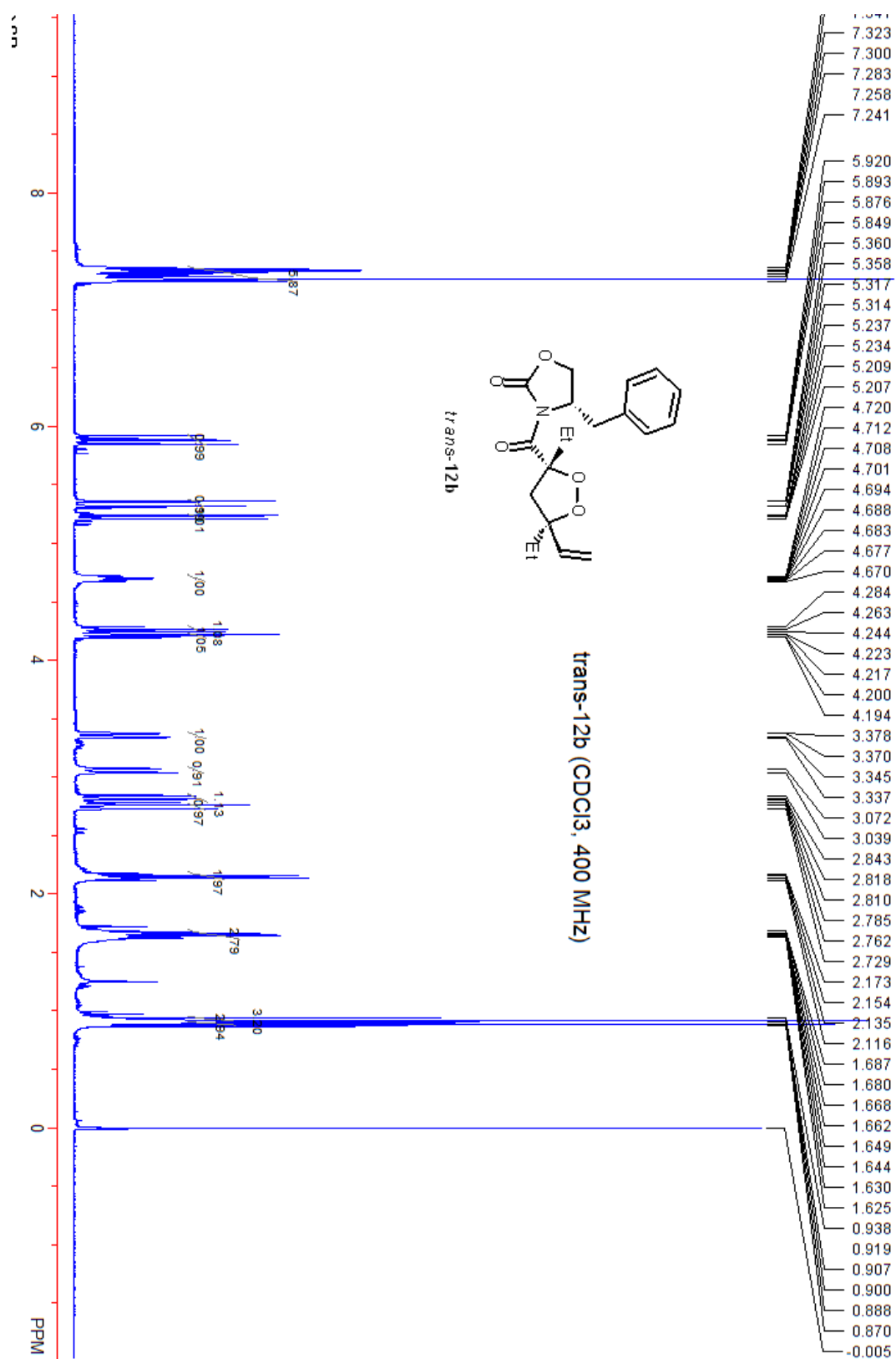


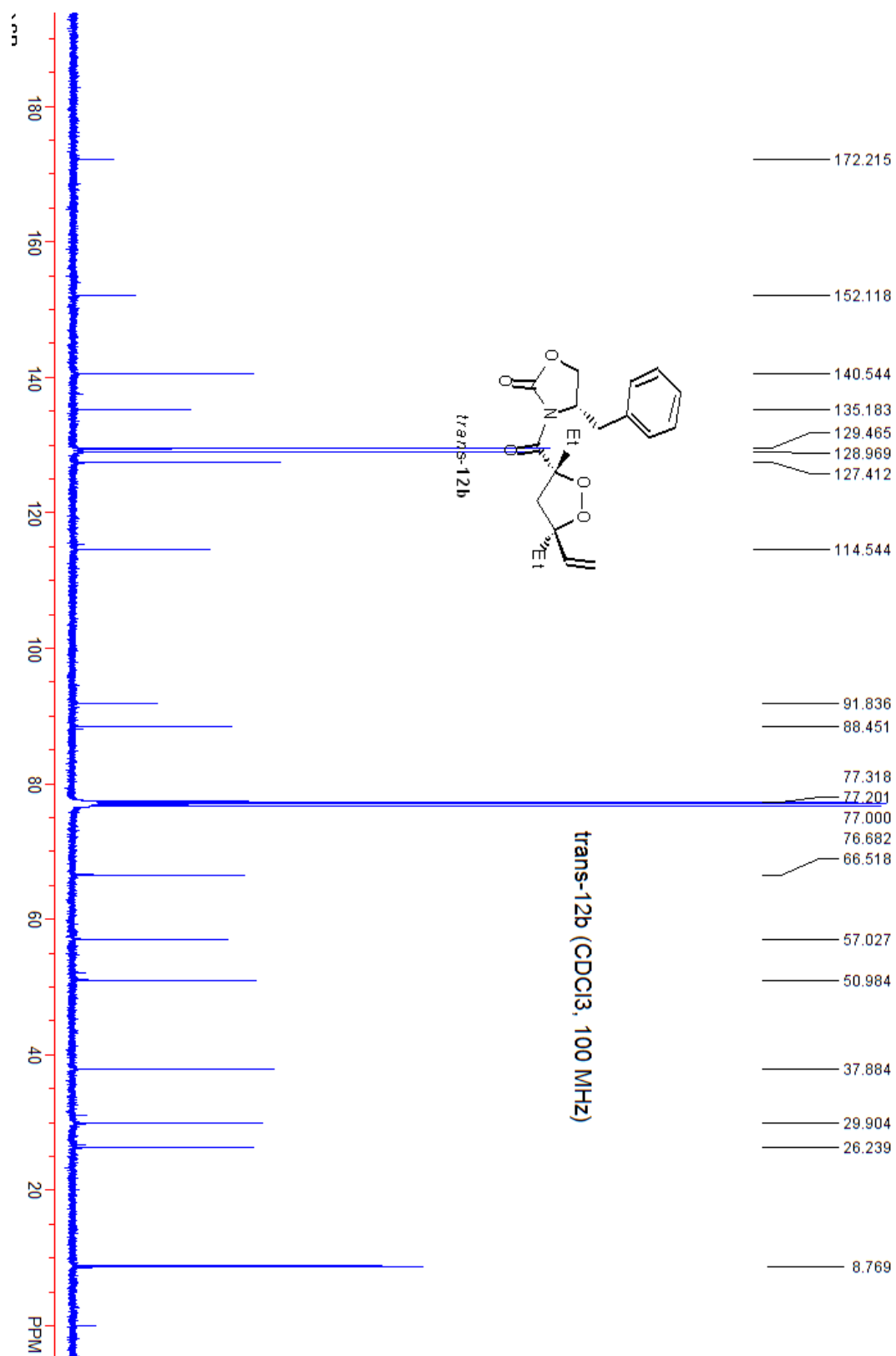


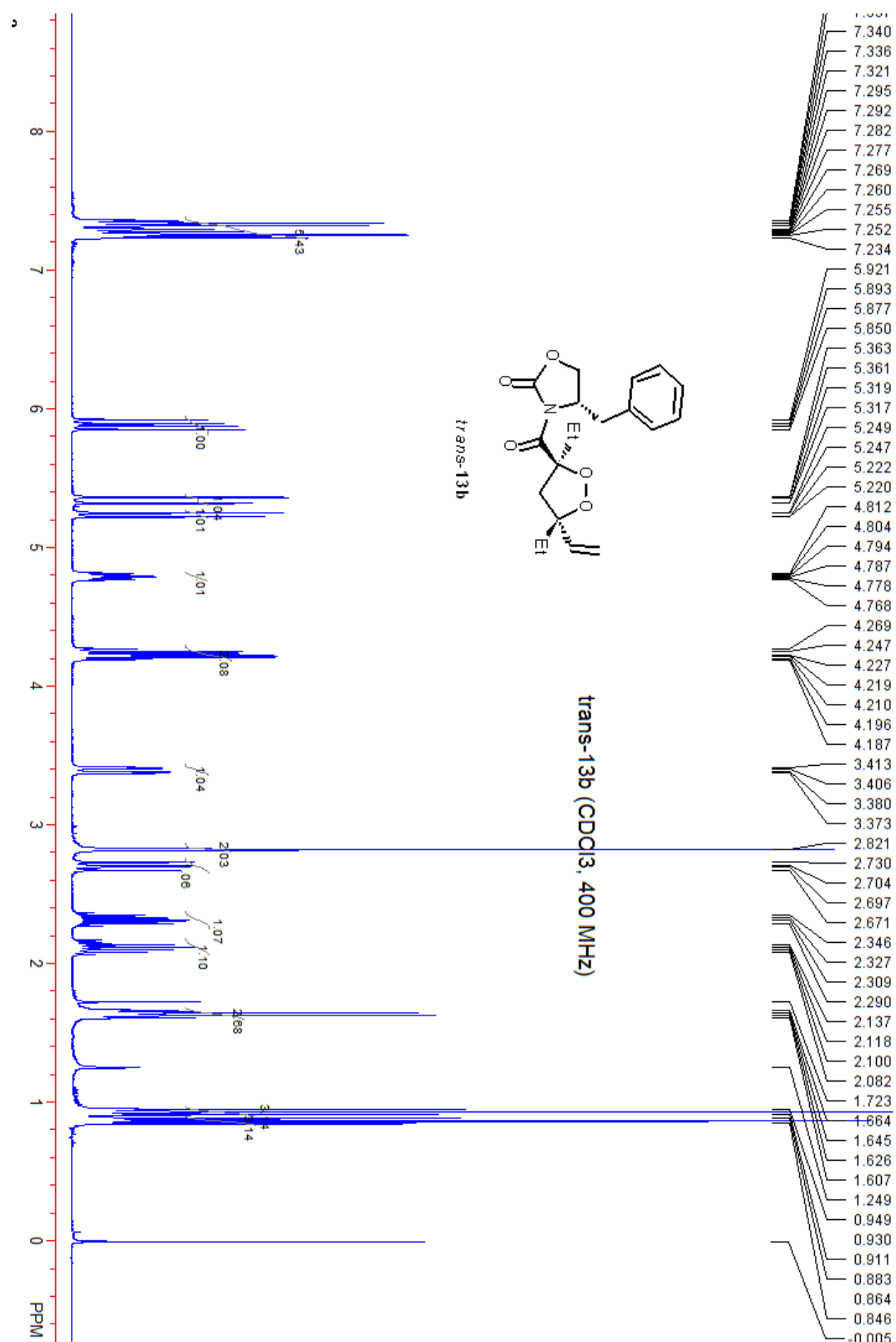


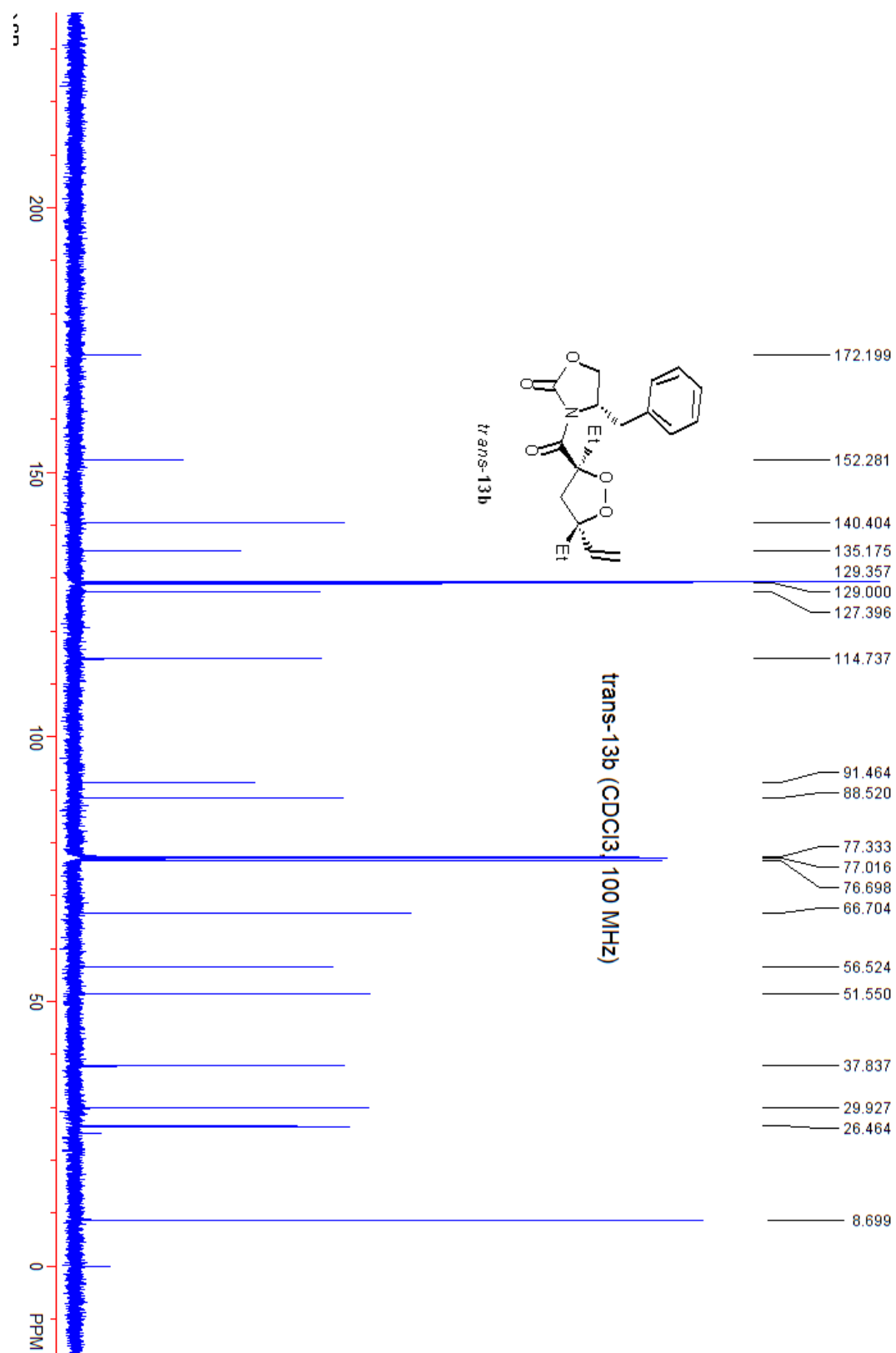


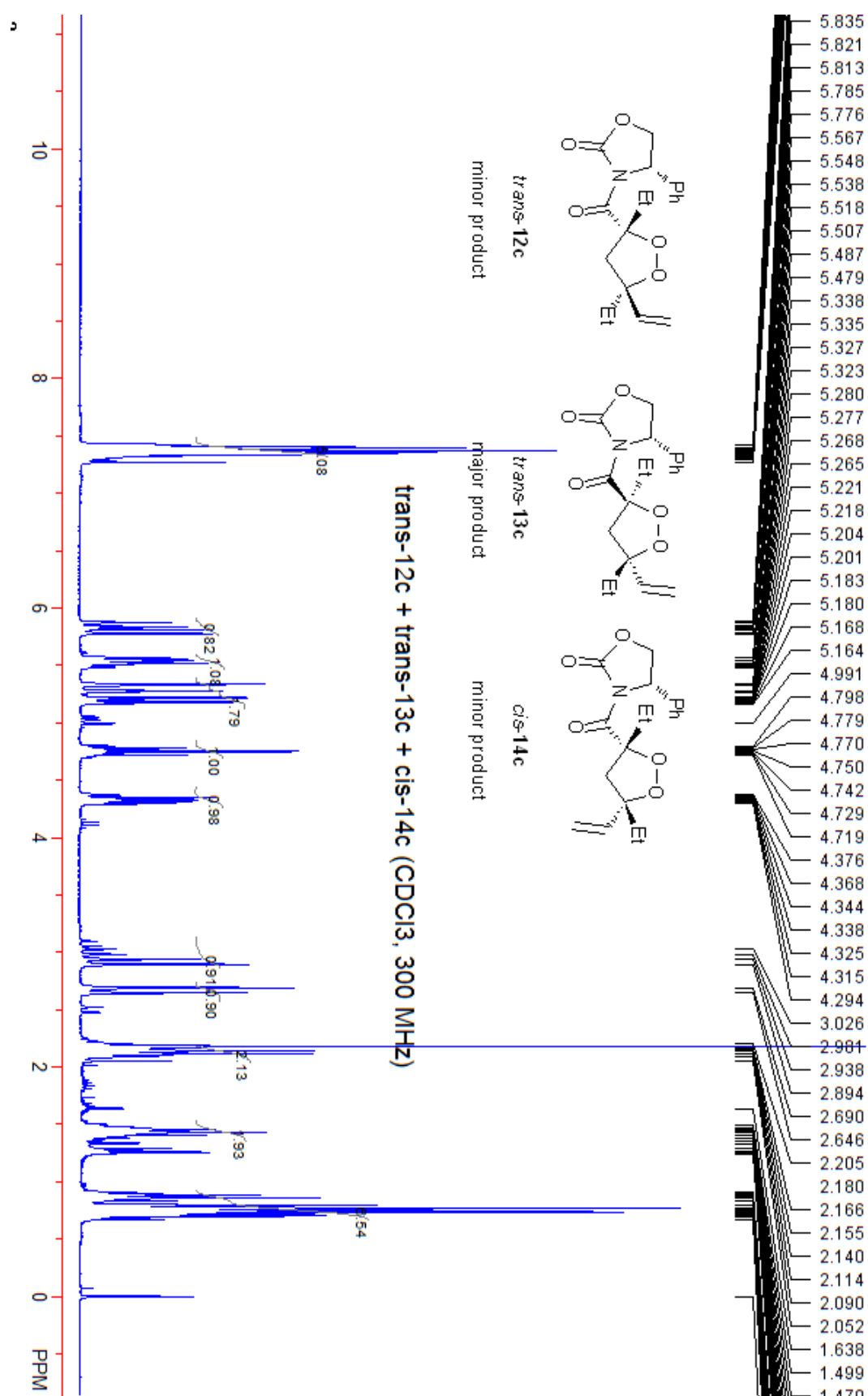


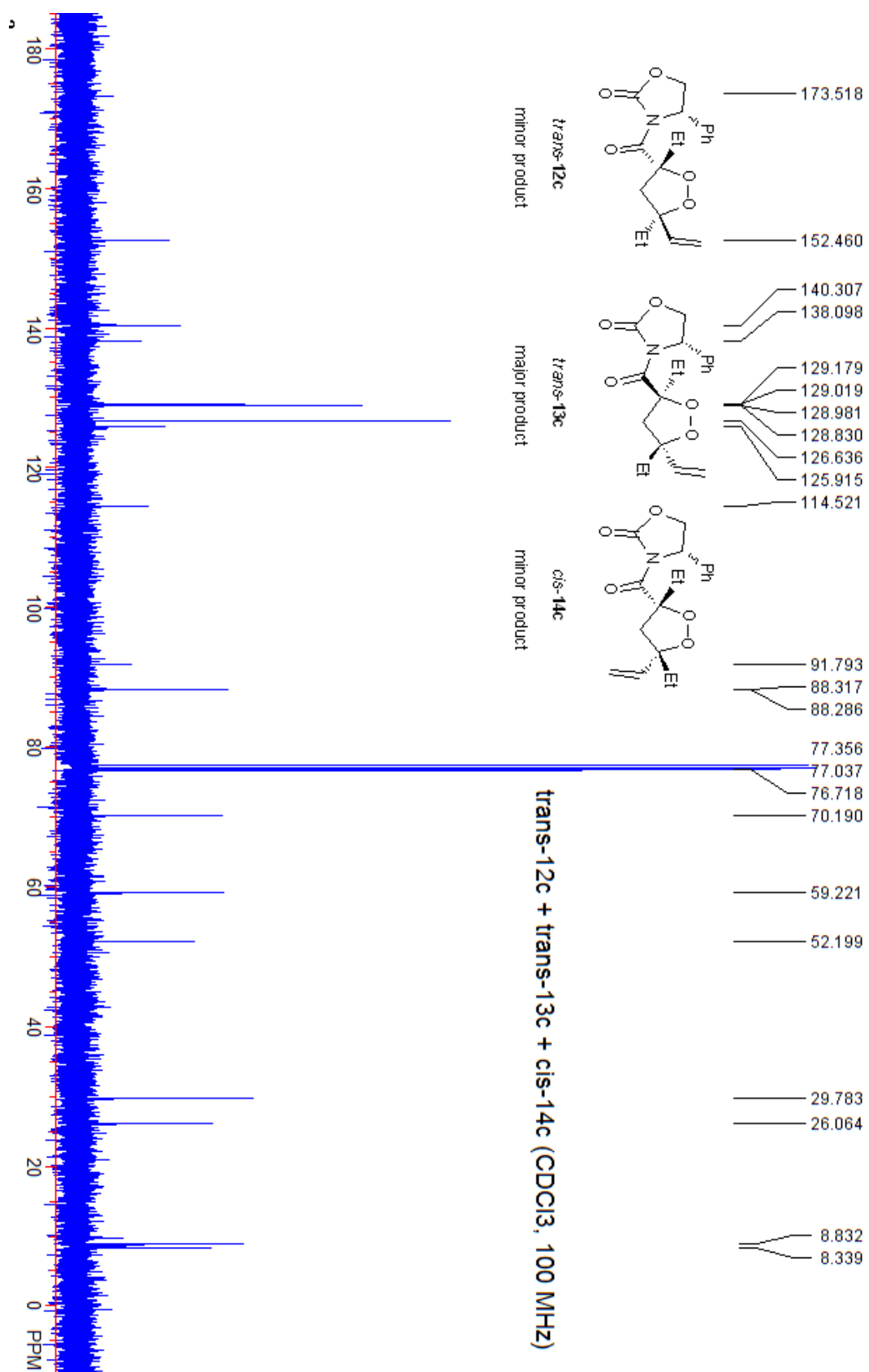


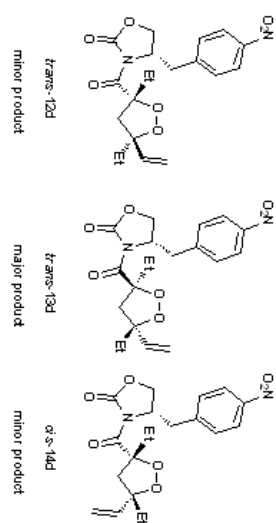
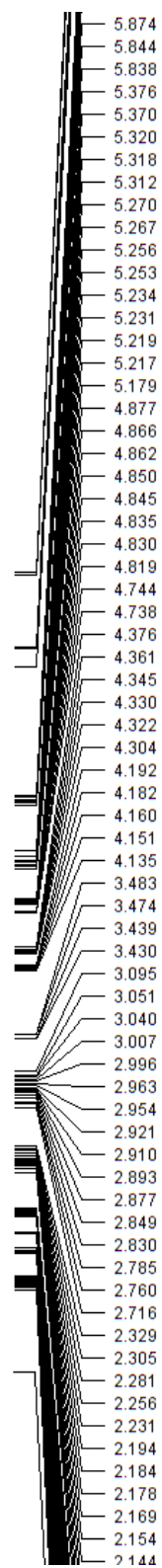




