1,1-Alkenylboration of diarylphosphino enynes: convenient synthetic entry to vicinal P/B Lewis pairs at extended conjugated π -frameworks

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\$ X-Ray crystal structure analyses.

Supporting Information

General Procedure. All syntheses involving air- and moisture sensitive compounds were carried out using standard Schlenk-type glassware (or in a glove box) under an atmosphere of argon. Solvents were dried and stored under an argon atmosphere. NMR spectra were recorded on a Agilent DD2-500 MHz (¹H: 500 MHz, ¹³C: 126 MHz, ¹⁹F: 470 MHz, ¹¹B: 160 MHz, ³¹P: 202 MHz) and on a Agilent DD2- 600 MHz (¹H: 600 MHz, ¹³C: 151 MHz, ¹⁹F: 564 MHz, ¹¹B: 192 MHz, ³¹P: 243 MHz). ¹H NMR and ¹³C NMR: chemical shifts are given relative to TMS and referenced to the solvent signal. ¹⁹F NMR: chemical shifts are given relative to CFCl₃ ($\delta = 0$, external reference), ¹¹B NMR: chemical shifts are given relative to BF₃·Et₂O ($\delta = 0$, external reference), ³¹P NMR: chemical shifts are given relative to H₃PO₄ (85% in D₂O) ($\delta = 0$, external reference). NMR assignments were supported by additional 2D NMR experiments. Elemental analyses were performed on an Elementar Vario El III. IR spectra were recorded on a Varian 3100 FT-IR (Excalibur Series). Melting points and decomposition points were obtained with a DSC 2010 (TA Instruments). HRMS was recorded on GTC Waters Micromass (Manchester, UK). X-Ray crystal structure analyses. Data sets were collected with a Nonius KappaCCD diffractometer. Programs used: data collection, COLLECT (Nonius B.V., 1998); data reduction Denzo-SMN (Z. Otwinowski, W. Minor, Methods Enzymol. 1997, 276, 307-326); absorption correction, Denzo (Z. Otwinowski, D. Borek, W. Majewski, W. Minor, Acta Crystallogr. 2003, A59, 228-234); structure solution SHELXS-97 (G. M. Sheldrick, Acta Crystallogr. 1990, A46, 467-473); structure refinement SHELXL-97 (G. M. Sheldrick, Acta Crystallogr. 2008, A64, 112-122) and graphics, XP (BrukerAXS, 2000). Thermal ellipsoids are shown with 30% probability, *R*-values are given for observed reflections, and wR^2 values are given for all reflections. Exceptions and special features: For compound 5a one disordered over two positions t-Bu group was found in the asymmetrical unit. Several restraints (SADI, SIMU, ISOR and SAME) were used in order to improve refinement stability. A half disordered pentane molecule was found in the asymmetrical unit of compound 5c could not be satisfactorily refined. For the compound 5d one half dichloromethane and one half pentane molecules were found in the asymmetric unit. The program SQUEEZE (A. L. Spek J. Appl. Cryst., 2003, 36, 7-13) was therefore used to remove mathematically the effect of the solvent. The quoted formula and derived parameters are not included the squeezed solvent molecule. One CH₂-CH₂ unit disordered over two positions was found in the asymmetrical unit of 7b. Several restraints (SADI, SAME, ISOR and SIMU) were used in order to improve refinement stability. In addition a badly disordered dichloromethane molecule was found in the asymmetrical unit and could not be satisfactorily refined. The program SQUEEZE (A. L. Spek J. Appl. Cryst., 2003, 36, 7-13) was therefore used to remove mathematically the effect of the solvent. The quoted formula and derived parameters are not included the squeezed solvent molecule. For compounds 10a and 11 one disordered over two positions dichloromethane molecule was found in the asymmetrical unit. Several restraints (SADI, SIMU, ISOR and SAME) were used in order to improve refinement stability. CCDC deposition numbers are 1016759 to 1016766.

Materials. Vinylboranes 1a, 1b [O. Ekkert, O. Tuschewitzki, C. G. Daniliuc, G. Kehr and G. Erker, *Chem. Commun.*, 2013, 49, 6992-6994; D. J. Parks, W. E. Piers and G. P. A. Yap, *Organometallics*, 1998, 17, 5492-5503.] and diarylphosphino-enynes 4a [M. S. Chattha, *J. Chem. Eng. Data.*, 1978, 23, 95-96] and 6 [C. Charrier, W. Chodkiewicz and P. Cadiot, *Bull. Soc. Chim. Fr.*, 1966, 3, 1002-1011.] were prepared according to the literature.

Synthesis of compound 4b.

¹**H NMR** (500 MHz, 299 K, CD₂Cl₂): $\delta = 6.82$ (dm, ⁴*J*_{PH} = 3.3 Hz, 4H, *m*-Mes), 5.23, 5.22 (each m, each 1H, =CH₂), 2.37 (m, 12H, *o*-CH₃^{Mes}), 2.25 (s, 6H, *p*-CH₃^{Mes}), 1.87 (dd, *J* = 1.6 Hz, *J* = 1.1 Hz, 3H, CH₃).

¹³C{¹H} NMR (126 MHz, 299 K, CD₂Cl₂): $\delta = 142.3$ (d, ²*J*_{PC} = 15.7 Hz, *o*-Mes), 138.8 (*p*-Mes), 130.2 (d, ³*J*_{PC} = 3.7 Hz, *m*-Mes), 130.0 (d, ¹*J*_{PC} = 12.1 Hz, *i*-Mes), 127.7 (d, ³*J*_{PC} = 1.7 Hz, =C^{Me}), 121.6 (d, ⁴*J*_{PC} = 2.8 Hz, =CH₂), 108.4 (d, ²*J*_{PC} = 8.6 Hz, =C), 86.9 (d, ¹*J*_{PC} = 6.2 Hz, =CP), 23.0 (d, ³*J*_{PC} = 14.4 Hz, *o*-CH₃^{Mes}), 22.8 (d, ⁴*J*_{PC} = 1.4 Hz, CH₃), 21.0 (*p*-CH₃^{Mes}).

³¹P{¹H} NMR (202 MHz, 299 K, CD₂Cl₂): $\delta = -56.7 (v_{1/2} \sim 1 \text{ Hz}).$

¹**H**, ¹**H**-**GCOSY** (500 MHz / 500 MHz, 299 K, CD_2Cl_2): δ ¹H / δ ¹H = 6.82 / 2.37,

2.25 (*m*-Mes / *o*-CH₃^{Mes}, *p*-CH₃^{Mes}), 5.23, 5.22 / 1.87 (=CH₂ / CH₃).

¹**H**, ¹³**C GHSQC** (500 MHz / 126 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹³C = 6.82 / 130.2 (*m*-Mes), 5.23 / 121.6 (=CH₂), 5.22 / 121.6 (=CH₂), 2.37 / 23.0 (*o*-CH₃^{Mes}), 2.25 / 21.0 (*p*-CH₃^{Mes}), 1.87 / 22.8 (CH₃).

¹H, ¹³C GHMBC (500 MHz / 126 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹³C = 6.82 / 130.2, 130.0, 23.0, 21.0 (*m*-Mes / *m*-Mes, *i*-Mes, *o*-CH₃^{Mes}, *p*-CH₃^{Mes}), 5.23, 5.22 / 127.7, 108.4, 22.8 (=CH₂, =CH₂ / =C^{Me}, ≡C, CH₃), 2.37 / 142.3, 130.2, 130.0 (*o*-CH₃^{Mes} / *o*-Mes, *m*-Mes, *i*-Mes), 2.25 / 138.8, 130.2 (*p*-CH₃^{Mes} / *p*-Mes, *m*-Mes), 1.87 / 127.7, 121.6, 108.4 (CH₃ / =C^{Me}, =CH₂, ≡C).







Synthesis of compound 5a.



The reaction mixture of the borane **1a** (0.214 g, 0.5 mmol, 1 eq) and the phosphane **4a** (0.125 g, 0.5 mmol, 1 eq) in toluene (5 mL) was heated at 60 °C for 3 d. Then all volatiles were removed in vacuo and the residue was washed with pentane (1 × 3 mL). After drying of the solid in vacuo compound **5a** (0.272 g, 0.40 mmol, 80 %) was obtained as a light yellow solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a dichloromethane solution of compound **5a** at -35 °C. **IR** (KBr): $\tilde{\nu}$ / cm⁻¹ = 3055, 2963, 1642, 1622, 1514, 1464, 1378, 1285, 1097, 971, 745, 693. **M.p.**: 152 °C. **Anal. Calc.** for C₃₅H₂₆BF₁₀P: C: 61.97; H: 3.86. Found: C: 61.70; H: 3.51.

¹**H NMR** (600 MHz, 299 K, C₆D₆): δ = 7.31 (m, 4H, *o*-Ph), 6.96 (dd, ³*J*_{HH} = 15.8 Hz, ⁴*J*_{PH} = 2.0 Hz, 1H, =CH), 6.90 (m, 2H, *p*-Ph), 6.82 (m, 4H, *m*-Ph), 6.18 (d, ³*J*_{HH} = 15.8 Hz, 1H, =CH^{*t*Bu}), 5.05, 4.91 (each m, each 1H, =CH₂), 1.87 (m, 3H, CH₃), 0.89 (s, 9H, ^{*t*}Bu).

¹³C{¹H} NMR (151 MHz, 299 K, C₆D₆): $\delta = 177.5$ (br, =CB), 155.2 (=CH^{*t*Bu}), 148.8 (dm, ¹*J*_{FC} ~ 240 Hz, C₆F₅), 140.2 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 140.1 (d, ²*J*_{PC} = 2.5 Hz, =C^{Me}), 137.5 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 132.2 (d, ²*J*_{PC} = 9.2 Hz, *o*-Ph), 131.7 (d, ⁴*J*_{PC} = 3.0 Hz, *p*-Ph), 129.6 (d, ¹*J*_{PC} = 56.2 Hz, =CP), 128.9 (d, ³*J*_{PC} = 10.5 Hz, *m*-Ph), 126.2 (d, ¹*J*_{PC} = 40.4 Hz, *i*-Ph), 123.2 (d, ³*J*_{PC} = 47.2 Hz, =CH), 118.2 (d, ³*J*_{PC} = 6.6 Hz, =CH₂), 117.2 (br, *i*-C₆F₅), 33.8 (d, ⁵*J*_{PC} = 0.8 Hz, ^{*t*}Bu), 28.9 (^{*t*}Bu), 23.3 (d, ³*J*_{PC} = 5.4 Hz, CH₃).

¹¹B{¹H} NMR (192 MHz, 299 K, C₆D₆): δ = -8.3 (v_{1/2} ~ 250 Hz).

³¹**P**{¹**H**} **NMR** (243 MHz, 299 K, C₆D₆): $\delta = 10.2 (v_{1/2} \sim 75 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 299 K, C₆D₆): δ = -129.2 (m, 2F, *o*-C₆F₅), -157.8 (tm, ³J_{FF} = 20.7 Hz, 1F, *p*-C₆F₅), -164.4 (m, 2F, *m*-C₆F₅) [Δδ¹⁹F_{m,p} = 6.6].

¹**H**, ¹**H GCOSY** (600 MHz / 600 MHz, 299 K, C₆D₆)[selective traces]: δ ¹H / δ ¹H = 7.31 / 6.82 (*o*-Ph / *m*-Ph), 6.96 / 6.18 (=CH / =CH^{*t*Bu}), 6.90 / 6.82 (*p*-Ph / *m*-Ph), 5.05 / 4.91, 1.87 (=CH₂ / =CH₂, CH₃).

