

Stereoselective Synthesis of 1,2-Diamine Containing Indolines by a Conjugate Addition Nitro-Mannich Reaction

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Supplementary Information

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Copies of ¹ H and ¹³ C NMR spectra:		
4aa	1H..... 13C.....	S6 S7
Table 1		
3aa Entry 1	1H..... 13C	S8 S9
3aa minor diastereoisomer	1H..... 13C	S10 S11
3 Entry 2	1H..... 13C	S12 S13
3 Entry 3	1H..... 13C	S14 S15
3 Entry 4	1H..... 13C	S16 S17
3 Entry 5	1H..... 13C	S18 S19
3 Entry 6	1H..... 13C	S20 S21
3 Entry 7	1H..... 13C	S22 S23
3 Entry 8	1H..... 13C	S24 S25
3 Entry 9	1H.....	S26

	13CS27
7bb	Entry 1 1H.....	.S28
	13CS29
7	Entry 2 1H.....	.S30
	13CS31
7	Entry 3 1H.....	.S32
	13CS33
7	Entry 5 1H.....	.S34
	13CS35
7	Entry 7 1H.....	.S36
	13CS37
7	Entry 8 1H.....	.S38
	13CS39
7	Entry 9 1H.....	.S40
	13CS41
9	1H.....	.S42
	13CS43
10	1H.....	.S44
	13CS45

General experimental

Unless otherwise stated, all reactions were carried out under an atmosphere of nitrogen. All glassware was flame dried and allowed to cool under a stream of nitrogen before use. Cooling to 0 °C was effected using an ice-water bath. Cooling to temperatures below 0 °C was effected using dry ice-acetone mixtures. Reactions were monitored by thin layer chromatography (TLC) using Polygram® SIL G/UV₂₅₄ 0.25 mm silica gel precoated plastic plates with fluorescent indicator. Sheets were visualised using ultra-violet light (254 nm) and/or anisaldehyde or KMnO₄ solutions, as appropriate. Flash column chromatography was performed using Fluorochrom silica gel 60, 35-70 µ. The liquid phase was analytical grade 40-60 petroleum ether (petrol) and ethyl acetate (EtOAc) unless otherwise noted.

Purification of Solvents and Reagents:

Commercial solvents and reagents were used as supplied or purified in accordance with standard procedures, as described below.

Tetrahydrofuran (THF) was pre-dried over sodium wire and distilled under an atmosphere of dry nitrogen from sodium benzophenone ketal or obtained from a solvent tower, where degassed THF was passed through two columns of activated alumina and a 7 micron filter under 4 bar pressure.

Diethyl ether (Et₂O) was pre-dried over sodium wire and distilled under an atmosphere of dry nitrogen from sodium benzophenone ketal or obtained from a solvent tower, where degassed Et₂O was passed through two columns of activated alumina and a 7 micron filter under 4 bar pressure.

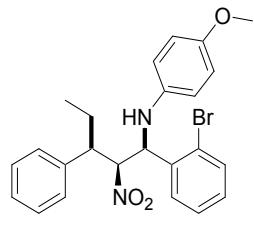
Toluene was obtained from a solvent tower, where degassed toluene was passed through two columns of activated alumina and a 7 micron filter under 4 bar pressure. Dichloromethane was distilled from calcium hydride powder or purchased as an analytical grade and stored over 4Å molecular sieves. Anhydrous MeCN was used as supplied.

Characterization:

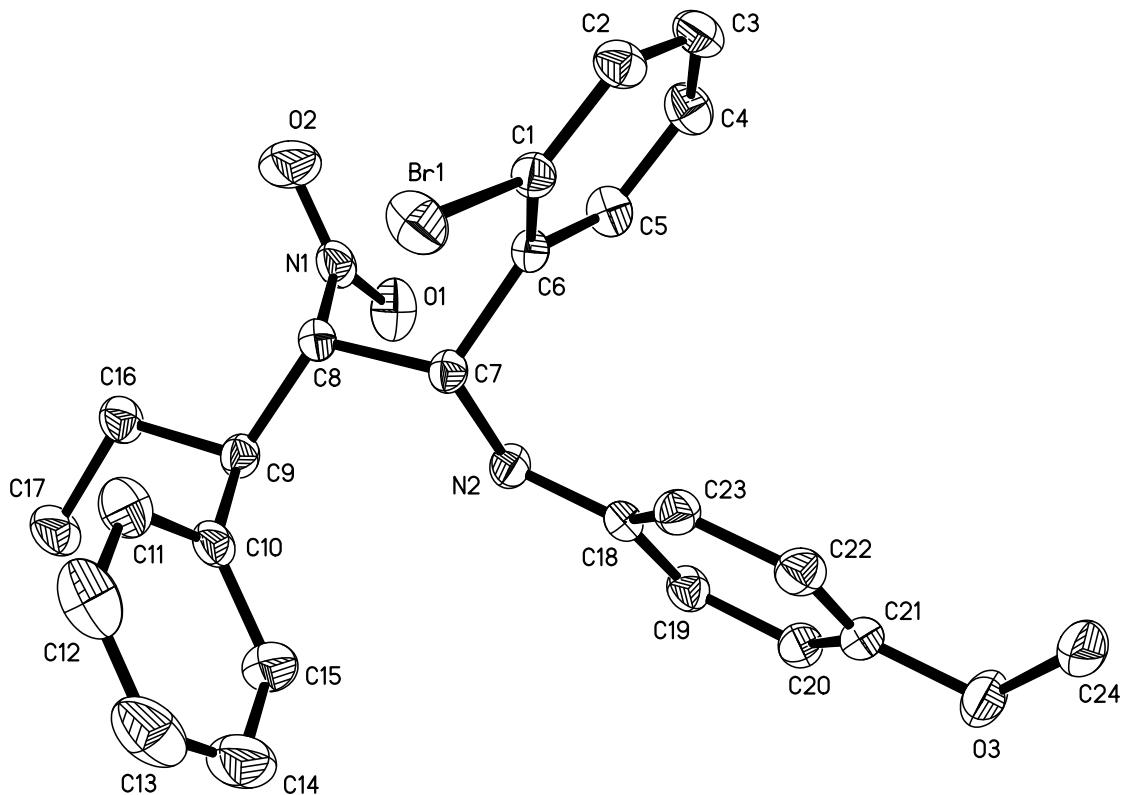
Melting points are uncorrected. Infrared spectra were recorded as a thin film on sodium chloride discs or as dilute chloroform solutions. ¹H and ¹³C NMR spectra were recorded as dilute solutions in deuteriochloroform unless otherwise stated. All chemical shifts (δ) are reported in parts per million (ppm) relative to residual solvent peaks. All chemical shifts are reported relative to chloroform ($\delta_H = 7.27$ ppm, $\delta_C = 77.1$ ppm). Coupling constants (J) are reported in Hertz and are recorded as observed in the spectrum without averaging. The multiplicity of an ¹H signal is designated by the following abbreviations: m = multiplet, s = singlet, d = doublet, t = triplet, q = quartet and quin = quintet, br = broad signal. ¹³C

multiplicities were assigned using a DEPT sequence. Where appropriate, HMQC and NOE experiments were carried out to aid assignment. For mass spectrometry, high resolution (HR) electrospray ionization (ES^+) mass data was obtained using a time-of-flight (TOF) analyzer, and HR chemical ionization (CI) or electron impact (EI) mass data were recorded using an ion trap.

Structure and X-ray structure:



3bb

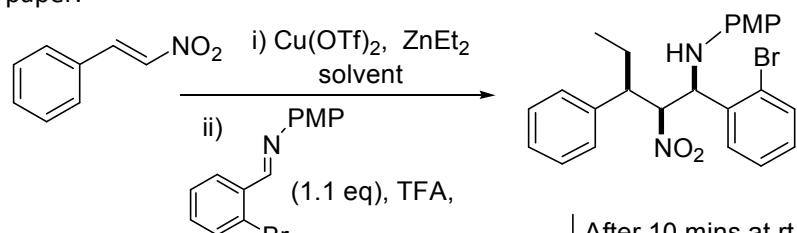


Crystallographic data (excluding structure factors) for this structure has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC 1020679. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-(0)1223-336033 or e-mail: deposit@ccdc.cam.ac.uk).

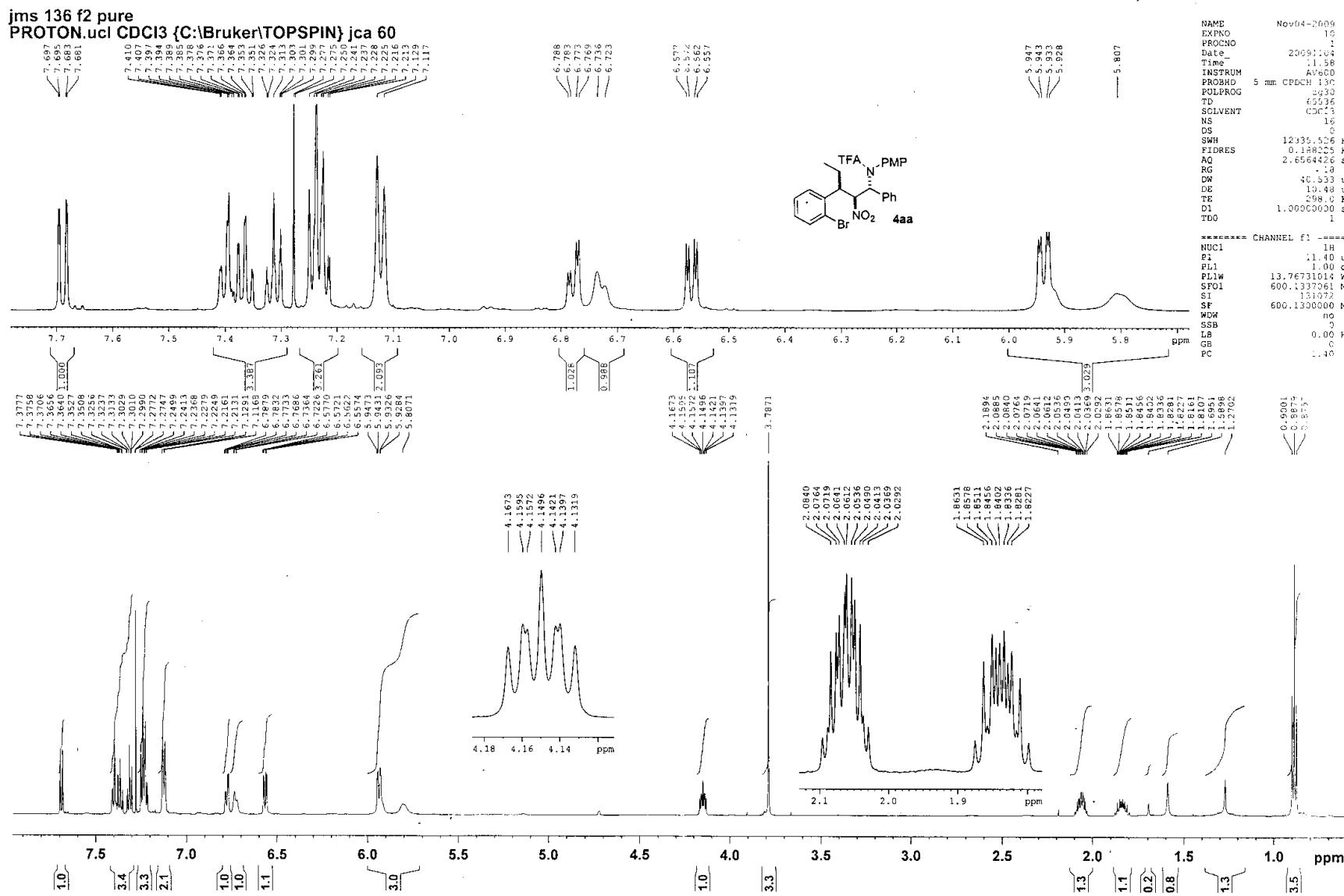
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Formula weight	469.37
Temperature	150(2) K
Radiation, wavelength	MoK α , 0.71073 Å
Crystal system, space group	monoclinic, P2 ₁ /c
Unit cell parameters	$a = 15.139(3)$ Å $\alpha = 90^\circ$ $b = 10.0225(18)$ Å $\beta = 103.866(3)^\circ$ $c = 30.107(6)$ Å $\gamma = 90^\circ$
Cell volume	4435.1(14) Å ³
Z	8
Calculated density	1.406 g/cm ³
Absorption coefficient μ	1.881 mm ⁻¹
F(000)	1936
Crystal colour and size	yellow, 0.48 × 0.42 × 0.14 mm ³
Data collection method	Bruker SMART APEX diffractometer ω/θ scans
θ range for data collection	2.46 to 28.31°
Index ranges	h –20 to 19, k –13 to 13, l –38 to 40
Completeness to $\theta = 26.00^\circ$	99.6 %
Reflections collected	36089
Independent reflections	10436 ($R_{int} = 0.0379$)
Reflections with $F^2 > 2\sigma$	8169
Absorption correction	semi-empirical from equivalents
Min. and max. transmission	0.4654 and 0.7787
Structure solution	direct methods
Refinement method	Full-matrix least-squares on F^2
Weighting parameters a, b	0.0640, 0.3959
Data / restraints / parameters	10436 / 0 / 541
Final R indices [$F^2 > 2\sigma$]	$R_1 = 0.0358$, $wR_2 = 0.0953$
R indices (all data)	$R_1 = 0.0504$, $wR_2 = 0.1023$
Goodness-of-fit on F^2	0.971
Largest and mean shift/su	0.001 and 0.000
Largest diff. peak and hole	0.696 and –0.564 e Å ⁻³