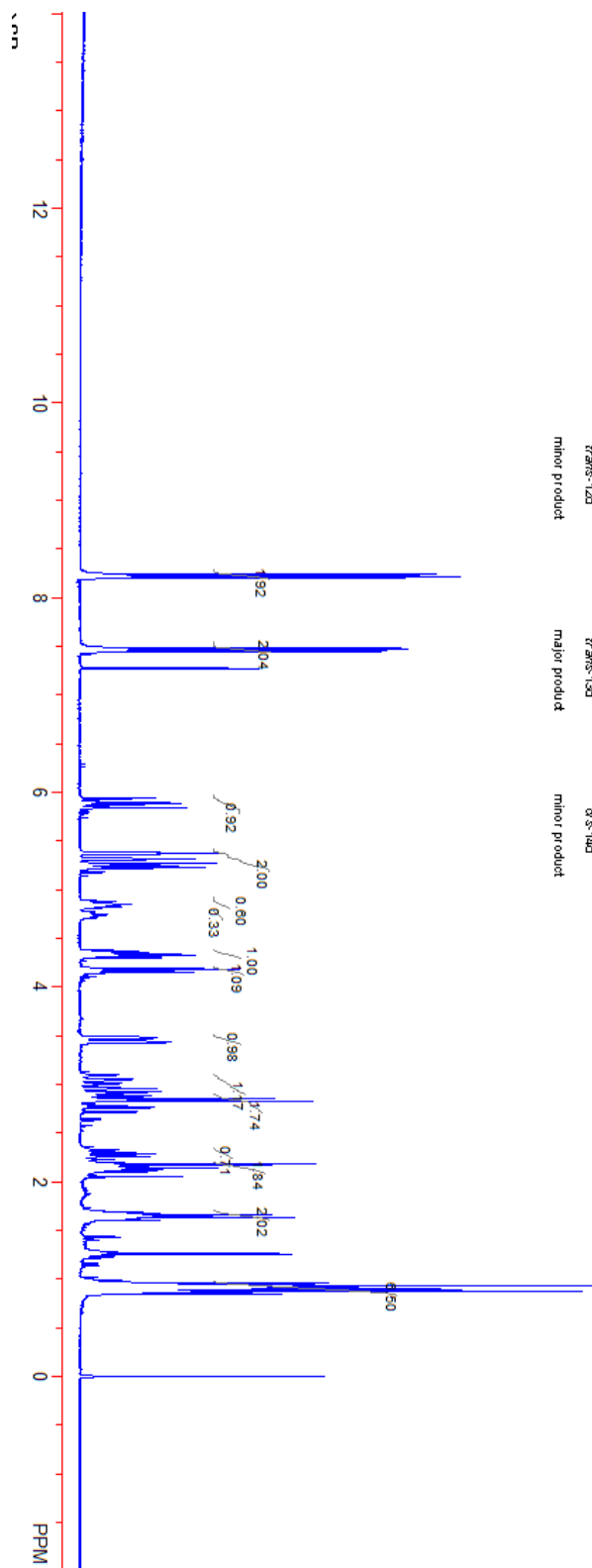


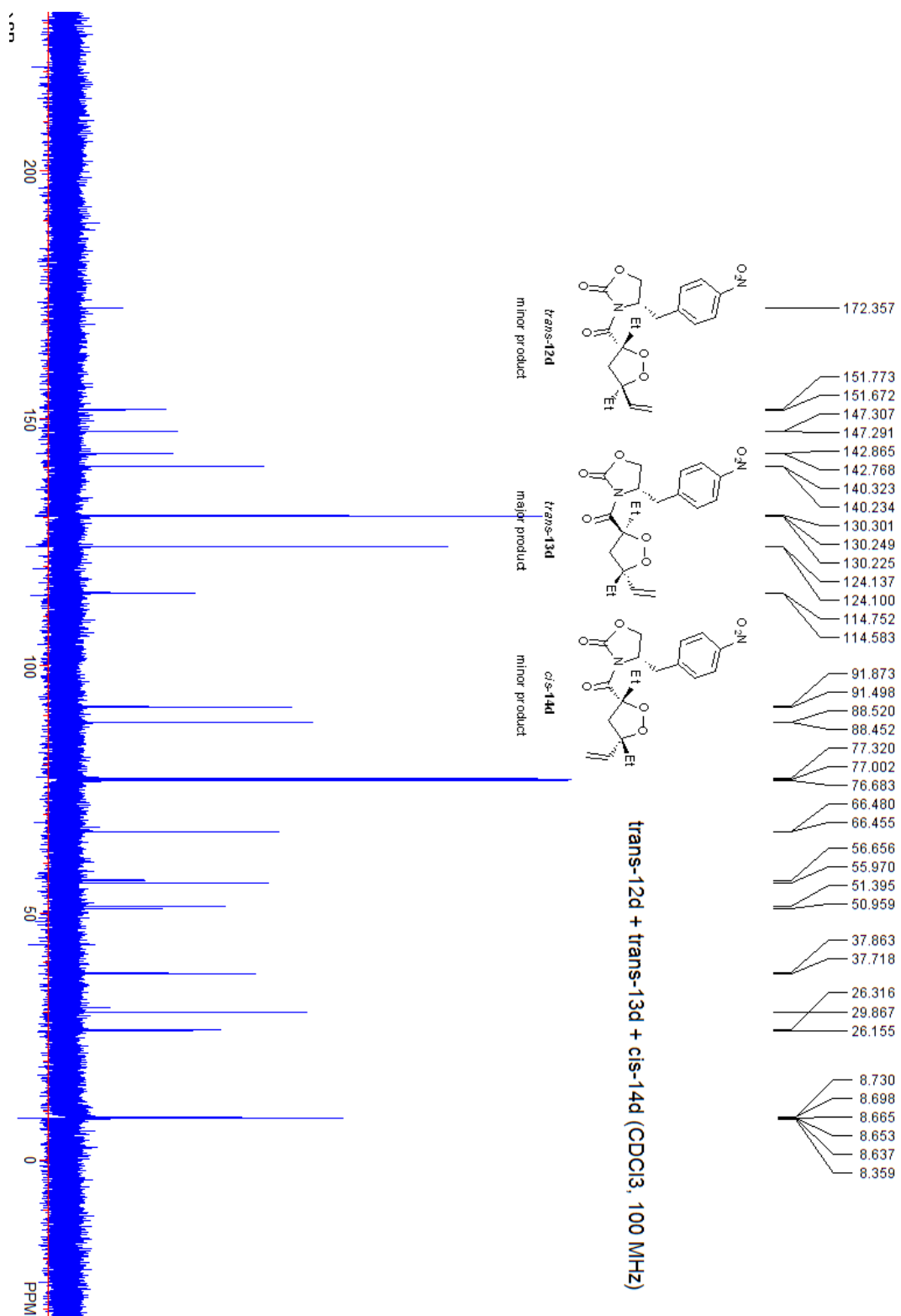


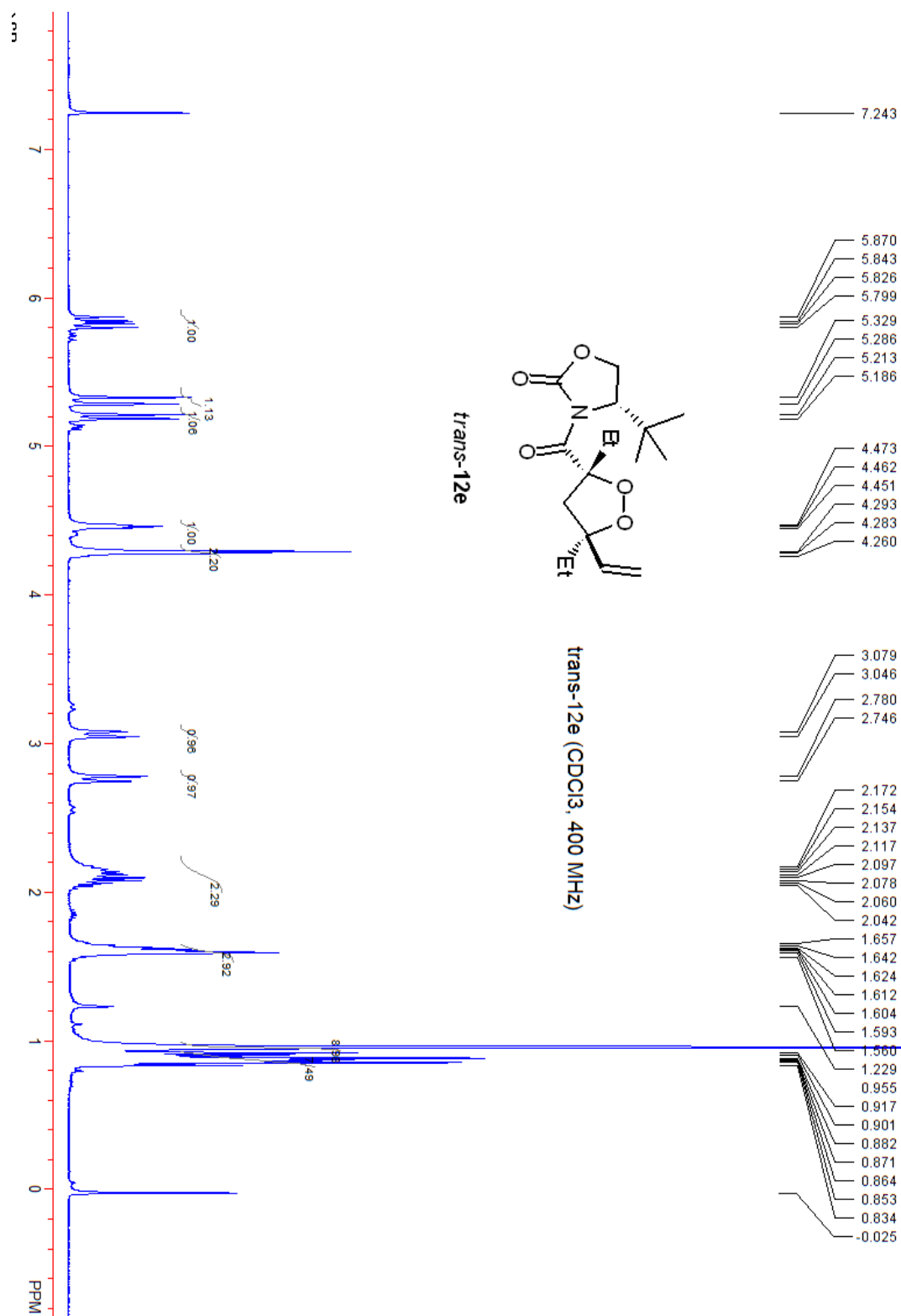


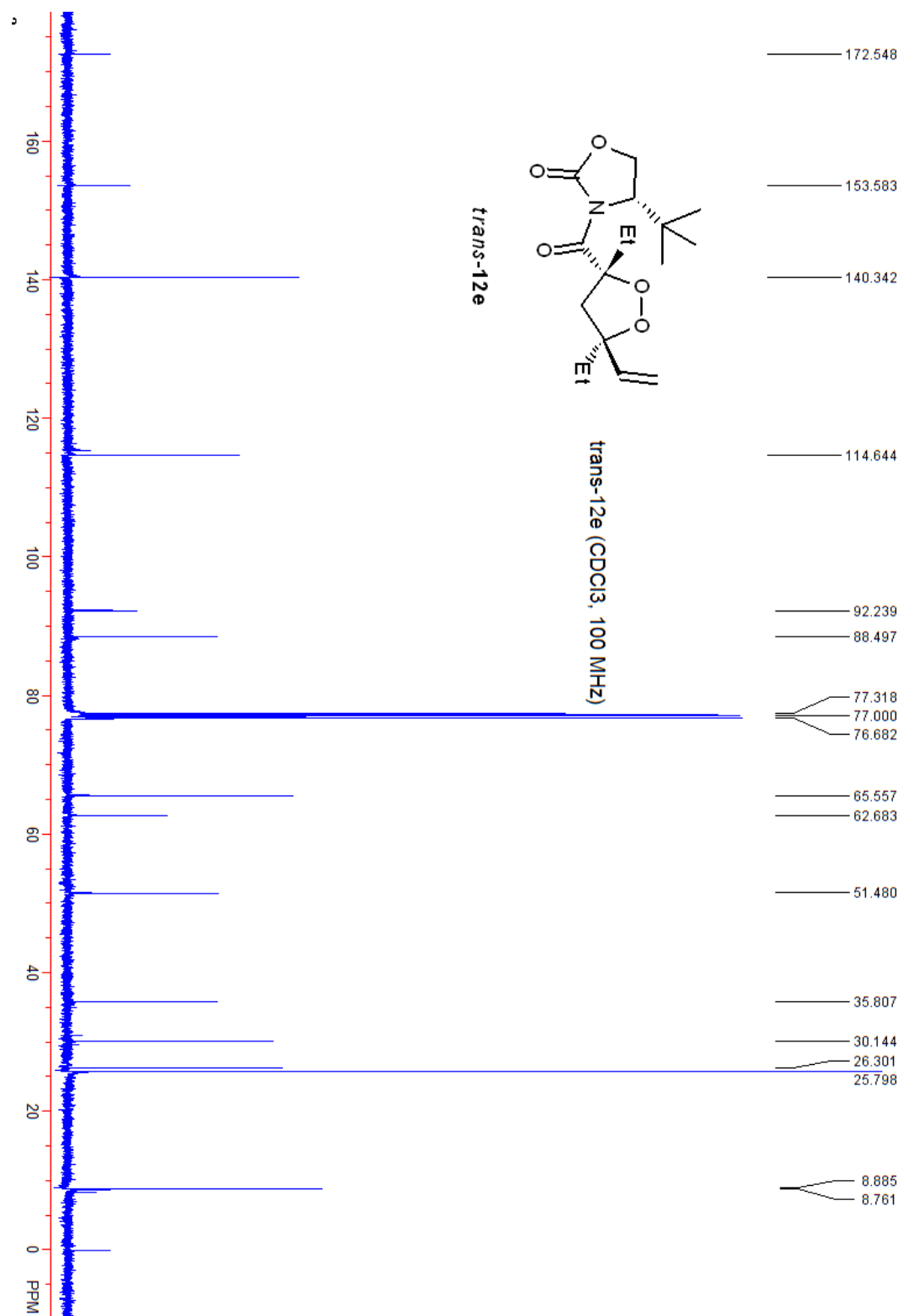


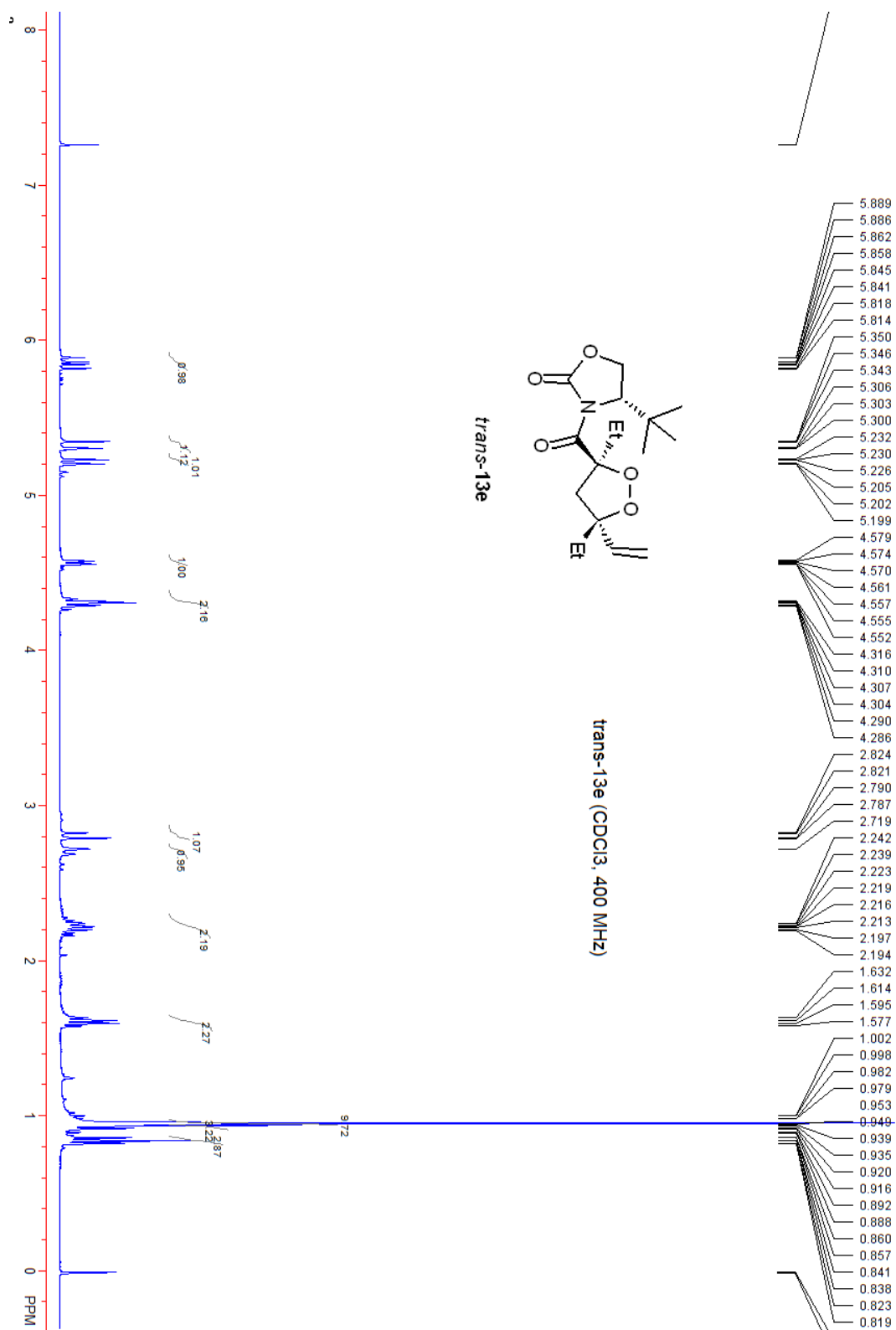
trans-12d + trans-13d + cis-14d (CDCl₃, 300 MHz)

