## Electronic supporting information

TFAA/ $/ \mathrm{H}_{3} \mathrm{PO}_{4}$ mediated unprecedented N -acylation of carbazole leading to small molecules possessing anti-proliferative activities against cancer cells

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General method: Unless stated otherwise, reactions were performed under nitrogen atmosphere using oven dried glassware. Reactions were monitored by thin layer chromatography (TLC) on silica gel plates ( 60 F 254 ), visualizing with ultraviolet light or iodine spray. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were recorded in $\mathrm{CDCl}_{3} / \mathrm{DMSO}-\mathrm{d}_{6}$ solution by using a 400 MHz spectrometer (VARIAN 400 MR ). Proton chemical shifts ( $\delta$ ) are relative to tetramethylsilane (TMS, $\delta=0.00$ ) as internal standard and expressed in ppm. Spin multiplicities are given as s (singlet), d (doublet), t (triplet), dd (doublet of doublet), td (triplet of doublet) and $m$ (multiplet) as well as bs (broad). Coupling constants (J) are given in hertz. Infrared spectra were recorded on a FT-IR spectrometer (FT/IR-4200, JASCO). Melting points were determined by using melting point apparatus (Buchi melting point B540) and are uncorrected. MS spectra were obtained on a mass spectrometer (AGILENT 6430 triple quardrupole LC-MS). Chromatographic purity by HPLC (Agilent 1200 series Chem Station software) was determined by using area normalization method and the condition specified in each case: column, mobile phase (range used), flow rate, detection wavelength, and retention time.

## General procedure for the preparation of compound 3:

A mixture of TFAA ( 2.093 mmol ) and benzoic acid ( 0.598 mmol ) was stirred for 20 min until the solid was dissolved. After stirring for additional 20 min , carbazole, $\mathbf{1}(0.598 \mathrm{mmol})$ was added in one portion. To this mixture was added $85 \% \mathrm{H}_{3} \mathrm{PO}_{4}(0.059 \mathrm{mmol})$ drop wise for a duration of 20 min . The mixture was then stirred for 2 h (monitored by TLC) and the excess of TFA/TFAA was distilled out at atmospheric pressure. The remaining liquid was partitioned between ethyl acetate $(30 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(15 \mathrm{~mL})$. The organic layer was separated and washed with $5 \% \mathrm{NaOH}(7 \mathrm{~mL})$ and then brine $(8 \mathrm{~mL})$. The mixture was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under vacuum. The residue was purified by column chromatography using EtOAc-n-hexane to give the desired product 3.

## (9H-carbazol-9-yl)(phenyl)methanone (3a) ${ }^{1}$



3a was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 a}$.
White solid; yield: $75 \%$; mp: $98-100{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}$ ( $1 \%$ EtOAc- $n$-Hexane) $0.36 ;{ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 8.03-8.01(\mathrm{~m}, 2 \mathrm{H}), 7.73-7.71(\mathrm{~m}, 2 \mathrm{H}), 7.66(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.55-7.50$ ( $\mathrm{m}, 4 \mathrm{H}$ ), 7.37-7.30 (m, 4H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 169.6,139.1,135.7,132.3$, 129.0 (2C), 128.8 (2C), 126.7 (3C), 126.0, 123.3 (3C), 119.8 (3C), 115.7; HPLC: 99.6\%, column: Symmetry C-18 75*4.6 mm, $3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase $\mathrm{B}: \mathrm{CH}_{3} \mathrm{CN}$ (gradient) $\mathrm{T} / \% \mathrm{~B}: 0 / 50,1.0 / 50,6 / 98,10 / 98,10.5 / 50,12 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm , retention time 6.19 min ; $\mathrm{IR}\left(\mathrm{KBr}, \mathrm{cm}^{-1}\right): 2958$, 2914, 1682, 1479, 1298; MS (ES mass): $m / z 272.2$ (M+1).

## (9H-carbazol-9-yl)(4-chlorophenyl)methanone (3b)


$\mathbf{3 b}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 b}$.
White solid; yield: $55 \%$; mp: $155-160{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}$ (Pure- $n$-Hexane) $0.27 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right)$ 8: 8.04-8.02 (m, 2H), $7.70(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.55-7.51(\mathrm{~m}, 4 \mathrm{H}), 7.41-7.35(\mathrm{~m}, 4 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 168.4,138.9,138.8,133.9,130.6$ (2C), 129.2 (3C), 126.8 (2C), 126.0, 123.5 (2C), 119.9 (2C), 115.6 (2C), 109.9; HPLC: $98.7 \%$, column: Symmetry C$1875 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, 1.0/50, 6/98, 10/98, 10.5/50, 12/50; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm , retention time 6.80 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2925, 2843, 1682, 1490, 1293; MS (ES mass): $m / z 306.1(\mathrm{M}+1)$; Elemental analysis found $\mathrm{C}, 74.31 ; \mathrm{H}, 3.95 ; \mathrm{N}, 4.70 ; \mathrm{C}_{19} \mathrm{H}_{12} \mathrm{ClNO}$ requires C, 74.64; H, 3.96; N, 4.58.
(9H-carbazol-9-yl)(2-iodophenyl)methanone (3c) ${ }^{2}$