¹**H**, ¹³**C-GHSQC** (600 MHz / 151 MHz, 299 K, C_6D_6): δ ¹H / δ ¹³C = 7.31 / 132.2 (*o*-Ph), 6.96 / 123.2 (=CH), 6.90 / 131.7 (*p*-Ph), 6.82 / 128.9 (*m*-Ph), 6.18 / 155.2 (=CH^{*t*Bu}), 5.05 / 118.2 (=CH₂), 4.91 / 118.2 (=CH₂), 1.87 / 23.3 (CH₃), 0.89 / 28.9 (^{*t*}Bu).

¹**H**, ¹³**C GHMBC** (600 MHz / 151 MHz, 299 K, C₆D₆)[selective traces]: δ ¹H / δ ¹³C = 7.31 / 132.2, 131.7 (*o*-Ph / *o*-Ph, *p*-Ph), 6.96 / 129.7, 33.9 (=CH / =CP, ^{*t*}Bu), 6.82 / 128.9, 126.2 (*m*-Ph / *m*-Ph, *i*-Ph), 6.18 / 177.5, 33.8, 28.9 (=CH^{*t*Bu} / =CB, ^{*t*}Bu, ^{*t*}Bu), 5.05, 4.91 / 129.6, 23.3 (=CH₂ / =CP, CH₃), 1.87 / 140.2, 129.6, 118.2 (CH₃ / =C^{Me}, =CP, =CH₂), 0.89 / 155.2, 33.8, 28.9 (^{*t*}Bu / =CH^{*t*Bu}, ^{*t*}Bu).

¹⁹**F**, ¹⁹**F GCOSY** (564 MHz / 564 MHz, 299 K, C_6D_6): $\delta^{19}F / \delta^{19}F = -129.2 / -164.4$ (*o*- C_6F_5 / m - C_6F_5), -157.8 / -164.4 (*p*- C_6F_5 / m - C_6F_5).





of compound 5a

X-ray crystal structure analysis of compound 5a: formula $C_{35}H_{26}BF_{10}P$, M = 678.34, colourless crystal, 0.24 x 0.19 x 0.14 mm, a = 11.5460(2), b = 12.3600(1), c = 12.7650(2) Å, $\alpha = 101.747(1)$, $\beta = 113.392(2)$, $\gamma = 95.129(1)^{\circ}$, V = 1607.29(4) Å³, $\rho_{calc} = 1.402$ gcm⁻³, $\mu = 1.501$ mm⁻¹, empirical absorption correction (0.714 $\leq T \leq 0.817$), Z = 2, triclinic, space group $P\overline{1}$ (No. 2), $\lambda = 1.54178$ Å, T = 223(2) K, ω and ϕ scans, 20182 reflections collected ($\pm h$, $\pm k$, $\pm l$), [(sin θ)/ λ] = 0.60 Å⁻¹, 5561 independent ($R_{int} = 0.045$) and 4965 observed reflections [$I > 2 \sigma$ (I)], 459 refined parameters, R = 0.040, $wR^2 = 0.106$, max. (min.) residual electron density 0.21 (-0.25) e.Å⁻³, hydrogen atoms calculated and refined as riding atoms.



Synthesis of compound 5b.



The reaction mixture of the borane **1b** (0.224 g, 0.5 mmol, 1 eq) and the phosphane **4a** (0.125 g, 0.5 mmol, 1 eq) in toluene (5 mL) was heated at 60 °C for 3 d. Then all volatiles were removed in vacuo and the residue was washed with pentane (1 × 3 mL). After drying in vacuo compound **5b** (0.298 g, 0.43 mmol, 85 %) was obtained as a light yellow solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a dichloromethane solution of compound **5b** at -35 °C. **IR** (KBr): $\tilde{\nu}$ / cm⁻¹ = 3053, 2926, 1644, 1598, 1516, 1457, 1382, 1286, 1100, 977, 750, 692. **Decomp.**: 198 °C. **Anal. Calc.** for C₃₇H₂₂BF₁₀P: C: 63.64; H: 3.18. Found: C: 63.19; H: 2.89.

¹**H NMR** (600 MHz, 299 K, C₆D₆): δ = 7.75 (dd, ³*J*_{HH} = 15.8 Hz, ⁴*J*_{PH} = 2.1 Hz, 1H, =CH), 7.31 (m, 4H, *o*-Ph^P), 7.28 (m, 2H, *o*-Ph), 7.08 (d, ³*J*_{HH} = 15.8 Hz, 1H, =CH^{Ph}), 6.98 (m, 2H, *m*-Ph), 6.97 (m, 1H, *p*-Ph), 6.90 (m, 2H, *p*-Ph^P), 6.82 (m, 4H, *m*-Ph^P), 5.10, 4.96 (each m, each 1H, =CH₂), 1.86 (m, 3H, CH₃).

¹³C{¹H} NMR (151 MHz, 299 K, C₆D₆): $\delta = 175.8$ (br, =CB), 148.7 (dm, ¹*J*_{FC} ~ 240 Hz, C₆F₅), 140.8 (=CH^{Ph}), 140.3 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 140.1 (d, ²*J*_{PC} = 2.3 Hz, =C^{Me}), 137.5 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 136.9 (d, ⁵*J*_{PC} = 1.2 Hz, *i*-Ph), 132.1 (d, ²*J*_{PC} = 9.2 Hz, *o*-Ph^P), 131.8 (d, ¹*J*_{PC} = 55.7 Hz, =CP), 131.8 (d, ⁴*J*_{PC} = 3.0 Hz, *p*-Ph^P), 129.1 (*p*-Ph), 129.0 (*m*-Ph), 128.9 (d, ³*J*_{PC} = 10.4 Hz, *m*-Ph^P), 127.7 (*o*-Ph), 126.2 (d, ³*J*_{PC} = 48.1 Hz, =CH), 126.0 (d, ¹*J*_{PC} = 40.7 Hz, *i*-Ph^P), 118.8 (d, ³*J*_{PC} = 6.8 Hz, =CH₂), 116.8 (br, *i*-C₆F₅), 23.3 (d, ³*J*_{PC} = 5.1 Hz, CH₃).

¹¹**B**{¹**H**} **NMR** (192 MHz, 299 K, C₆D₆): $\delta = -8.1 (v_{1/2} \sim 350 \text{ Hz}).$

³¹**P**{¹**H**} **NMR** (243 MHz, 299 K, C_6D_6): $\delta = 10.6 (v_{1/2} \sim 60 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 299 K, C₆D₆): δ = -129.2 (m, 2F, *o*-C₆F₅), -157.3 (tm, ³J_{FF} = 20.8 Hz, 1F, *p*-C₆F₅), -164.0 (m, 2F, *m*-C₆F₅) [Δδ¹⁹F_{m,p} = 6.7].

¹**H**, ¹**H GCOSY** (600 MHz / 600 MHz, 299 K, C₆D₆)[selective traces]: δ ¹H / δ ¹H = 7.75 / 7.08 (=CH / =CH^{Ph}), 7.31 / 6.90, 6.82 (*o*-Ph^P / *p*-Ph^P, *m*-Ph^P), 7.28 / 6.97 (*o*-Ph / *m*-Ph, *p*-Ph), 5.10 / 4.96, 1.86 (=CH₂ / =CH₂, CH₃).

¹**H**, ¹³**C GHSQC** (600 MHz / 151 MHz, 299 K, C₆D₆): δ ¹H / δ ¹³C = 7.75 / 126.2 (=CH), 7.31 / 132.1 (*o*-Ph^P), 7.28 / 127.7 (*o*-Ph), 7.08 / 140.8 (=CH^{Ph}), 6.97 / 129.1 (*p*-Ph), 6.97 / 129.0 (*m*-Ph), 6.90 / 131.8 (*p*-Ph^P), 6.82 / 128.9 (*m*-Ph^P), 5.10, 4.96 / 118.8 (=CH₂), 1.86 / 23.3 (CH₃).

¹H, ¹³C GHMBC (600 MHz / 151 MHz, 299 K, C₆D₆)[selective traces]: δ ¹H / δ ¹³C = 7.75 / 136.9, 131.8 (=CH / *i*-Ph, =CP), 7.31 / 132.1, 131.8 (*o*-Ph^P / *o*-Ph^P, *p*-Ph^P), 7.28 / 140.8, 129.1, 127.7 (*o*-Ph / =CH^{Ph}, *p*-Ph, *o*-Ph), 7.08 / 175.8, 136.9, 127.7 (=CH^{Ph} / =CB, *i*-Ph, *o*-Ph), 6.97 / 136.9, 129.0 (*m*-Ph / *i*-Ph, *m*-Ph), 6.82 / 128.9, 126.0 (*m*-Ph^P / *m*-Ph^P, *i*-Ph^P), 5.10, 4.96 / 131.8, 23.3 (=CH₂ / =CP, CH₃), 1.86 / 140.1, 131.8, 118.8 (CH₃ / =C^{Me}, =CP, =CH₂).

¹⁹**F**, ¹⁹**F GCOSY** (564 MHz / 564 MHz, 299 K, C_6D_6): $\delta^{19}F / \delta^{19}F = -129.2 / -164.0$ (*o*- C_6F_5 / m - C_6F_5), -157.3 / -164.0 (*p*- C_6F_5 / m - C_6F_5).



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290 270 250 230 210 190 170 150 130 110 10 -10 -30 90 70 50 30 -50 -70 -90 ¹¹B{¹H} NMR (192 MHz, 299 K, C_6D_6) and ³¹P{¹H} NMR (243 MHz, 299 K, C_6D_6) of compound 5b

X-ray crystal structure analysis of compound 5b: formula $C_{37}H_{22}BF_{10}P \cdot 0.5 \times CH_2Cl_2$, M = 740.79, colourless crystal, 0.24 x 0.16 x 0.08 mm, a = 23.2576(5), b = 18.1844(5), c = 16.9234(5) Å, $\beta = 111.378(2)^{\circ}$, V = 6664.9(3) Å³, $\rho_{calc} = 1.477$ gcm⁻³, $\mu = 2.224$ mm⁻¹, empirical absorption correction (0.617 $\leq T \leq 0.842$), Z = 8, monoclinic, space group C2/c (No. 15), $\lambda = 1.54178$ Å, T = 223(2) K, ω and ϕ scans, 24503 reflections collected ($\pm h$, $\pm k$, $\pm l$), [($\sin\theta$)/ λ] = 0.60 Å⁻¹, 5865 independent ($R_{int} = 0.046$) and 5223 observed reflections [$I > 2 \sigma(I)$], 465 refined parameters, R = 0.051, $wR^2 = 0.140$, max. (min.) residual electron density 1.17 (-0.75) e.Å⁻³, the hydrogen atom at C6 was refined freely; others were calculated and refined as riding atoms.



Synthesis of compound 5c.



The reaction mixture of the borane **1a** (0.214 g, 0.5 mmol, 1 eq) and the phosphane **4b** (0.167 g, 0.5 mmol, 1 eq) in toluene (5 mL) was heated at 40 °C for 3 d. Then all volatiles were removed in vacuo and the residue was washed with pentane (1 × 3 mL). After drying in vacuo compound **5c** (0.270 g, 0.35 mmol, 71 %) was obtained as a light yellow solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a dichloromethane solution of compound **5c** at -35 °C. **IR** (KBr): $\tilde{\nu}$ / cm⁻¹ = 3023, 2962, 2866, 1642, 1605, 1516, 1469, 1380, 1287, 1107, 975. **M.p.**: 173 °C. **Anal. Calc.** for C₄₁H₃₈BF₁₀P: C: 64.58; H: 5.02. Found: C: 64.35; H: 4.58.

