

Supporting Information for:

Synthesis of directly fused porphyrin dimers through Fe(OTf)₃-mediated oxidative coupling

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1. Instrumentations and Materials

All NMR solvents were used as received. Chemical shifts of NMR spectra were reported in ppm down field from internal Me₄Si. UV-Vis-NIR spectra were recorded on spectrophotometer (Shimadzu, Japan). High-resolution mass spectra (HRMS) were recorded on a VG ZAB-HS mass spectrometer under electron spray ionization (ESI) and a Bruker ultra fleXtreme MALDI-TOF/TOF spectrometer. EPR spectra were collected with a Bruker A300 spectrometer equipped with a liquid helium cryostat. The electrochemical experiments were carried out by cyclic voltammetry (CV) using a BAS-100B electrochemical apparatus in deaerated CH₂Cl₂ under argon atmosphere at 298 K. And *n*-Bu₄NPF₆ (0.1 M) in CH₂Cl₂ was employed as the supporting electrolyte. A standard three-electrode cell consists of a glassy carbon disk as work electrode, a platinum wire as a counter electrode, and 0.1 M AgNO₃/Ag (in 0.1 M *n*-Bu₄NPF₆ in acetonitrile) as reference electrode. The ferrocenium/ferrocene redox couple (Fc⁺/Fc) was taken as the internal standard. All of the solvents were purified and distilled according to the standard procedure. The commercially obtained materials were used directly without further purification unless otherwise noted. Fe(OTf)₃ (98%) was purchased from Aldrich.

2. The Procedure for the Synthesis of Singly Linked Dimers 4d, 4e, 4g, and 5.

General method for the *meso-meso* linked dimers 4

Under the nitrogen atmosphere, to a stirred solution of porphyrin monomer (0.06 mmol) in dry CH₂Cl₂ (50 mL) was added dropwise a solution of PIFA (13 mg, 0.03 mmol, 0.5 equiv) in dry CH₂Cl₂ (10 mL) over 20 min at RT. The reaction mixture was stirred for additional 50 min at the same temperature. Et₃N (0.2 mL) was added to the reaction mixture. The reaction mixture was poured into water and extracted with CHCl₃, then washed with saturated sodium bicarbonate aqueous solution twice. The combined extract was dried over anhydrous Na₂SO₄ and then evaporated to dryness. The crude residue was purified by column chromatography (silica-gel). Compound 4 was obtained after recrystallization from CHCl₃/CH₃OH.

4d: ¹H NMR (400 MHz, CDCl₃, RT): δ = 8.83 – 8.79 (m, 8 H), 8.60 (d, J = 5.0 Hz, 4H), 8.12 – 8.08 (m, 8H), 7.88 (d, J = 1.6 Hz, 8H), 7.73 – 7.68 (m, 6H), 7.65 (s, 4H), 1.40 (s, 72H). ¹³C NMR (100 MHz, CDCl₃, RT): δ = 148.87, 146.71, 143.44, 142.55, 142.51, 141.17, 139.88, 133.88, 133.67, 132.54, 132.34, 132.04, 128.78, 127.72, 126.86, 121.10, 121.07, 119.45, 115.66, 34.94, 31.61.

HRMS (MALDI) m/z [M]⁺ calcd. for C₁₀₈H₁₁₀N₈Ni₂: 1636.7519, found 1636.7565; Ultraviolet-visible absorption : (in CHCl₃, λ_{max} /nm, ϵ /mol⁻¹dm³cm⁻¹): 419 (1.89 × 10⁵), 449 (2.16 × 10⁵), 539 (5.22 × 10⁴).

4e: ¹H NMR (400 MHz, CDCl₃, RT): δ = 9.58 (d, J = 5.0 Hz, 4 H), 8.85 (d, J = 5.0 Hz, 4 H), 8.53 (d, J = 5.0 Hz, 4 H), 8.02 (d, J = 5.0 Hz, 4 H), 7.83 (d, J = 1.6 Hz, 8 H), 7.66 (s, 4 H), 1.41 (s, 72 H). ¹³C NMR (100 MHz, CDCl₃, RT): δ = 149.00, 146.84, 143.95, 142.72, 142.26, 139.41, 134.08, 133.54, 133.10, 133.00, 128.71, 121.62, 121.30, 115.83, 102.97, 34.94, 31.60.

HRMS (MALDI) m/z [M]⁺ calcd. for C₉₆H₁₀₀Br₂N₈Ni₂: 1643.5158, found 1643.5175; Ultraviolet-visible absorption : (in CHCl₃, λ_{max} /nm, ϵ /mol⁻¹dm³cm⁻¹): 421 (1.64 × 10⁵), 449 (1.84 × 10⁵), 539 (4.20 × 10⁴).

4g: ¹H NMR (400 MHz, CDCl₃, RT): δ = 8.90 – 8.86 (m, 8 H), 8.61 (d, J = 5.0 Hz, 4H), 8.25 (d, J = 5.5 Hz, 4H), 8.10 – 8.00 (m, 12H), 7.82 – 7.76 (m, 6H), 7.69 (s, 4H), 1.43 (s, 72H). ¹³C NMR (100 MHz, CDCl₃, RT): δ = 148.71, 145.80, 142.30, 142.00, 141.42, 141.41, 140.66, 134.06, 132.57, 131.49, 131.21, 130.92, 129.30, 127.78, 126.74, 123.92, 122.23, 121.05, 118.94, 34.99, 31.66.

HRMS (MALDI) m/z [M]⁺ calcd. for C₁₀₈H₁₁₀N₈Pd₂: 1732.6929, found 1732.6968; Ultraviolet-visible absorption : (in CHCl₃, λ_{max} /nm, ϵ /mol⁻¹dm³cm⁻¹): 423 (3.22 × 10⁵), 447 (4.81 × 10⁵), 539 (1.31 × 10⁵).

For the *meso-β* singly linked 5

Under the nitrogen atmosphere, to a stirred solution of **1e** (49 mg, 0.06 mmol) and BF₃·Et₂O (4 mg, 0.03 mmol, 0.5 equiv) in dry CH₂Cl₂ (50 mL) was added dropwise a solution of PIFA (13 mg, 0.03 mmol, 0.5 equiv) in dry CH₂Cl₂ (10 mL) over 20 min at RT. The reaction mixture was stirred for additional 10 min at the same temperature. Et₃N (0.2 mL) was added to the reaction mixture. The reaction mixture was poured into water and extracted with CHCl₃, then washed with saturated sodium bicarbonate aqueous solution twice. The combined extract was dried over anhydrous Na₂SO₄ and then evaporated to dryness. The crude residue was purified by column chromatography (silica-gel). **5** (39 mg, 80%) was obtained after recrystallization from CHCl₃/CH₃OH.

¹H NMR (400 MHz, CDCl₃, RT): δ = 9.66 (d, J = 5.0 Hz, 1H), 9.63 (s, 1H), 9.60 – 9.56 (m, 3H), 9.07 – 9.04 (m, 2H), 8.91 – 8.86 (m, 3H), 8.76 (d, J = 5.0 Hz, 2H), 8.66 (bs, 3H), 8.46 (d, J = 4.8 Hz, 1H), 8.10 (d, J = 1.7 Hz, 2H), 7.80 (d, J = 1.7 Hz, 2H), 7.71 – 7.68 (m, 4H), 1.48 – 1.40 (m, 72H). ¹H NMR (400 MHz, CDCl₃, -40°C): δ = 9.66 (s, 1H), 9.68 (d, J = 5.0 Hz, 1H), 9.62 – 9.58 (m, 3H), 9.14 (d, J = 5.0 Hz, 1H), 9.01 (s, 1H), 8.96 (d, J = 5.0 Hz, 2H), 8.92 (d, J = 5.0 Hz, 1H), 8.80 (d, J = 5.0 Hz, 2H), 8.73 – 8.67 (m, 3H), 8.47 (d, J = 4.8 Hz, 1H), 8.17 – 8.12 (m, 4H), 7.80 (d, J = 1.8 Hz, 2H), 7.71 (bs, 1H), 7.69 (s, 3H), 7.59 (s, 2H), 1.51 – 1.46 (m, 32H), 1.44 (s, 18H), 1.40 (s, 18H). ¹³C NMR (100 MHz, CDCl₃, RT): δ = 149.24, 149.09, 149.04, 145.90, 144.42, 143.90, 143.85, 143.78, 143.62, 143.56, 143.34, 143.05, 142.43, 142.38, 142.24, 141.38, 139.55, 139.35, 137.88, 133.72, 133.68, 133.64, 133.20, 133.16, 133.12, 133.07, 132.52, 129.27, 128.90,

128.74, 121.46, 121.32, 121.21, 121.12, 120.54, 112.16, 106.74, 102.76, 102.66, 35.00, 34.96, 31.67, 31.62.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{96}H_{100}Br_2N_8Ni_2$: 1642.5080, found 1642.5073; Ultraviolet-visible absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 417 (2.14×10^5), 431 (1.80×10^5), 534 (3.57×10^4).

3. The General Procedure for the Synthesis of Fused Dimers 2 and 3.

Under the nitrogen atmosphere, to a stirred solution of **1** (0.03 mmol) in dry 25 mL CH_2Cl_2/CH_3NO_2 (4:1) was added 75 mg $Fe(OTf)_3$ (0.15 mmol, 5 equiv) at RT. The reaction mixture was stirred for additional 2 h. Saturated aqueous sodium bicarbonate (10 mL) was added, and the mixture was extracted with $CHCl_3$. The combined organic layer was washed with saturated sodium bicarbonate aqueous solution and dried over anhydrous Na_2SO_4 , then evaporated to dryness. The crude residue was purified by silica-gel column chromatography. Compounds **2** and **3** were obtained after recrystallization from $CHCl_3/CH_3OH$.

2a: HRMS (ESI) m/z $[M]^+$ calcd. for $C_{108}H_{106}N_8Cu_2$: 1642.7117, found 1642.7131; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 410 (5.11×10^5), 558 (3.83×10^4), 574 (4.04×10^4), 780 (0.58×10^4), 995 (1.00×10^4).

