

Supporting information

Divergent total synthesis of 1,6,8a-tri-*epi*-castanospermine and 1-deoxy-6,8a-di-*epi*-castanospermine from substituted azetidin-2-one (β -lactam), involving a cascade sequence of reactions as key step

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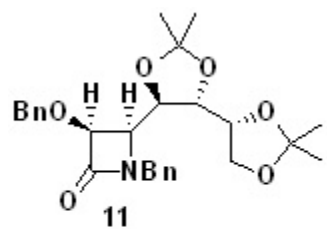
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^b Organic Chemistry Division, National Chemical Laboratory, Homi Bhabha Road, Pune, 411008, India

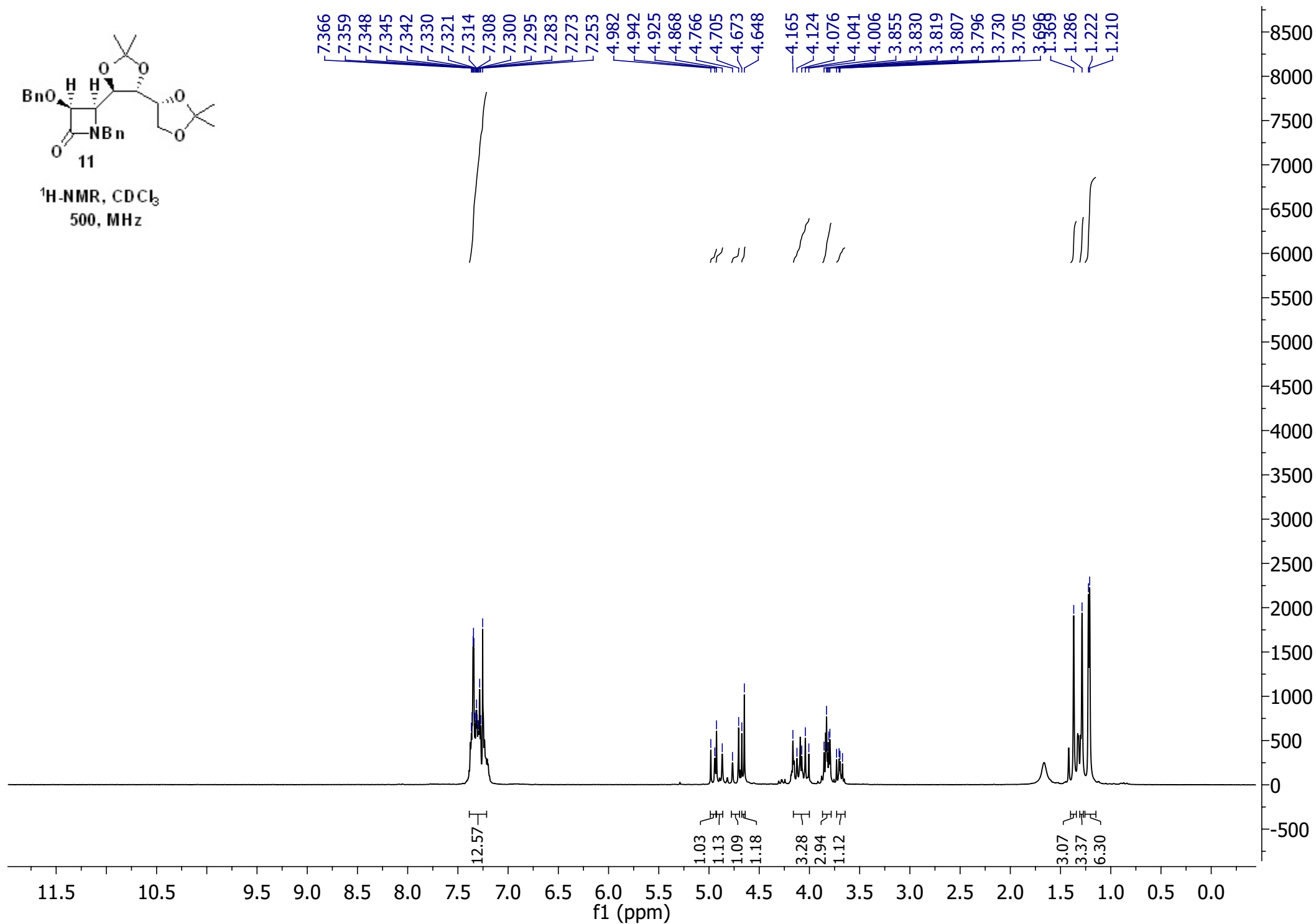
^c Centre for Materials Characterization, National Chemical Laboratory, Homi Bhabha Road, Pune, 411008, India

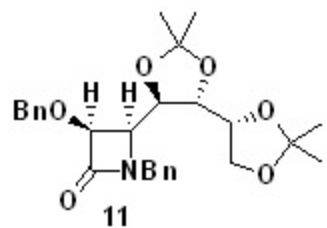
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$^1\text{H-NMR}$, CDCl_3
500, MHz





$^{13}\text{C-NMR}$, CDCl_3
125 MHz

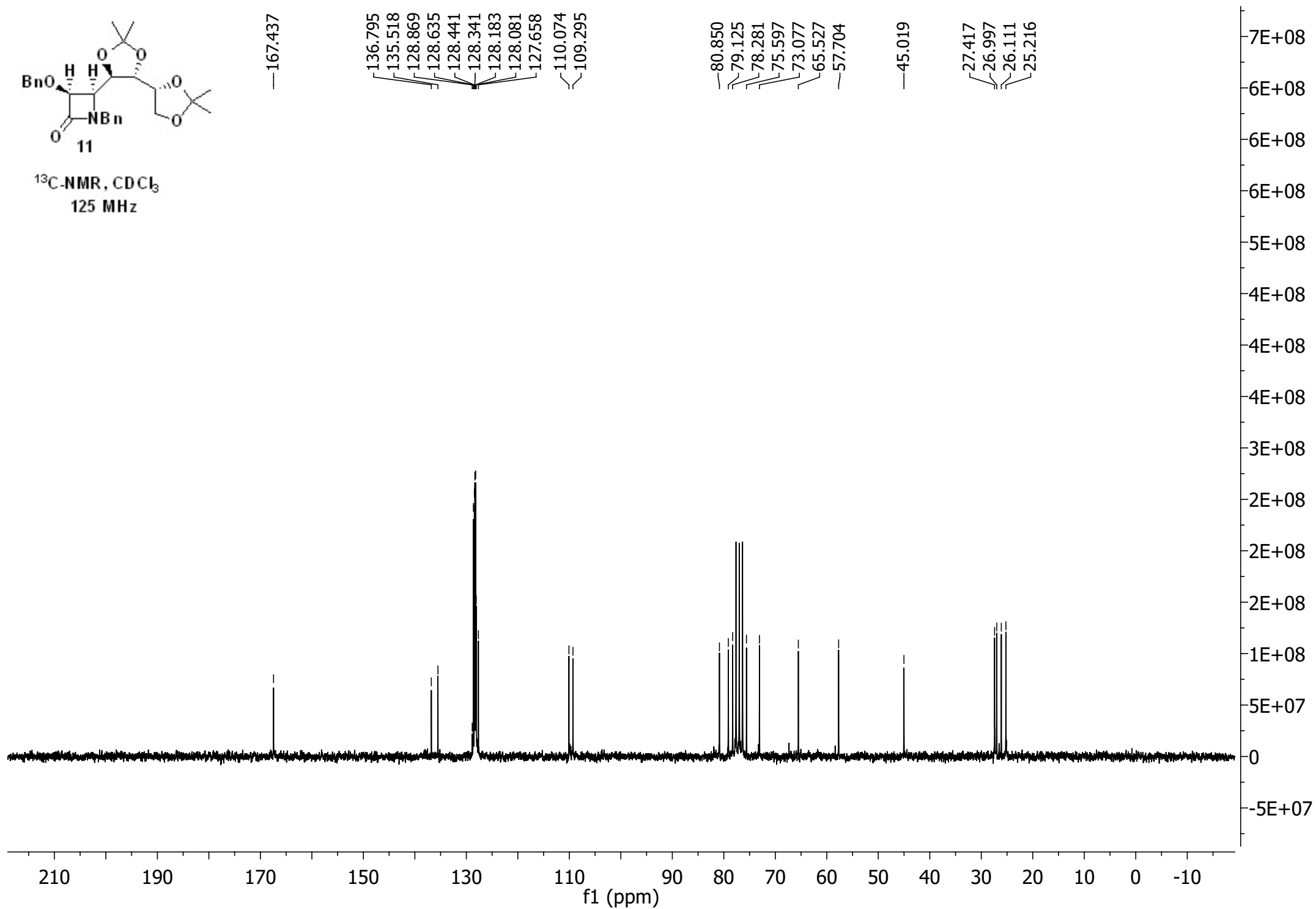
—167.437

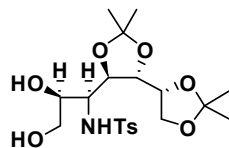
136.795
135.518
128.869
128.635
128.441
128.341
128.183
128.081
127.658
110.074
109.295

80.850
79.125
78.281
75.597
73.077
65.527
57.704

—45.019

27.417
26.997
26.111
25.216

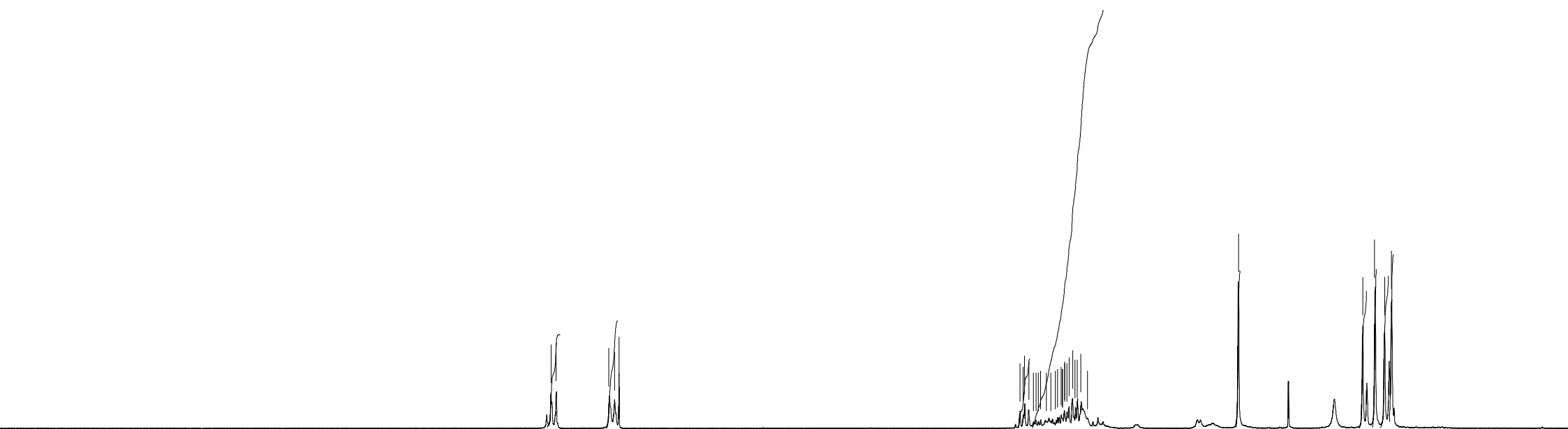




13
¹H-NMR, CDCl₃
 500, MHz

7.783
 7.742
 7.329
 7.286
 7.252

4.126
 4.101
 4.090
 4.059
 3.833
 3.805
 3.795
 3.795
 3.780
 3.780
 3.759
 3.745
 3.717
 3.693
 3.678
 3.648
 3.599
 2.423
 1.455
 1.358
 1.283
 1.229
 4.021
 4.002
 3.984
 3.964
 3.918
 3.887
 3.845

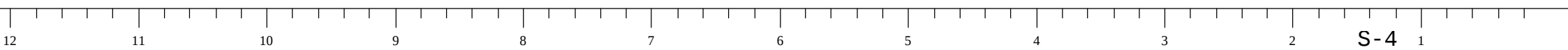


1.8
 2.0

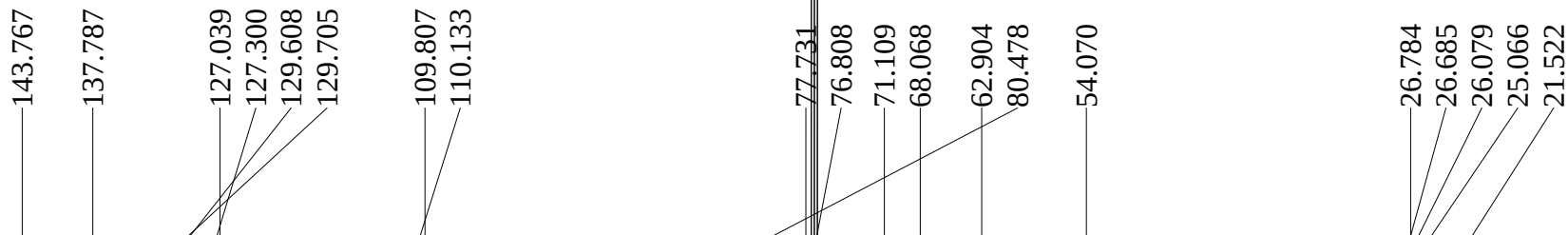
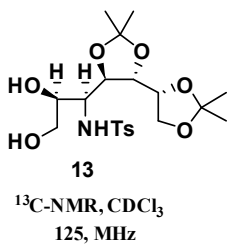
1.3
 8.0

3.0

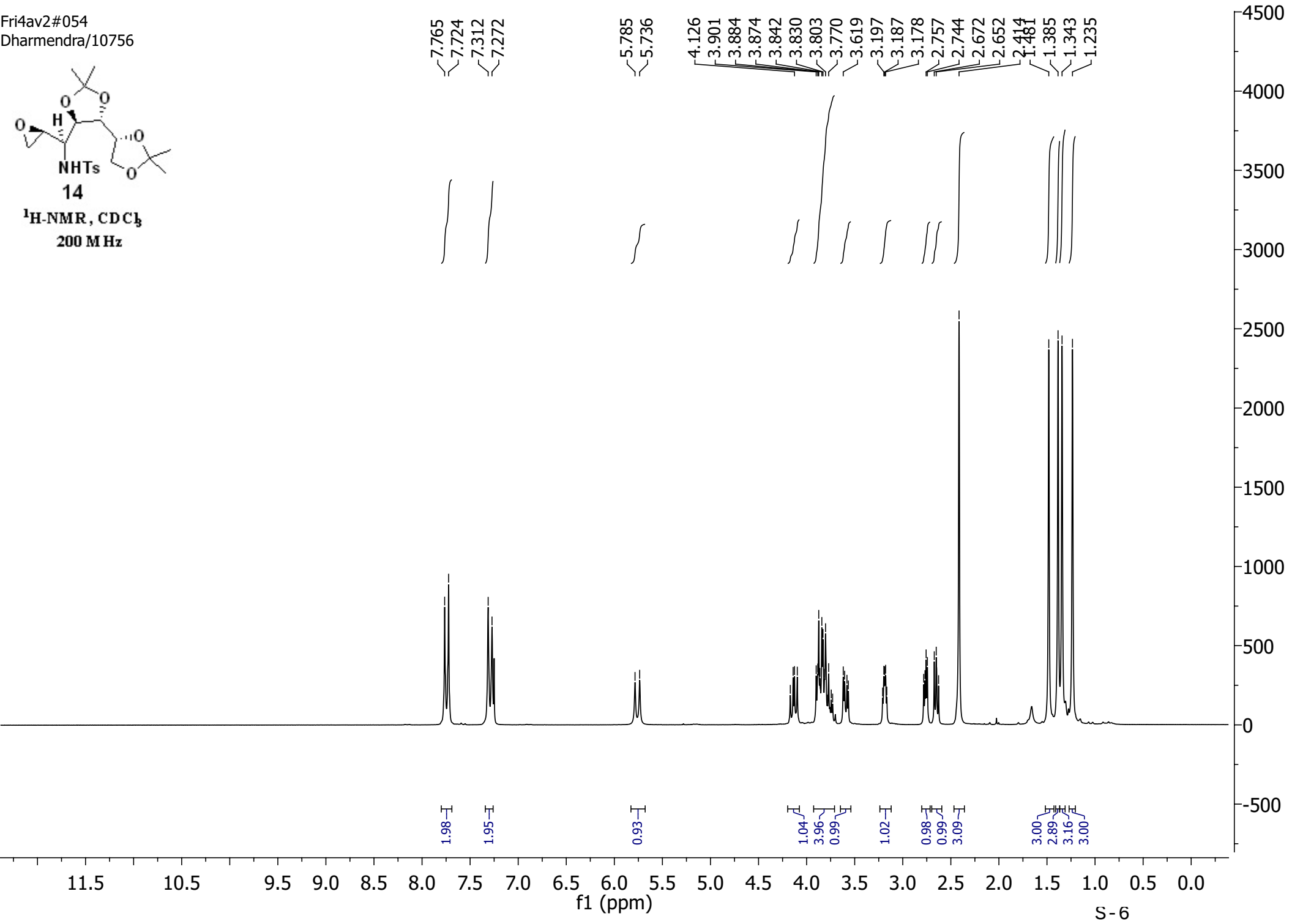
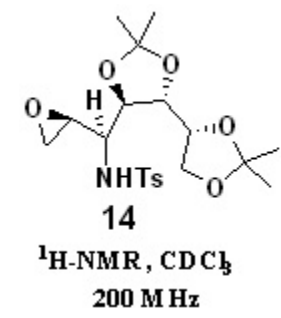
2.6
 3.3
 3.3