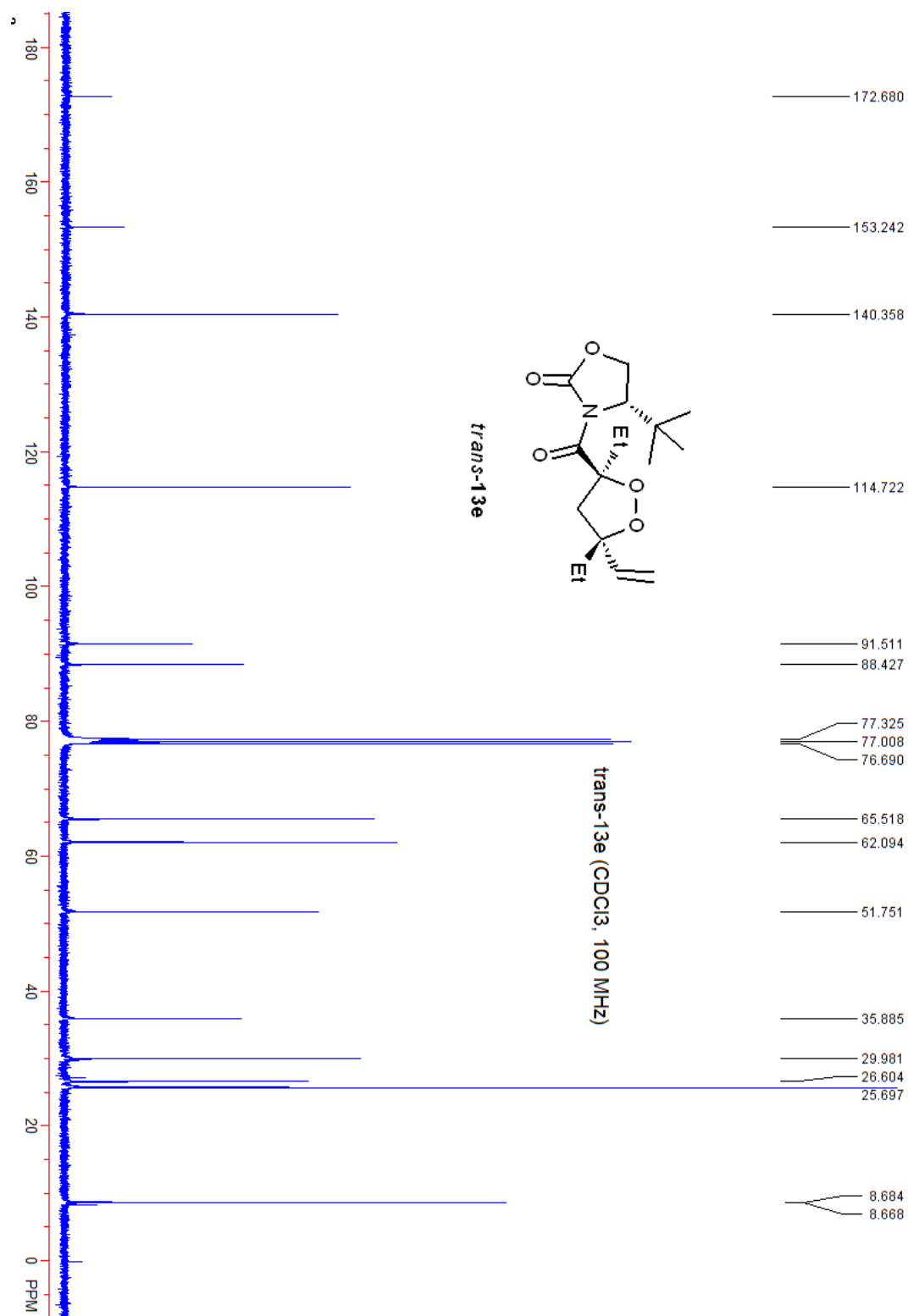


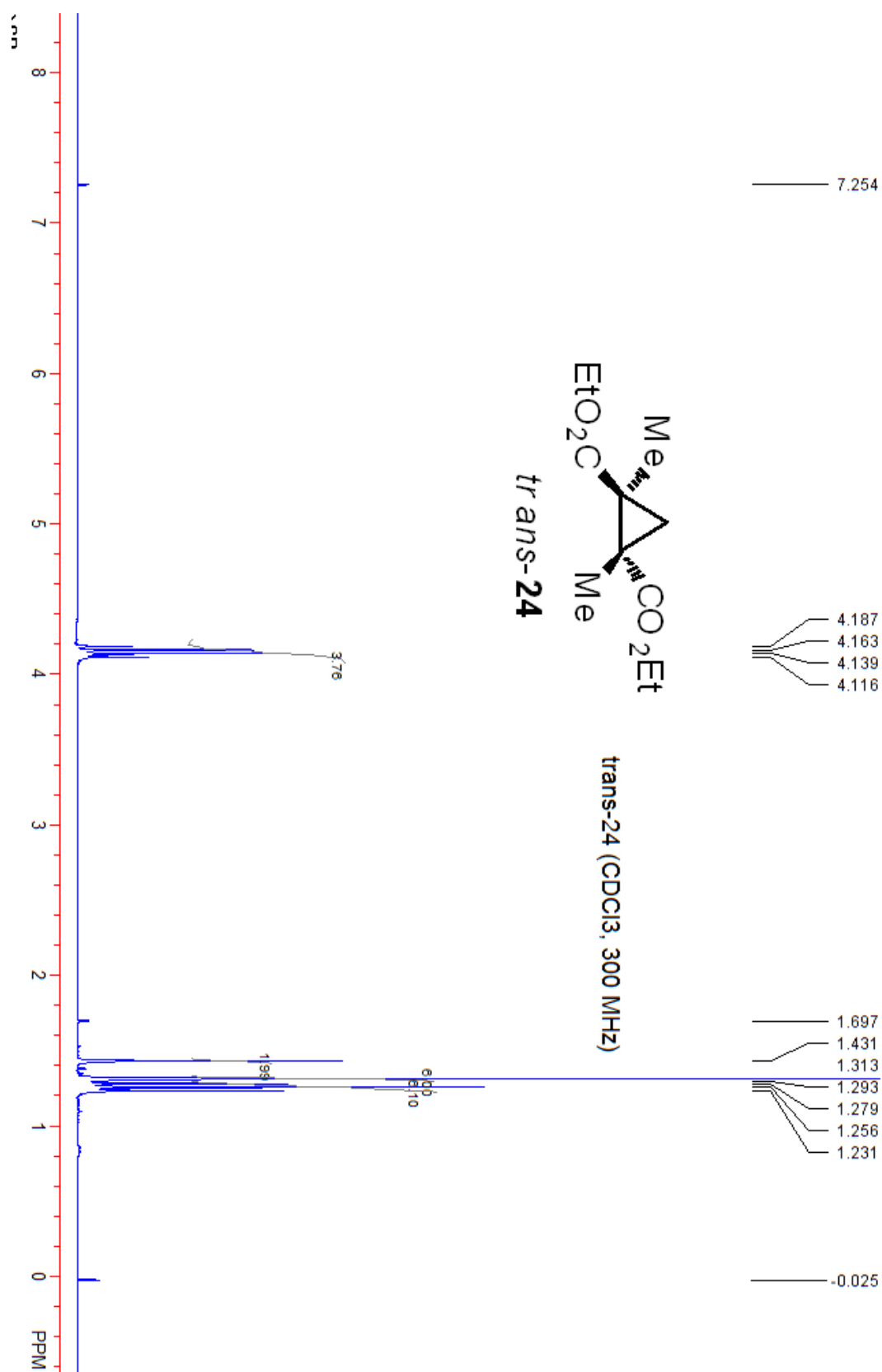


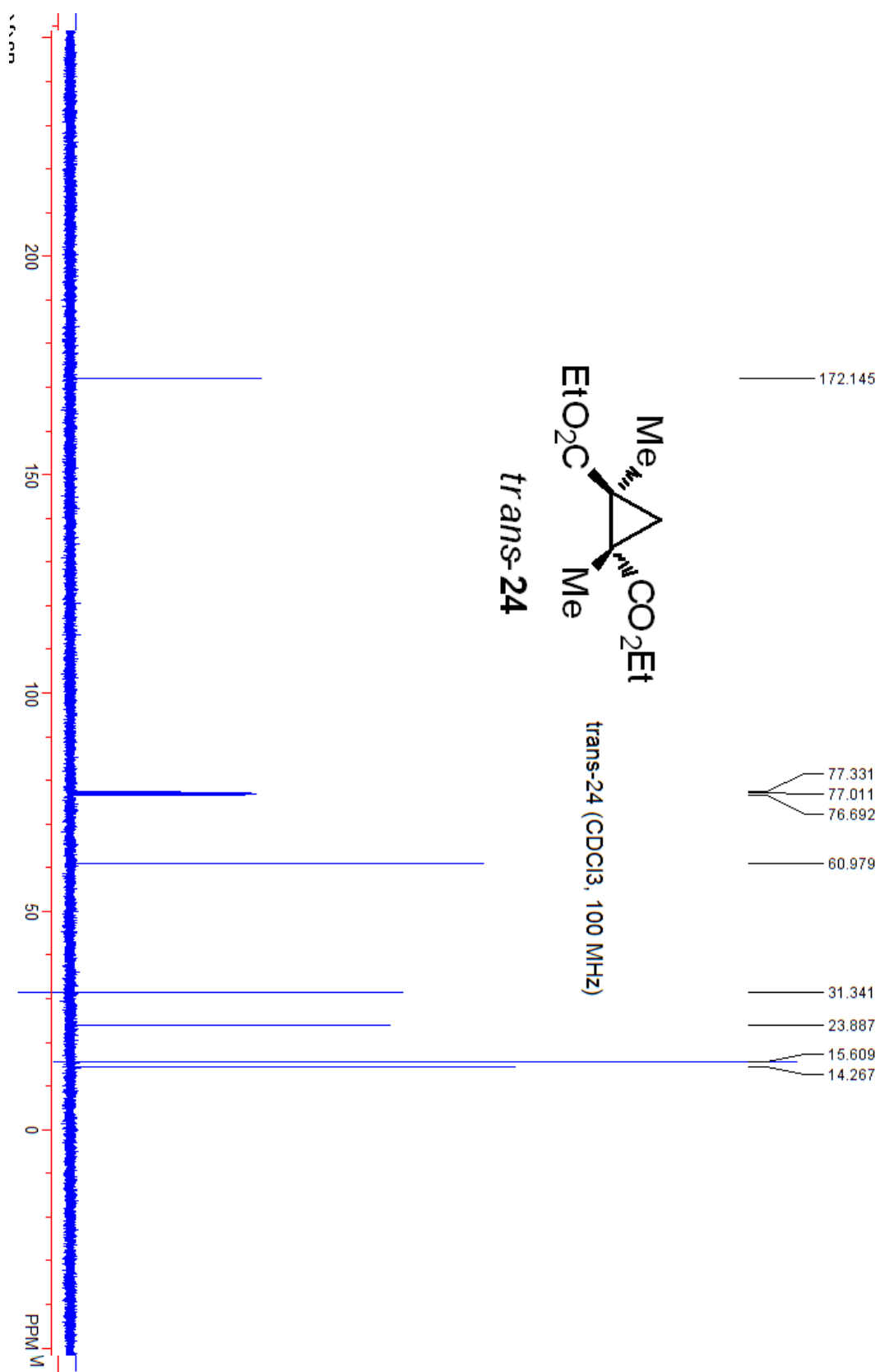


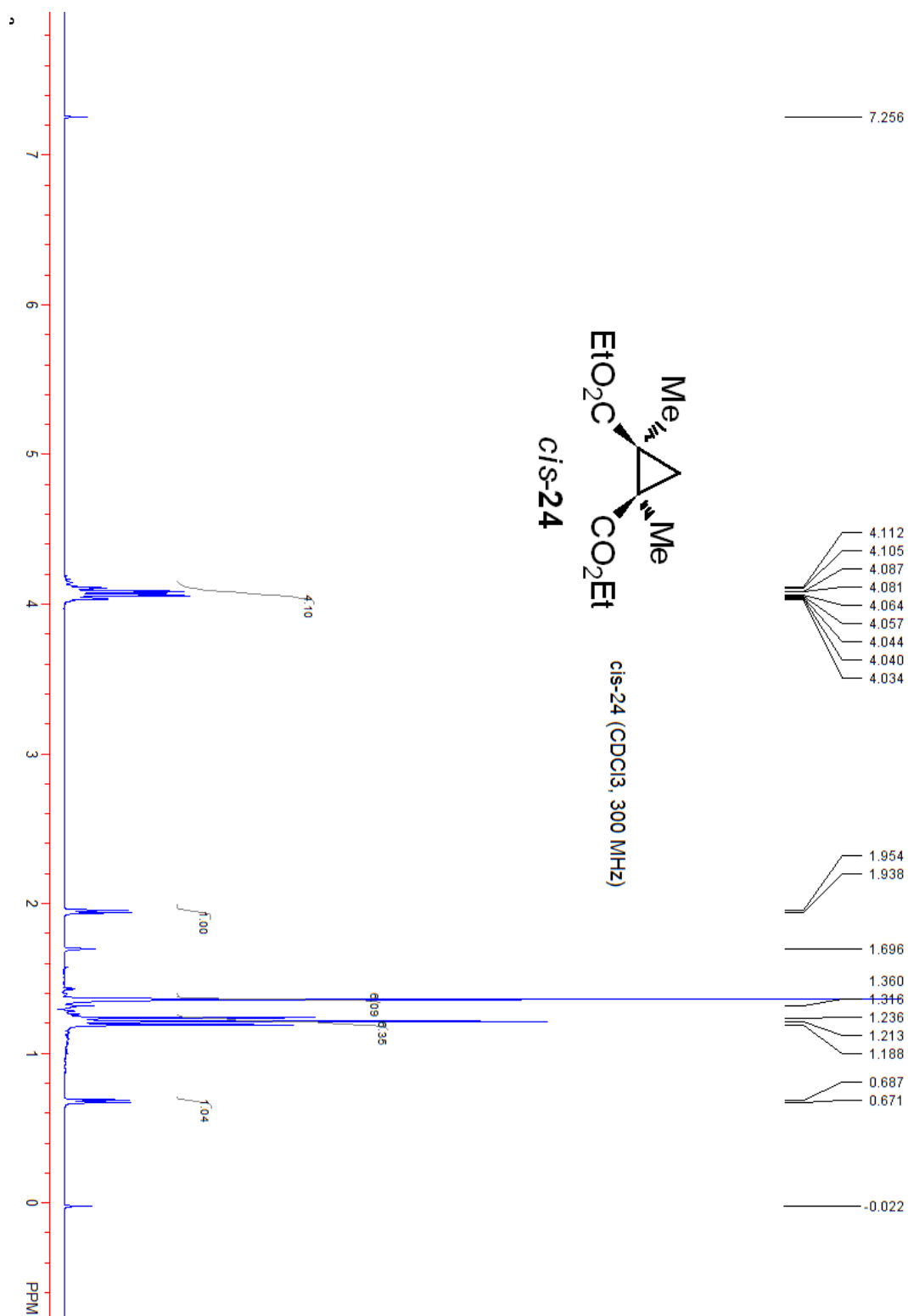


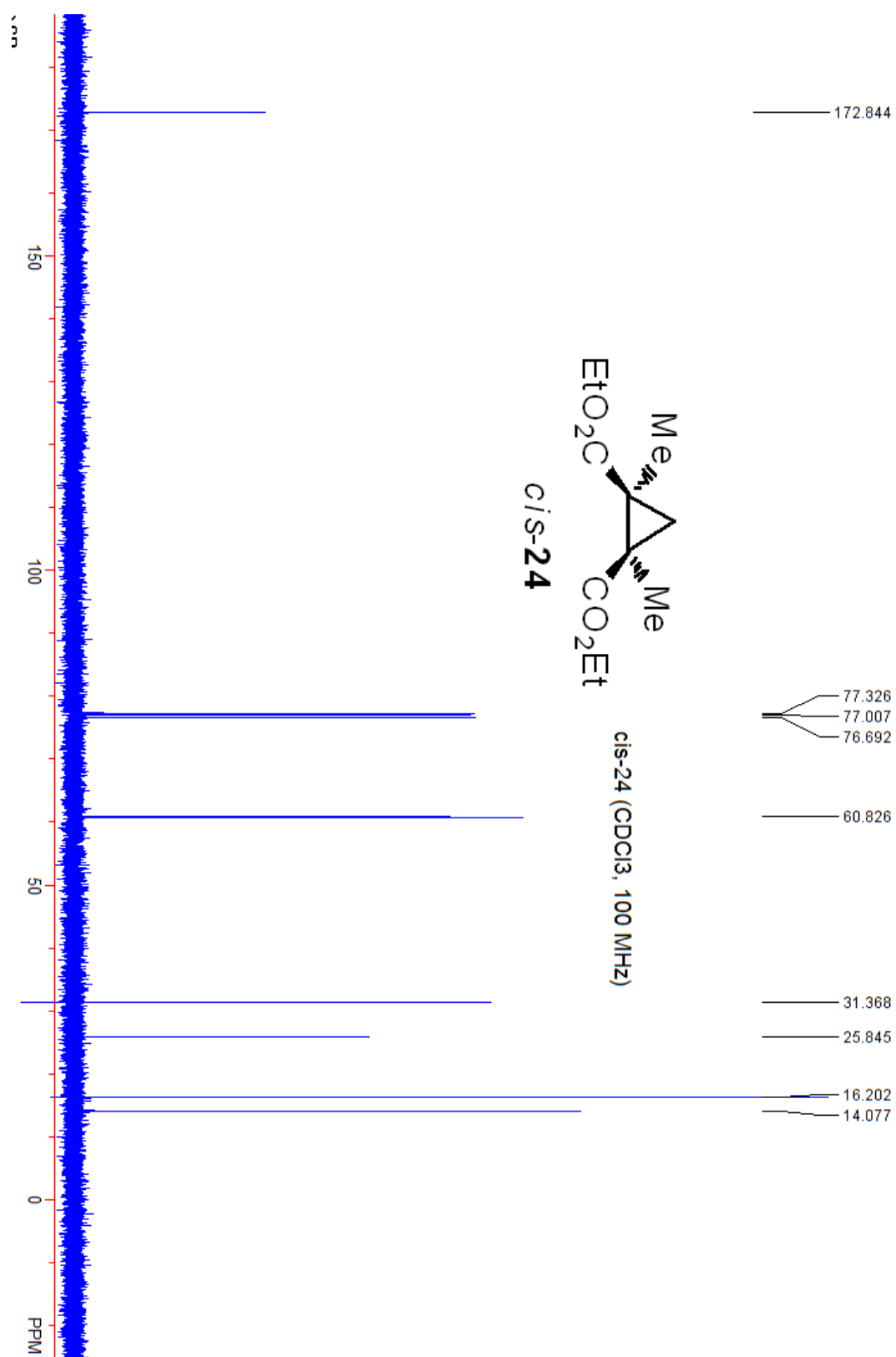


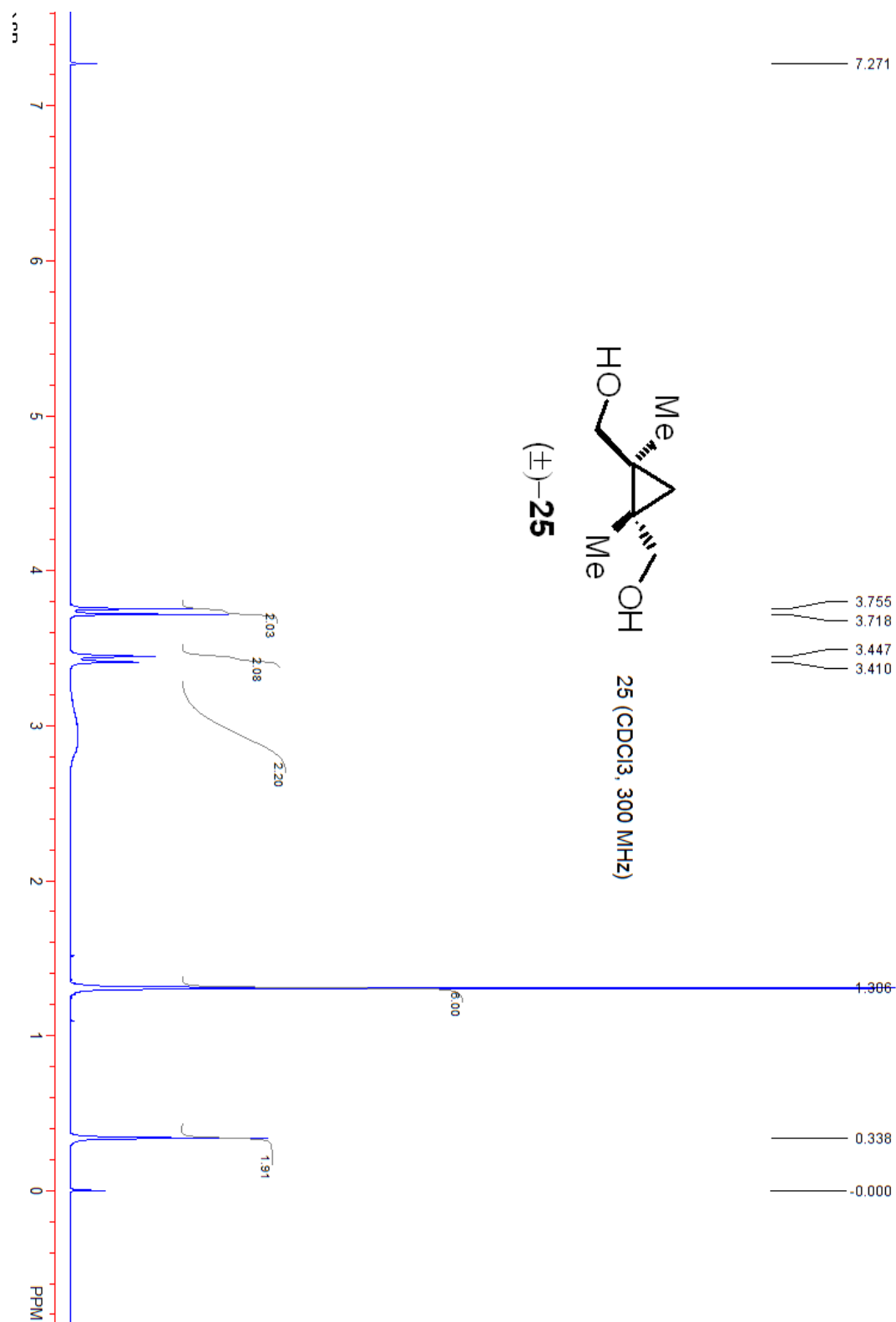


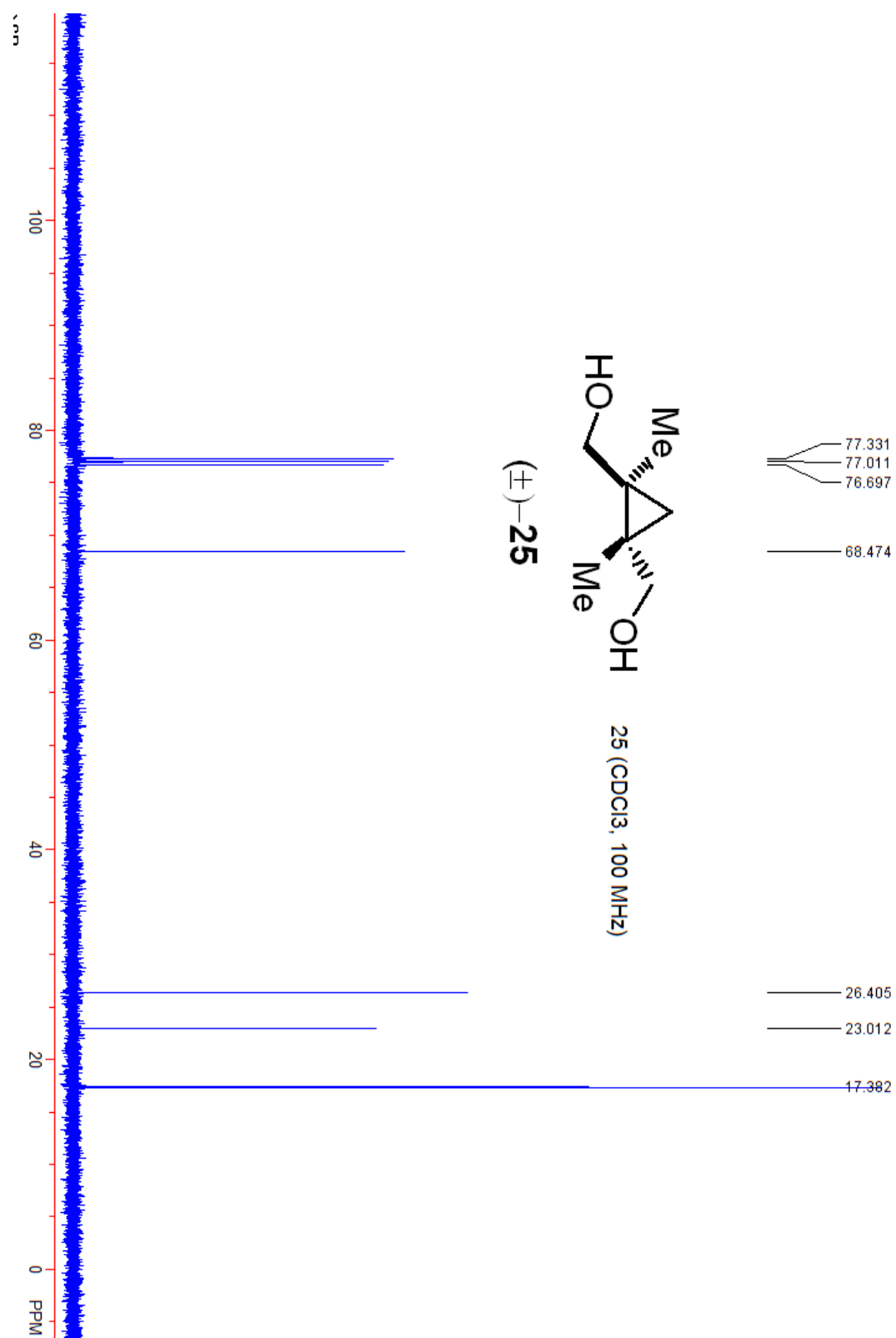


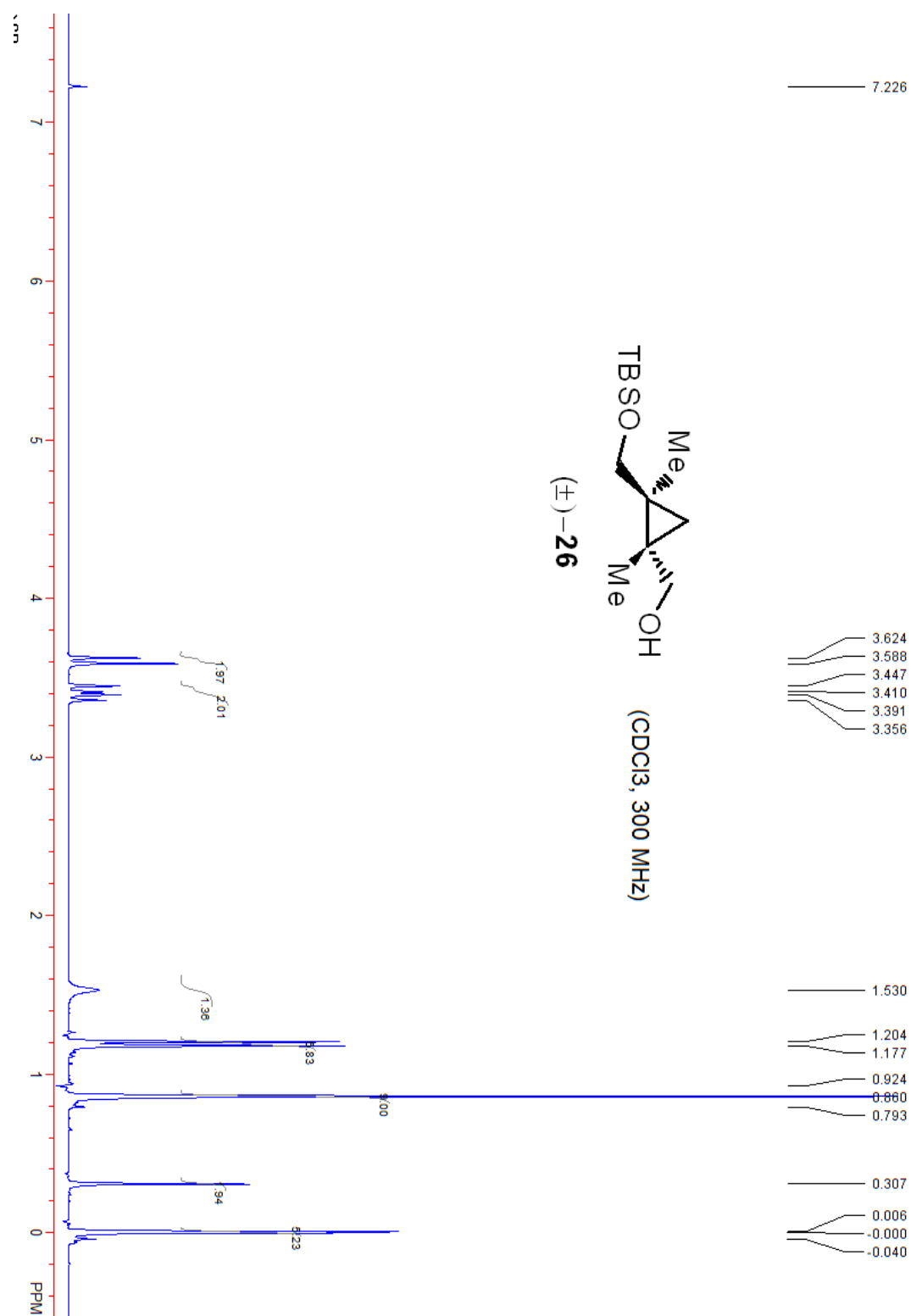


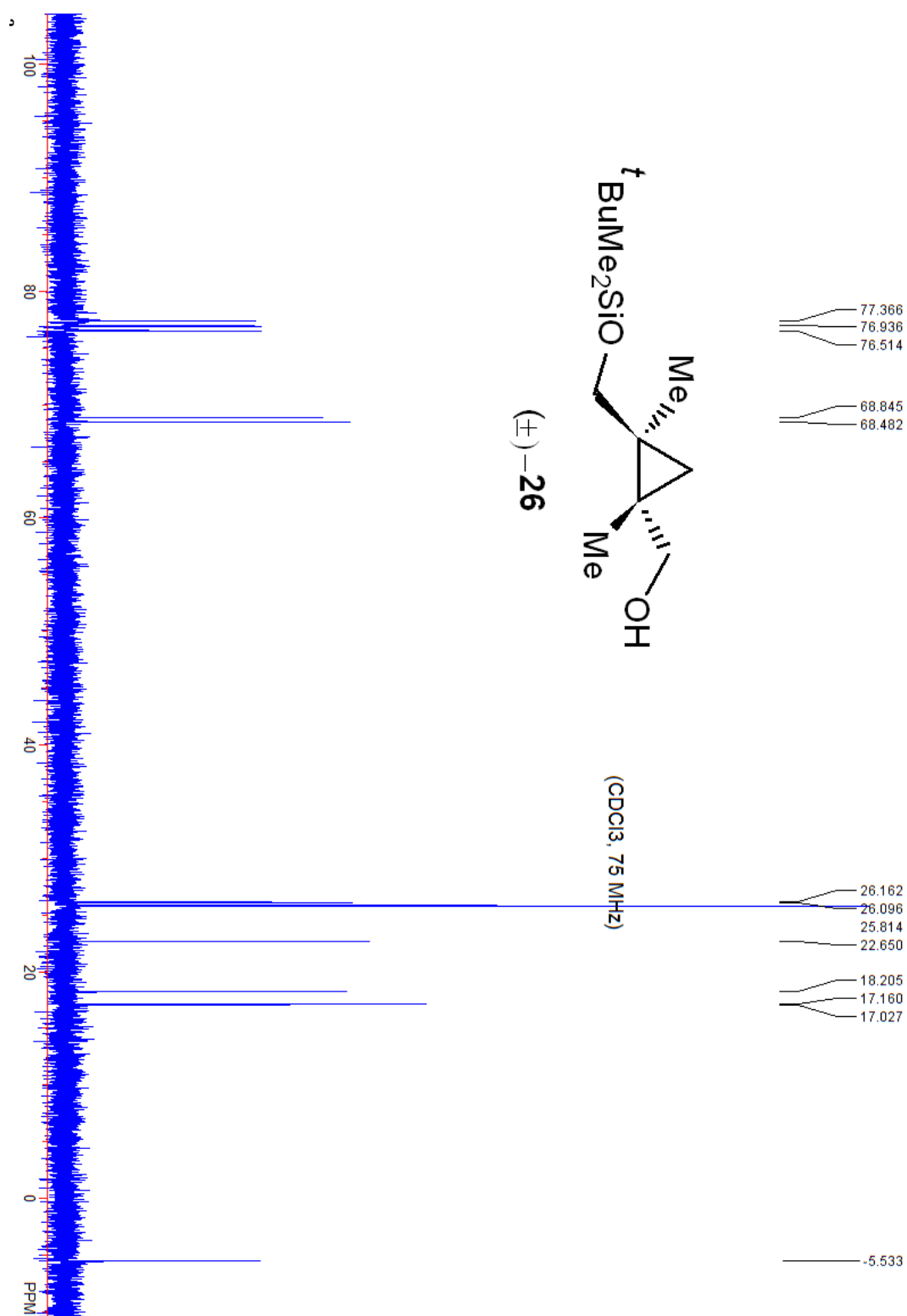


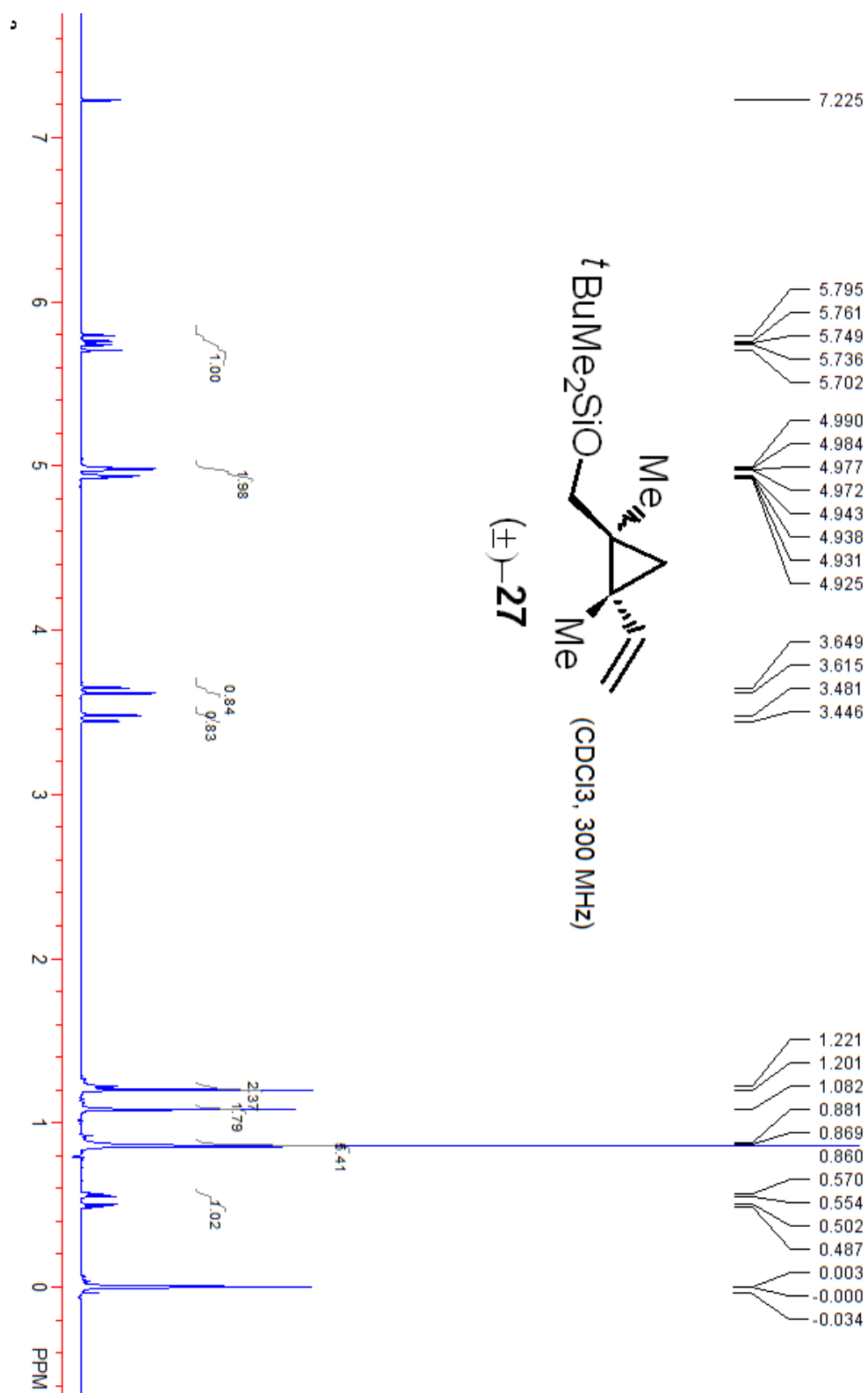


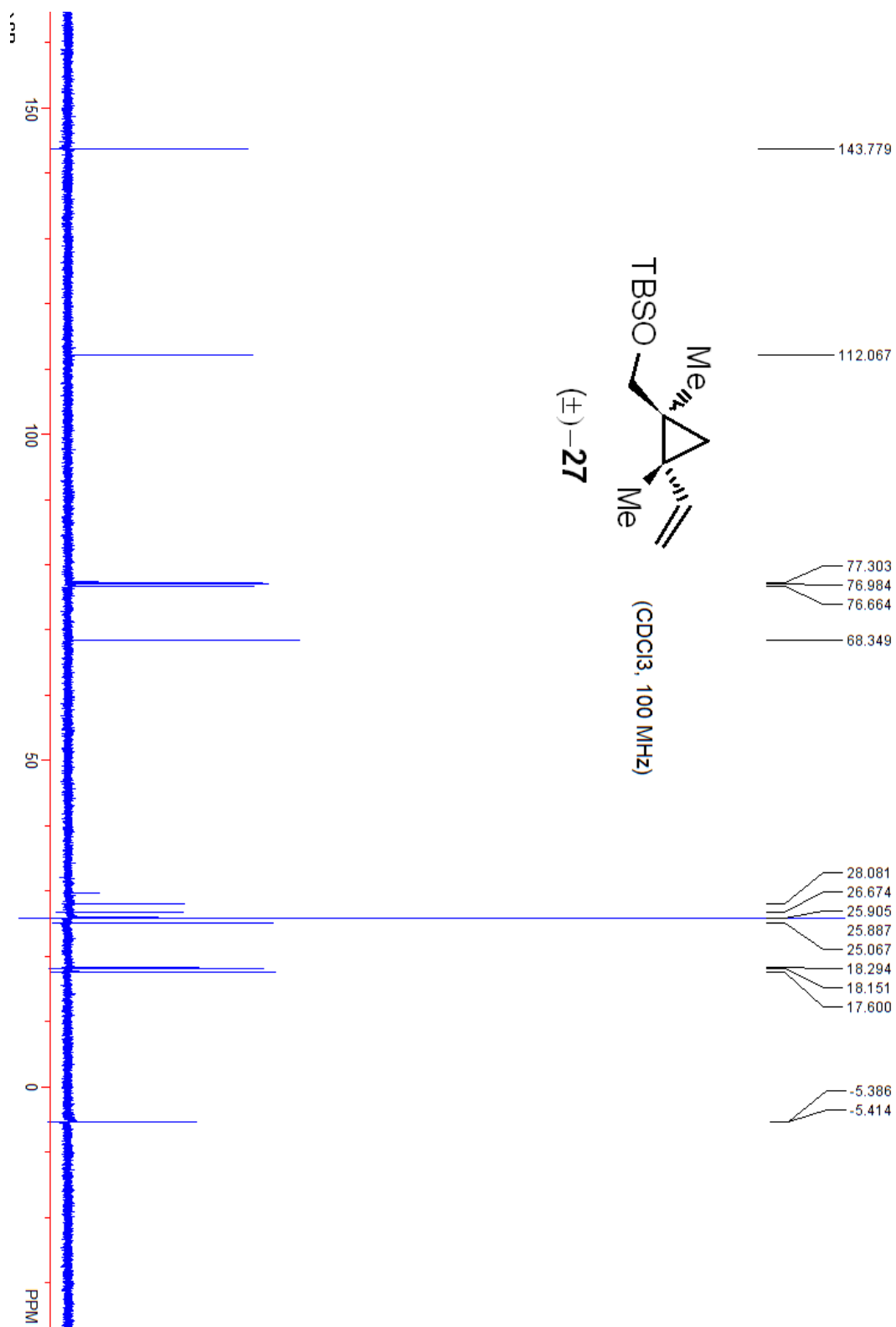