Solvent screen

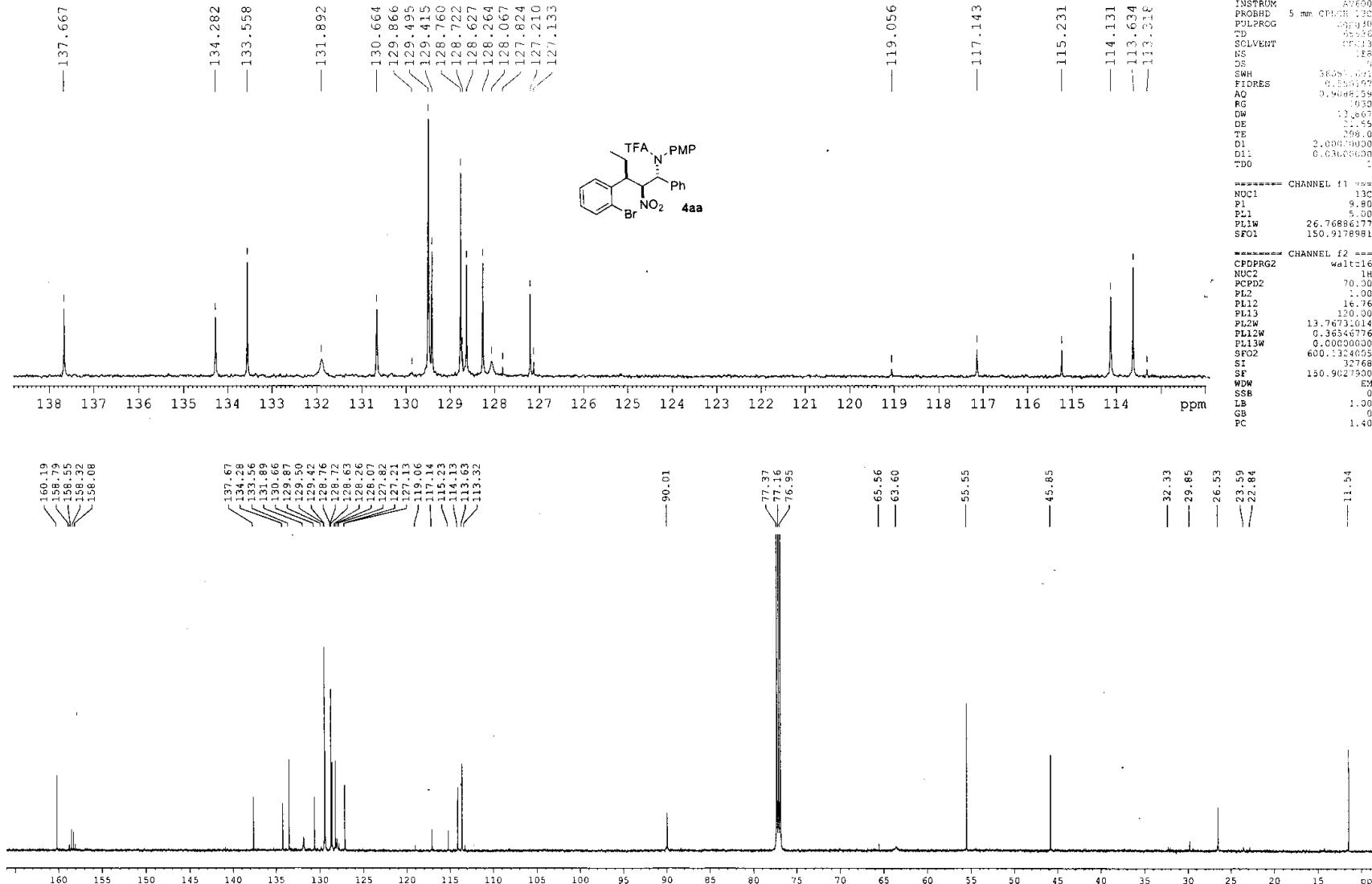
Method used was 'General Procedure for the Synthesis of *syn, syn*-β-Nitroamines 3 (Table 1)' in main paper.



Solvent	After 10 mins at rt
	<i>dr</i> - conversion
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THF	50 : 50 - 100%
Toluene	80 : 20 - 80%
DME	65 : 35 - 100%
DCM	75 : 25 - 100%