$\mathbf{3 c}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 c}$.
Liquid; yield: $60 \%$; $\mathrm{R}_{f}\left(2 \%\right.$ EtOAc- $n$-Hexane) $0.36 ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 7.98(\mathrm{t}, J=$ $7.2 \mathrm{~Hz}, 4 \mathrm{H}), 7.55(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.47(\mathrm{dd}, J=7.6,1.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.38-7.24(\mathrm{~m}, 8 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta: 168.6,142.3,139.9$ (2C), 138.5, 131.8 (2C), 128.9, 128.4 (2C), 127.3 (2C), 126.6, 124.1 (2C), 119.7 (2C), 116.0, 92.6. HPLC: 99.0\%, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase $\mathrm{B}: \mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, 1.0/50, 6/98, 10/98, 10.5/50, 12/50; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time $6.26 \mathrm{~min} ; \operatorname{IR}\left(\mathrm{KBr}, \mathrm{cm}^{-1}\right): 2923,2845,1640,1432,1256 ;$ MS (ES mass): $m / z$ $397.5(\mathrm{M}+1)$.

## (5-Bromo-2-chlorophenyl)(9H-carbazol-9-yl)methanone (3d)



3d was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 d}$.
White solid; yield: $64 \%$; mp: $98-100{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(7 \%\right.$ EtOAc- $n$-Hexane) $0.57 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.00(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.69-7.66(\mathrm{~m}, 2 \mathrm{H}), 7.41-7.32(\mathrm{~m}, 7 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ) $\delta: 164.8,138.2,137.8,134.9$ (2C), 131.9 (2C), 131.5 (2C), 130.3, 127.4 (2C), 126.7, 124.4 (2C), 121.3, 119.9 (2C), 115.6; HPLC: 98.8\%, column: Symmetry C-18 75*4.6 mm, $3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: $0 / 50,1.0 / 50,6 / 98,10 / 98,10.5 / 50,12 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.93 min; IR (KBr, $\mathrm{cm}^{-1}$ ): 2920, 2841, 1687, 1446, 1391, 1298; MS (ES mass): m/z 385.5 (M+1); Elemental analysis found C, 59.21; H, 2.89; N, 3.77; $\mathrm{C}_{19} \mathrm{H}_{11} \mathrm{BrClNO}$ requires C, 59.33; H, 2.88; N, 3.64.

## (2-Bromophenyl)(9H-carbazol-9-yl)methanone (3e) ${ }^{2}$


$\mathbf{3 e}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 e}$.
Semi solid; yield: 60\%; $\mathrm{R}_{f}\left(5 \%\right.$ EtOAc- $n$-Hexane) $0.46 ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 8.00(\mathrm{~d}, J$ $=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.73(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.52-7.48(\mathrm{~m}, 4 \mathrm{H}), 7.38-7.28(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(100$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 167.2,138.4,138.3,133.5$ (2C), 131.9 (2C), 128.7 (2C), 128.2 (2C), 127.2 (2C), 126.6, 124.1 (2C), 119.7 (2C), 115.8; HPLC: 99.4\%, column: Symmetry C-18 75*4.6 mm, 3.5 mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) $\mathrm{T} / \% \mathrm{~B}: 0 / 50$, 1.0/50, 6/98, 10/98, 10.5/50, 12/50; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.21 min ; IR (KBr, $\mathrm{cm}^{-1}$ ): 2920, 2849, 1676, 1594, 1446, 1210; MS (ES mass): m/z $349.5(\mathrm{M}+1)$.

## (9H-carbazol-9-yl)(3-chlorophenyl)methanone (3f)



3f was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 f}$.
White solid; yield: $75 \%$; mp: $105-110{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}$ (Pure- $n$-Hexane) $0.27 ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 8.02(\mathrm{~d}, J=8.0,2 \mathrm{H}), 7.73(\mathrm{~s}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.52(\mathrm{~d}$, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.48(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39-7.32(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta:$ $167.9,138.8,137.3,135.0,132.3,130.2,128.9,127.0$ (2C), 126.8 (2C), 126.1, 123.7 (2C), 119.9 (2C), 115.7 (2C), 109.9; HPLC: 99.5\%, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, 1.0/50, $6 / 98,10 / 98,10.5 / 50,12 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.77 min ; IR $\left(\mathrm{KBr}, \mathrm{cm}^{-1}\right): 2920,2865,2849,1649,1561,1364$; MS (ES mass): m/z $305.5(\mathrm{M}+1)$; Elemental analysis found $\mathrm{C}, 74.78 ; \mathrm{H}, 3.97 ; \mathrm{N}, 4.32 ; \mathrm{C}_{19} \mathrm{H}_{12} \mathrm{ClNO}$ requires $\mathrm{C}, 74.64 ; \mathrm{H}, 3.96 ; \mathrm{N}, 4.58$.