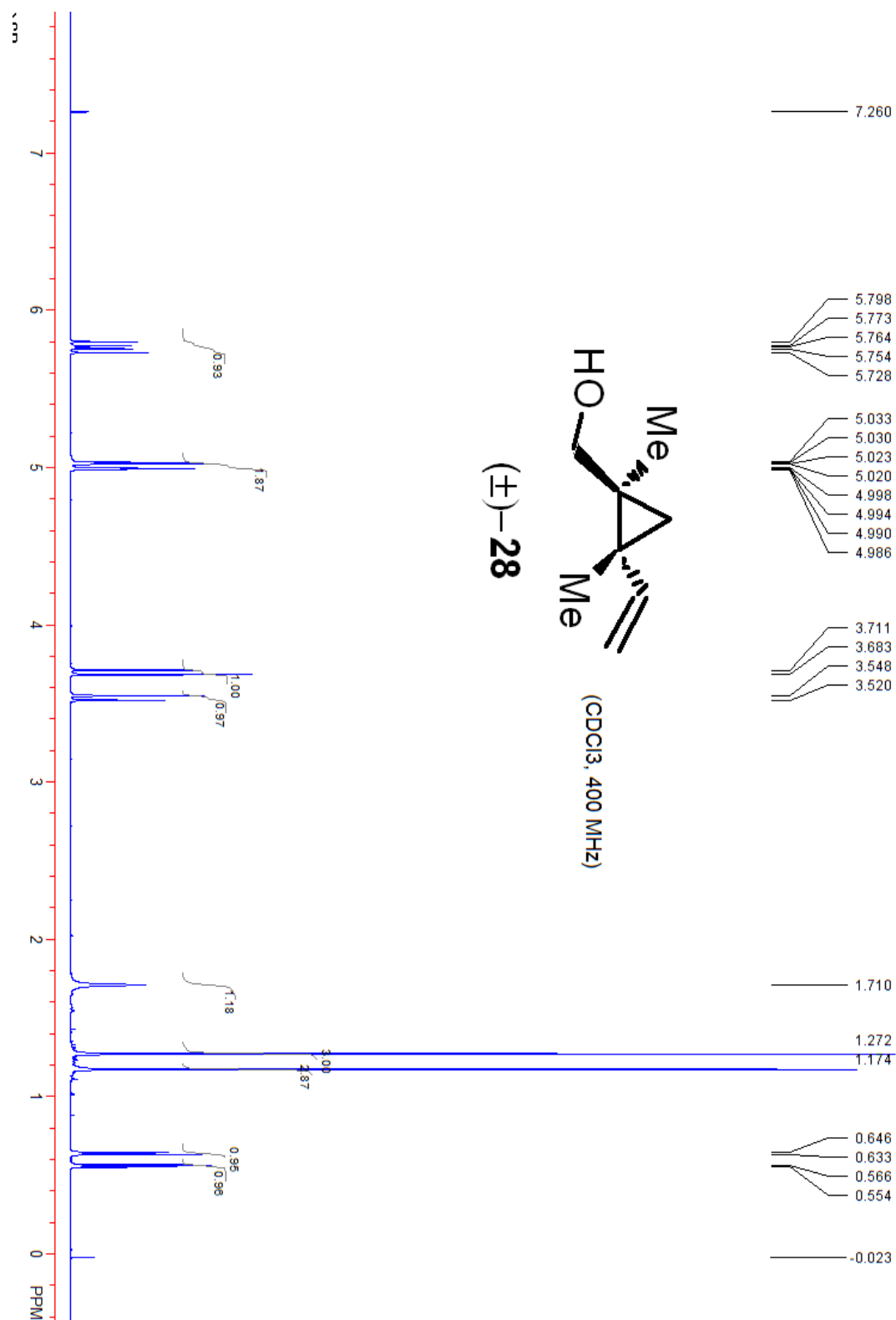


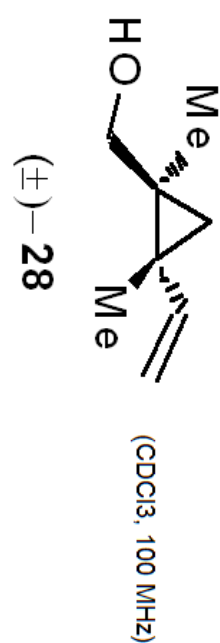










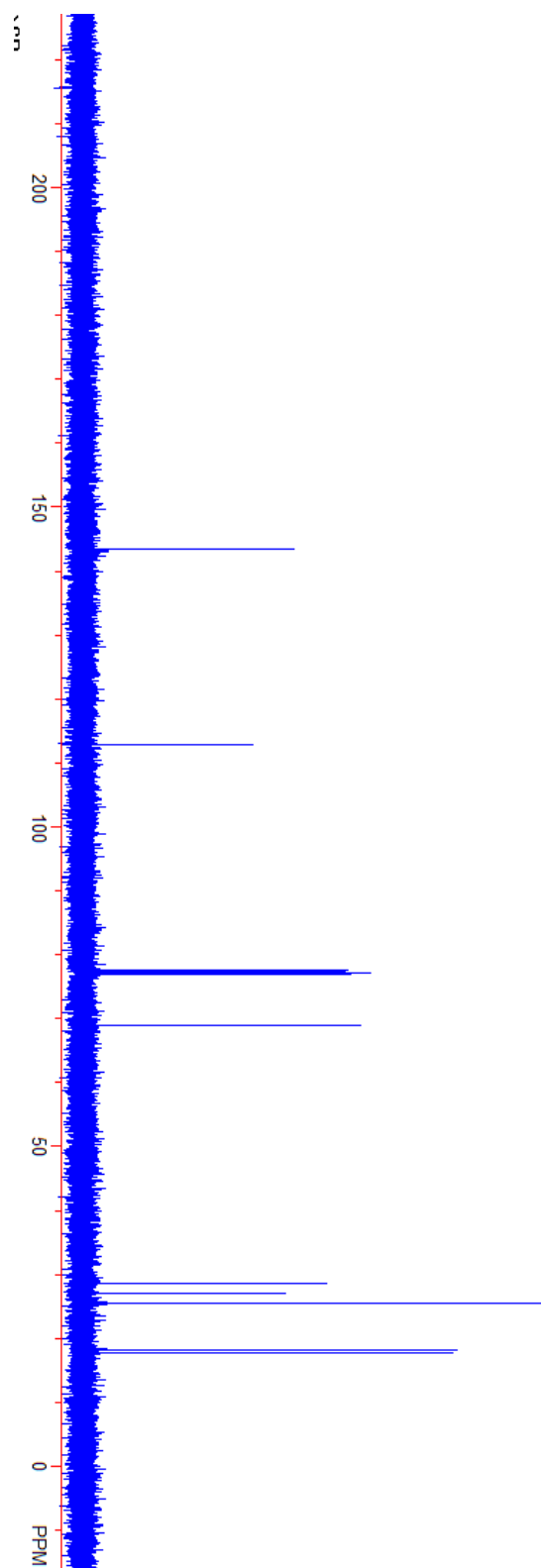


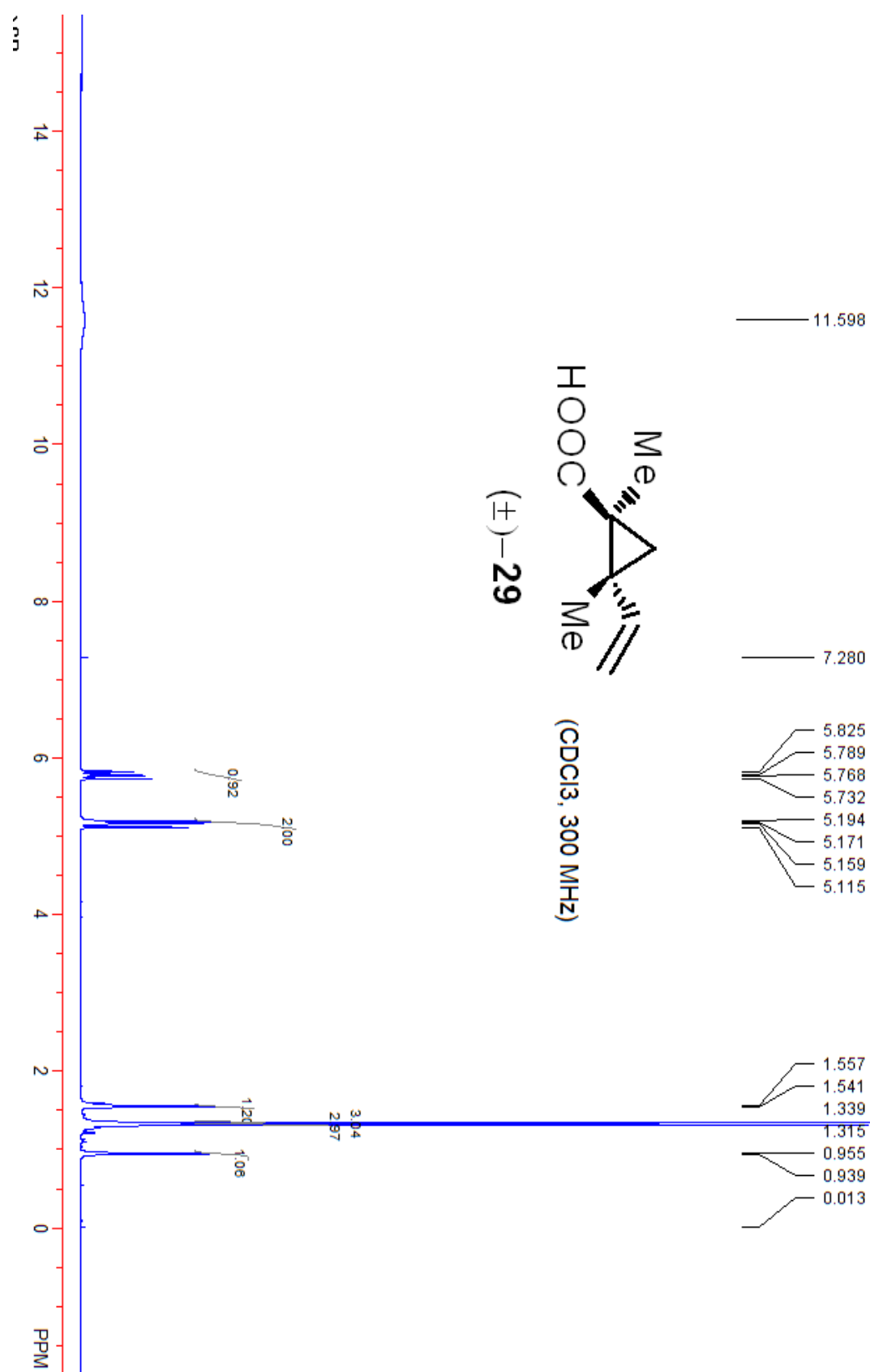
143.362

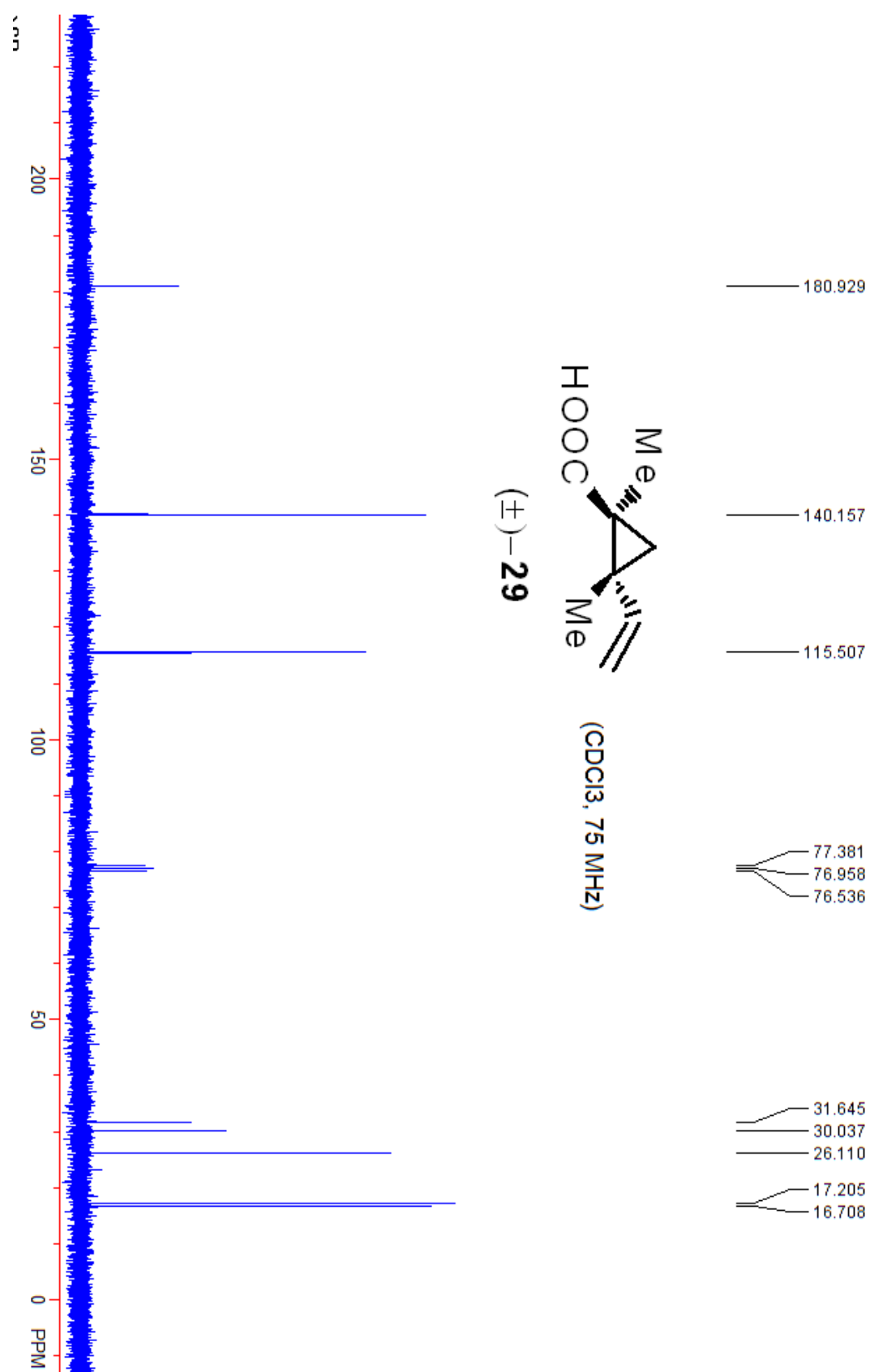
112.985

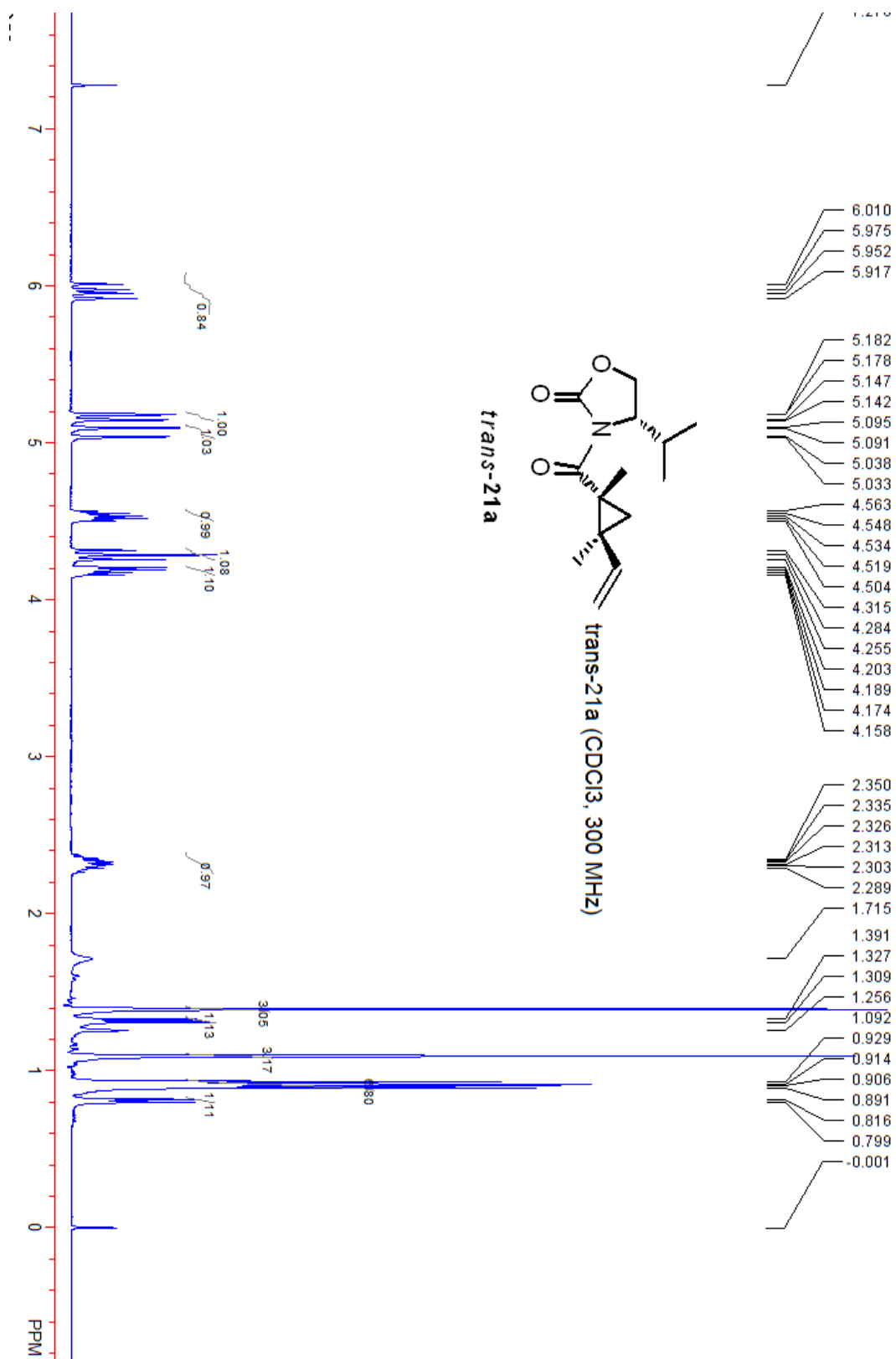
77.652
77.336
77.020
69.030

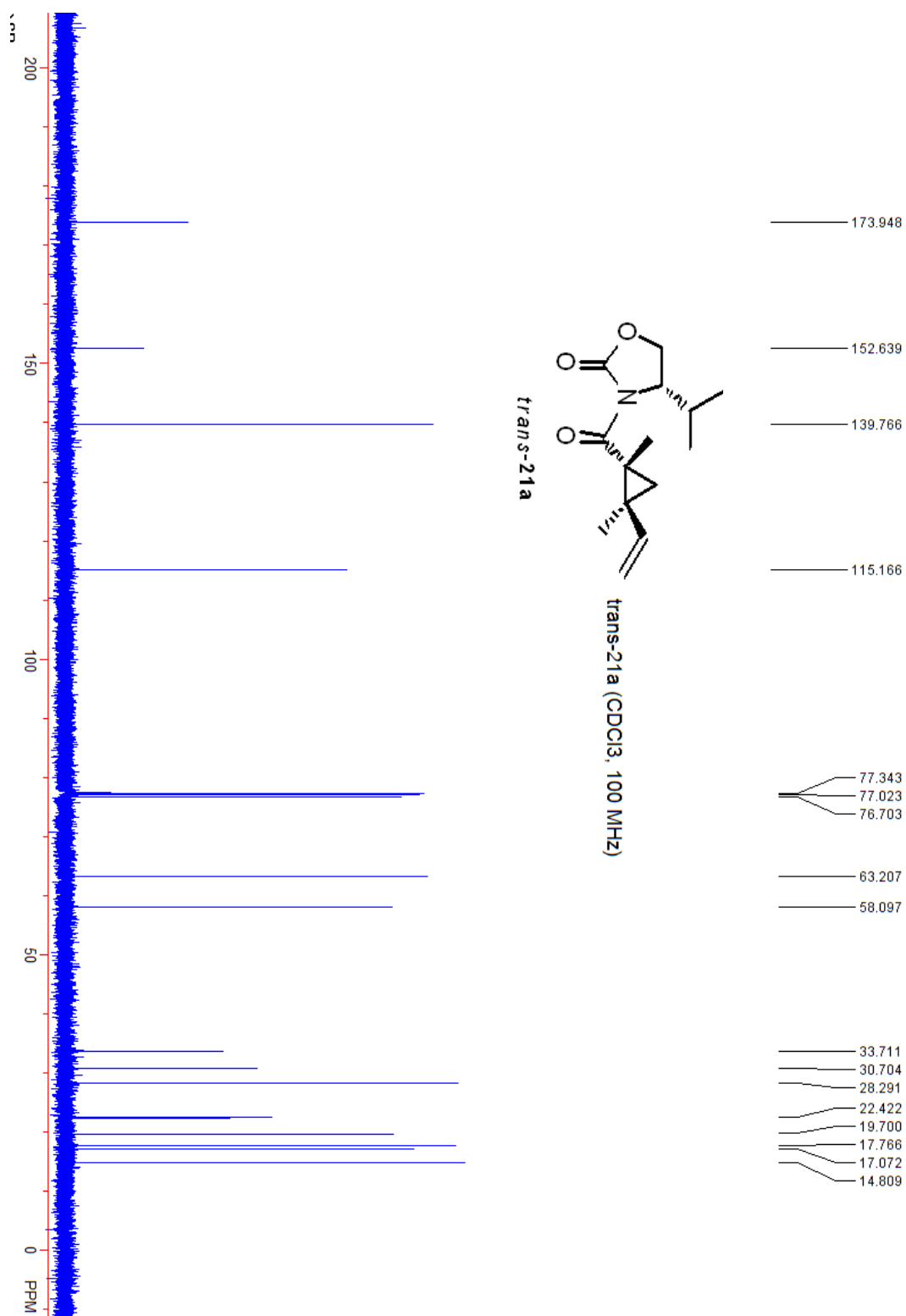
28.640
27.193
25.676
18.288
17.939

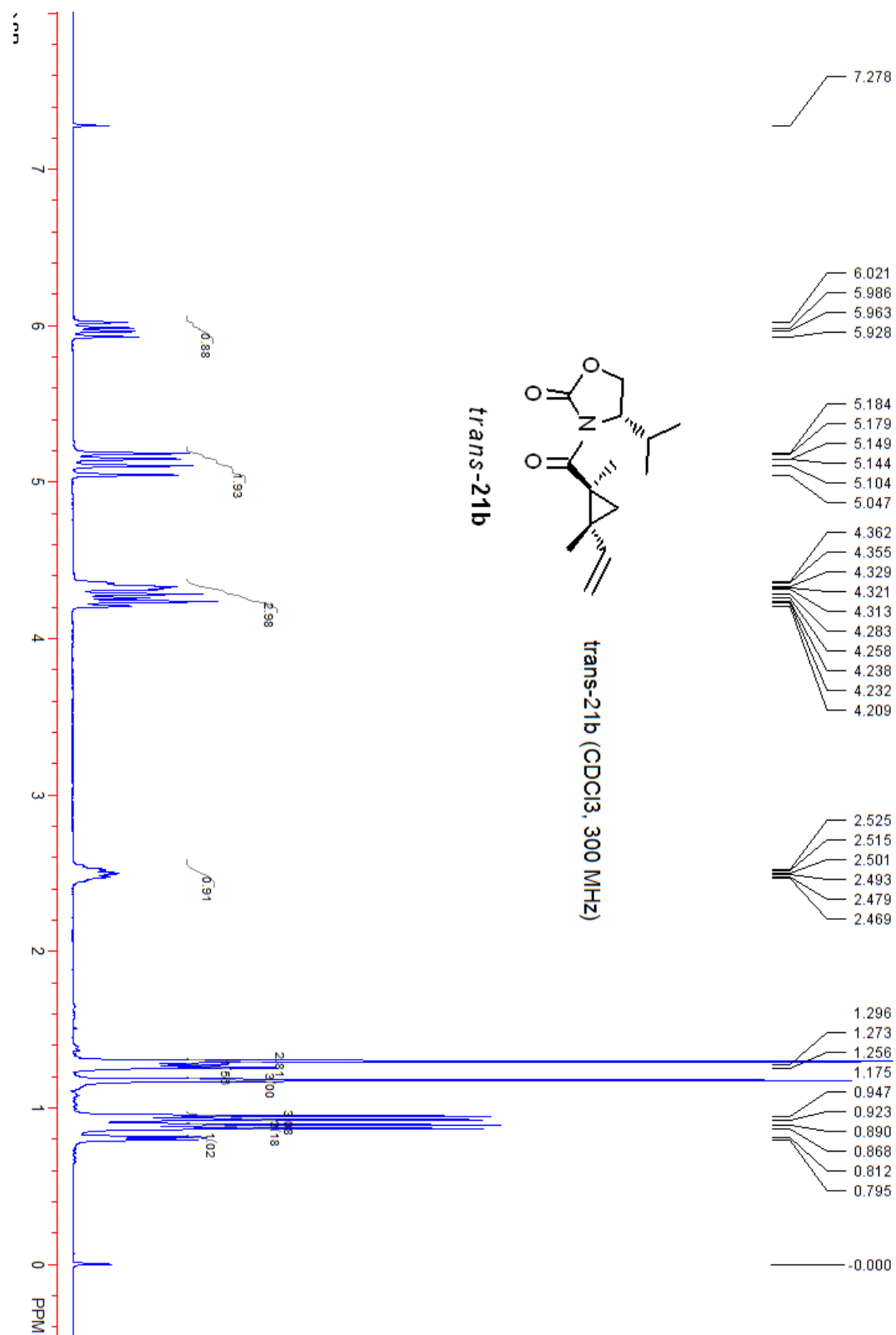