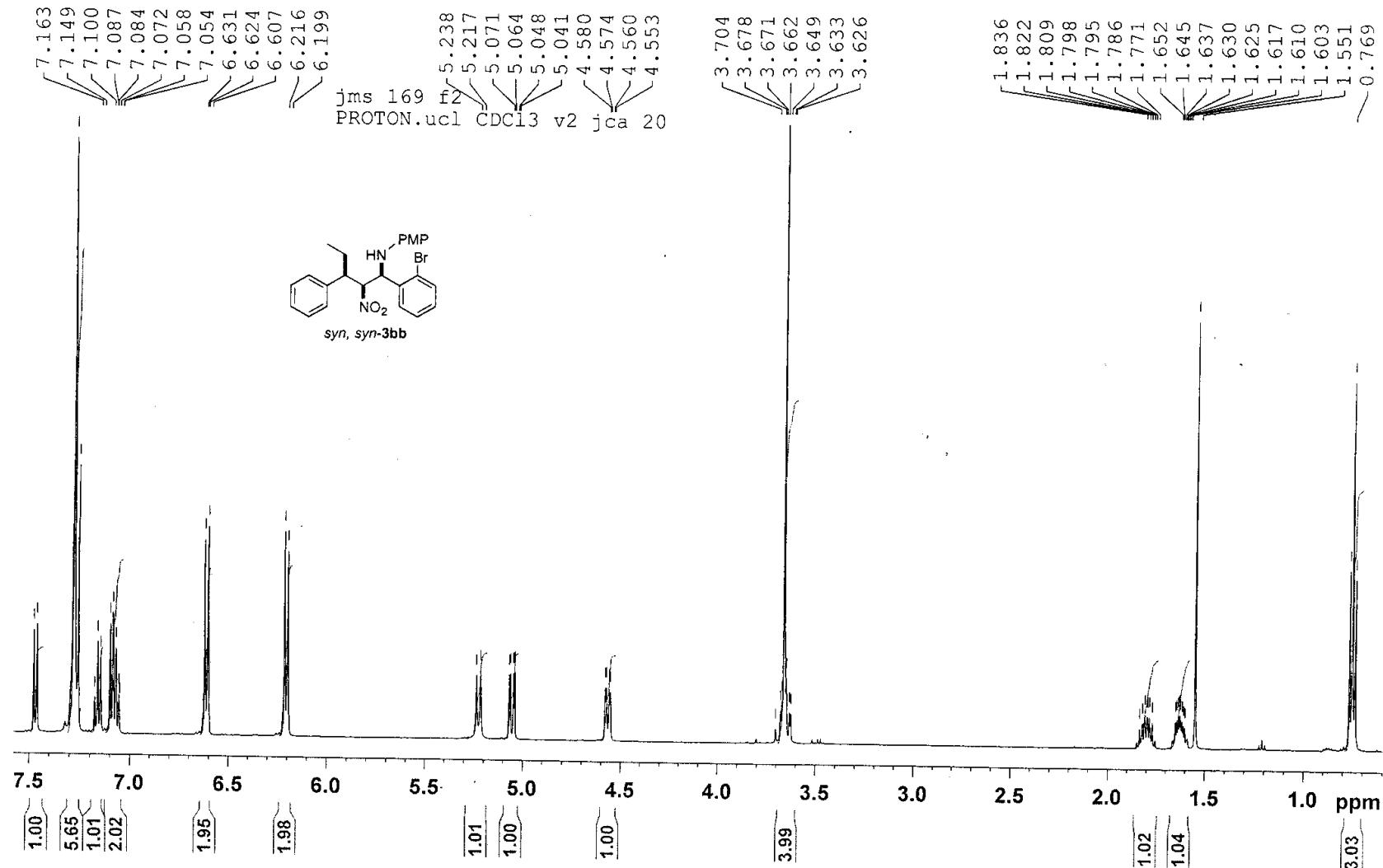


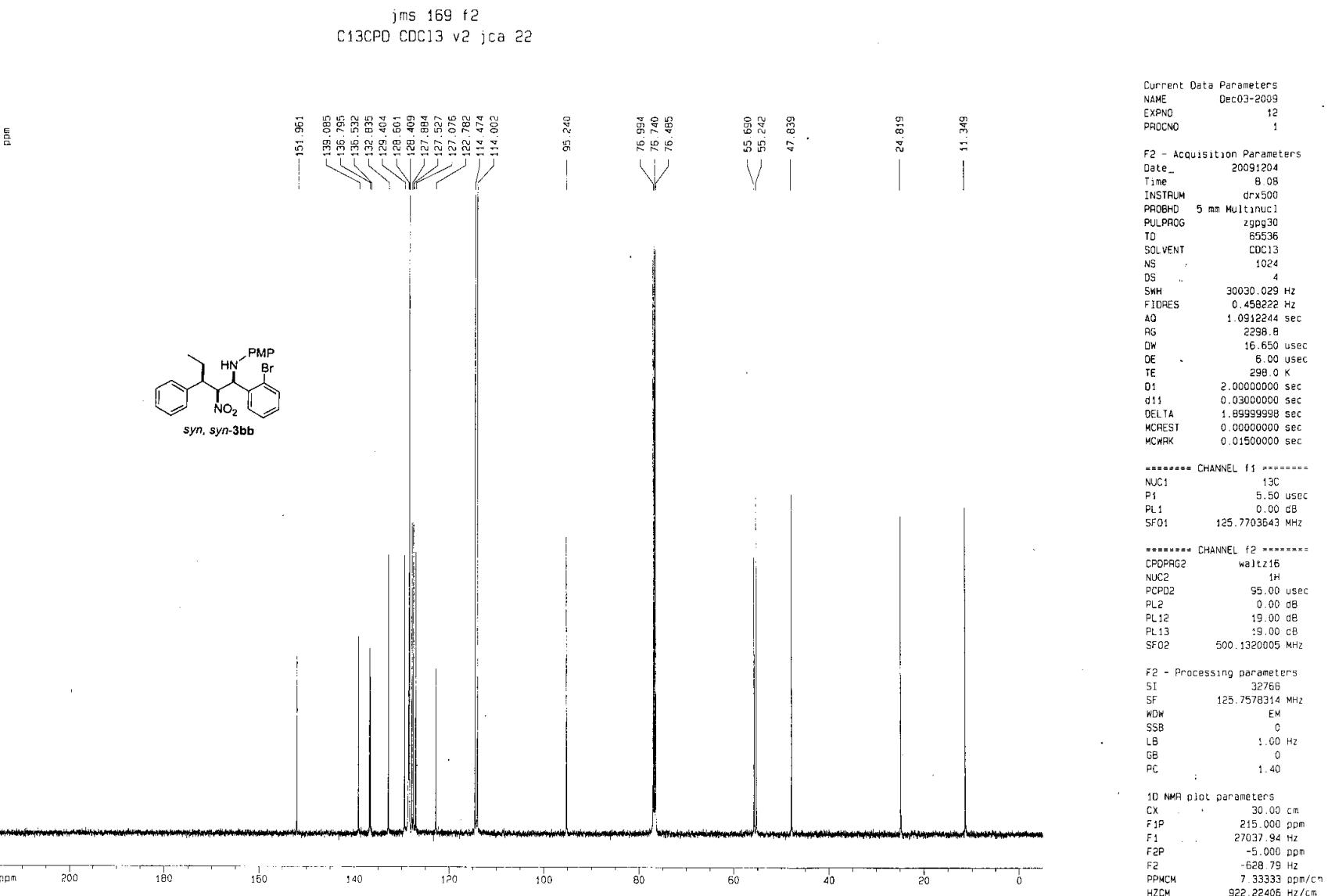
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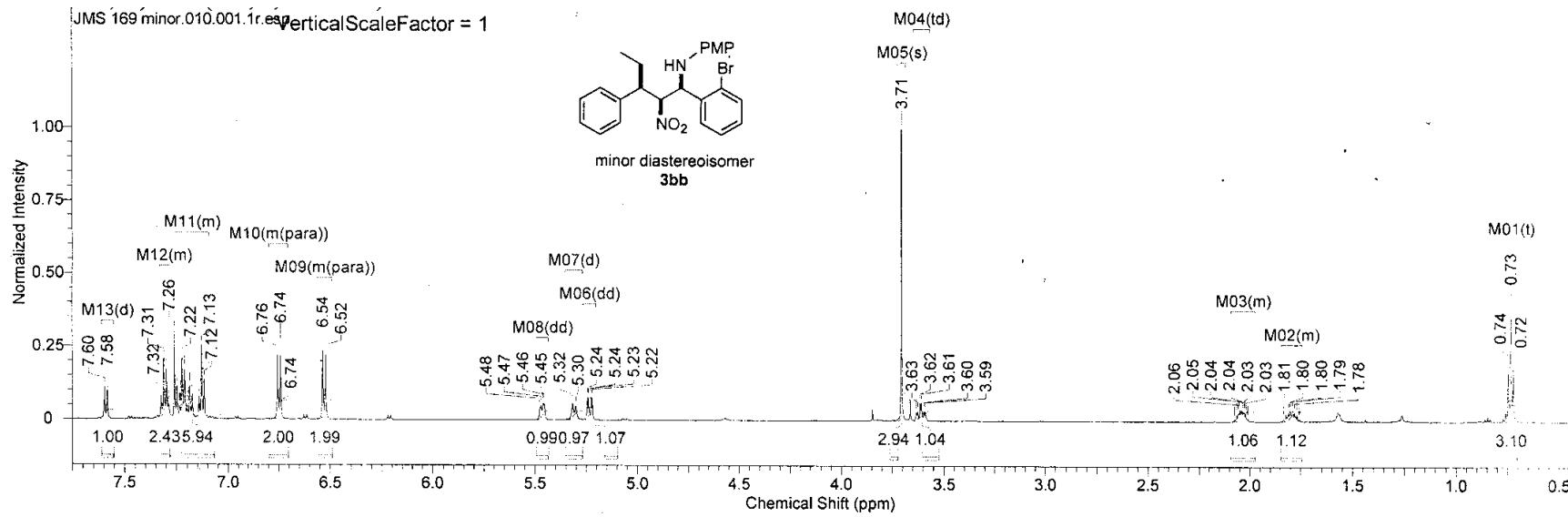


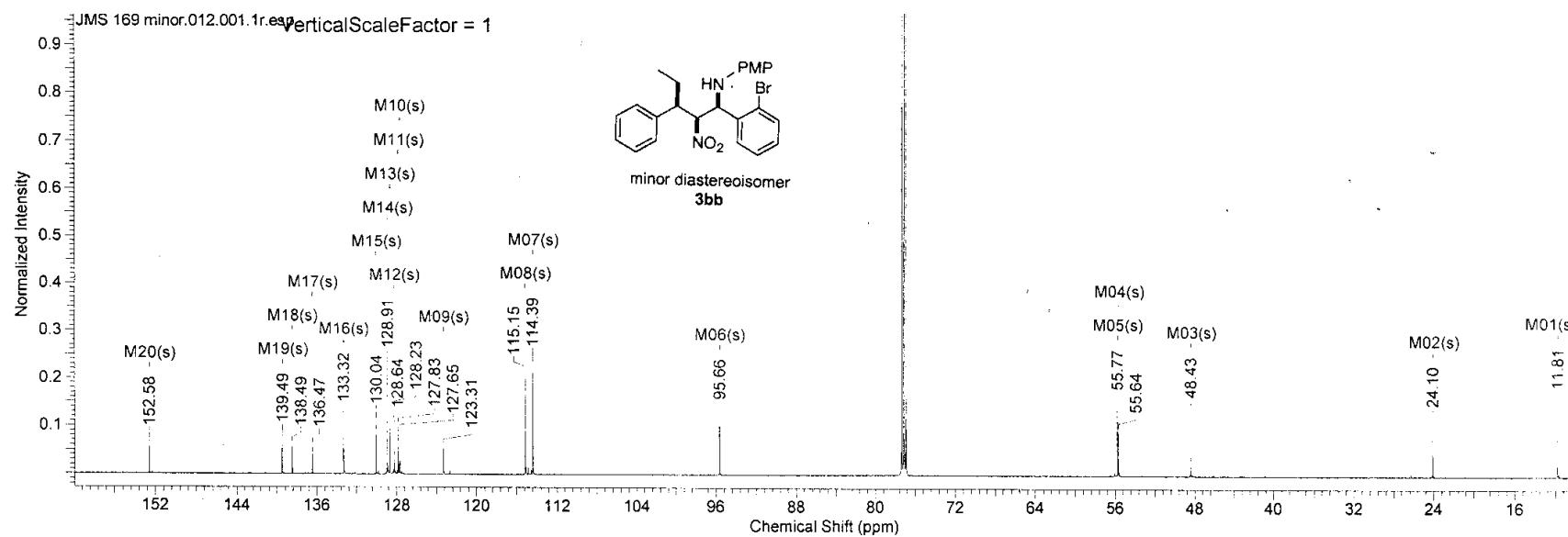
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PL1 5.00 dB
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SFOL 150.9178981 MHz

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CRDPGZ waltz16
NUC2 ¹H
PCPD2 70.00 usec
PL2 1.00 dB
PL12 16.76 dB
PL13 120.00 dB
PL2W 13.76731014 W
PL12W 0.35545764 W
PL13W 0.00030000 W
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LB 1.00 Hz
GB 0
PC 1.40



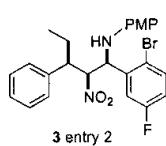
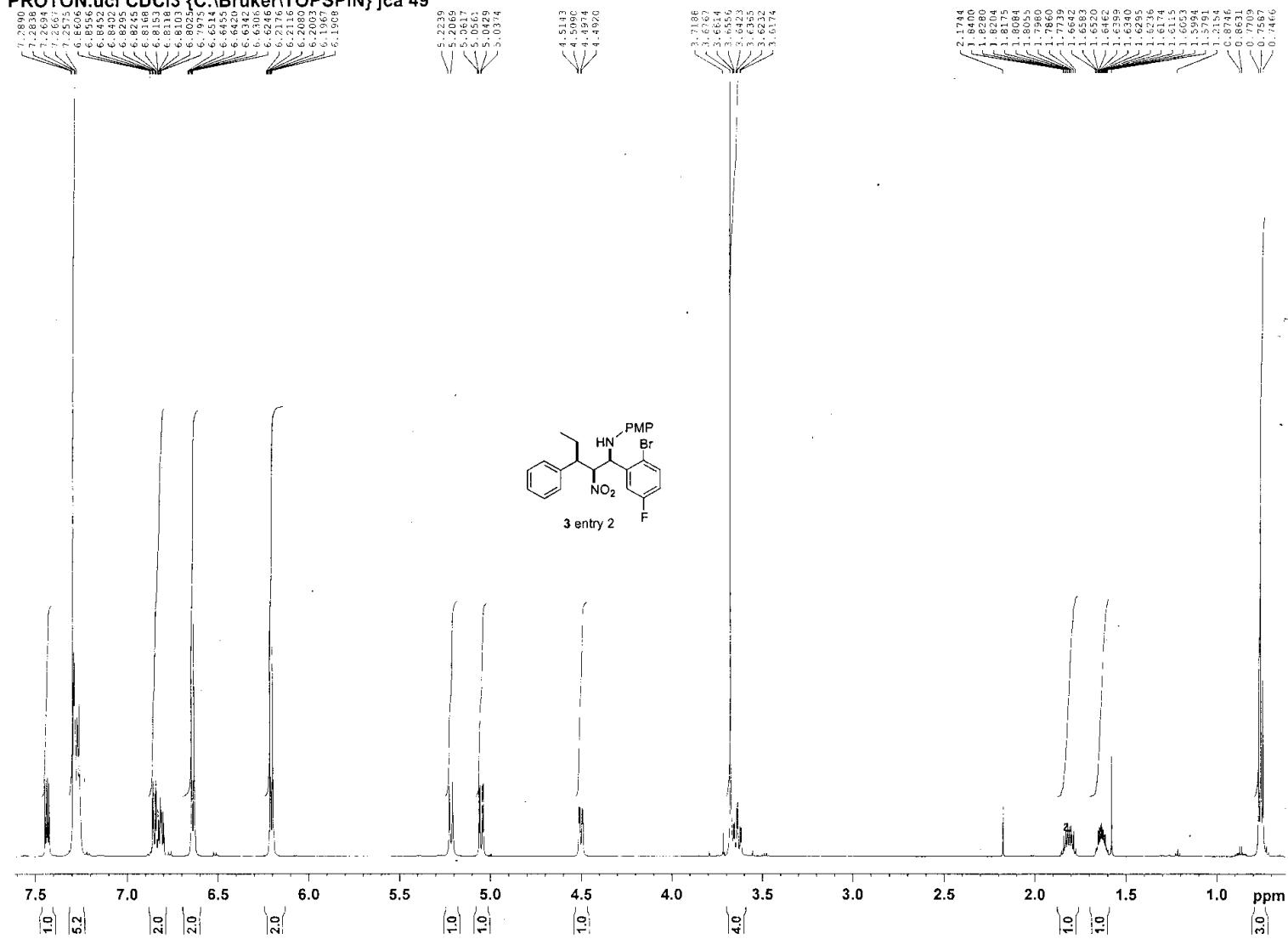






