

Electronic Supplementary Material (ESI) for Organic & Biomolecular Chemistry
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NBD-based Fluorescent Chemosensor for the Selective Quantification of Copper and Sulfide in Aqueous Solution and Living Cells

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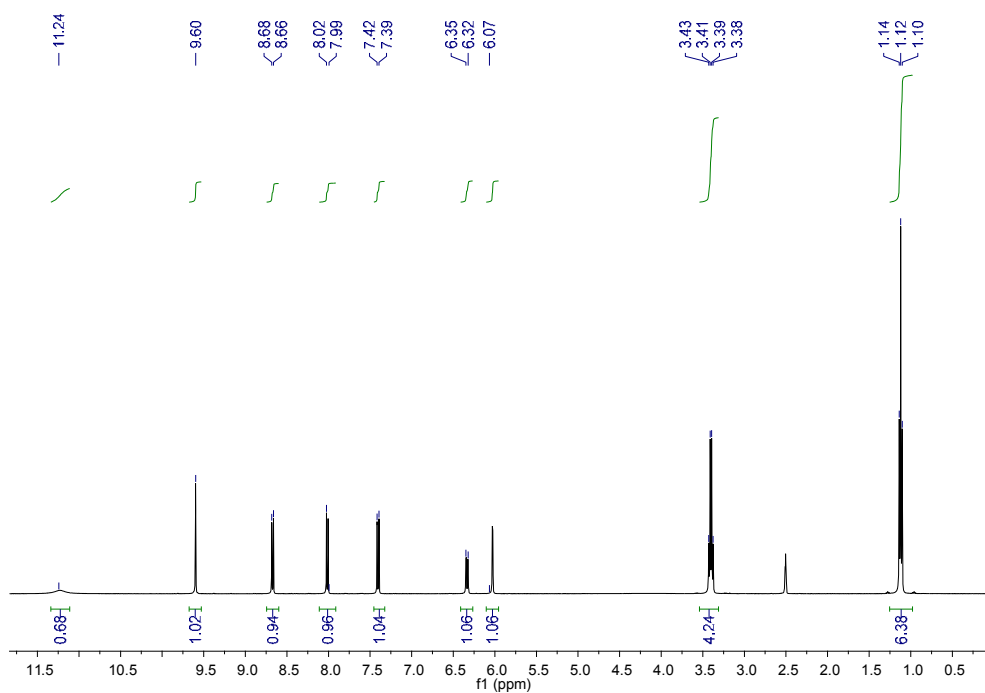


Fig. S1 $^1\text{H-NMR}$ of NL ($\text{DMSO-}d_6$)