2b: HRMS (MALDI) m/z $[M]^+$ calcd. for $C_{96}H_{96}Br_2N_8Cu_2$: 1650.4835, found 1650.4868; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 416 (1.06×10^5), 563 (9.34×10^4), 885 (1.74×10^4), 997 (2.79×10^4).

2d: 1H NMR (400 MHz, $CDCl_3$, RT): δ = 7.75 – 7.70 (m, 12H), 7.61 (s, 4H), 7.59 (s, 8H), 7.56 – 7.52 (m, 10H), 1.43 (s, 72H). ^{13}C NMR (100 MHz, $CDCl_3$, RT): δ = 149.16, 146.97, 146.49, 145.86, 145.38, 140.14, 138.77, 134.74, 132.44, 131.03, 130.76, 127.77, 127.74, 127.20, 127.07, 124.55, 123.76, 121.19, 113.26, 34.93, 31.63.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{108}H_{106}N_8Ni_2$: 1632.7206, found 1632.7218; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 406 (7.81×10^4), 573 (8.70×10^4), 647 (1.09×10^4), 873 (1.34×10^4), 932 (1.37×10^4).

2e: 1H NMR (400 MHz, $V(CDCl_3)/V(CS_2)$ = 4:1, RT): δ = 8.52 (d, J = 4.9 Hz, 4H), 7.77 (d, J = 4.9 Hz, 4H), 7.60 (s, 4H), 7.54 (s, 8H), 7.47 (s, 4H), 1.43 (s, 72H). ^{13}C NMR (100 MHz, $V(CDCl_3)/V(CS_2)$ = 4:1, RT): δ = 149.15, 147.45, 146.53, 145.52, 145.47, 138.37, 134.76, 132.04, 131.70, 127.65, 127.49, 125.01, 121.40, 113.31, 107.89, 34.81, 31.59.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{96}H_{96}Br_2N_8Ni_2$: 1638.4767, found 1638.4787; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 411 (1.58×10^5), 572 (1.23×10^5), 651 (1.37×10^4), 864 (1.72×10^4), 937 (2.10×10^4).

2f: 1H NMR (400 MHz, $V(CDCl_3)/V(CS_2)$ = 4:1, RT): δ = 7.83 – 7.79 (m, 4H), 7.79 – 7.76 (m, 4H), 7.74 – 7.72 (m, 4H), 7.71 – 7.69 (m, 8H), 7.65 – 7.61 (m, 8H), 7.59 – 7.55 (m, 6H), 1.46 (s, 72H). ^{13}C NMR (100 MHz, $V(CDCl_3)/V(CS_2)$ = 4:1, RT): δ = 148.91, 144.77, 144.51, 143.86, 143.45, 140.49, 139.30, 134.01, 132.70, 130.21, 130.13, 128.19, 127.71, 127.11, 126.36, 125.55, 121.15, 116.02, 110.88, 34.82, 31.63.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{108}H_{106}N_8Pd_2$: 1728.6616, found 1728.6632; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 407 (6.51×10^4), 567 (8.70×10^4), 604 (3.06×10^4), 638 (9.20×10^3), 931 (6.62×10^3).

3a: HRMS (ESI) m/z $[M]^+$ calcd. for $C_{108}H_{108}N_8Cu_2$: 1644.7273, found 1644.7291; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 418 (5.72×10^4), 543 (8.12×10^4), 717 (1.68×10^4), 777 (5.56×10^4).

3b: HRMS (MALDI) m/z $[M]^+$ calcd. for $C_{96}H_{98}Br_2N_8Cu_2$: 1648.4678, found 1648.4708; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 419 (7.62×10^4), 541 (5.43×10^4), 779 (3.48×10^4).

3c: HRMS (MALDI) m/z $[M]^+$ calcd. for $C_{96}H_{98}N_{10}Cu_2O_4$: 1582.6348, found 1582.6375; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 420 (5.54×10^4), 538 (3.92×10^4), 780 (2.53×10^4).

3d: 1H NMR (400 MHz, $CDCl_3$, RT): δ = 9.50 (d, J = 5.0 Hz, 2H), 9.06 (s, 2 H), 8.80 (d, J = 4.9 Hz, 2 H), 8.41 – 8.39 (m, 4 H), 8.36 – 8.33 (m, 4H), 7.96 – 7.94 (m, 8H), 7.84 (d, J = 1.8 Hz, 4 H), 7.76 (bs, 2 H), 7.72 (bs, 2 H), 7.76 – 7.62 (m, 6 H), 1.54 (s, 36H), 1.51 (s, 36H). ^{13}C NMR (100 MHz, $CDCl_3$, RT): δ = 149.32, 149.28, 144.30, 144.15, 143.67, 143.48, 143.22, 142.87, 142.23, 141.83, 140.29, 139.95, 139.24, 139.03, 134.56, 133.67, 133.24, 132.62, 132.13, 132.03, 131.95, 131.54, 129.00, 128.65, 127.77, 127.08, 122.57, 122.09, 121.43, 121.21, 120.86, 109.82, 35.08, 35.03, 31.74, 31.68.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{108}H_{108}N_8Ni_2$: 1634.7363, found 1634.7359; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 413 (6.45×10^4), 502 (4.82×10^4), 544 (4.87×10^4), 754 (3.52×10^4)

3e: 1H NMR (400 MHz, $CDCl_3$, RT): δ = 9.41 (s, 2H), 9.12 – 9.11 (m, 4H), 8.96 (s, 2H), 8.74 (d, J = 3.8 Hz, 2H), 8.44 (d, J = 5.0 Hz, 2H), 8.39 (d, J = 4.8 Hz, 2H), 7.91 (s, 4H), 7.84 – 7.65 (m, 8H), 1.54 (s, 36H), 1.51 (s, 36H). ^{13}C NMR (100 MHz, $CDCl_3$, RT): δ = 149.44, 149.41, 144.68, 144.61, 143.45, 143.37, 143.31, 142.84, 142.39, 142.11, 140.22, 138.83, 138.61, 134.96, 134.00, 133.70, 133.21, 132.91, 132.41, 132.27, 128.90, 128.55, 123.15, 122.55, 121.66, 121.43, 110.10, 104.24, 35.09, 35.04, 31.73, 31.68.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{96}H_{98}Br_2N_8Ni_2$: 1640.4924, found 1640.4943; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 419 (8.56×10^4), 498 (5.85×10^4), 533 (5.62×10^4), 684 (1.16×10^4), 756 (4.66×10^4).

3f: 1H NMR (400 MHz, $CDCl_3$, RT): δ = 9.38 (d, J = 4.9 Hz, 2H), 8.94 (s, 2H), 8.88 (bs, 4H), 8.73 (d, J = 4.9 Hz, 2H), 8.50 (d, J = 5.1 Hz, 2H), 8.45 (d, J = 5.1 Hz, 2H), 7.88 (d, J = 1.4 Hz, 4H), 7.77 (bs, 8H), 1.54 (s, 36H), 1.52 (s, 36H). ^{13}C NMR (100 MHz, $CDCl_3$, RT): δ = 149.70, 149.69, 144.56, 144.46, 144.19, 144.16, 143.15, 141.55, 139.75, 139.14, 139.10, 138.26, 137.98, 135.27, 135.06, 134.99, 134.96, 133.01, 129.61, 129.07, 128.84, 128.79, 128.40, 125.05, 124.18, 122.08, 121.85, 112.81, 35.11, 35.07, 31.71, 31.65.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{96}H_{98}N_{10}Ni_2O_4$: 1572.6438, found 1572.6453; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 442 (6.14×10^4), 533 (5.56×10^4), 765 (4.10×10^4)

3g: 1H NMR (400 MHz, $CDCl_3$, RT): δ = 9.74 (d, J = 5.0 Hz, 2H), 9.33 (s, 2H), 8.78 (d, J = 5.0 Hz, 2H), 8.57 (d, J = 4.9 Hz, 2H), 8.51 (d, J = 4.9 Hz, 2H), 8.48 – 8.35 (m, 4H), 8.15 (s, 4H), 8.11 – 8.05 (m, 4H), 8.01 (d, J = 1.4 Hz, 4H), 7.84 (s, 2H), 7.79 (s, 2H), 7.70 (d, J = 6.1 Hz, 6H), 1.60 (s, 36H), 1.56 (s, 36H). ^{13}C NMR (100 MHz, $CDCl_3$, RT): δ = 149.18, 149.09, 143.11, 143.00, 142.66, 142.60, 142.41, 141.37, 141.27, 141.21, 140.35, 140.30, 140.27, 140.07, 133.90, 133.61, 133.53, 133.38, 131.11, 131.09, 130.82, 130.79, 129.68, 129.52, 127.79, 126.95, 125.47, 124.98, 123.47, 121.40, 121.12, 111.80, 35.17, 35.11, 31.83, 31.74.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{108}H_{108}N_8Pd_2$: 1730.6773, found 1730.6790; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 417 (7.51×10^4), 543 (6.53×10^4), 606 (1.07×10^4), 684 (1.44×10^4), 747 (4.06×10^4).

3h: 1H NMR (400 MHz, $V(CDCl_3)/V(CS_2)$ = 4:1, RT): δ = 9.61 (d, J = 4.9 Hz, 2 H), 9.21 (s, 2 H), 8.84 (d, J = 5.1 Hz, 4 H), 8.72 (d, J = 5.0 Hz, 2 H), 8.67 (d, J = 5.1 Hz, 2 H), 8.61 (d, J = 5.1 Hz, 2 H), 8.10 (d, J = 1.3 Hz, 4 H), 7.95 (d, J = 1.4 Hz, 4 H), 7.86 (s, 2 H), 7.82 (s, 2 H), 1.63 (s, 36 H), 1.59 (s, 36 H). ^{13}C NMR (100 MHz, $VCDCl_3/VCS_2$ = 4:1, RT): δ = 149.34, 149.24, 143.62, 143.42, 143.39, 142.87, 141.08, 141.00, 140.35, 139.32, 139.13, 137.51, 137.38, 135.26, 133.75, 133.44, 133.32, 132.59, 130.32, 129.55, 129.27, 127.52, 127.43, 127.29, 126.57, 121.89, 121.66, 114.37, 34.96, 34.89, 31.69, 31.63.