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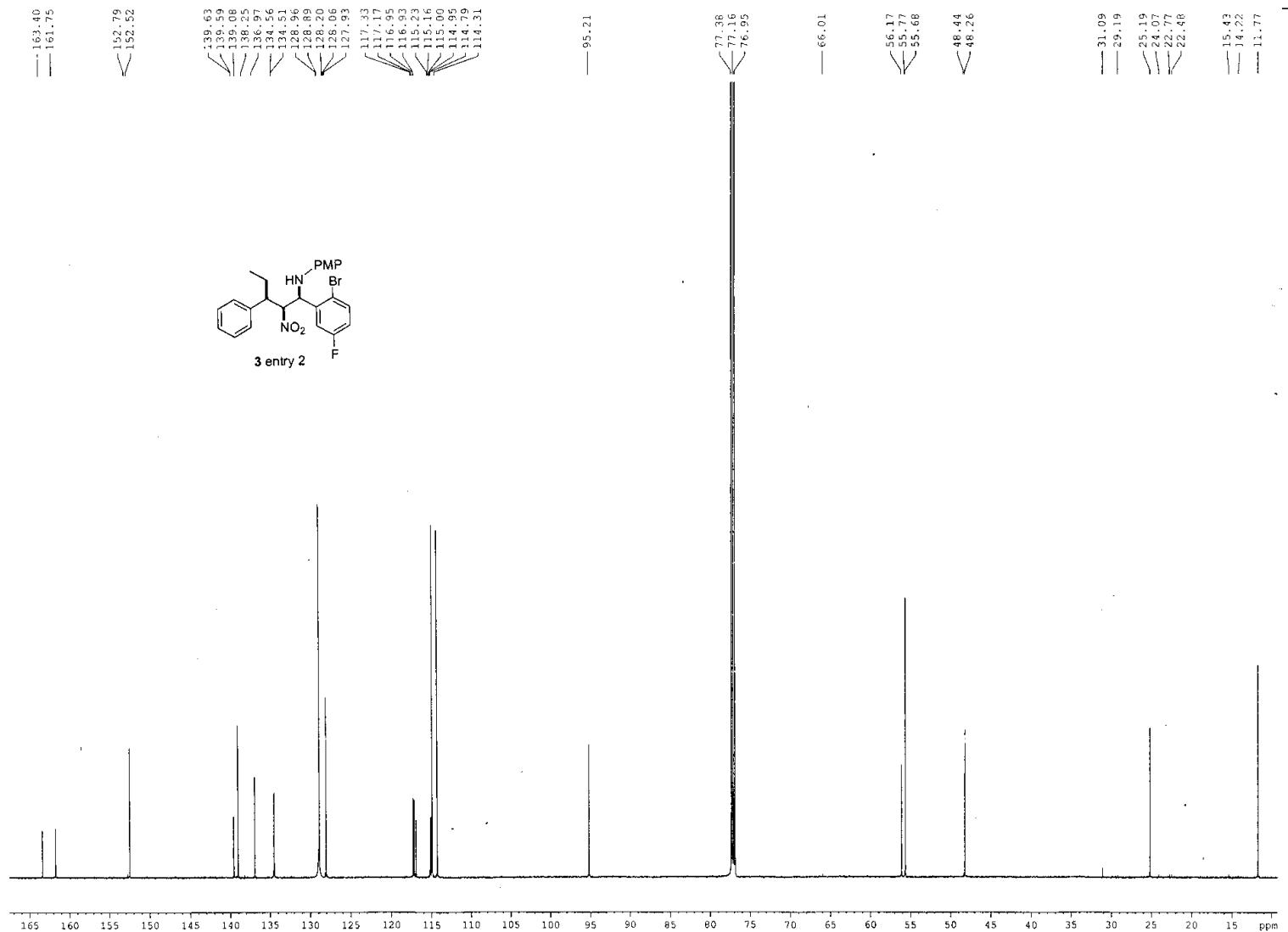
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S-12

jms 245 F2 major
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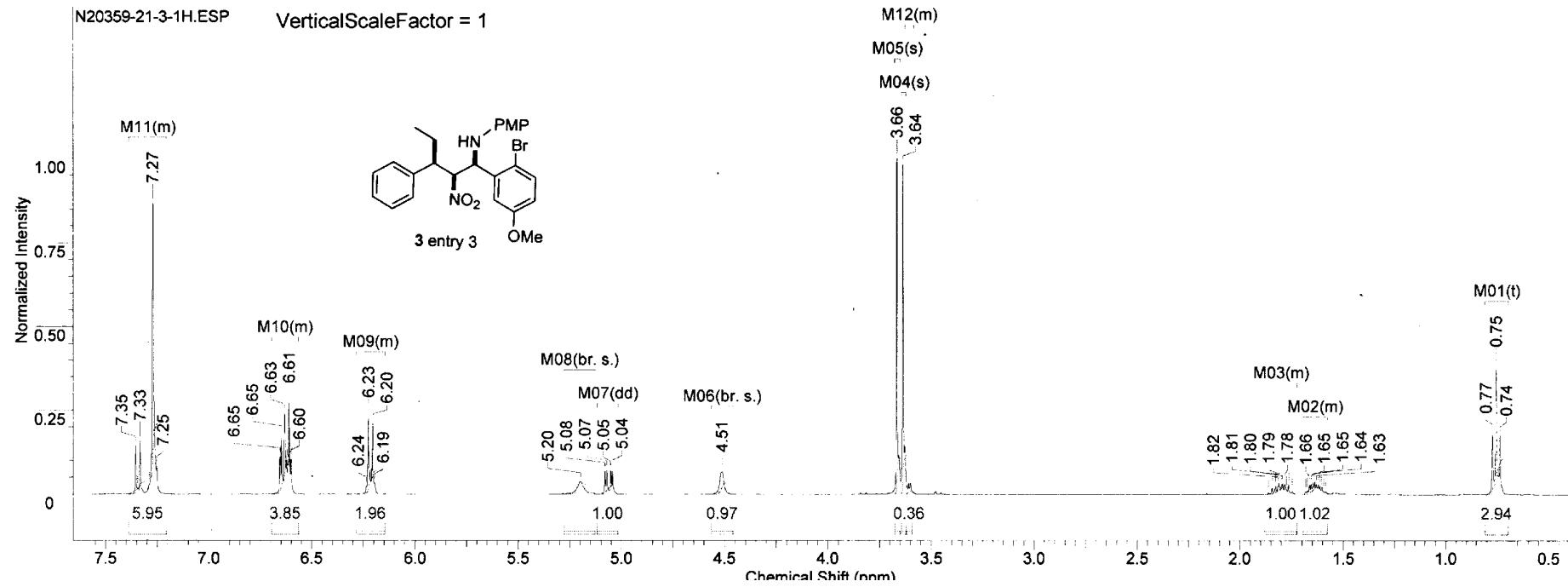
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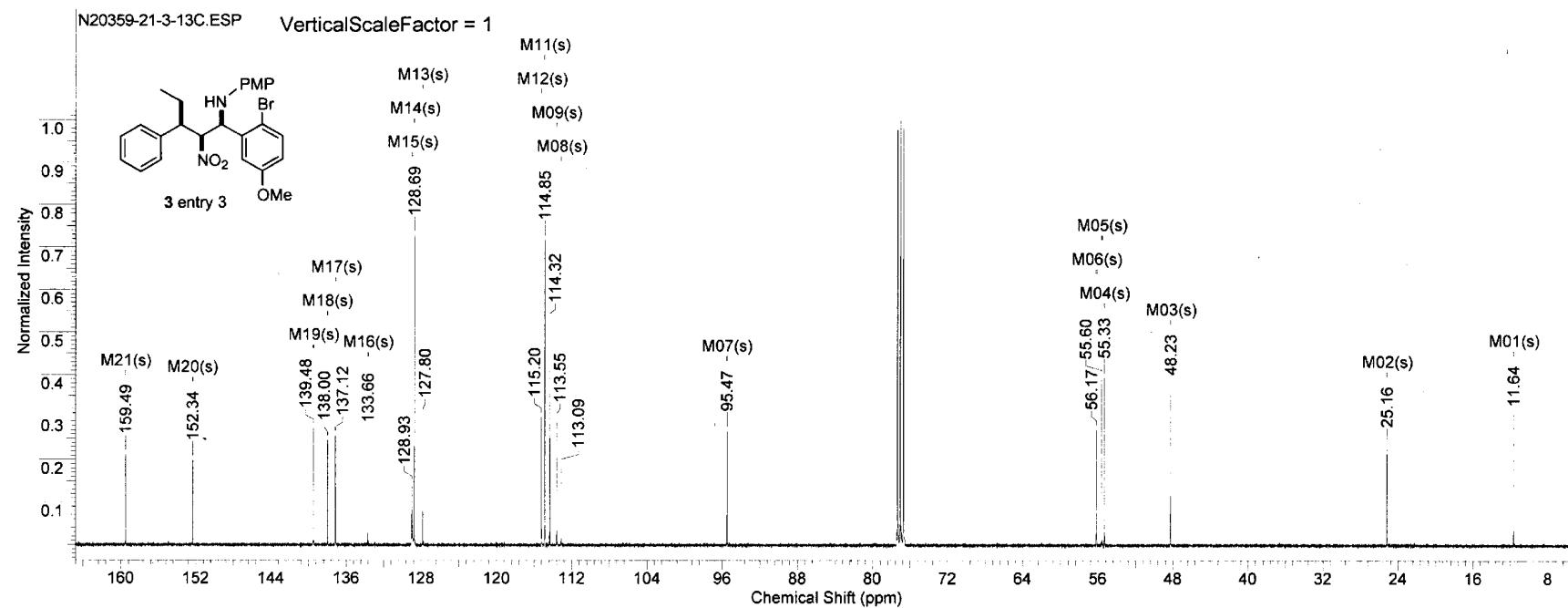