¹**H NMR** (600 MHz, 213 K, CD₂Cl₂): $\delta = 6.96$ (m, 1H, *m*-Mes^A), 6.80 (m, 1H, *m*'-Mes^A), 6.77 (m, 1H, *m*-Mes^B), 6.36 (m, 1H, *m*'-Mes^B), 6.19 (d, ³*J*_{HH} = 16.0 Hz, 1H, =CH), 5.65 (d, ³*J*_{HH} = 16.0 Hz, 1H, =CH^{*t*Bu}), 5.27, 4.99 (each br m, each 1H, =CH₂), 2.54 (s, 3H, *o*-CH₃^{MesB}), 2.45 (br s, 3H, *o*-CH₃^{MesA}), 2.22 (s, 3H, *p*-CH₃^{MesA}), 2.16 (s, 3H, *o*'-CH₃^{MesA}), 2.10 (s, 3H, *p*-CH₃^{MesB}), 1.58 (s, 3H, *o*'-CH₃^{MesB}), 1.53 (s, 3H, CH₃), 0.75 (s, 9H, ^{*t*}Bu).

¹³C{¹H} **NMR** (151 MHz, 213 K, CD₂Cl₂): $\delta = 168.6$ (d, ²*J*_{PC} = 28.7 Hz, =CB), 152.6 (=CH^{*t*Bu}), 143.3 (d, ²*J*_{PC} = 17.7 Hz, *o*-Mes^A), 142.9 (*o*²-Mes^A), 142.5 (d, ²*J*_{PC} = 3.3 Hz, *o*-Mes^B), 141.2 (d, ⁴*J*_{PC} = 2.3 Hz, *p*-Mes^A), 139.8 (d, ⁴*J*_{PC} = 2.5 Hz, *p*-Mes^B), 139.72 (d, ²*J*_{PC} = 13.8 Hz, *o*²-Mes^B), 139.67 (d, ²*J*_{PC} = 2.9 Hz, =C^{Me}), 134.7 (d, ¹*J*_{PC} = 52.4 Hz, =CP), 130.2 (d, ³*J*_{PC} = 8.2 Hz, *m*-Mes^B), 129.6 (d, ³*J*_{PC} = 5.8 Hz, *m*²-Mes^A), 129.4 (d, ³*J*_{PC} = 9.1 Hz, *m*-Mes^A), 129.2 (m, *m*²-Mes^B), 125.1 (d, ¹*J*_{PC} = 25.8 Hz, *i*-Mes^A), 124.4 (d, ¹*J*_{PC} = 40.2 Hz, *i*-Mes^B), 120.1 (dm, ³*J*_{PC} = 45.3 Hz, =CH), 118.7 (*i*-C₆F₅), 118.0 (m, =CH₂), 116.4 (*i*-C₆F₅), 33.4 (^{*i*}Bu), 27.9 (^{*i*}Bu), 24.8 (*o*-CH₃^{MesB}), 23.6 (*o*-CH₃^{MesA}), 20.1 (*p*-CH₃^{MesA}), [C₆F₅ not listed]

¹¹B{¹H} NMR (192 MHz, 213 K, CD₂Cl₂): $\delta = -2.0 (v_{1/2} \sim 1800 \text{ Hz}).$

¹¹B{¹H} NMR (192 MHz, 213 K, CD₂Cl₂): $\delta = -0.5 (v_{1/2} \sim 380 \text{ Hz}).$

³¹**P**{¹**H**} **NMR** (243 MHz, 213 K, CD₂Cl₂): $\delta = 11.3$ (partial relaxed 1:1:1:1 q, $J_{PB} \sim 25$ Hz).

³¹P{¹H} NMR (243 MHz, 299 K, CD₂Cl₂): $\delta = 12.8 (v_{1/2} \sim 40 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 213 K, CD₂Cl₂): δ = -125.9 (m, *o*), -131.3 (m, *o*'), -158.5 (t, ${}^{3}J_{FF}$ = 21.4 Hz, *p*), -164.7 (m, *m*), -164.8 (m, *m*')(each 1F, C₆F₅^A) [Δδ¹⁹F_{m,p} = 6.2, 6.3], -128.7 (m, *o*), -130.6 (m, *o*'), -159.7 (br t, ${}^{3}J_{FF}$ = 19.8 Hz, *p*), -165.8 (m, *m*'), -166.0 (m, *m*)(each 1F, C₆F₅^B) [Δδ¹⁹F_{m,p} = 6.1, 6.3].

¹**H**, ¹**H GCOSY** (600 MHz / 600 MHz, 213 K, CD₂Cl₂)[selected traces]: δ ¹H / δ ¹H = 6.96 / 6.80, 2.45, 2.22, 2.16 (*m*-Mes^A / *m*²-Mes^A, *o*-CH₃^{MesA}, *p*-CH₃^{MesA}, *o*²-CH₃^{MesA}), 6.77 / 6.36, 2.54, 2.10, 1.58 (*m*-Mes^B / *m*²-Mes^B, *o*-CH₃^{MesB}, *p*-CH₃^{MesB}, *o*²-CH₃^{MesB}), 6.19 / 5.65 (=CH / =CH^{*t*Bu}), 5.27 / 4.99, 1.53 (=CH₂ / =CH₂, CH₃).

¹H, ¹³C GHSQC (600 MHz / 151 MHz, 213 K, CD₂Cl₂): δ ¹H / δ ¹³C = 6.96 / 129.4 (*m*-Mes^A), 6.80 / 129.6 (*m*'-Mes^A), 6.77 / 130.2 (*m*-Mes^B), 6.36 / 129.2 (*m*'-Mes^B), 6.19 / 120.1 (=CH), 5.65 / 152.6 (=CH^{*t*Bu}), 5.27, 4.99 / 118.0 (=CH₂), 2.54 / 24.8 (*o*-CH₃^{MesB}), 2.45 / 23.6 (*o*-CH₃^{MesA}), 2.22 / 20.7 (*p*-CH₃^{MesA}), 2.16 / 22.6 (*o*'-CH₃^{MesA}), 2.10 / 20.1 (*p*-CH₃^{MesB}), 1.58 / 22.0 (*o*'-CH₃^{MesB}), 1.53 / 21.9 (CH₃), 0.75 / 27.9 (^{*t*}Bu).

¹H, ¹³C GHMBC (600 MHz / 151 MHz, 213 K, CD₂Cl₂): δ ¹H / δ ¹³C = 6.96 / 129.6, 125.1, 23.6, 20.7 (*m*-Mes^A / *m*'-Mes^A, *i*-Mes^A, *o*-CH₃^{MesA}, *p*-CH₃^{MesA}), 6.80 / 129.4, 125.1, 22.6, 20.7 (*m*'-Mes^A / *m*-Mes^A, *i*-Mes^A, *o*'-CH₃^{MesA}, *p*-CH₃^{MesA}), 6.77 / 129.2, 124.4, 24.8, 20.0 (*m*-Mes^B / *m*'-Mes^B, *i*-Mes^B, *o*-CH₃^{MesB}, *p*-CH₃^{MesB}), 6.36 / 130.2, 124.4, 22.0, 20.0 (*m*'-Mes^B / *m*-Mes^B, *i*-Mes^B, *o*'-CH₃^{MesB}, *p*-CH₃^{MesB}), 6.19 / 134.7, 33.5 (=CH / =CP, 'Bu), 5.65 / 168.6, 33.5, 27.9 (=CH'^{Bu} / =CB, 'Bu, 'Bu), 5.27, 4.99 / 134.7, 21.9 (=CH₂ / =CP, CH₃), 2.54 / 142.5, 130.2, 124.4 (*o*-CH₃^{MesB} / *o*-Mes^B, *m*-Mes^B, *i*-Mes^B), 2.45 / 143.3, 129.4, 125.1 (*o*-CH₃^{MesA} / *o*-Mes^A, *m*-Mes^A, *i*-Mes^A), 2.16 / 142.9, 129.6, 125.1 (*o*'-CH₃^{MesA} / *o*'-Mes^A, *m*'-Mes^A, *i*-Mes^A), 2.16 / 142.9, 129.6, 125.1 (*o*'-CH₃^{MesA} / *o*'-Mes^A, *i*-Mes^A), 2.10 / 139.8, 130.2, 129.2 (*p*-CH₃^{MesB} / *p*-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*'-Mes^A, *m*'-Mes^A, *m*-Mes^A), 2.16 / 142.9, 129.6, 125.1 (*o*'-CH₃^{MesA} / *o*'-Mes^A, *m*'-Mes^A), 2.10 / 139.8, 130.2, 129.2 (*p*-CH₃^{MesB} / *p*-Mes^B, *m*'-Mes^B), 1.58 / 139.72, 129.2, 124.5 (*o*'-CH₃^{MesB} / *o*'-Mes^B / *o*'-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*'-Mes^B, *m*'-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*-Mes^B, *m*'-Mes^B, *m*-Mes^A, *m*-Mes^A,

0.75 / 152.6, 33.5, 27.9 (^tBu / =CH^{tBu}, ^tBu, ^tBu).

¹⁹**F**, ¹⁹**F GCOSY** (564 MHz / 564 MHz, 213 K, CD₂Cl₂): δ^{19} **F** / δ^{19} **F** = -125.9 / 164.7 (*o*-C₆F₅^A / *m*-C₆F₅^A), -128.7 / -166.0 (*o*-C₆F₅^B / *m*-C₆F₅^B), -130.6 / -165.8 (*o*'-C₆F₅^B / *m*'-C₆F₅^B), -131.3 / -164.8 (*o*'-C₆F₅^A / *m*'-C₆F₅^A), -158.5 / -164.7, -164.8 (*p*-C₆F₅^A / *m*-C₆F₅^A), -158.5 / -164.7, -164.8 (*p*-C₆F₅^A / *m*-C₆F₅^A), -159.7 / -165.8, -166.0 (*p*-C₆F₅^B / *m*'-C₆F₅^B, *m*-C₆F₅^B).







290 270 250 230 210 190 170 150 130 110 90 70 50 30 10 -10 -30 -50 -70 -90

¹¹B{¹H} NMR (192 MHz, 213 K, CD₂Cl₂) and ³¹P{¹H} NMR (243 MHz, 213 K, CD₂Cl₂) of compound 5c



¹¹B{¹H} NMR (192 MHz, CD₂Cl₂) of compound **5c** at 299K (1) and 213K (2)



³¹P{¹H} NMR (243 MHz, CD₂Cl₂) of compound **5c** at 299K (1) and 213K (2)

Dynamic ¹H NMR (600 MHz, CD₂Cl₂):





Dynamic ¹⁹F NMR (564 MHz, CD₂Cl₂)

 $\Delta G^{\ddagger}[T_{c}, \Delta v(T)] = RT_{c}(22.96 + \ln(T_{c}/\Delta v)) \text{ [J/mol]}$ $T_{c} = \text{coalescence temperature [K]: 278 K (^{19}F, p-BC_{6}F_{5})}$ $\Delta v = \text{chemical shift difference [Hz] (^{19}F, p-BC_{6}F_{5}, 213 \text{ K}): 675 \text{ Hz}}$ R = 8.314 J/(mol K); 1 J = 0.239 cal $\Delta G^{\ddagger}[278K, \Delta v(213 \text{ K}) = 675 \text{ Hz}] = 51017 \text{ J/mol} = 12.2 \pm 0.3 \text{ kcal/mol}}$

X-ray crystal structure analysis of compound 5c: formula C₄₁H₃₈BF₁₀P, M = 762.49, colourless crystal, 0.18 x 0.11 x 0.07 mm, a = 22.3123(3), b = 15.8067(2), c = 23.2517(4) Å, $\beta = 106.554(1)^{\circ}$, V = 7860.6(2) Å³, $\rho_{calc} = 1.289$ gcm⁻³, $\mu = 0.145$ mm⁻¹, empirical absorption correction (0.974 \leq T \leq 0.989), Z = 8, monoclinic, space group $P2_1/n$ (No. 14), $\lambda = 0.71073$ Å, T = 223(2) K, ω and ϕ scans, 58267 reflections collected ($\pm h$, $\pm k$, $\pm l$), [(sin θ)/ λ] = 0.62 Å⁻¹, 15885 independent ($R_{int} = 0.069$) and 9981 observed reflections [$I > 2 \sigma$ (I)], 975 refined parameters, R = 0.073, $wR^2 = 0.182$, max. (min.) residual electron density 0.40 (-0.26) e.Å⁻³, hydrogen atoms calculated and refined as riding atoms.