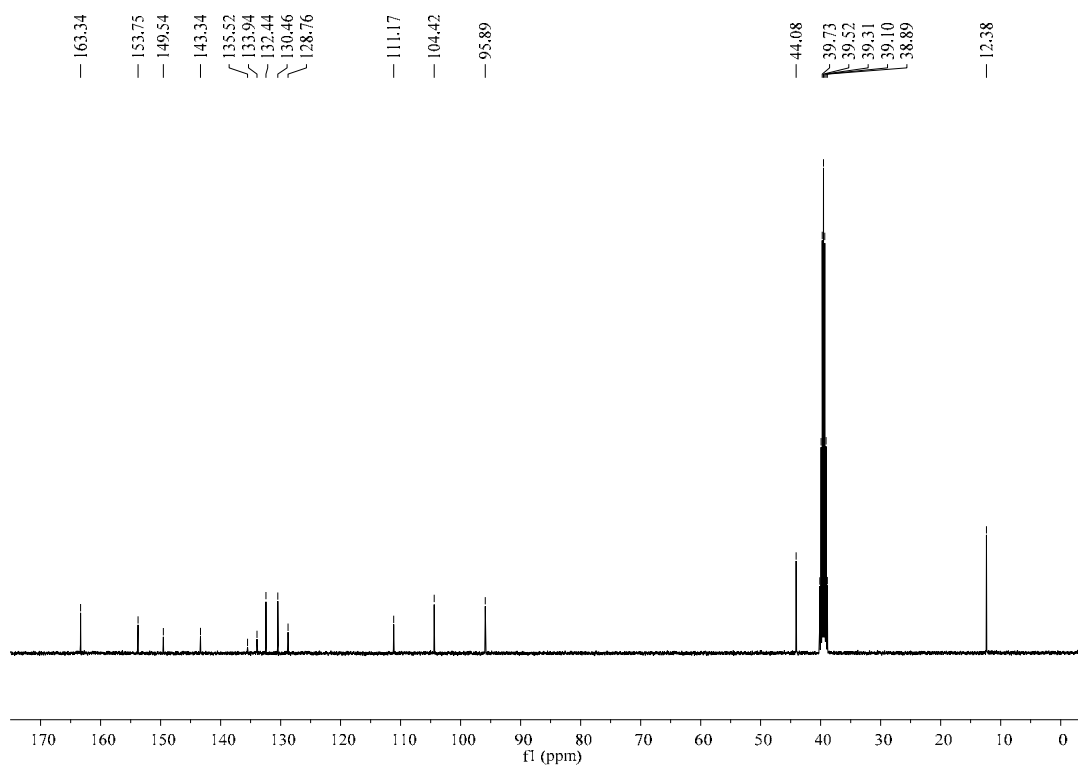


Fig. S2 $^{13}\text{C-NMR}$ of NL ($\text{DMSO-}d_6$)