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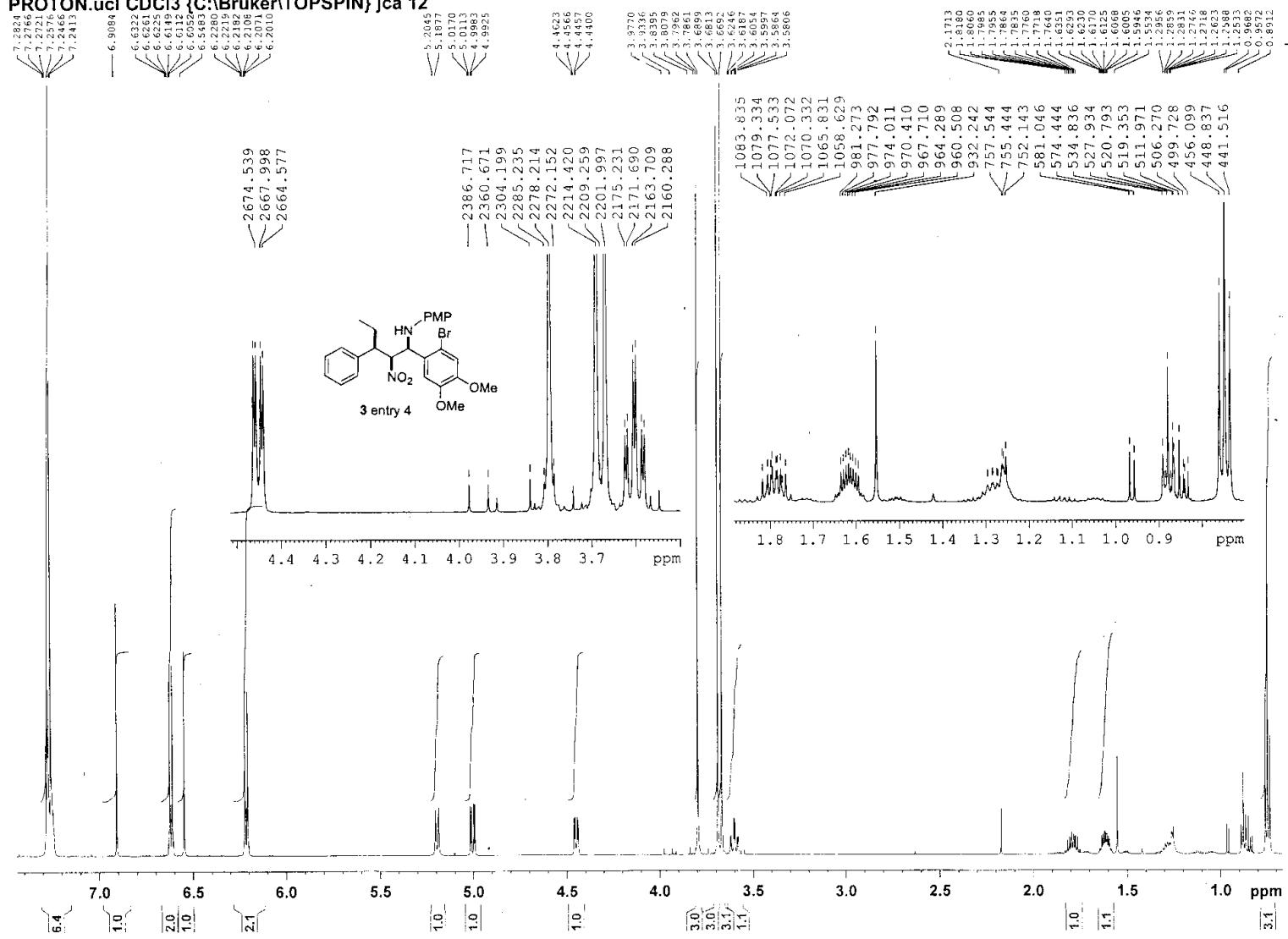
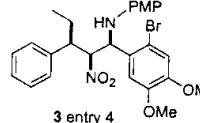
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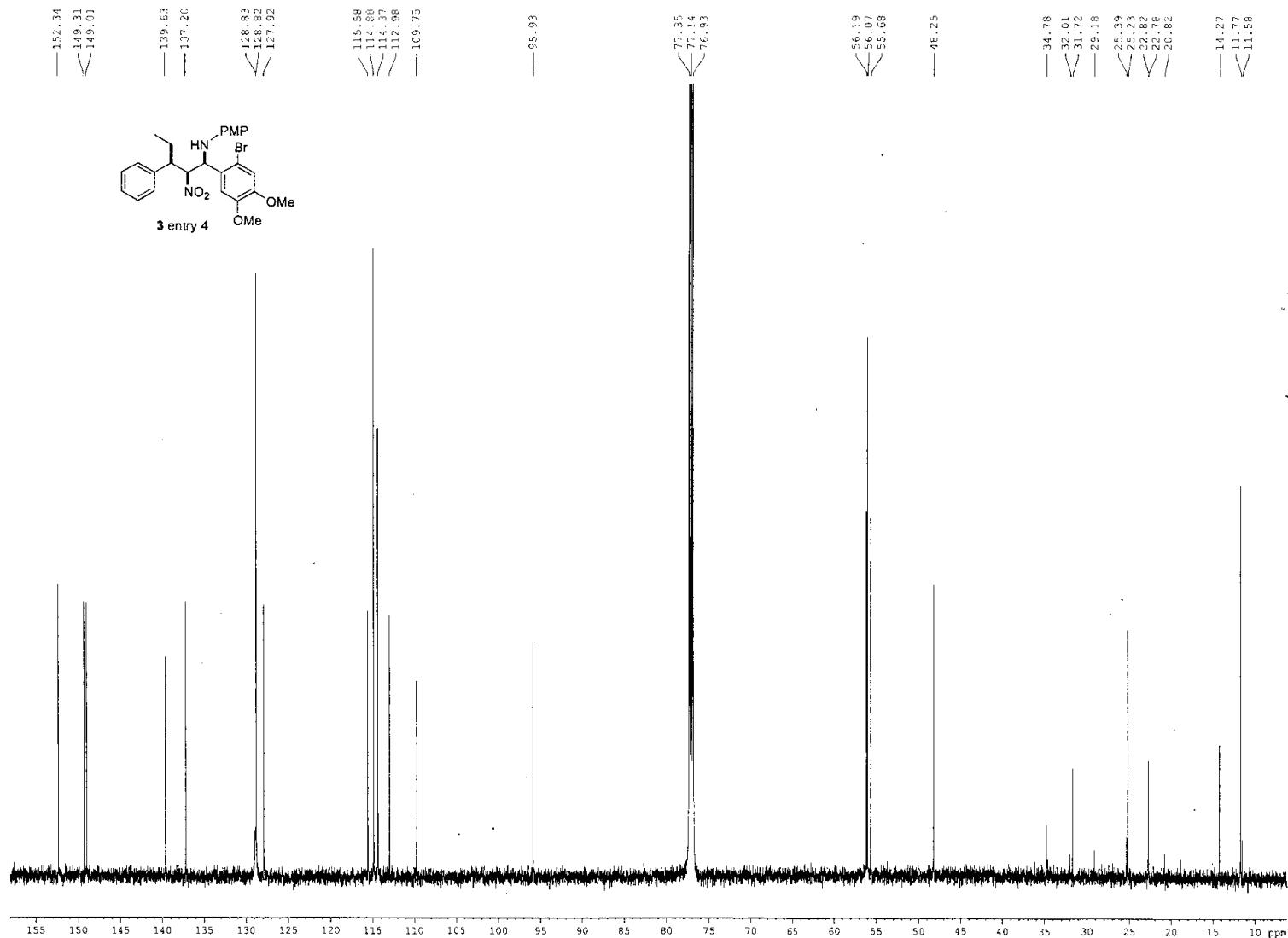
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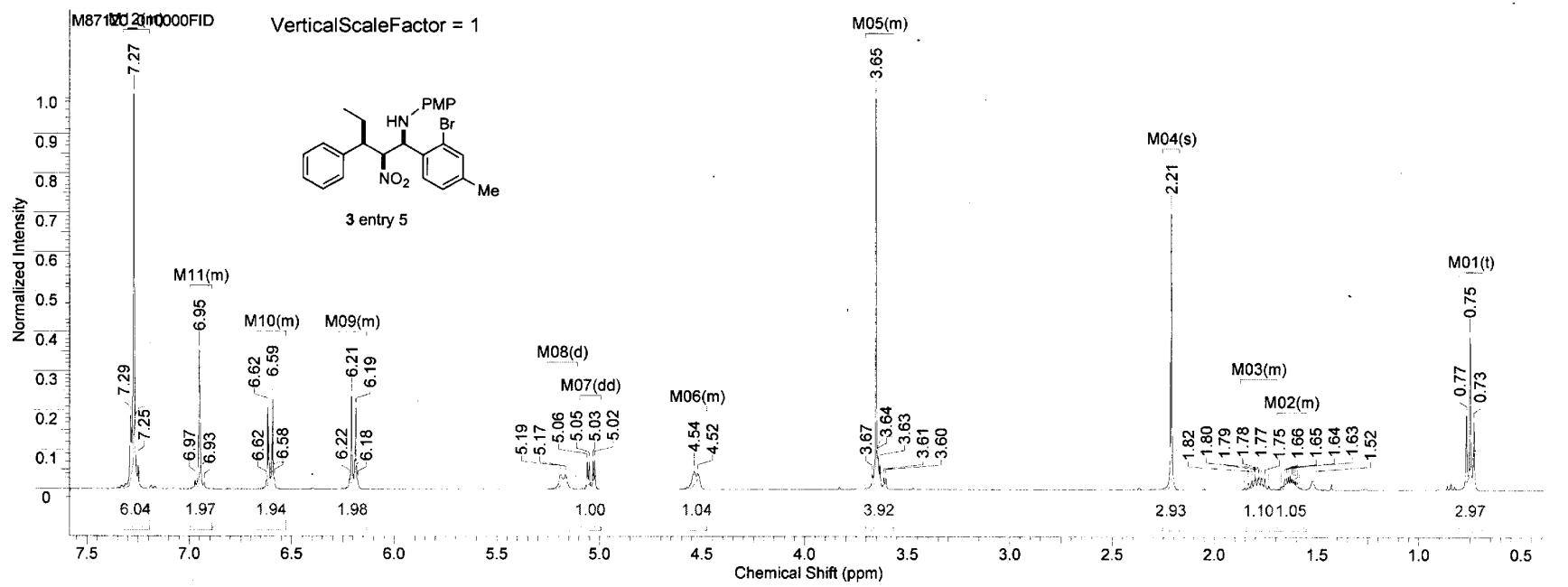