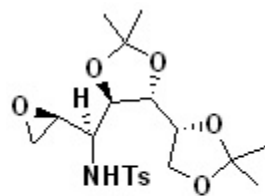
S-4



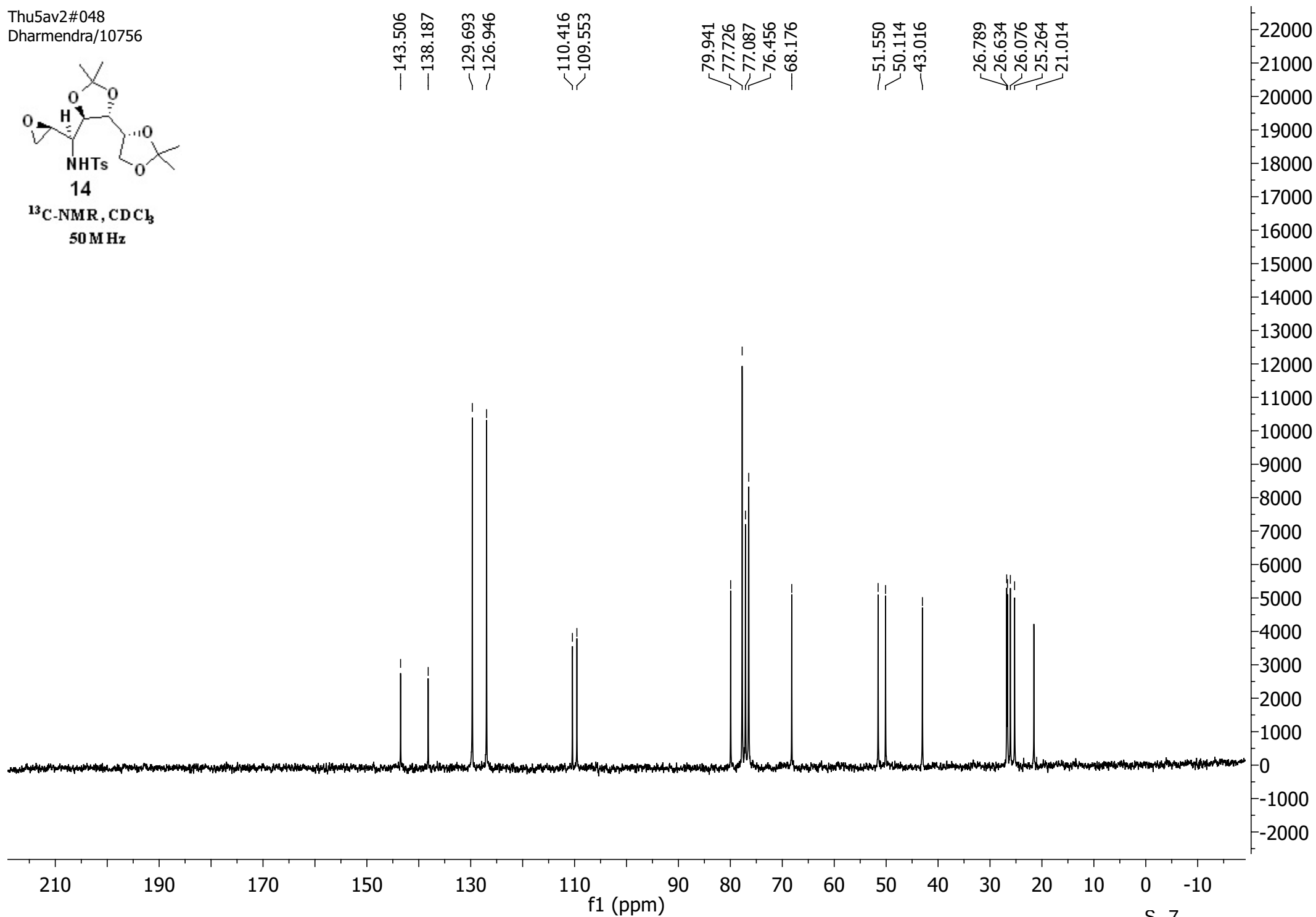
Fri4av2#054
Dharmendra/10756



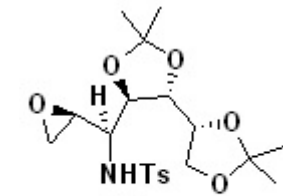
Thu5av2#048
Dharmendra/10756



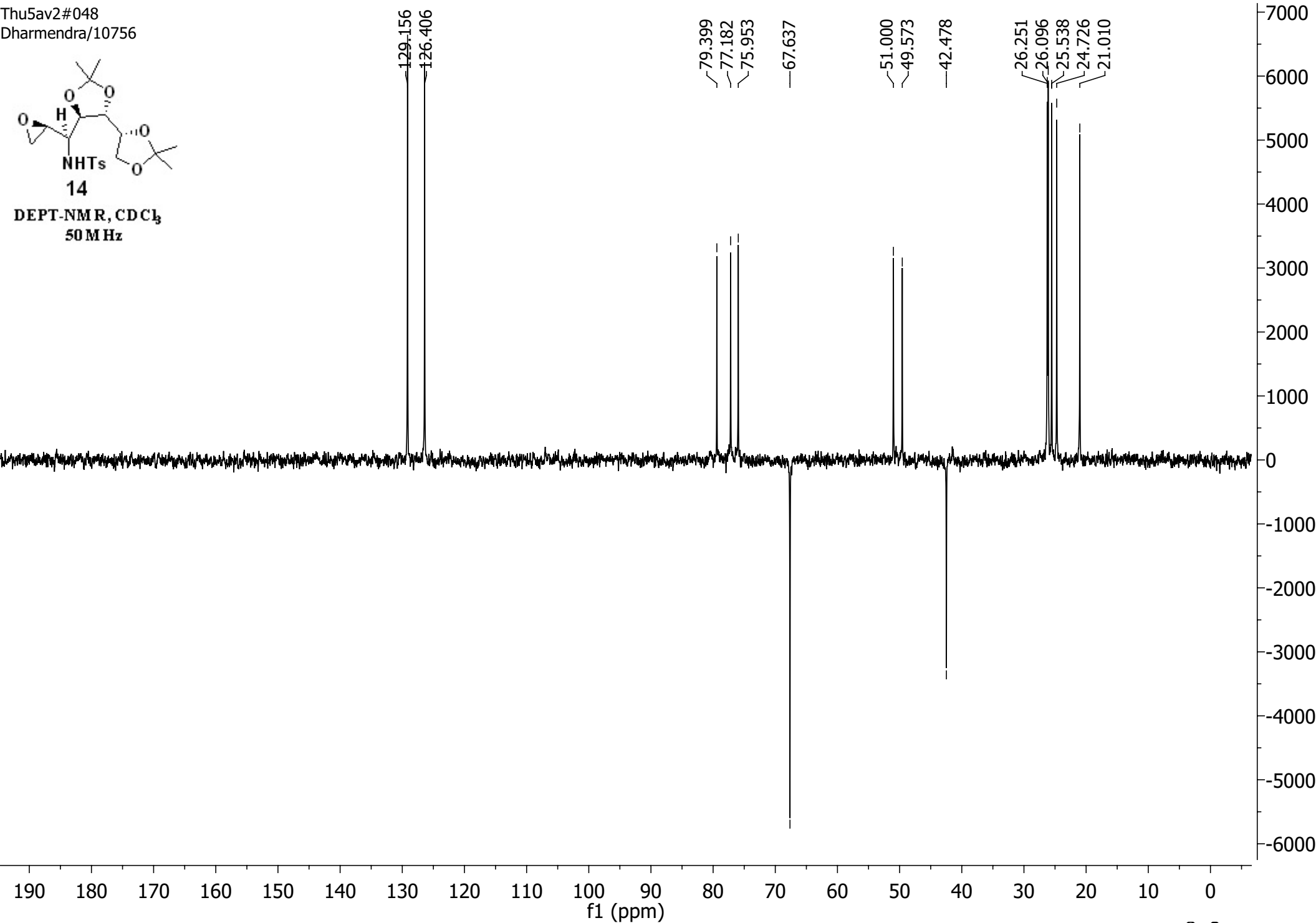
14
¹³C-NMR, CDCl₃
 50 MHz

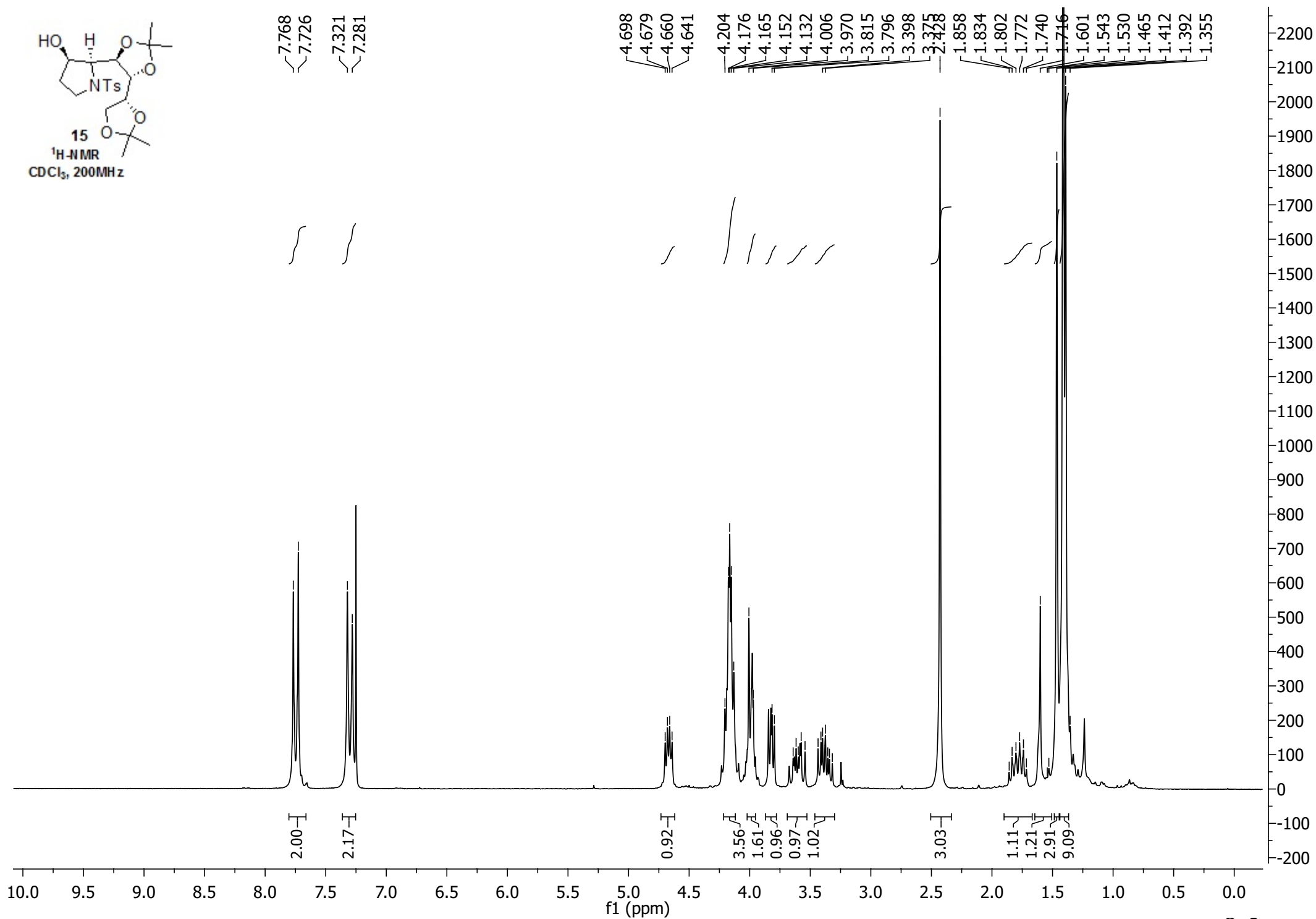
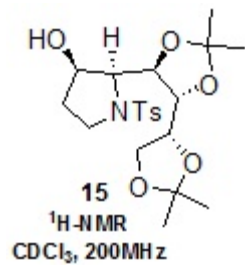


