

Supporting Information

The base discriminating potential of pyrrolidinyl PNA demonstrated by magnetic Fe_xO_y particles

Claudia Stubinitzky,^a Tirayut Vilaivan^{*b} and Hans-Achim Wagenknecht^{*a}

Institute of Organic Chemistry, Karlsruhe Institute of Technology (KIT), Fritz-Haber-Weg 6, 76131 Karlsruhe, Germany. Fax: +49-721-608-44825; Tel: +49-721-608-47486; E-mail: Wagenknecht@kit.edu

Organic Synthesis Research Unit, Department of Chemistry, Faculty of Science, Chulalongkorn University, Phayathai Road, Patumwan, Bangkok 10330, Thailand, Fax: 02-2187598 Tel: 02-2187627 ext 101 E-mail: vtirayut@chula.ac.th

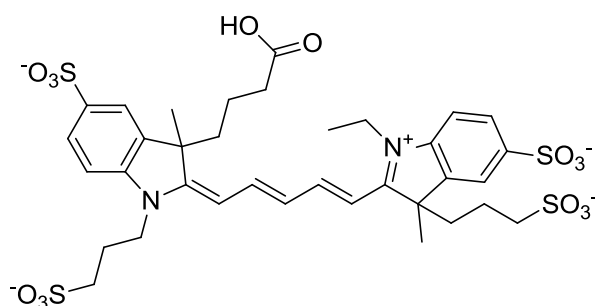
Contents	Page
1. Materials and methods	2
2. PNA/DNA sequences and melting temperatures	8
3. Spectroscopic analysis of capture experiments	10
3.1. General remarks	10
3.2. Immobilisation of biotinylated oligonucleotides	10
3.3. UV/Vis absorption spectroscopy of capture experiments	10
4. References	12

1. Materials and Methods

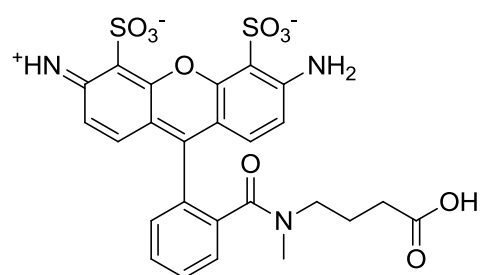
DNA0-DNA5 have been purchased from metabion international AG, DNA6-9 from biomers.net GmbH and pierce streptavidin magnetic beads from Thermo Fisher Scientific (Streptavidin monolayer covalently coupled to magnetic bead surface, bead concentration: 10mg/ml, binding capacity: ~ 3500 pmol biotinylated fluorescein/mg of beads).