HRMS (ESI) m/z $[M]^+$ calcd. for $C_{96}H_{98}N_{10}Pd_2O_4$: 1668.5878, found 1668.5880; Ultraviolet-visible-infrared absorption: (in $CHCl_3$, λ_{max}/nm , $\epsilon/mol^{-1}dm^3cm^{-1}$): 426 (1.16×10^5), 528 (1.07×10^5), 749 (1.37×10^4).

4. NMR Spectra

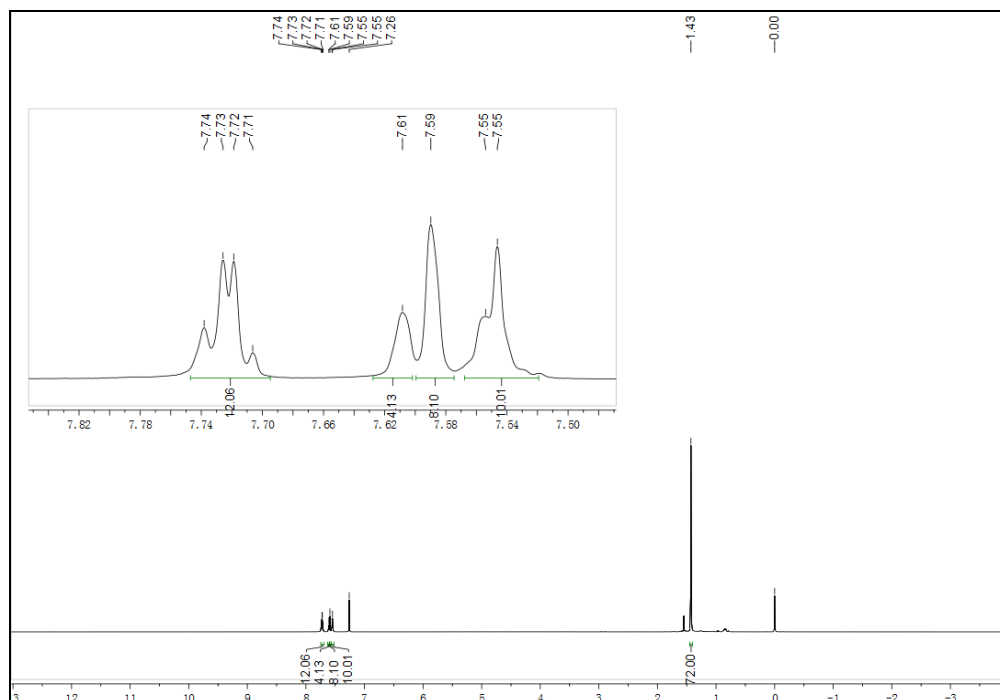


Figure S1. ¹H NMR spectrum of **2d** (400 MHz, CDCl₃ at 25°C).

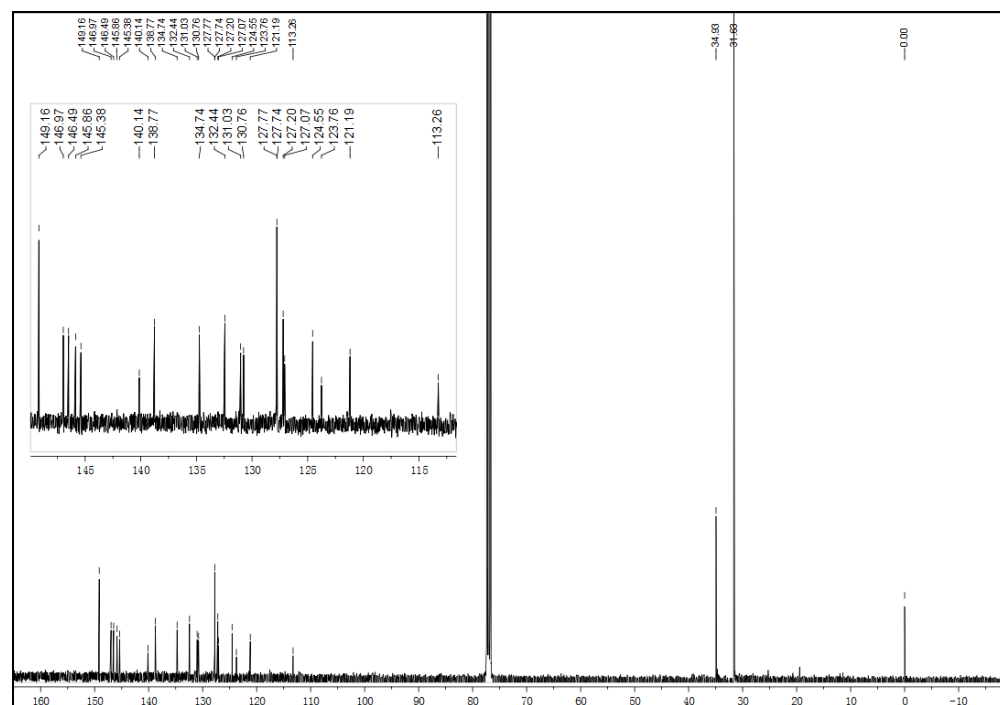


Figure S2. ¹³C NMR spectrum of **2d** (100 MHz, CDCl₃ at 25°C).

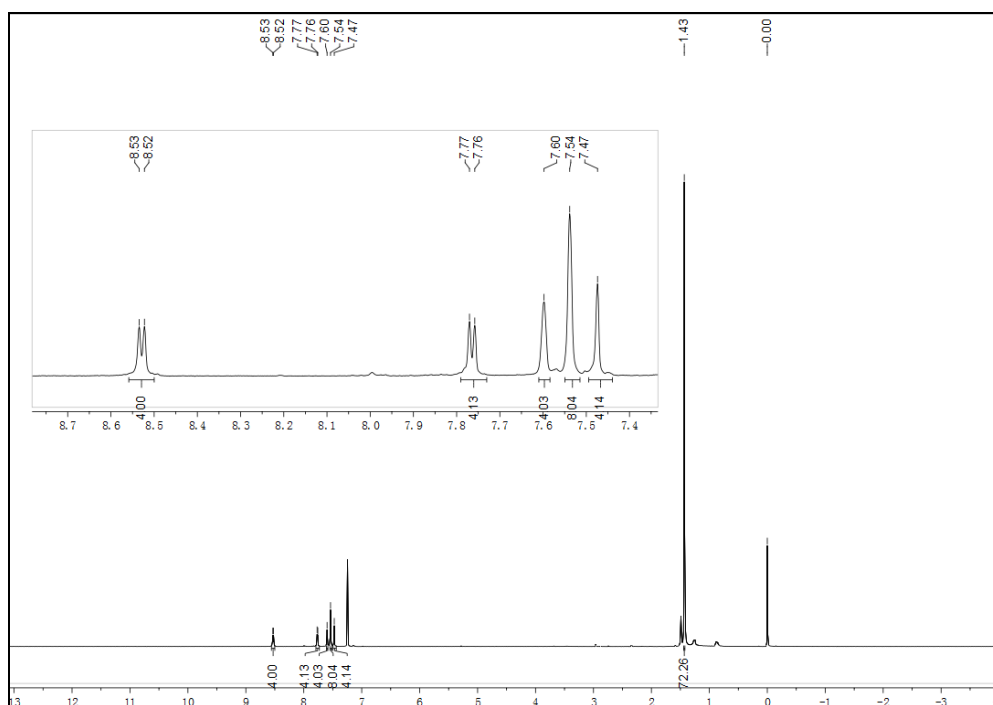


Figure S3. ¹H NMR spectrum of **2e** (400 MHz, V_{CDCl₃}/V_{CS₂} = 4:1 at 25°C).