Synthesis of compound 5d.



The reaction mixture of the borane **1b** (0.224 g, 0.5 mmol, 1 eq) and the phosphane **4b** (0.167 g, 0.5 mmol, 1 eq) in toluene (5 mL) was heated at 40 °C for 3 d. Then all volatiles were removed in vacuo and the residue was washed with pentane (1 × 3 mL). After drying in vacuo compound **5d** (0.304 g, 0.39 mmol, 78 %) was obtained as a light yellow solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a dichloromethane solution of compound **5d** at -35 °C. **IR** (KBr): $\tilde{\nu}$ / cm⁻¹ = 3026, 2965, 2927, 1643, 1606, 1517, 1465, 1381, 1288, 1113, 978. **Decomp**.: 190 °C. **Anal. Calc.** for C₄₃H₃₄BF₁₀P: C: 66.00; H: 4.38. Found: C: 66.06; H: 4.09.

¹**H NMR** (600 MHz, 213 K, CD₂Cl₂): δ = 7.28 (m, 2H, *o*-Ph), 7.23 (m, 2H, *m*-Ph), 7.18 (m, 1H, *p*-Ph), 7.17 (d, ³*J*_{HH} = 16.3 Hz, 1H, =CH), 6.98 (m, 1H, *m*-Mes^A), 6.82 (m, 1H, *m*²-Mes^A), 6.79 (m, 1H, *m*-Mes^B), 6.42 (d, ³*J*_{HH} = 16.3 Hz, 1H, =CH^{Ph}), 6.40 (m, 1H, *m*²-Mes^B), 5.38, 5.12 (each m, each 1H, =CH₂), 2.57 (s, 3H, *o*-CH₃^{MesB}), 2.45 (br s, 3H, *o*-CH₃^{MesA}), 2.23 (s, 3H, *p*-CH₃^{MesA}), 2.20 (s, 3H, *o*²-CH₃^{MesA}), 2.11 (s, 3H, *p*-CH₃^{MesB}), 1.63 (s, 3H, *o*²-CH₃^{MesB}), 1.58 (s, 3H, CH₃).

¹³C{¹H} NMR (151 MHz, 213 K, CD₂Cl₂): $\delta = 166.9$ (d, ²*J*_{PC} = 29.2 Hz, =CB), 143.3 (d, ²*J*_{PC} = 17.8 Hz, *o*-Mes^A), 142.9 (*o*'-Mes^A), 142.4 (d, ²*J*_{PC} = 3.2 Hz, *o*-Mes^B), 141.4 (d, ⁴*J*_{PC} = 2.3 Hz, *p*-Mes^A), 140.0 (d, ⁴*J*_{PC} = 2.7 Hz, *p*-Mes^B), 139.84 (d, ²*J*_{PC} = 13.9 Hz, *o*'-Mes^B), 139.78 (d, ²*J*_{PC} = 2.9 Hz, =C^{Me}), 137.8 (d, ¹*J*_{PC} = 51.9 Hz, =CP), 137.0 (br, =CH^{Ph}), 136.0 (*i*-Ph), 130.3 (d, ³*J*_{PC} = 8.3 Hz, *m*-Mes^B), 129.6 (d, ³*J*_{PC} = 5.9 Hz, *m*'-Mes^A), 129.4 (d, ³*J*_{PC} = 9.3 Hz, *m*-Mes^A), 129.3 (m, *m*'-Mes^B), 128.35 (*m*-Ph), 128.29 (*p*-Ph), 126.7 (*o*-Ph), 124.8 (d, ¹*J*_{PC} = 26.2 Hz, *i*-Mes^A), 124.3 (d, ¹*J*_{PC} = 40.4 Hz, *i*-Mes^B), 124.2 (d, ³*J*_{PC} = 46.4 Hz, =CH), 118.4 (m, =CH₂), 117.9 (br, *i*-C₆F₅), 116.1 (br, *i*-C₆F₅), 24.9 (m, *o*-CH₃^{MesA}), 20.1 (*p*-CH₃^{MesA}), [C₆F₅ not listed].

¹¹B{¹H} NMR (192 MHz, 213 K, CD₂Cl₂): $\delta = -1.1 (v_{1/2} \sim 1800 \text{ Hz}).$

¹¹B{¹H} NMR (192 MHz, 299 K, CD₂Cl₂): $\delta = 0.1 (v_{1/2} \sim 500 \text{ Hz}).$

³¹**P**{¹**H**} **NMR** (243 MHz, 213 K, CD₂Cl₂): $\delta = 12.4$ (partial relaxed 1:1:1:1 q, $J_{PB} \sim 25$ Hz).

³¹P{¹H} NMR (243 MHz, 299 K, CD₂Cl₂): $\delta = 13.8 (v_{1/2} \sim 35 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 213 K, CD₂Cl₂): δ = -125.5 (m, *o*), -131.7 (m, *o*'), -157.8 (t, ³J_{FF} = 21.3 Hz, *p*), -163.7 (m, *m*'), -164.5 (m, *m*)(each 1F, C₆F₅^A) [Δδ¹⁹F_{m,p} = 5.9, 6.7], -128.3 (m, *o*), -130.5 (m, *o*'), -159.2 (br t, ³J_{FF} ~ 20 Hz, *p*), -165.4 (m, *m*'), -165.8 (m, *m*)(each 1F, C₆F₅^B) [Δδ¹⁹F_{m,p} = 6.2, 6.6].

¹**H**, ¹**H GCOSY** (600 MHz / 600 MHz, 213 K, CD₂Cl₂)[selected traces]: δ ¹H / δ ¹H = 7.28 / 7.23 (*o*-Ph / *m*-Ph), 7.23 / 7.18 (*m*-Ph / *p*-Ph), 7.17 / 6.42 (=CH / =CH^{Ph}), 6.98 / 6.82, 2.45, 2.23, 2.20 (*m*-Mes^A / *m*'-Mes^A, *o*-CH₃^{MesA}, *p*-CH₃^{MesA}, *o*'-CH₃^{MesA}), 6.79 / 6.40, 2.57, 2.11, 1.63 (*m*-Mes^B / *m*'-Mes^B, *o*-CH₃^{MesB}, *p*-CH₃^{MesB}, *o*'-CH₃^{MesB}), 5.38 / 5.12, 1.58 (=CH₂ / =CH₂, CH₃).

¹H, ¹³C GHSQC (600 MHz / 151 MHz, 213 K, CD₂Cl₂): δ ¹H / δ ¹³C = 7.28 / 126.7 (*o*-Ph), 7.23 / 128.35 (*m*-Ph), 7.18 / 128.29 (*p*-Ph), 7.17 / 124.2 (=CH), 6.98 / 129.4 (*m*-Mes^A), 6.82 / 129.6 (*m*'-Mes^A), 6.79 / 130.3 (*m*-Mes^B), 6.42 / 137.0 (=CH^{Ph}), 6.40 / 129.3 (m, 1H, *m*'-Mes^B), 5.38 / 118.4 (=CH₂), 5.12 / 118.4 (=CH₂), 2.57 / 24.9 (*o*-CH₃^{MesB}), 2.45 / 23.7 (*o*-CH₃^{MesA}), 2.23 / 20.7 (*p*-CH₃^{MesA}), 2.20 / 22.6 (*o*'-CH₃^{MesA}), 2.11 / 20.1 (*p*-CH₃^{MesB}), 1.63 / 22.0 (*o*'-CH₃^{MesB}), 1.58 / 22.0 (CH₃).

¹**H**, ¹³**C GHMBC** (600 MHz / 151 MHz, 213 K, CD₂Cl₂)[selected traces]: δ ¹**H** / δ ¹³**C** = 7.28 / 137.0, 128.29, 126.7 (*o*-Ph / =CH^{Ph}, *p*-Ph, *o*-Ph), 7.23 / 136.0, 128.35 (*m*-Ph / *i*-Ph, *m*-Ph), 7.17 / 137.8, 136.0 (=CH / =CP, *i*-Ph), 6.98 / 129.6, 124.8, 23.7, 20.7 (*m*-Mes^A / *m*'-Mes^A, *i*-Mes^A, *o*-CH₃^{MesA}, *p*-CH₃^{MesA}), 6.82 / 129.4, 124.8, 22.6, 20.7 (*m*'-Mes^A / *m*-Mes^A, *i*-Mes^A, *o*-CH₃^{MesA}, *p*-CH₃^{MesA}), 6.79 / 129.3, 124.3, 24.9, 20.1 (*m*-Mes^B / *m*'-Mes^B, *i*-Mes^B, *o*-CH₃^{MesB}, *p*-CH₃^{MesB}), 6.42 / 166.9, 136.0, 126.6 (=CH^{Ph} / =CB, *i*-Ph, *o*-Ph), 6.40 / 130.3, 124.3, 22.0, 20.1 (*m*'-Mes^B / *m*-Mes^B, *i*-Mes^B, *i*-Mes^B, *i*-Mes^B, *i*-Mes^B, *i*-Mes^A, *i*-Mes^A / *o*-Ph), 6.40 / 130.3, 124.3, 22.0, 20.1 (*m*'-Mes^B / *m*-Mes^B, *i*-Mes^B, *i*-Mes^B, *i*-Mes^A, *i*-Mes^A, *i*-Mes^A, *i*-Mes^A, *i*-Mes^A, *i*-Mes^A, *i*-Mes^A, *i*-Mes^B, *m*-Mes^B, *i*-Mes^B, *i*-Mes^B, *i*-Mes^B, *i*-Mes^B, *i*-Mes^A, *i*-Mes

¹⁹**F**, ¹⁹**F GCOSY** (564 MHz / 564 MHz, 213 K, CD₂Cl₂): $\delta^{19}F / \delta^{19}F = -125.5 / 164.4$ (*o*-C₆F₅^A / *m*-C₆F₅^A), -128.3 / -165.8 (*o*-C₆F₅^B / *m*-C₆F₅^B), -130.5 / -165.4 (*o*'-C₆F₅^B / *m*'-C₆F₅^B), -131.7 / -163.7 (*o*'-C₆F₅^A / *m*'-C₆F₅^A), -157.8 / -163.7, -164.4 (*p*-C₆F₅^A / *m*'-C₆F₅^A, *m*-C₆F₅^A), -159.2 / -165.4, -165.8 (*p*-C₆F₅^B / *m*'-C₆F₅^B, *m*-C₆F₅^B).