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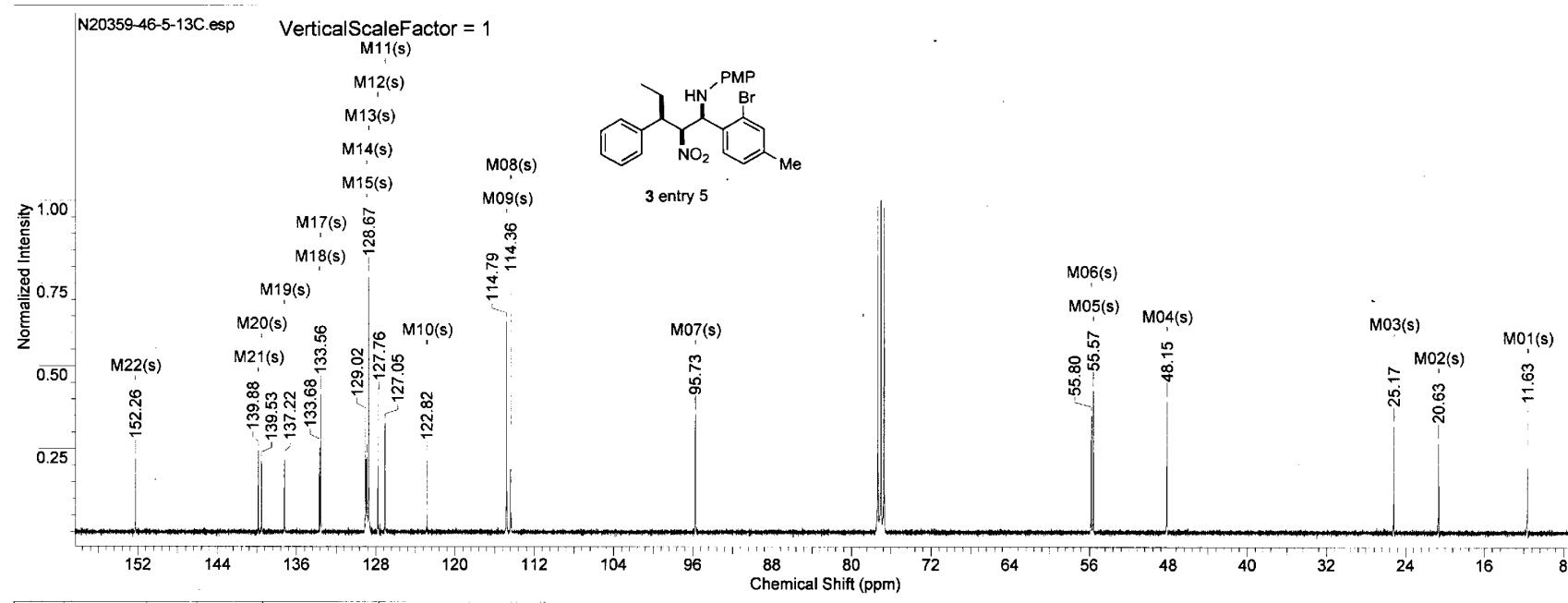
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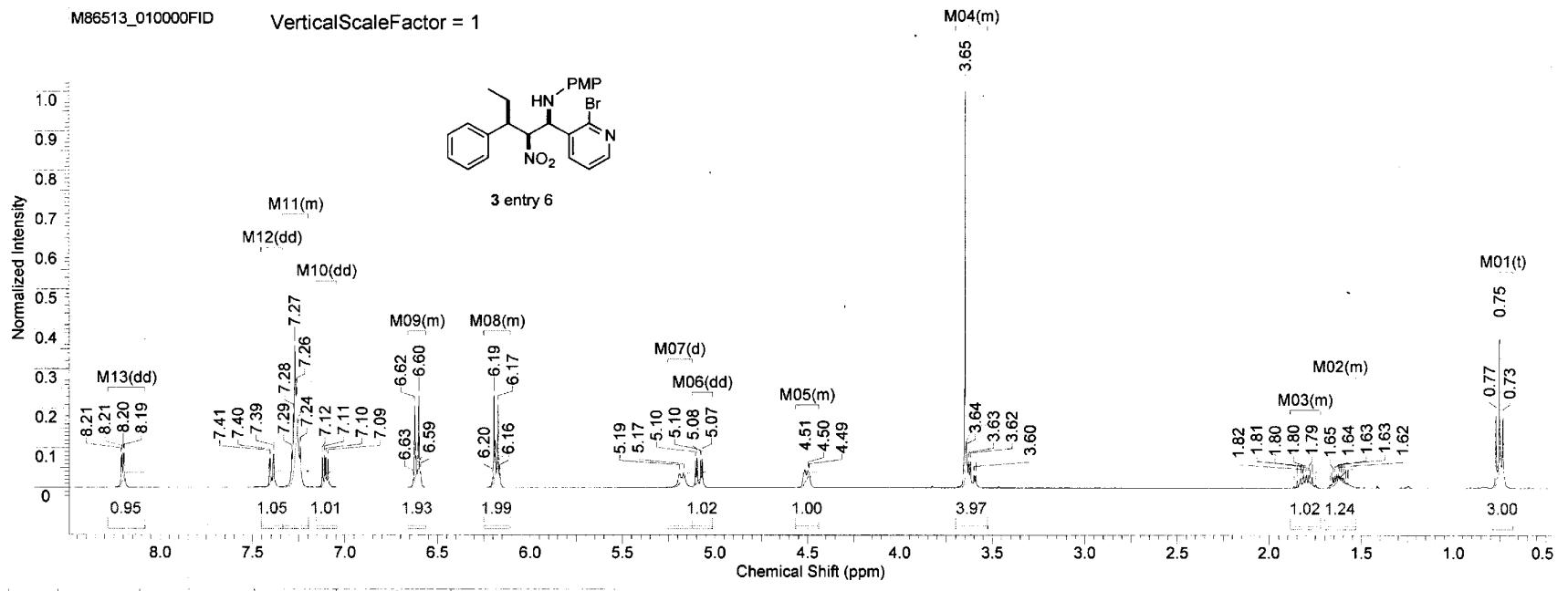
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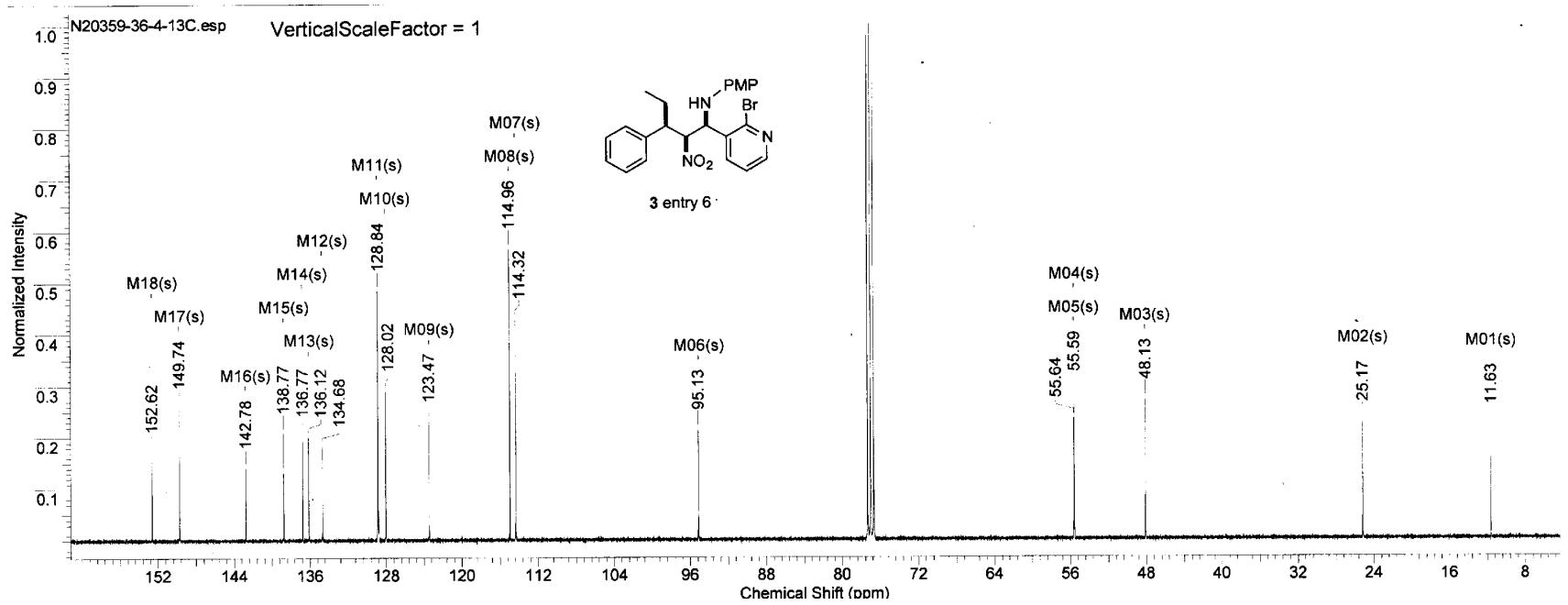
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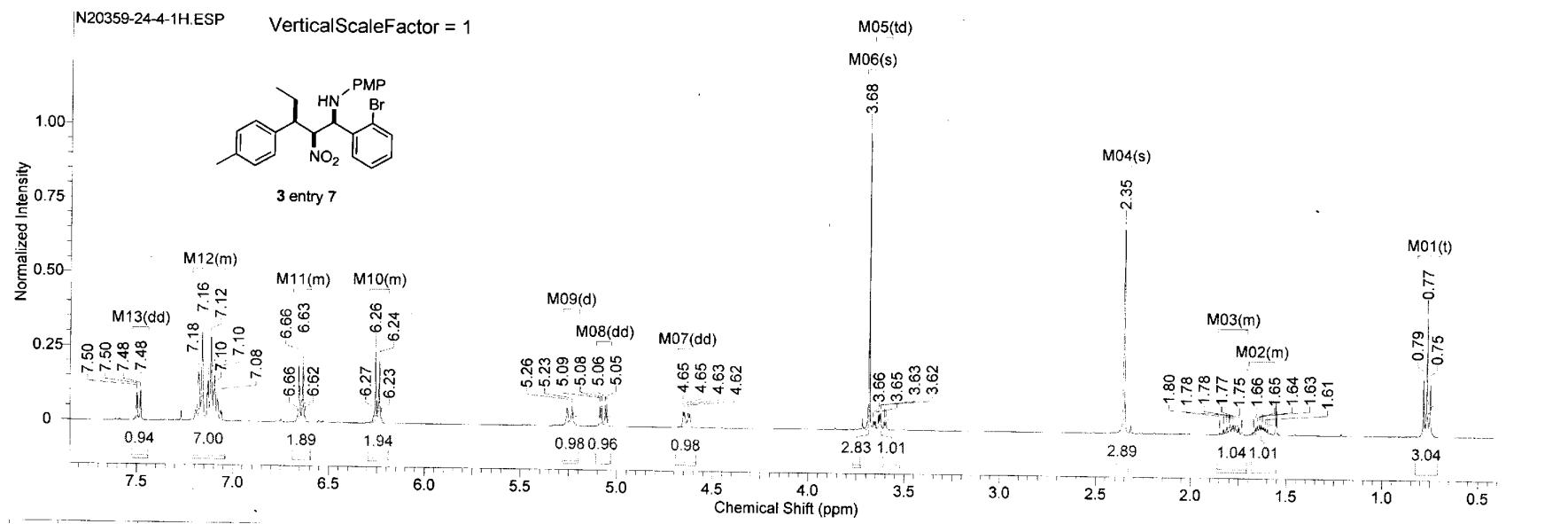