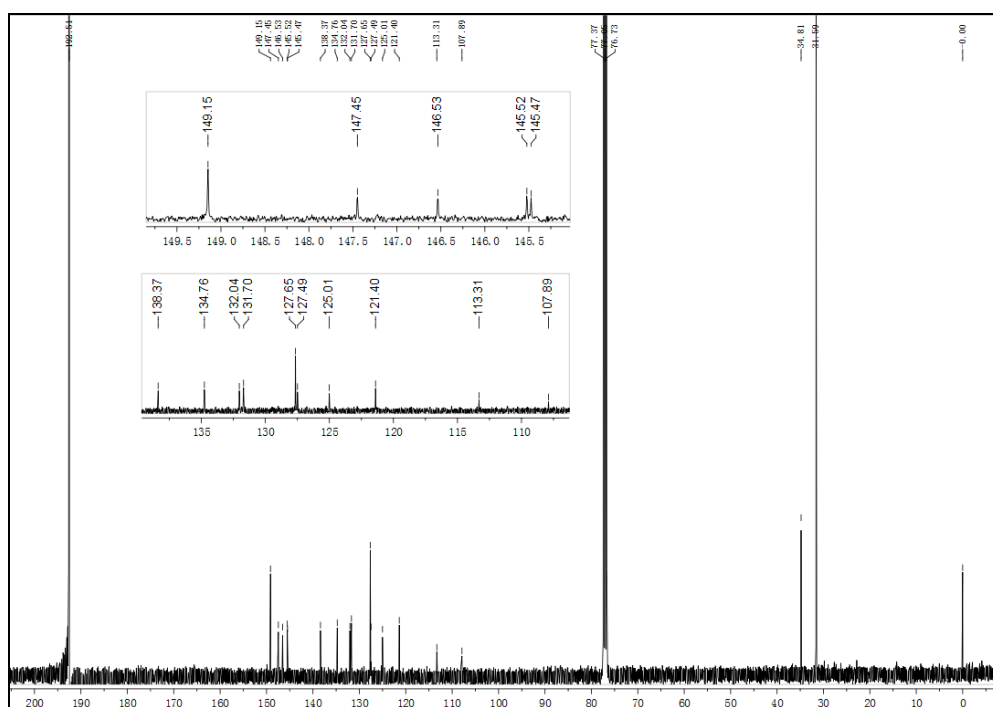
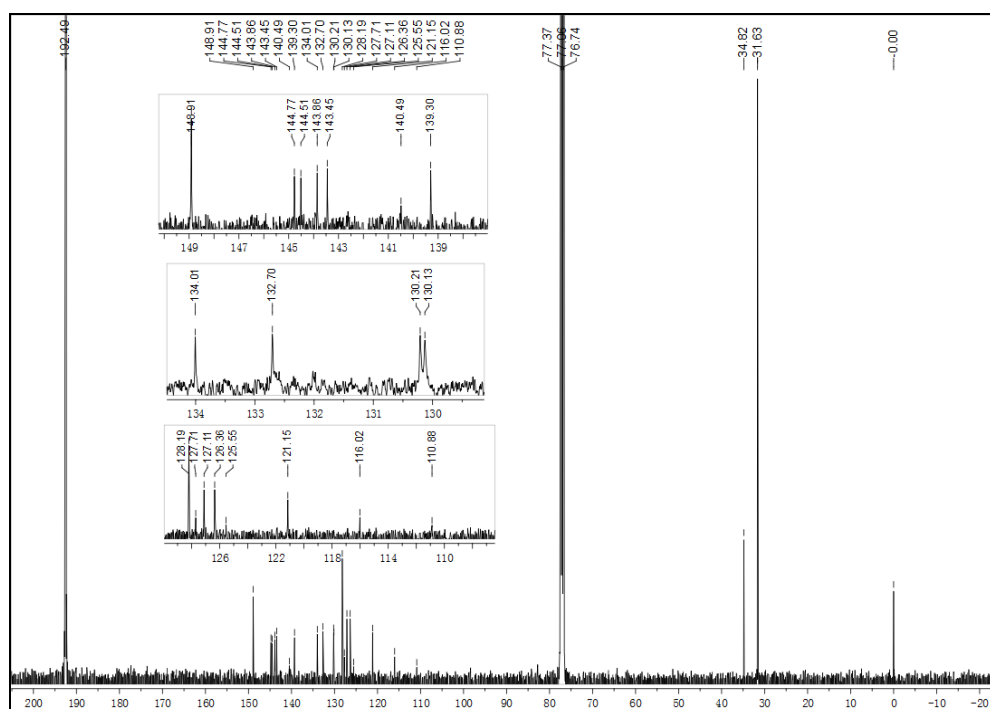
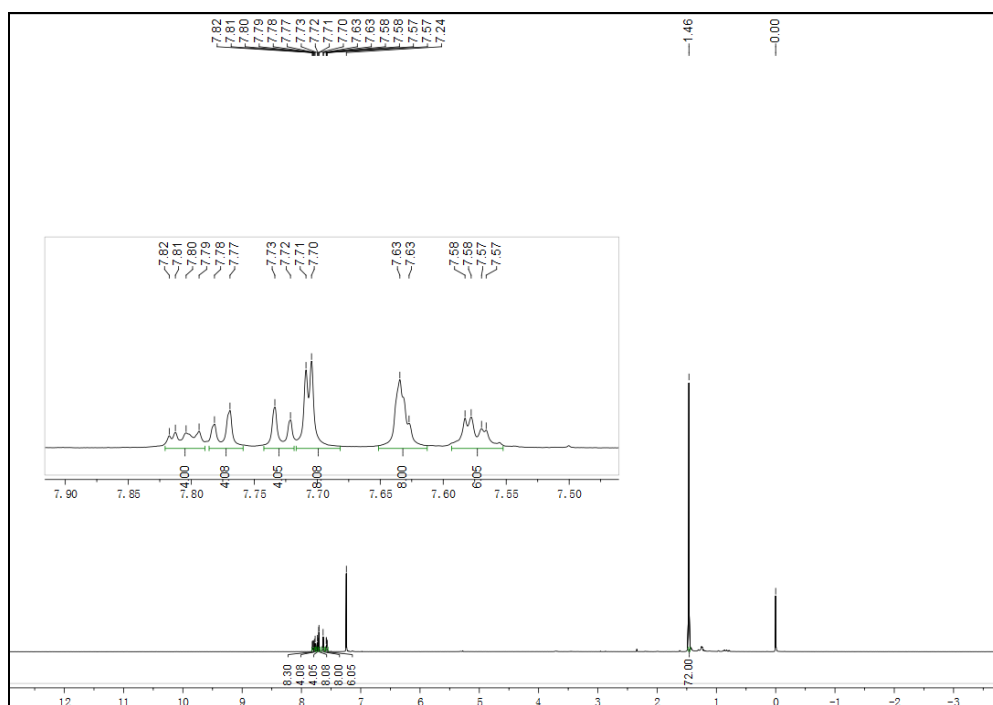
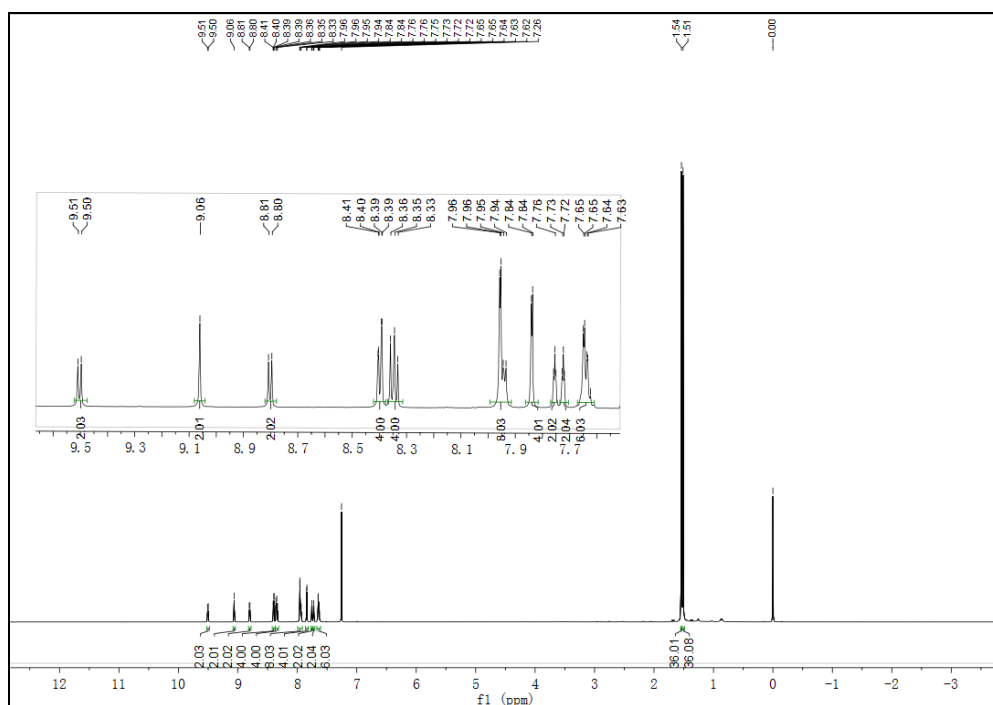


Figure S4. ¹³C NMR spectrum of **2e** (100 MHz, V_{CDCl₃}/V_{CS₂} = 4:1 at 25°C, δ (¹³C_{CS₂}) = 192.51).





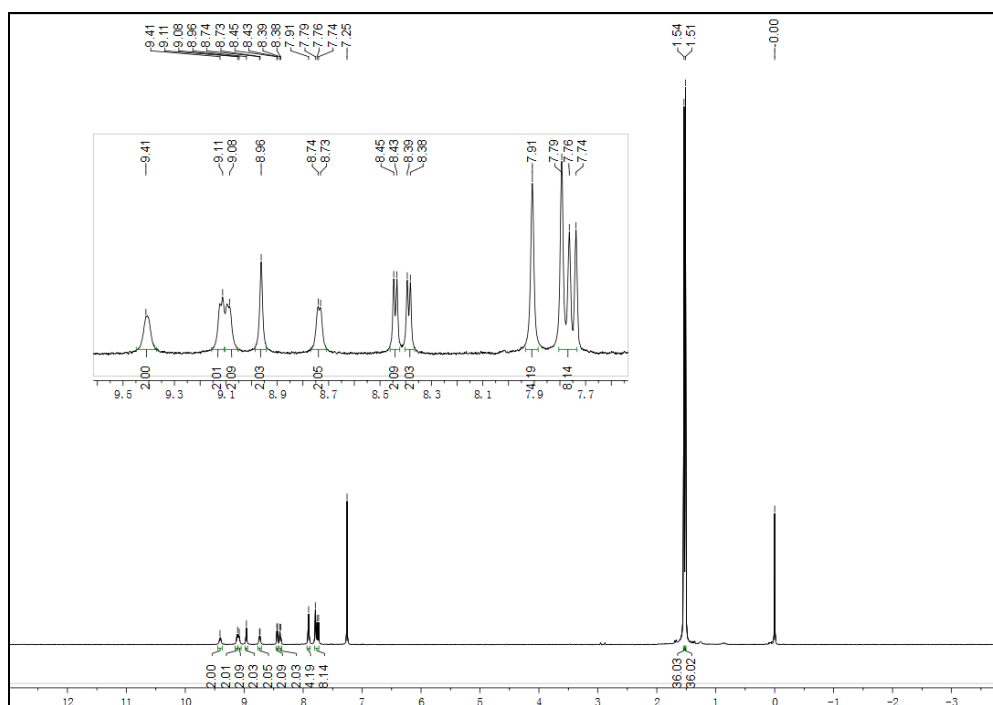


Figure S9. ¹H NMR spectrum of **3e** (400 MHz, CDCl₃ at 25°C).

