

Supporting Information:

Developing principles for predicting ionic liquid effects on reaction outcome. A demonstration using a simple condensation reaction

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Representative plots for the determination of rate constants

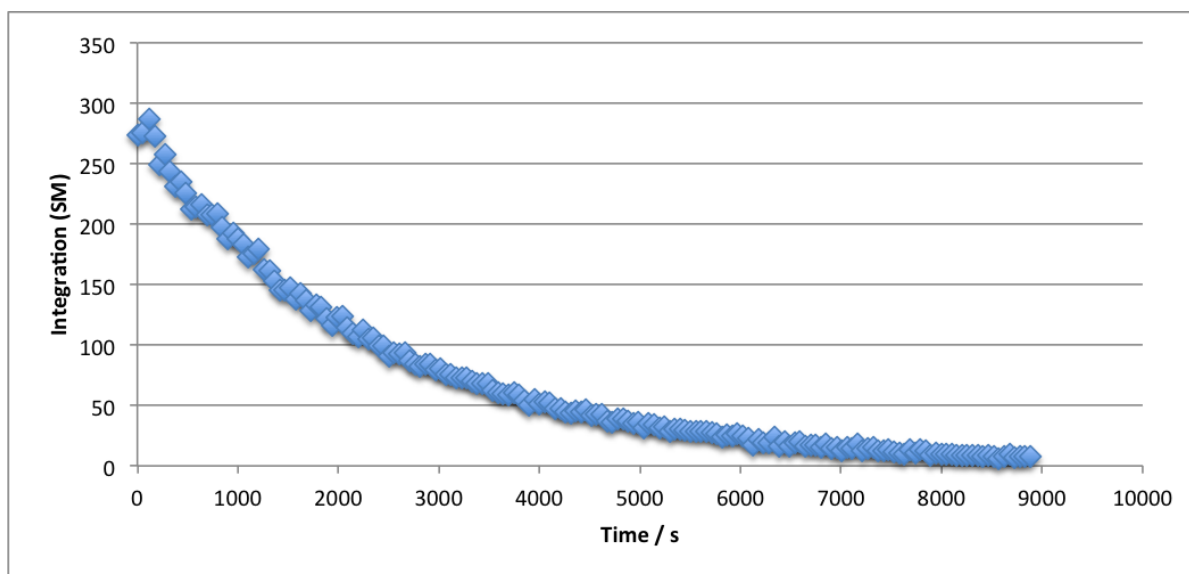


Figure S1: Plot of the integration of the signal due to the starting material **1a** in the ^1H NMR spectrum vs time for the reaction of species **1a** and **2**, in a mixture of the ionic liquid **4** in acetonitrile at a mole fraction of 0.16 at 8.1°C . Using exponential fitting in Graphpad Prism the R^2 value is 0.998 and the rate constant determined was $(4.17 \pm 0.04) \times 10^{-4} \text{ s}^{-1}$.