## (9H-carbazol-9-yl)(2-chlorophenyl)methanone (3g) ${ }^{2}$


$\mathbf{3 g}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 g}$.
White solid; yield: $60 \%$; mp: $110-115{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}(7 \%$ EtOAc- $n$-Hexane) $0.37 ; 8.00$ (d, $J=7.6 \mathrm{~Hz}$, 2 H ), 7.58-7.54 (m, 3H), 7.50-7.46 (m, 2H), 7.39-7.32 (m, 5H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta:$ $166.5,138.4,136.2,131.8$ (2C), 131.3, 130.4 (2C), 128.7 (2C), 127.6, 127.2 (2C), 126.5, 124.1 (2C), 119.7 (2C), 115.7; HPLC: $99.6 \%$, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, 1.0/50, 6/98, $10 / 98,10.5 / 50,12 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.19 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-}$ ${ }^{1}$ ): $2840,2835,1676,1435,1331,1161$; MS (ES mass): $m / z 305.6$ (M+1).

## (9H-carbazol-9-yl)(0-tolyl)methanone (3h)



3h was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 h}$.
White solid; yield: $60 \%$; mp: $75-80{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(2 \%\right.$ EtOAc- $n$-Hexane) $0.33 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.00(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.52(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.47-7.39(\mathrm{t}, J=8.8 \mathrm{~Hz}, 3 \mathrm{H}), 7.37-$ $7.26(\mathrm{~m}, 6 \mathrm{H}), 2.26(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 169.8,138.6,136.5,135.5,131.1$ (2C), 130.8 (2C), 127.3, 127.1 (2C), 126.5, 126.3, 123.7 (2C), 119.6 (2C), 115.8 (2C), 19.0; HPLC: $99.7 \%$, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: $0 / 50,1.0 / 50,6 / 98,10 / 98,10.5 / 50,12 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 230 nm retention time 6.56 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2834, 2367, 1649, 1331, 1062, 750; MS (ES mass): m/z 285.6 (M+1); Elemental analysis found C, 84.07; H, 5.30; $\mathrm{N}, 4.99 ; \mathrm{C}_{20} \mathrm{H}_{15} \mathrm{NO}$ requires $\mathrm{C}, 84.19 ; \mathrm{H}, 5.30 ; \mathrm{N}, 4.91$.
(9H-carbazol-9-yl)(p-tolyl)methanone (3i)

$\mathbf{3 i}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 i}$.
White solid; yield: $68 \%$; mp: $135-140{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}$ ( $2 \%$ EtOAc- $n$-Hexane) $0.36 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.03-8.01(\mathrm{~m}, 2 \mathrm{H}), 7.65(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.55-7.53(\mathrm{~m}, 2 \mathrm{H}), 7.37-7.26(\mathrm{~m}, 6 \mathrm{H})$, $2.49(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 169.6,143.2,139.1,132.7,129.4$ (3C), 129.3 (3C), 126.5 (2C), 125.8, 123.1 (2C), 119.7 (2C), 115.6 (2C), 21.7; HPLC: 99.7\%, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, 1.0/50, 6/98, 10/98, 10.5/50, 12/50; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.74 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2914, 1676, 1605, 1435, 1293; MS (ES mass): m/z 285.6 $(\mathrm{M}+1)$; Elemental analysis found $\mathrm{C}, 84.27 ; \mathrm{H}, 5.29 ; \mathrm{N}, 4.70 ; \mathrm{C}_{20} \mathrm{H}_{15} \mathrm{NO}$ requires C, 84.19; H, 5.30; N, 4.91.
(9H-carbazol-9-yl)(4-iodophenyl)methanone (3j) ${ }^{3}$

$\mathbf{3 j}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2} \mathbf{j}$.
Yellow; yield: $70 \%$; mp: $145-147{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(3 \%\right.$ EtOAc- $n$-Hexane) $0.47 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.02(\mathrm{dd}, J=5.6,2.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.90(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.54-7.51(\mathrm{~m}, 2 \mathrm{H}), 7.47(\mathrm{~d}, J$ $=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.38-7.34(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 168.6,138.8(2 \mathrm{C}), 138.1$ (2C), 134.9, 130.5 (2C), 126.8 (2C), 126.0 (2C), 123.5 (2C), 119.8 (2C), 115.6 (2C), 99.6; HPLC: $96.9 \%$, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, $1 / 50,6 / 98,12 / 98,13 / 50,15 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 7.16 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2914, 1676, 1435, 1293; MS (ES mass): m/z 397.7 (M+1).