-10 -30 -50 -70 -90

 $^{11}B\{^1H\}$ NMR (192 MHz, 213 K, CD₂Cl₂) and $^{31}P\{^1H\}$ NMR (243 MHz, 213 K,

CD₂Cl₂) of compound **5d**





14.9 14.7 14.5 14.3 14.1 13.9 13.7 13.5 13.3 13.1 12.9 12.7 12.5 12.3 12.1 11.9 11.7 11.5

 $^{31}P{^{1}H} NMR$ (243 MHz, CD₂Cl₂) of compound **5d** at 299K (1) and 213K (2)



Dynamic ¹H NMR (600 MHz, CD₂Cl₂):

Dynamic ¹⁹F NMR (564 MHz, CD₂Cl₂):



 $\Delta G^{\ddagger}[T_{c}, \Delta v(T)] = RT_{c}(22.96 + \ln(T_{c}/\Delta v)) [J/mol]$ $T_{c} = \text{coalescence temperature [K]: 290 K (^{19}F, p-BC_{6}F_{5})$ $\Delta v = \text{chemical shift difference [Hz] (^{19}F, p-BC_{6}F_{5}, 213 \text{ K}): 842 \text{ Hz}}$ R = 8.314 J/(mol K); 1 J = 0.239 cal $\Delta G^{\ddagger}[290\text{K}, \Delta v(213 \text{ K}) = 842 \text{ Hz}] = 57928 \text{ J/mol} = 13.8 \pm 0.3 \text{ kcal/mol}$

X-ray crystal structure analysis of compound 5d: formula C₄₃H₃₄BF₁₀P, M = 782.48, colourless crystal, 0.20 x 0.13 x 0.10 mm, a = 21.8974(5), b = 20.2120(5), c = 20.4155(7) Å, $\beta = 108.587(1)^{\circ}$, V = 8564.4(4) Å³, $\rho_{calc} = 1.214$ gcm⁻³, $\mu = 1.195$ mm⁻¹, empirical absorption correction (0.796 \leq T \leq 0.889), Z = 8, monoclinic, space group C2/c (No. 15), $\lambda = 1.54178$ Å, T = 223(2) K, ω and ϕ scans, 29400 reflections collected ($\pm h$, $\pm k$, $\pm l$), [($\sin\theta$)/ λ] = 0.60 Å⁻¹, 7537 independent ($R_{int} = 0.049$) and 5941 observed reflections [$I > 2 \sigma(I)$], 503 refined parameters, R = 0.045, $wR^2 = 0.130$, max. (min.) residual electron density 0.21 (-0.23) e.Å⁻³, hydrogen atoms calculated and refined as riding atoms.



Synthesis of compound 7a.



The reaction mixture of the borane **1a** (0.214 g, 0.5 mmol, 1 eq) and the phosphane **6** (0.145 g, 0.5 mmol, 1 eq) in toluene (5 mL) was heated at 60 °C for 3 d. Then all volatiles were removed in vacuo and the residue was washed with pentane (1 × 3 mL). After drying in vacuo compound **7a** (0.293 g, 0.41 mmol, 82 %) was obtained as a light yellow solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to dichloromethane solution of compound **7a** at -35 °C. **IR** (KBr): $\tilde{\gamma}$ / cm⁻¹ = 2952, 2864, 1639, 1516, 1457, 1384 1282, 1097, 972, 749, 693. **M.p.**: 155 °C. **Anal. Calc.** for C₃₈H₃₀BF₁₀P: C: 63.53; H: 4.21. Found: C: 63.32; H: 3.87.

¹**H NMR** (600 MHz, 299 K, C₆D₆): δ = 7.35 (m, 4H, *o*-Ph), 6.97 (dd, ³*J*_{HH} = 15.8 Hz, ⁴*J*_{PH} = 2.1 Hz, 1H, =CH), 6.91 (m, 2H, *p*-Ph), 6.85 (m, 4H, *m*-Ph), 6.16 (d, ³*J*_{HH} = 15.8 Hz, 1H, =CH^{*t*Bu}), 5.92 (m, 1H, =CH^{cy}), 2.25 (2H), 1.84 (2H), 1.39 (4H)(each m, CH₂^{cy}), 0.92 (s, 9H, ^{*t*}Bu).

¹³C{¹H} NMR (151 MHz, 299 K, C₆D₆): $\delta = 175.0$ (br, =CB), 154.2 (=CH^{*i*Bu}), 148.7 (dm, ¹*J*_{FC} ~ 240 Hz, C₆F₅), 140.1 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 137.4 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 134.1 (d, ²*J*_{PC} = 2.4 Hz, =C^{cy}), 132.1 (d, ²*J*_{PC} = 9.1 Hz, *o*-Ph), 131.6 (d, ⁴*J*_{PC} = 3.0 Hz, *p*-Ph), 130.7 (d, ¹*J*_{PC} = 55.3 Hz, =CP), 130.4 (d, ³*J*_{PC} = 7.2 Hz, =CH^{cy}), 128.8 (d, ³*J*_{PC} = 10.4 Hz, *m*-Ph), 126.7 (d, ¹*J*_{PC} = 49.7 Hz, *i*-Ph), 123.5 (d, ³*J*_{PC} = 47.9 Hz, =CH), 117.4 (br, *i*-C₆F₅), 33.8 (d, ⁵*J*_{PC} = 0.9 Hz, ^{*t*}Bu), 29.4 (d, ³*J*_{PC} = 5.0 Hz), 25.8, 22.9, 22.0 (CH₂^{cy}), 29.0 (^{*t*}Bu),.

¹¹B{¹H} NMR (192 MHz, 299 K, C₆D₆): δ = -8.5 (v_{1/2} ~ 300 Hz).

³¹**P**{¹**H**} **NMR** (243 MHz, 299 K, C₆D₆): $\delta = 10.1 (v_{1/2} \sim 65 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 299 K, C₆D₆): δ = -129.2 (m, 2F, *o*-C₆F₅), -158.1 (tm, ³J_{FF} = 20.8 Hz, 1F, *p*-C₆F₅), -164.6 (m, 2F, *m*-C₆F₅) [Δδ¹⁹F_{m,p} = 6.5].

³¹**P**{¹**H**} **NMR** (243 MHz, 299 K, C_6D_6): $\delta = 10.1 (v_{1/2} \sim 60 \text{ Hz})$.

¹**H**, ¹**H-GCOSY** (600 MHz / 600 MHz, 299 K, C₆D₆): δ ¹H / δ ¹H = 7.35 / 6.85 (*o*-Ph / *m*-Ph), 6.97 / 6.16 (=CH / =CH^{'Bu}), 6.91 / 6.85 (*p*-Ph / *m*-Ph), 5.92 / 2.25, 1.84 (=CH^{2-cy} / CH₂^{6-cy}, CH₂^{3-cy}), 2.25 / 1.84, 1.39 (CH₂^{6-cy} / CH₂^{3-cy}, CH₂^{5-cy}), 1.84 / 1.39 (CH₂^{3-cy} / CH₂^{4-cy}).

¹**H**, ¹³**C-GHSQC** (600 MHz / 151 MHz, 299 K, C₆D₆): δ ¹H / δ ¹³C = 7.35 / 132.1 (*o*-Ph), 6.97 / 123.5 (=CH), 6.91 / 131.6 (*p*-Ph), 6.85 / 128.8 (*m*-Ph), 6.16 / 154.2 (=CH^{*t*Bu}), 5.92 / 130.4 (=CH^{cy}), 2.25 / 29.4 (CH₂^{6-cy}), 1.84 / 25.8 (CH₂^{3-cy}), 1.39 / 22.9 (CH₂^{5-cy}), 1.39 / 22.0 (CH₂^{4-cy}), 0.92 / 29.0 (^{*t*}Bu).

¹H, ¹³C-GHMBC (600 MHz / 151 MHz, 299 K, C₆D₆)[selected traces]: δ ¹H / δ ¹³C = 7.35 / 132.1, 131.6 (*o*-Ph / *o*-Ph, *p*-Ph), 6.97 / 130.8, 33.8 (=CH / =CP, ^tBu), 6.85 / 128.8, 126.7 (*m*-Ph / *m*-Ph, *i*-Ph), 6.16 / 175.0, 33.8, 29.0 (=CH^{tBu} / =CB, ^tBu, ^tBu), 5.92 / 25.8, 22.0 (=CH^{cy} / CH₂^{3-cy}, CH₂^{4-cy}), 2.25 / 134.1, 130.4, 22.9, 22.0 (CH₂^{6-cy} / =C^{cy}, =CH^{cy}, CH₂^{5-cy}, CH₂^{4-cy}), 0.92 / 154.2, 33.8, 29.0 (^tBu / =CH^{tBu}, ^tBu, ^tBu).

¹⁹**F**, ¹⁹**F**-**GCOSY** (564 MHz / 564 MHz, 299 K, C_6D_6): $\delta^{19}F / \delta^{19}F = -129.2 / -164.6$ (*o*- C_6F_5 / m - C_6F_5), -158.1 / -164.4 (*p*- C_6F_5 / m - C_6F_5).



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¹¹B{¹H} NMR (192 MHz, 299 K, C₆D₆) and ³¹P{¹H} NMR (243 MHz, 299 K, C₆D₆) of compound **7a**

X-ray crystal structure analysis of compound 7a: formula $C_{38}H_{30}BF_{10}P$, M = 718.40, colourless crystal, 0.27 x 0.20 x 0.17 mm, a = 10.1386(2), b = 10.9523(2), c = 16.0070(4) Å, $\alpha = 94.704(1)$, $\beta = 90.060(1)$, $\gamma = 104.481(2)^{\circ}$, V = 1714.74(6) Å³, $\rho_{calc} = 1.391$ gcm⁻³, $\mu = 0.162$ mm⁻¹, empirical absorption correction (0.957 $\leq T \leq 0.973$), Z = 2, triclinic, space group $P\bar{1}$ (No. 2), $\lambda = 0.71073$ Å, T = 223(2) K, ω and ϕ scans, 16073 reflections collected ($\pm h$, $\pm k$, $\pm l$), [($\sin\theta$)/ λ] = 0.60 Å⁻¹, 6788 independent ($R_{int} = 0.039$) and 6089 observed reflections [$I > 2 \sigma$ (I)], 454 refined parameters, R = 0.049, $wR^2 = 0.129$, max. (min.) residual electron density 0.37 (-0.34) e.Å⁻³, hydrogen atoms calculated and refined as riding atoms.



Synthesis of compound 7b.



The reaction mixture of the borane **1b** (0.224 g, 0.5 mmol, 1 eq) and the phosphane **6** (0.145 g, 0.5 mmol, 1 eq) in toluene (5 mL) was heated at 60 °C for 3 d. Then all volatiles were removed in vacuo and the residue was washed with pentane (1×3 mL). After drying in vacuo compound **7b** (0.318 g, 0.43 mmol, 86 %) was obtained as a light yellow solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to dichloromethane solution of compound **7b** at

-35 °C. **IR** (KBr): $\tilde{\nu}$ / cm⁻¹ = 3059, 2933, 2859, 1641, 1516, 1456, 1381, 1287, 1102, 972, 743, 693. **Decomp**.: 183 °C. **Anal. Calc.** for C₄₀H₂₆BF₁₀P: C: 65.06; H: 3.55. Found: C: 64.72; H: 3.21.

¹**H NMR** (600 MHz, 299 K, C₆D₆): δ = 7.77 (dd, ³*J*_{HH} = 16.0 Hz, ⁴*J*_{PH} = 2.2 Hz, 1H, =CH), 7.35 (m, 4H, *o*-Ph^P), 7.32 (m, 2H, *o*-Ph), 7.07 (d, ³*J*_{HH} = 16.0 Hz, 1H, =CH^{Ph}), 6.99 (m, 2H, *m*-Ph), 6.96 (m, 1H, *p*-Ph), 6.91 (m, 2H, *p*-Ph^P), 6.85 (m, 4H, *m*-Ph^P), 5.96 (m, 1H, =CH^{cy}), 2.23 (2H), 1.85 (2H), 1.40 (4H)(each m, CH₂^{cy}).