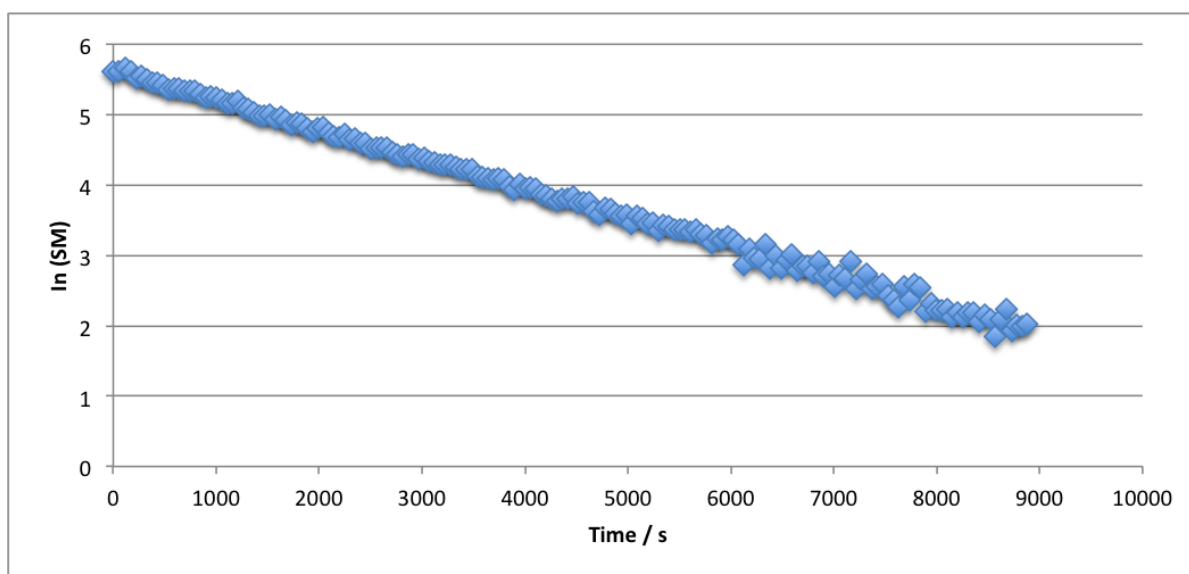


Figure S2: Plot of the natural logarithm of the integration of the signal due to the starting material **1a** in the ^1H NMR spectrum vs time, for the case shown in Figure S1 above (the reaction of species **1a** and **2** in a mixture of the ionic liquid **4** in acetonitrile at a mole fraction of 0.16 at 8.1°C). Using the LINEST function in Excel, the R^2 value is 0.995 and the rate constant determined was $(4.13 \pm 0.02) \times 10^{-4} \text{ s}^{-1}$.

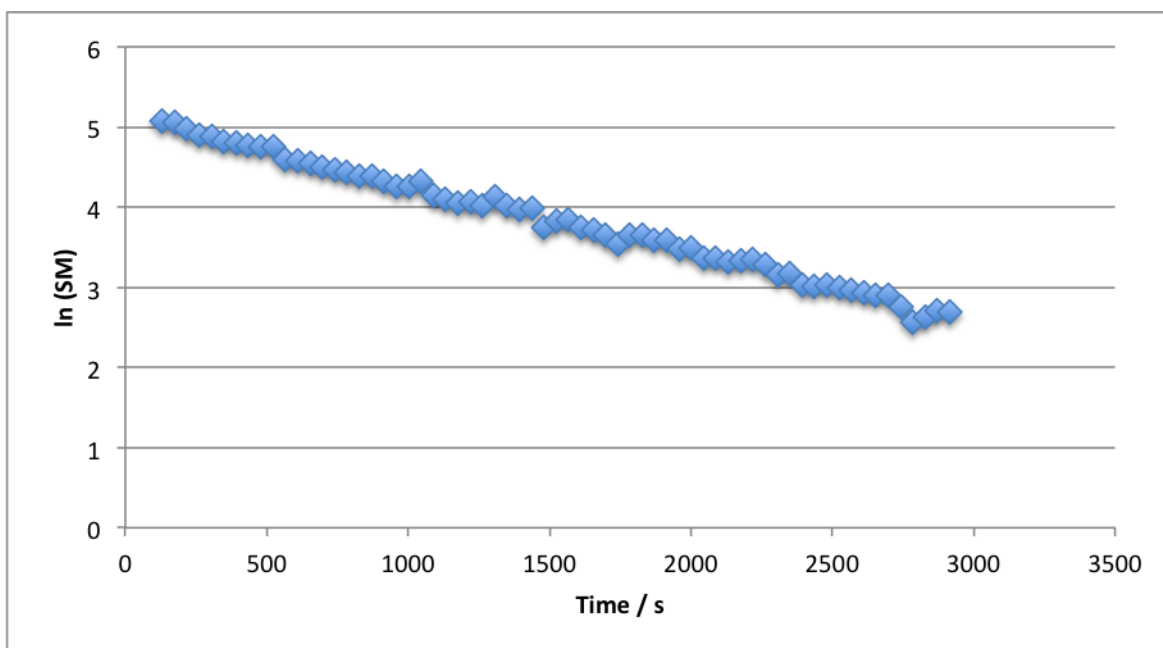


Figure S3: Plot of the integration of the signal due to the starting material **1a** in the ^1H NMR spectrum vs time for the reaction of species **1a** and **2**, in a mixture of the ionic liquid **4** in acetonitrile at a mole fraction of 0.90 at 8.1°C Using the LINEST function in Excel, the R^2 value is 0.994 and the rate constant determined was $(8.49 \pm 0.08) \times 10^{-4} \text{ s}^{-1}$.