$\mathbf{3 k}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 k}$. White soild; yield: $60 \%$; mp: $130-132{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(2 \%\right.$ EtOAc- $n$-Hexane) $0.25 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.89(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.02(\mathrm{t}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.60-7.56(\mathrm{~m}, 1 \mathrm{H}), 7.56-7.34(\mathrm{~m}$, $4 \mathrm{H}), 7.34(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.26(\mathrm{~s}, 1 \mathrm{H}), 7.14(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ) $\delta: 163.3,142.1,138.8,137.2,135.9,132.4,131.6,128.6$ (2C), 128.0, 127.1 (2C), 126.5, $124.9,124.0,120.3,119.4,118.1,112.9$; HPLC: $97.5 \%$, column: Symmetry C-18 75*4.6 mm, $3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: $0 / 50,1 / 50,6 / 98,12 / 98,13 / 50,15 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.53 min ; IR (KBr, $\mathrm{cm}^{-1}$ ): 2914, 1676, 1435, 1293; MS (ES mass): m/z 339.9 (M+1); Elemental analysis found $\mathrm{C}, 66.90 ; \mathrm{H}, 3.19 ; \mathrm{N}, 4.30 ; \mathrm{C}_{19} \mathrm{H}_{11} \mathrm{Cl}_{2} \mathrm{NO}$ requires $\mathrm{C}, 67.08 ; \mathrm{H}, 3.26 ; \mathrm{N}, 4.12$.

## (9H-carbazol-9-yl)(3,4-dichlorophenyl)methanone (31)



31 was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 l}$.
White soild; yield: $62 \%$; mp: $128-130{ }^{\circ} \mathrm{C} ; \mathrm{R}_{f}$ ( $n$-Hexane) $0.28 ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ : $8.02(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.86(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.53(\mathrm{dd}, J=4.00$, $1.6 \mathrm{~Hz}, 3 \mathrm{H}), 7.38(\mathrm{~s}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 167.0,138.7,136.9,135.2,135.2$, 133.5, 131.0, 130.9, 128.2, 126.9 (2C), 126.8, 126.1, 123.8 (2C), 120.0 (2C), 115.6 (2C); HPLC: $92.2 \%$, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, $1 / 50,6 / 98,12 / 98,13 / 50,15 / 50$; flow rate: 1.0 $\mathrm{mL} / \mathrm{min}$; UV 225 nm retention time 7.25 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2934, 2843, 1676, 1435, 1293; MS (ES mass): m/z 339.9 ( $\mathrm{M}+1$ ); Elemental analysis found $\mathrm{C}, 66.85 ; \mathrm{H}, 3.28$; N , 4.26; $\mathrm{C}_{19} \mathrm{H}_{11} \mathrm{Cl}_{2} \mathrm{NO}$ requires $\mathrm{C}, 67.08 ; \mathrm{H}, 3.26 ; \mathrm{N}, 4.12$.

## (9H-carbazol-9-yl)(2,4-dichlorophenyl)methanone (3m)


$\mathbf{3 m}$ was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 m}$.
White soild; yield: $70 \%$; mp: $134-136{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(1 \%\right.$ EtOAc- $n$-Hexane) $0.18 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.00(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.57(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.51-7.48(\mathrm{~m}, 3 \mathrm{H}), 7.41-7.26(\mathrm{~m}$, $5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 165.5,138.3,137.4,134.7,132.4,130.4$ (2C), 129.8, 128.1 (2C), 127.3 (2C), 126.6, 124.3 (2C), 119.8 (2C), 115.6, 109.9; HPLC: 99.0\%, column: Symmetry C-18 75*4.6 mm, $3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, $1 / 50,6 / 98,12 / 98,13 / 50,15 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.89 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2924, 2860, 1662, 1421, 1163; MS (ES mass): m/z $339.8(\mathrm{M}+1)$; Elemental analysis found $\mathrm{C}, 67.01 ; \mathrm{H}, 3.20 ; \mathrm{N}, 4.30 ; \mathrm{C}_{19} \mathrm{H}_{11} \mathrm{Cl}_{2} \mathrm{NO}$ requires C, 67.08; H, 3.26; N, 4.12.

## (9H-carbazol-9-yl)(2,5-dichlorophenyl)methanone (3n)



3n was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 n}$.
White soild; yield: $64 \%$; mp: $115-118{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(2 \%\right.$ EtOAc- $n$-Hexane) $0.36 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.01(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.55-7.47(\mathrm{~m}, 4 \mathrm{H}), 7.41-7.35(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ) $\delta: 164.8,138.2,137.6,133.7,131.9$ (2C), 131.6 (2C), 128.6, 127.4 (2C), 126.6, 124.4 (2C), 119.9 (2C), 115.6, 115.6, 105.2; HPLC: 93.1\%, column: Symmetry C-18 75*4.6 mm, $3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: $0 / 50,1 / 50,6 / 98,12 / 98,13 / 50,15 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 6.79 min; IR (KBr, cm ${ }^{-1}$ ): 2954, 2856, 1675, 1469, 1245; MS (ES mass): m/z 339.8 (M+1); Elemental analysis found $\mathrm{C}, 67.29 ; \mathrm{H}, 3.24 ; \mathrm{N}, 4.01 ; \mathrm{C}_{19} \mathrm{H}_{11} \mathrm{Cl}_{2} \mathrm{NO}$ requires $\mathrm{C}, 67.08 ; \mathrm{H}, 3.26 ; \mathrm{N}, 4.12$.