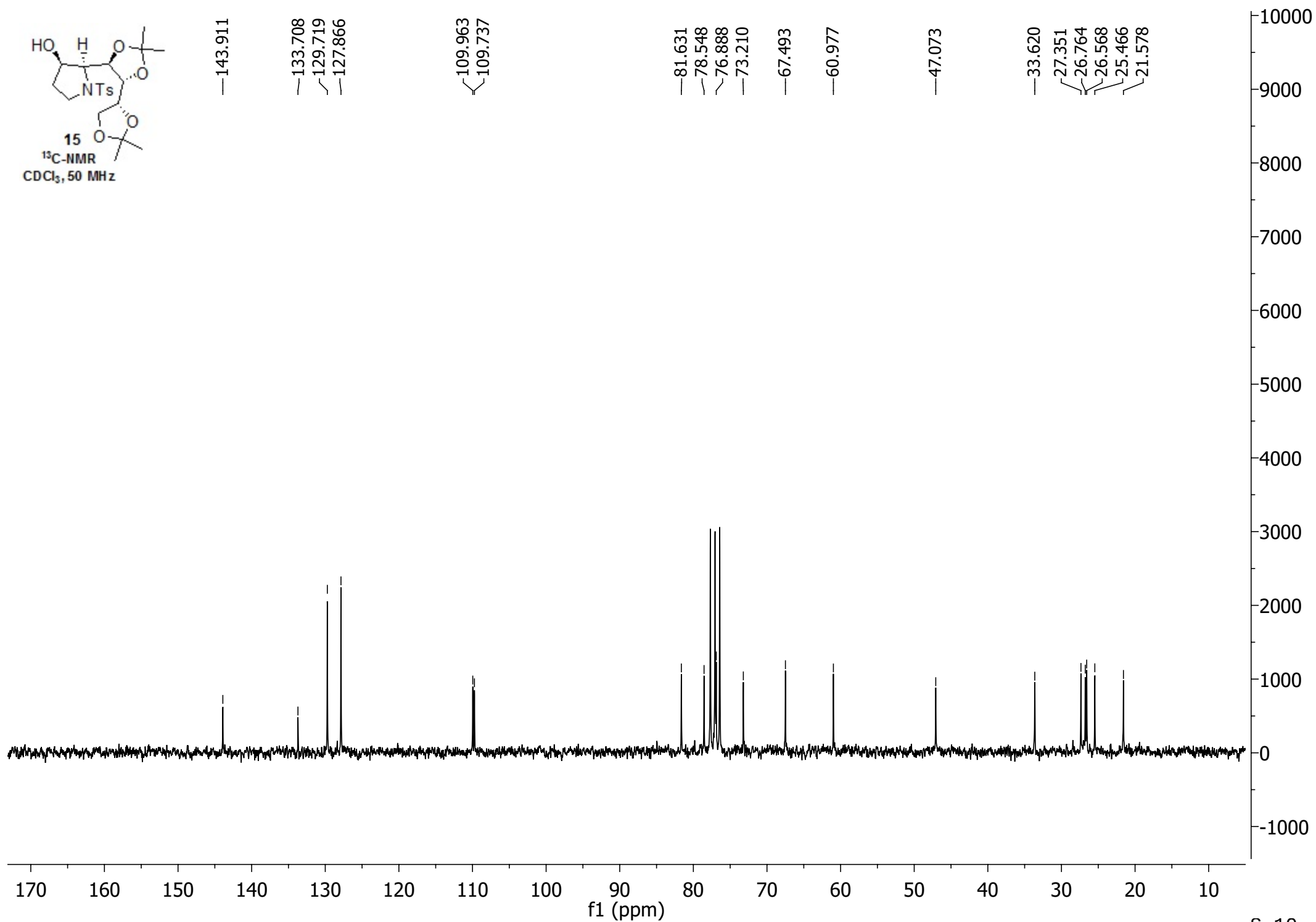
Thu5av2#048
Dharmendra/10756

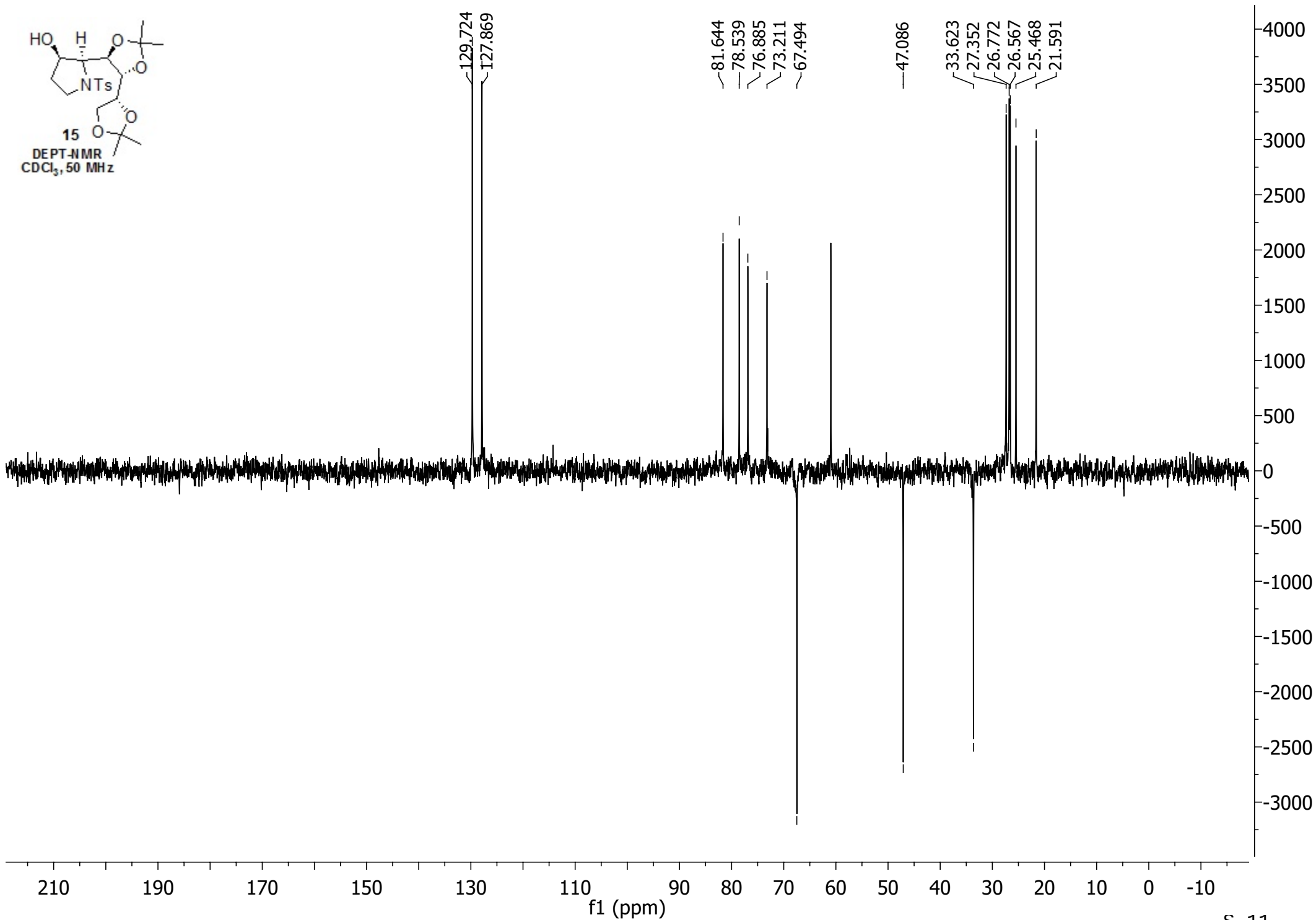


14
DEPT-NMR, CDCl₃
50 MHz

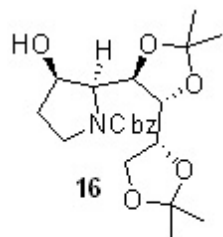
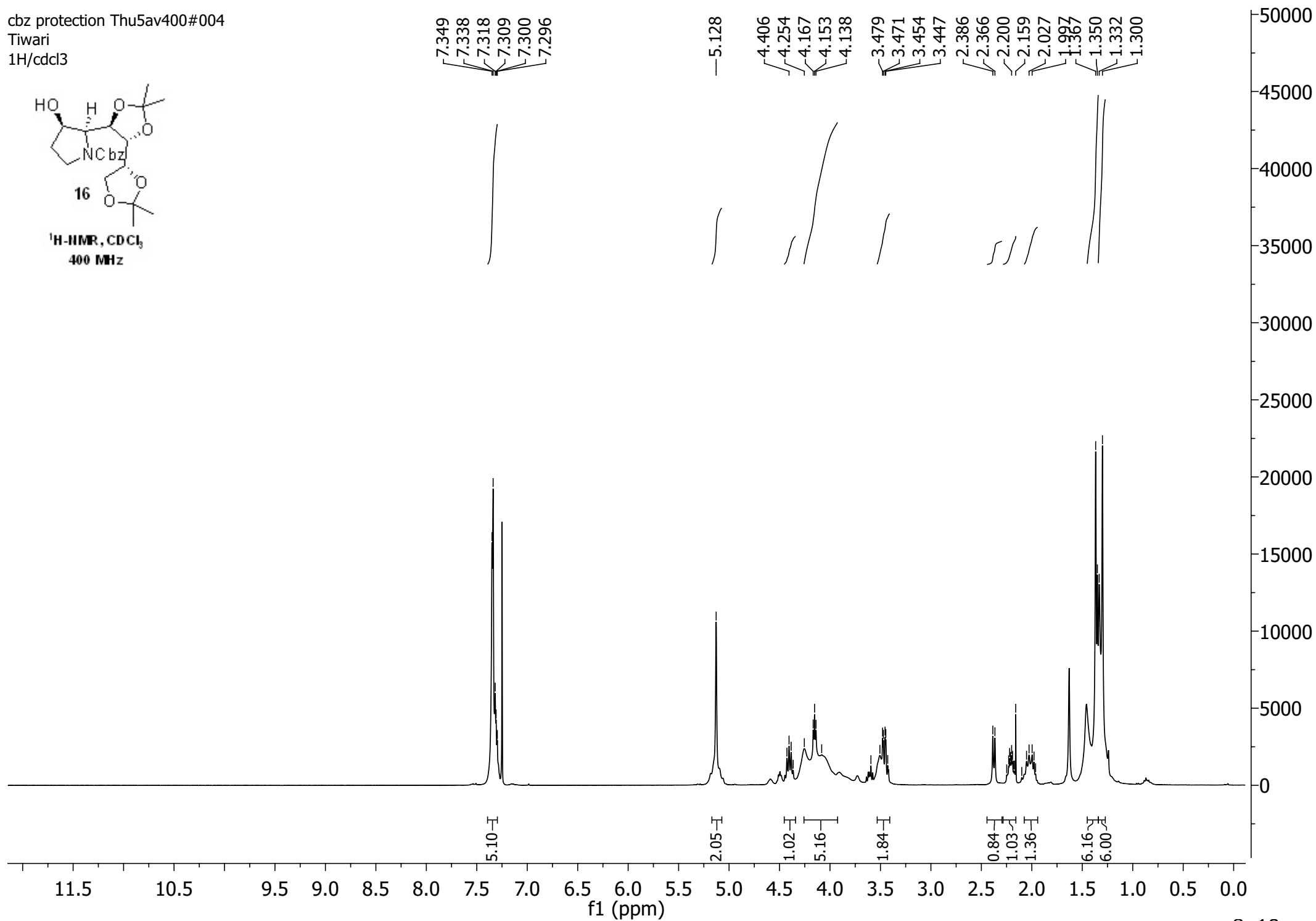


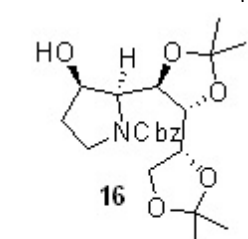
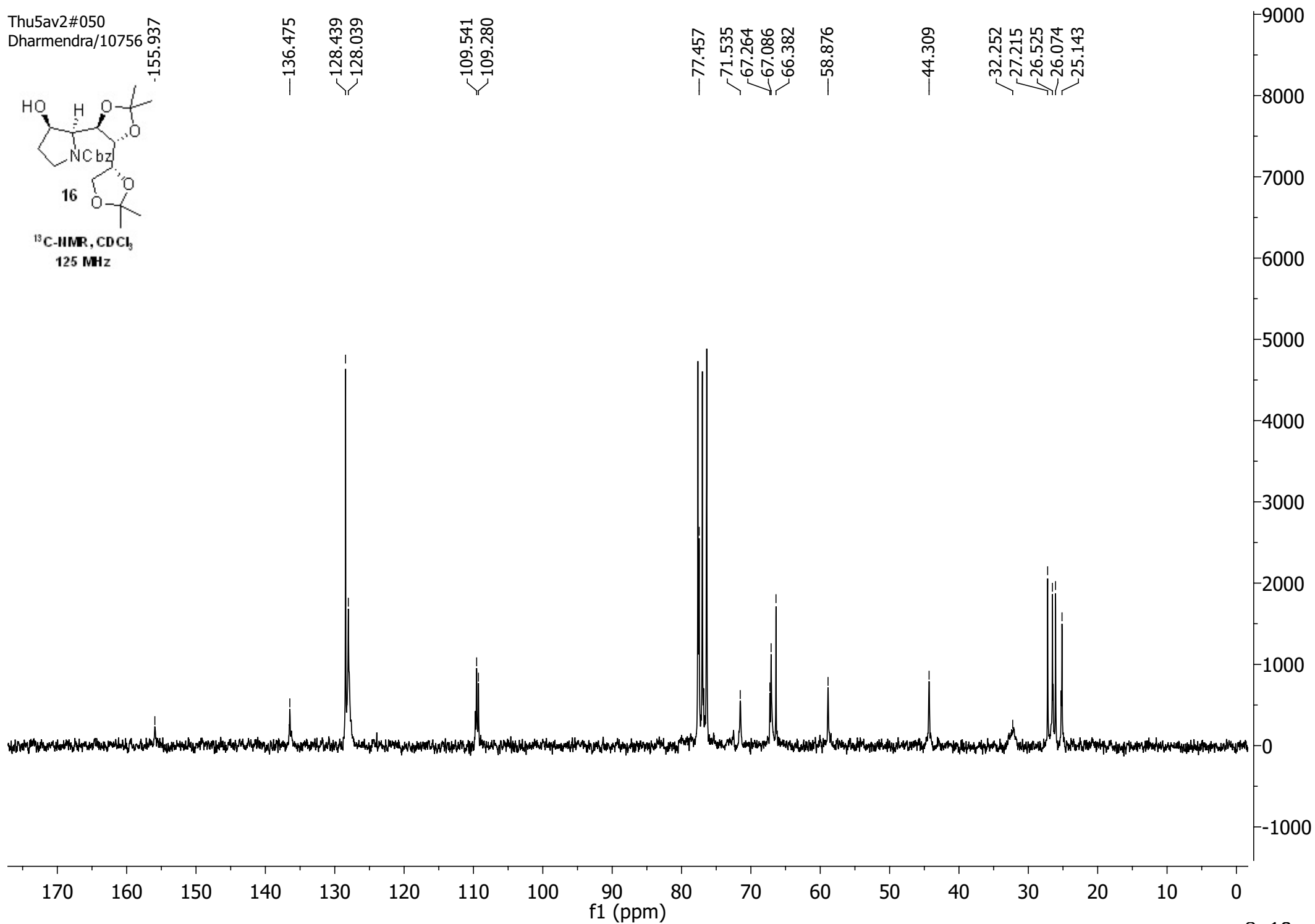




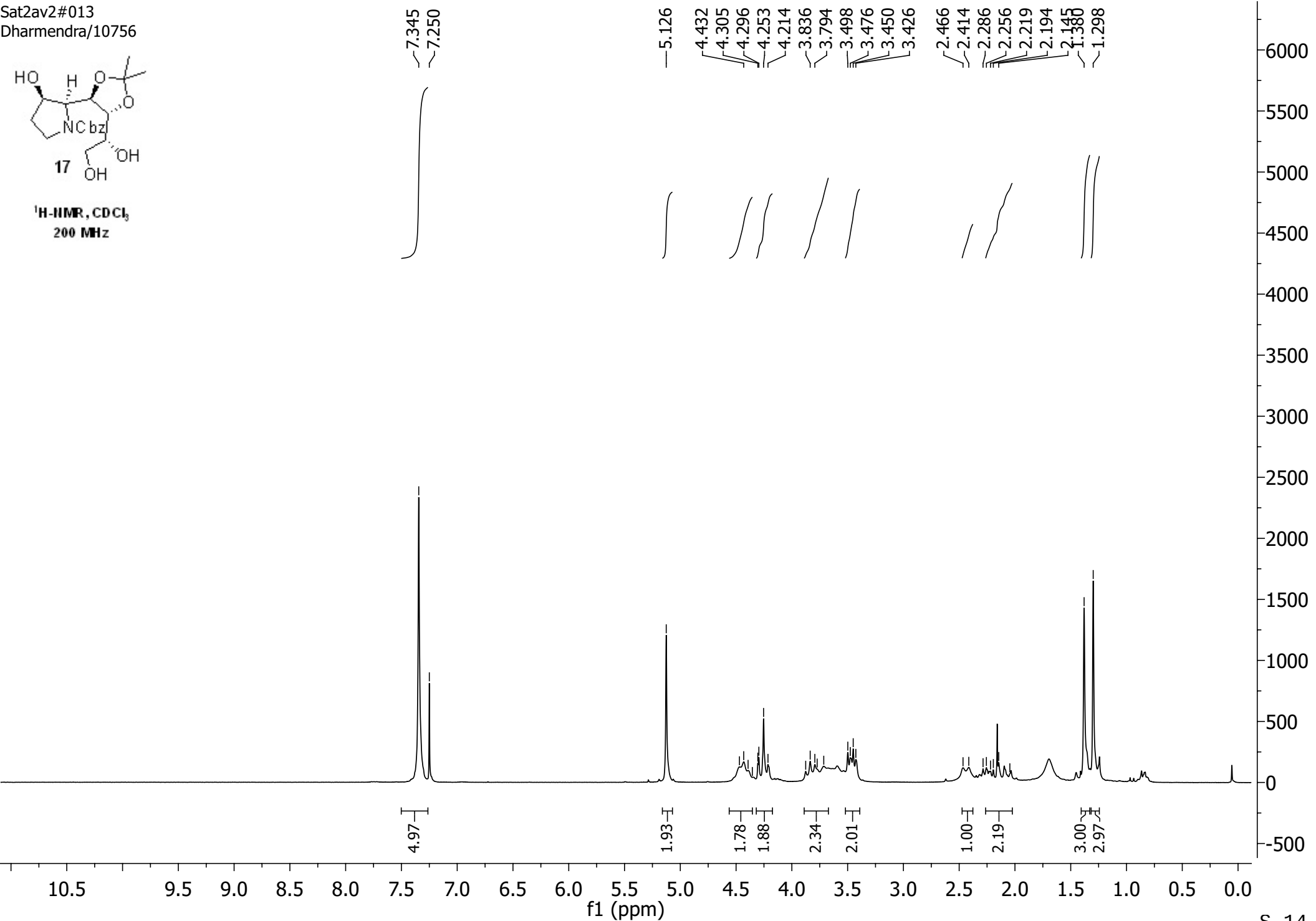
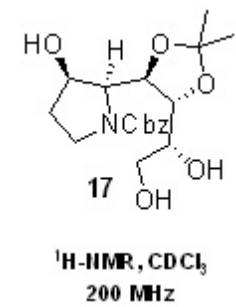


1H/cdcl3

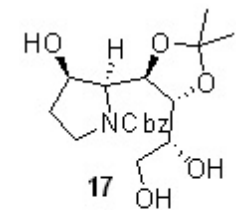
¹H-NMR, CDCl₃
400 MHz

 $^{13}\text{C-NMR}$, CDCl_3
125 MHz

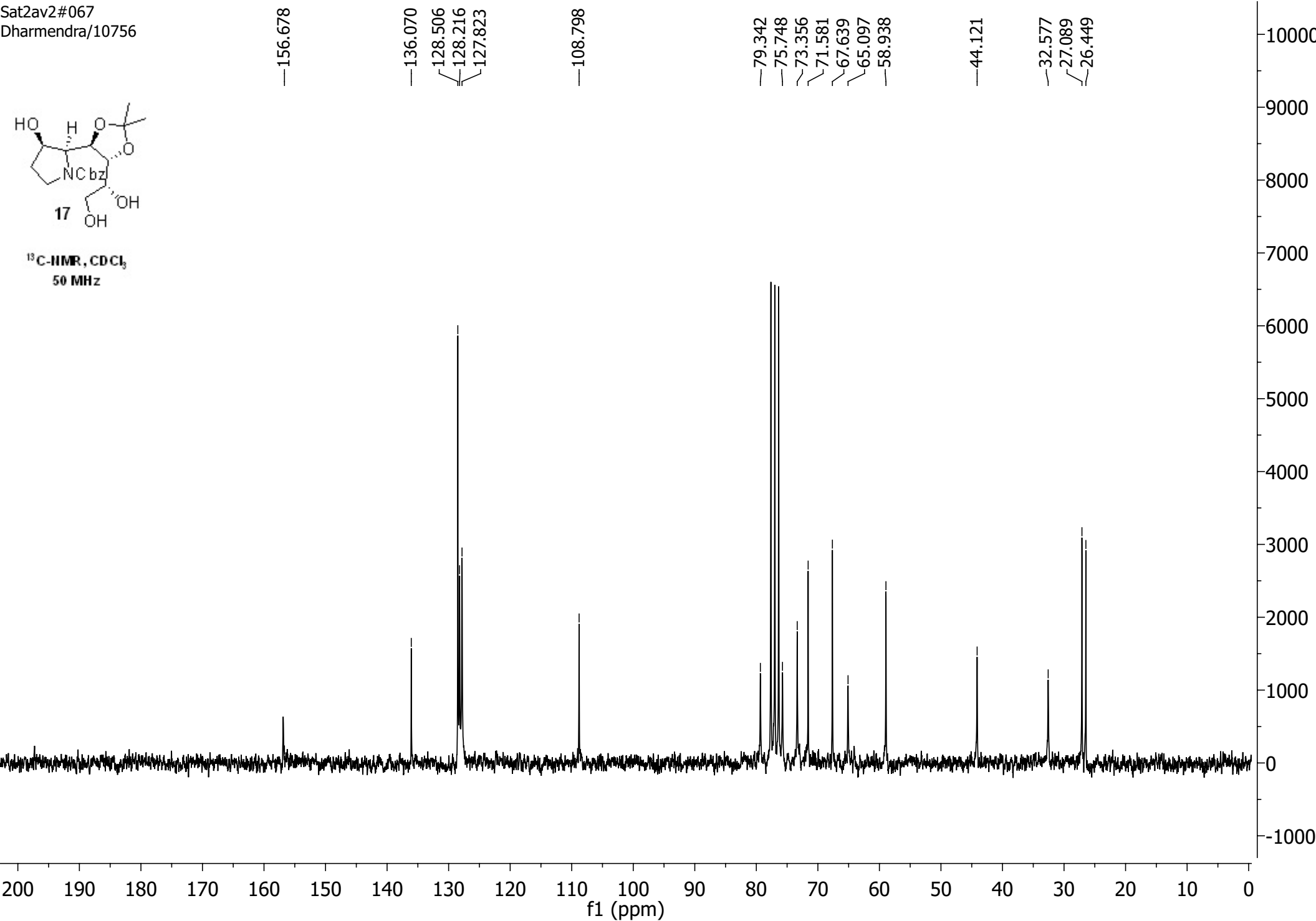
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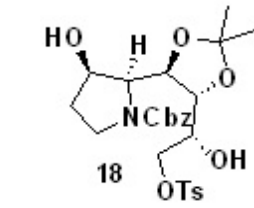


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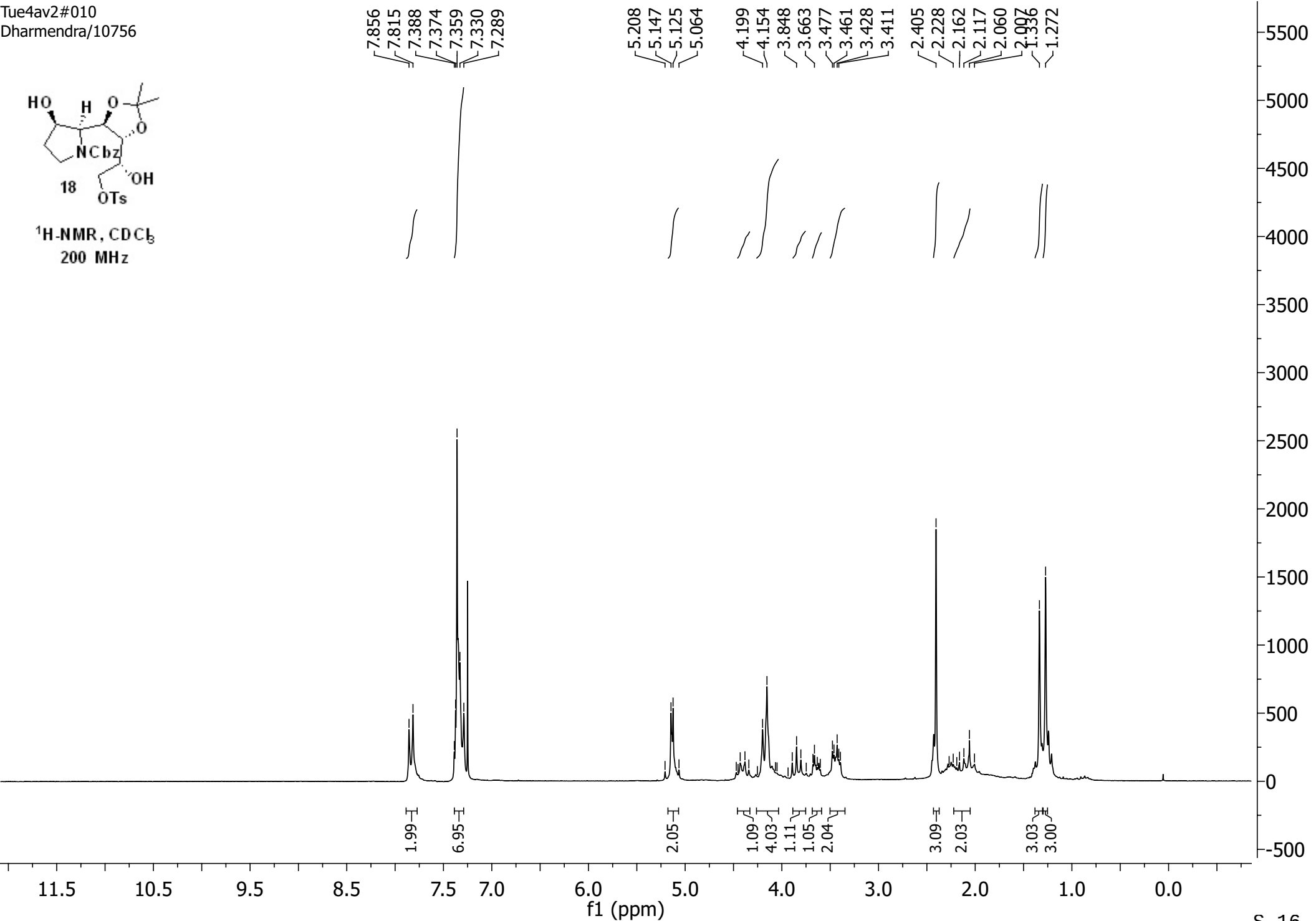