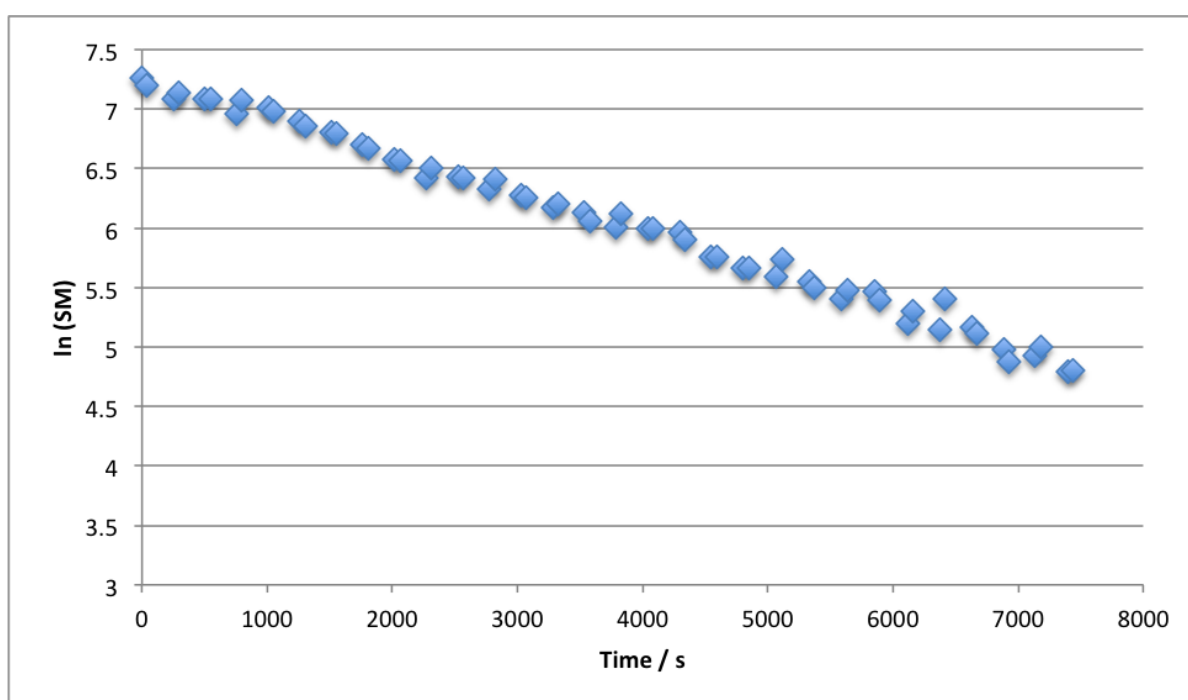


Figure S4: Plot of the integration of the signal due to the starting material **1b** in the ^1H NMR spectrum vs time for the reaction of species **1b** and **2**, in a mixture of the ionic liquid **8** at a mole fraction of 0.87 (diluted only by reagents) at 36.6°C Using the LINEST function in Excel, the R^2 value is 0.993 and the rate constant determined was $(3.23 \pm 0.04) \times 10^{-4} \text{ s}^{-1}$.

Benzaldehyde **1a** and hexan-1-amine **2**

Mole fraction plot: [Bmim][N(CF₃SO₂)₂] **4** in acetonitrile, at 8.1°C

Mole Fraction of 4	[Nu] / mol L ⁻¹	$k_{\text{obs}} / 10^{-3} \text{ s}^{-1}$	$k_2 / 10^{-3} \text{ L mol}^{-1} \text{ s}^{-1}$	Average k_2 / $10^{-3} \text{ L mol}^{-1} \text{ s}^{-1}$
0.00	0.144	0.099	0.69	0.75 (0.06)
		0.115	0.80	
		0.110	0.76	
0.01	0.151	0.187	1.24	1.17 (0.08)
		0.178	1.18	
		0.162	1.07	
0.05	0.156	0.349	2.23	2.22 (0.08)
		0.360	2.30	
		0.335	2.14	
0.16	0.131	0.435	3.31	3.18 (0.11)
		0.408	3.10	
		0.413	3.14	
0.25	0.129	0.586	4.56	4.48 (0.31)
		0.610	4.75	
		0.533	4.15	
0.39	0.129	0.679	5.28	5.40 (0.11)
		0.708	5.50	
		0.697	5.42	
0.49	0.151	0.706	4.67	4.79 (0.10)
		0.734	4.85	
		0.732	4.84	
0.51	0.134	0.589	4.40	4.42 (0.25)
		0.626	4.68	
		0.559	4.18	
0.65	0.126	0.770	6.10	6.01 (0.15)
		0.737	5.84	
		0.769	6.09	
0.80	0.143	0.858	6.01	6.02 (0.23)
		0.893	6.25	
		0.828	5.80	
0.91	0.131	0.903	6.90	6.72 (0.21)
		0.849	6.48	
		0.888	6.78	
0.97	0.099	0.694	7.02	6.80 (0.20)
		0.657	6.64	
		0.667	6.74	