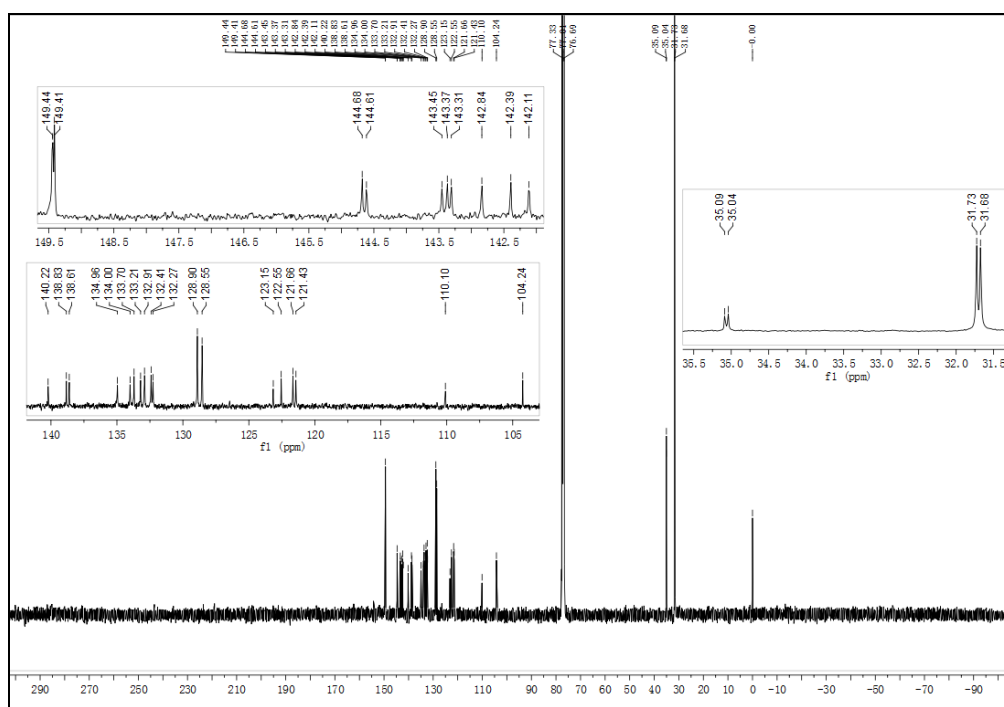


Figure S10. ¹³C NMR spectrum of **3e** (100 MHz, CDCl₃ at 25°C).

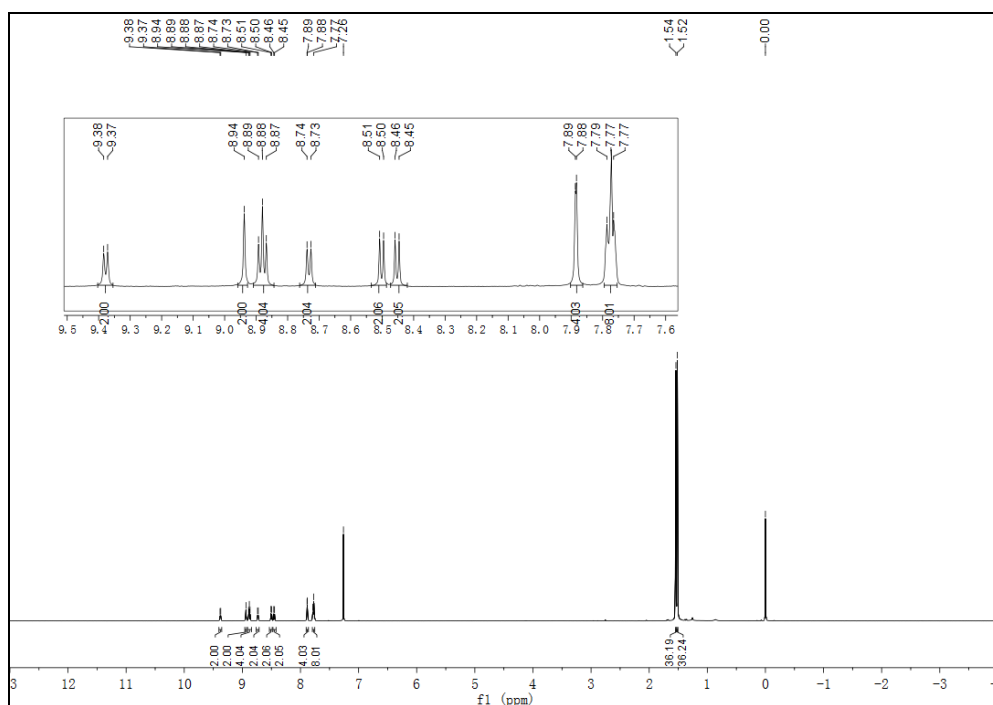


Figure S11. ¹H NMR spectrum of **3f** (400 MHz, CDCl₃ at 25°C).

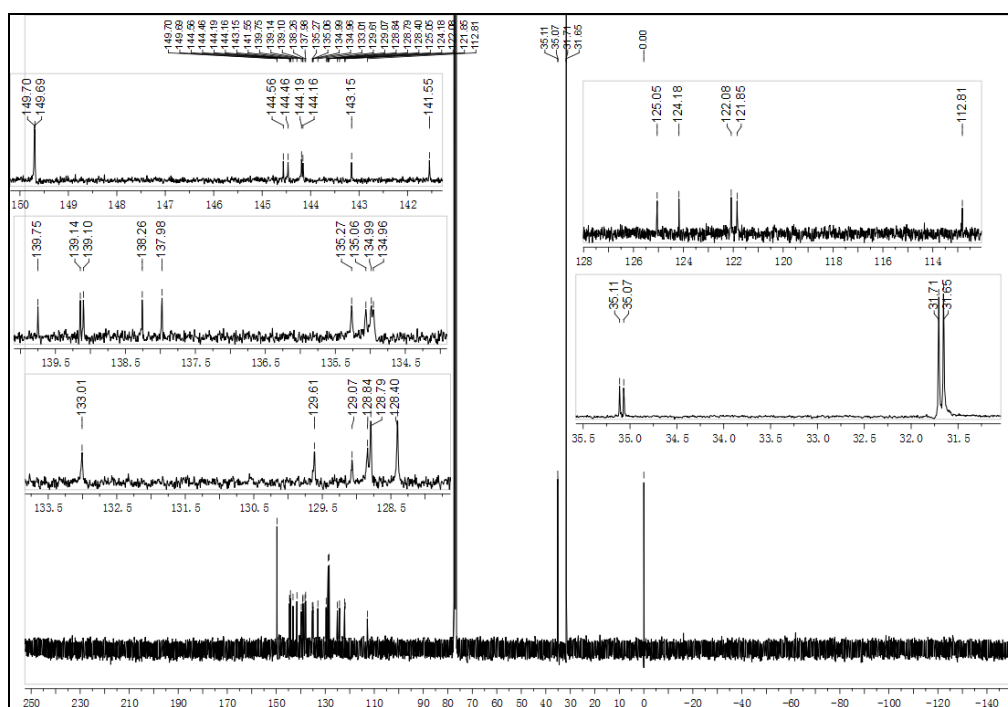
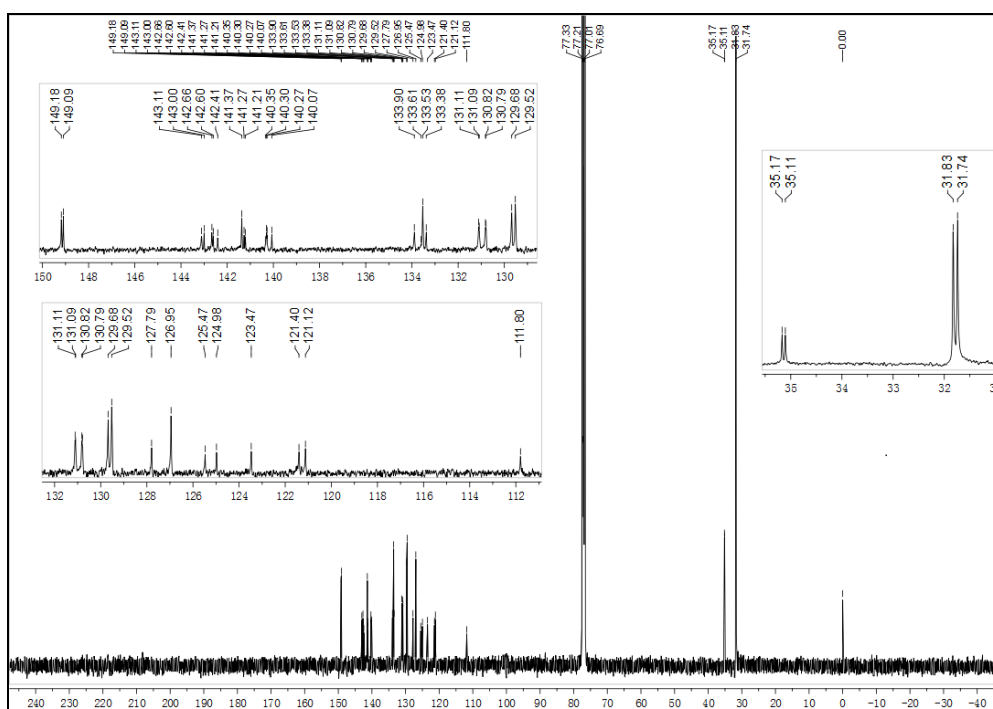
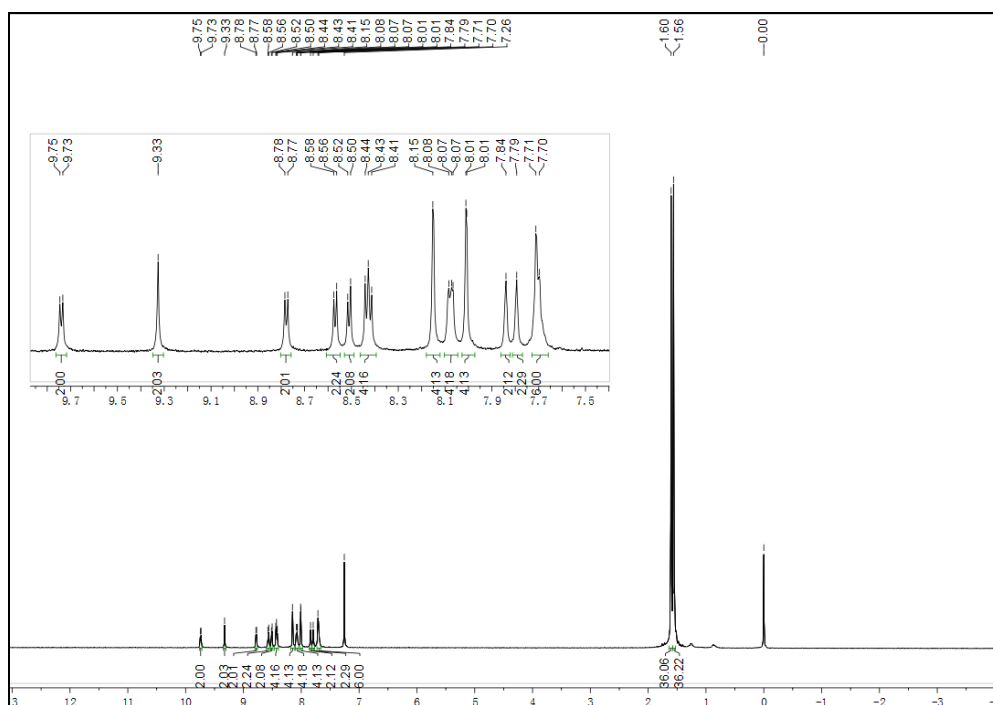