¹³C{¹H} NMR (151 MHz, 299 K, C₆D₆): $\delta = 173.4$ (br, =CB), 148.8 (dm, ¹*J*_{FC} ~ 240 Hz, C₆F₅), 140.2 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 139.9 (=CH^{Ph}), 137.4 (dm, ¹*J*_{FC} ~ 250 Hz, C₆F₅), 137.2 (d, ⁵*J*_{PC} = 1.2 Hz, *i*-Ph), 134.2 (d, ²*J*_{PC} = 2.0 Hz, =C^{cy}), 132.9 (d, ¹*J*_{PC} = 54.7 Hz, =CP), 132.1 (d, ²*J*_{PC} = 9.1 Hz, *o*-Ph^P), 131.7 (d, ⁴*J*_{PC} = 3.0 Hz, *p*-Ph^P), 131.3 (dm, ³*J*_{PC} = 7.5 Hz, =CH^{cy}), 129.0 (*m*-Ph), 128.89 (d, ³*J*_{PC} = 10.4 Hz, *m*-Ph^P), 128.86 (*p*-Ph), 127.6 (*o*-Ph), 126.8 (d, ³*J*_{PC} = 48.8 Hz, =CH), 126.5 (d, ¹*J*_{PC} = 40.1 Hz, *i*-Ph^P), 117.1 (br, *i*-C₆F₅), 29.5 (d, ³*J*_{PC} = 4.8 Hz), 25.9, 22.9, 22.0 (CH₂^{cy}).

¹¹B{¹H} NMR (192 MHz, 299 K, C₆D₆): $\delta = -8.2 (v_{1/2} \sim 350 \text{ Hz}).$

³¹**P**{¹**H**} **NMR** (243 MHz, 299 K, C_6D_6): $\delta = 10.4 (v_{1/2} \sim 50 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 299 K, C₆D₆): δ = -129.2 (m, 2F, *o*-C₆F₅), -157.5 (tm, ³J_{FF} = 20.8 Hz, 1F, *p*-C₆F₅), -164.2 (m, 2F, *m*-C₆F₅)[Δδ¹⁹F_{m,p} = 6.7].

³¹**P**{¹**H**} **NMR** (243 MHz, 299 K, C₆D₆): $\delta = 10.4 (v_{1/2} \sim 50 \text{ Hz}).$

¹**H**, ¹**H-GCOSY** (600 MHz / 600 MHz, 299 K, C₆D₆): δ ¹H / δ ¹H = 7.77 / 7.07 (=CH / =CH^{Ph}), 7.35 / 6.85 (*o*-Ph^P / *m*-Ph^P), 7.32 / 6.99 (*o*-Ph / *m*-Ph), 6.99 / 6.96 (*m*-Ph / *p*-Ph), 6.91 / 6.85 (*p*-Ph^P / *m*-Ph^P), 5.96 / 2.23, 1.85 (=CH^{cy} / CH₂^{6-cy}, CH₂^{3-cy}), 2.23 / 1.84, 1.40 (CH₂^{6-cy} / CH₂^{3-cy}, CH₂^{5-cy}), 1.85 / 1.40 (CH₂^{3-cy} / CH₂^{4-cy}).

¹H, ¹³C-GHSQC (600 MHz / 151 MHz, 299 K, C₆D₆): δ ¹H / δ ¹³C = 7.77 / 126.8 (=CH), 7.35 / 132.1 (*o*-Ph^P), 7.32 / 127.6 (*o*-Ph), 7.07 / 139.9 (=CH^{Ph}), 6.99 / 129.0 (*m*-Ph), 6.96 / 128.86 (*p*-Ph), 6.91 / 131.7 (*p*-Ph^P), 6.85 / 128.89 (*m*-Ph^P), 5.96 / 131.3 (=CH^{cy}), 2.23 / 29.5 (CH₂^{6-cy}), 1.85 / 25.9 (CH₂^{3-cy}), 1.40 / 22.9 (CH₂^{5-cy}), 1.40 / 22.0 (CH₂^{4-cy}).

¹H, ¹³C-GHMBC (600 MHz / 151 MHz, 299 K, C₆D₆)[selected traces]: δ ¹H / δ ¹³C = 7.77 / 137.2, 132.9 (=CH / *i*-Ph, =CP), 7.35 / 132.1, 131.7 (*o*-Ph^P / *o*-Ph^P, *p*-Ph^P), 7.32

/ 139.9, 128.86, 127.6 (*o*-Ph / =CH^{Ph}, *p*-Ph, *o*-Ph), 7.07 / 173.4, 122.6 (=CH^{Ph} / =CB, *o*-Ph), 6.99 / 137.2, 129.0 (*m*-Ph / *i*-Ph, *m*-Ph), 6.85 / 128.89, 126.5 (*m*-Ph^P / *m*-Ph^P, *i*-Ph^P), 5.96 / 25.9, 22.0 (=CH^{cy} / CH₂^{3-cy}, CH₂^{4-cy}), 2.23 / 134.2, 131.3, 22.9, 22.0 (CH₂^{6-cy} / =C^{cy}, =CH^{cy}, CH₂^{5-cy}, CH₂^{4-cy}),.

¹⁹**F**, ¹⁹**F**-**GCOSY** (564 MHz / 564 MHz, 299 K, C_6D_6): $\delta^{19}F / \delta^{19}F = -129.2 / -164.2$ (*o*- C_6F_5 / m - C_6F_5), -157.5 / -164.2 (*p*- C_6F_5 / m - C_6F_5).





290 270 250 230 210 190 170 150 130 110 90 70 50 30 10 -10 -30 -50 -70 -90

¹¹B{¹H} NMR (192 MHz, 299 K, C₆D₆) and ³¹P{¹H} NMR (243 MHz, 299 K, C₆D₆) of compound **7b**



170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 ${}^{13}C{}^{1}H$ NMR (151 MHz, 299 K, C₆D₆) of compound **7b**



X-ray crystal structure analysis of compound 7b: formula $C_{40}H_{26}BF_{10}P$, M = 738.39, colourless crystal, 0.35 x 0.20 x 0.08 mm, a = 12.0744(3), b = 28.1047(6), c = 11.3117(2) Å, $\beta = 99.395(1)^{\circ}$, V = 3787.1(1) Å³, $\rho_{calc} = 1.295$ gcm⁻³, $\mu = 0.149$ mm⁻¹, empirical absorption correction (0.949 $\leq T \leq 0.988$), Z = 4, monoclinic, space group $P2_1/c$ (No. 14), $\lambda = 0.71073$ Å, T = 223(2) K, ω and ϕ scans, 17587 reflections collected ($\pm h$, $\pm k$, $\pm l$), [($\sin\theta$)/ λ] = 0.60 Å⁻¹, 6486 independent ($R_{int} = 0.056$) and 5195 observed reflections [$I > 2 \sigma(I)$], 488 refined parameters, R = 0.073, $wR^2 = 0.211$, max. (min.) residual electron density 0.48 (-0.34) e.Å⁻³, hydrogen atoms calculated and refined as riding atoms.



Synthesis of compounds 10a/10a'.



A solution of compound **5a** (0.271 g, 0.4 mmol) in toluene (4 mL) was heated at 100 °C for 3 d. Then all volatiles were removed in vacuo and the residue was washed with pentane (1 × 3 mL). After drying in vacuo a light yellow solid (0.244 g, 90 %) was obtained. In CD_2Cl_2 solution two isomers were observed (ca. 80:20 [¹H]).

Major isomer 10a.

¹H NMR (600 MHz, 299 K, CD₂Cl₂)[selected resonances]: $\delta = 6.33$ (m, 1H, =CH),

2.34 (td, ${}^{2}J_{HH} = 17.4$ Hz, ${}^{4}J_{PH} = 5.2$ Hz, 1H, CH₂), 2.25 (dd, ${}^{2}J_{HH} = 17.0$ Hz, ${}^{3}J_{HH} = 6.5$ Hz, 1H, CH₂), 2.02 (ddm, ${}^{3}J_{HH} = 17.7$ Hz, ${}^{3}J_{HH} = 6.5$ Hz, 1H, CH), 1.56 (d, ${}^{4}J_{PH} = 2.9$ Hz, 3H, CH₃), 0.99 (s, 9H, ${}^{t}Bu$).

¹³C{¹H} **NMR** (151 MHz, 299 K, CD₂Cl₂): $\delta = 160.3$ (dd, ²*J*_{PC} = 8.9 Hz, ⁵*J*_{FC} = 1.8 Hz, =C^{Me}), 134.5 (d, ⁴*J*_{PC} = 3.1 Hz), 134.1 (d, ⁴*J*_{PC} = 3.2 Hz)(*p*-Ph), 133.0 (dd, ²*J*_{PC} = 11.2 Hz, ⁵*J*_{FC} = 1.9 Hz), 132.6 (dd, ²*J*_{PC} = 11.5 Hz, ⁵*J*_{FC} = 2.3 Hz)(*o*-Ph), 130.3 (d, ³*J*_{PC} = 13.3 Hz), 130.0 (d, ³*J*_{PC} = 13.1 Hz)(*m*-Ph), 127.8 (m, =CH), 125.1 (dd, ¹*J*_{PC} = 87.0 Hz, ⁴*J*_{FC} = 4.1 Hz), 119.2 (d, ¹*J*_{PC} = 85.2 Hz)(*i*-Ph), 114.3 (dd, ¹*J*_{PC} = 85.1 Hz, ⁴*J*_{FC} = 6.2 Hz, =CP), 42.9 (d, ⁴*J*_{PC} = 2.1 Hz, CH), 35.6 (d, ³*J*_{PC} = 13.2 Hz, CH₂), 32.5 (^tBu), 27.6 (^tBu), 25.1 (d, ³*J*_{PC} = 9.1 Hz, CH₃). [C₆F₅, C₆F₄ not listed]

¹¹B{¹H} NMR (192 MHz, 299 K, CD₂Cl₂): δ = -1.5 (v_{1/2} ~ 200 Hz).

³¹**P**{¹**H**} **NMR** (243 MHz, 299 K, CD₂Cl₂): $\delta = -1.5$ (v_{1/2} ~ 20 Hz).

¹⁹**F NMR** (564 MHz, 299 K, CD₂Cl₂): δ = -126.8, -131.4, -147.1, -158.9 (each m, each 1F, C₆F₄), -132.8 (m, 2F, *o*), -161.0 (t, ³*J*_{FF} = 20.1 Hz, 1F, *p*), -165.8 (m, 2F, *m*)(C₆F₅)[Δδ¹⁹F_{m,p} = 4.8], -199.2 (br, 1F, BF).

¹**H**, ¹**H**-**GCOSY** (600 MHz / 600 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹H = 7.70 / 7.57 (*p*-Ph^A / *m*-Ph^A), 7.67 / 7.49 (*o*-Ph^B / *m*-Ph^B), 6.33 / 2.25, 2.02, 1.56 (=CH / CH₂, CH, CH₃), 2.34 / 2.25, 1.56 (CH₂ / CH₂, CH₃), 2.25 / 2.02 (CH₂ / CH).

¹**H**, ¹³**C-GHSQC** (600 MHz / 151 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹³C = 7.70 / 134.1 (*p*-Ph^A), 7.67 / 134.5 (*p*-Ph^B), 7.57 / 132.6, 130.3 (*o*-,*m*-Ph^A / *o*-Ph^A, *m*-Ph^A), 7.49 / 133.0, 130.0 (*o*-,*m*-Ph^B / *o*-Ph^B, *m*-Ph^B), 6.33 / 127.8 (=CH), 2.34 / 35.6 (CH₂), 2.25 / 35.6 (CH₂), 2.02 / 42.9 (CH), 1.56 / 25.1 (CH₃), 0.99 / 27.6 (^{*t*}Bu).