¹³C-NMR, CDCl₃
50 MHz



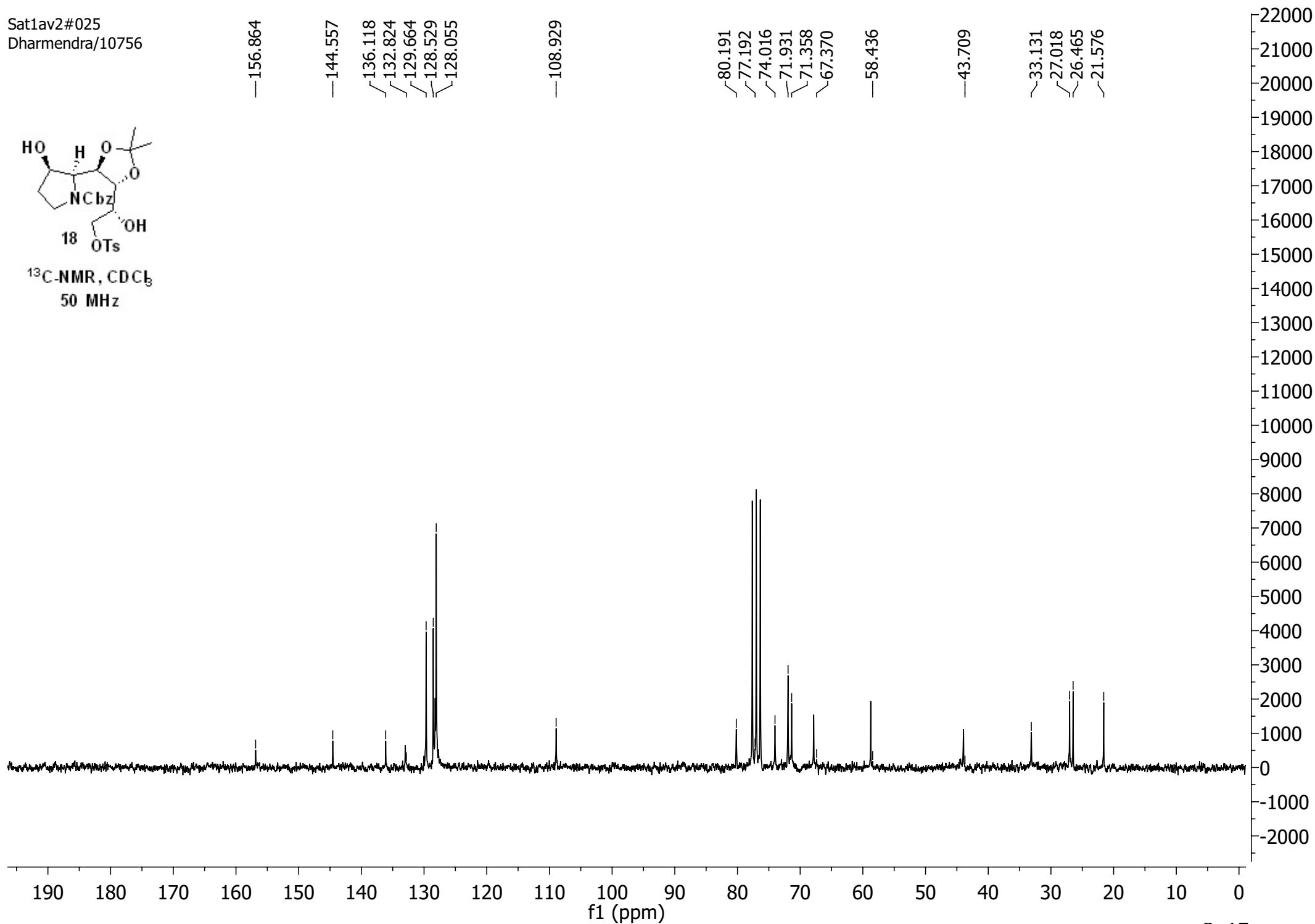


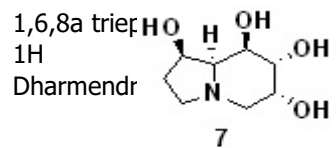
¹H-NMR, CDCl₃
200 MHz



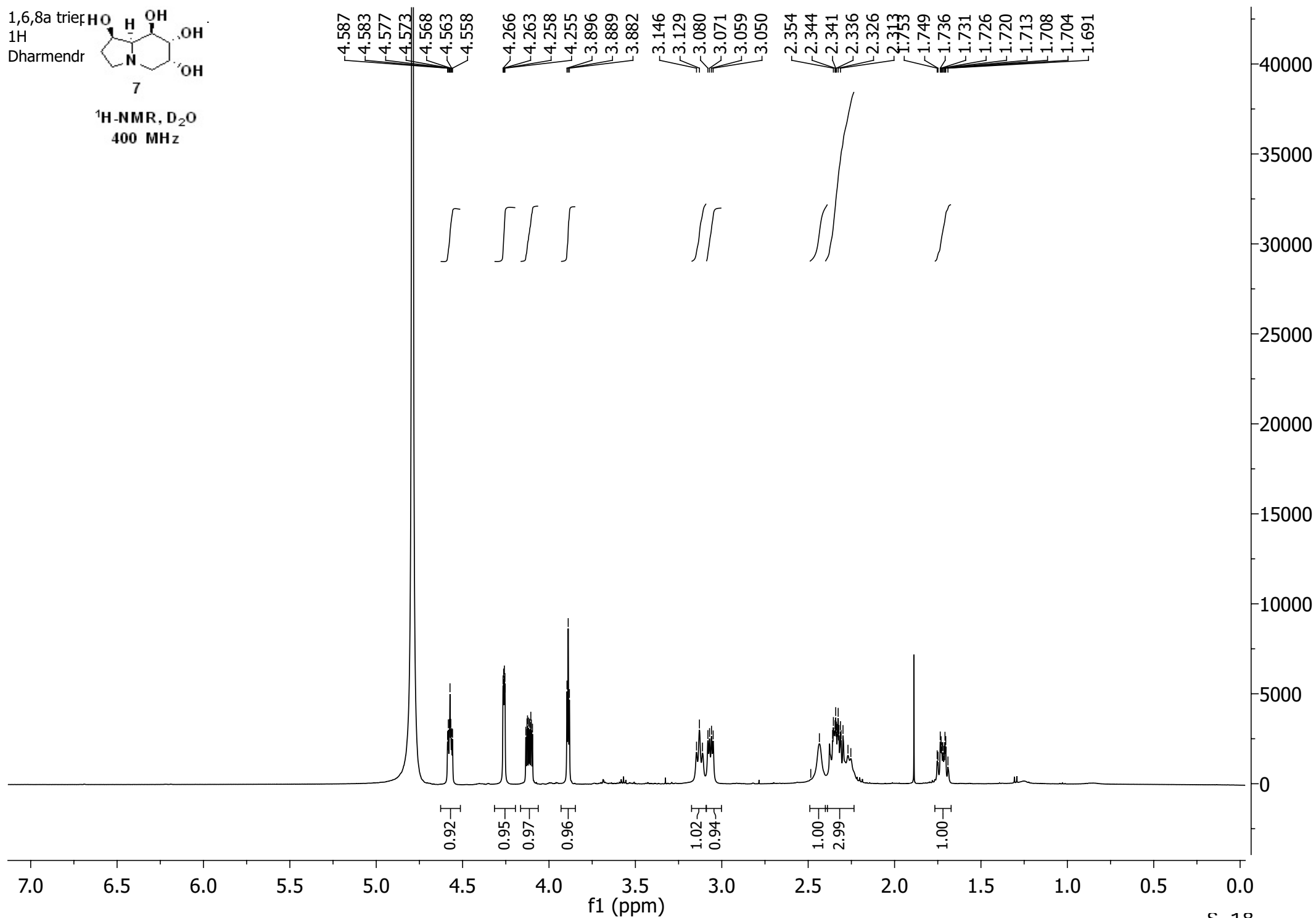


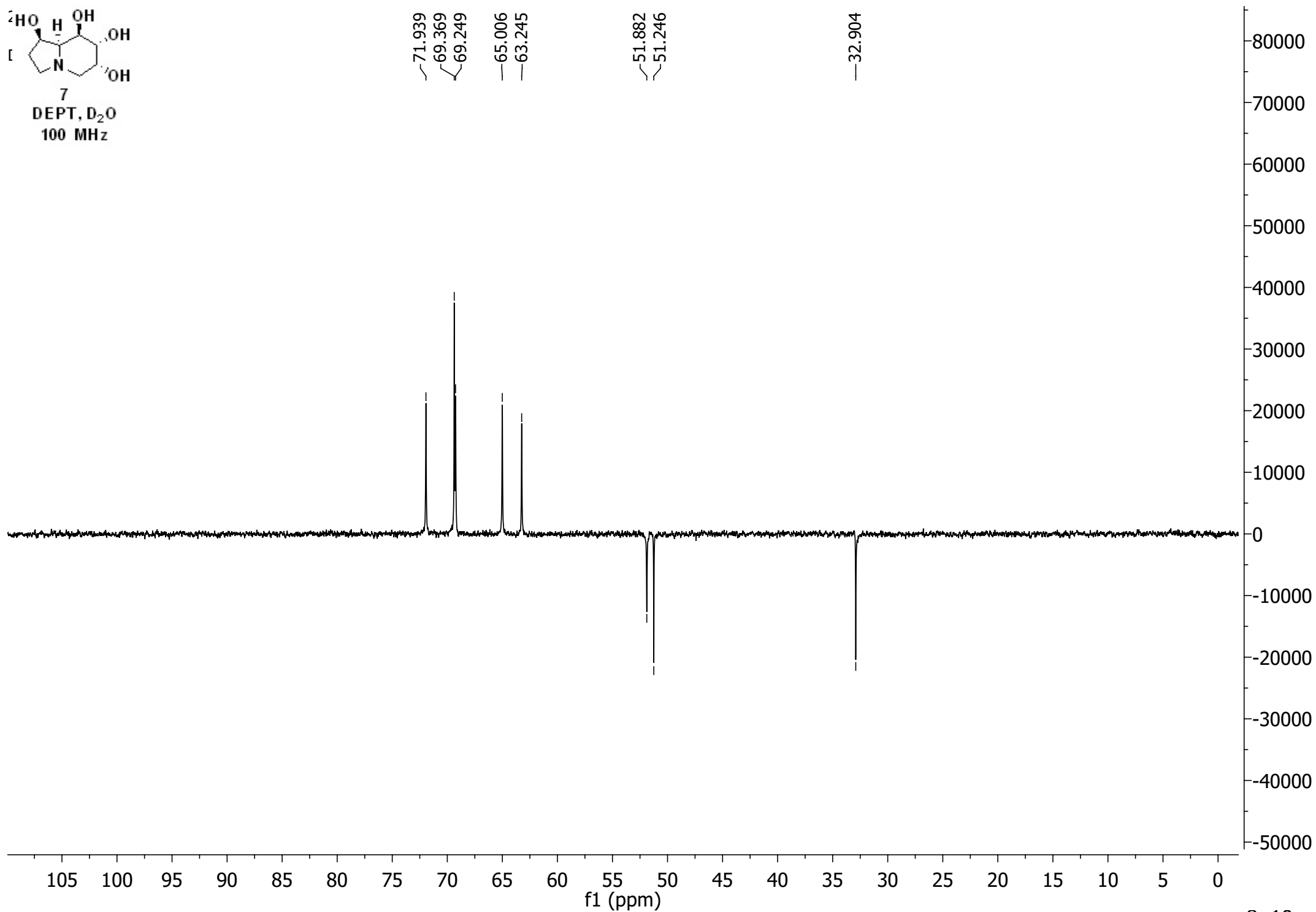
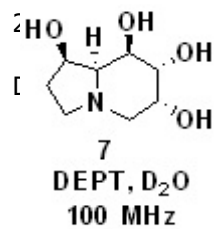
¹³C-NMR, CDCl₃
 50 MHz

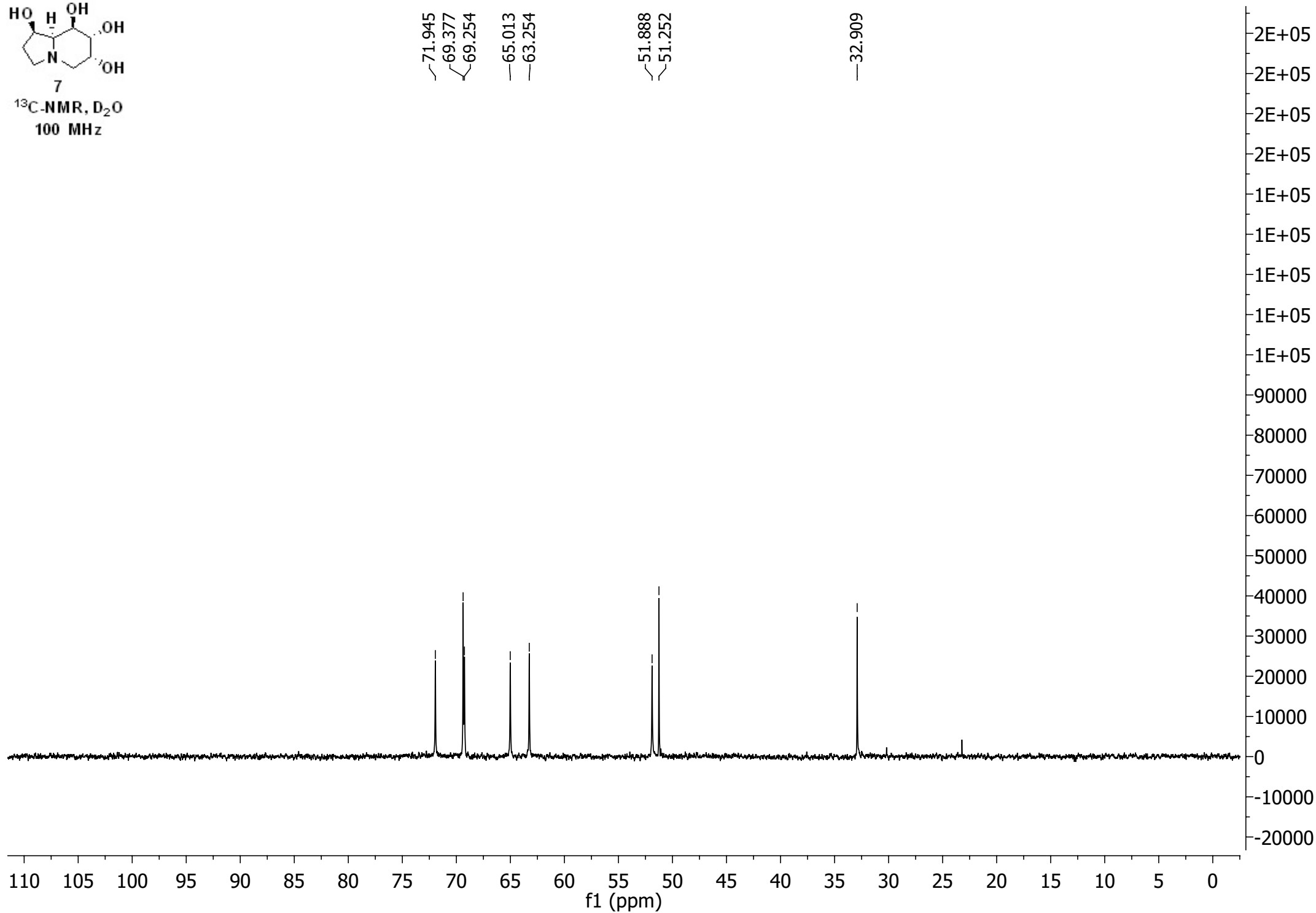
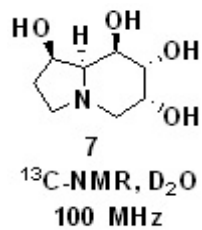


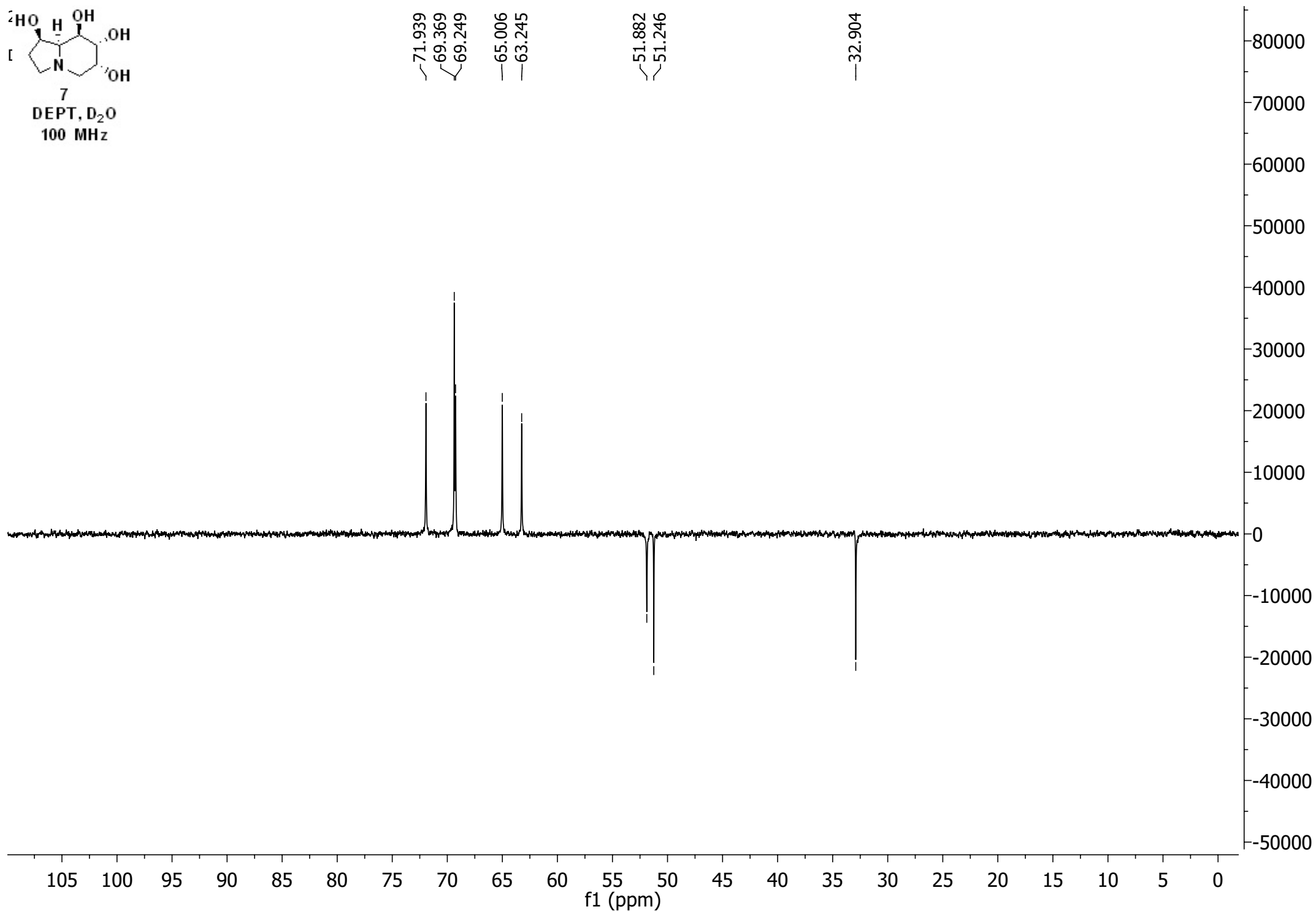
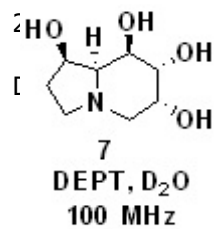