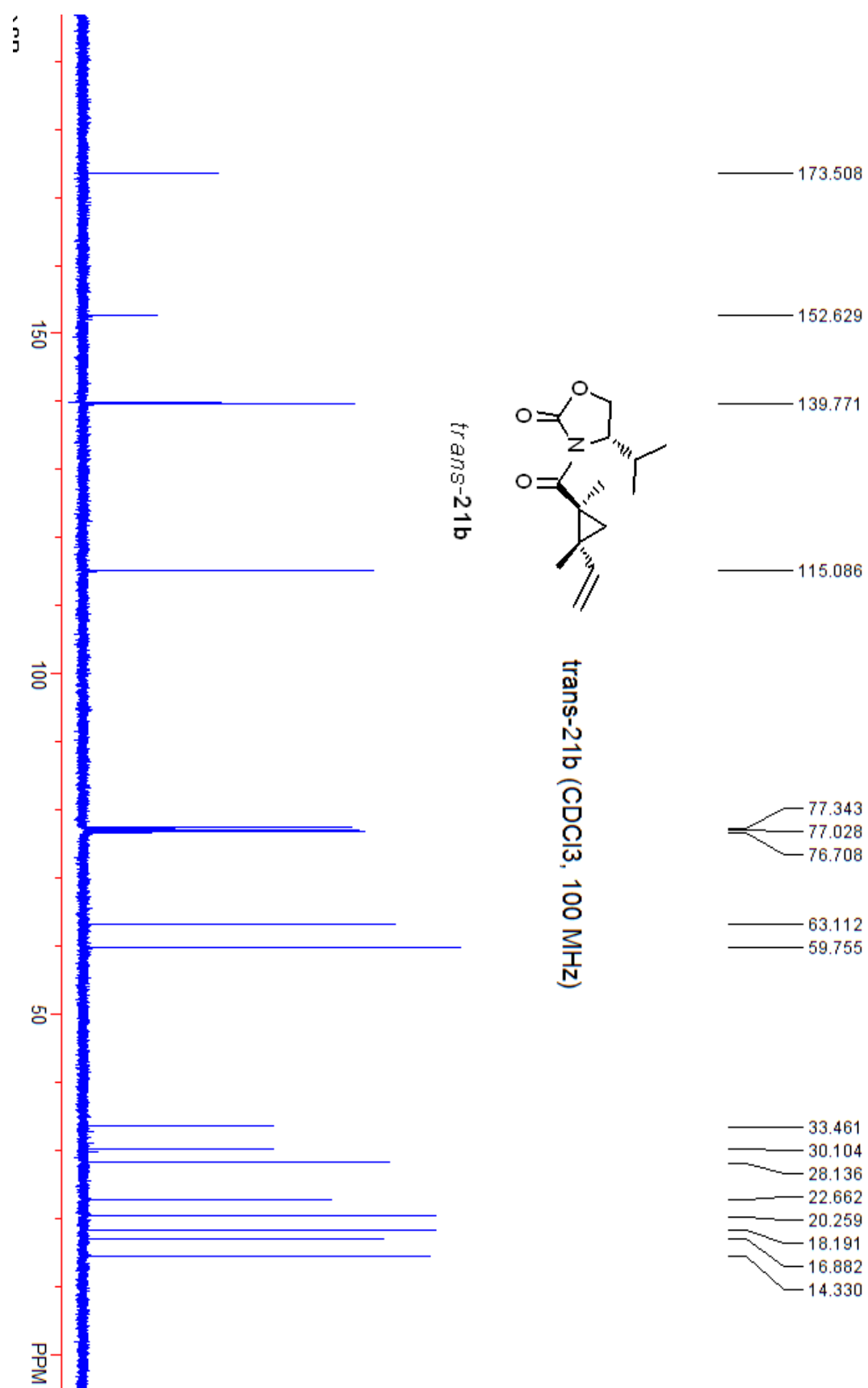


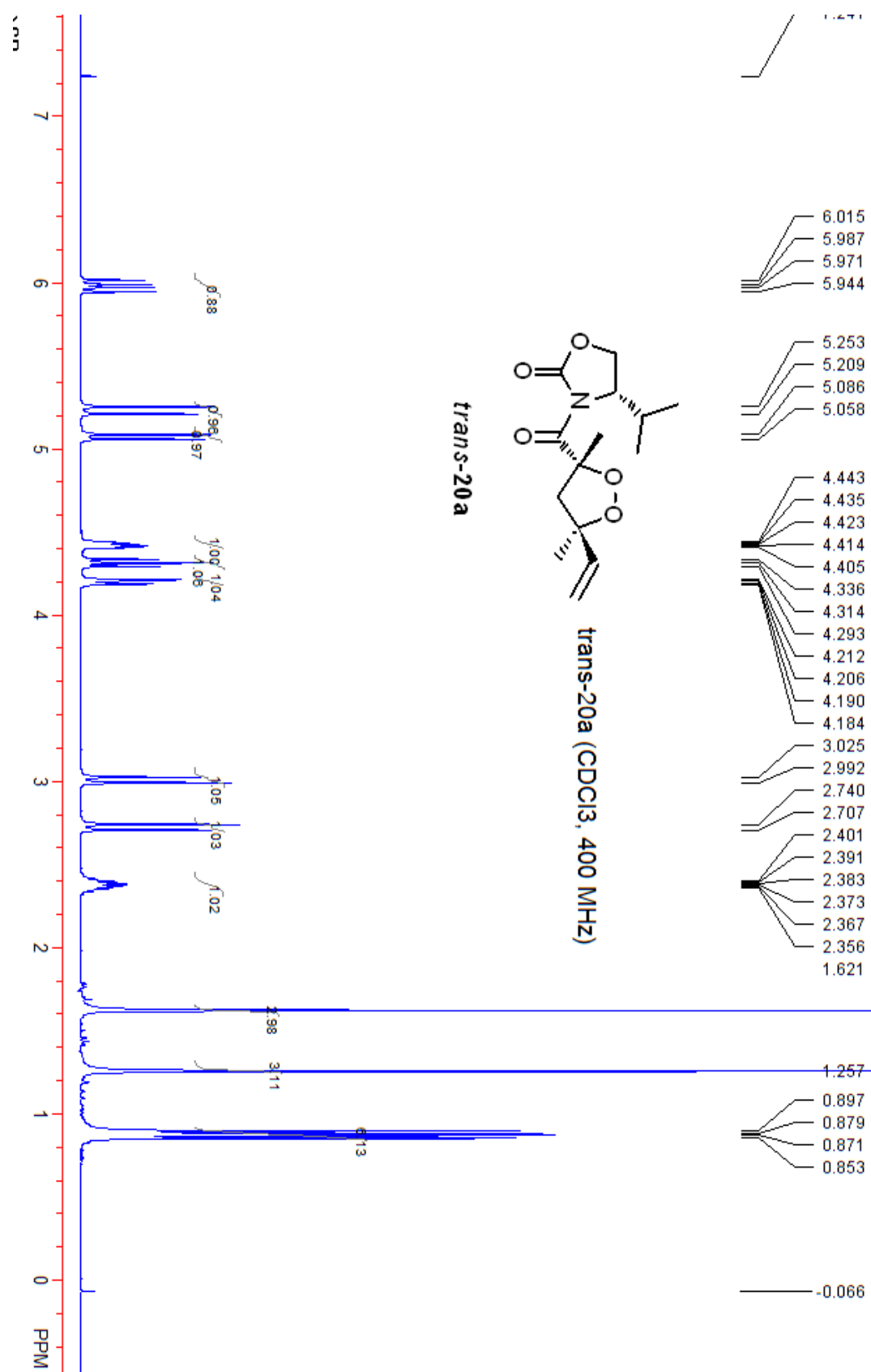


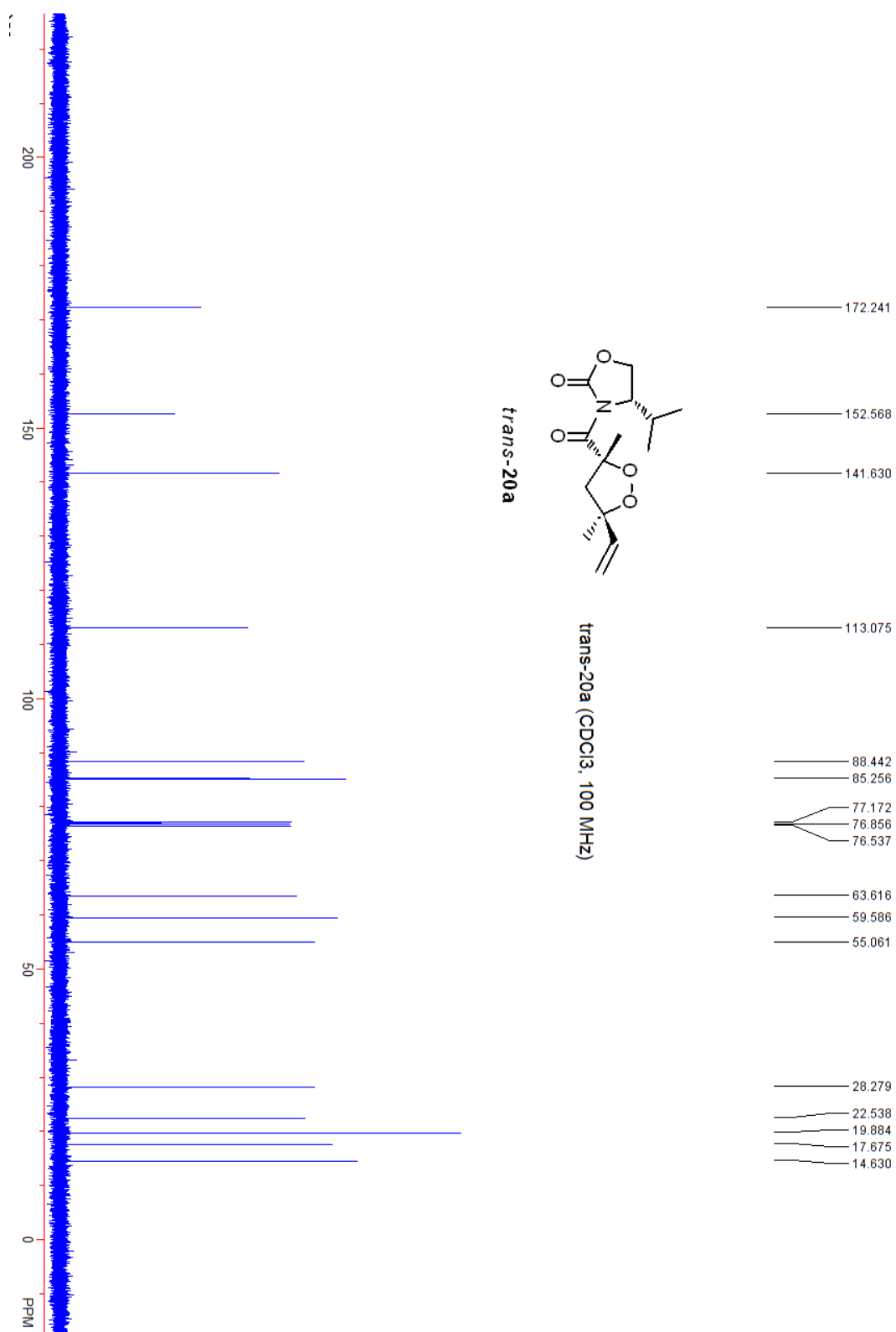


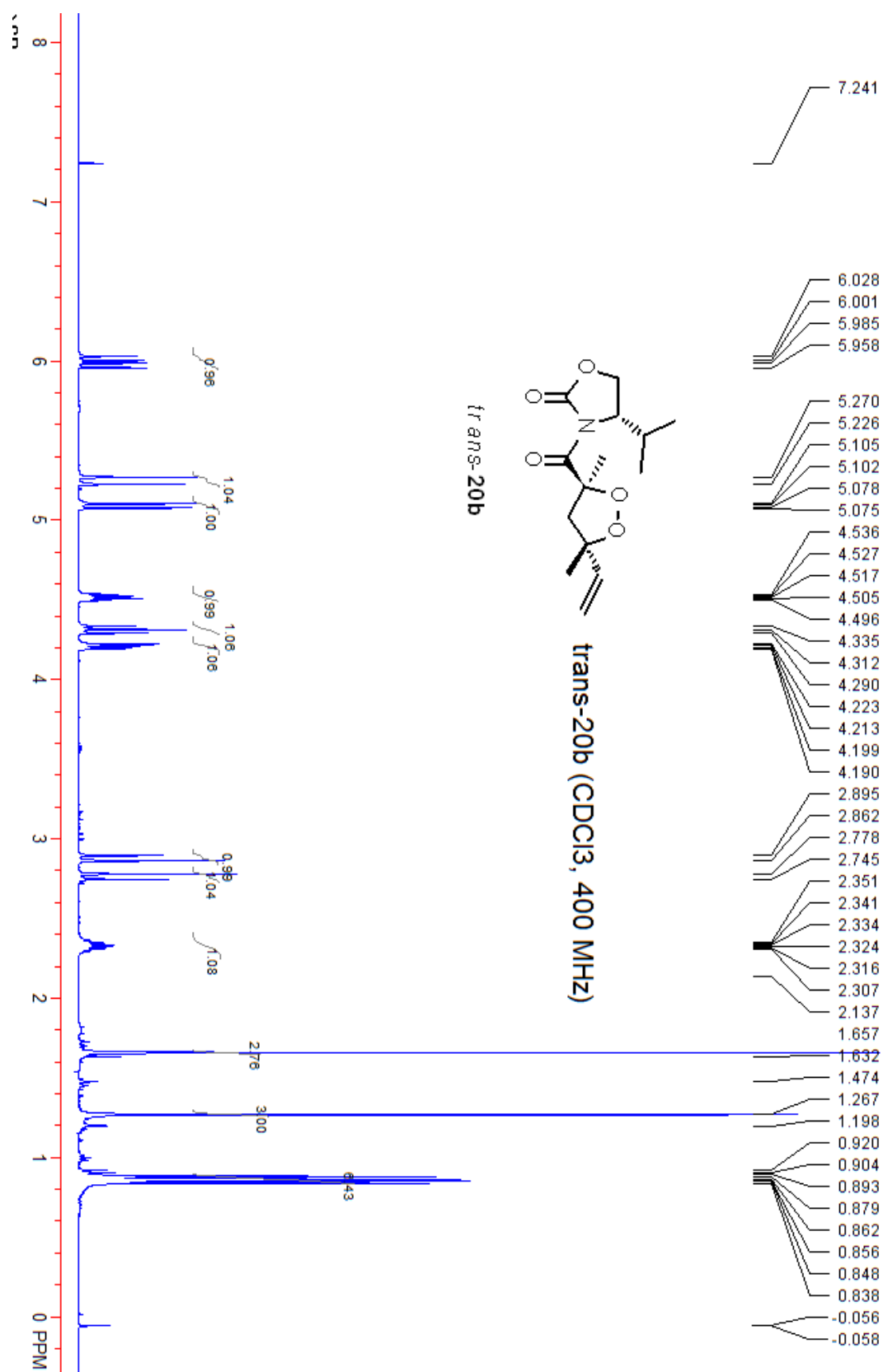


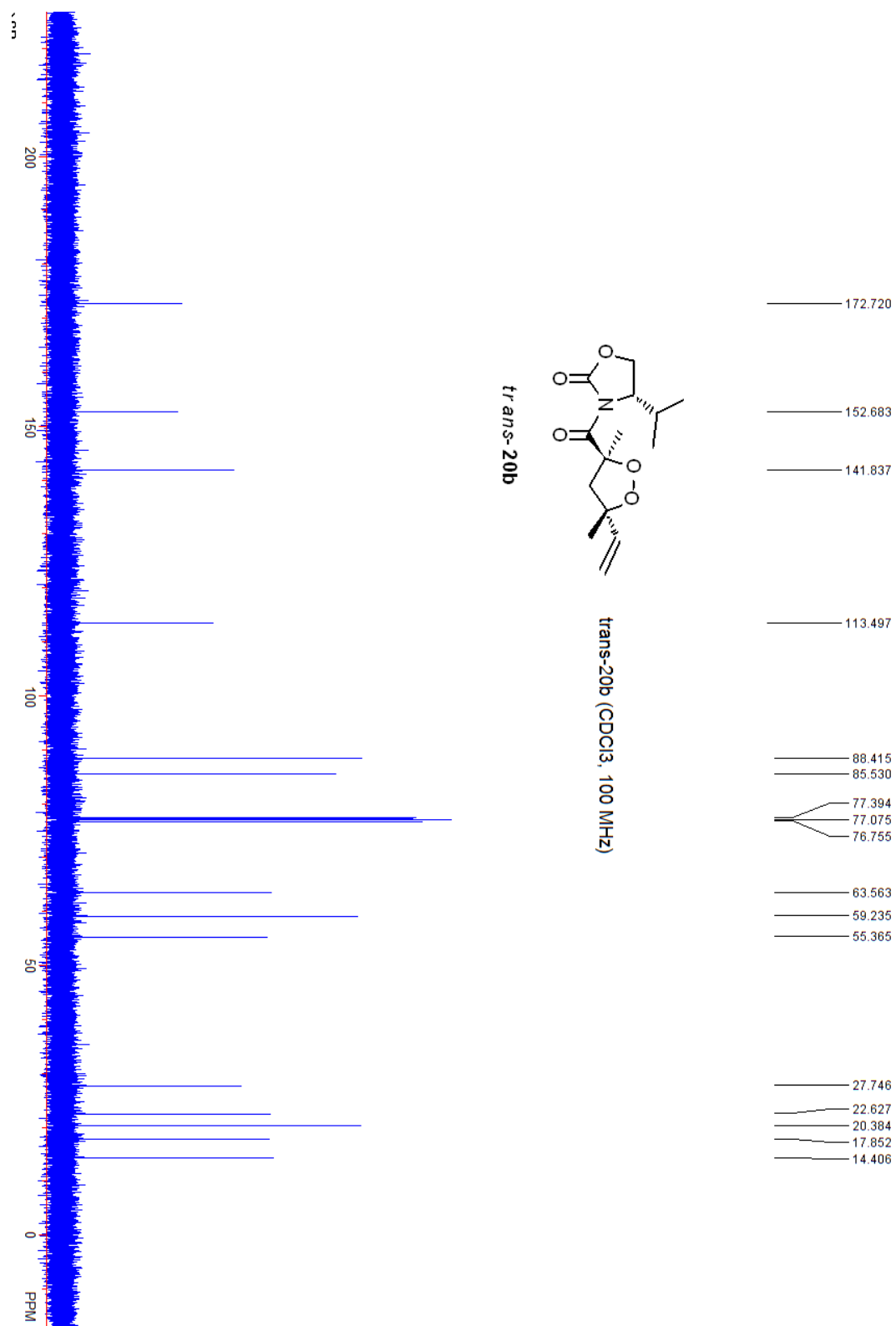


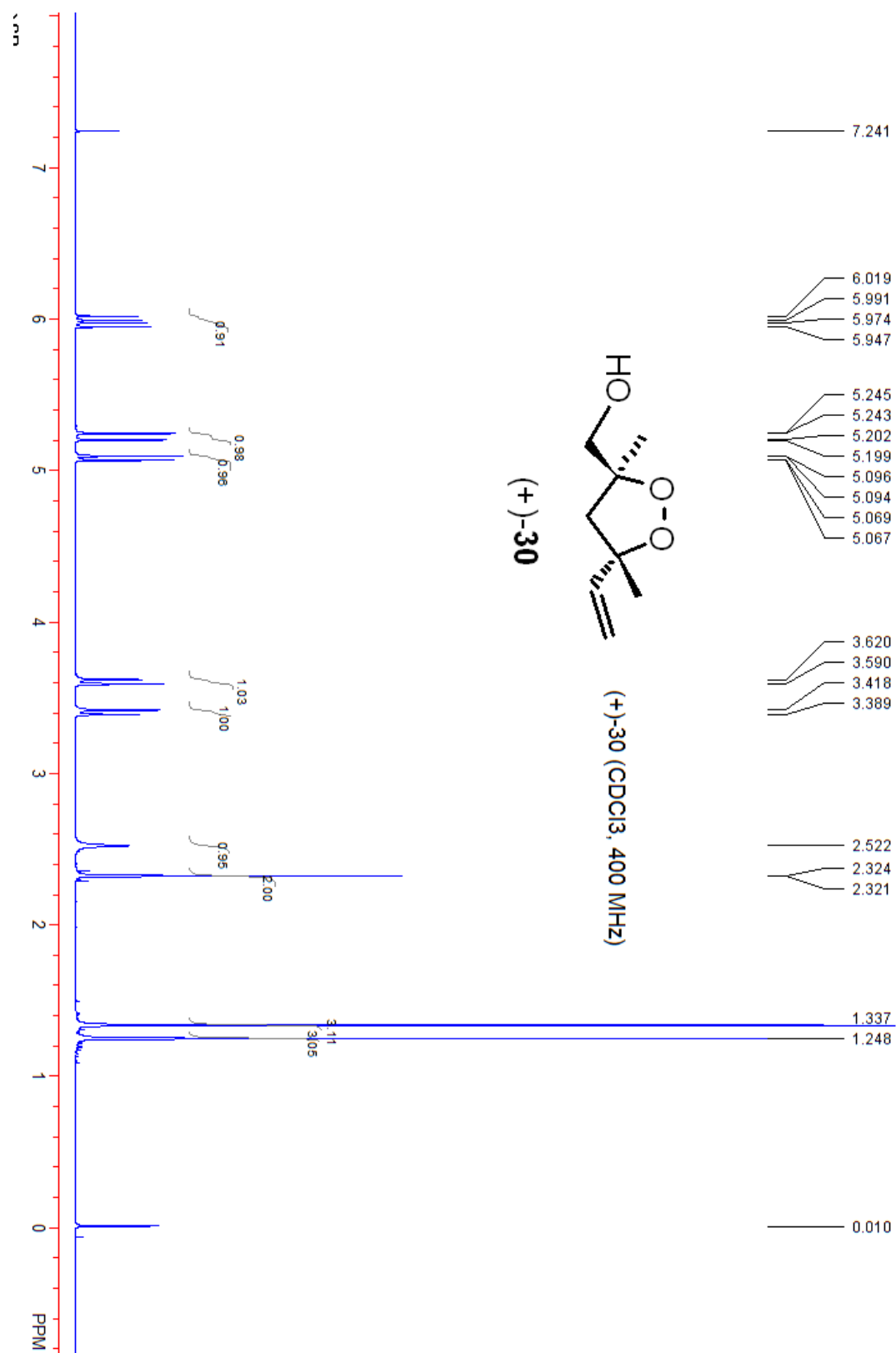


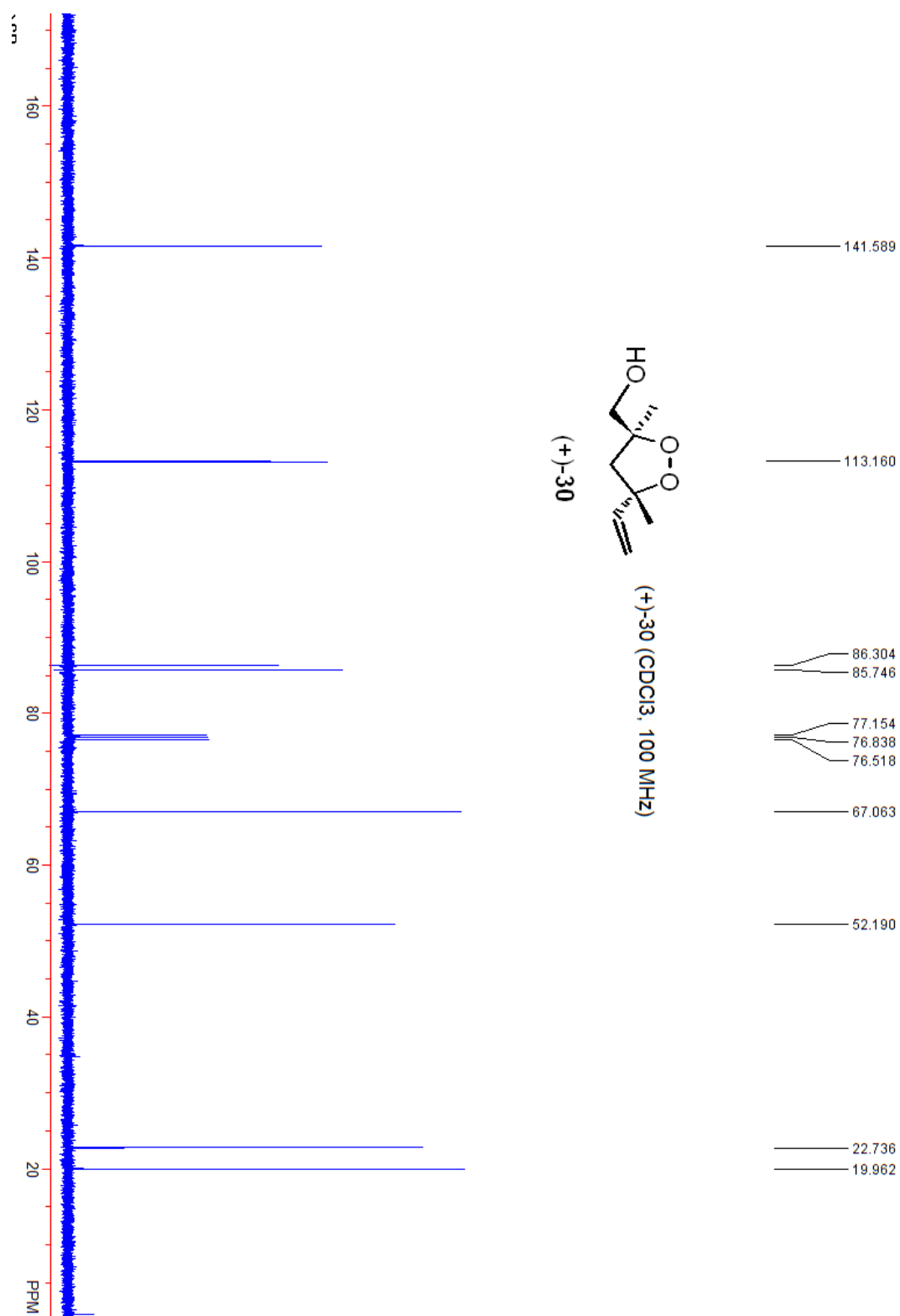


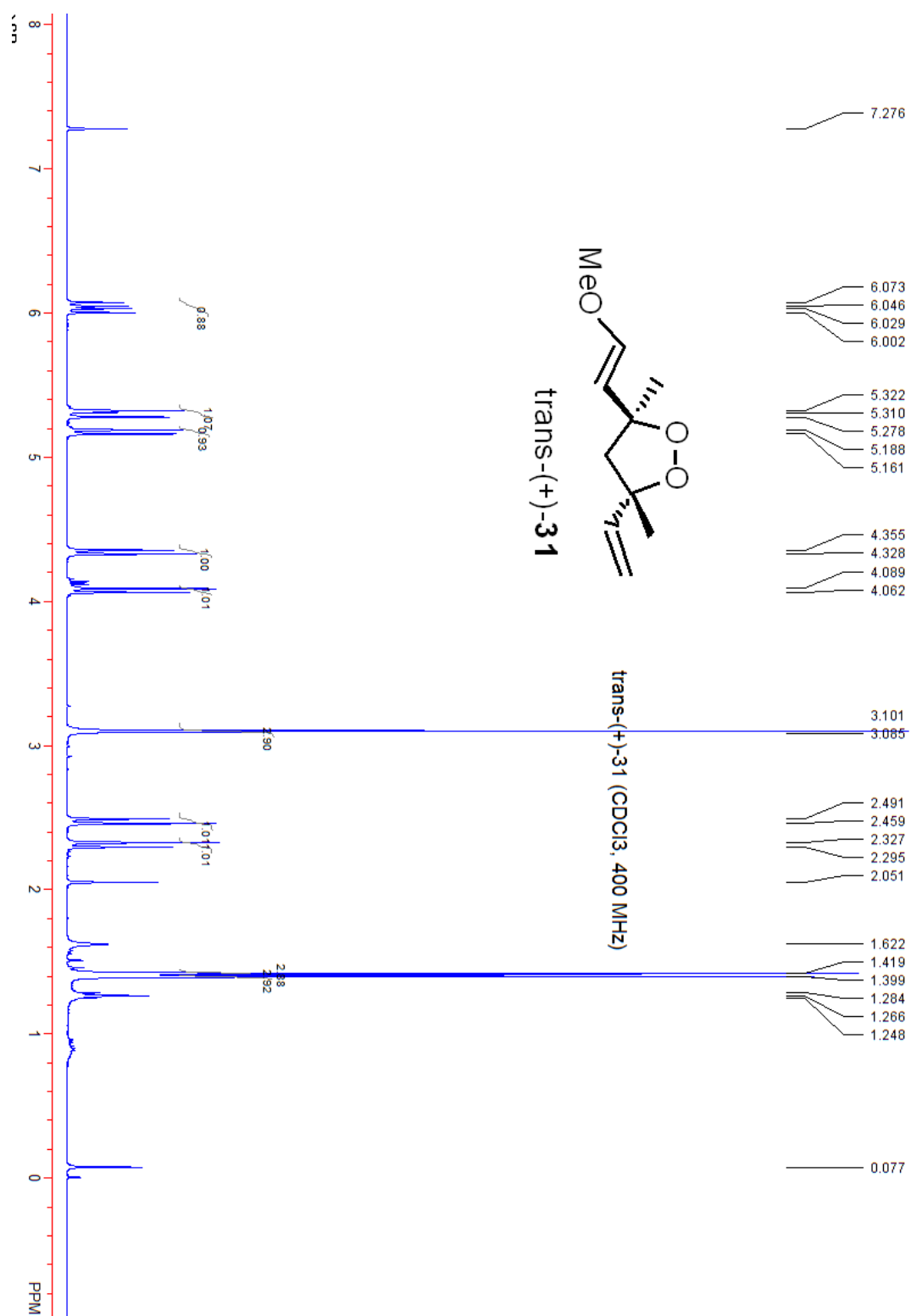


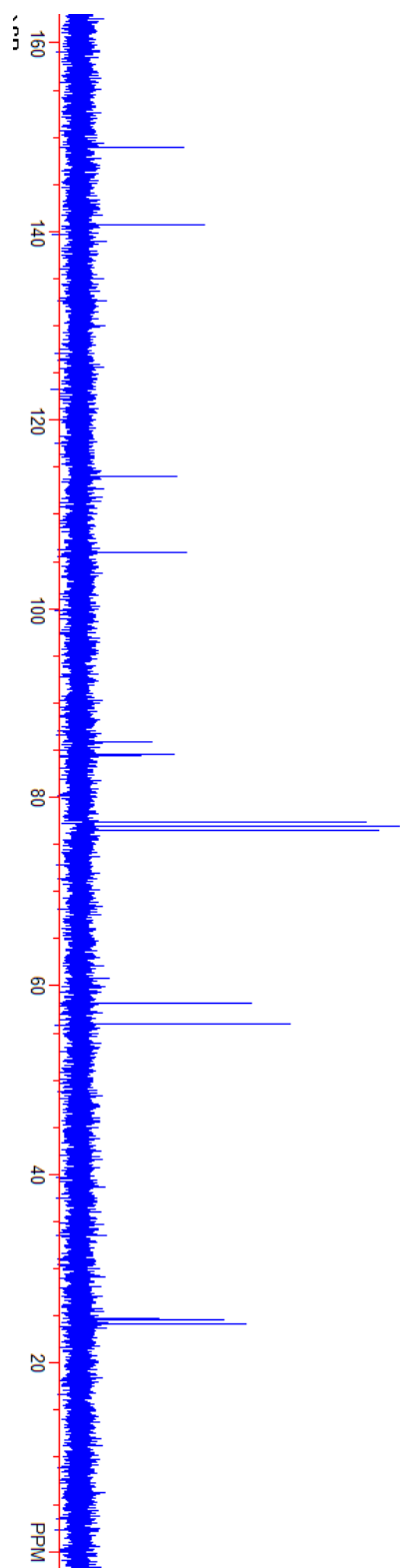
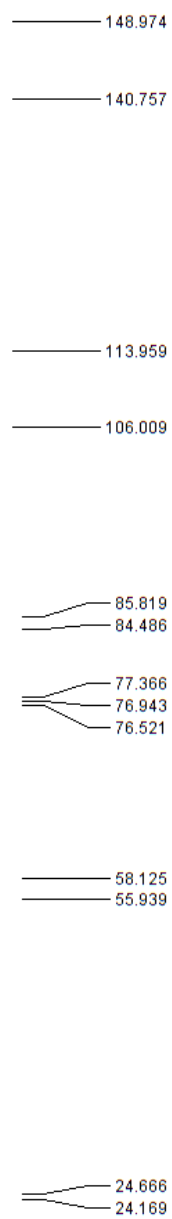
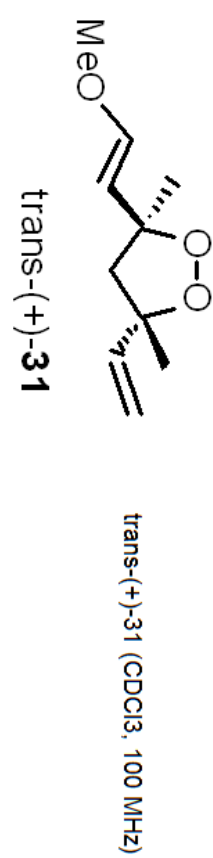