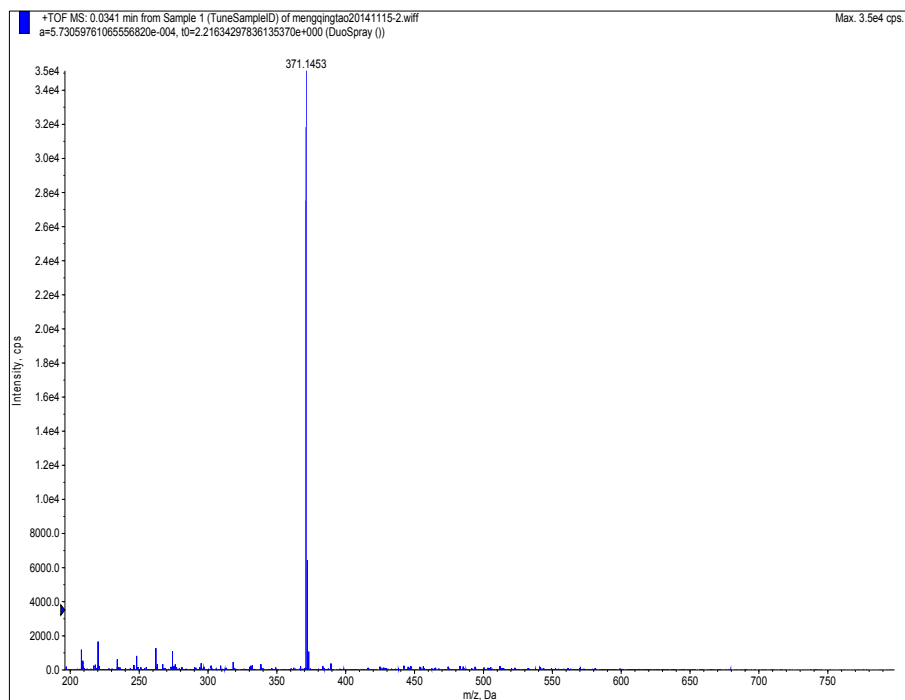


Fig. S3 High resolution TOF-MS of NL

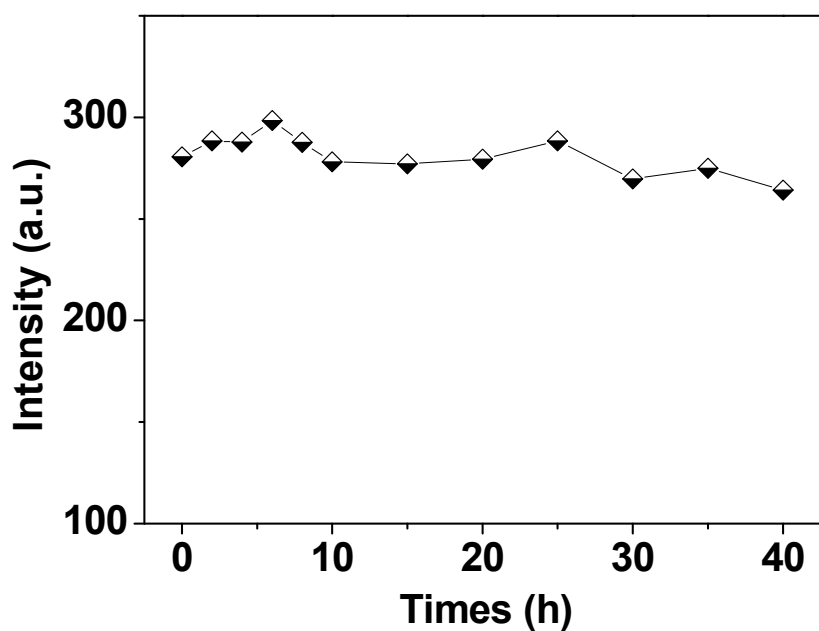


Fig. S4 Time-dependent fluorescence changes of NL in HEPES aqueous buffer (THF: H₂O = 3:7, 20 mM, pH = 7.4). Excitation at 430 nm.

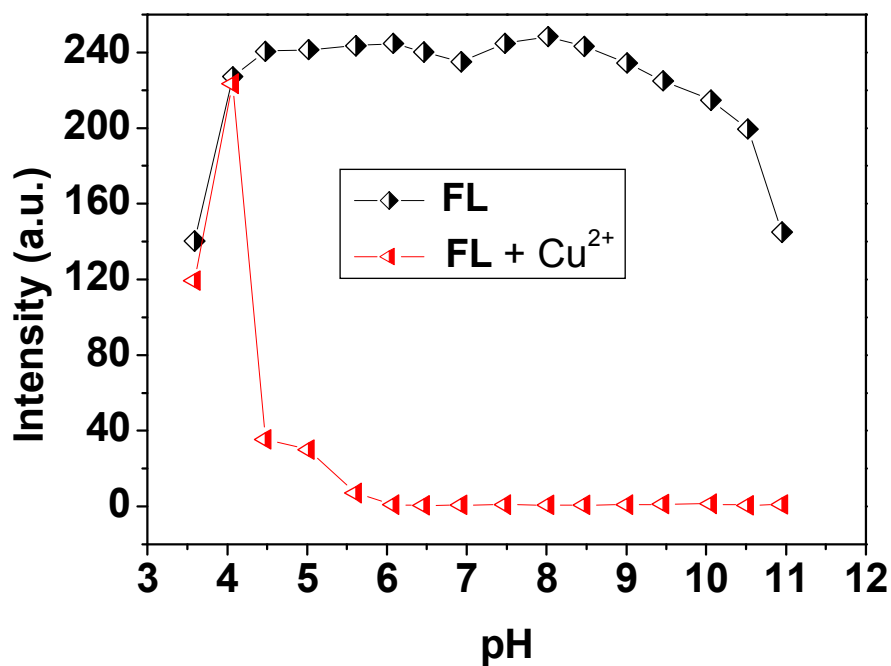


Fig. S5 Variations of fluorescence intensity at 519 nm of NL (10 μM) in aqueous solution with (bottom) and without (up) Cu^{2+} (0–20 μM) as a function of pH. Excitation at 430 nm.