¹H-NMR, D₂O
400 MHz

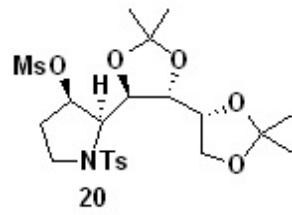




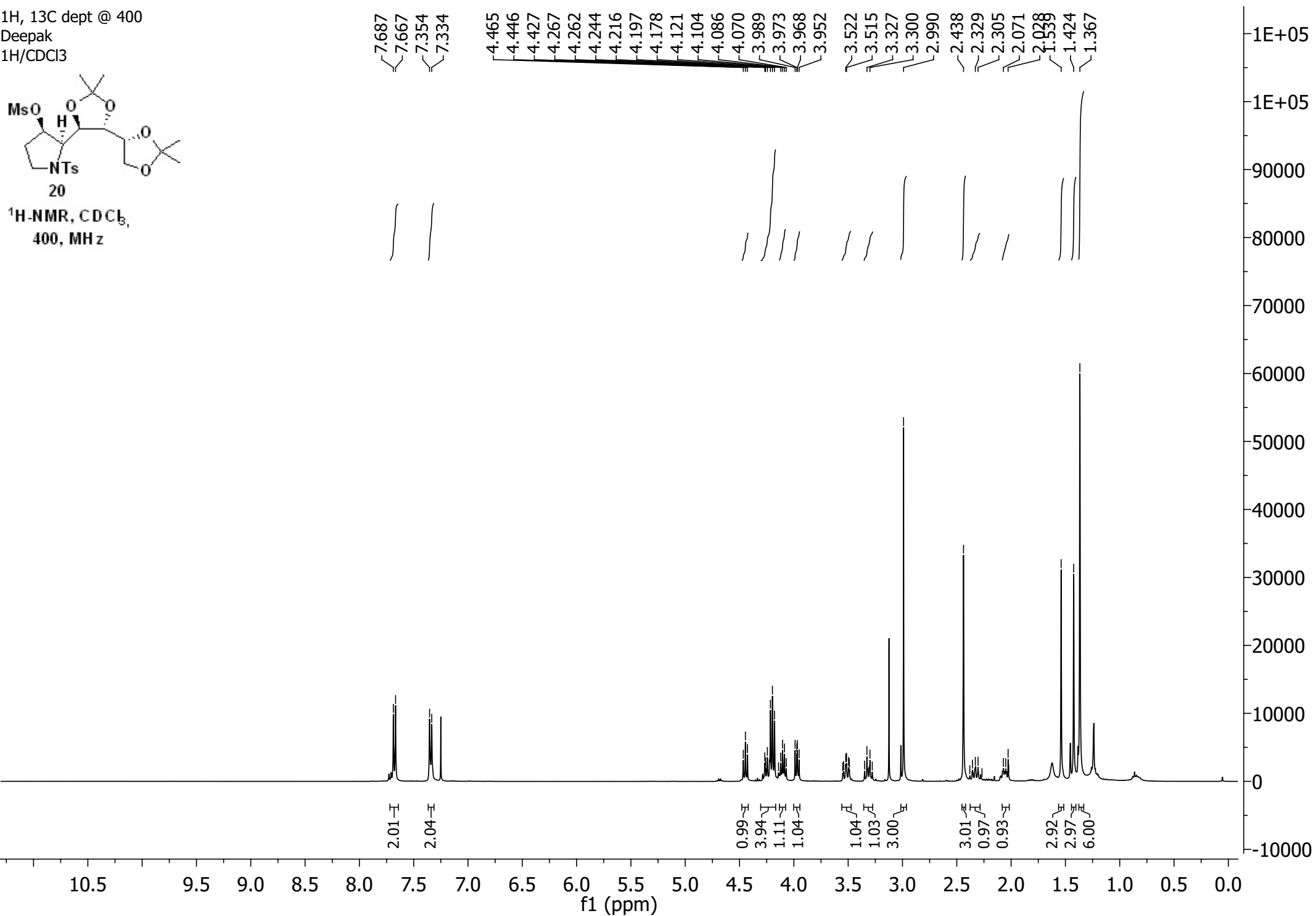


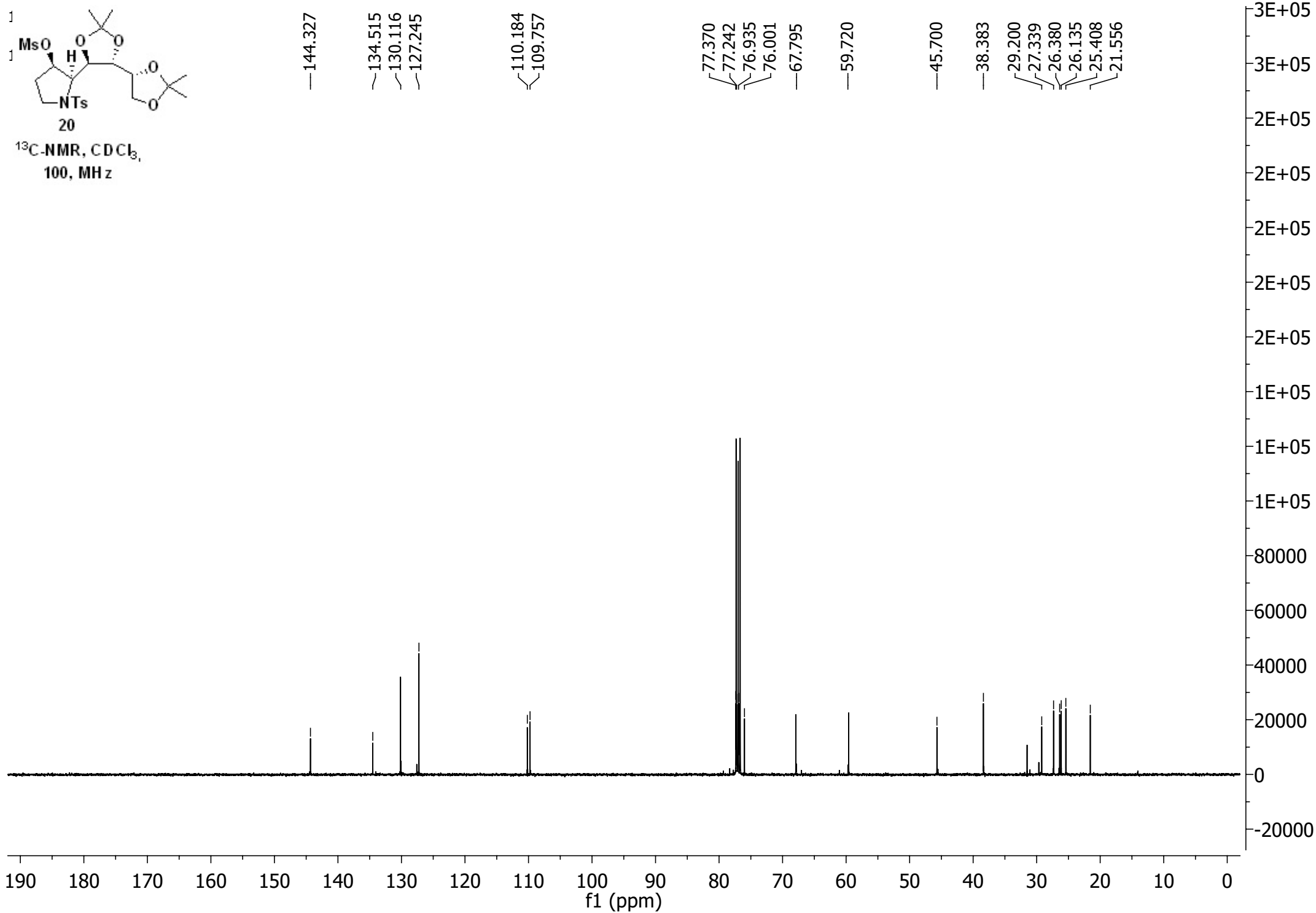


¹H, ¹³C dept @ 400
Deepak
¹H/CDCl₃

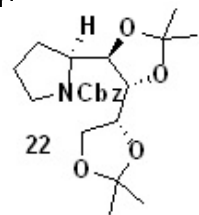


¹H-NMR, CDCl₃,
400, MHz

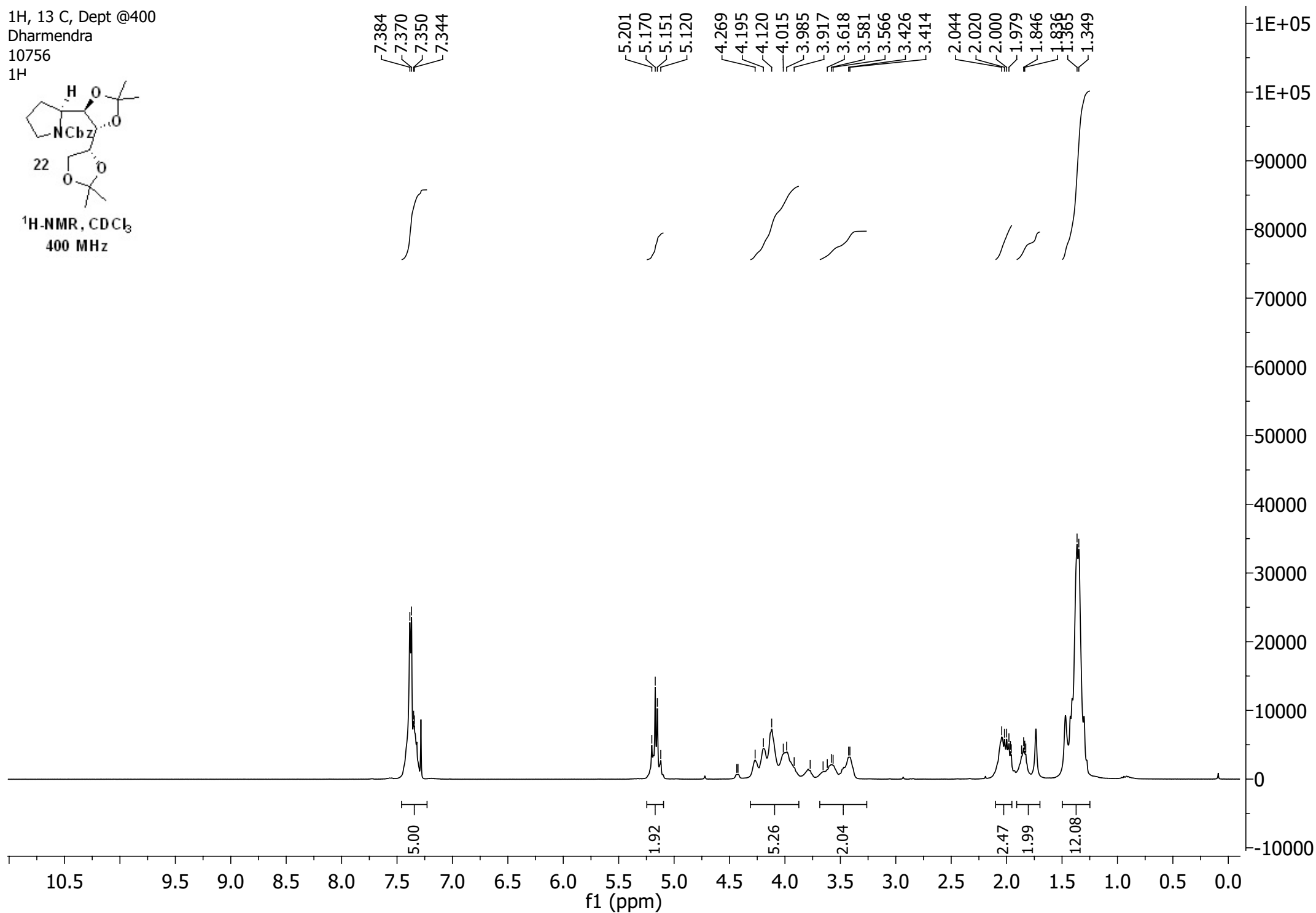




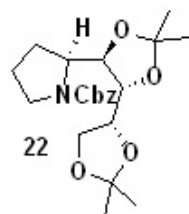
¹H, ¹³C, Dept @400
Dharmendra
10756
¹H



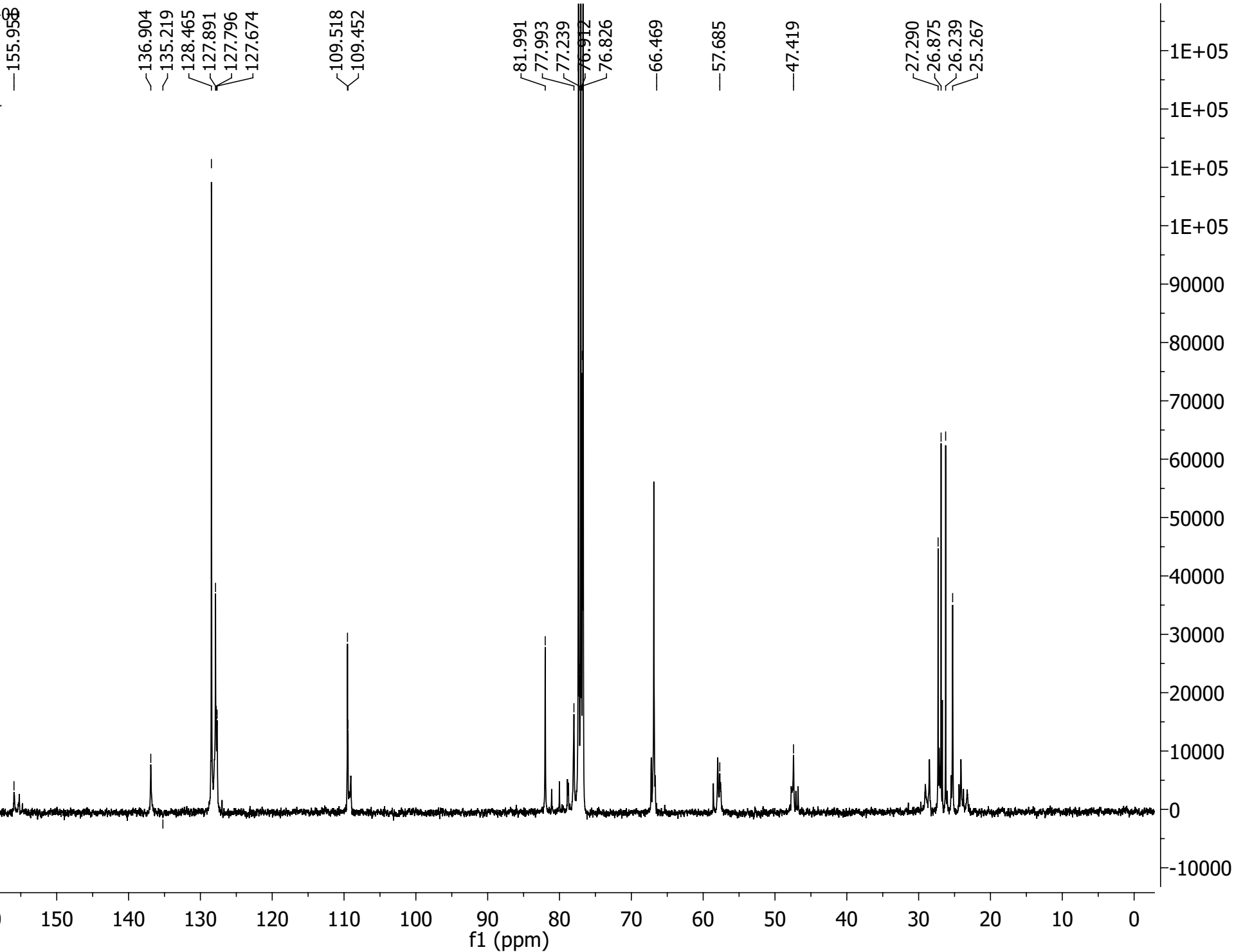
¹H-NMR, CDCl₃
400 MHz



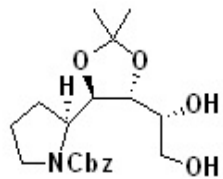
¹H, ¹³C, Dept @400



¹³C-NMR, CDCl₃
100 MHz

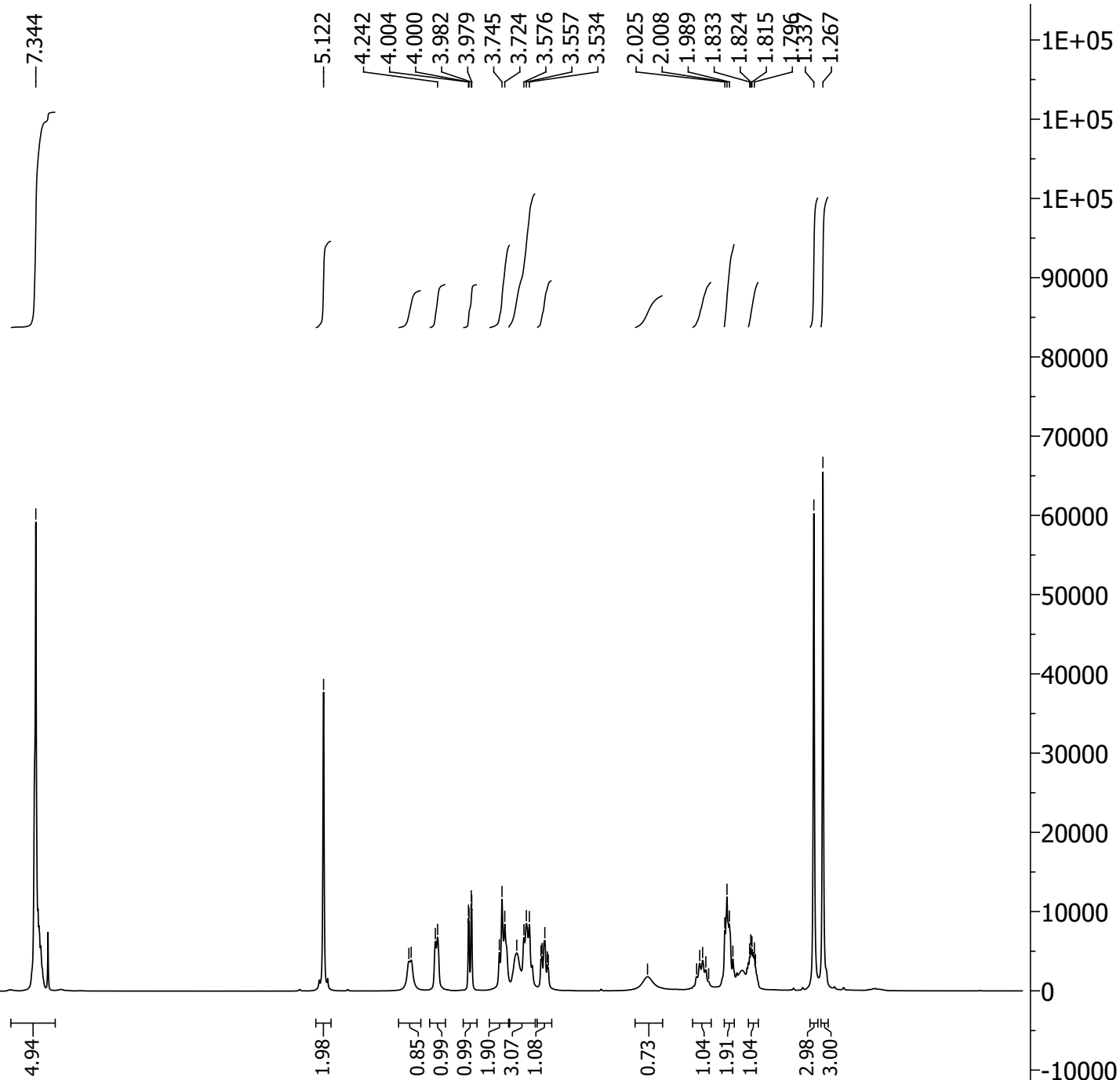


11

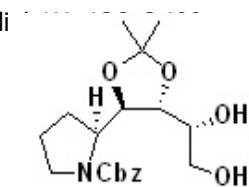


23

f1 (ppm)