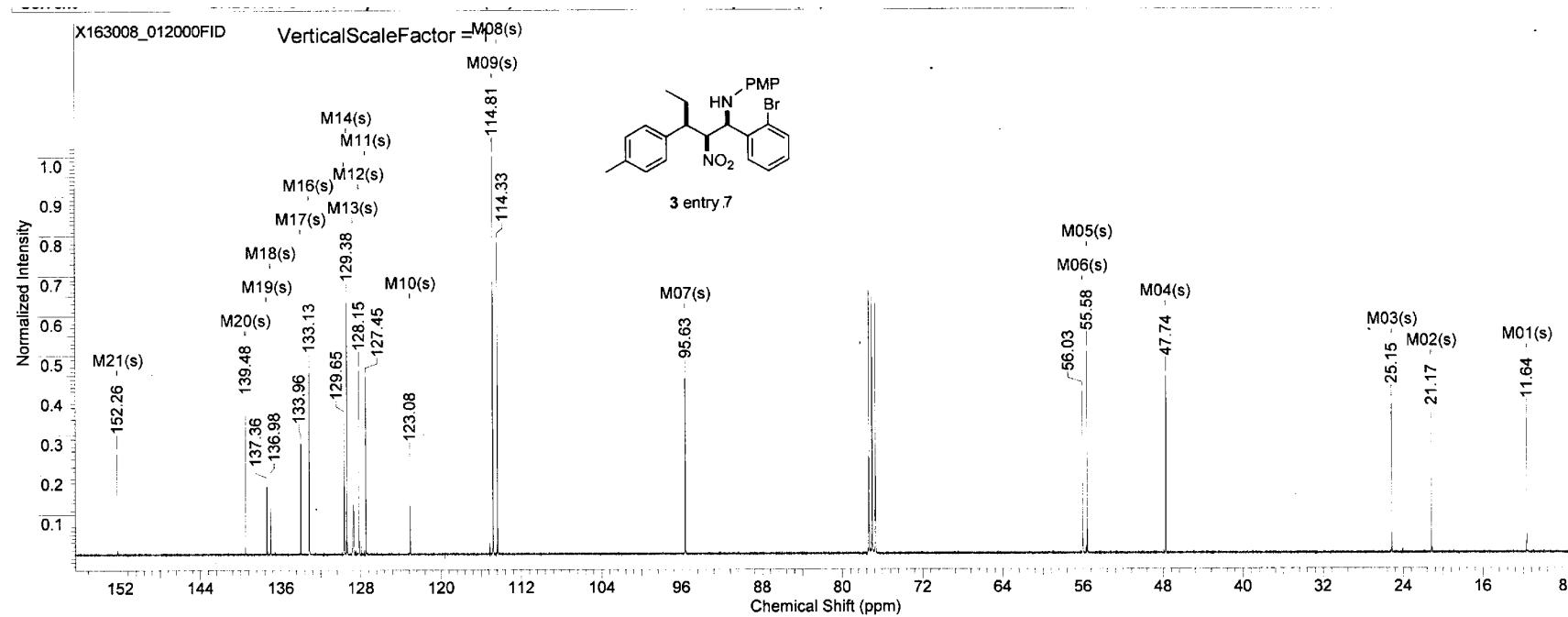


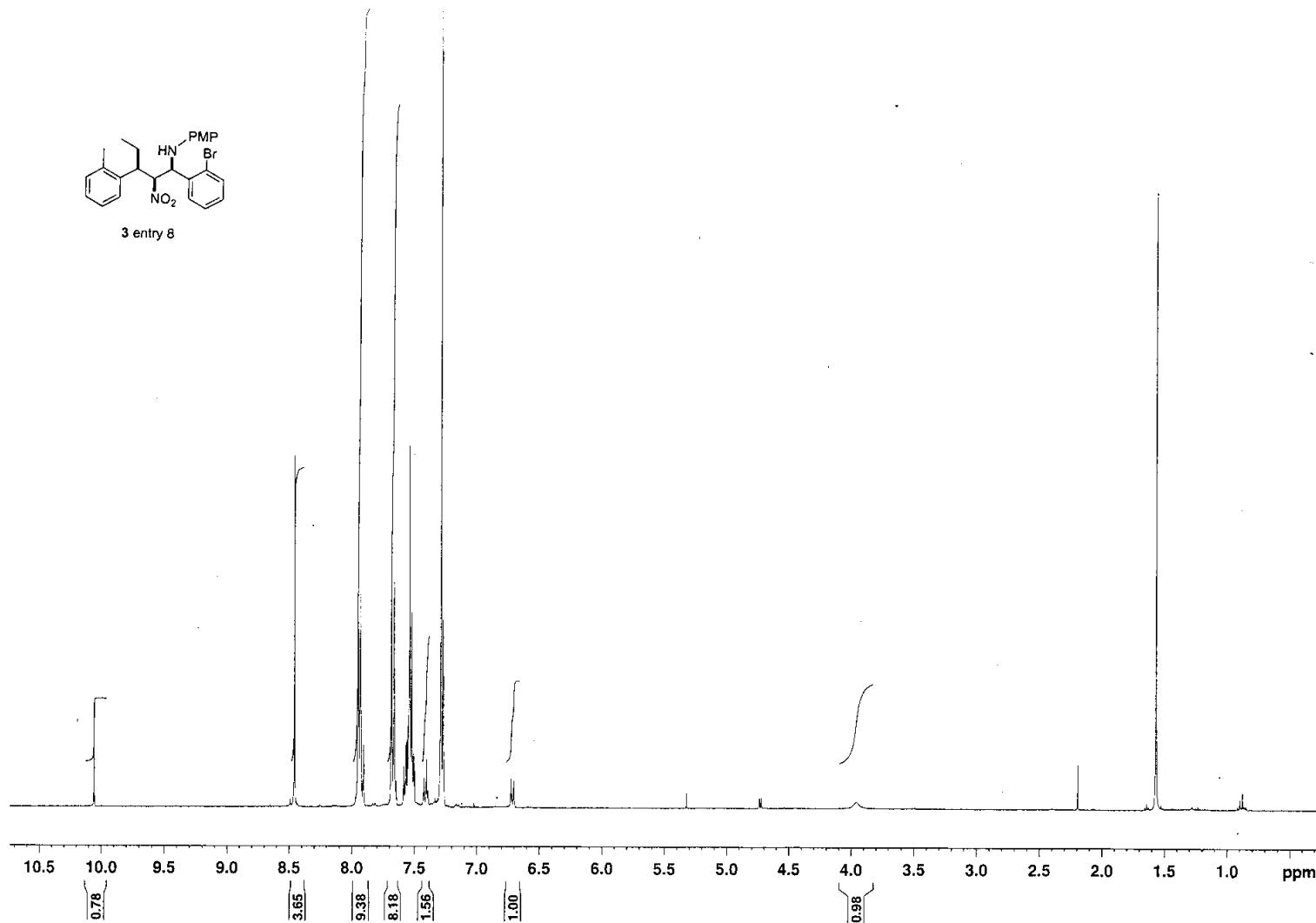




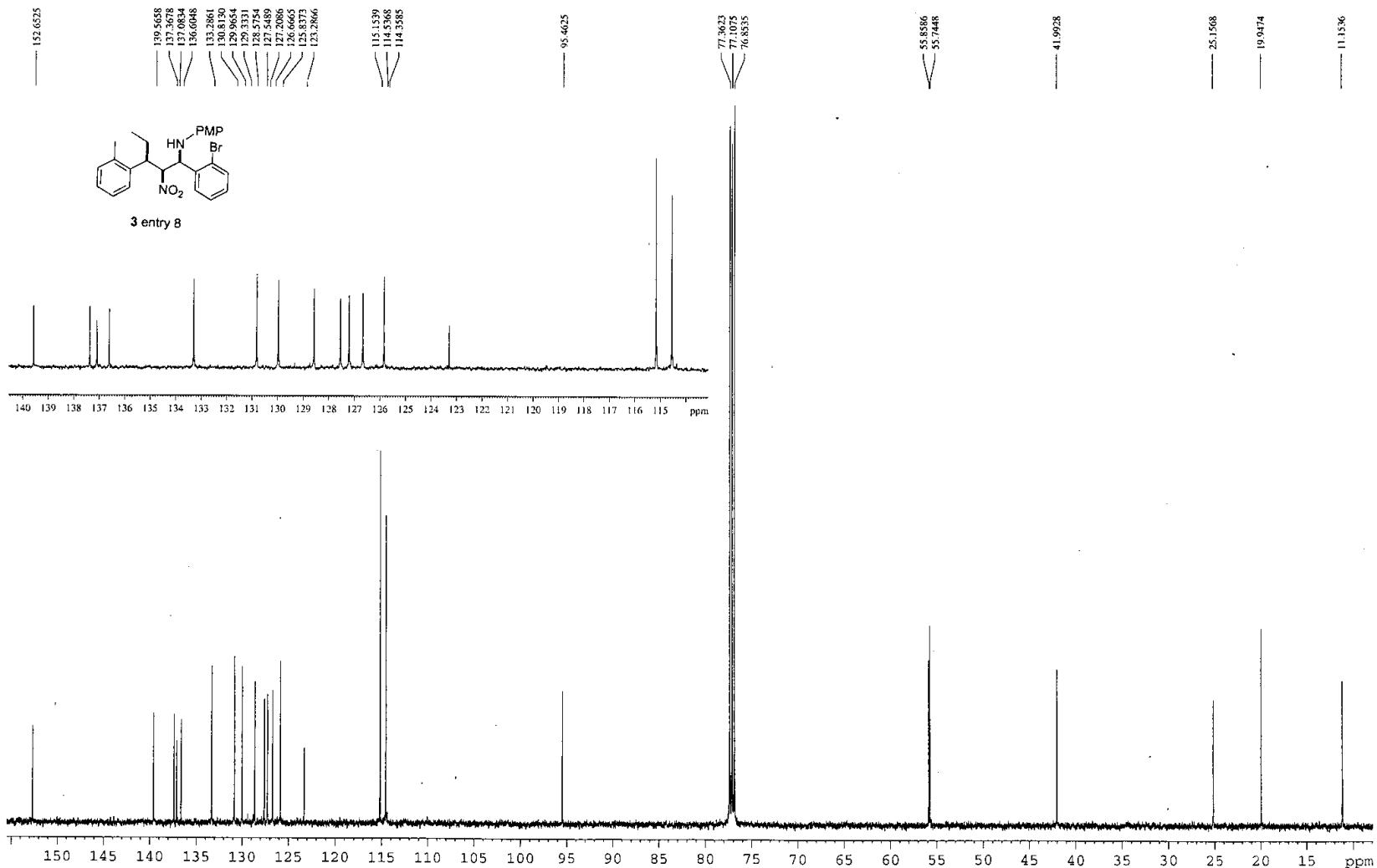


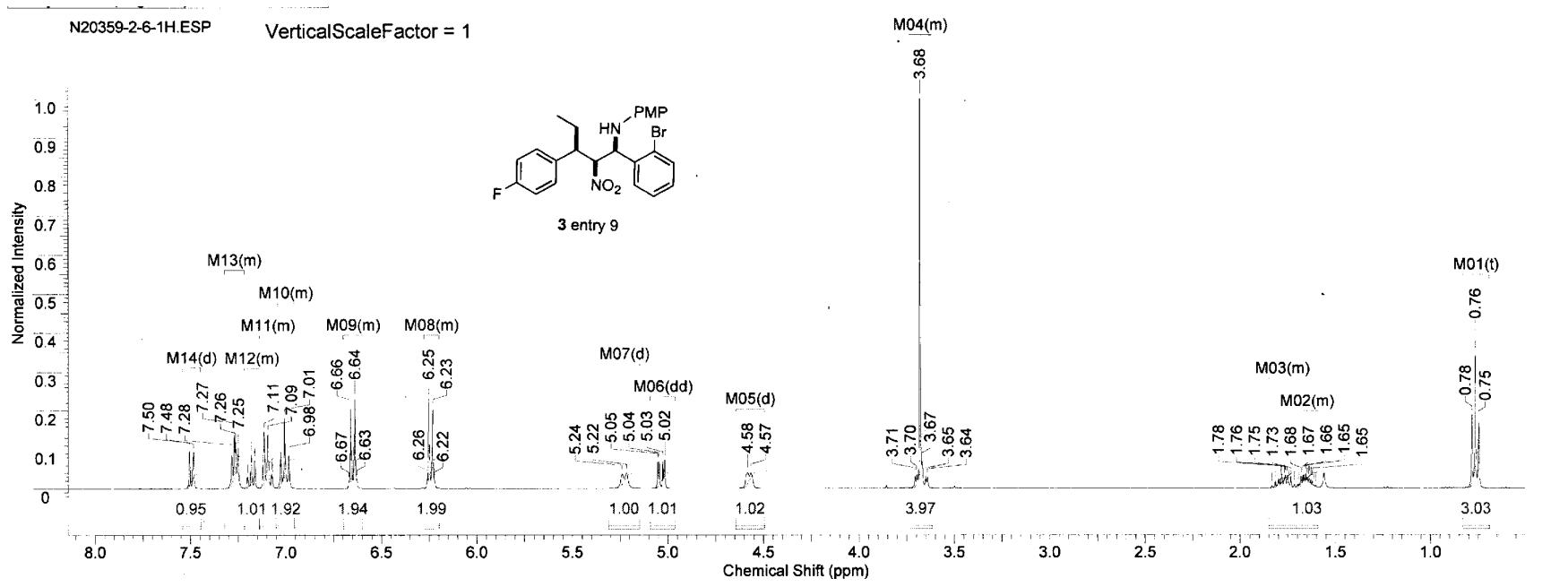


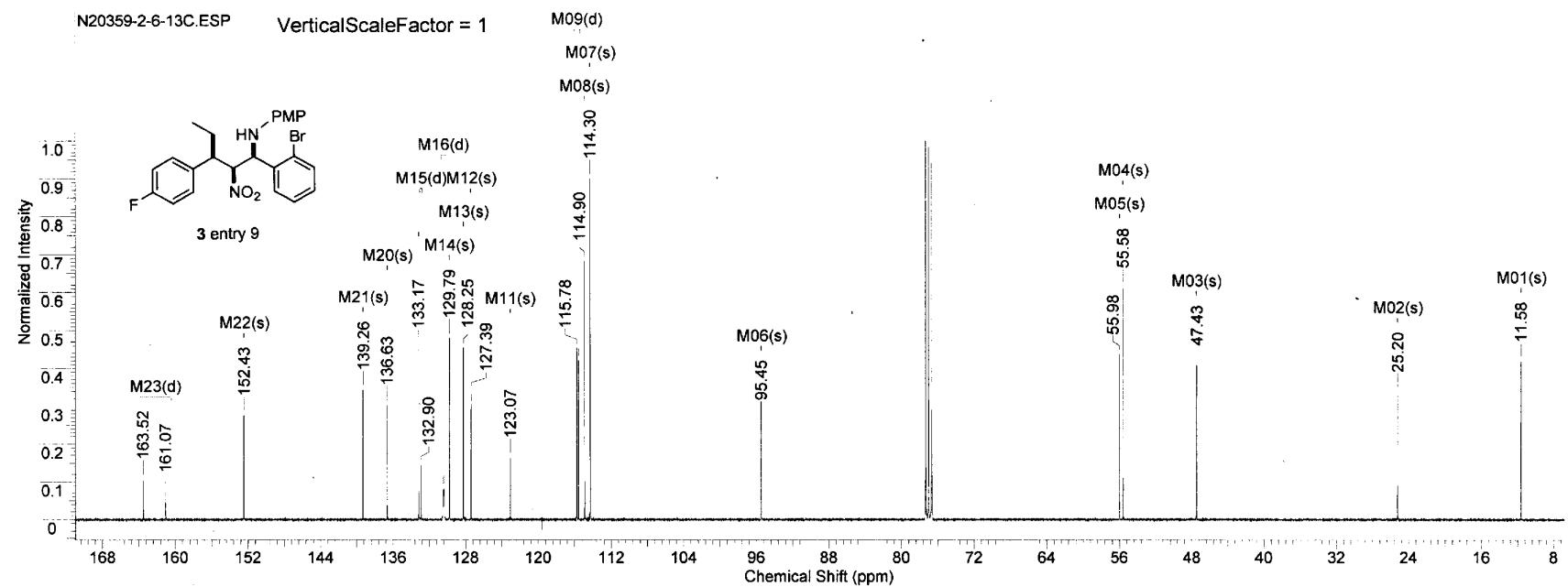




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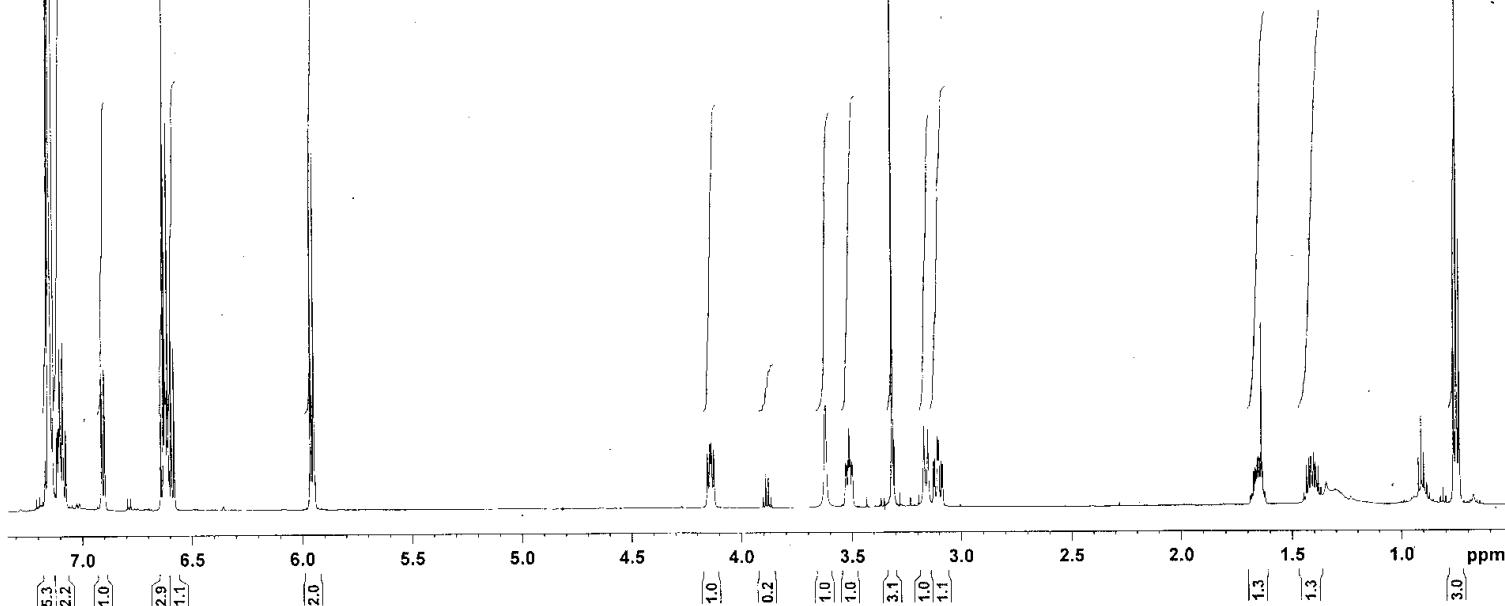
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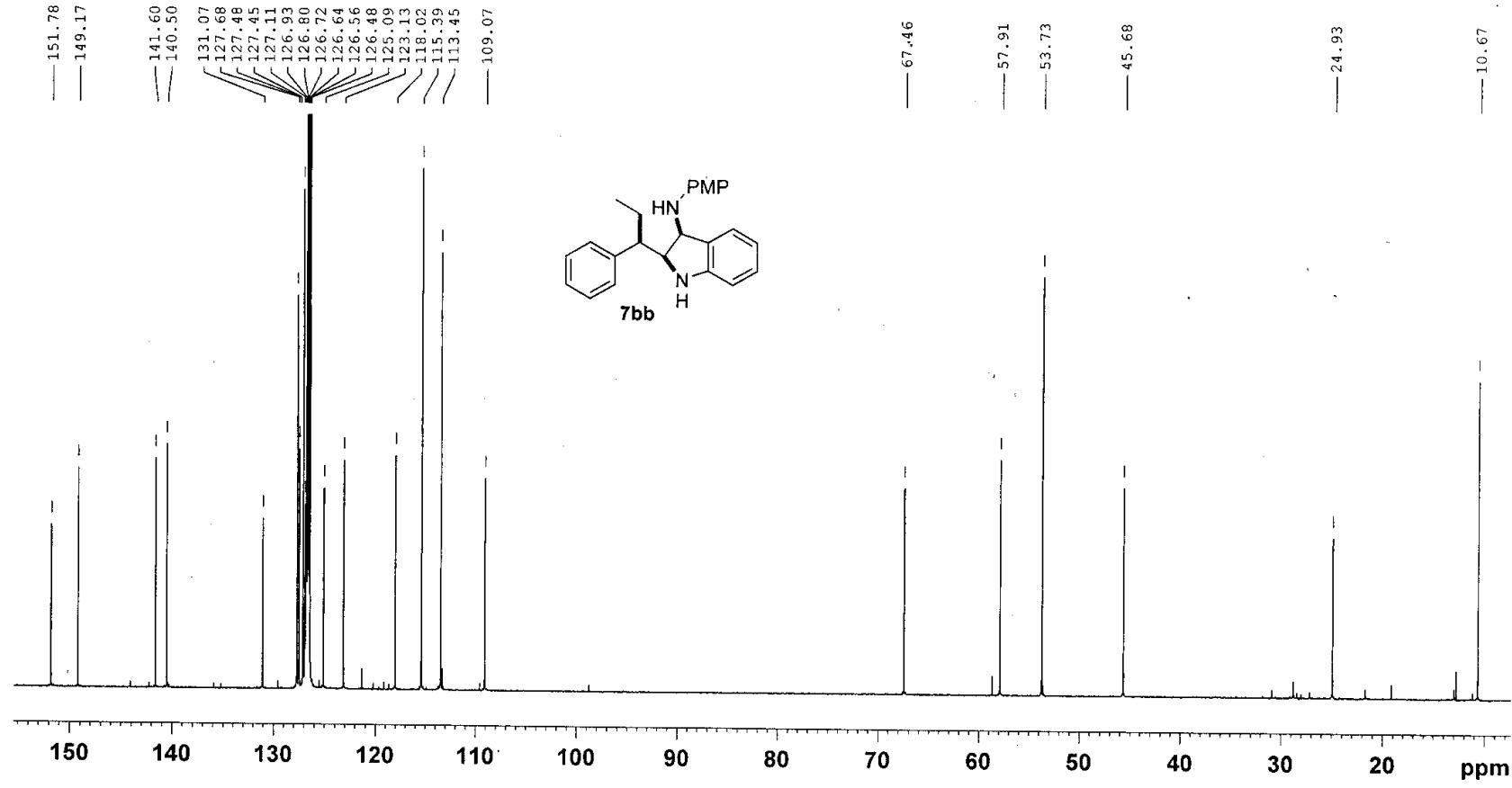


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DE 10.48 usec
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TD0 1

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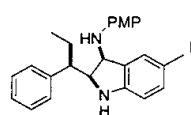


