

Supplementary Information

Structure-Alkali Metal Cation Complexation Relationships for Macrocyclic PNP-Lariat Ether Ligands

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Table S2 Combustion analysis and ¹H NMR spectral data for monoaryloxy-substituted PNP-lariat Ethers **3c-f**.

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Table S1 ^{31}P NMR spectral data for monoaryloxy-substituted PNP-lariat ethers **3c-f**.

Compound	$^{31}\text{P}\{-^1\text{H}\}$ NMR, AMX spin system						$^{31}\text{P}\{=^1\text{H}\}$ NMR		
	δ_{P_i} , ppm ^a			$J_{\text{P-P}}$, Hz ^b			$J_{\text{P-H}}$, Hz		
	P_A	P_M	P_X	A-M	A-X	M-X	P_AH	P_MH	P_XH
3c	5.90	27.60	21.10	75.0	70.4	77.3	12.3 ^c	^d	25.8 ^e
3d	6.40	27.30	21.30	73.5	68.6	80.8	12.4 ^c	^d	23.5 ^e
3e	6.46	27.48	20.64	75.8	70.7	80.6	12.5		23.6 ^e
3f	5.65	27.32	20.86	77.7	70.4	79.8	12.6 ^e		24.5 ^e

^aAMX spin system: $\text{P}_A = \text{P}(\text{OCH}_2\sim)[\text{OAr}(z)]$; $\text{P}_M = \text{PCl}_2$; $\text{P}_X = \text{P}(\text{OCH}_2\sim)\text{Cl}$. ^b $J_{\text{P-P}}$ values calculated according to ref. 27. ^cFor each peak of the double doublet split into triplets. ^dNo proton coupling was evident. ^eFor each peak of the double doublet split into quintets.

Table S2 Combustion analysis and ^1H NMR spectral data for monoaryloxy-substituted PNP-lariat ethers **3c-f**.

Formula	Elemental analysis, % ^a			^1H NMR δ_{H} , ppm ^b								
	C	H	N	(CH ₂) OP	(CH ₂) OC	(CH) _{Ar}	CHO	CH ₃	(CH ₂) OC=O	(CH ₂) C	(CH ₃) ₂ C	(CH ₃) _x C
3c C ₁₇ H ₂₆ O ₈ N ₃ P ₃ Cl ₂	<u>34.35</u> 34.11	<u>4.47</u> 4.35	<u>6.97</u> 7.02	4.67- 4.11 (m,4H)	3.85- 3.61 (3m,12H)	8.05- 6.75 (2m,4H)		1.40- 1.30 (t,3H)	4.36- 4.29 (qt,2H)	1.69- 1.66 (2s,2H)	1.31- 1.23 (2s,6H)	0.68 (s,9H)
3d C ₂₂ H ₃₈ O ₆ N ₃ P ₃ Cl ₂	<u>41.30</u> 41.38	<u>5.88</u> 5.96	<u>6.64</u> 6.58	4.45- 4.08 (m,4H)	3.85- 3.57 (2m,12H)	7.31- 6.71 (3m,4H)						
3e C ₁₅ H ₂₁ O ₇ N ₃ P ₃ Cl ₂	<u>32.55</u> 32.46	<u>3.83</u> 3.79	<u>7.48</u> 7.57	4.58- 4.15 (2m,2H)	3.91- 3.44 (m,12H)	7.95 6.93 (4m,4H)	10.50- 10.40 (S,1H)					
3f C ₁₅ H ₂₁ O ₇ N ₃ P ₃ Cl ₂	<u>32.60</u> 32.46	<u>3.88</u> 3.79	<u>7.49</u> 7.57	4.58- 4.09 (2m,4H)	4.00- 3.53 (m,12H)	8.09- 7.16 (2m,4H)	10.02 9.90 (2s,1H)					

^aCalcd./found. ^bIn CDCl₃

Table S3 ^{31}P NMR data for tetraaryloxy-substituted PNP-lariat ethers **4c-f**.

Compound	$^{31}\text{P}\{-1\text{H}\}$ NMR, A_2B spin system ^a		$J_{\text{P,P}}$, Hz ^b
	δ_{p} , ppm		
	P_A^{c}	P_B^{d}	
4c	12.80 (d)	8.90 (t)	93.5
4d	13.70 (d)	9.40 (t)	89.6
4e	13.70 (d)	9.67 (t)	91.0
4f	13.47 (d)	9.71 (t)	91.2

^a A_2B spin system: $\text{P}_A = \text{P}(\text{OCH}_2\sim)[\text{OAr}]$; $\text{P}_B = \text{PCl}_2$;
d - double doublet, corresponds to P_A (2 equally substituted
P atoms), t - triplet, corresponds to P_B . ^b $J_{\text{P,P}}$ and δ_{p} values
calculated according to ref. 28. ^cProton coupling causes
collapse into a broad doublet. ^dProton coupling causes
weak broadening.