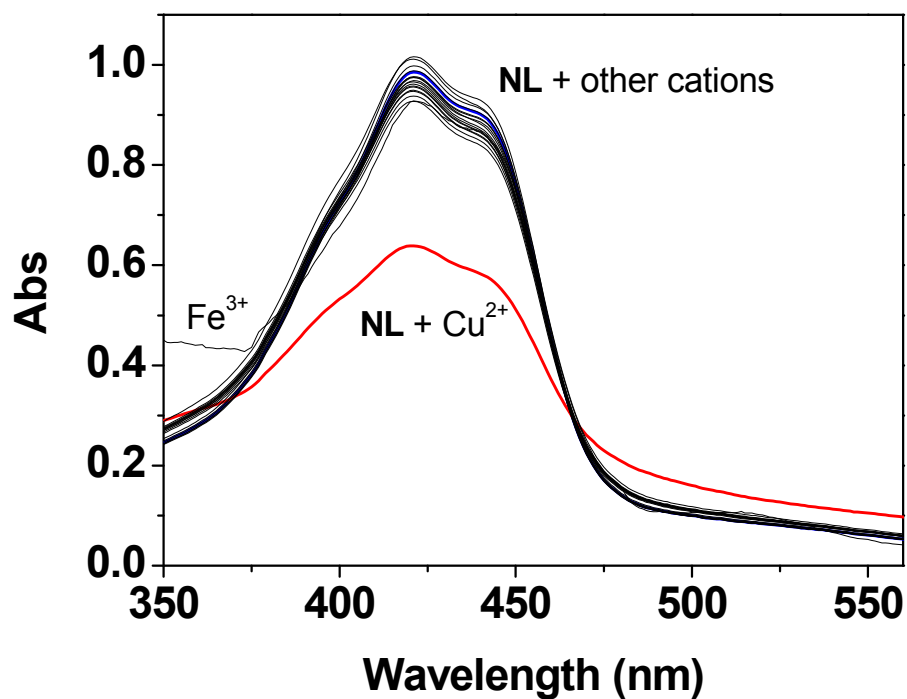


Fig. S6 Absorption spectra of NL (10 μM) in HEPES aqueous buffer (THF: H_2O = 3:7, 20 mM, pH = 7.4) upon addition of various metal ions (20 μM).

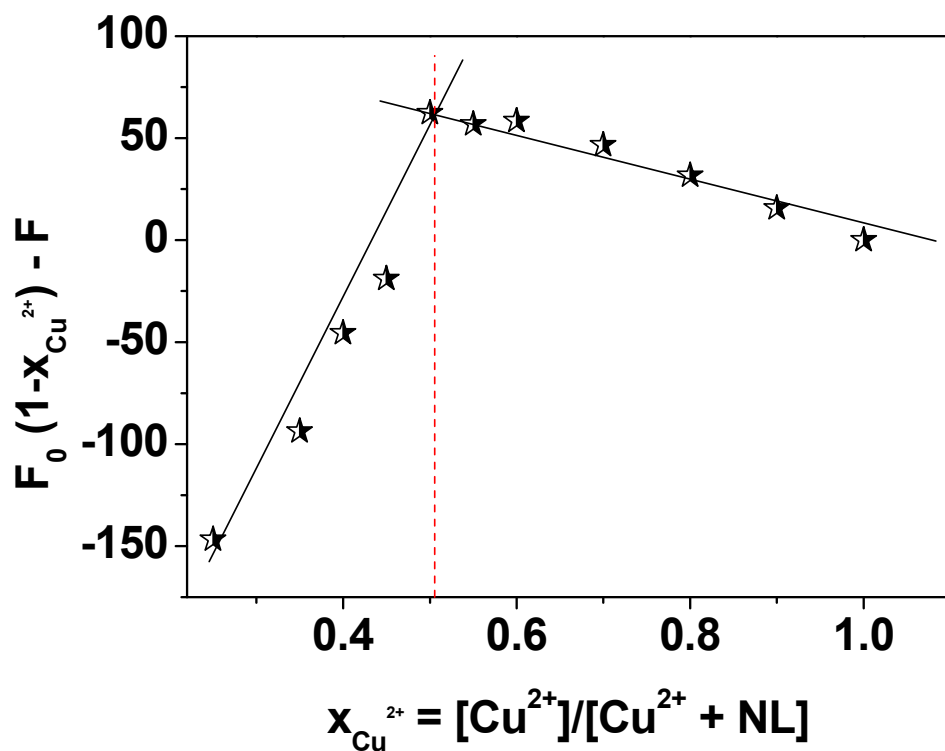


Fig. S7. The Job's plot of NL toward Cu^{2+} .