Mole fraction plot: [Bmpy][N(CF₃SO₂)₂] 5 in acetonitrile, at 8.1°C

Mole Fraction of 5	[Nu] / mol L ⁻¹	$k_{\text{obs}} / 10^{-3} \text{ s}^{-1}$	$k_2 / 10^{-3} \text{ L mol}^{-1} \text{ s}^{-1}$	Average $k_2 / 10^{-3} \text{ L mol}^{-1} \text{ s}^{-1}$
0.00	0.144	0.099	0.69	0.75 (0.06)
		0.115	0.80	
		0.110	0.76	
0.06	0.148	0.343	2.32	2.35 (0.04)
		0.354	2.39	
		0.346	2.33	
0.13	0.152	0.635	4.18	4.19 (0.04)
		0.629	4.15	
		0.642	4.23	
0.24	0.164	0.974	5.93	6.16 (0.24)
		0.991	6.03	
		1.024	6.23	
		1.062	6.46	
0.28	0.276	1.343	4.86	4.88 (0.02)
		1.349	4.88	
		1.355	4.91	
0.41	0.209	1.528	7.33	7.15 (0.29)
		1.421	6.81	
		1.525	7.31	
0.51	0.203	1.764	8.71	8.72 (0.09)
		1.749	8.63	
		1.785	8.81	
0.61	0.215	2.448	11.39	12.34 (1.00)
	0.200	2.452	12.26	
		2.677	13.38	
0.71	0.177	2.122	12.00	12.02 (0.38)
		2.196	12.41	
		2.060	11.64	
0.79	0.203	2.726	13.46	14.18 (0.81)
		3.051	15.06	
		2.843	14.03	
0.86	0.215	2.842	13.19	13.28 (0.28)
		2.813	13.06	
		2.928	13.59	
0.91	0.178	3.137	17.59	17.33 (0.56)
		2.976	16.69	
		3.159	17.71	
0.94	0.189	4.168	22.00	21.33 (0.65)
		4.029	21.27	
		3.924	20.71	

Mole fraction plot: [Bm₂im][N(CF₃SO₂)₂] **6** in acetonitrile, at 8.1°C

Mole Fraction of 6	[Nu] / mol L ⁻¹	$k_{\text{obs}} / 10^{-3} \text{ s}^{-1}$	$k_2 / 10^{-3} \text{ L mol}^{-1} \text{ s}^{-1}$	Average k_2 / $10^{-3} \text{ L mol}^{-1} \text{ s}^{-1}$
0.00	0.144	0.099	0.69	0.75 (0.06)
		0.115	0.80	
		0.110	0.76	
0.10	0.160	0.410	2.57	2.55 (0.04)
		0.399	2.50	
		0.411	2.58	
0.20	0.112	0.378	3.38	3.34 (0.07)
		0.378	3.37	
		0.365	3.26	
0.30	0.136	0.655	4.82	4.65 (0.15)
		0.621	4.57	
		0.618	4.55	
0.39	0.151	0.779	5.15	5.16 (0.03)
		0.786	5.20	
		0.776	5.14	
0.49	0.167	0.740	4.43	4.49 (0.05)
		0.758	4.54	
		0.749	4.49	
0.60	0.138	0.782	5.65	5.68 (0.09)
		0.775	5.60	
		0.800	5.78	
0.67	0.159	0.968	6.08	6.04 (0.04)
		0.955	6.00	
		0.959	6.03	
0.86	0.166	1.052	6.33	6.31 (0.03)
		1.045	6.29	
0.95	0.157	1.062	6.76	6.89 (0.34)
		1.145	7.28	
		1.045	6.64	