Structures of Atto488 and DY649

DY649



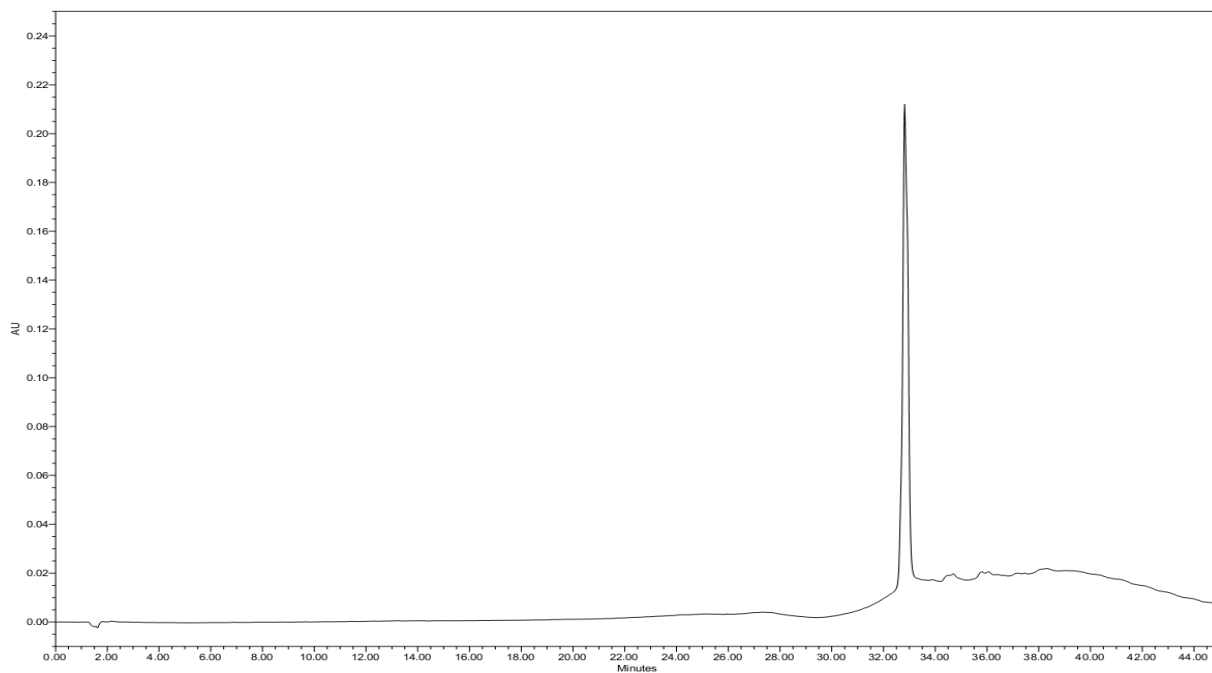
Atto488



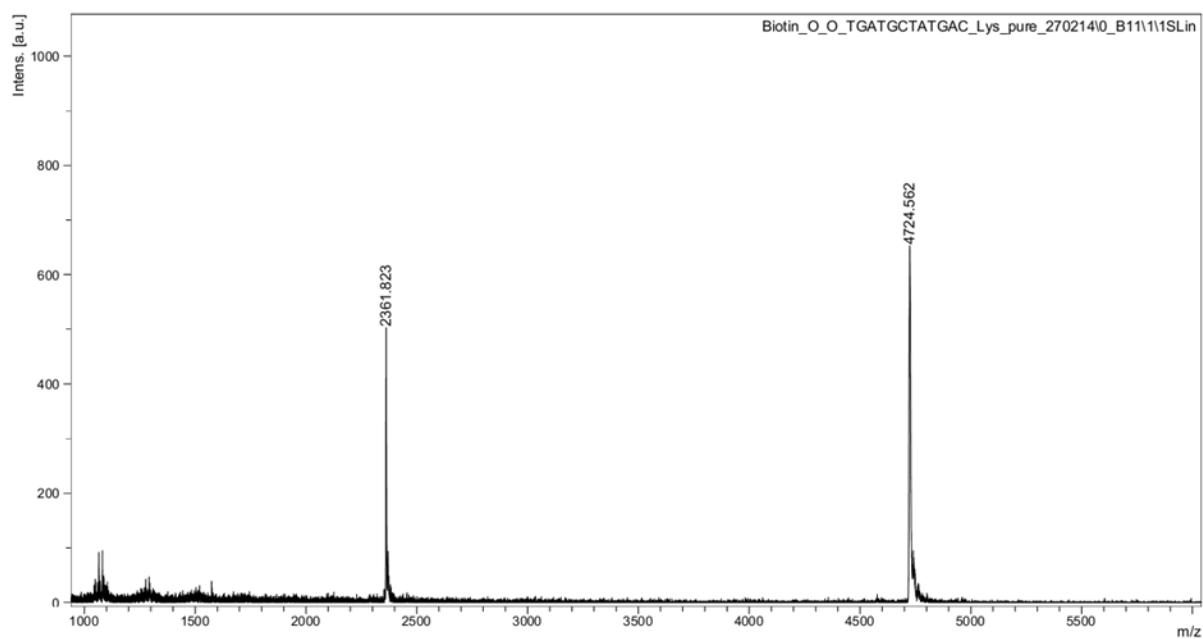
Synthesis of PNA1-PNA5

The PNAs were synthesized manually by Fmoc-solid phase peptide synthesis on TentaGel S RAM resin (0.24 mmol/g substitution, 1.5 μ mol for PNA2 and 1 μ mol for PNA1,3-5) preloaded with Lys(Boc) using the standard protocol.^[1,2] Apcp PNA was modified at the N-terminus with biotin via two aminoethoxyethoxyacetyl linkers. The nucleobase protecting groups (Bz and Ibu) were removed by NH_4OH :dioxane (1:1) treatment at 60 $^\circ\text{C}$ overnight. Cleavage of the PNA oligomers from the resin was achieved by treatment with trifluoroacetic acid (TFA). After precipitation with diethyl ether, the PNAs were purified by reverse phase HPLC using the following conditions: A = 0.1% TFA in water; B = 0.1% TFA in MeOH; gradient = 0-80% B over 70 min. The collected fractions were lyophilized and quantified by their absorbance at 260 nm. Identity was verified by MS (MALDI) in the linear negative mode (matrix: saturated CCA solution with H_2O :acetonitrile (1:1) + 0.1 % TFA) and purity was verified by analytical reverse phase HPLC.

PNA1



(HPLC conditions: C18 column 4.6×50 mm 3 μ , gradient H₂O:MeOH containing 0.1% TFA 90:10 for 5 min then linear gradient to 10:90 over 30 min, 260 nm)



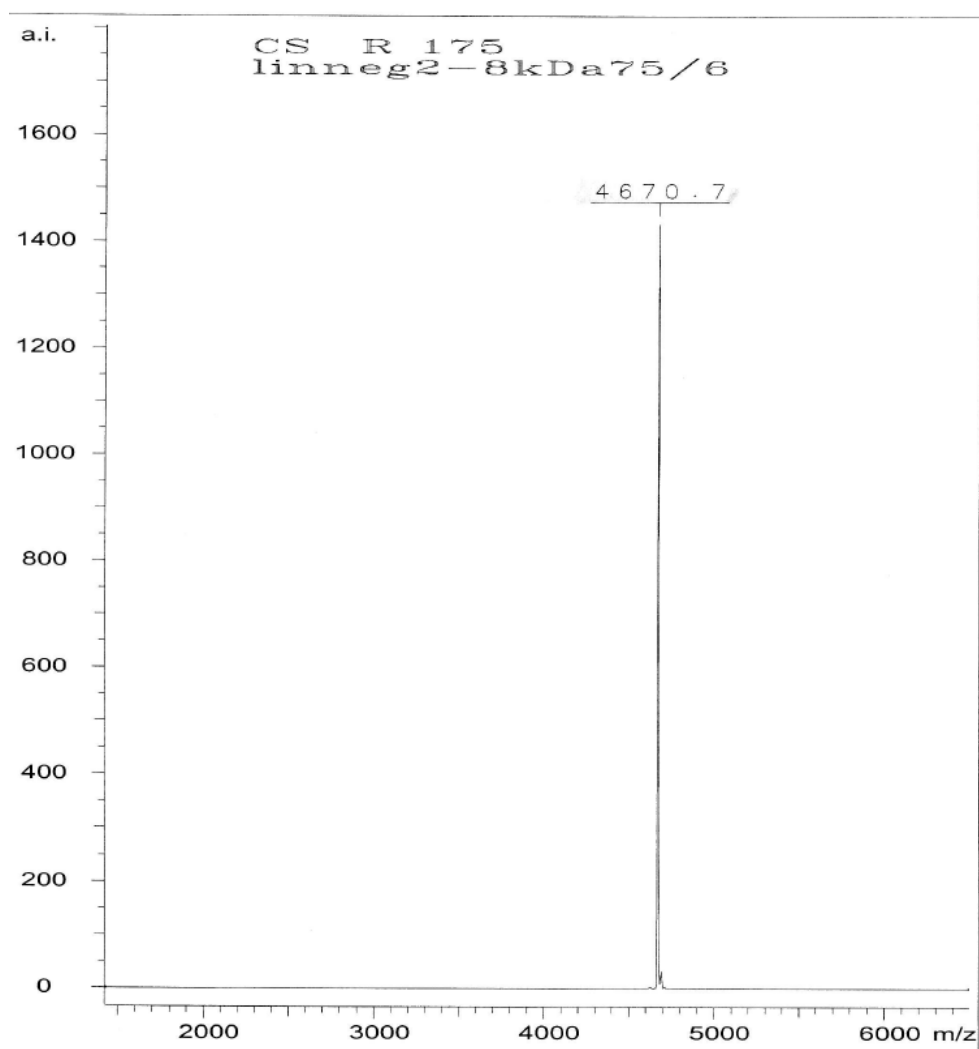
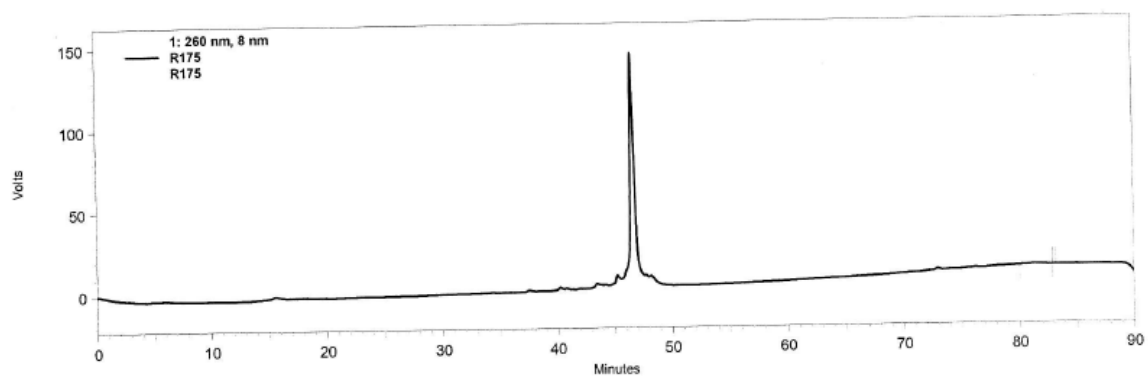
$M \cdot H^+$ (calculated) = 4723.1 g/mol