¹**H**, ¹³**C-GHMBC** (600 MHz / 151 MHz, 299 K, CD₂Cl₂)[selected traces]: δ ¹H / δ ¹³C = 7.57 / 134.1, 132.6 (*o*-Ph^A / *p*-Ph^A, *o*-Ph^A), 7.57 / 130.3, 125.4 (*m*-Ph^A / *m*-Ph^A, *i*-Ph^A), 7.49 / 134.5, 133.0 (*o*-Ph^B / *p*-Ph^B, *o*-Ph^B), 7.49 / 130.0, 119.2 (*m*-Ph^B / *m*-Ph^B, *i*-Ph^B), 2.25 / 160.3, 127.8, 114.3, 42.9, 32.5, 25.1 (CH₂ / =C^{Me}, =CH, =CP, CH, ^{*t*}Bu, CH₃), 0.99 / 42.9, 32.5, 27.6 (^{*t*}Bu / CH, ^{*t*}Bu, ^{*t*}Bu).

¹⁹**F**, ¹⁹**F**-**GCOSY** (564 MHz / 564 MHz, 299 K, CD₂Cl₂): δ ¹⁹**F** / δ ¹⁹**F** = -126.8 / -147.1, 158.9 (C₆F₄ / C₆F₄, C₆F₄), -131.4 / -147.1 (C₆F₄ / C₆F₄), -132.8 / -165.8 (*o*-C₆F₅ / *m*-C₆F₅), -147.1 / -158.9 (C₆F₄ / C₆F₄), -161.0 / -165.8 (*p*-C₆F₅ / *m*-C₆F₅).

Minor isomer 10a'.

¹**H** NMR (600 MHz, 299 K, CD₂Cl₂)[selected resonances]: $\delta = 6.11$ (m, 1H, =CH), 2.55 (dddm, ²*J*_{HH} = 17.8 Hz, ³*J*_{HH} = 9.8 Hz, ⁴*J*_{PH} = 3.4 Hz, 1H, CH₂), 2.39 (dd, ²*J*_{HH} = 17.8 Hz, ³*J*_{HH} = 7.6 Hz, 1H, CH₂), 2.07 (m, 1H, CH), 1.55 (d, ⁴*J*_{PH} = 2.9 Hz, 3H, CH₃), 0.85 (s, 9H, ^{*t*}Bu).

¹³C{¹H} NMR (151 MHz, 299 K, CD₂Cl₂): δ = n.o. (=CB), 159.2 (dd, ²*J*_{PC} = 9.0 Hz, ⁵*J*_{FC} = 0.8 Hz, =C^{Me}), 134.3 (d, ⁴*J*_{PC} = 3.2 Hz), 134.2 (d, ⁴*J*_{PC} = 3.2 Hz)(*p*-Ph), 132.89 (dd, ²*J*_{PC} = 11.2 Hz, ⁵*J*_{FC} = 1.6 Hz), 132.87 (dd, ²*J*_{PC} = 11.6 Hz, ⁵*J*_{FC} = 2.2 Hz)(*o*-Ph), 130.2 (d, ³*J*_{PC} = 13.3 Hz), 130.1 (d, ³*J*_{PC} = 13.0 Hz)(*m*-Ph), 128.0 (m, =CH), 123.7 (dd, ¹*J*_{PC} = 86.6 Hz, ⁴*J*_{FC} = 3.1 Hz), 121.0 (d, ¹*J*_{PC} = 85.9 Hz)(*i*-Ph), 113.9 (dd, ¹*J*_{PC} = 85.8 Hz, ⁴*J*_{FC} = 4.2 Hz, =CP), 42.2 (d, ⁴*J*_{PC} = 1.9 Hz, CH), 35.5 (d, ³*J*_{PC} = 13.2 Hz, CH₂), 34.0 (^{*i*}Bu), 27.4 (^{*i*}Bu), 25.0 (d, ³*J*_{PC} = 9.2 Hz, CH₃). [C₆F₅, C₆F₄ not listed] ¹¹B{¹H} NMR (192 MHz, 299 K, CD₂Cl₂): δ = -1.5 (v_{1/2} ~ 200 Hz).

³¹P{¹H} NMR (243 MHz, 299 K, CD₂Cl₂): $\delta = -0.9 (v_{1/2} \sim 20 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 299 K, CD₂Cl₂): δ = -125.9, -130.4, -147.0, -158.7 (each m, each 1F, C₆F₄), -133.7 (m, 2F, *o*), -161.4 (t, ³J_{FF} = 20.1 Hz,1F, *p*), -166.1 (m, 2F, *m*)(C₆F₅)[Δδ¹⁹F_{m,p} = 4.7], -192.3 (br, 1F, BF).













-10 -90 -30 -50 -70

¹¹B{¹H} NMR (192 MHz, 299 K, CD₂Cl₂) and ³¹P{¹H} NMR (243 MHz, 299 K, CD₂Cl₂) of compounds **10a/10a**'



 $^1\text{H},\,^{13}\text{C-GHSQC}$ (600 MHz / 151 MHz, 299 K, CD₂Cl₂) of compounds 10a/10a'

Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a dichloromethane solution of the yellow solid at -35 $^{\circ}$ C.

The ¹H and ¹⁹F NMR data obtained from a solution of the crystal from the X-ray crystal structure analysis in CD_2Cl_2 are consistent with those listed for the major isomer **10a** (see above).



¹**H** NMR (600 MHz, 299 K, CD_2Cl_2) of (2) isomer mixure **10a/10a**' [light yellow solid, see above)] and (1) the crystal from the X-ray crystal structure analysis.



X-ray crystal structure analysis of compound 10a: formula $C_{35}H_{26}BF_{10}P \cdot CH_2Cl_2$, M = 763.26, colourless crystal, 0.18 x 0.04 x 0.02 mm, a = 20.7710(9), b = 10.2846(5), c = 18.6572(9) Å, $\beta = 120.235(4)^{\circ}$, V = 3443.4(3) Å³, $\rho_{calc} = 1.472$ gcm⁻³, $\mu = 2.862$ mm⁻¹, empirical absorption correction (0.626 $\leq T \leq 0.945$), Z = 4, monoclinic, space group Cc (No. 9), $\lambda = 1.54178$ Å, T = 223(2) K, ω and ϕ scans, 12481 reflections collected ($\pm h$, $\pm k$, $\pm l$), [($\sin\theta$)/ λ] = 0.60 Å⁻¹, 5034 independent ($R_{int} = 0.060$) and 4282 observed reflections [$I > 2 \sigma$ (I)], 484 refined parameters, R = 0.054, $wR^2 = 0.137$, max. (min.) residual electron density 0.35 (-0.27) e.Å⁻³, hydrogen atoms calculated and refined as riding atoms.



Synthesis of compound 11.



A solution of compound **5b** (0.175 g, 0.25 mmol) in benzene (5 mL) was heated at 120 $^{\circ}$ C for 1 d. After cooling to room temperature, the reaction mixture was photolyzed (UV lamp: HPK 125, Pyrex filter) for 1 d. Then all volatiles of the

reaction mixture were removed in vacuo and the residue was washed with pentane (1 × 3 mL). After drying in vacuo compound **11** (0.145 g, 82 %) was obtained as a light yellow solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a dichloromethane solution of compound **11** at -35 °C. **IR** (KBr): $\tilde{\nu}$ / cm⁻¹ = 3062, 1636, 1559, 1512, 1500, 1460, 1429, 1318, 1273, 1108, 1085, 1054, 962. **Decomp**.: 250 °C. **Anal. Calc.** for C₃₇H₂₂BF₁₀P: C: 63.64; H: 3.18. Found: C: 63.34; H: 3.12.

¹**H NMR** (600 MHz, 299 K, CD₂Cl₂): δ = 7.80, 7.69 (each m, each 1H, *p*-Ph^P), 7.79, 7.75 (each m, each 2H, *o*-Ph^P), 7.64, 7.54 (each m, each 2H, *m*-Ph^P), 7.42 (br m, 2H, *o*-Ph), 7.26 (m, 2H, *m*-Ph), 7.20 (m, 1H, *p*-Ph), 6.43 (dd, ³*J*_{HH} = 12.7 Hz, ⁵*J*_{PH} = 1.2 Hz, 1H, =CH^{Ph}), 6.15 (br d, ³*J*_{HH} = 12.7 Hz, 1H, =CH), 4.61, 3.53 (each m, each 1H, =CH₂), 1.20 (m, 3H, CH₃).

¹³C{¹H} NMR (151 MHz, 299 K, CD₂Cl₂): $\delta = 180.4$ (br, =CB), 139.6 (d, ²*J*_{PC} = 13.4 Hz, =C^{Me}), 138.8 (*i*-Ph), 134.8 (d, ⁴*J*_{PC} = 3.0 Hz), 134.4 (d, ⁴*J*_{PC} = 3.1 Hz)(*p*-Ph^P), 134.14 (d, ²*J*_{PC} = 10.0 Hz), 134.09(dd, ²*J*_{PC} = 10.9 Hz, *J* = 1.3 Hz)(*o*-Ph^P), 133.6 (d, ³*J*_{PC} = 21.8 Hz, =CH), 130.4 (br, =CH^{Ph}), 129.9 (d, ³*J*_{PC} = 12.7 Hz), 129.8 (d, ³*J*_{PC} = 12.9 Hz)(*m*-Ph^P), 128.8 (d, *J* = 3.2 Hz, *o*-Ph), 128.2 (*m*-Ph), 127.1 (*p*-Ph), 121.8 (d, ¹*J*_{PC} = 90.2 Hz), 118.3 (d, ¹*J*_{PC} = 90.1 Hz)(*i*-Ph^P), 121.2 (d, ³*J*_{PC} = 6.3 Hz, =CH₂), 117.5 (dd, ¹*J*_{PC} = 83.7 Hz, ³*J*_{FC} = 4.4 Hz,=CP), 23.9 (d, ³*J*_{PC} = 2.5 Hz, CH₃), [C₆F₅, C₆F₄ not listed].

¹¹**B**{¹**H**} **NMR** (192 MHz, 299 K, CD₂Cl₂): $\delta = -1.9$ (d, ¹*J*_{BF} ~ 50 Hz).

³¹P{¹H} NMR (243 MHz, 299 K, CD₂Cl₂): $\delta = -5.9 (v_{1/2} \sim 30 \text{ Hz}).$

¹⁹**F NMR** (564 MHz, 299 K, CD₂Cl₂): δ = -125.0, -129.0, -147.0, -157.6 (each m, each 1F, C₆F₄), -135.8 (br, 2F, *o*), -161.1 (t, ³*J*_{FF} = 20.1 Hz, 1F, *p*), -165.6 (br m, 2F, *m*)(C₆F₅)[Δδ¹⁹F_{m,p} = 4.5], -197.0 (br, 1F, BF).

¹H, ¹H-GCOSY (600 MHz / 600 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹H = 7.80 / 7.64 (*p*-Ph^{PB} / *m*-Ph^{PB}), 7.79 / 7.64 (*o*-Ph^{PB} / *m*-Ph^{PB}), 7.75 / 7.54 (*o*-Ph^{PA} / *m*-Ph^{PA}), 7.69 / 7.54 (*p*-Ph^{PA} / *m*-Ph^{PA}), 7.42 / 7.26 (*o*-Ph / *m*-Ph), 7.26 / 7.20 (*m*-Ph / *p*-Ph), 6.43 / 6.15 (=CH^{Ph} / =CH), 4.61 / 3.53, 1.20 (=CH₂ / =CH₂, CH₃), 3.53 / 1.20 (=CH₂ / CH₃). ¹H, ¹³C-GHSQC (600 MHz / 151 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹³C = 7.80 / 134.8 (*p*-Ph^{PB}), 7.79 / 134.14 (*o*-Ph^{PB}), 7.75 / 134.09 (*o*-Ph^{PA}), 7.69 / 134.4 (*p*-Ph^{PA}), 7.64 / 129.9 (*m*-Ph^{PB}), 7.54 / 129.8 (*m*-Ph^{PA}), 7.42 / 128.8 (*o*-Ph), 7.26 / 128.2 (*m*-Ph), 7.20 / 127.1 (*p*-Ph), 6.43 / 130.4 (=CH^{Ph}), 6.15 / 133.6 (=CH), 4.61, 3.53 / 121.2 (=CH₂), 1.20 / 23.9 (CH₃).