di



23

^{13}C -NMR, CDCl_3 ,
100, MHz

—156.990

—136.345

—128.499

—128.133

—127.778

—108.212

—83.917

—76.063

—73.224

—67.511

—65.471

—57.926

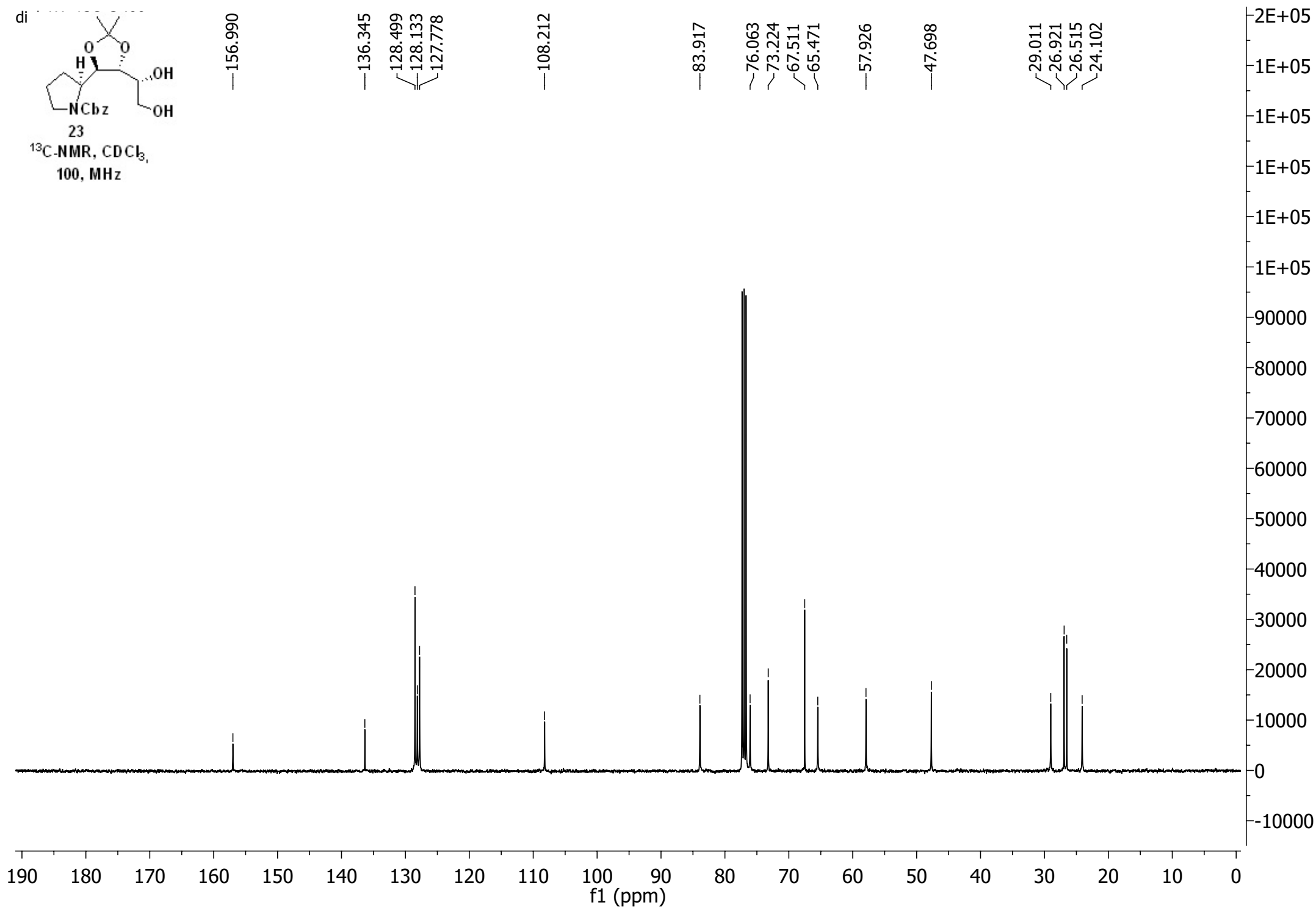
—47.698

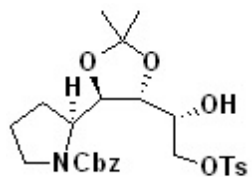
—29.011

—26.921

—26.515

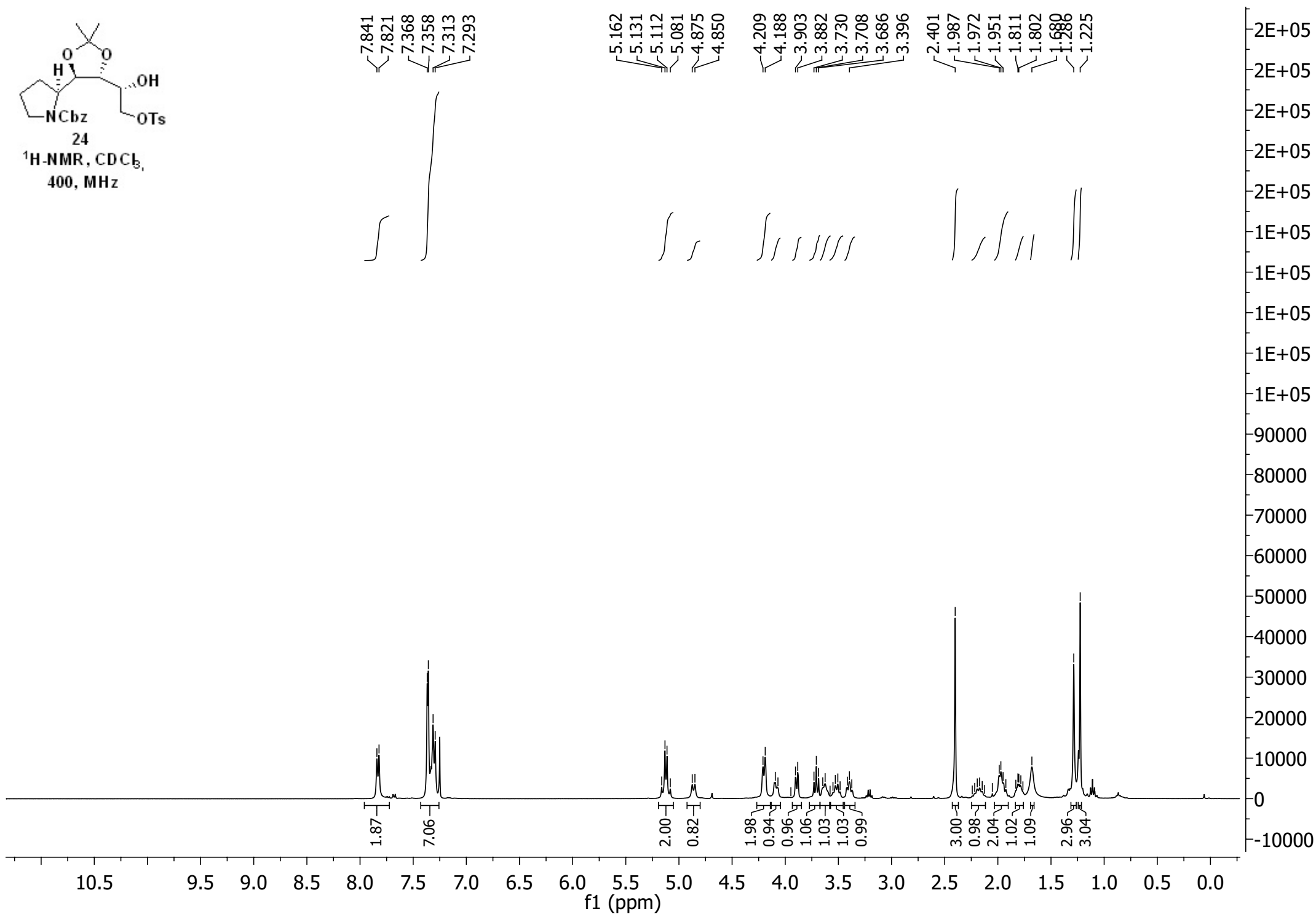
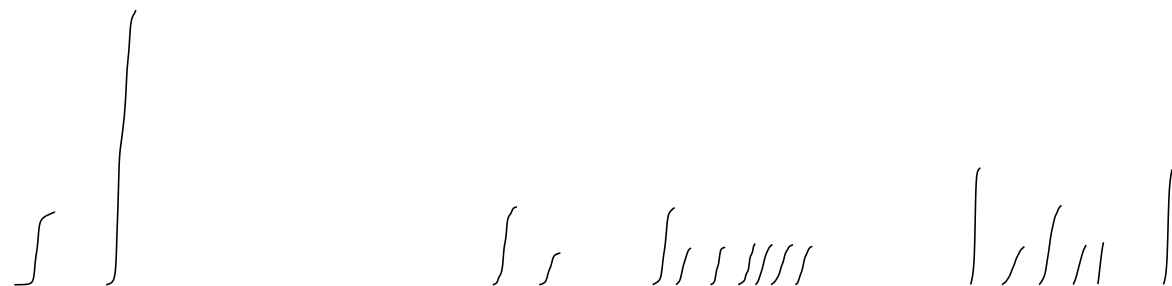
—24.102

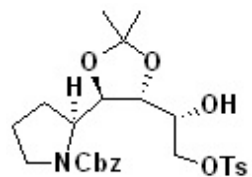




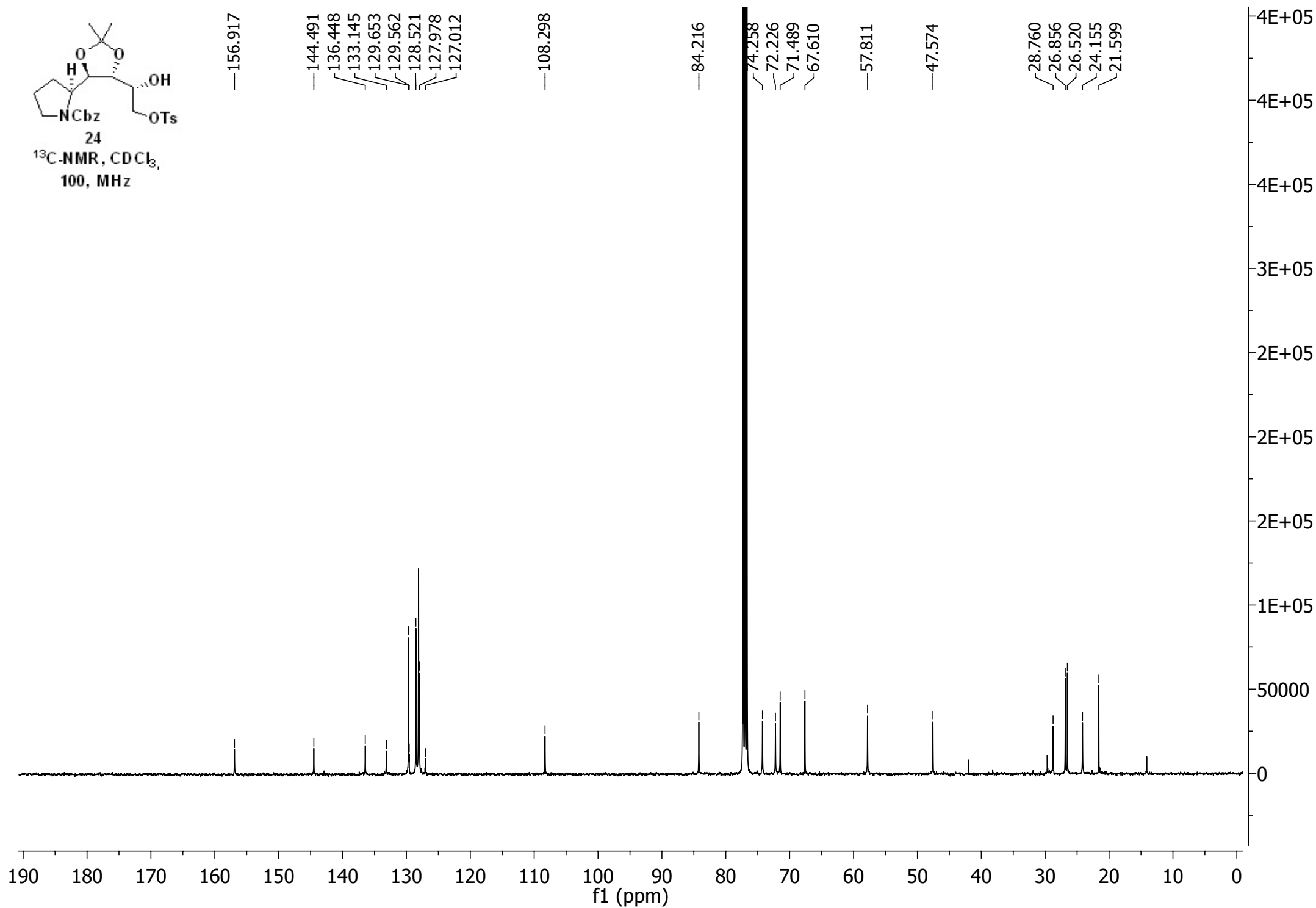
24
¹H-NMR, CDCl₃,
 400, MHz

7.841 7.821 7.368 7.358 7.313 7.293
 5.162 5.131 5.112 5.081 4.875 4.850
 4.209 4.188 3.903 3.882 3.730 3.708 3.686 3.396
 2.401 1.987 1.972 1.951 1.811 1.802 1.680 1.586 1.225

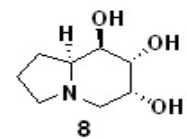




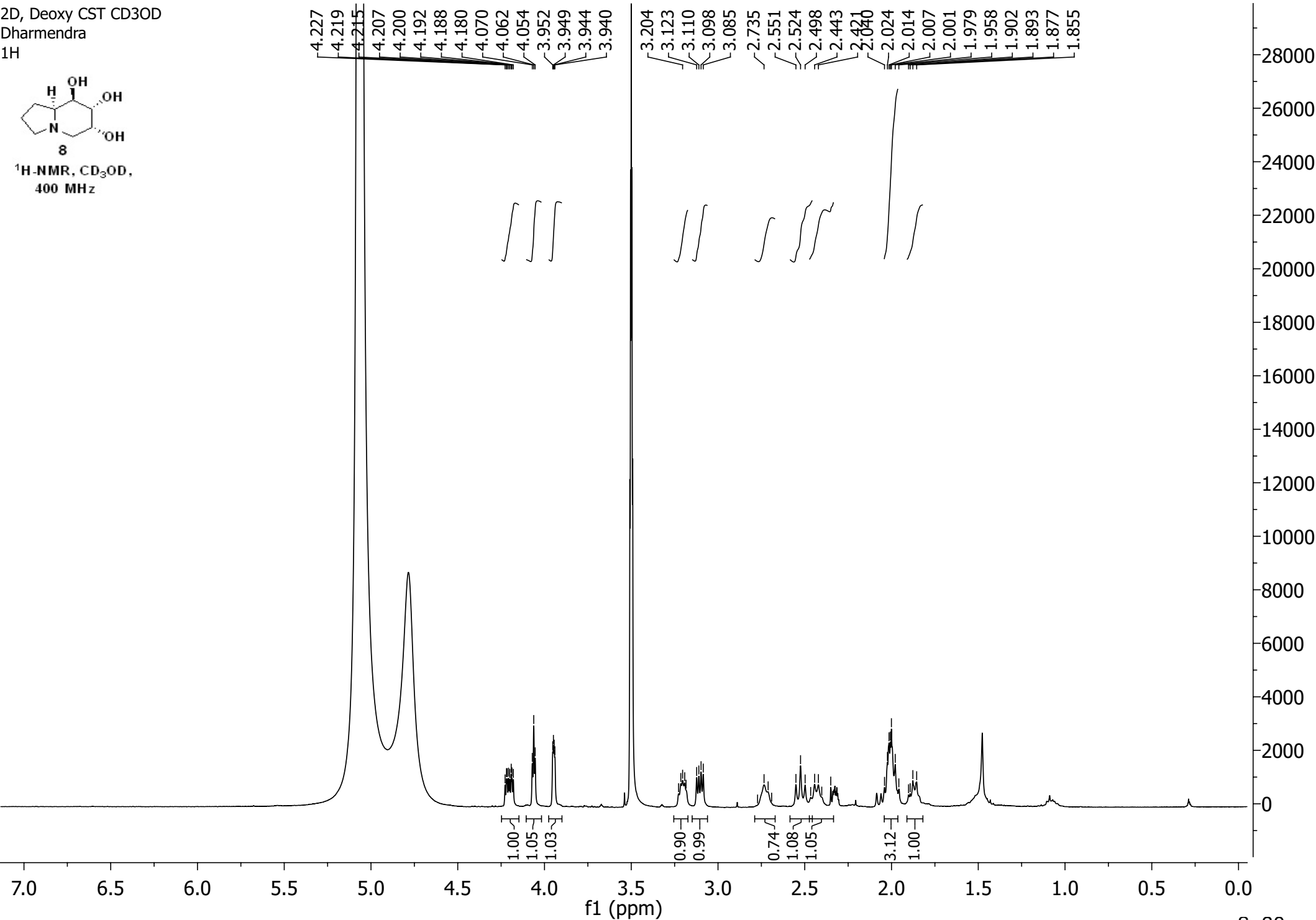
24
 $^{13}\text{C-NMR}$, CDCl_3 ,
 100, MHz