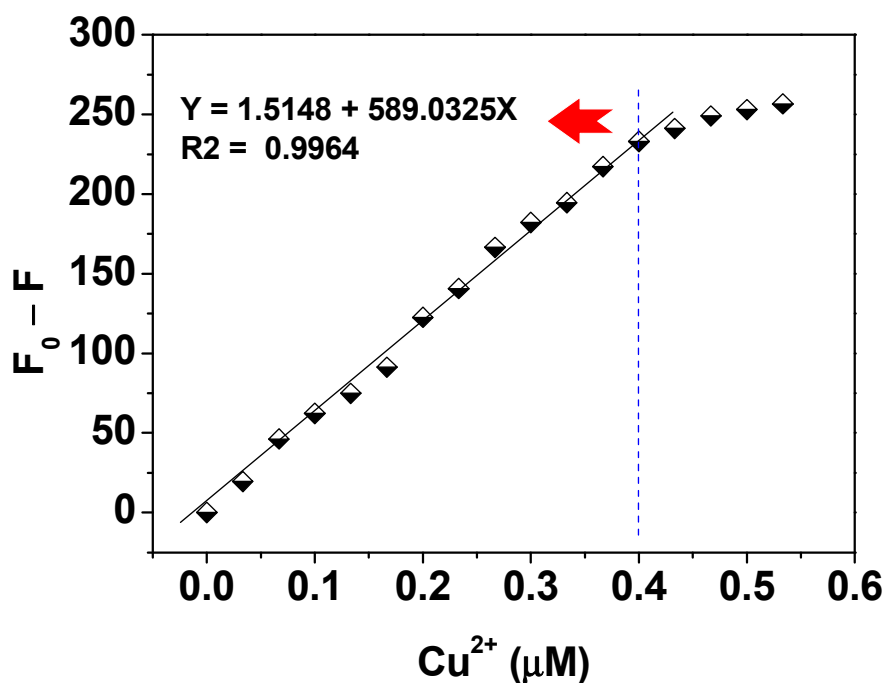


Fig. S8 Linear relationship between fluorescence intensity of NL (1 μM) at 519 nm versus the concentration of Cu^{2+} in HEPES buffer (THF: $H_2O = 3:7$, 20 mM, pH = 7.4). Excitation at 430 nm.