¹H, ¹³C-GHMBC (600 MHz / 151 MHz, 299 K, CD₂Cl₂)[selected traces]: δ ¹H / δ ¹³C = 7.79 / 134.8 (*o*-Ph^{PB} / *p*-Ph^{PB}),7.75 / 134.4 (*o*-Ph^{PA} / *p*-Ph^{PA}), 7.64 / 134.14, 129.9, 118.3 (*m*-Ph^{PB} / *o*-Ph^{PB}, *m*-Ph^{PB}, *i*-Ph^{PB}), 7.54 / 134.09, 129.8, 121.8 (*m*-Ph^{PA} / *o*-Ph^{PA}, *m*-Ph^{PA}, *i*-Ph^{PA}), 7.42 / 128.8, 128.2, 127.1 (*o*-Ph / *o*-Ph, *m*-Ph, *p*-Ph), 7.26 / 138.8, 128.8, 128.2 (*m*-Ph / *i*-Ph, *o*-Ph, *m*-Ph), 6.43 / 180.4, 128.8 (=CH^{Ph} / =CB, *o*-Ph), 4.61, 3.53 / 117.5, 23.9 (=CH₂ / =CP, CH₃), 1.20 / 139.6, 121.2, 117.5 (CH₃ / =C^{Me}, =CH₂, =CP).

¹⁹**F**, ¹⁹**F**-**GCOSY** (564 MHz / 564 MHz, 299 K, CD₂Cl₂): δ ¹⁹F / δ ¹⁹F = -125.0 / 157.6 (C₆F₄ / C₆F₄), -129.0 / -147.0 (C₆F₄ / C₆F₄), -161.1 / -165.6 (*p*-C₆F₅ / *m*-C₆F₅).









^{7.85} 7.80 7.75 7.70 7.65 7.60 7.55 7.50 7.45 7.40 7.35 7.30 7.25 7.20 7.15 7.10 ¹**H**, ¹³**C-GHSQC** (600 MHz / 151 MHz, 299 K, CD₂Cl₂) of compound **11**:

(1) $\delta^{1}H_{irr}$: 7.64 (*m*-Ph^P). (2) $\delta^{1}H_{irr}$: 7.54 (*m*-Ph^P). (3) $\delta^{1}H_{irr}$: 7.20 (*p*-Ph).



 ^{19}F NMR (564 MHz, 299 K, $CD_2Cl_2)$ of compound 11



290 270 250 230 210 190 170 150 130 110 90 70 10 -10 -30 -50 -70 50 30 ¹¹B{¹H} NMR (192 MHz, 299 K, CD₂Cl₂) and ³¹P{¹H} NMR (243 MHz, 299 K, CD₂Cl₂) of compound **11**

X-ray crystal structure analysis of compound 11: formula $C_{37}H_{22}BF_{10}P \cdot CH_2Cl_2$, M = 783.25, colourless crystal, 0.14 x 0.11 x 0.06 mm, a = 16.2406(5), b = 10.2753(3), c = 21.4263(8) Å, $\beta = 102.915(2)^{\circ}$, V = 3485.1(2) Å³, $\rho_{calc} = 1.493$ gcm⁻³, $\mu = 2.848$ mm⁻¹, empirical absorption correction (0.691 $\leq T \leq 0.847$), Z = 4, monoclinic, space group $P2_1/c$ (No. 14), $\lambda = 1.54178$ Å, T = 223(2) K, ω and ϕ scans, 26271 reflections collected ($\pm h$, $\pm k$, $\pm l$), [($\sin\theta$)/ λ] = 0.60 Å⁻¹, 5989 independent ($R_{int} = 0.087$) and 4390 observed reflections [$I > 2 \sigma$ (I)], 506 refined parameters, R = 0.050, $wR^2 = 0.137$, max. (min.) residual electron density 0.22 (-0.31) e.Å⁻³, the hydrogen atoms at C1 and C2 were refined freely; others were calculated and refined as riding atoms.



Generation of compounds 10b/10b'

A solution of compound **5b** (135.6 mg, 0.2 mmol) in toluene (2 mL) was heated at 120 °C for 1 d. Then all volatiles of the reaction mixture were removed in vacuo and the residue was washed with pentane (3 × 1 mL). After drying in vacuo a light yellow solid (117.4 mg, 86 %) was obtained. **Anal. Calc.** for $C_{37}H_{22}BF_{10}P$: C: 63.64; H: 3.18. Found: C: 63.56; H: 3.10.

In CD₂Cl₂ solution two isomers were observed (ca. 54:46 [¹H]). After characterization of the reaction mixture by NMR experiments the solution was photolyzed (UV lamp: HPK 125, Pyrex filter) at room temperature for 1 day to finally give compound **11**. *Major isomer 10b*

¹**H** NMR (500 MHz, 299 K, CD₂Cl₂)[selected resonaces]: $\delta = 6.41$ (m, 1H, =CH), 3.62 (br dm, ${}^{3}J_{\text{HH}} = 16.5$ Hz, 1H, CH), 2.73 (br ddd, ${}^{2}J_{\text{HH}} = 17.1$ Hz, ${}^{3}J_{\text{HH}} = 16.5$ Hz, ${}^{4}J_{PH} = 4.1$ Hz), 2.52 (dd, ${}^{2}J_{HH} = 17.1$ Hz, ${}^{3}J_{HH} = 6.8$ Hz)(each 1H, CH₂), 1.59 (d, ${}^{4}J_{PH} = 2.8$ Hz, 3H, CH₃).

¹³C{¹H} NMR (126 MHz, 299 K, CD₂Cl₂)[selected resonaces]: $\delta = 158.9$ (d, ²*J*_{PC} = 9.0 Hz, =C^{Me}), 129.7 (m, =CH), 114.9 (dm, ¹*J*_{PC} = 85.8 Hz, =CP), 42.7 (d, ³*J*_{PC} = 13.3 Hz, CH₂), 39.5 (d, ⁴*J*_{PC} = 2.0 Hz, CH), 24.9 (d, ³*J*_{PC} = 9.0 Hz, CH₃).

¹¹**B**{¹**H**} **NMR** (160 MHz, 299 K, CD₂Cl₂): δ = -1.4 (v_{1/2} ~ 200 Hz).

³¹**P**{¹**H**} **NMR** (202 MHz, 299 K, CD₂Cl₂): $\delta = -1.0 (v_{1/2} \sim 20 \text{ Hz}).$

¹⁹**F NMR** (470 MHz, 299 K, CD₂Cl₂): δ = -126.6, -130.7, -146.6, -158.47 (each m, each 1F, C₆F₄), -132.8 (m, 2F, *o*), -160.72 (t, ³*J*_{FF} = 19.7 Hz, 1F, *p*), -165.5 (m, 2F, *m*)(C₆F₅)[Δδ¹⁹F_{m,p} = 4.78], -197.7 (br, 1F, BF).

¹**H**, ¹**H**-**GCOSY** (500 MHz / 500 MHz, 299 K, CD_2Cl_2): δ ¹H / δ ¹H = 6.41 / 3.62, 2.52, 1.59 (=CH / CH₂, CH₃), 3.62 / 2.73, 2.52 (CH / CH₂, CH₂), 2.73 / 2.52, 1.59 (CH₂ / CH₂, CH₃). [Ph not listed]

¹H, ¹³C-GHSQC (500 MHz / 126 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹³C = 6.41 / 129.7 (=CH), 3.62 / 39.5 (CH), 2.73 / 42.7 (CH₂) , 2.52 / 42.7 (CH₂) , 1.59 / 24.9 (CH₃). [Ph not listed]

¹H, ¹³C-GHMBC (500 MHz / 126 MHz, 299 K, CD₂Cl₂): δ ¹H / δ ¹³C = 2.73 / 158.9, 129.7 (CH₂ / =C^{Me}, =CH), 2.52 / 158.9, 129.7, 114.9, 39.5, 24.9 (CH₂ / =C^{Me}, =CH, =CP, CH, CH₃), 1.59 / 158.9, 114.9, 42.7 (CH₃ / =C^{Me}, =CP, CH₂). [Ph not listed]

¹⁹**F**, ¹⁹**F**-**GCOSY** (470 MHz / 470 MHz, 299 K, CD₂Cl₂): δ ¹⁹F / δ ¹⁹F = -126.6 / 158.5 (C₆F₄ / C₆F₄), -130.7 / -146.6 (C₆F₄ / C₆F₄), -132.8 / -165.5 (*o*-C₆F₅ / *m*-C₆F₅).

Minor isomer **10b** '

¹**H NMR** (500 MHz, 299 K, CD₂Cl₂)[selected resonances]: $\delta = 6.60$ (m, 1H, =CH), 3.69 (m, 1H, CH), 3.00 (m), 2.70 (dd, ²*J*_{HH} = 17.2 Hz, ³*J*_{HH} = 4.7 Hz)(each 1H, CH₂), 1.47 (d, ⁴*J*_{PH} = 2.9 Hz, 3H, CH₃).

¹³C{¹H} NMR (126 MHz, 299 K, CD₂Cl₂)[selected resonaces]: $\delta = 158.1$ (d, ² $J_{PC} = 8.8$ Hz, =C^{Me}), 127.3 (m, =CH), 114.9 (dm, ¹ $J_{PC} = 86.8$ Hz, =CP), 41.0 (d, ³ $J_{PC} = 13.2$ Hz, CH₂), 36.0 (d, ⁴ $J_{PC} = 2.0$ Hz, CH), 25.5 (d, ³ $J_{PC} = 9.1$ Hz, CH₃).

¹¹B{¹H} NMR (160 MHz, 299 K, CD₂Cl₂): δ = -1.4 (v_{1/2} ~ 200 Hz).

³¹P{¹H} NMR (202 MHz, 299 K, CD₂Cl₂): $\delta = -1.1 (v_{1/2} \sim 20 \text{ Hz}).$

¹⁹**F NMR** (470 MHz, 299 K, CD₂Cl₂): δ = -126.6, -130.7, -146.6, -158.53 (each m, each 1F, C₆F₄), -133.1 (m, 2F, *o*), -160.67 (t, ${}^{3}J_{FF}$ = 19.8 Hz, 1F, *p*), -165.5 (m, 2F, *m*)(C₆F₅) [Δδ¹⁹F_{m,p} = 4.83], -197.6 (br, 1F, BF).



¹**H NMR** (500 MHz, 299 K, CD₂Cl₂) of compounds **10b/10b**' (# major isomer, * minor isomer)



⁷⁰ 165 160 155 150 145 140 135 130 125 120 115 110 105 100 95 90 85 80 75 70 65 60 55 50 45 40 35 30 25 20 1 ¹³C{¹H} NMR (126 MHz, 299 K, CD_2Cl_2) of compounds **10b/10b**' (# major isomer, * minor isomer)



Control experiment

1) We have synthesized a selected example of compounds **3** according to reference [10]. Then a solution of compound **3** in C_6D_6 was heated at 120 °C for 2 days. No nucleophilic substitution of the PPh₂ substituent to a C_6F_5 ring at boron was observed.



¹**H NMR** (299 K, C_6D_6) of compound **3** (600 MHz, bottom) and after heating (500

MHz, top)





2) A solution of compound **11** in C_6D_6 was heated at 120 °C for 2 days. No reaction was observed.





MHz, top)