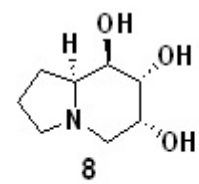
2D, Deoxy CST CD3OD
Dharmendra
1H



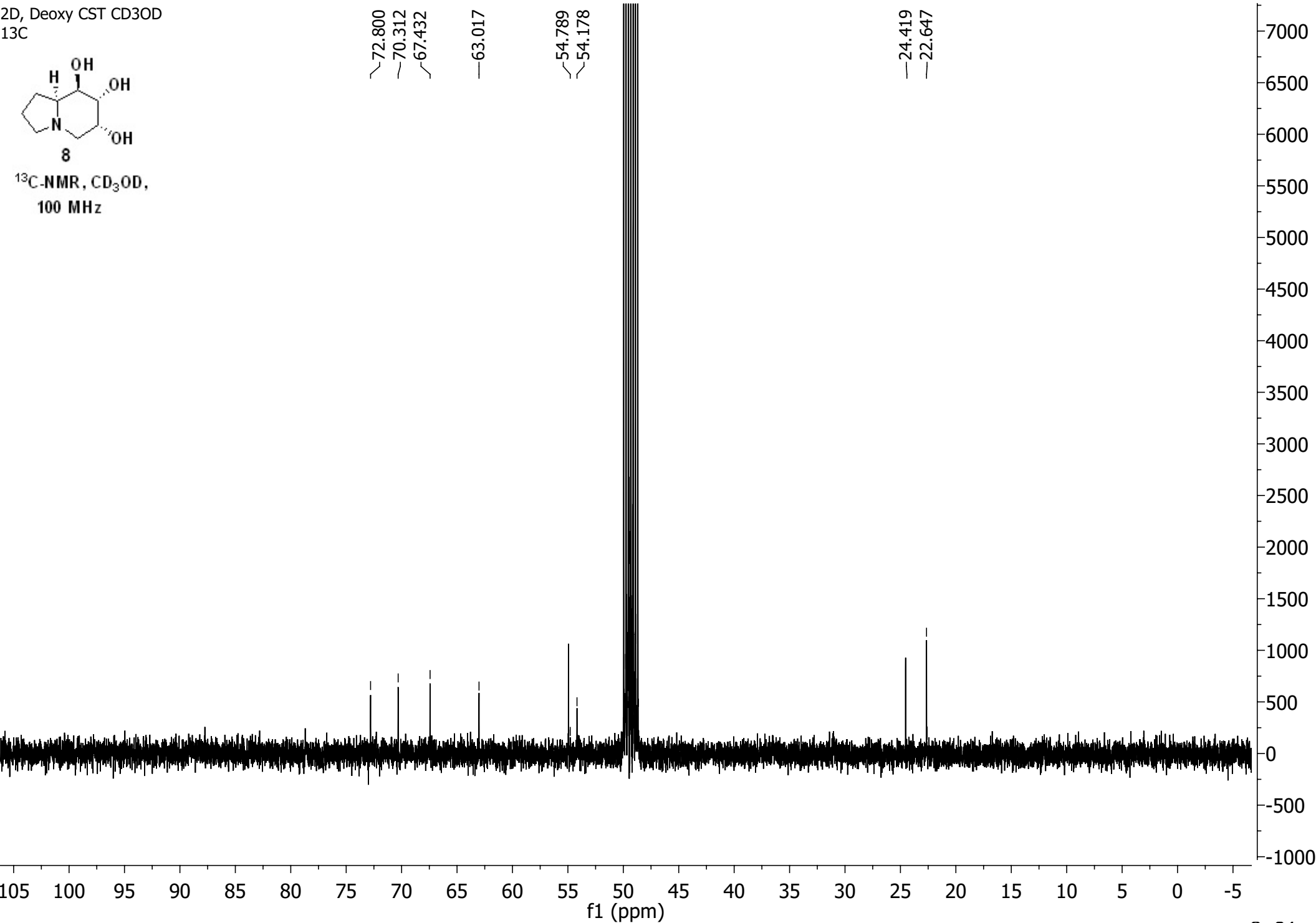
¹H-NMR, CD₃OD,
400 MHz



2D, Deoxy CST CD3OD
13C

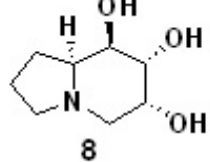


¹³C-NMR, CD₃OD,
100 MHz



2D DEPTW CST CD3OD

D

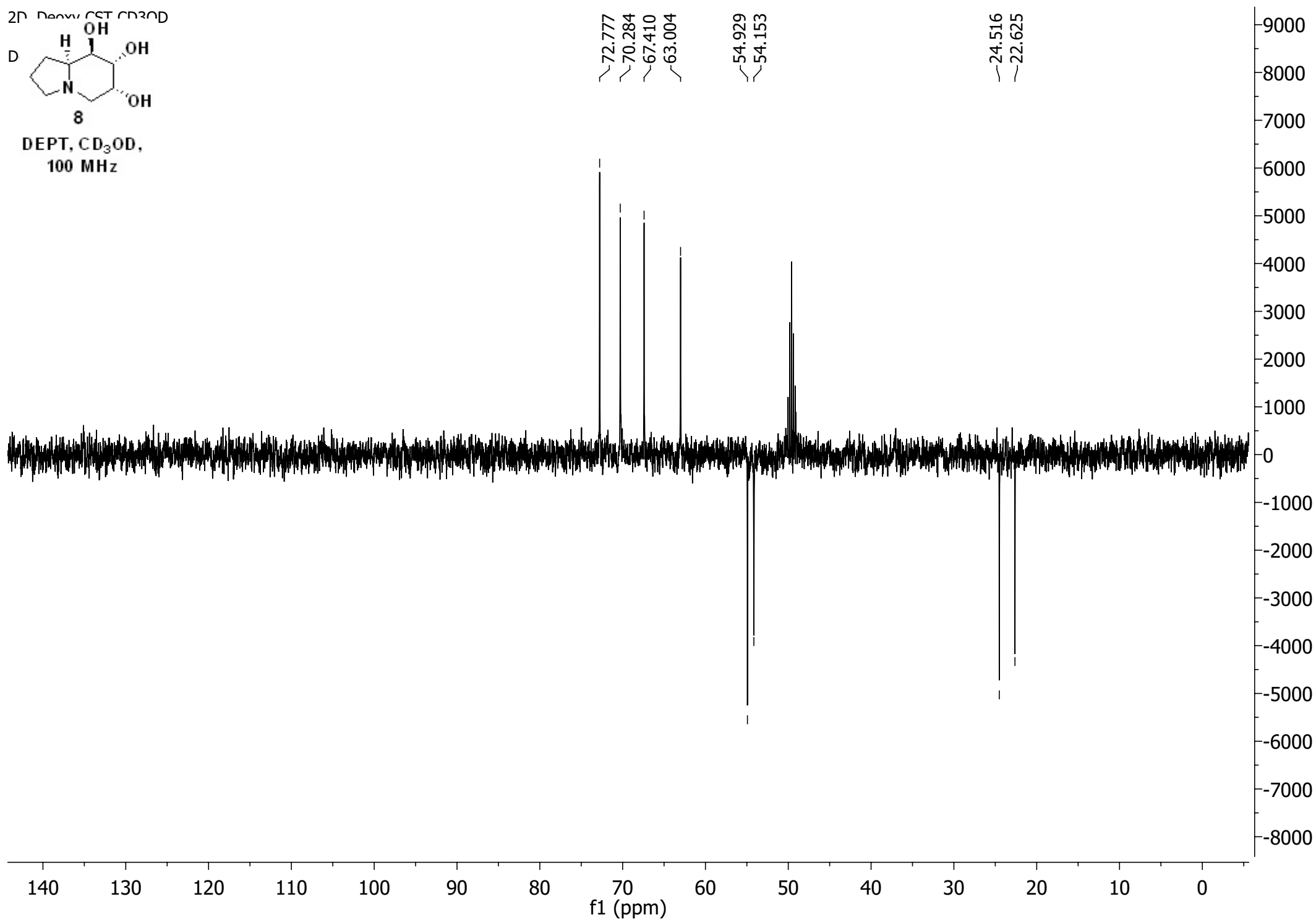


DEPT, CD₃OD,
100 MHz

72.777
70.284
67.410
63.004

54.929
54.153

24.516
22.625

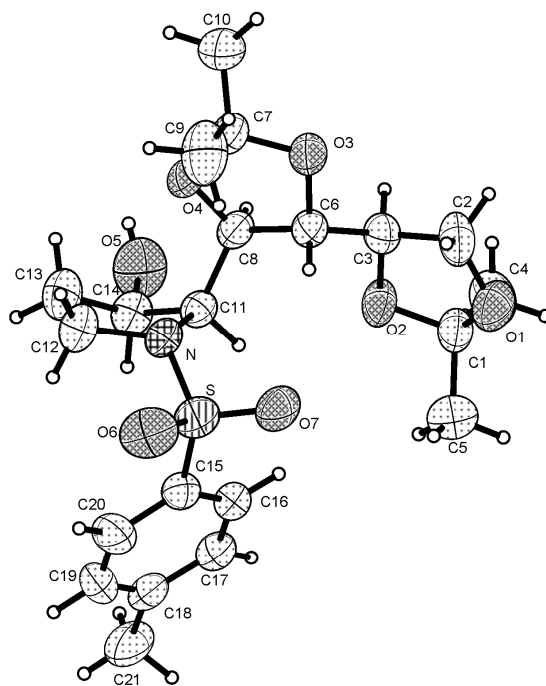
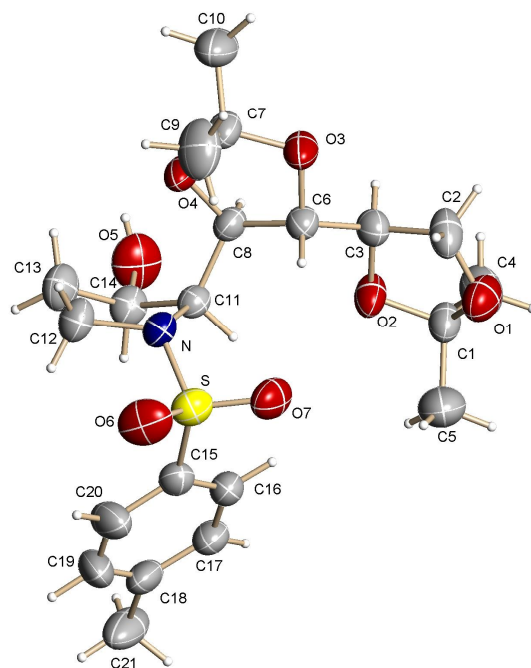


X-ray Crystal Structure Analysis For DTI-357

Crystal Data: Single crystals of the compound were grown by slow evaporation of the solution in dichloromethane and pet-ether. Colourless crystal of approximate size 0.23 x 0.09 x 0.02 mm, was used for data collection on *Bruker SMART APEX* CCD diffractometer using Mo K α radiation with fine focus tube with 50kV and 30mA. Crystal to detector distance 6.05 cm, 512 x 512 pixels / frame, hemisphere data acquisition. Total frames = 1271, Oscillation / frame -0.3°, exposure / frame = 15.0 sec / frame, maximum detector swing angle = $\sim$ 30.0°, beam center = (260.2, 252.5), in plane spot width = 1.24, SAINT integration, θ range = 2.09 to 25.0 °, completeness to θ of 25.0 ° is 100.0 %. SADABS correction applied, C₂₁ H₃₁ N O₇ S, $M = 441.53$. Crystals belong to Orthorhombic, space group P2₁2₁2₁, $a = 6.7501(5)$, $b = b = 17.434(1)$, $c = c = 19.510(1)$ Å, $V = 2296.1(3)$ Å³, $Z = 4$, $D_c = 1.277$ g /cc, μ (MoK α) = 0.181 mm⁻¹, $T = 293(2)$ K, 11648 reflections measured, 4023 unique [$I > 2\sigma(I)$], R value 0.0448, wR2 = 0.0997. All the data were corrected for Lorentzian, polarisation and absorption effects. SHELX-97 (ShelxTL)^{ref} was used for structure solution and full matrix least squares refinement on F². Hydrogen atoms were included in the refinement as per the riding model. Largest diff. peak and hole 0.269 and -0.135 e. Å⁻³. Data collection and refinement parameters are listed in table 1.

X-ray analysis revealed the relative conformation of the molecule as C3, C6, c8, C11 and C14 as R, R, R, S and R configuration respectively.

Caption to Fig 1.: ORTEP diagram of the molecule. Ellipsoids are drawn at 40% probability.



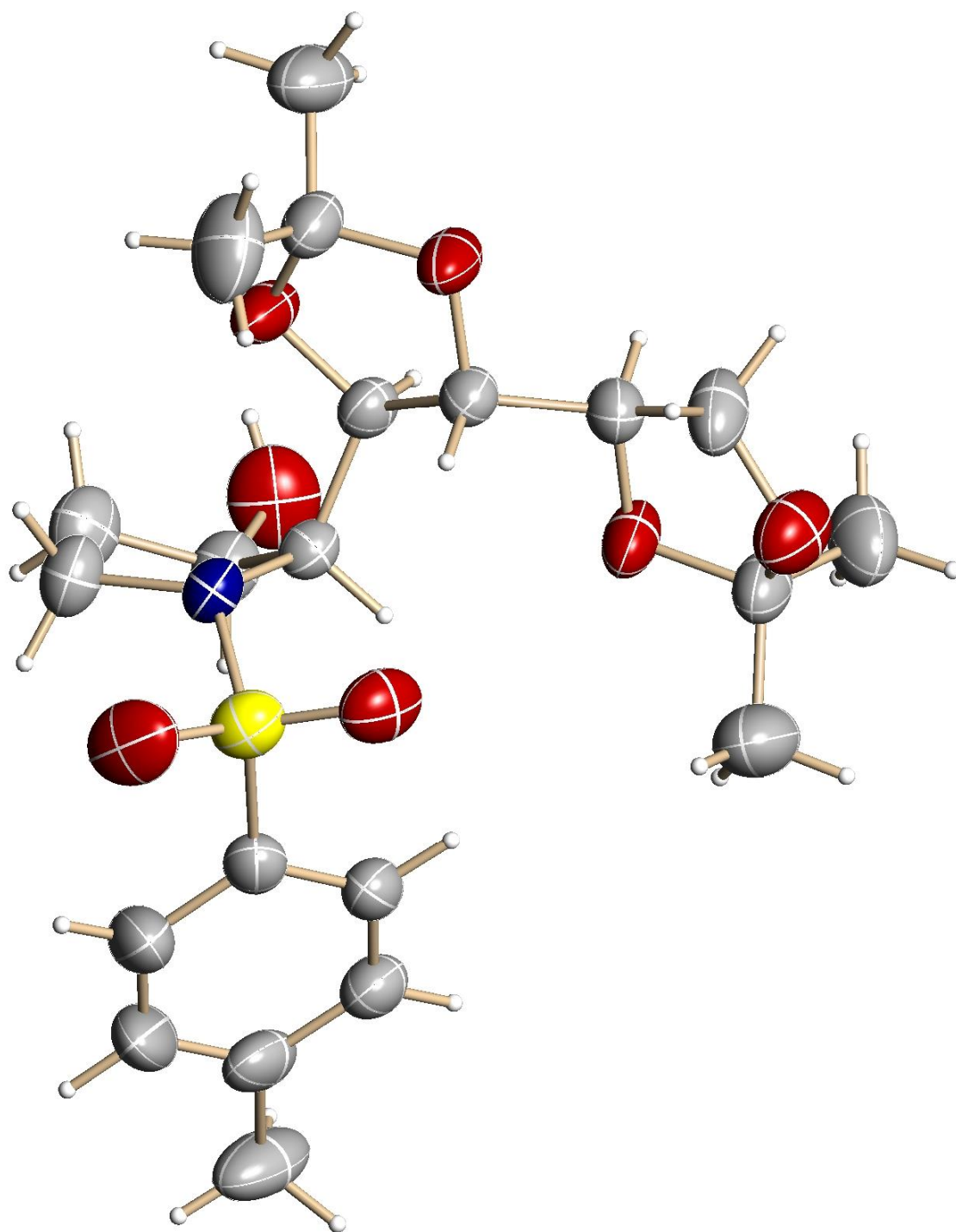


Table 1. Crystal data and structure refinement for dti357.