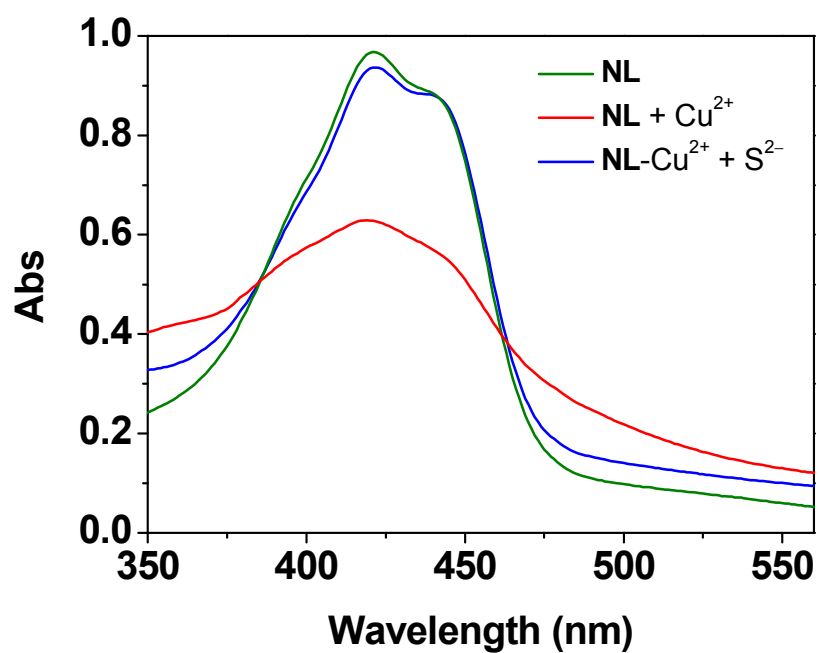


Fig. S9 UV-vis spectra of NL (10 μM) sequential in the presence of Cu^{2+} (20 μM) and S^{2-} (20 μM) in HEPES aqueous buffer (THF: H_2O = 3:7, 20 mM, pH = 7.4).

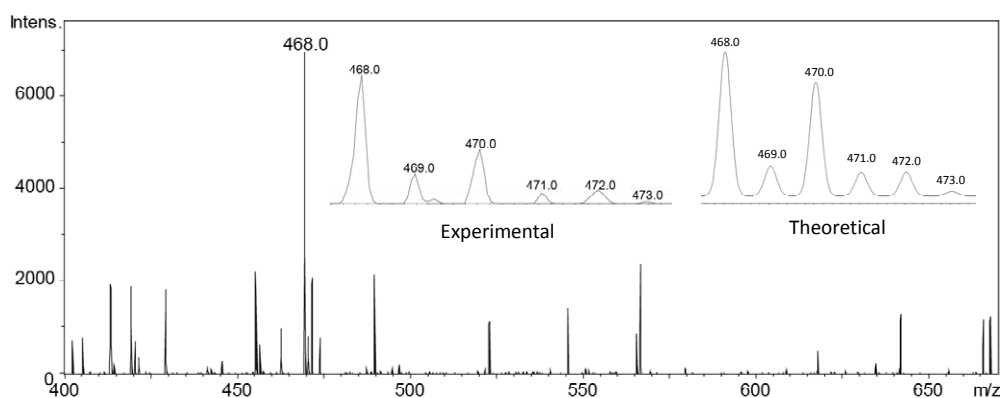


Figure S10 ESI-mass spectra of NL-Cu^{2+} ensemble

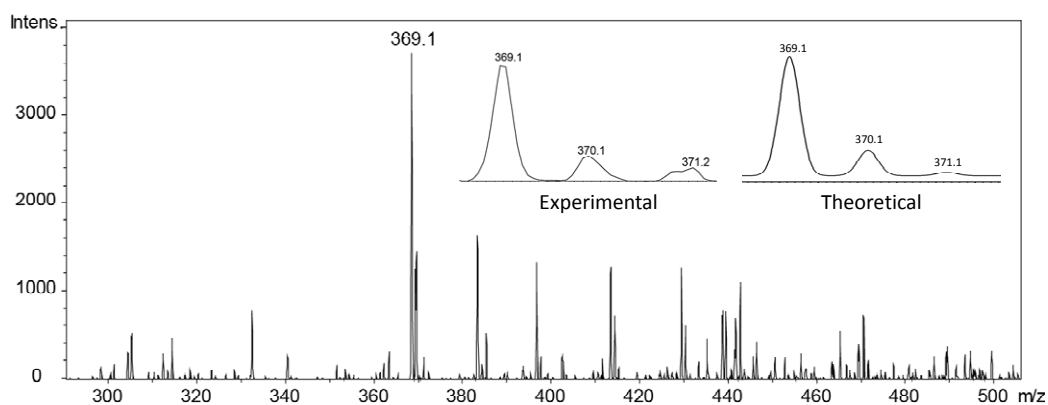


Figure S11 ESI-mass spectra of NL-Cu^{2+} ensemble in the presence of S^{2-} .