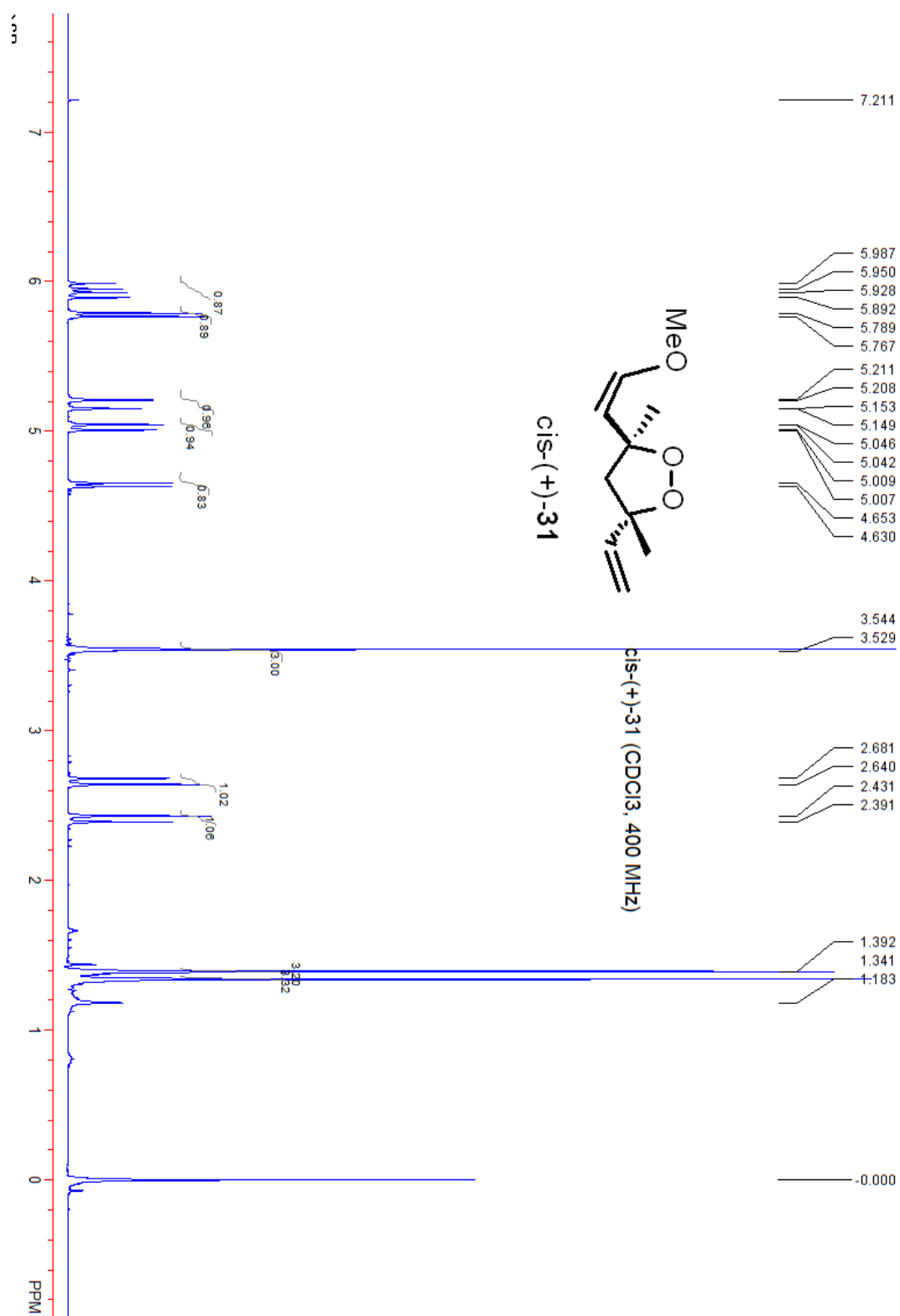












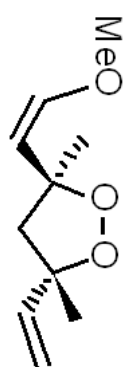
146.194
141.805

113.144
111.524

85.409
85.006
77.294
76.974
76.660

59.961
57.781

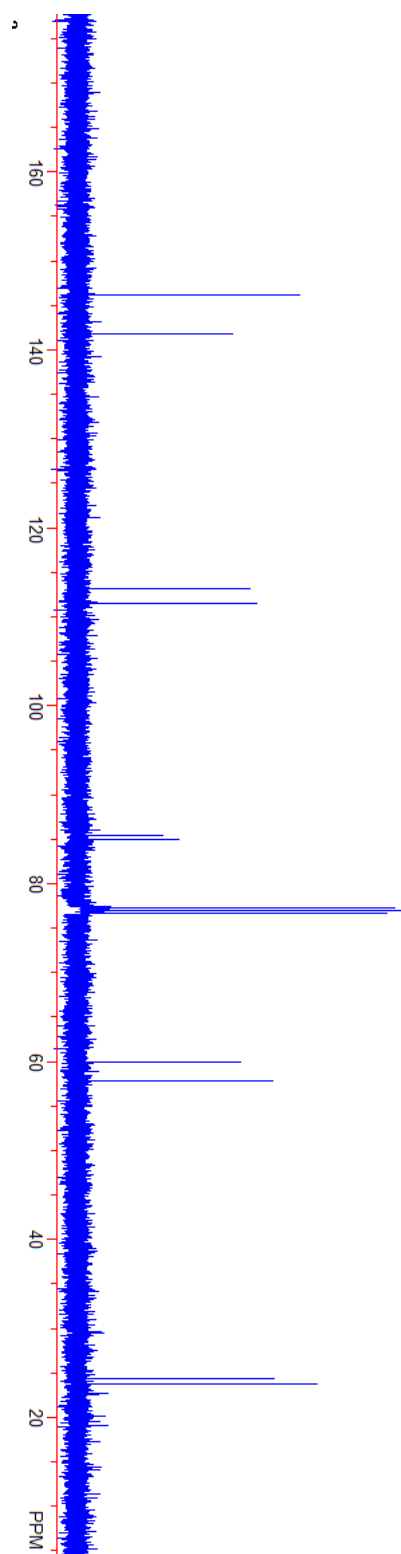
24.365
23.791

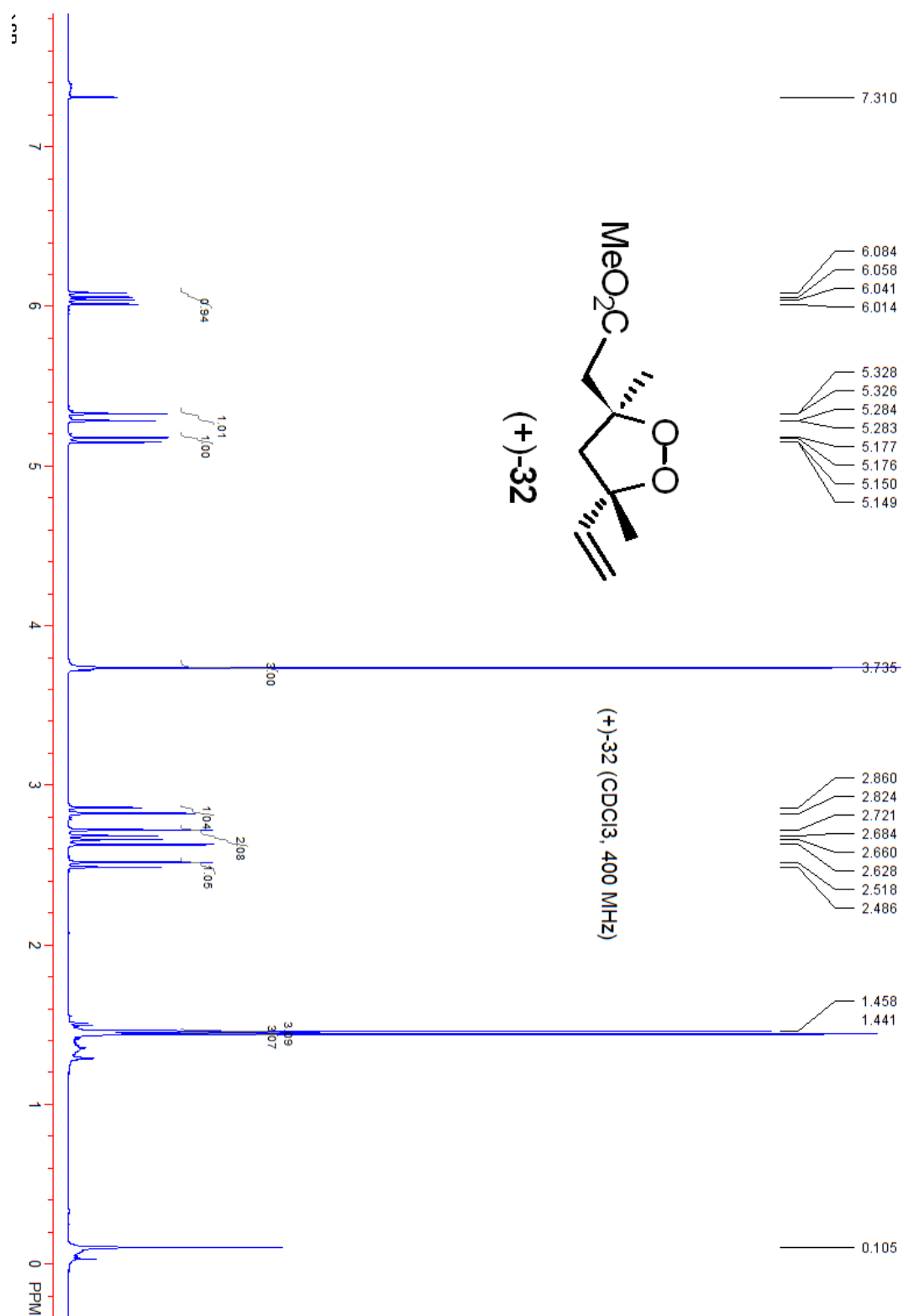


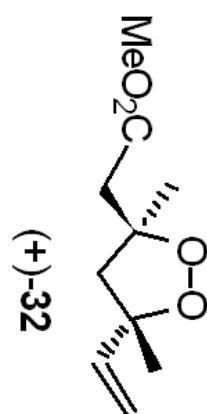
cis-(+)-**31**

cis-(+)-**31** (CDCl₃, 100 MHz)

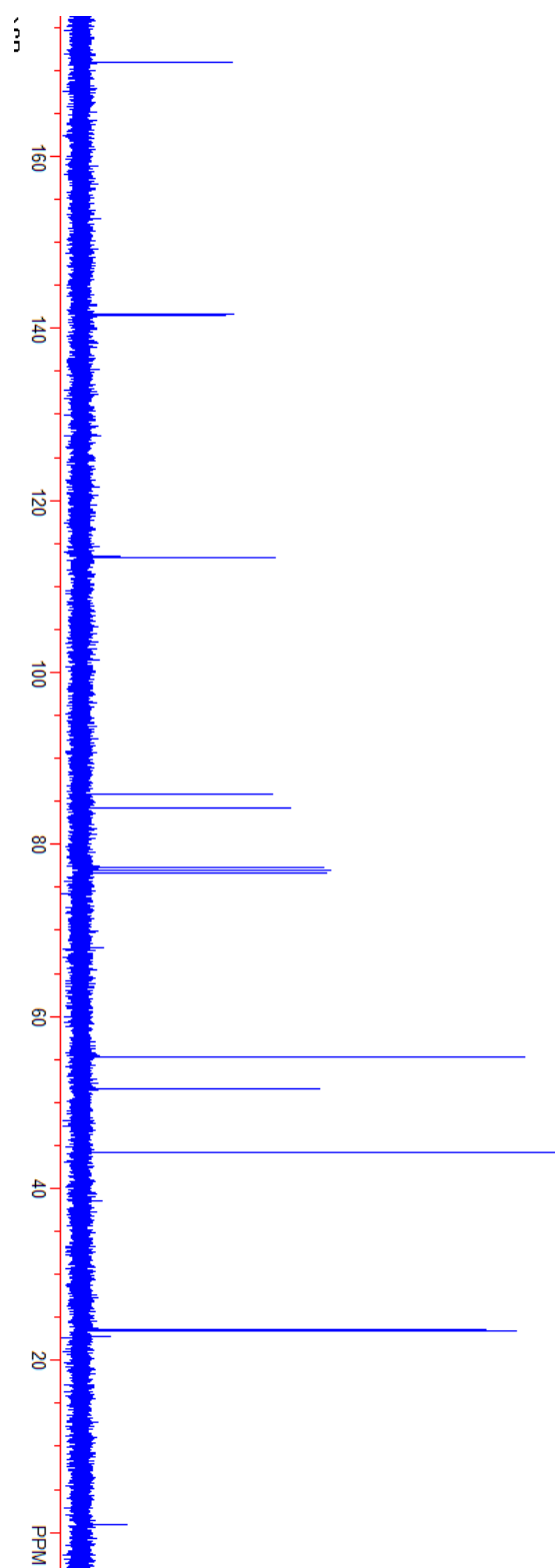
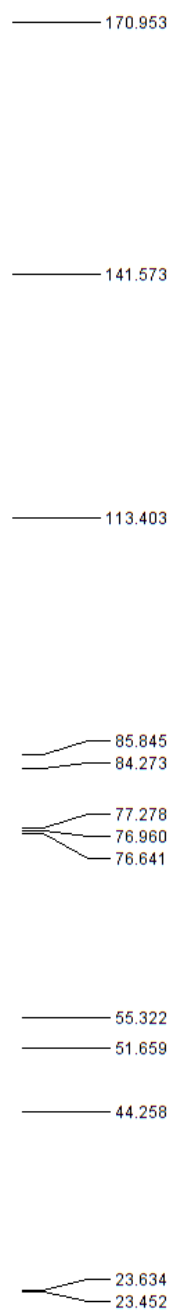
; blank line