$M \cdot H^+$ (found) = 4724.6 g/mol

$[M \cdot 2H]^{2+}$ (calculated) = 2362.1 g/mol

$[M \cdot 2H]^{2+}$ (found) = 2361.8 g/mol

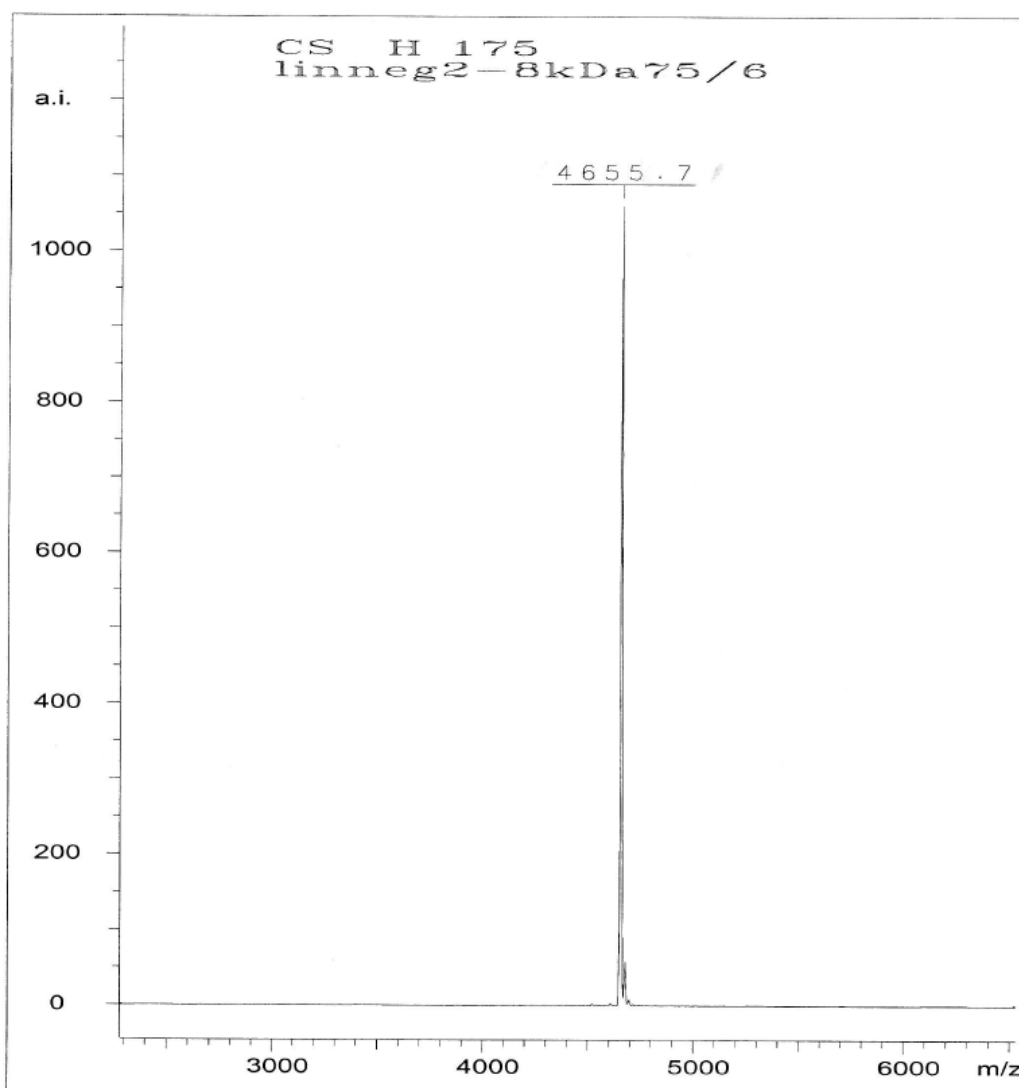
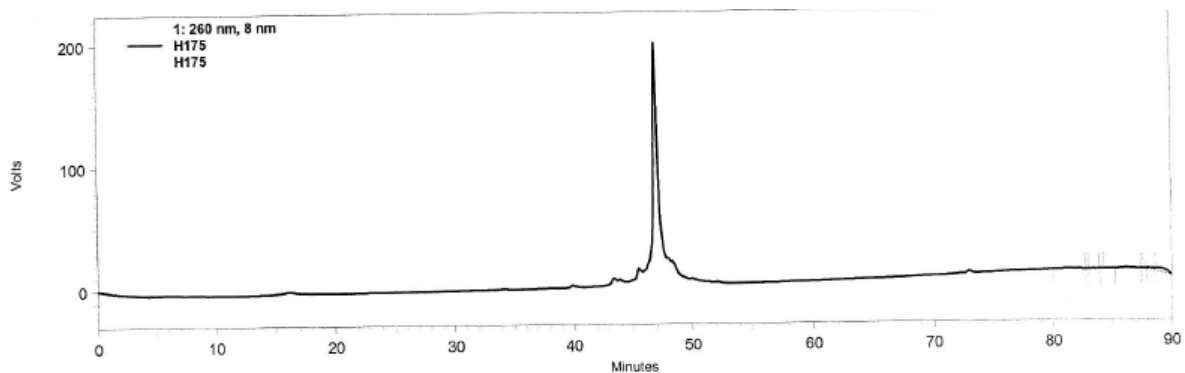
PNA2