Figure S12. ¹³C NMR spectrum of **3f** (100 MHz, CDCl₃ at 25°C).



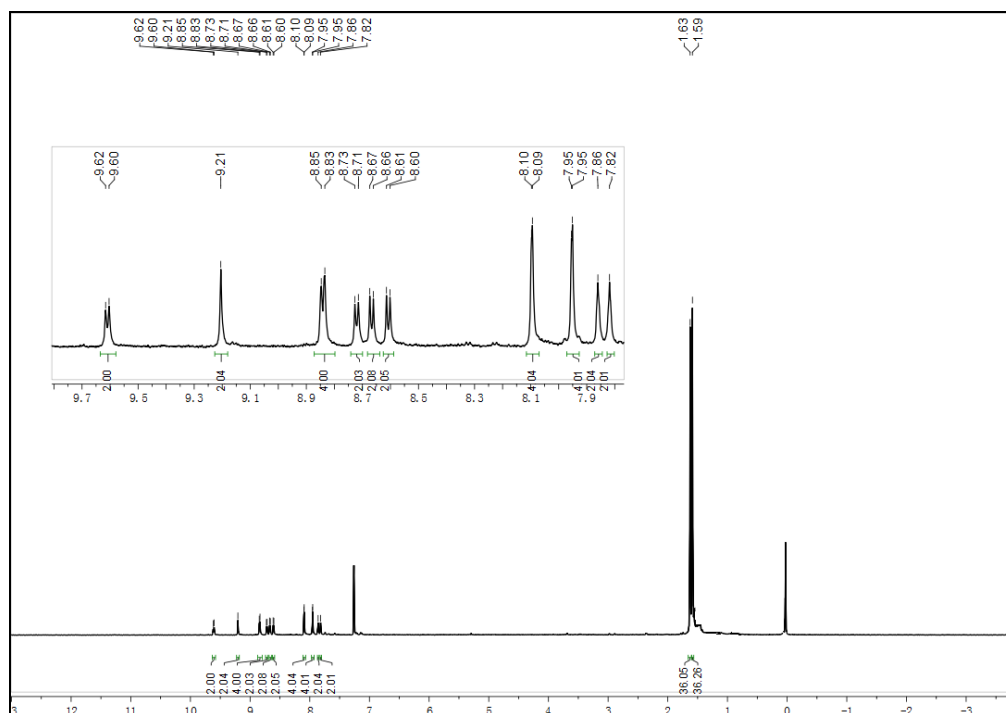


Figure S15. ¹H NMR spectrum of **3h** (400 MHz, V_{CDCl₃}/V_{CS₂} = 4:1 at 25°C).

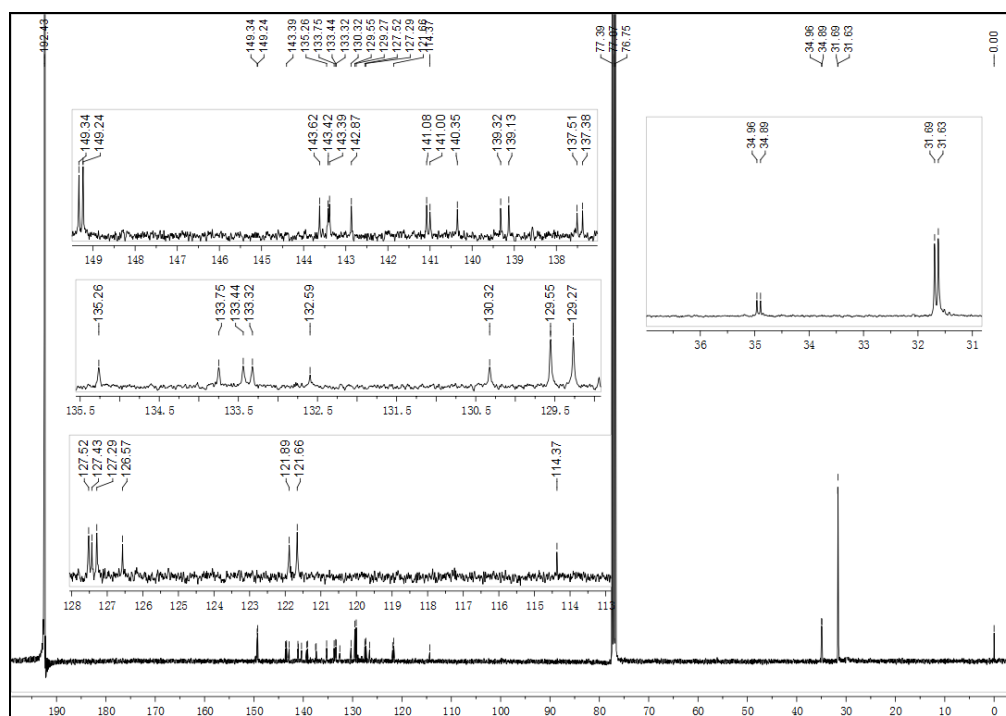


Figure S16. ¹³C NMR spectrum of **3h** (100 MHz, V_{CDCl₃}/V_{CS₂} = 4:1 at 25°C, δ (¹³C_{CS₂}) = 192.43).

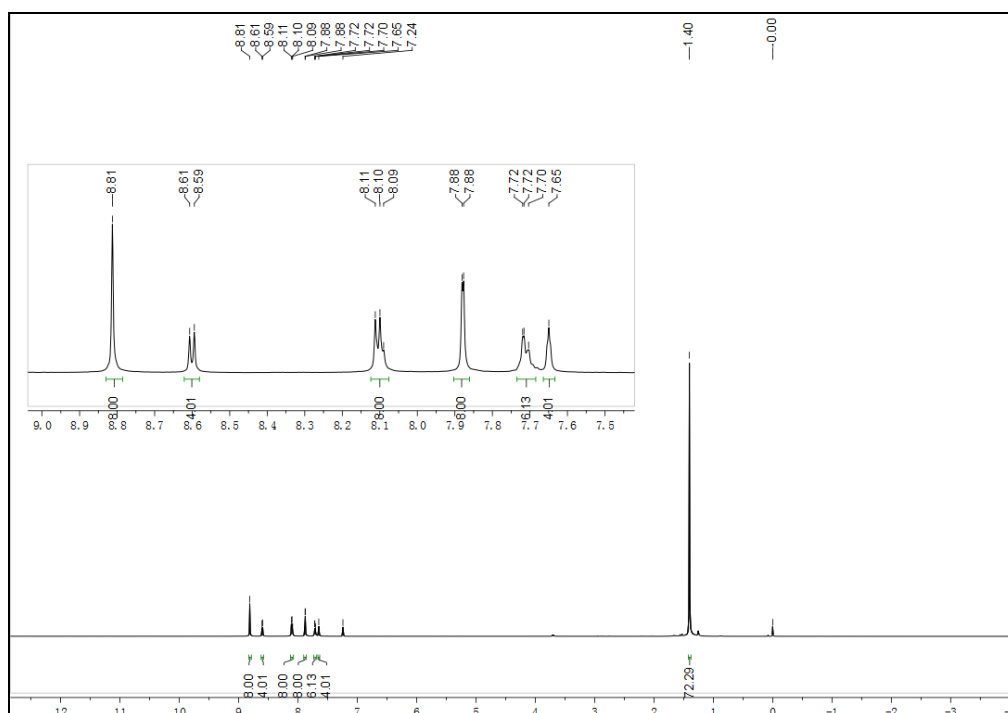


Figure S17. ¹H NMR spectrum of **4d** (400 MHz, CDCl₃ at 25°C).