Mole fraction plot: [Bm₄im][N(CF₃SO₂)₂] 7 in acetonitrile, at 8.1°C

Mole Fraction of 7	[Nu] / mol L⁻¹	<i>k</i>_{obs} / 10⁻³ s⁻¹	<i>k</i>₂ / 10⁻³ L mol⁻¹ s⁻¹	Average <i>k</i>₂ / 10⁻³ L mol⁻¹ s⁻¹
0.00	0.144	0.099	0.69	0.75 (0.06)
		0.115	0.80	
		0.110	0.76	
0.05	0.150	0.696	4.64	4.60 (0.05)
		0.696	4.64	
		0.682	4.54	
0.12	0.150	1.383	9.21	9.47 (0.24)
		1.452	9.67	
		1.433	9.54	

Eyring Data: Acetonitrile

Temperature / °C	[Nu] / mol L ⁻¹	k_{obs} / 10 ⁻³ s ⁻¹	k_2 / 10 ⁻³ L mol ⁻¹ s ⁻¹
8.1	0.144	0.099	0.69
		0.095	0.66
		0.110	0.76
21.2	0.144	0.188	1.31
		0.188	1.35
		0.204	1.46
27.8	0.144	0.240	1.67
		0.245	1.70
		0.264	1.89
15.4	0.144	0.166	1.19
		0.142	1.01
		0.171	1.22

Eyring Data: [Bmim][N(CF₃SO₂)₂] **4** in acetonitrile

Mole Fraction of 4	Temperature / °C	[Nu] / mol L ⁻¹	<i>k</i> _{obs} / 10 ⁻³ s ⁻¹	<i>k</i> ₂ / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.23	21.4	0.139	1.028	7.42
			1.021	7.37
			1.033	7.45
			1.032	7.45
	15.4		0.759	5.48
			0.745	5.37
			0.768	5.54
	8.1		0.494	3.56
			0.503	3.63
			0.494	3.56
			0.338	2.44
	2.3		0.334	2.41
			0.339	2.44
			0.50	21.1
1.631		11.03		
1.661		11.23		
1.655	11.19			
15.3	1.120	7.57		
	1.191	8.05		
	1.200	8.11		
8.1	0.780	5.27		
	0.769	5.20		
	0.827	5.59		
	0.551	3.73		
1.7	0.548	3.71		
	0.558	3.77		
	0.97	15.4	1.011	10.22
0.929			9.39	
0.987			9.97	
1.281			12.94	
21.4		1.311	13.25	
		1.524	15.40	
		0.694	7.02	
		0.657	6.64	
8.1		0.667	6.74	
		0.416	4.20	
		0.432	4.36	
0		0.391	3.95	

Activation parameters: Benzaldehyde **1a** and hexan-1-amine **2** in [Bmim][N(CF₃SO₂)₂] **4** / acetonitrile mixtures

Mole Fraction [Bmim][N(CF₃SO₂)₂] 4 in acetonitrile	ΔH^\ddagger / kJ mol⁻¹	ΔS^\ddagger / J K⁻¹ mol⁻¹
0	27.2 (2.5)	-272.1 (8.6)
0.23	34.5 (0.5)	-233.2 (1.7)
0.50	32.4 (0.8)	-237.0 (2.8)
0.97	32.4 (1.3)	-235.2 (4.6)

Eyring Data: [Bmpy][N(CF₃SO₂)₂] **5** in acetonitrile

Mole Fraction of 5	Temperature / °C	[Nu] / mol L ⁻¹	k_{obs} / 10 ⁻³ s ⁻¹	k_2 / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.24	8.1	0.164	0.974	5.93
			0.991	6.03
			1.024	6.23
			1.062	6.46
	1.6		0.660	4.02
			0.670	4.08
			0.662	4.03
			0.405	2.47
	-5.2		0.395	2.41
			0.396	2.41
			0.247	1.51
			0.246	1.50
	-12.7		0.246	1.50
			8.1	4.168
4.029		21.27		
3.924		20.71		
2.4		2.110		11.14
		2.758		14.56
		2.328		12.29
-4.6		1.420	7.50	
		1.330	7.02	
		1.415	7.47	
	-12.6	0.729	3.85	
0.736		3.88		
0.729		3.85		

Activation parameters: Benzaldehyde **1a** and hexan-1-amine **2** in [Bmpy][N(CF₃SO₂)₂] **5** / acetonitrile mixtures

Mole Fraction [Bmpy][N(CF ₃ SO ₂) ₂] 5 in acetonitrile	ΔH^\ddagger / kJ mol ⁻¹	ΔS^\ddagger / J K ⁻¹ mol ⁻¹
0	27.2 (2.5)	-272.1 (8.6)
0.24	37.3 (0.6)	-218.7 (2.4)
0.94	45.3 (1.7)	-180.2 (6.2)