## (9H-carbazol-9-yl)(3,5-dichlorophenyl)methanone (3o)



30 was prepared according to the general procedure presented above by using $\mathbf{1}$ and $\mathbf{2 0}$.
White soild; yield: $60 \%$; mp: $158-160{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(1 \%\right.$ EtOAc- $n$-Hexane) $0.24 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.02(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.63(\mathrm{~s}, 1 \mathrm{H}), 7.59(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.53(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, 2H), 7.41-7.35 (m, 4H); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 168.6,148.1,138.6$ (2C), 138.3, 135.8 (2C), 132.1 (2C), 127.2, 127.0, 126.2 (2C), 124.0 (2C), 120.0 (2C), 115.6 (2C); HPLC: 97.7\%, column: Symmetry C-18 $75 * 4.6 \mathrm{~mm}, 3.5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Acetic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/50, $1 / 50,6 / 98,12 / 98,13 / 50,15 / 50$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; UV 225 nm retention time 7.40 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2926, 2834, 1669, 1450, 1167. MS (ES mass): $m / z 339.8(\mathrm{M}+1)$; Elemental analysis found $\mathrm{C}, 67.21 ; \mathrm{H}, 3.27 ; \mathrm{N}, 4.03$; $\mathrm{C}_{19} \mathrm{H}_{11} \mathrm{Cl}_{2} \mathrm{NO}$ requires $\mathrm{C}, 67.08 ; \mathrm{H}, 3.26 ; \mathrm{N}, 4.12$.

## 1-(9H-carbazol-9-yl)-2,2-diphenylethanone (3p)



White solid; yield: $60 \%$; mp: $178-180{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(10 \%\right.$ EtOAc- $n$-Hexane) $0.64 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.18(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 8.01-7.99(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.36(\mathrm{~m}, 12 \mathrm{H}), 7.31(\mathrm{q}, J=4.4 \mathrm{~Hz}$, $2 \mathrm{H}), 6.14(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 172.3,139.0,138.4,129.2,128.8,127.5$, $127.4,126.5,123.8,120.2,119.8,116.3,58.5$; MS (ES mass): $m / z 362.0(\mathrm{M}+1)$; Elemental analysis found $\mathrm{C}, 86.62 ; \mathrm{H}, 5.35 ; \mathrm{N}, 3.59 ; \mathrm{C}_{26} \mathrm{H}_{19} \mathrm{NO}$ requires $\mathrm{C}, 86.40 ; \mathrm{H}, 5.30 ; \mathrm{N}, 3.88$.

## 1-(9H-Carbazol-9-yl)butan-1-one (3q)



Semi solid; yield: $63 \%$; $\mathrm{R}_{f}\left(10 \%\right.$ EtOAc- $n$-Hexane) $0.66 ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 8.24(\mathrm{~d}$, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 8.01(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.49(\mathrm{t}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.39(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.14$ ( $\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 2.02-1.93 (m, 2H), $1.13(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta:$ 173.1, 138.6, 127.3, 126.4, 123.5, 119.8, 116.5, 41.1, 18.2, 13.9; MS (ES mass): $m / z 238.0$ (M+1); Elemental analysis found C, 80.75; H, 6.39; N, 6.03; $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{NO}$ requires C, 80.98; H, 6.37; N, 5.90.

## (3-Bromo-9H-carbazol-9-yl)(phenyl)methanone (3r)



White solid; yield: $65 \%$; mp: $120-122{ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}\left(10 \%\right.$ EtOAc- $n$-Hexane) $0.68 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.11(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.96-7.94(\mathrm{~m}, 1 \mathrm{H}), 7.70-7.68(\mathrm{~m}, 2 \mathrm{H}), 7.67-7.64(\mathrm{~m}, 1 \mathrm{H})$, $7.54(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.45-7.40(\mathrm{~m}, 3 \mathrm{H}), 7.37-7.30(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ : $169.3,139.4,137.9,135.3,132.6,129.4,129.0,128.9,127.8,127.4,124.7,123.6,122.6,120.0$, 117.2, 116.5, 115.7; MS (ES mass): m/z 351.9 (M+1); Elemental analysis found C, 65.31; H, 3.57; N, 3.89; $\mathrm{C}_{19} \mathrm{H}_{12} \mathrm{BrNO}$ requires C, 65.16; H, 3.45; N, 4.00.