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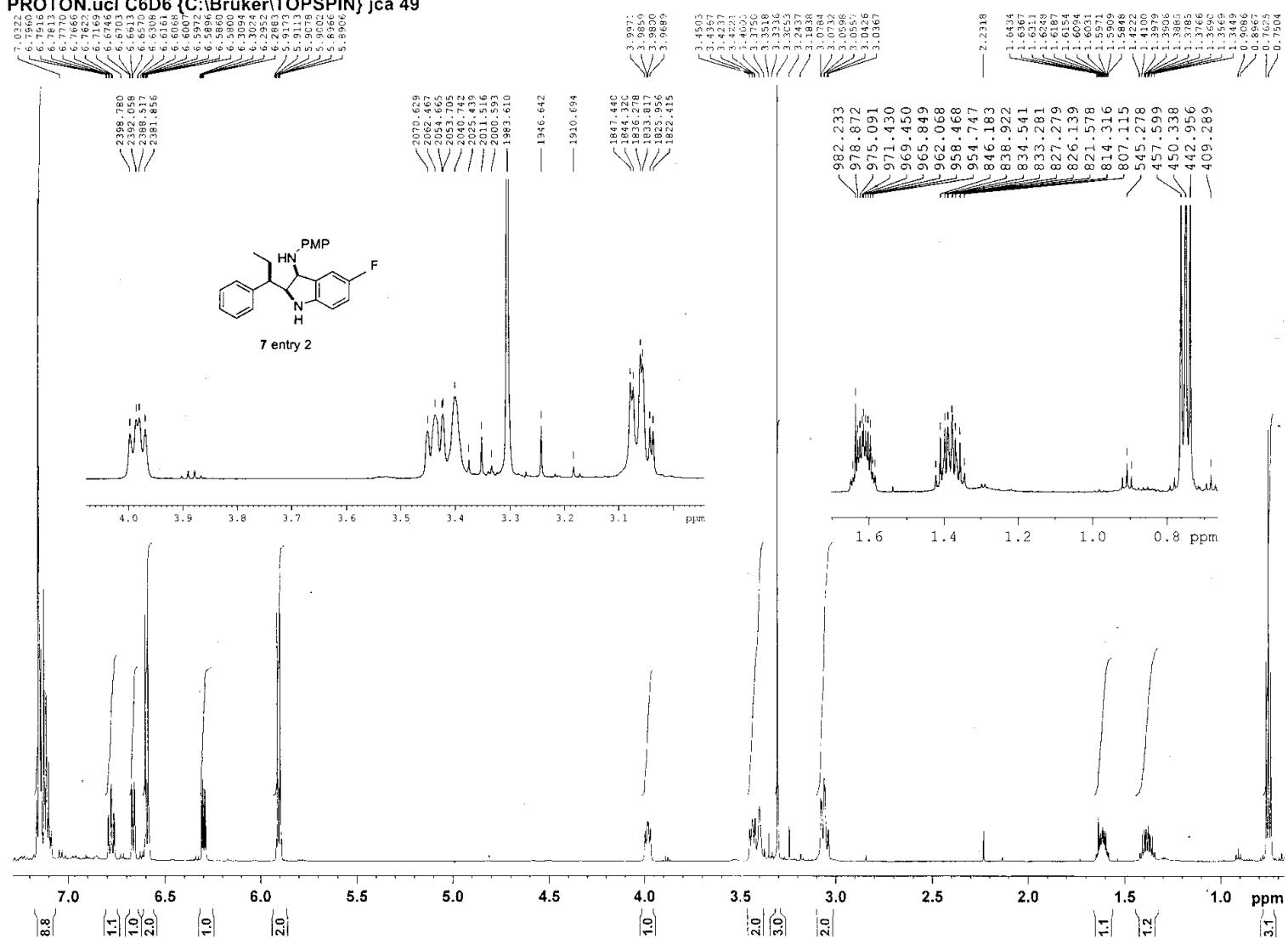


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7 entry 2



01. /504
UCL

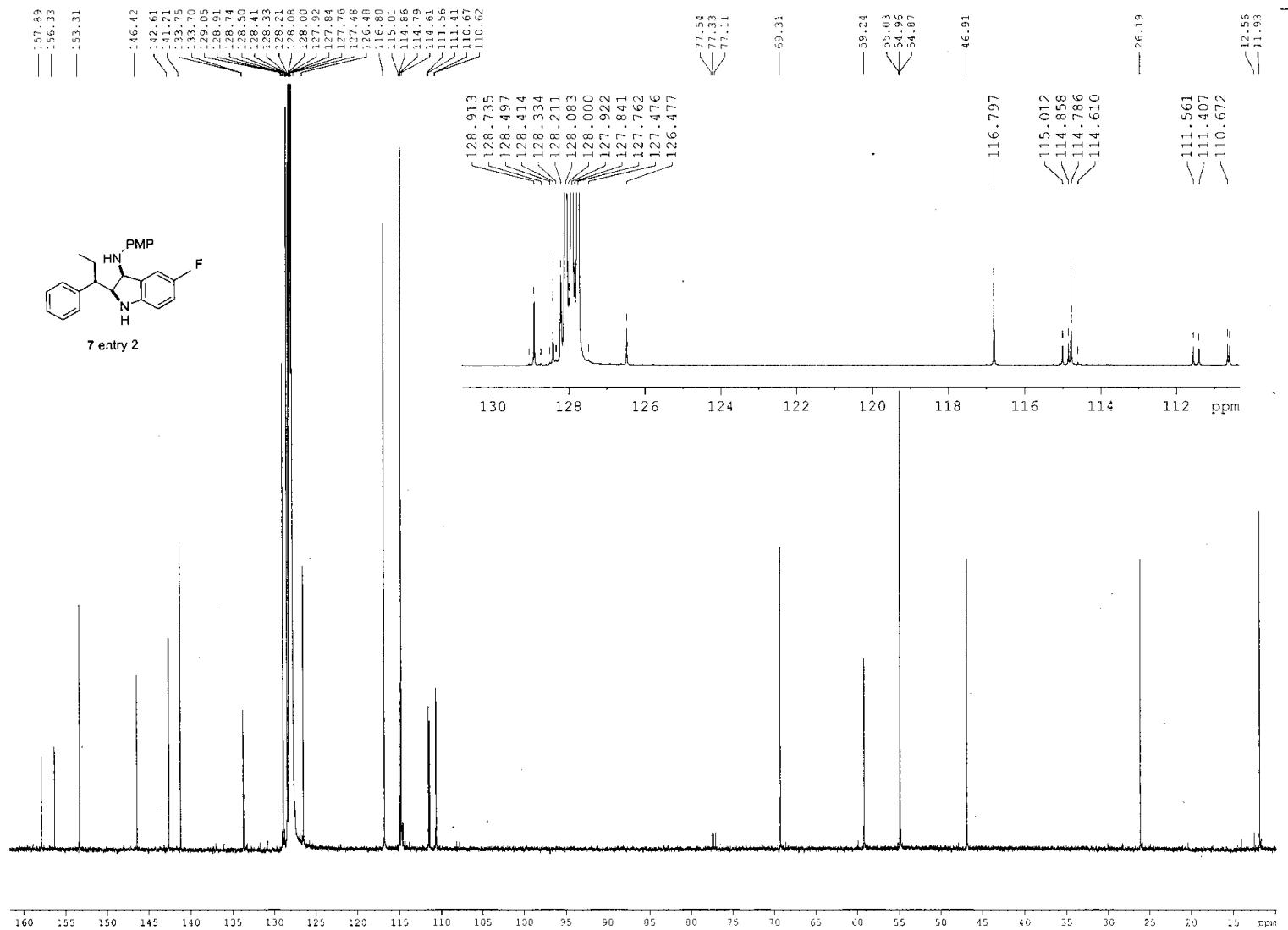
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TE         768.6 K
D1      1.0000000 sec
TDO           1

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jms 248 f1a
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TDC 1

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PL12W 0.36E4.776 W
PL13W 0.0E0.0000 W
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SI 1E-74.8
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