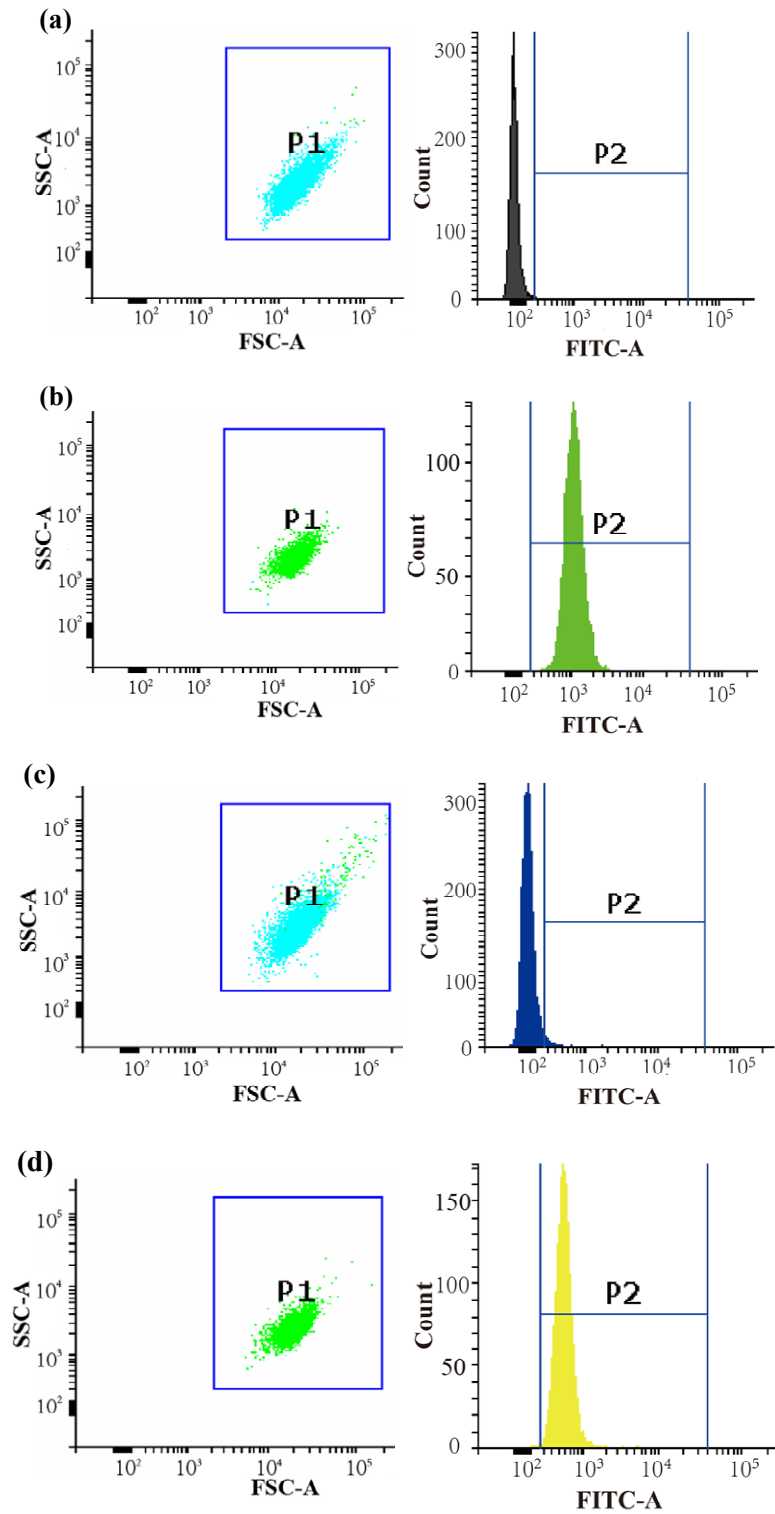


Fig. S12 Flow cytometric analysis of MDA-MB-231 cells stained with NL and its fluorescent response to Cu^{2+} and S^{2-} . (a) Control group: MDA-MB-231 cells only, (b) cells stained with 10 μ M NL for 20 min, (c) NL stained cells treated with 20 μ M Cu^{2+} for 15 min, (d) then incubated with 40 μ M S^{2-} for another 30 min.