## Synthesis of (9H-carbazole-3,9-diyl)bis(phenylmethanone) (4a) ${ }^{4}$

A mixture of TFAA ( 1.291 mmol ) and benzoic acid ( 0.369 mmol ) was stirred for 20 min until the solid was dissolved. After stirring for additional 20 min , (9H-carbazol-9yl )(phenyl)methanone 3a ( 0.369 mmol ) was added in one portion. To this mixture was added $85 \% \mathrm{H}_{3} \mathrm{PO}_{4}(0.036 \mathrm{mmol})$ drop wise for a duration of 20 min . The mixture was then stirred for

2h (monitored by TLC) and the excess of TFA/TFAA was distilled out at atmospheric pressure. The remaining liquid was partitioned between ethyl acetate ( 30 mL ) and $\mathrm{H}_{2} \mathrm{O}(15 \mathrm{~mL})$. The organic layer was separated and washed with $5 \% \mathrm{NaOH}(7 \mathrm{~mL})$ and then brine ( 8 mL ). The mixture was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under vacuum. The residue was purified by column chromatography using EtOAc-n-hexane to give the desired product $\mathbf{4 a}$.


White solid; yield: $72 \%$; mp: $162-165{ }^{\circ} \mathrm{C}$ ( lit $^{4} 170{ }^{\circ} \mathrm{C}$ ); $\mathrm{R}_{f}$ ( $8 \%$ EtOAc- $n$-Hexane) $0.8 ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 8.54(\mathrm{~d}, J=1.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.07(\mathrm{dd}, J=6.9,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.88-7.82(\mathrm{~m}, 3 \mathrm{H})$, 7.79-7.75 (m, 2H), $7.70(\mathrm{dd}, J=10.6,4.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.64-7.51(\mathrm{~m}, 7 \mathrm{H}), 7.43-7.37(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta: 196.2,169.5,141.6,139.7,138.0,135.0,132.9$ (2C), 132.6, 132.2, 130.0 (2C), 129.2 (2C), 129.0 (2C), 128.3 (2C), 127.4, 125.8, 125.4, 123.8, 122.4, 120.2, 115.7, 115.1; HPLC: $95.39 \%$, column: X BRIDGE C-18 $150 * 4.6 \mathrm{~mm}, 5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Formic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: 0/20, 3/20, 6/98, 12/100, 18/100; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; Diluent: $\mathrm{CH}_{3} \mathrm{CN}$ UV 254 nm retention time $13.39 \mathrm{~min} ; \mathrm{IR}\left(\mathrm{KBr}, \mathrm{cm}^{-1}\right)$ : 3073, 2915, 1698, 1649, 1309, 756; MS (ES mass): $m / z 376.0$ (M+1).

## Synthesis of (9-benzoyl-9H-carbazol-3-yl)(p-tolyl)methanone (4b)

A mixture of TFAA ( 1.291 mmol ) and 4-methyl benzoic acid ( 0.369 mmol ) was stirred for 20 min until the solid was dissolved. After stirring for additional 20 min , ( 9 H -carbazol-9yl )(phenyl)methanone 3a ( 0.369 mmol ) was added in one portion. To this mixture was added $85 \% \mathrm{H}_{3} \mathrm{PO}_{4}(0.036 \mathrm{mmol})$ drop wise for a duration of 20 min . The mixture was then stirred for 2h (monitored by TLC) and the excess of TFA/TFAA was distilled out at atmospheric pressure. The remaining liquid was partitioned between ethyl acetate ( 30 mL ) and $\mathrm{H}_{2} \mathrm{O}(15 \mathrm{~mL})$. The organic layer was separated and washed with $5 \% \mathrm{NaOH}(7 \mathrm{~mL})$ and then brine ( 8 mL ). The mixture was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under vacuum. The residue was purified by column chromatography using EtOAc-n-hexane to give the desired product $\mathbf{4 b}$.


White solid; yield: 70\%; mp: 152-155 ${ }^{\circ} \mathrm{C}$; $\mathrm{R}_{f}$ (8\% EtOAc- $n$-Hexane) $0.8 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.50(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.09-8.02(\mathrm{~m}, 1 \mathrm{H}), 7.80(\mathrm{dd}, J=8.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.76(\mathrm{dd}, J$ $=7.5,3.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.69(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.60-7.49(\mathrm{~m}, 4 \mathrm{H}), 7.41-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.32(\mathrm{~d}, J=$ $7.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 196.0,169.5,143.1,141.5,139.7$, $135.2,135.1,133.0,132.9,130.2$ (2C), 129.2 (2C), 129.1, 129.0, 129.0, 128.9, 127.3, 125.8, $125.5,123.7,122.3$ (2C), 120.2, 115.7, 115.1, 21.6; HPLC: 93.00\%, column: X BRIDGE C-18 $150 * 4.6 \mathrm{~mm}, 5 \mu \mathrm{~m}$, mobile phase A: $0.1 \%$ Formic acid in water mobile phase B: $\mathrm{CH}_{3} \mathrm{CN}$ (gradient) T/\%B: $0 / 20,3 / 20,6 / 98,12 / 100,18 / 100$; flow rate: $1.0 \mathrm{~mL} / \mathrm{min}$; Diluent: $\mathrm{CH}_{3} \mathrm{CN}$ UV 254 nm retention time 13.80 min ; IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2926, 2854, 1687, 1654, 1446, 1358, 1315; MS (ES mass): m/z 390.0 (M+1); Elemental analysis found C, 83.15; H, 4.91; N, 3.73; $\mathrm{C}_{27} \mathrm{H}_{19} \mathrm{NO}_{2}$ requires C, 83.27; $\mathrm{H}, 4.92$; $\mathrm{N}, 3.60$.