$M \cdot H^+$ (calculated) = 4670.1 g/mol

$M \cdot H^+$ (found) = 4670.7 g/mol

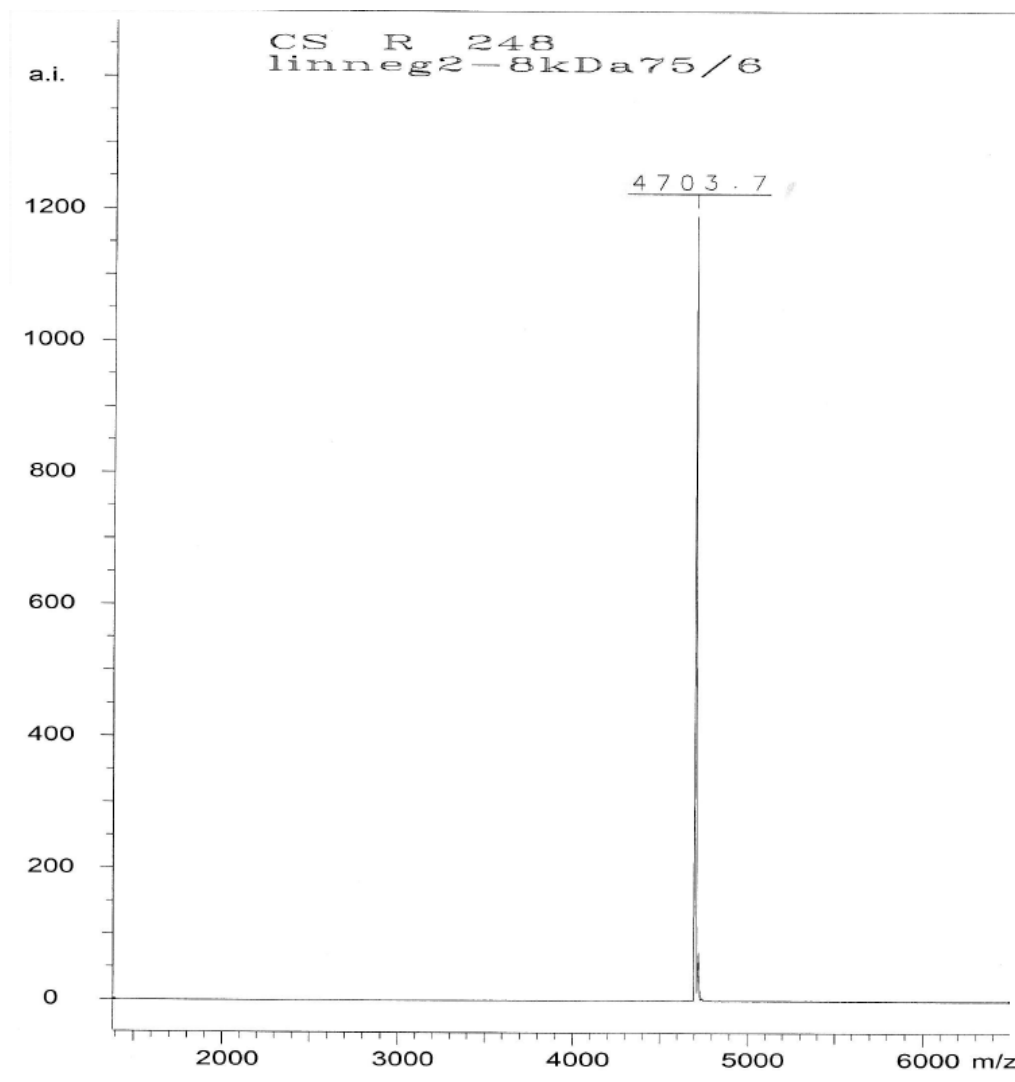
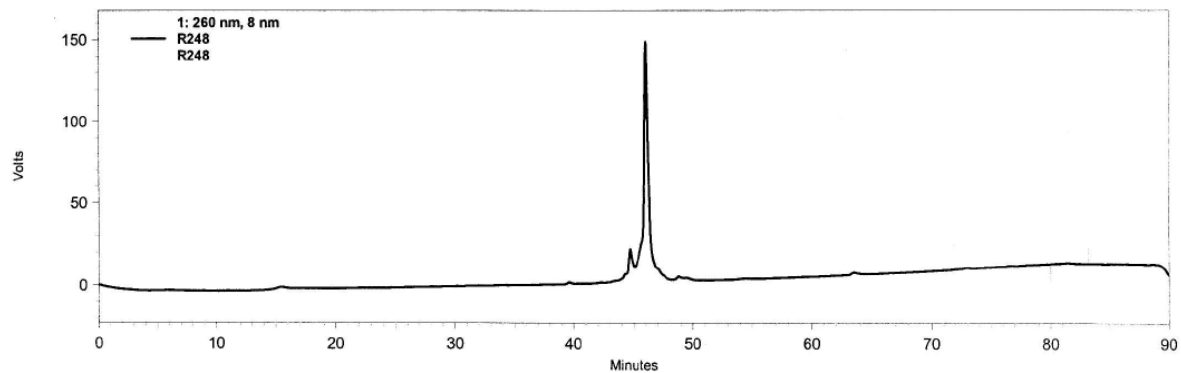
PNA3