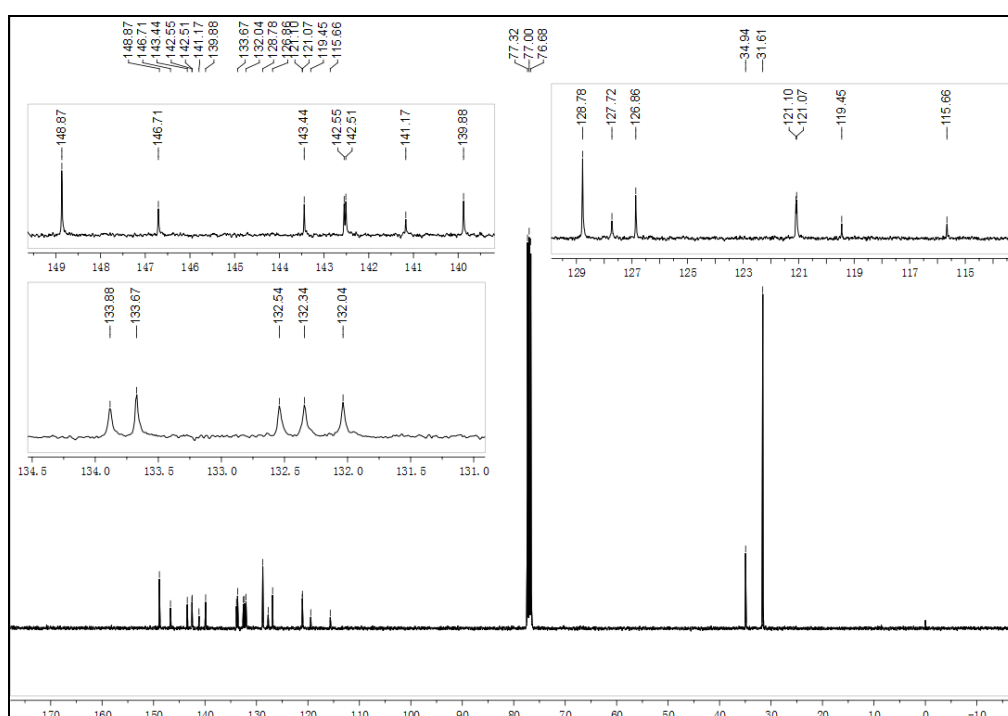


Figure S18. ¹³C NMR spectrum of **4d** (100 MHz, CDCl₃ at 25°C).

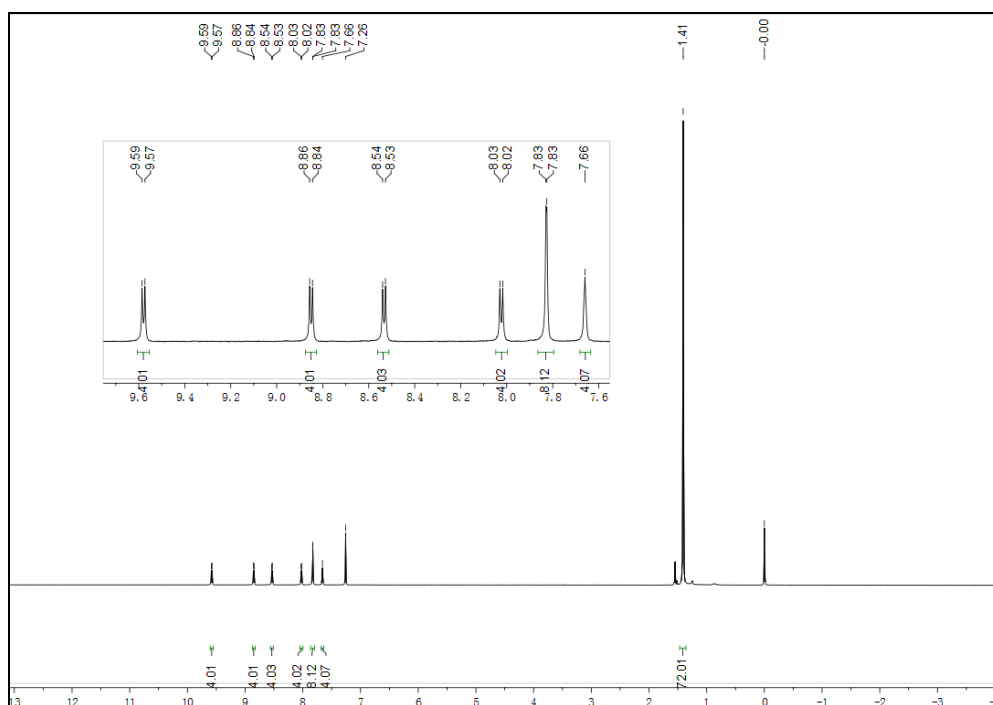


Figure S19. ¹H NMR spectrum of **4e** (400 MHz, CDCl₃ at 25°C).

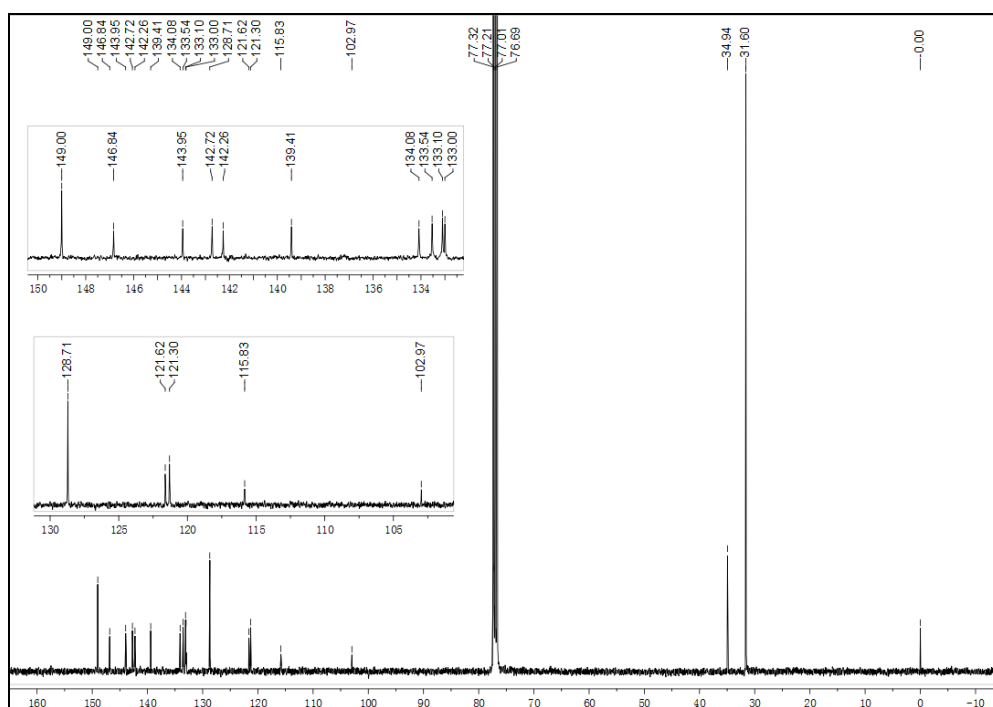


Figure S20. ¹³C NMR spectrum of **4e** (100 MHz, CDCl₃ at 25°C).

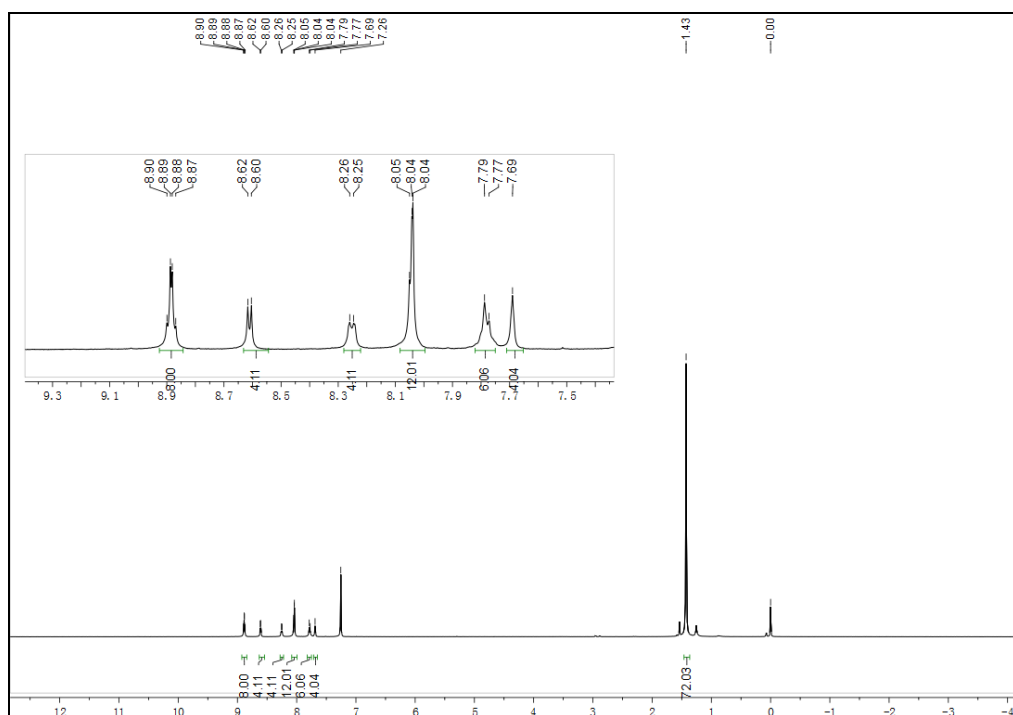


Figure S21. ¹H NMR spectrum of **4g** (100 MHz, CDCl₃ at 25°C).