## Synthesis of (9H-carbazol-3-yl)(phenyl)methanone (5a) ${ }^{5}$

To the solution of amide, $\mathbf{4 a}(0.256 \mathrm{mmol})$ in methanol $(5 \mathrm{~mL})$, $\mathrm{DBU}(0.512 \mathrm{mmol})$ was added and the reaction mixture was refluxed for 2 h . Then the reaction mixture was cooled to room temperature and solvent evaporated under vacuum. ${ }^{6}$ Water added to this reaction mixture and the formed solid was filtered to give compound 5a as Light yellow solid.

yield: $76 \%$; mp: $200-202{ }^{\circ} \mathrm{C}\left(\right.$ lit $\left.^{5 \mathrm{~b}} 203-205{ }^{\circ} \mathrm{C}\right)$; $\mathrm{R}_{f}$ ( $40 \%$ EtOAc- $n$-Hexane) $0.4 ;{ }^{1} \mathrm{H}$ NMR ( 400 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 8.60(\mathrm{~s}, 1 \mathrm{H}), 8.46(\mathrm{bs}, 1 \mathrm{H}, \mathrm{NH}), 8.10(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.00(\mathrm{~d}, J=8.3 \mathrm{~Hz}$, $1 \mathrm{H}), 7.86(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.61(\mathrm{t}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.54-7.48(\mathrm{~m}, 5 \mathrm{H}), 7.29-7.28(\mathrm{~m}, 1 \mathrm{H}) ;$ ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 196.7,142.1,140.1,139.9,138.8,131.7,129.9,129.8,129.0$,
$128.5,128.1,126.6,126.5,123.9,122.9,120.6,120.3,110.9,110.2 ;$ MS (ES mass): m/z 272.0 $(\mathrm{M}+1)$.

## Synthesis of 9H-carbazol-3-yl)(p-tolyl)methanone (5b)

To the solution of amide, $\mathbf{4 b}(0.256 \mathrm{mmol})$ in methanol $(5 \mathrm{~mL})$, DBU $(0.512 \mathrm{mmol})$ was added and the reaction mixture was refluxed for 2 h . Then the reaction mixture was cooled to room temperature and solvent evaporated under vacuum. ${ }^{6}$ Water added to this reaction mixture and the formed solid was filtered to give compound $\mathbf{5 b}$ as off white solid.


Yield: 68\%; mp: 222-224 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 8.64-8.54(\mathrm{~m}, 1 \mathrm{H}), 8.47-8.36(\mathrm{~m}$, $1 \mathrm{H}), 8.15-8.05(\mathrm{~m}, 1 \mathrm{H}), 8.03-7.94(\mathrm{~m}, 1 \mathrm{H}), 7.83-7.72(\mathrm{~m}, 2 \mathrm{H}), 7.48(\mathrm{~s}, 3 \mathrm{H}), 7.31(\mathrm{dd}, J=8.9$, $7.8 \mathrm{~Hz}, 3 \mathrm{H}$ ), $2.48(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 196.5$, 130.2 (2C), 128.8 (2C), 128.5 (2C), 126.5, 126.5, 123.8 (2C), 123.4, 122.8, 120.6, 120.3 (2C), 110.9 (2C), 110.1, 21.6; MS (ES mass): $m / z 285.0(\mathrm{M}+1)$; Elemental analysis found $\mathrm{C}, 84.41 ; \mathrm{H}, 5.32 ; \mathrm{N}, 4.67$; $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{NO}$ requires C, 84.19; H, 5.30; N, 4.91.

## Synthesis of 9-(prop-2-ynyl)-9H-carbazole (7) ${ }^{7}$

To a solution of carbazole $\mathbf{1}(10 \mathrm{mmol})$ in acetone ( 30 mL ) was added $\mathrm{K}_{2} \mathrm{CO}_{2}(20 \mathrm{mmol})$. After stirring the mixture for 15 min , 3-bromopropyne ( $80 \mathrm{wt} \%$ solution in toluene, 15 mmol ) was added drop wise for 30 min . Then the mixture was stirred for 3 h at room temperature. The mixture was filtered and the filtrate was evaporated under reduced pressure. The crude product obtained was purified by chromatography on silica gel to give 9 -(prop-2-ynyl)-9H-carbazole (7) as a white solid.