$M \cdot H^+$ (calculated) = 4654.1 g/mol

$M \cdot H^+$ (found) = 4655.7 g/mol

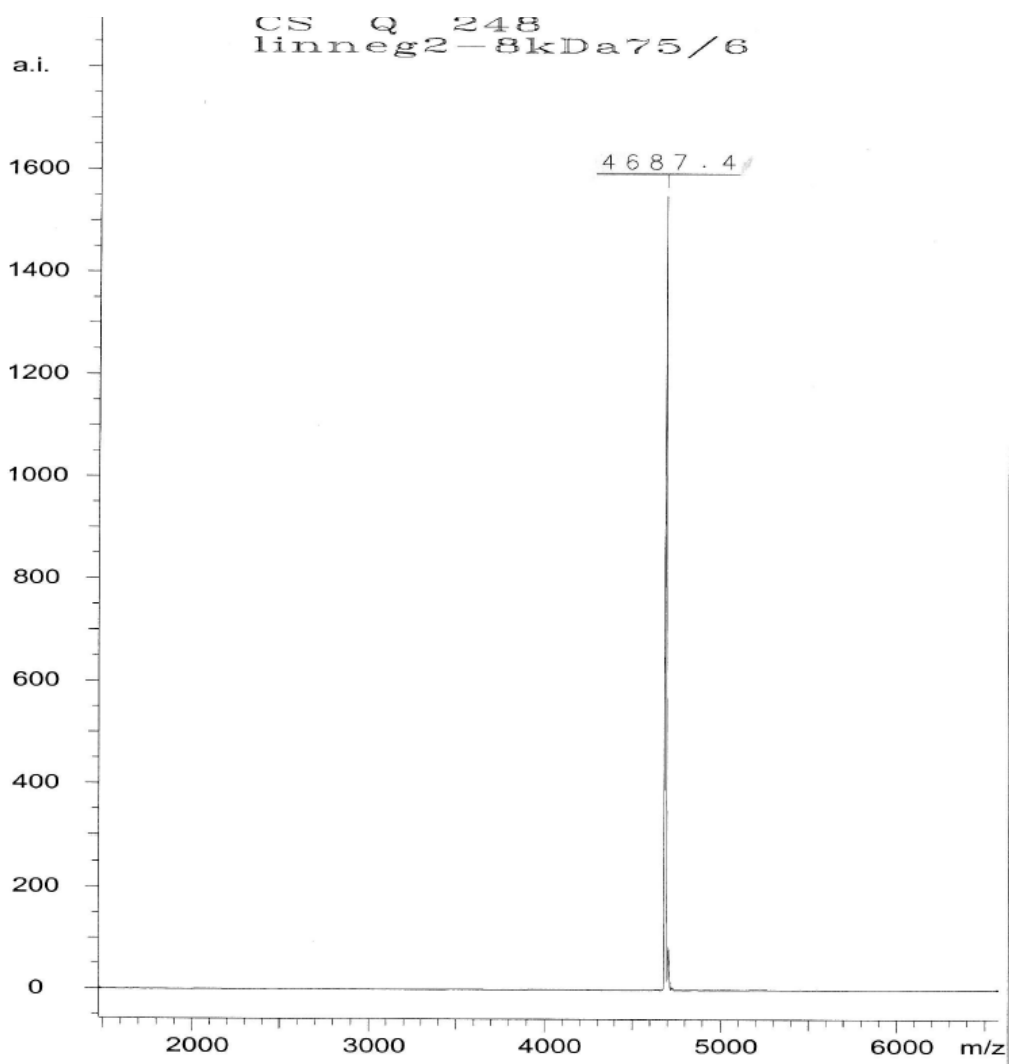
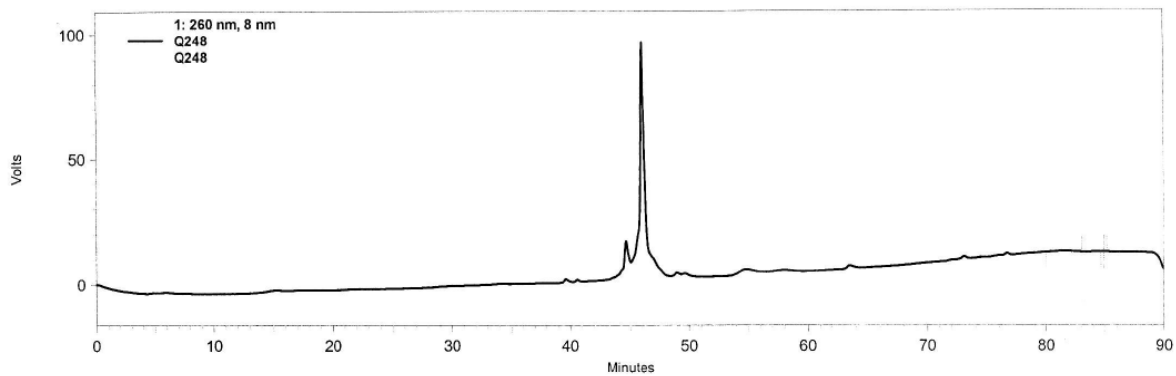
PNA4