Eyring Data: [Bm₂im][N(CF₃SO₂)₂] **6** in acetonitrile

Mole Fraction of 6	Temperature / °C	[Nu] / mol L ⁻¹	<i>k</i> _{obs} / 10 ⁻³ s ⁻¹	<i>k</i> ₂ / 10 ⁻³ L mol ⁻¹ s ⁻¹			
0.20	16.0	0.112	0.592	5.29			
			0.612	5.46			
			0.578	5.16			
	8.1		0.378	3.38			
			0.378	3.37			
			0.365	3.26			
	-8.1		0.143	1.28			
			0.139	1.24			
			0.136	1.21			
	0.0		0.0	0.219	1.95		
				0.212	1.90		
				0.222	1.98		
				0.95	8.1	1.045	6.64
						1.062	6.76
1.145		7.28					
1.1		0.755			4.80		
	0.763	4.85					
	0.798	5.08					
-5	-5	0.533	3.39				
		0.563	3.58				
		0.513	3.26				
		0.313	1.99				
-12.5	-12.5	0.312	1.99				
		0.315	2.00				

Activation parameters: Benzaldehyde **1a** and hexan-1-amine **2** in [Bm₂im][N(CF₃SO₂)₂] **6** / acetonitrile mixtures

Mole Fraction [bm ₂ im][N(CF ₃ SO ₂) ₂] 6 in acetonitrile	ΔH [‡] / kJ mol ⁻¹	ΔS [‡] / J K ⁻¹ mol ⁻¹
0	27.2 (2.5)	-272.1 (8.6)
0.20	34.1 (0.8)	-234.9 (2.7)
0.95	32.3 (1.1)	-235.2 (4.1)

Eyring Data: [TOA][N(CF₃SO₂)₂] **8**

Mole Fraction of #	Temperature / °C	[Nu] / mol L ⁻¹	<i>k</i> _{obs} / 10 ⁻³ s ⁻¹	<i>k</i> ₂ / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.87	36.1	0.200	0.938	4.70
			1.113	5.57
			1.195	5.99
	41.7		1.233	6.18
			1.558	7.81
			1.570	7.87
	47.4		1.742	8.73
			2.049	10.26
			1.776	8.90
	53.8		2.377	11.91
			2.432	12.18
			2.110	10.57

Activation parameters: Benzaldehyde and hexan-1-amine **2** in [TOA][N(CF₃SO₂)₂] **8**

Mole Fraction [TOA][N(CF ₃ SO ₂) ₂] 8	Δ <i>H</i> [‡] / kJ mol ⁻¹	Δ <i>S</i> [‡] / J K ⁻¹ mol ⁻¹
0	27.2 (2.5)	-272.1 (8.6)
0.87	30.8 (3.7)	-254.1 (11.8)

Activation Parameter Summary:

Solvent	Mole Fraction	Δ <i>H</i> [‡] / kJ mol ⁻¹	Δ <i>S</i> [‡] / J K ⁻¹ mol ⁻¹
Acetonitrile	-	27.2 (2.5)	-272.1 (8.6)
[TOA][N(CF ₃ SO ₂) ₂] 8	0.87	30.8 (3.7)	-254.1 (11.8)
[Bm ₂ im][N(CF ₃ SO ₂) ₂] 6	0.20	34.1 (0.8)	-234.9 (2.7)
	0.95	32.3 (1.1)	-235.2 (4.1)
[Bmim][N(CF ₃ SO ₂) ₂] 4	0.23	34.5 (0.5)	-233.2 (1.7)
	0.50	32.4 (0.8)	-237.0 (2.8)
	0.97	32.4 (1.3)	-235.2 (4.6)
[Bmpy][N(CF ₃ SO ₂) ₂] 5	0.24	37.3 (0.6)	-218.7 (2.4)
	0.94	45.3 (1.7)	-180.2 (6.2)

4-Methoxybenzaldehyde **1b** and hexan-1-amine **2**

Eyring Data: Acetonitrile

Temperature / °C	[Nu] / mol L ⁻¹	k_{obs} / 10 ⁻³ s ⁻¹	k_2 / 10 ⁻³ L mol ⁻¹ s ⁻¹
8.1	0.196	0.0303	0.154
		0.0232	0.118
63.6	0.153	0.0185	0.121
		0.0708	0.463
		0.0744	0.486
		0.0831	0.543
47.8		0.0574	0.375
		0.0493	0.322
		0.0552	0.361
36.6		0.0349	0.228
		0.0461	0.301
		0.0388	0.254
54.0		0.0644	0.421
		0.0672	0.439
		0.0621	0.406