White solid; yield: $80 \%$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 8.11$ (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.64-7.44 (m, 4H), 7.41-7.17 (m, 2H), 5.06 (d, $J=2.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.26(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ) $\delta: 139.8,125.9,123.2,120.4,119.5,108.7,77.8,72.2,32.2$. MS (ES mass): $m / z 204.6$ $(\mathrm{M}+1)$.

## Synthesis of phenyl(9-(prop-2-ynyl)-9H-carbazol-3-yl)methanone (8)

A mixture of TFAA ( 1.704 mmol ) and benzoic acid $(0.487 \mathrm{mmol})$ was stirred for 20 min until the solid was dissolved. After stirring for additional 20 min the compound $7(0.487 \mathrm{mmol})$ was added in one portion. To this mixture was added $85 \% \mathrm{H}_{3} \mathrm{PO}_{4}(0.048 \mathrm{mmol})$ drop wise for a duration of 20 min . The mixture was then stirred for 2 h (monitored by TLC) and the excess of TFA/TFAA was distilled out at atmospheric pressure. The remaining liquid was partitioned between ethyl acetate $(30 \mathrm{ml})$ and $\mathrm{H}_{2} \mathrm{O}(15 \mathrm{ml})$. The organic layer was separated and washed with $5 \% \mathrm{NaOH}(7 \mathrm{ml})$ and then brine ( 8 ml ). The mixture was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under vacuum. The residue was purified by column chromatography using EtOAc-n-hexane to give the desired product 8.


White soild; yield: $70 \%$; mp: $168-170{ }^{\circ} \mathrm{C} ; \mathrm{R}_{f}\left(10 \%\right.$ EtOAc- $n$-Hexane) $0.62 ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta: 8.63(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.14(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.08(\mathrm{dd}, J=8.4,1.6 \mathrm{~Hz}, 1 \mathrm{H})$, 7.87-7.85 (m, 2H), $7.64(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.57-7.52(\mathrm{~m}, 5 \mathrm{H}), 7.36-7.32(\mathrm{~m}, 1 \mathrm{H}), 5.11(\mathrm{~d}, J=$ $2.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.33(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 196.5,142.3,140.4,138.8$, 131.7, 129.9 (2C), 129.2, 128.6, 128.2 (2C), 126.7, 123.8, 123.3, 122.9, 120.8, 120.6, 109.2, 108.3, 77.3, 72.8, 32.5; MS (ES mass): $m / z 310.2$ (M+1); Elemental analysis found C, 85.60; H, 4.87; N, 4.41; $\mathrm{C}_{22} \mathrm{H}_{15} \mathrm{NO}$ requires C, 85.41; H, 4.89; N, 4.53.

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## Cell Proliferation Assay

The anti-proliferative activity and cancer cell selectivity of the synthesized compounds on normal and cancer cells was evaluated using the SRB (Sulforhodamine B) cell proliferation assay. This assay was chosen because of its sensitivity, large dynamic range and the ability to measure cell proliferation over three days with normalization to initial cell number as well as to vehicle-treated cells. Further, this assay is the standardized assay of choice for anticancer compound screening at the National Cancer Institute (NIH). The SRB assay provides a colorimetric readout which can be spectrophotometrically measured and does not involve antibodies or toxic reagents. The assay is based on detection of total protein content of cells, which increases or decreases in proportion with cell number.

In brief, the assay was performed as follows: Cancer(Cal 27(oral cancer cell line) and MDAMB231(breast cancer cell line)) and non-cancer (Human Embryonic Kidney (HEK) 293T cell line) cells were seeded in 96 -well plates and incubated overnight. The optimum cell numbers to be seeded were determined by a growth curve analysis for each cell line. In the initial (single dose) screen, compounds (dissolved in $100 \%$ DMSO to a stock concentration of 100 mM ) were added to the adhered cells at a final concentration of 10 uM . After 72 h of treatment, the cells were washed with phosphate-buffered saline and ice-cold $10 \%$ trichloroacetic acid added to the cells to precipitate all proteins for 1 h at $4^{\circ} \mathrm{C}$. The cells were then washed with water and airdried. Cellular proteins were then stained using $0.4 \%$ SRB solution in $1 \%$ acetic acid for 10 min at room temperature. The unbound dye was washed away by destaining with $1 \%$ acetic acid and
bound dye solubilized with 10 mM Tris solution. Absorbance of solubilized dye was measured at a wavelength of 590 nm . Percentage growth was determined by the formula [(At-A0/Ac-A0)] X 100 , where $\mathrm{At}=$ absorbance after 72 h of test compound treatment, $\mathrm{A} 0=$ Absorbance at time 0 , $\mathrm{Ac}=$ Absorbance after 72 h without treatment. (Compounds which resulted in $<50 \%$ growth of cancer cells were considered potentially anti-proliferative. Among such compounds, those which retained $>75 \%$ growth of non-cancerous cells were considered potentially selective to cancer cells). The Graphpad Prism 5 Demo software was used to generate the dose response curve using the sequence: select table and graph - enter data - analyze - transform data - nonlinear regression - inhibition vs response - results.

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