$M \cdot H^+$ (calculated) = 4703.1 g/mol

$M \cdot H^+$ (found) = 4703.7 g/mol

PNA5



$M \cdot H^+$ (calculated) = 4687.1 g/mol

$M \cdot H^+$ (found) = 4687.4 g/mol

2. PNA/DNA sequences and melting temperatures

Table S1: Sequences concerning R175H mutation

PNA2:	Biotin-O-O-AGG CGCT GGCCCC-LysNH ₂
PNA3:	Biotin-O-O-AGG CACT GGCCCC-LysNH ₂
DNA10	5'-TGCGGGGCA CGC CCT-3'
DNA11	5'-TGCGGGGCA GTG CCT-3'
DNA12	5'-AGG CGCT GGCCCC-3'
DNA13	5'-AGG CACT GGCCCC-3'

Table S2: Sequences concerning R248Q mutation

PNA4	Biotin-O-O-AAC CGG AGGCCCC-LysNH ₂
PNA5	Biotin-O-O-AAC CAG AGGCCCC-LysNH ₂
DNA14	5'-TGCGGGCCT CCG GTT-3'
DNA15	5'-TGCGGGCCT CTG GTT-3'
DNA16	5'-AAC CGG AGGCCCC-3'
DNA17	5'-AAC CAG AGGCCCC-3'

Melting temperatures

T_m were recorded with a Varian Cary 100 spectrometer using 2.5 μ M PNA-DNA / DNA-DNA solutions in 10 mM NaP_i buffer (pH =7) and 250 mM NaCl: λ = 260 nm, 10-90 °C, interval: 0.7 °C/min.

Table S3: T_m of PNA2-3, DNA12-13 with DNA10-11

PNA	DNA	T_m [°C]	
PNA2	DNA10	69.2	} $\Delta T_m = 16.0$ °C
PNA2	DNA11	53.2	
PNA3	DNA10	36.3	} $\Delta T_m = 33.5$ °C
PNA3	DNA11	69.8	
DNA12	DNA10	66.3	} $\Delta T_m = 10.6$ °C
DNA12	DNA11	55.8	
DNA13	DNA10	43.1	} $\Delta T_m = 19.2$ °C
DNA13	DNA11	62.3	

Table S4: T_m of PNA2-3 with DNA2-3/DNA6-7

PNA	DNA	T_m [°C]	
PNA2	DNA2	65.1	} $\Delta T_m = 15.5$ °C
PNA2	DNA3	49.6	
PNA2	DNA6	67.4	} $\Delta T_m = 11.4$ °C
PNA2	DNA7	56.0	
PNA3	DNA2	29.3	} $\Delta T_m = 34.8$ °C
PNA3	DNA3	64.1	
PNA3	DNA6	35.9	} $\Delta T_m = 38.0$ °C
PNA3	DNA7	73.9	

Table S5: T_m of PNA4-5, DNA16-17 with DNA14-15

PNA	DNA	T_m [°C]	
PNA4	DNA14	70.0	} $\Delta T_m = 13.4$ °C
PNA4	DNA15	56.6	
PNA5	DNA14	47.2	} $\Delta T_m = 21.1$ °C
PNA5	DNA15	68.3	
DNA16	DNA14	63.9	} $\Delta T_m = 10.4$ °C
DNA16	DNA15	53.5	
DNA17	DNA14	45.9	} $\Delta T_m = 15.2$ °C
DNA17	DNA15	61.1	

Table S6: T_m of PNA4-5 with DNA4-5/DNA8-9

PNA	DNA	T_m [°C]	
PNA4	DNA4	70.7	} $\Delta T_m = 15.1$ °C
PNA4	DNA5	55.6	
PNA4	DNA8	70.5	} $\Delta T_m = 13.4$ °C
PNA4	DNA9	57.1	
PNA5	DNA4	43.3	} $\Delta T_m = 25.5$ °C
PNA5	DNA5	68.8	
PNA5	DNA8	46.3	} $\Delta T_m = 23.1$ °C
PNA5	DNA9	69.4	

3. Spectroscopic analysis of capture experiments

3.1. General remarks

All absorption spectra were recorded with a Varian Cary 100 spectrometer using 2.5 μM solutions ($V = 1 \text{ mL}$) in 10 mM NaP_i buffer ($\text{pH} = 7$) and 250 mM NaCl at 20°C , except otherwise mentioned. Oligonucleotide solutions were added to particles, after shaking they were kept for a certain time (specified in spectra). After capturing the Fe_xO_y particles with a magnet, the absorbance of the supernatant was measured spectrophotometrically.

3.2. Immobilisation of biotinylated oligonucleotides and biotinylated PNA

150 μL of magnetic streptavidin coated Fe_xO_y particles (marked with “P” in the spectra) were added into an Eppendorf tube and washed three times with milliQ water ($3 \times 1 \text{ mL}$). Afterwards the biotinylated oligonucleotide (DNA0 or biotinylated PNA) was added for immobilization for 15min.

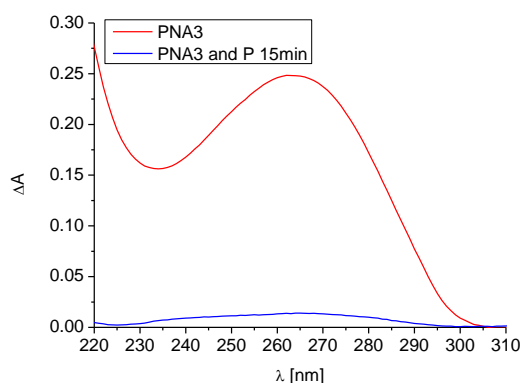


Figure S1: Immobilisation of **PNA3** on Fe_xO_y particles.