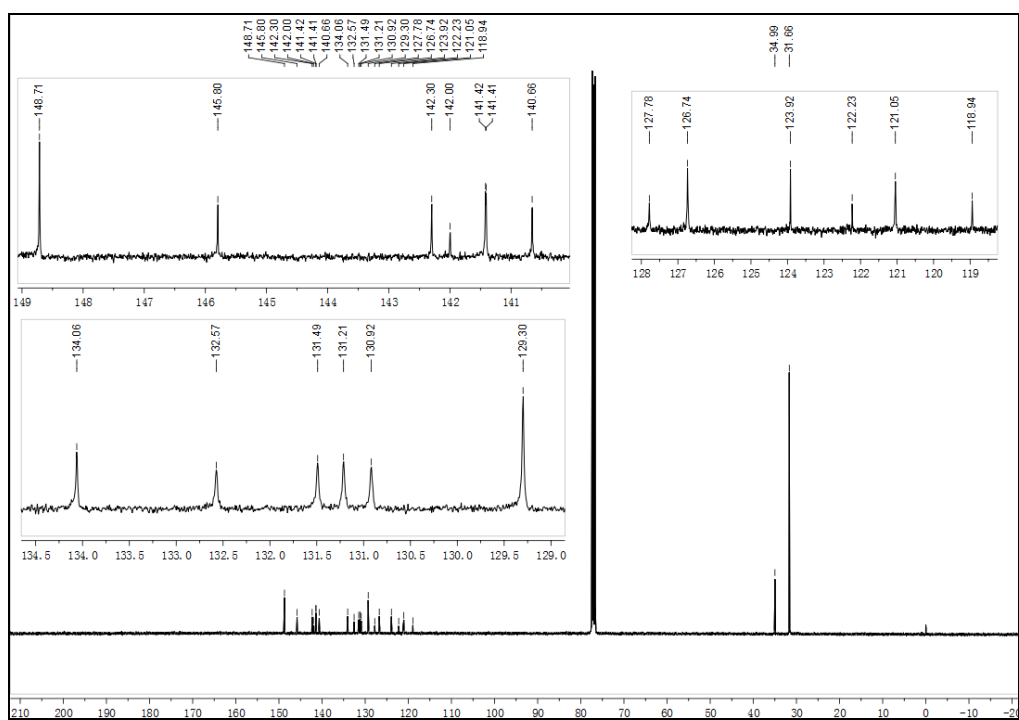
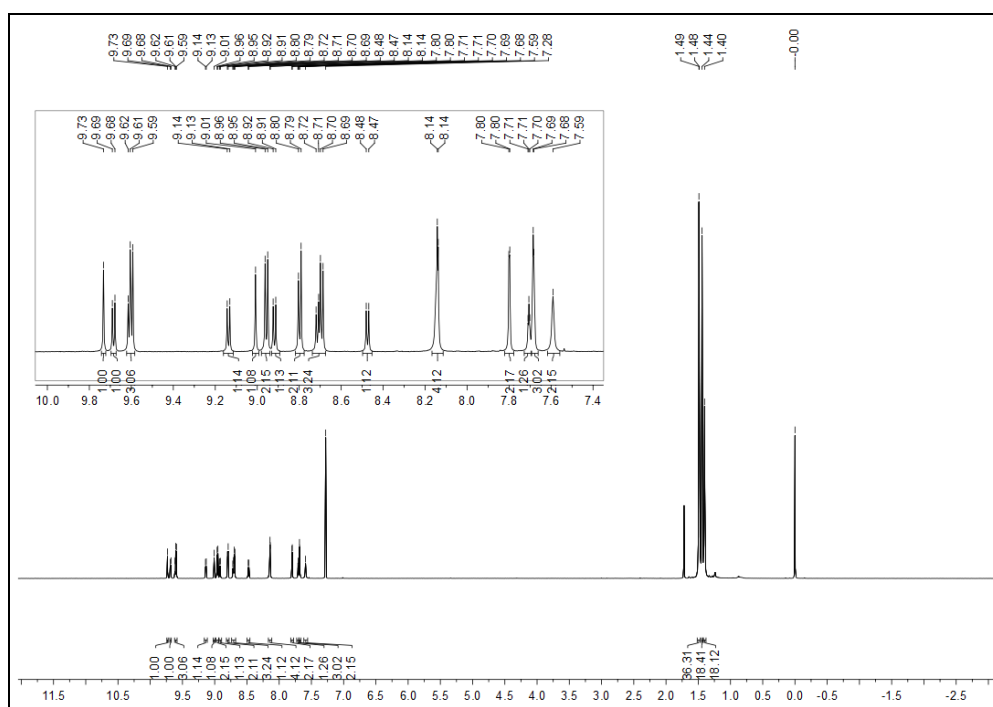
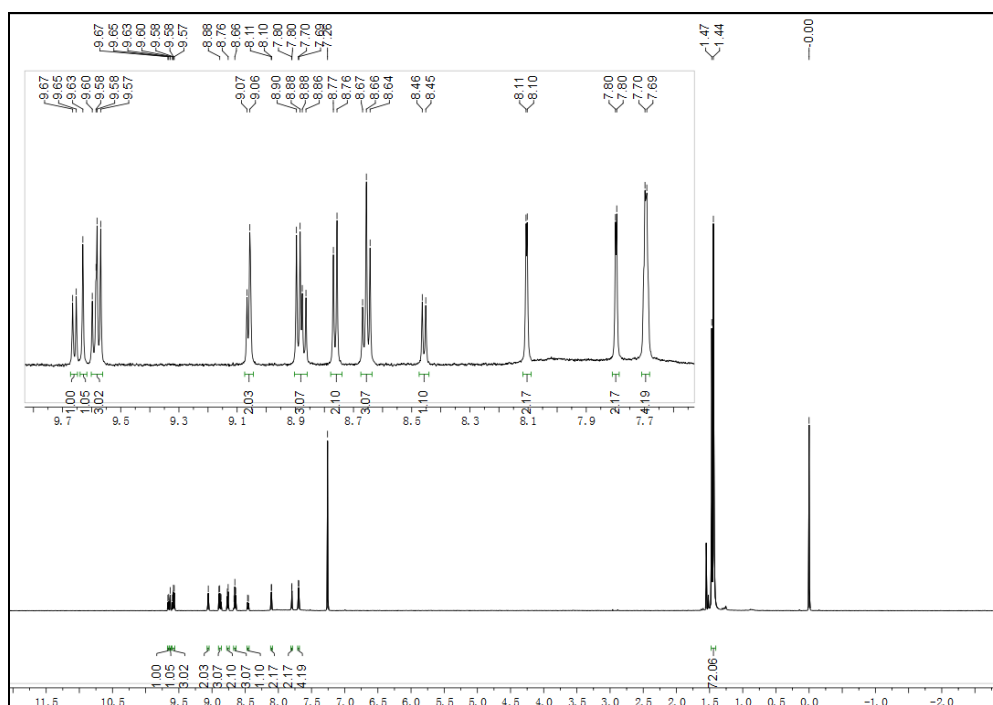


Figure S22. ¹³C NMR spectrum of **4g** (400 MHz, CDCl₃ at 25°C).



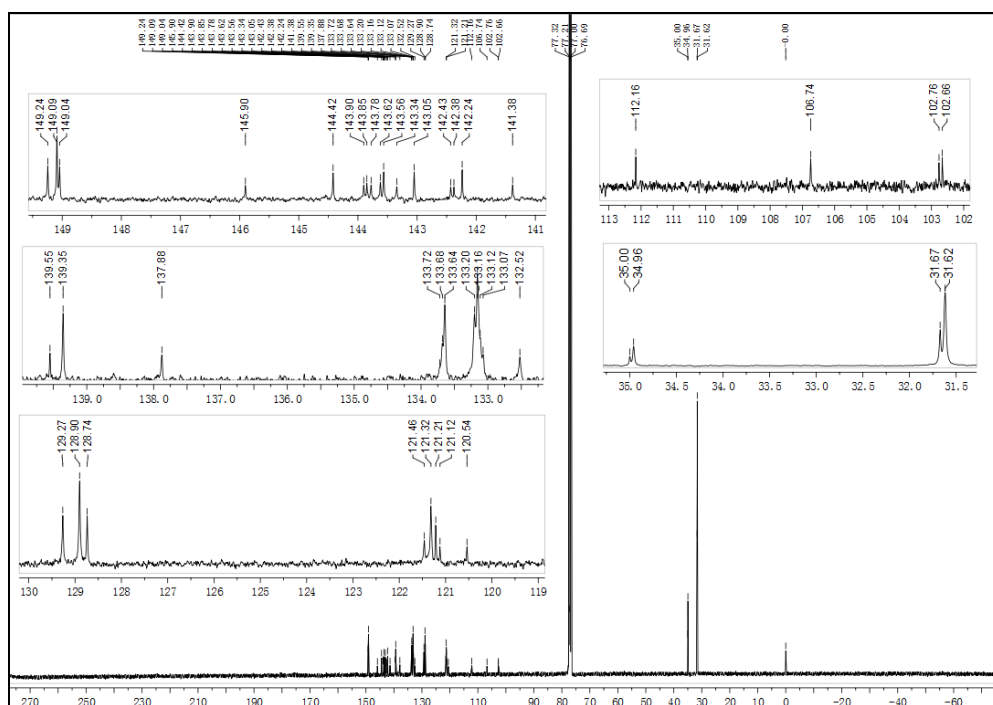


Figure S25. ^{13}C NMR spectrum of **5** (100 MHz, CDCl_3 at 25°C).

5. ESR Spectrum

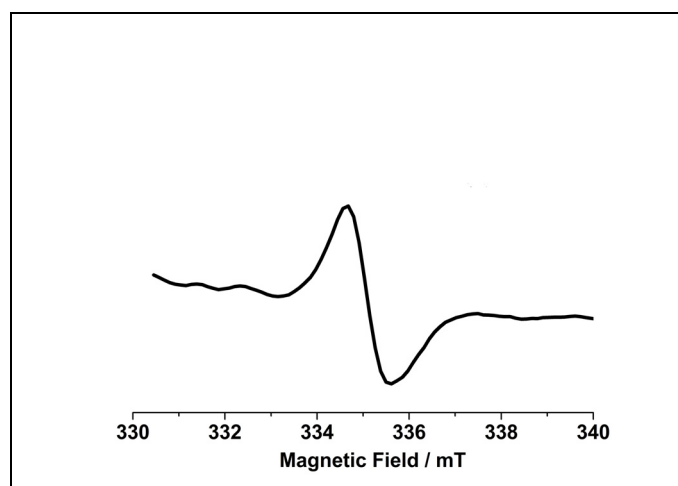


Figure S26. ESR spectrum of $\text{Fe}(\text{OTf})_3$ -mediated oxidative coupling of **1f**. ($g = 2.0082$)

6. The CV Spectra of 1a-1h