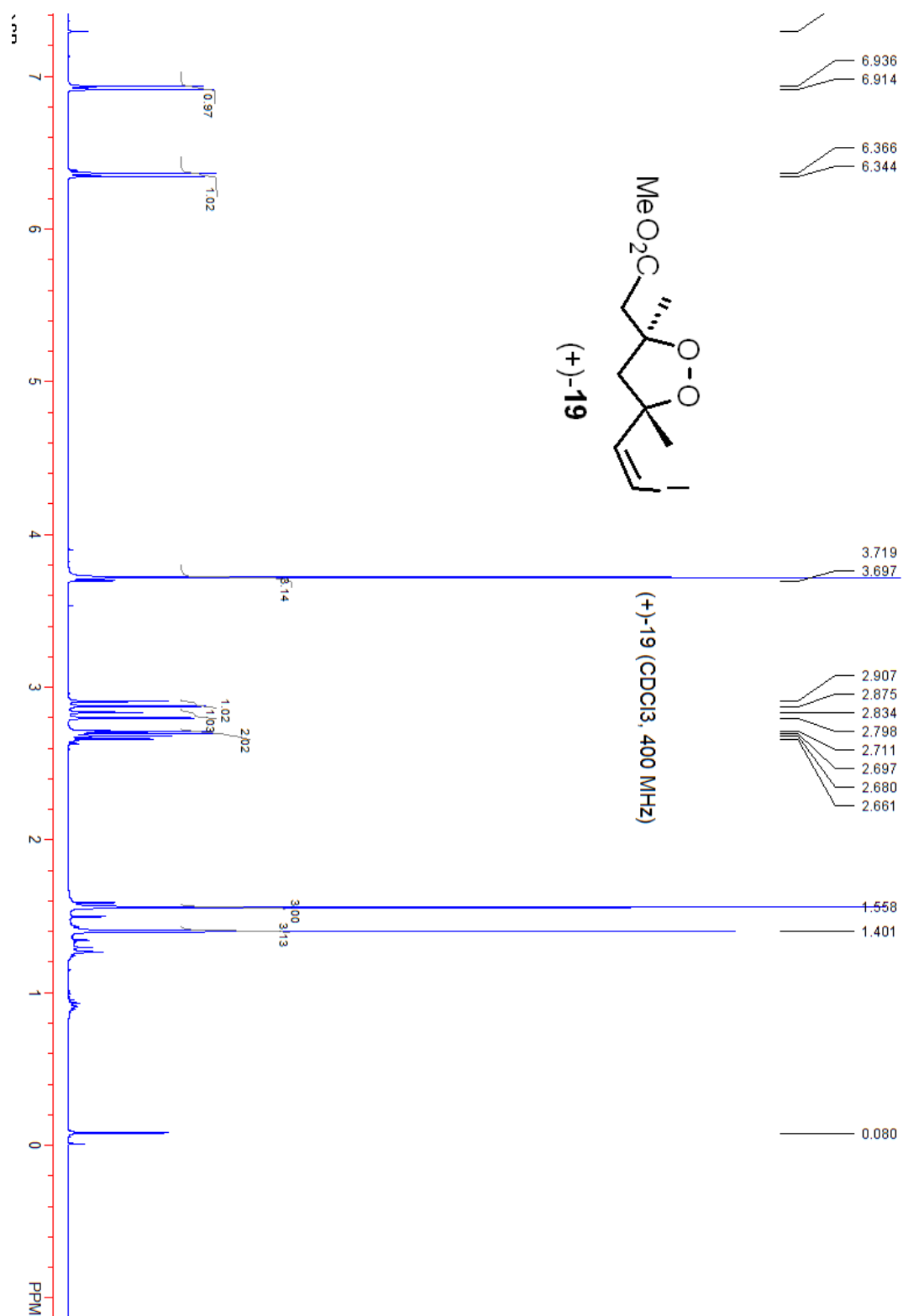


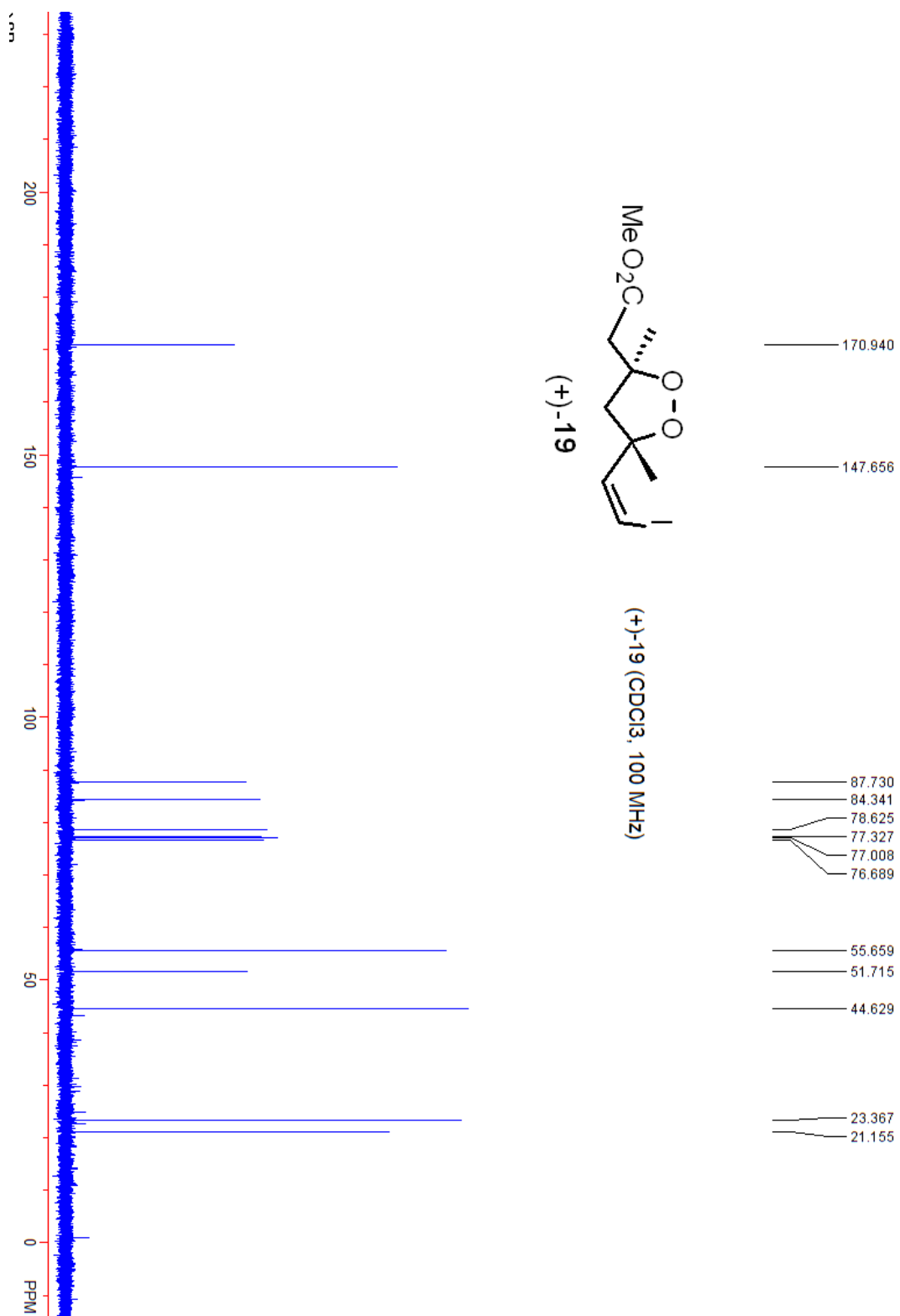


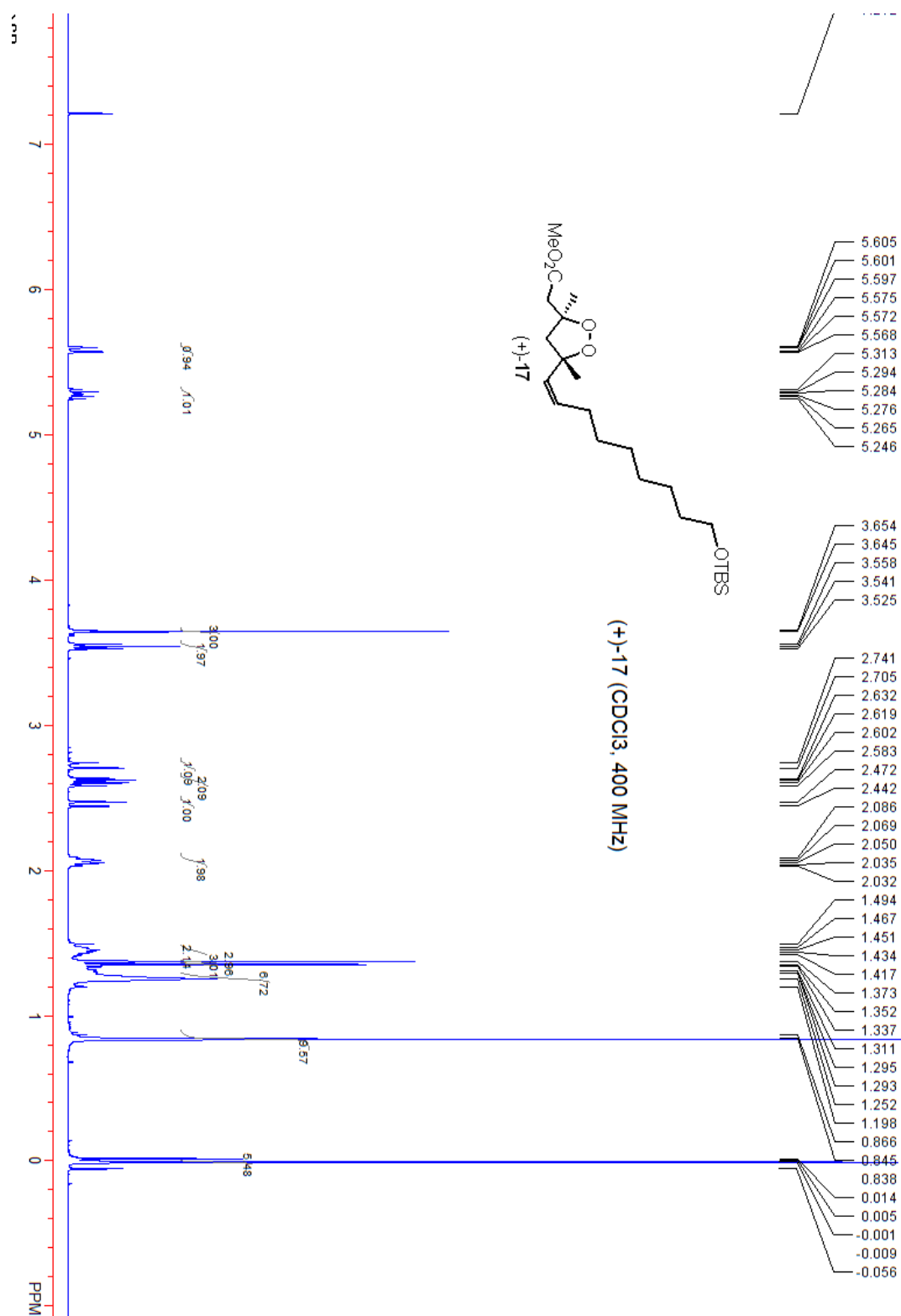


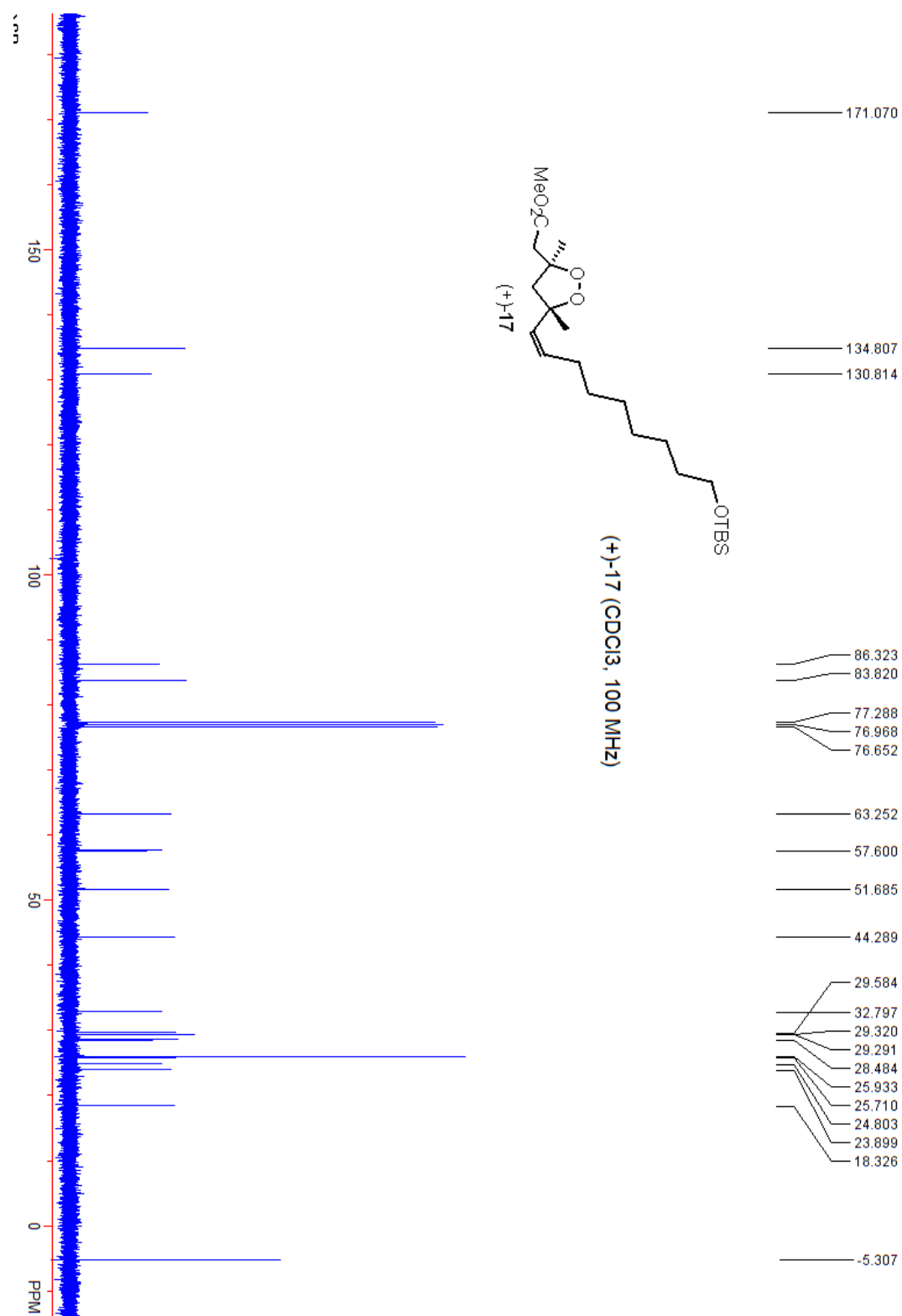
(+)-32 (CDCl₃, 100 MHz)