3.3. UV/Vis absorption spectroscopy of capture experiments

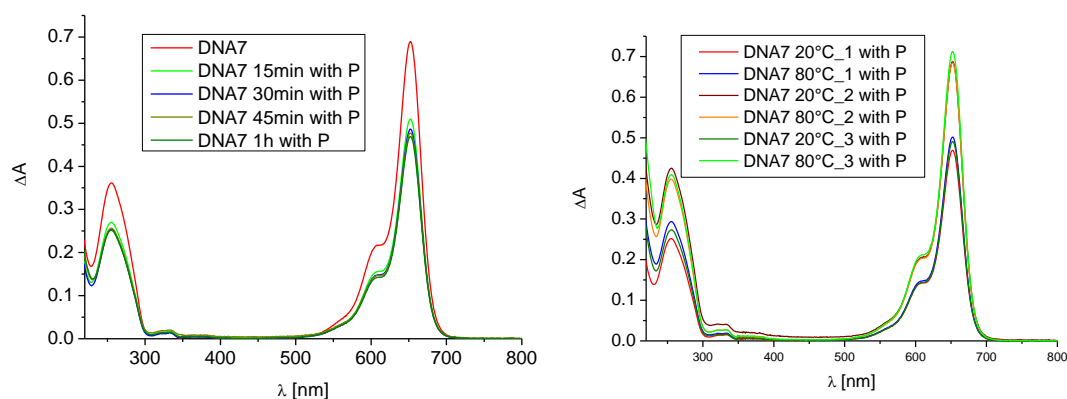


Figure S2: Measurements of **DNA7** with **PNA3** bearing particles.

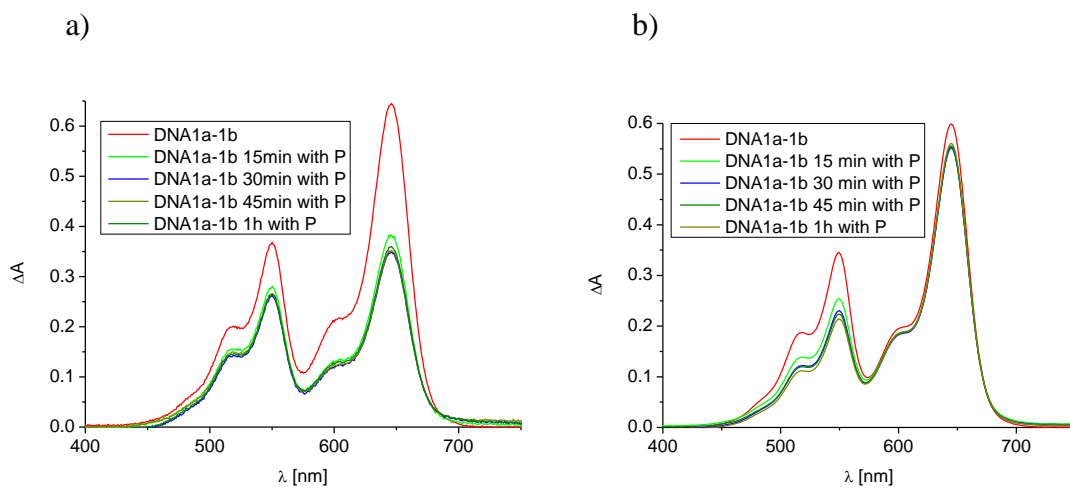


Figure S3: Measurements with a) **DNA0** and b) **PNA1** bearing particles.

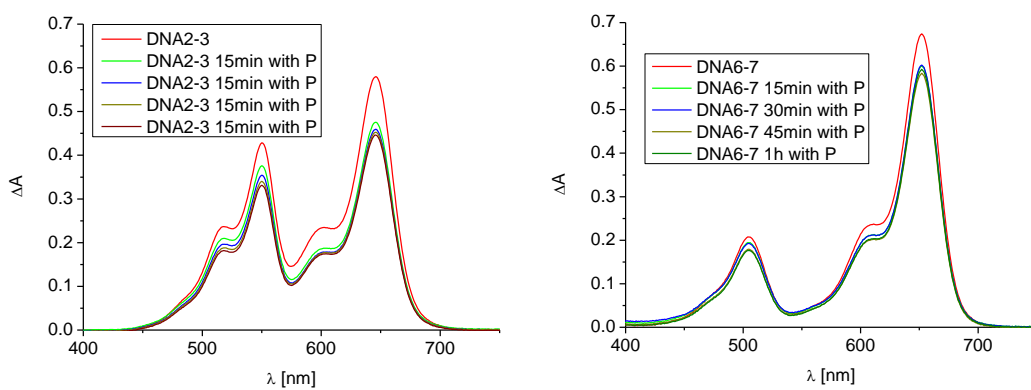


Figure S4: Measurements with **PNA2** bearing particles.

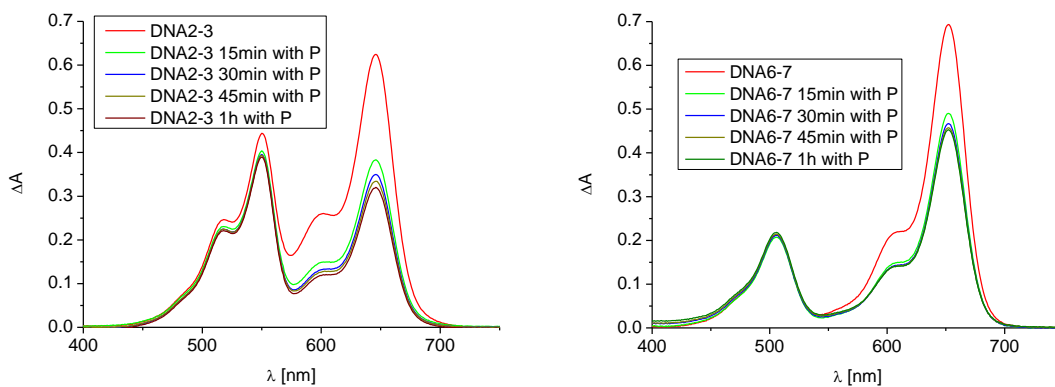


Figure S5: Measurements with **PNA3** bearing particles.