Empirical formula	C ₂₁ H ₃₁ N O ₇ S	
Formula weight	441.53	
Temperature	296(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P2 ₁ 2 ₁ 2 ₁	
Unit cell dimensions	a = 6.7501(5) Å	α = 90°.
	b = 17.4344(13) Å	β = 90°.
	c = 19.510(1) Å	γ = 90°.
Volume	2296.1(3) Å ³	
Z	4	
Density (calculated)	1.277 g/cc	
Absorption coefficient	0.181 mm ⁻¹	
F(000)	944	
Crystal size	0.23 x 0.09 x 0.02 mm ³	
Theta range for data collection	2.09 to 25.00°.	
Index ranges	-7 ≤ h ≤ 8, -20 ≤ k ≤ 19, -23 ≤ l ≤ 22	
Reflections collected	11648	
Independent reflections	4023 [R(int) = 0.0344]	
Completeness to theta = 25.00°	100.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9957 and 0.9588	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	4023 / 0 / 277	
Goodness-of-fit on F ²	1.024	
Final R indices [I > 2σ(I)]	R1 = 0.0448, wR2 = 0.0997	
R indices (all data)	R1 = 0.0739, wR2 = 0.1110	
Absolute structure parameter	-0.02(10)	
Largest diff. peak and hole	0.269 and -0.135 e.Å ⁻³	

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for dti357. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	$U(\text{eq})$
S	921(1)	789(1)	1026(1)	64(1)
O(1)	4885(4)	-1188(1)	-837(1)	85(1)
O(2)	5799(3)	-669(1)	168(1)	68(1)
O(3)	2700(3)	-1970(1)	1111(1)	70(1)
O(4)	3031(4)	-1217(1)	2060(1)	72(1)
O(5)	6808(4)	88(2)	2325(2)	103(1)
O(6)	-1015(4)	986(1)	1268(1)	93(1)
O(7)	1171(4)	388(1)	398(1)	77(1)
N	1904(4)	266(1)	1629(1)	53(1)
C(1)	6295(5)	-694(2)	-534(1)	66(1)
C(2)	4201(6)	-1699(2)	-338(2)	82(1)
C(3)	4945(5)	-1383(2)	341(2)	60(1)
C(4)	8375(6)	-995(2)	-622(2)	96(1)
C(5)	6059(9)	100(2)	-824(2)	109(1)
C(6)	3368(4)	-1244(1)	875(1)	52(1)
C(7)	2039(6)	-1884(2)	1797(2)	66(1)
C(8)	4152(5)	-863(1)	1527(1)	52(1)
C(9)	-133(7)	-1756(3)	1826(2)	109(2)
C(10)	2752(7)	-2577(2)	2197(2)	98(1)
C(11)	3960(5)	6(1)	1547(1)	47(1)
C(12)	1583(5)	510(2)	2353(2)	72(1)
C(13)	3512(5)	409(2)	2705(2)	77(1)
C(14)	5015(5)	406(2)	2141(2)	61(1)
C(15)	2332(5)	1634(2)	977(2)	57(1)
C(16)	4078(5)	1643(2)	605(1)	60(1)
C(17)	5312(6)	2266(2)	632(1)	66(1)
C(18)	4856(6)	2896(2)	1027(2)	76(1)
C(19)	3081(8)	2891(2)	1375(2)	99(1)
C(20)	1841(6)	2270(2)	1358(2)	88(1)
C(21)	6290(8)	3565(2)	1078(2)	109(2)

Table 3. Bond lengths [Å] and angles [°] for dti357.

S-O(7)	1.421(2)
S-O(6)	1.431(3)
S-N	1.630(2)
S-C(15)	1.756(3)
O(1)-C(2)	1.399(4)
O(1)-C(1)	1.413(4)
O(2)-C(1)	1.409(3)
O(2)-C(3)	1.413(3)
O(3)-C(7)	1.418(4)
O(3)-C(6)	1.421(3)
O(4)-C(8)	1.427(3)
O(4)-C(7)	1.436(4)
O(5)-C(14)	1.378(4)
O(5)-H(5)	0.8200
N-C(11)	1.469(4)
N-C(12)	1.490(4)
C(1)-C(5)	1.504(5)
C(1)-C(4)	1.509(5)
C(2)-C(3)	1.519(4)
C(2)-H(2A)	0.9700
C(2)-H(2B)	0.9700
C(3)-C(6)	1.510(4)
C(3)-H(3)	0.9800
C(4)-H(4A)	0.9600
C(4)-H(4B)	0.9600
C(4)-H(4C)	0.9600
C(5)-H(5A)	0.9600
C(5)-H(5B)	0.9600
C(5)-H(5C)	0.9600
C(6)-C(8)	1.530(4)
C(6)-H(6)	0.9800
C(7)-C(9)	1.484(5)
C(7)-C(10)	1.518(5)
C(8)-C(11)	1.521(3)

C(8)-H(8)	0.9800
C(9)-H(9A)	0.9600
C(9)-H(9B)	0.9600
C(9)-H(9C)	0.9600
C(10)-H(10A)	0.9600
C(10)-H(10B)	0.9600
C(10)-H(10C)	0.9600
C(11)-C(14)	1.529(4)
C(11)-H(11)	0.9800
C(12)-C(13)	1.482(5)
C(12)-H(12A)	0.9700
C(12)-H(12B)	0.9700
C(13)-C(14)	1.497(5)
C(13)-H(13A)	0.9700
C(13)-H(13B)	0.9700
C(14)-H(14)	0.9800
C(15)-C(20)	1.376(4)
C(15)-C(16)	1.384(4)
C(16)-C(17)	1.370(4)
C(16)-H(16)	0.9300
C(17)-C(18)	1.375(4)
C(17)-H(17)	0.9300
C(18)-C(19)	1.377(5)
C(18)-C(21)	1.519(5)
C(19)-C(20)	1.369(5)
C(19)-H(19)	0.9300
C(20)-H(20)	0.9300
C(21)-H(21A)	0.9600
C(21)-H(21B)	0.9600
C(21)-H(21C)	0.9600
O(7)-S-O(6)	120.72(15)
O(7)-S-N	107.44(13)
O(6)-S-N	105.59(14)
O(7)-S-C(15)	107.47(14)
O(6)-S-C(15)	108.15(15)

N-S-C(15)	106.71(13)
C(2)-O(1)-C(1)	108.6(2)
C(1)-O(2)-C(3)	107.6(2)
C(7)-O(3)-C(6)	108.1(2)
C(8)-O(4)-C(7)	109.7(2)
C(14)-O(5)-H(5)	109.5
C(11)-N-C(12)	109.3(2)
C(11)-N-S	118.54(18)
C(12)-N-S	117.7(2)
O(2)-C(1)-O(1)	105.4(2)
O(2)-C(1)-C(5)	108.2(3)
O(1)-C(1)-C(5)	109.3(3)
O(2)-C(1)-C(4)	110.0(3)
O(1)-C(1)-C(4)	111.5(3)
C(5)-C(1)-C(4)	112.1(3)
O(1)-C(2)-C(3)	105.5(2)
O(1)-C(2)-H(2A)	110.6
C(3)-C(2)-H(2A)	110.6
O(1)-C(2)-H(2B)	110.6
C(3)-C(2)-H(2B)	110.6
H(2A)-C(2)-H(2B)	108.8
O(2)-C(3)-C(6)	108.1(2)
O(2)-C(3)-C(2)	104.2(2)
C(6)-C(3)-C(2)	115.3(3)
O(2)-C(3)-H(3)	109.7
C(6)-C(3)-H(3)	109.7
C(2)-C(3)-H(3)	109.7
C(1)-C(4)-H(4A)	109.5
C(1)-C(4)-H(4B)	109.5
H(4A)-C(4)-H(4B)	109.5
C(1)-C(4)-H(4C)	109.5
H(4A)-C(4)-H(4C)	109.5
H(4B)-C(4)-H(4C)	109.5
C(1)-C(5)-H(5A)	109.5
C(1)-C(5)-H(5B)	109.5
H(5A)-C(5)-H(5B)	109.5

C(1)-C(5)-H(5C)	109.5
H(5A)-C(5)-H(5C)	109.5
H(5B)-C(5)-H(5C)	109.5
O(3)-C(6)-C(3)	107.7(2)
O(3)-C(6)-C(8)	103.1(2)
C(3)-C(6)-C(8)	113.6(2)
O(3)-C(6)-H(6)	110.7
C(3)-C(6)-H(6)	110.7
C(8)-C(6)-H(6)	110.7
O(3)-C(7)-O(4)	106.1(2)
O(3)-C(7)-C(9)	111.3(3)
O(4)-C(7)-C(9)	109.1(3)
O(3)-C(7)-C(10)	107.5(3)
O(4)-C(7)-C(10)	108.2(3)
C(9)-C(7)-C(10)	114.4(3)
O(4)-C(8)-C(11)	111.6(2)
O(4)-C(8)-C(6)	103.5(2)
C(11)-C(8)-C(6)	115.1(2)
O(4)-C(8)-H(8)	108.8
C(11)-C(8)-H(8)	108.8
C(6)-C(8)-H(8)	108.8
C(7)-C(9)-H(9A)	109.5
C(7)-C(9)-H(9B)	109.5
H(9A)-C(9)-H(9B)	109.5
C(7)-C(9)-H(9C)	109.5
H(9A)-C(9)-H(9C)	109.5
H(9B)-C(9)-H(9C)	109.5
C(7)-C(10)-H(10A)	109.5
C(7)-C(10)-H(10B)	109.5
H(10A)-C(10)-H(10B)	109.5
C(7)-C(10)-H(10C)	109.5
H(10A)-C(10)-H(10C)	109.5
H(10B)-C(10)-H(10C)	109.5
N-C(11)-C(8)	113.0(2)
N-C(11)-C(14)	102.5(2)
C(8)-C(11)-C(14)	115.7(2)

N-C(11)-H(11)	108.5
C(8)-C(11)-H(11)	108.5
C(14)-C(11)-H(11)	108.5
C(13)-C(12)-N	106.1(3)
C(13)-C(12)-H(12A)	110.5
N-C(12)-H(12A)	110.5
C(13)-C(12)-H(12B)	110.5
N-C(12)-H(12B)	110.5
H(12A)-C(12)-H(12B)	108.7
C(12)-C(13)-C(14)	104.8(3)
C(12)-C(13)-H(13A)	110.8
C(14)-C(13)-H(13A)	110.8
C(12)-C(13)-H(13B)	110.8
C(14)-C(13)-H(13B)	110.8
H(13A)-C(13)-H(13B)	108.9
O(5)-C(14)-C(13)	113.9(3)
O(5)-C(14)-C(11)	115.0(3)
C(13)-C(14)-C(11)	104.1(3)
O(5)-C(14)-H(14)	107.8
C(13)-C(14)-H(14)	107.8
C(11)-C(14)-H(14)	107.8
C(20)-C(15)-C(16)	118.6(3)
C(20)-C(15)-S	121.0(3)
C(16)-C(15)-S	120.0(2)
C(17)-C(16)-C(15)	120.5(3)
C(17)-C(16)-H(16)	119.8
C(15)-C(16)-H(16)	119.8
C(16)-C(17)-C(18)	121.2(3)
C(16)-C(17)-H(17)	119.4
C(18)-C(17)-H(17)	119.4
C(17)-C(18)-C(19)	117.7(3)
C(17)-C(18)-C(21)	120.5(4)
C(19)-C(18)-C(21)	121.8(3)
C(20)-C(19)-C(18)	121.7(3)
C(20)-C(19)-H(19)	119.1
C(18)-C(19)-H(19)	119.1

C(19)-C(20)-C(15)	120.2(3)
C(19)-C(20)-H(20)	119.9
C(15)-C(20)-H(20)	119.9
C(18)-C(21)-H(21A)	109.5
C(18)-C(21)-H(21B)	109.5
H(21A)-C(21)-H(21B)	109.5
C(18)-C(21)-H(21C)	109.5
H(21A)-C(21)-H(21C)	109.5
H(21B)-C(21)-H(21C)	109.5

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for dti357. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
S	55(1)	70(1)	65(1)	10(1)	-6(1)	-2(1)
O(1)	104(2)	92(2)	59(1)	-5(1)	-4(1)	-23(2)
O(2)	80(2)	74(1)	50(1)	-11(1)	19(1)	-23(1)
O(3)	98(2)	54(1)	57(1)	-4(1)	18(1)	-22(1)
O(4)	115(2)	54(1)	47(1)	2(1)	11(1)	-22(1)
O(5)	73(2)	136(2)	99(2)	-14(2)	-29(2)	-2(2)
O(6)	49(1)	111(2)	118(2)	23(2)	0(1)	9(1)
O(7)	96(2)	78(1)	56(1)	3(1)	-20(1)	-15(1)
N	53(2)	60(1)	45(1)	5(1)	4(1)	-4(1)
C(1)	82(3)	64(2)	51(2)	-9(1)	15(2)	-13(2)
C(2)	93(3)	84(2)	67(2)	-25(2)	28(2)	-27(2)
C(3)	65(2)	57(2)	56(2)	-3(1)	14(2)	2(2)
C(4)	85(3)	96(3)	108(3)	-17(2)	37(2)	-8(2)
C(5)	157(4)	77(2)	93(3)	14(2)	17(3)	2(3)
C(6)	57(2)	50(2)	48(2)	0(1)	7(1)	-4(1)
C(7)	87(3)	61(2)	50(2)	2(1)	15(2)	-11(2)
C(8)	60(2)	56(2)	42(2)	5(1)	1(1)	-2(2)
C(9)	84(3)	146(4)	96(3)	-21(3)	29(2)	-13(3)
C(10)	155(4)	62(2)	78(2)	6(2)	3(3)	-10(2)
C(11)	46(2)	50(2)	44(2)	3(1)	0(1)	-6(2)
C(12)	71(2)	91(2)	53(2)	-1(2)	10(2)	-2(2)
C(13)	85(3)	92(2)	56(2)	-12(2)	-6(2)	-8(2)
C(14)	52(2)	71(2)	59(2)	-3(2)	-4(2)	-11(2)
C(15)	62(2)	55(2)	52(2)	5(1)	2(2)	10(2)
C(16)	78(2)	54(2)	48(2)	3(1)	8(2)	3(2)
C(17)	85(3)	61(2)	53(2)	8(2)	8(2)	-7(2)
C(18)	122(3)	61(2)	46(2)	12(2)	2(2)	-9(2)
C(19)	171(5)	51(2)	75(3)	-8(2)	38(3)	8(2)
C(20)	113(3)	60(2)	90(3)	7(2)	40(2)	17(2)
C(21)	182(5)	75(2)	69(2)	9(2)	-7(3)	-52(3)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^{-3}$) for dti357.

	x	y	z	U(eq)
H(5)	6615	-291	2565	154
H(2A)	4723	-2209	-418	98
H(2B)	2765	-1723	-341	98
H(3)	5966	-1724	528	71
H(4A)	9292	-653	-403	144
H(4B)	8681	-1029	-1101	144
H(4C)	8475	-1494	-417	144
H(5A)	4689	247	-805	164
H(5B)	6498	104	-1292	164
H(5C)	6837	455	-562	164
H(6)	2267	-948	682	62
H(8)	5550	-1000	1584	63
H(9A)	-494	-1362	1506	163
H(9B)	-502	-1601	2280	163
H(9C)	-808	-2223	1710	163
H(10A)	2258	-3036	1986	147
H(10B)	2275	-2547	2660	147
H(10C)	4174	-2587	2198	147
H(11)	4477	214	1115	56
H(12A)	1166	1042	2372	86
H(12B)	573	195	2568	86
H(13A)	3542	-70	2956	93
H(13B)	3756	828	3021	93
H(14)	5256	939	2006	73
H(16)	4416	1224	335	72
H(17)	6480	2264	379	79
H(19)	2717	3320	1628	119
H(20)	662	2278	1604	105
H(21A)	7449	3407	1327	163
H(21B)	5664	3982	1314	163

H(21C)

6665

3727

626

163

Table 6. Torsion angles [°] for dti357.

O(7)-S-N-C(11)	53.2(2)
O(6)-S-N-C(11)	-176.8(2)
C(15)-S-N-C(11)	-61.8(2)
O(7)-S-N-C(12)	-171.6(2)
O(6)-S-N-C(12)	-41.5(3)
C(15)-S-N-C(12)	73.4(2)
C(3)-O(2)-C(1)-O(1)	30.6(3)
C(3)-O(2)-C(1)-C(5)	147.5(3)
C(3)-O(2)-C(1)-C(4)	-89.8(3)
C(2)-O(1)-C(1)-O(2)	-26.2(3)
C(2)-O(1)-C(1)-C(5)	-142.4(3)
C(2)-O(1)-C(1)-C(4)	93.2(3)
C(1)-O(1)-C(2)-C(3)	12.0(4)
C(1)-O(2)-C(3)-C(6)	-145.7(3)
C(1)-O(2)-C(3)-C(2)	-22.6(3)
O(1)-C(2)-C(3)-O(2)	6.5(3)
O(1)-C(2)-C(3)-C(6)	124.8(3)
C(7)-O(3)-C(6)-C(3)	151.5(3)
C(7)-O(3)-C(6)-C(8)	31.1(3)
O(2)-C(3)-C(6)-O(3)	-172.5(2)
C(2)-C(3)-C(6)-O(3)	71.4(3)
O(2)-C(3)-C(6)-C(8)	-59.0(3)
C(2)-C(3)-C(6)-C(8)	-175.1(2)
C(6)-O(3)-C(7)-O(4)	-23.3(4)
C(6)-O(3)-C(7)-C(9)	95.1(3)
C(6)-O(3)-C(7)-C(10)	-138.9(3)
C(8)-O(4)-C(7)-O(3)	4.9(3)
C(8)-O(4)-C(7)-C(9)	-115.0(3)
C(8)-O(4)-C(7)-C(10)	120.0(3)
C(7)-O(4)-C(8)-C(11)	138.0(3)
C(7)-O(4)-C(8)-C(6)	13.7(3)
O(3)-C(6)-C(8)-O(4)	-27.0(3)
C(3)-C(6)-C(8)-O(4)	-143.2(2)
O(3)-C(6)-C(8)-C(11)	-149.1(3)

C(3)-C(6)-C(8)-C(11)	94.7(3)
C(12)-N-C(11)-C(8)	102.7(3)
S-N-C(11)-C(8)	-118.6(2)
C(12)-N-C(11)-C(14)	-22.5(3)
S-N-C(11)-C(14)	116.3(2)
O(4)-C(8)-C(11)-N	-46.3(3)
C(6)-C(8)-C(11)-N	71.4(3)
O(4)-C(8)-C(11)-C(14)	71.4(4)
C(6)-C(8)-C(11)-C(14)	-171.0(3)
C(11)-N-C(12)-C(13)	1.4(3)
S-N-C(12)-C(13)	-137.7(2)
N-C(12)-C(13)-C(14)	21.0(4)
C(12)-C(13)-C(14)-O(5)	-161.0(3)
C(12)-C(13)-C(14)-C(11)	-34.9(3)
N-C(11)-C(14)-O(5)	160.4(3)
C(8)-C(11)-C(14)-O(5)	37.0(4)
N-C(11)-C(14)-C(13)	35.0(3)
C(8)-C(11)-C(14)-C(13)	-88.3(3)
O(7)-S-C(15)-C(20)	154.3(3)
O(6)-S-C(15)-C(20)	22.5(3)
N-S-C(15)-C(20)	-90.7(3)
O(7)-S-C(15)-C(16)	-32.3(3)
O(6)-S-C(15)-C(16)	-164.1(2)
N-S-C(15)-C(16)	82.7(3)
C(20)-C(15)-C(16)-C(17)	1.7(4)
S-C(15)-C(16)-C(17)	-171.8(2)
C(15)-C(16)-C(17)-C(18)	0.0(4)
C(16)-C(17)-C(18)-C(19)	-2.3(5)
C(16)-C(17)-C(18)-C(21)	177.0(3)
C(17)-C(18)-C(19)-C(20)	2.9(5)
C(21)-C(18)-C(19)-C(20)	-176.4(3)
C(18)-C(19)-C(20)-C(15)	-1.2(6)
C(16)-C(15)-C(20)-C(19)	-1.2(5)
S-C(15)-C(20)-C(19)	172.3(3)