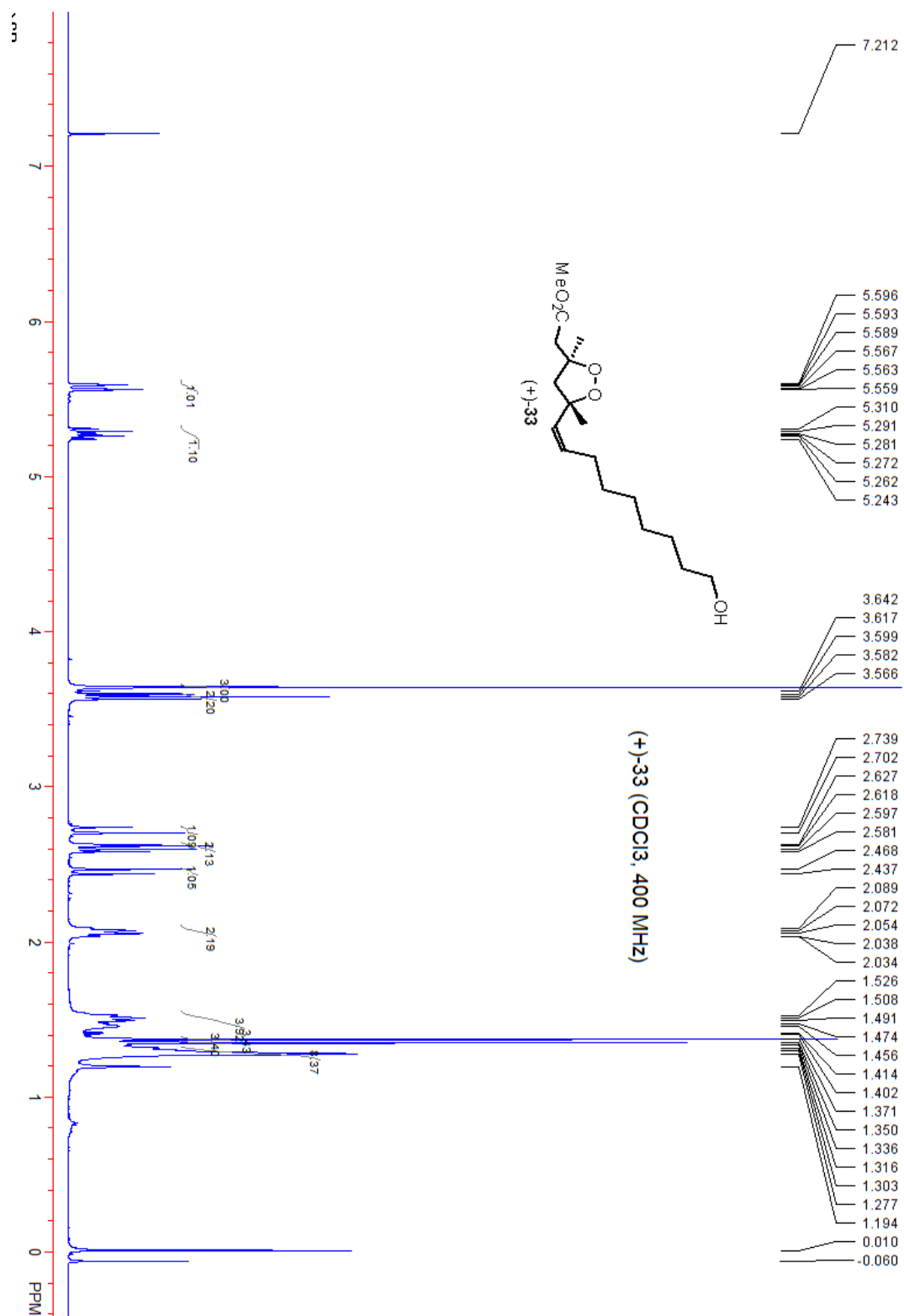


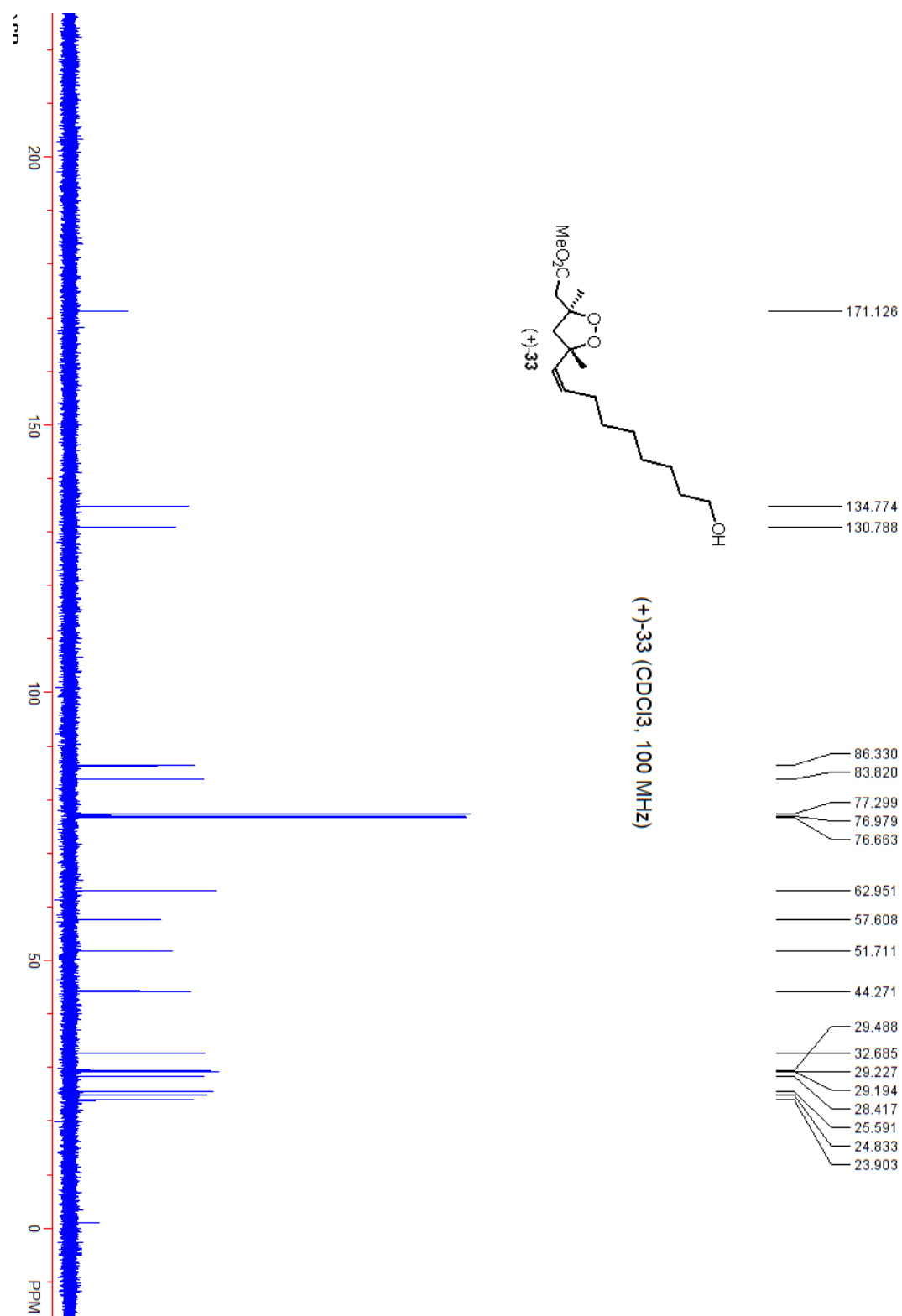


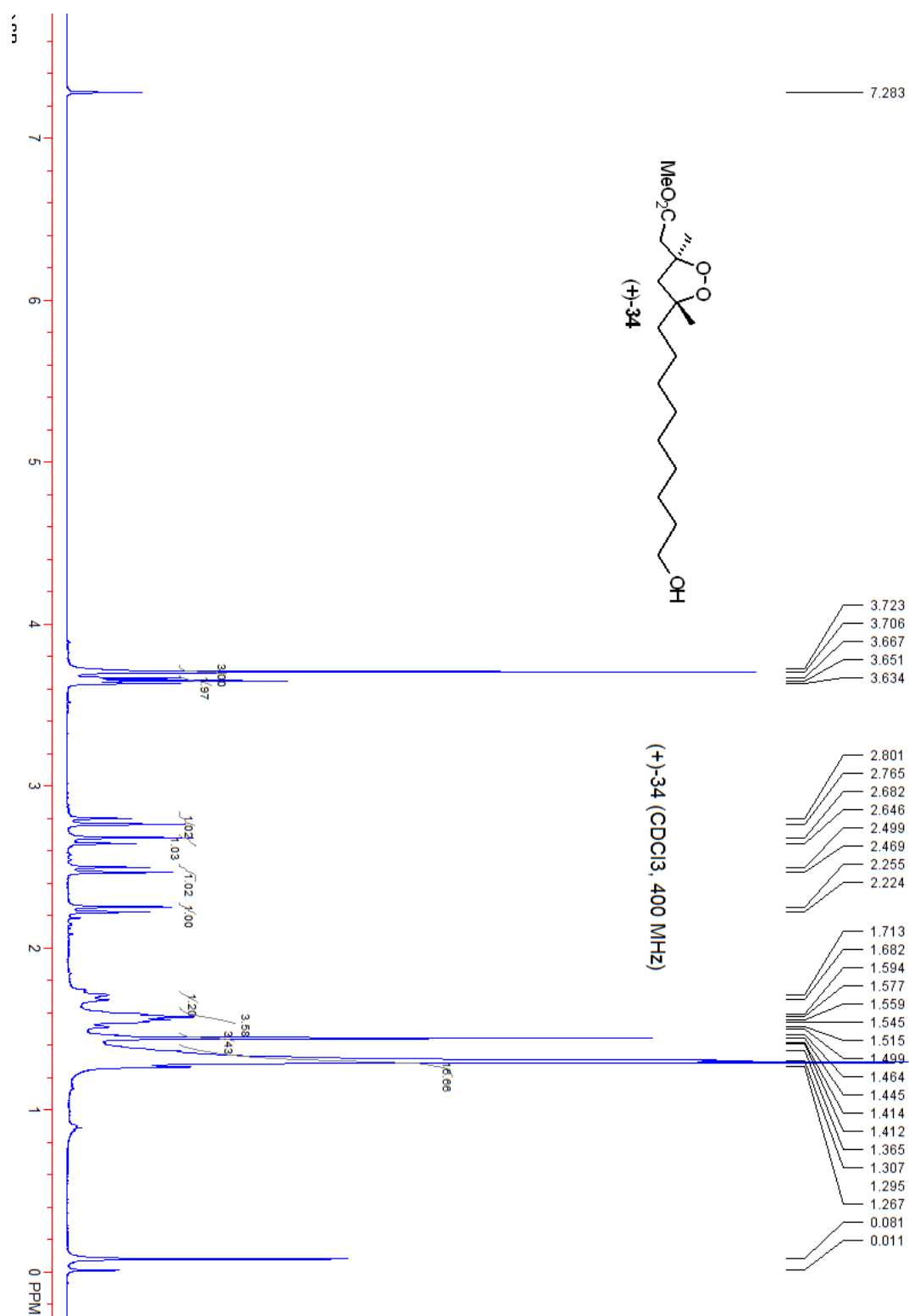


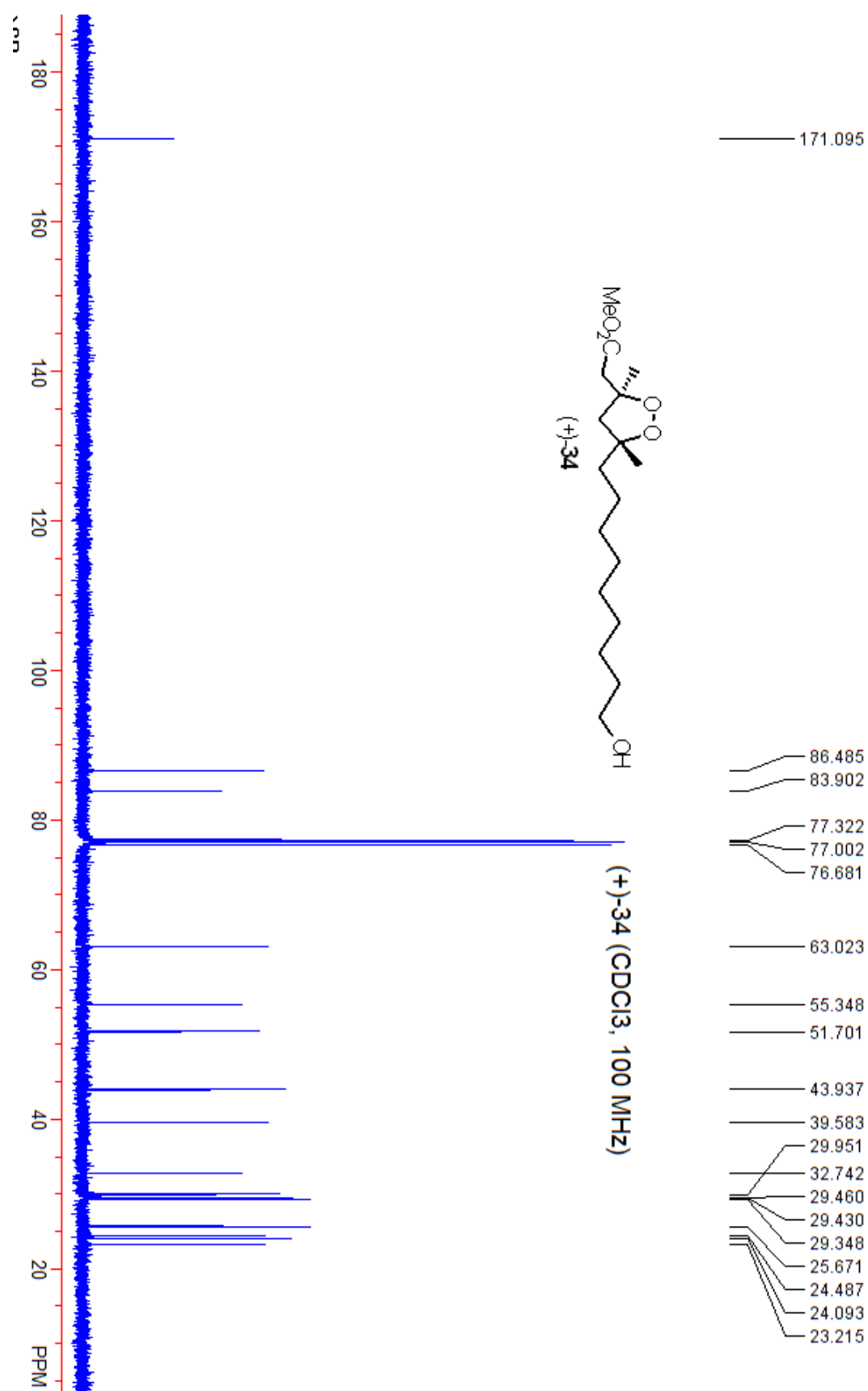


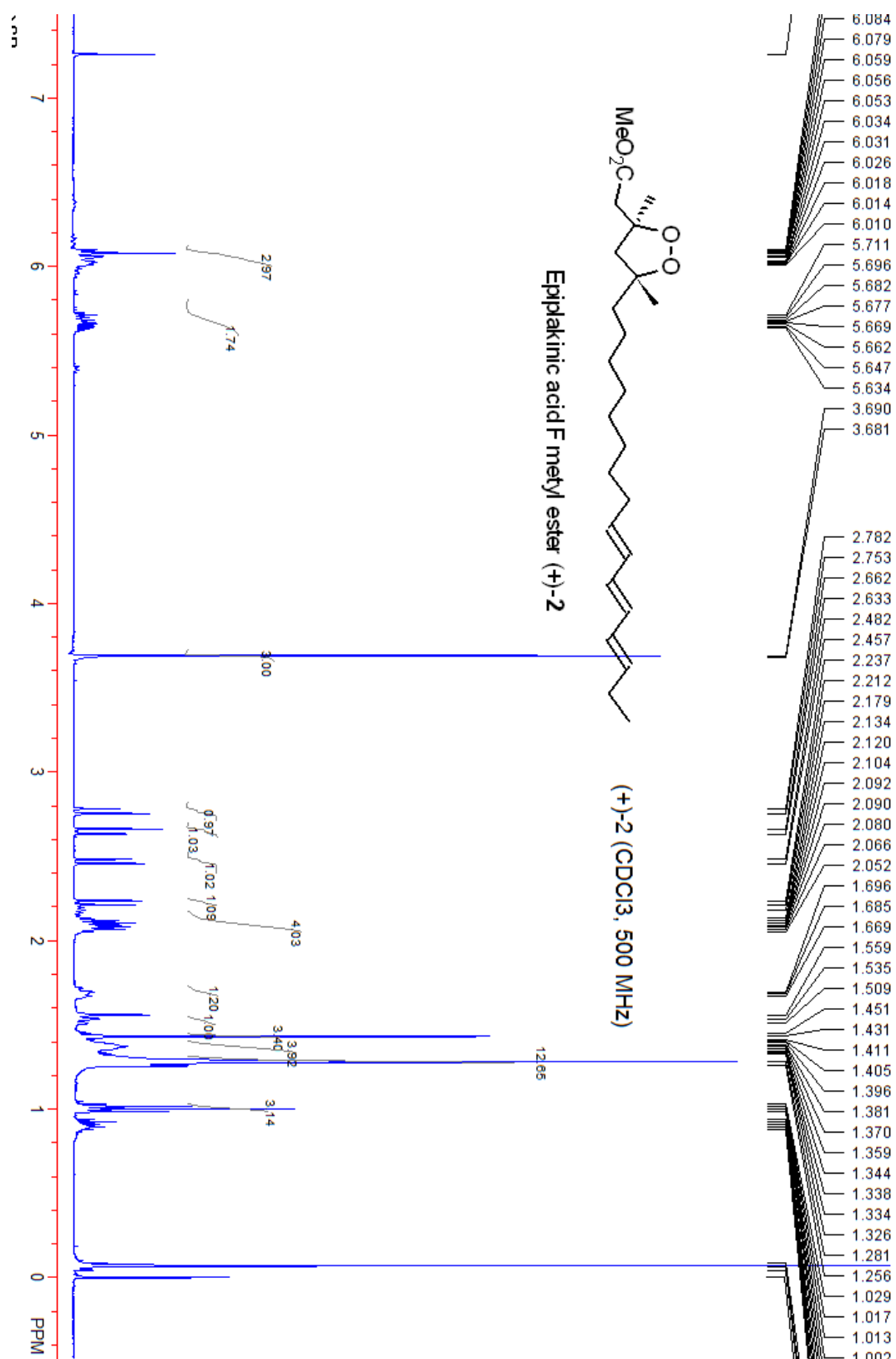


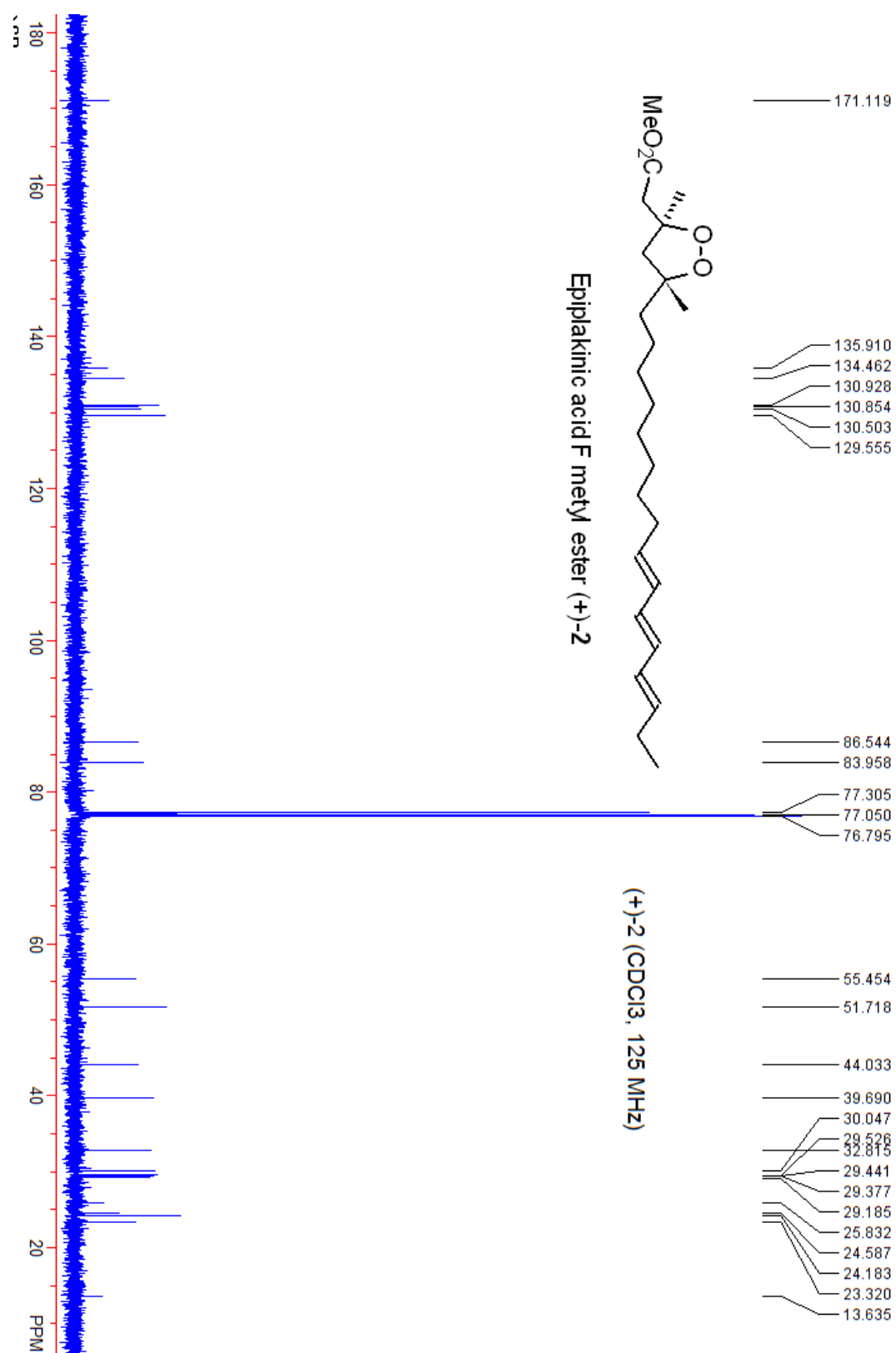


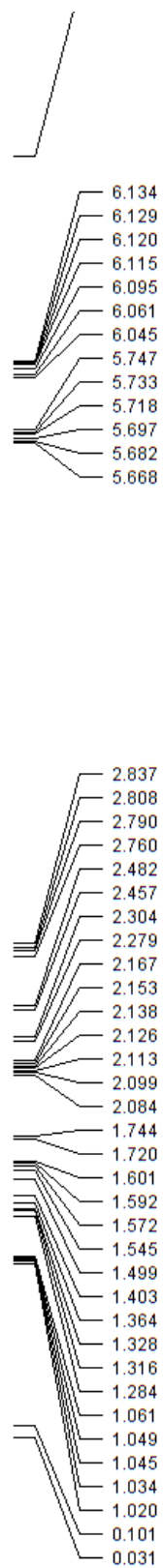




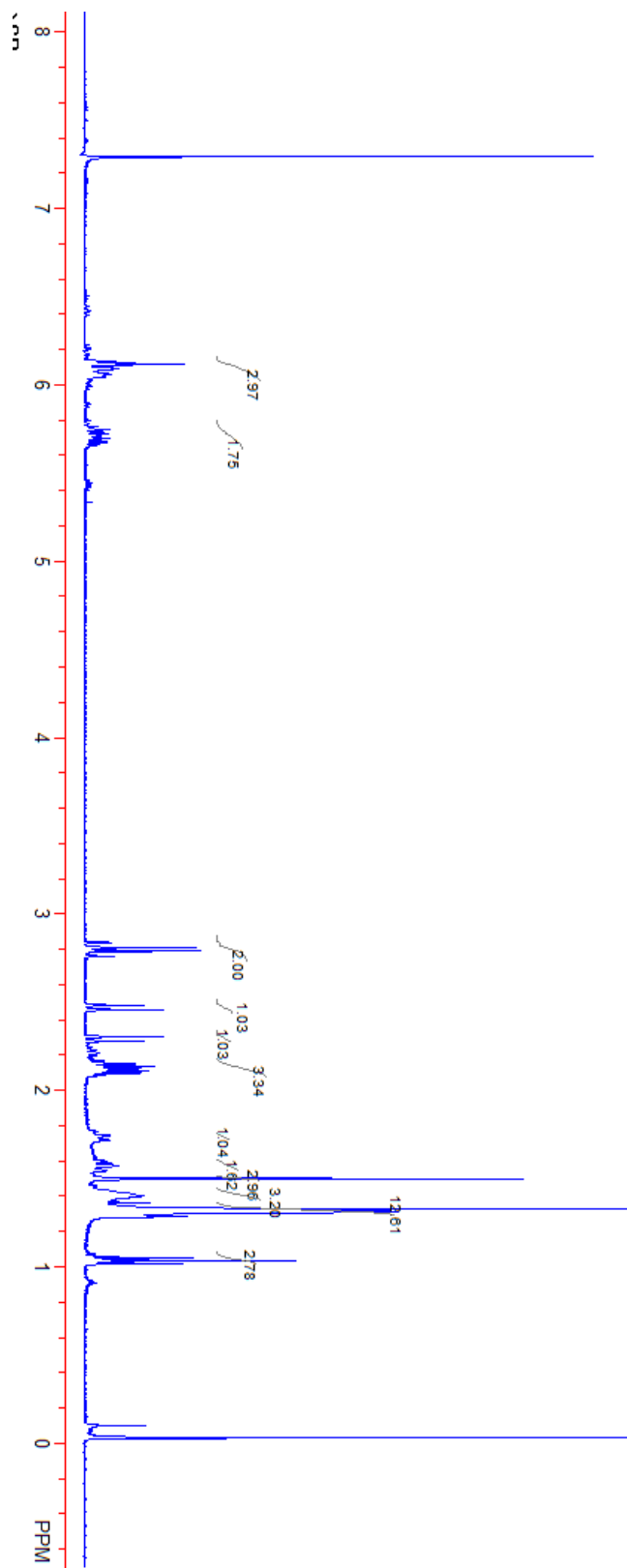


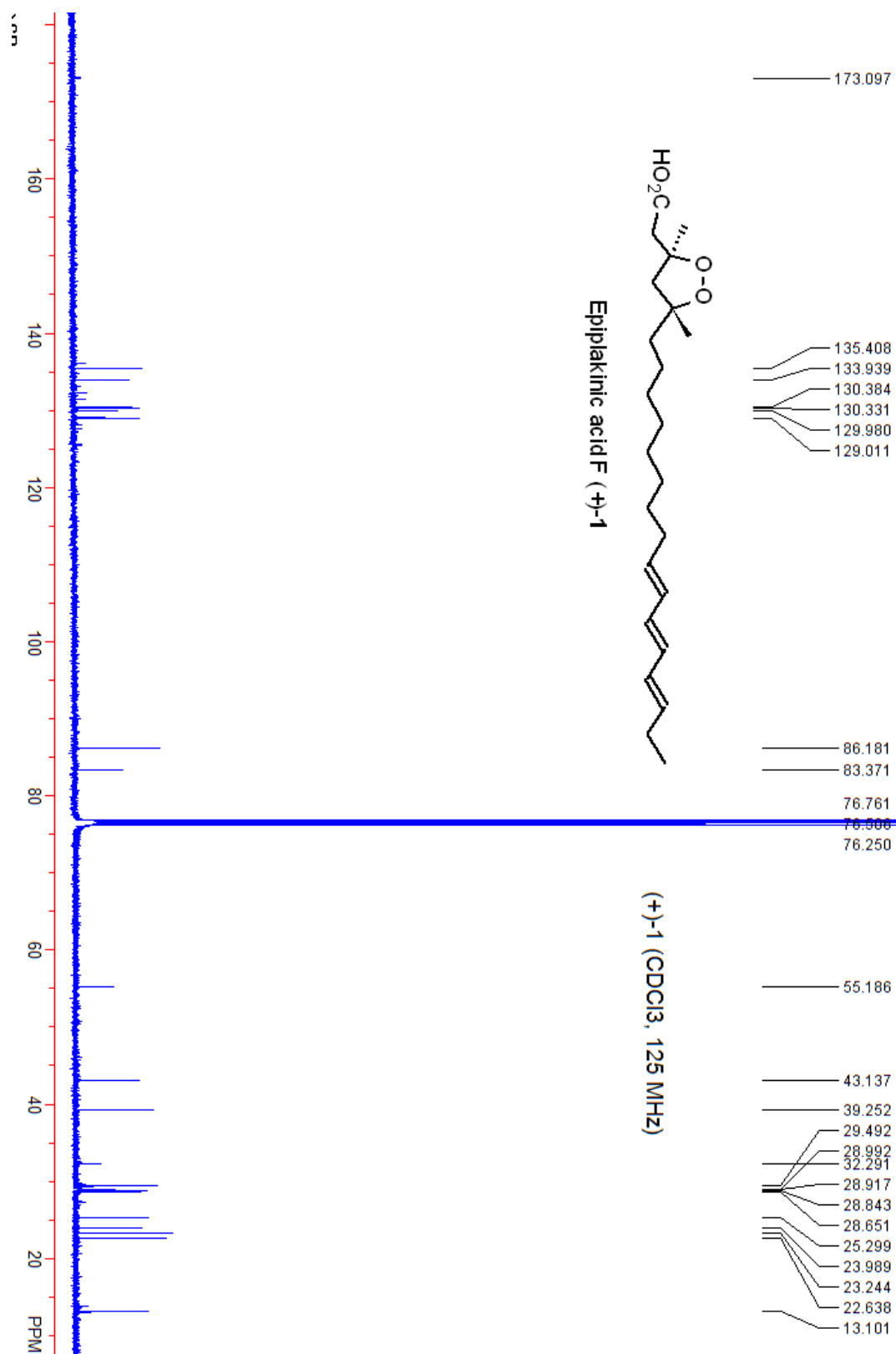






Epipakinic acid F (+)-1





HPLC Report

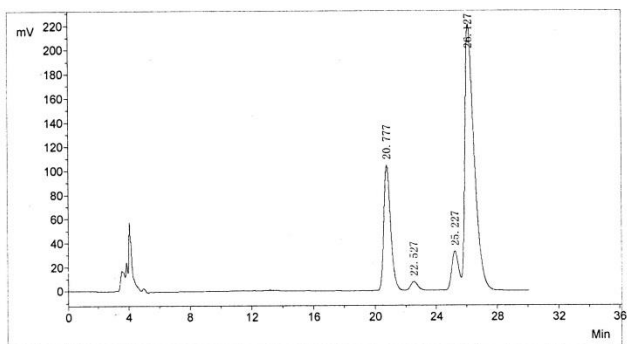
Sample Name:

Data File: TXY.che

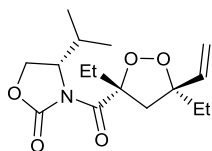
Operator:

Date: 2011-03-29

Time: 13:42

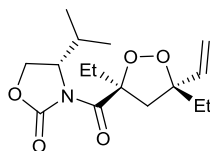


No.	PeakNo	R. Time	PeakHeight	PeakArea	PerCent
1	1	20.777	104082.9	3223432.0	23.4025
2	2	22.527	7302.6	242468.0	1.7603
3	3	25.227	32830.5	1018985.9	7.3980
4	4	26.127	220508.4	9288967.7	67.4391
Total			364724.6	13773853.5	100.0000



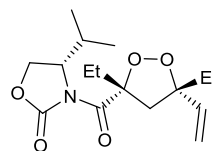
trans-12a

minor product



trans-13a

major product



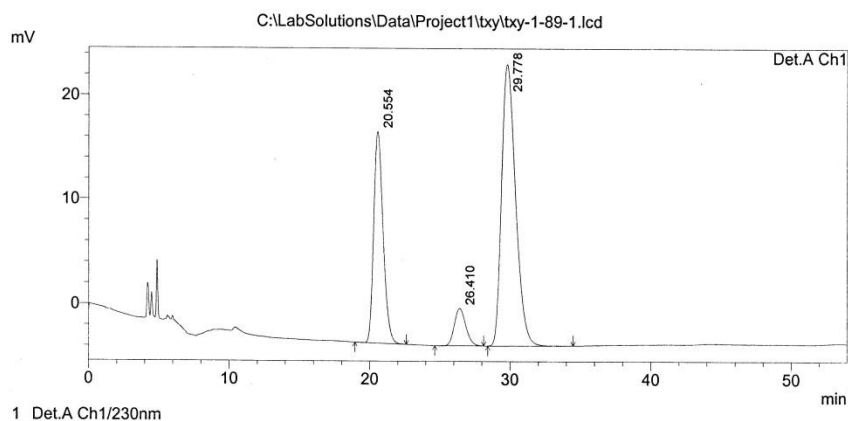
cis-14a

minor product

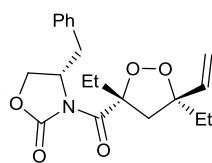
==== Shimadzu LCsolution Analysis Report =====

Acquired by : Admin
 Sample Name : txy
 Method : od-h,80/20,0.7,230
 Injection Volume : 1 uL
 Data File Name : txy-1-89-1.lcd
 Method File Name : 111.lcm
 Report File Name : Default.lcr
 Data Acquired : 2009-12-9 10:06:18
 Data Processed : 2009-12-9 11:00:20

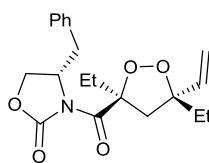
<Chromatogram>



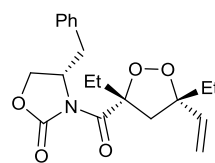
PeakTable					
Detector A Ch1 230nm					
Peak#	Ret. Time	Area	Height	Area %	Height %
1	20.554	942934	20277	31.312	39.887
2	26.410	202276	3589	6.717	7.059
3	29.778	1866222	26971	61.971	53.054
Total		3011433	50837	100.000	100.000



trans-12b
minor product



trans-13b
major product



cis-14b
minor product

C:\LabSolutions\Data\Project1\txy\txy-1-89-1.lcd

HPLC REPORT

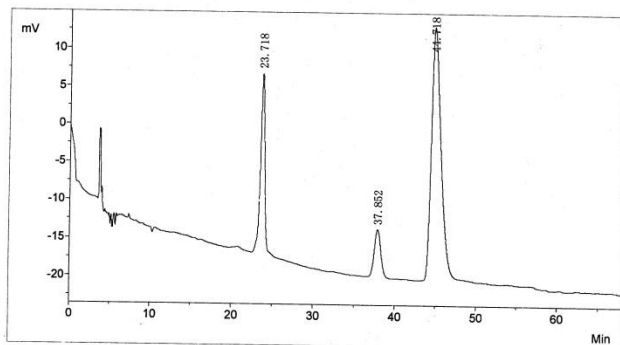
Data File: TXY-2-32-2+- IC 82 214 0.7. che

Sample name:

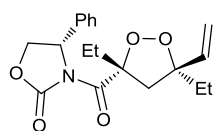
Date: 2010-01-05

Time : 14:15

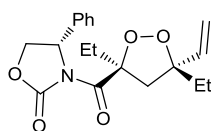
Operator:



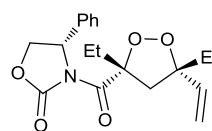
No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		23.718	23875.2	874980.2	22.3477
2	2		37.852	6279.2	347357.8	8.8718
3	3		44.718	33244.7	2692962.9	68.7805
Total				63399.1	3915300.9	100.0000



trans-12c
minor product



trans-13c
major product

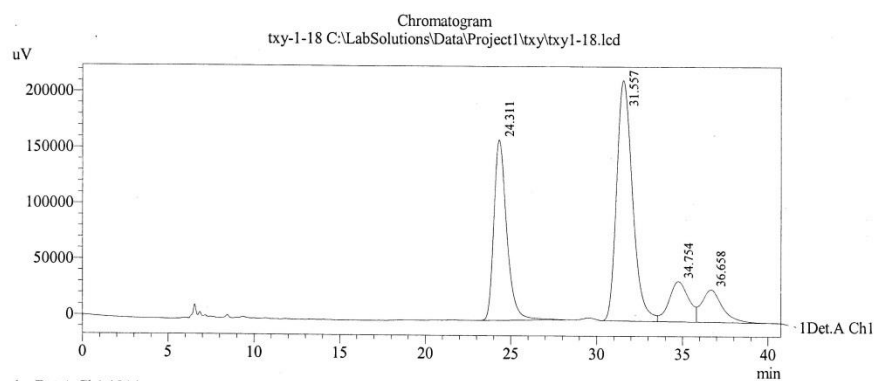


cis-14c
minor product

==== Shimadzu Lcsolution Calibration Curve ====

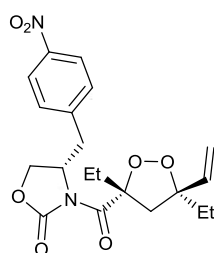
Sample Information

Acquired by : Admin
 Sample Name : ~~txy-1-18~~
 Sample ID : ic,50/50,0.5,214
 Vial# :
 Injection Volume : 1 uL
 Data Filename : txy1-18.lcd
 Method Filename : 111.lcm
 Batch Filename :
 Report Filename : Default.lcr

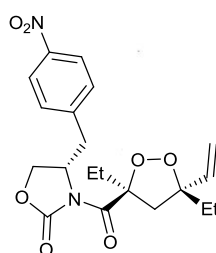


PeakTable

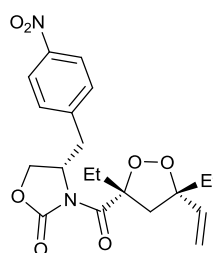
Peak#	Ret. Time	Area	Height	Area %	Height %
1	24.311	8438734	161197	30.550	36.549
2	31.557	13843914	214881	50.118	48.722
3	34.754	2870220	35925	10.391	8.145
4	36.658	2469886	29036	8.941	6.584
Total		27622754	441039	100.000	100.000



trans-12d
minor product



trans-13d
major product



cis-14d
minor product

C:\LabSolutions\Data\Project1\txy\txy1-18.lcd

HPLC Report

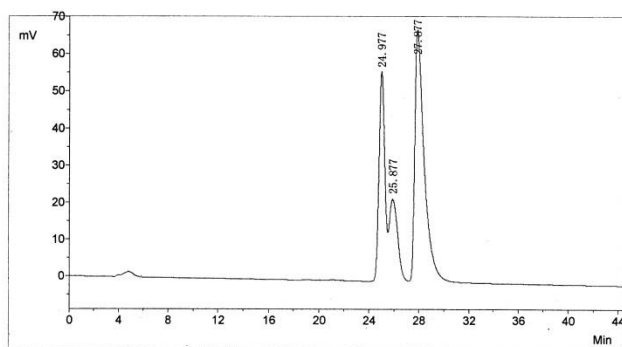
Sample Name:

Data File: TXY4-60-1+-AD-H9820. 7214. che

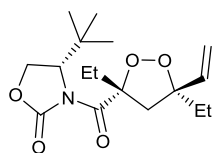
Operator:

Date: 2010-12-16

Time: 08:53

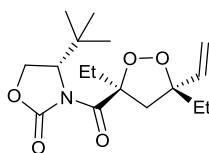


No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		24.977	56416.9	1845525.2	28.5066
2	2		25.877	22314.3	1043026.9	16.1109
3	3		27.877	68027.4	3585480.1	55.3825
Total				146758.7	6474032.1	100.0000



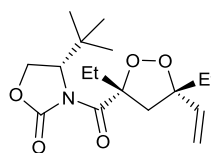
trans-12e

minor product



trans-13e

major product



cis-14e

minor product

HPLC Report

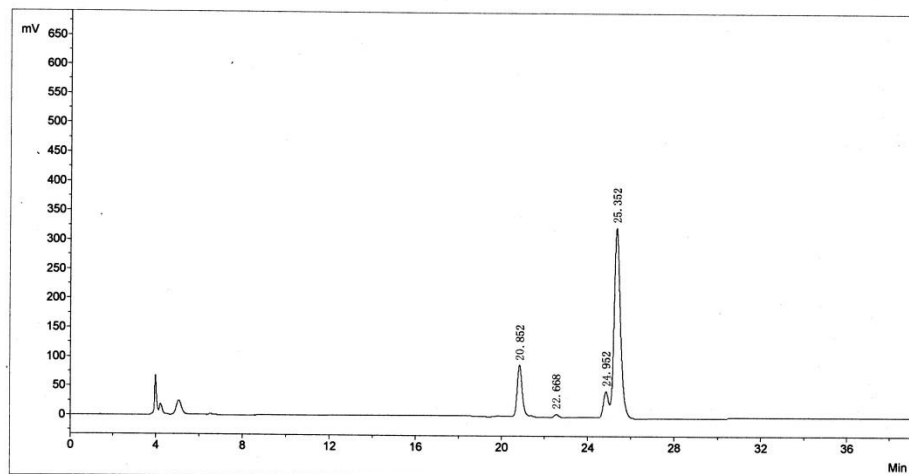
Sample Name:

Data File: TXY-2-84-1.che

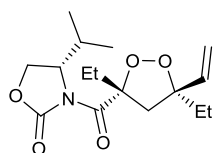
Operator:

Date: 2010-04-06

Time: 19:36

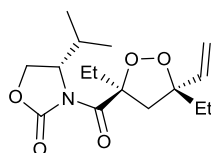


No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		20.852	87958.1	1365530.2	15.5149
2	2		22.668	5273.3	80960.5	0.9199
3	3		24.952	45232.7	663641.1	7.5402
4	4		25.352	327329.6	6691255.6	76.0250
Total				465793.7	8801387.3	100.0000



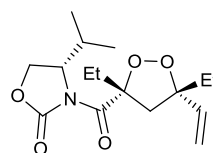
trans-12a

minor product



trans-13a

major product



cis-14a

minor product

reaction in Et₂O as the solvent and in the presence of Sc(OTf)₃ as Lewis acid

HPLC Report

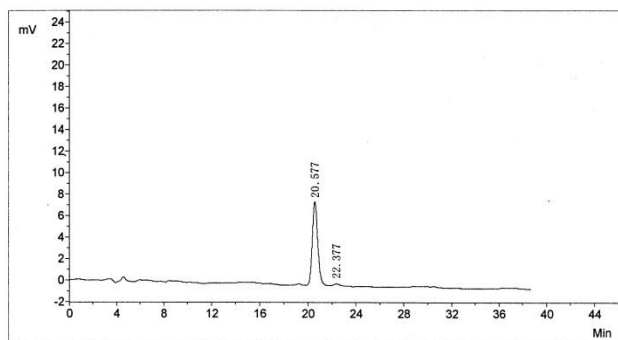
Sample Name:

Data File: TXY2-74-2. che

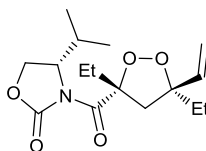
Operator:

Date: 2011-05-30

Time: 09:07



No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		20.577	7803.3	244751.8	97.3077
2	2		22.377	194.7	6771.8	2.6923
Total				7998.0	251523.6	100.0000

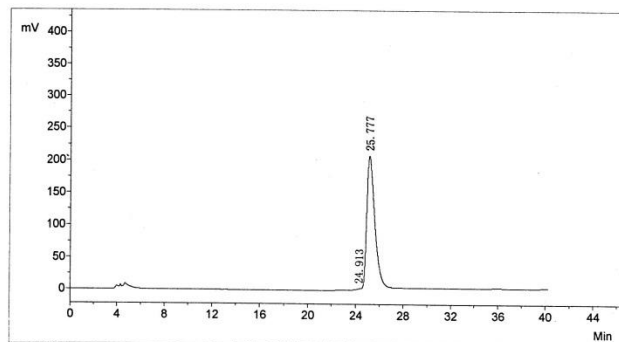


trans-12a

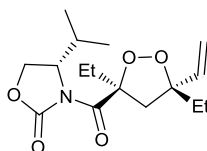
minor product

HPLC Report

Sample Name: Data File: TXY-2-74-3. che
 Operator: Date: 2011-05-30
 Time: 09:48



No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		24.913	1126.1	29160.0	0.2948
2	2		25.777	206483.6	9863090.1	99.7052
Total				207609.6	9892250.1	100.0000



trans-13a

major product

HPLC Report

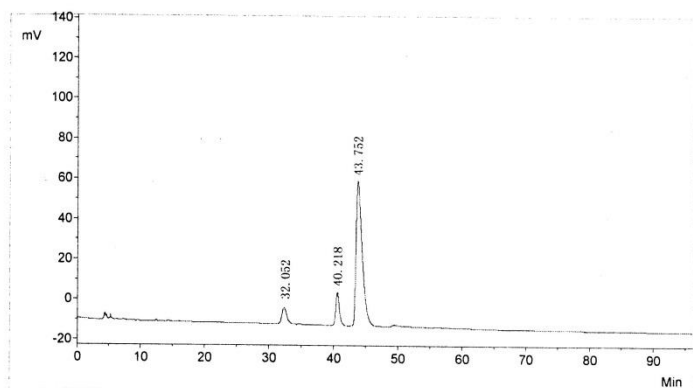
Sample Name:

Data File: TXY-6-3MIX AD-H 982 214 0.7. che

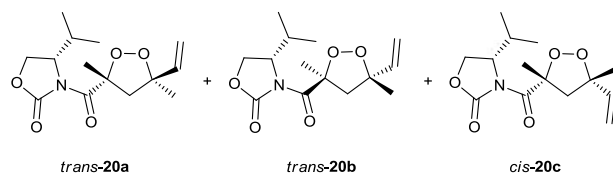
Operator:

Date: 2013-12-16

Time: 10:02



No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		32.052	8212.7	427358.7	7.3014
2	2		40.218	16628.4	651801.0	11.1872
3	3		43.752	72058.7	4770949.9	81.5114
Total				96899.8	5853109.6	100.0000



major product

HPLC Report

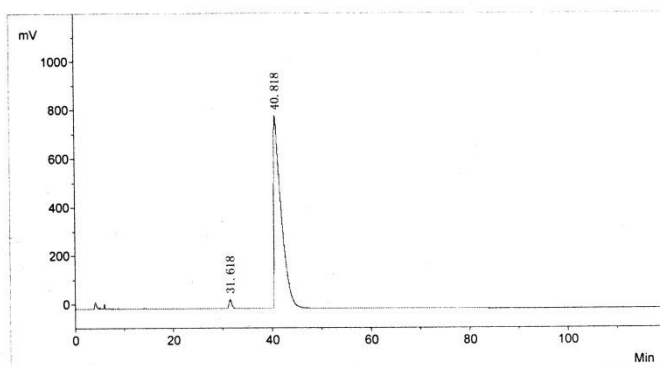
Sample Name:

Data File: TXY-6-3-1.che

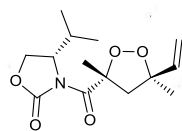
Operator:

Date: 2013-12-16

Time: 12:47



No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		31.618	38031.2	1479995.2	1.7826
2	2		40.818	792953.4	81543572.7	98.2174
Total				830984.7	83023567.9	100.0000



trans-20a

de > 96%

HPLC Report

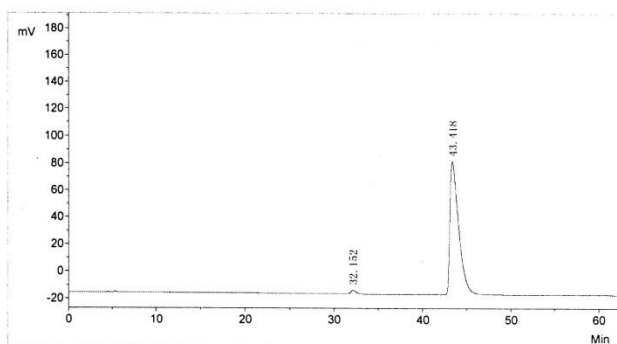
Sample Name:

Data File: TXY-6-3-2.che

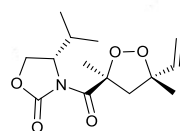
Operator:

Date: 2013-12-16

Time: 11:43



No.	PeakNo	ID. Name	R. Time	PeakHeight	PeakArea	PerCent
1	1		32.152	2779.3	107768.8	1.5657
2	2		43.418	98610.4	6775526.1	98.4343
Total				101419.7	6883294.9	100.0000



trans-20b

major product

de > 96%

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Notes and References

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