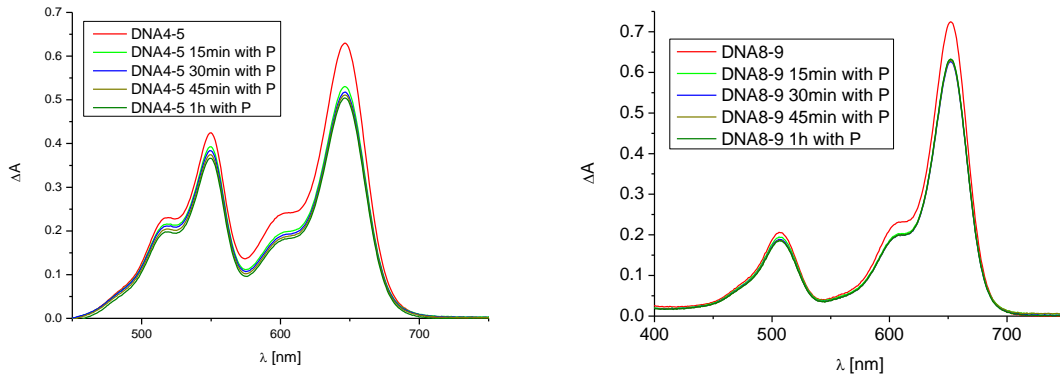


Figure S6: Measurements with **PNA4** bearing particles.

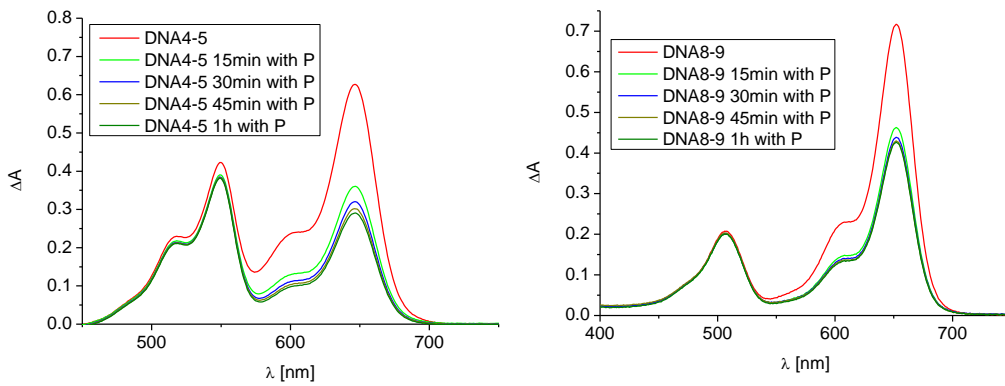


Figure S7: Measurements with **PNA5** bearing particles.

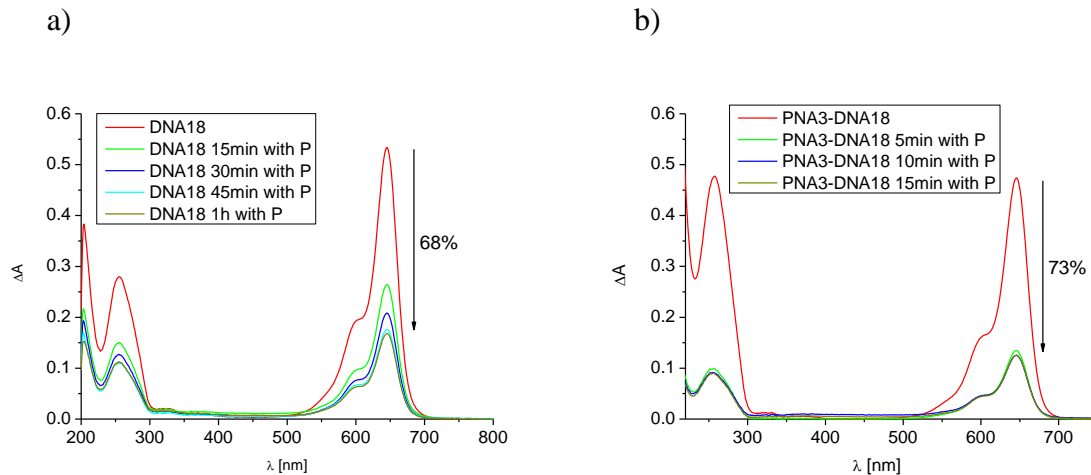


Figure S8: Hybridization control experiment of **DNA18** (5'-Cy5-GGGGCAGTGCCT-3') and **PNA3**: (a) Equimolar hybridization of **DNA18** with **PNA3** bearing particles, (b) Hybridization of **DNA18** (1 equiv.) with excess of **PNA3** (1.2 equiv.) and subsequent immobilization on the streptavidin coated particles.

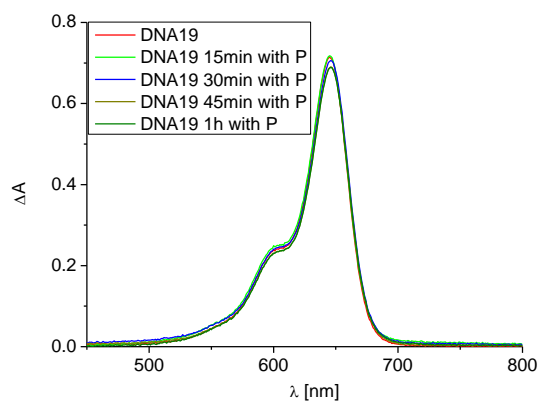


Figure S9: Control experiment using „non-sense“ **DNA19** (5′-Cy5-GATAGTCTCATCGTC-3′) and magnetic particles bearing non-complementary **DNA20** (5′-Biotin-GTACTGTGACTGATGCTGTGACGCA-3′).

4. References

- [1] T. Vilaivan and C. Srisuwannaket, *Org. Lett.* **2006**, *8*, 1897.
- [2] C. Vilaivan, C. Srisuwannaket, C. Ananthanawat, C. Suparpprom, J. Kawakami, Y. Yamaguchi, Y. Tanaka and T. Vilaivan, *Artificial DNA: PNA & XNA* **2011**, *2*, 50.