Eyring Data: [Bmim][N(CF₃SO₂)₂] **4**

Mole Fraction of 4	Temperature / °C	[Nu] / mol L ⁻¹	k_{obs} / 10 ⁻³ s ⁻¹	k_2 / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.93	29.4	0.237	0.578	2.44
			0.543	2.29
			0.536	2.26
	36.6		0.930	3.93
			0.932	3.94
			0.858	3.62
	44.3		1.131	4.78
			1.275	5.39
			1.264	5.34
19.3	0.333	1.41		
	0.316	1.33		
	0.329	1.39		

Eyring Data: [Bmpy][N(CF₃SO₂)₂] **5**

Mole Fraction of 5	Temperature / °C	[Nu] / mol L ⁻¹	<i>k</i> _{obs} / 10 ⁻³ s ⁻¹	<i>k</i> ₂ / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.93	22.3	0.227	0.798	3.51
			0.823	3.62
			0.776	3.41
	29.4		1.144	5.03
			1.133	4.98
			1.389	6.11
	14.5		0.450	1.98
			0.511	2.25
			0.471	2.07
	36.6		1.895	8.33
			2.033	8.94

Eyring Data: [Bm₂im][N(CF₃SO₂)₂] **6**

Mole Fraction of 6	Temperature / °C	[Nu] / mol L ⁻¹	<i>k</i> _{obs} / 10 ⁻³ s ⁻¹	<i>k</i> ₂ / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.95	36.6	0.147	0.958	6.52
			1.062	7.22
			1.010	6.87
	48.1		1.934	13.16
			1.832	12.46
			1.921	13.07
	17.7		0.476	3.24
			0.442	3.01
			0.446	3.04
	63.6		4.146	28.21
			4.191	28.52
			3.592	24.45

Eyring Data: [Bm₄im][N(CF₃SO₂)₂] **7**

Mole Fraction of 7	Temperature / °C	[Nu] / mol L ⁻¹	<i>k</i> _{obs} / 10 ⁻³ s ⁻¹	<i>k</i> ₂ / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.94	36.6	0.171	5.707	33.38
			5.561	32.52
			5.595	32.72
	41.6		7.248	42.40
			6.519	38.13
			6.536	38.23
	25.9		3.477	20.33
			3.541	20.71
			3.426	20.04
			3.426	20.04
	30.5		4.241	24.80
			4.194	24.53
			3.969	23.22

Eyring Data: [TOA][N(CF₃SO₂)₂] **8**

Mole Fraction of 8	Temperature / °C	[Nu] / mol L ⁻¹	<i>k</i> _{obs} / 10 ⁻³ s ⁻¹	<i>k</i> ₂ / 10 ⁻³ L mol ⁻¹ s ⁻¹
0.87	54.0	0.165	0.349	2.12
			0.476	2.70
	63.6		0.394	2.29
			0.524	3.05
			0.578	3.36
			0.532	3.10
	47.8		0.388	2.21
			0.448	2.60
			0.433	2.52
			0.433	2.52
	36.6		0.235	1.37
			0.263	1.53
			0.323	1.88

Activation Parameter Summary:

Solvent	Mole Fraction	$\Delta H^\ddagger / \text{kJ mol}^{-1}$	$\Delta S^\ddagger / \text{J K}^{-1} \text{mol}^{-1}$
Acetonitrile	-	14.2 (1.0)	-333.1 (3.3)
[TOA][N(CF ₃ SO ₂) ₂] 8	0.87	15.8 (3.2)	-312.9 (10.0)
[Bm ₄ im][N(CF ₃ SO ₂) ₂] 7	0.94	28.8 (1.4)	-246.1 (4.6)
[Bm ₂ im][N(CF ₃ SO ₂) ₂] 6	0.95	33.3 (1.3)	-243.4 (4.0)
[Bmim][N(CF ₃ SO ₂) ₂] 4	0.93	36.8 (1.5)	-238.5 (5.0)
[Bmpy][N(CF ₃ SO ₂) ₂] 5	0.93	41.9 (1.9)	-214.8 (6.5)