Table S4 Combustion analysis and ¹H NMR spectral data for tetraaryloxy-substituted PNP-lariat ethers **4c-f**.

Formula	Elemental analysis, % ^a			¹ H NMR δ _H , ppm ^b					
	C	H	N	(CH ₂) OP	(CH ₂) OC	(CH) _{Ar}	CHO	CH ₃	(CH ₂) OC=O
4c C ₄₄ H ₆₀ O ₉ N ₃ P ₃	<u>53.30</u> 53.44	<u>5.98</u> 6.07	<u>4.33</u> 4.25	4.39- 4.05 (m,4H)	3.71- 3.58 (2m,12H)	7.83- 6.81 (4m,16H)		1.39- 1.28 (t,12H)	4.37- 4.29 (qt,8H)
4d C ₅₀ H ₁₀₄ O ₉ N ₃ P ₃	<u>52.41</u> 52.26	<u>9.20</u> 9.06	<u>3.57</u> 3.66	4.90- 4.15 (m,4H)	3.75- 3.61 (2m,12H)	7.94- 6.84 (2m,16H)		1.39- 1.32 (t,12H)	4.36- 4.28 (qt,8H)
4e C ₃₆ H ₃₆ O ₁₃ N ₃ P ₃	<u>53.09</u> 53.27	<u>4.32</u> 4.44	<u>5.30</u> 5.18	4.40- 4.00 (2m,4H)	3.89- 3.52 (2m,12H)	8.00- 6.94 (5m,16H)	10.40- 9.89 (3s,4H)		
4f C ₃₆ H ₃₆ O ₁₃ N ₃ P ₃	<u>53.13</u> 53.27	<u>4.40</u> 4.44	<u>5.24</u> 5.18	4.27- 4.07 (m,4H)	3.80- 3.57 (2m,12H)	8.00- 7.03 (6M,16H)	10.04 9.83 (2s,4H)		

^a-Calcd./found. ^bIn CDCl₃

Table S5 ^{31}P NMR spectral data for per-(2-naphthyloxy) *gem*-polymethylenediamino-bridged PNP-bis-lariat ethers **10a-f** and monomeric PNP-lariat ether analogue **7**.

Compound	n ^a	$^{31}\text{P}\{-^1\text{H}\}$ NMR, AMX spin system						$^{31}\text{P}\{^1\text{H}\}$ NMR		
		δ_{p} , ppm ^b			$J_{\text{p-p}}$, Hz ^c			$J_{\text{p-H}}$, Hz		
		P _A	P _M	P _X	A-M	A-X	M-X	P _A H ^e	P _M H ^f	P _X H ^g
7	3	16.87	9.24	26.36	71.9	66.6	58.1	8.5	7.2	
10a	2	14.28	10.75	21.38	72.6	72.2	85.3	8.2	7.3	
10b	3	14.45	10.96	21.49	71.8	71.8	85.2	9.4	7.1	
10c	6	14.57	11.10	21.65	71.5	71.7	85.9	8.4	7.4	
10d	7	14.54	11.06	21.60	71.5	71.5	85.9	8.7	7.5	
10e	10	14.64	11.18	21.61	71.4	71.4	85.9	8.9	7.3	
10f	12	14.63	11.16	21.60	71.4	71.4	85.9	8.5	7.3	

^aNumber of CH₂ groups in the oligomethylene(di)amino substituent ^b δ_{p} and $J_{\text{p-p}}$ values for the AMX coupling system were calculated according to ref. 27. For **7**: P_A = P(OCH₂~)(NHPr); P_M = P(OCH₂~)(ONaph) P_X = P(ONaph)₂. For **10a-f**: P_A = P(OCH₂~)(NHR); P_M = P(OCH₂~)(ONaph) P_X = P(ONaph)₂. ^cDegenerate double doublet (collapse into a triplet due to overlap of the inner lines). ^dWell-resolved double doublets. ^eEach line splits into 6 signals on proton coupling. ^fEach line splits into a triplet. ^gLack of noticeable proton coupling.

Table S6 Combustion analysis and ^1H NMR spectral data for heterogeneously disubstituted PNP-lariat ether **7** and bis-PNP-lariat ethers **10a-f**.

Formula	Elemental analysis, % ^a			^1H NMR δ_{H} , ppm ^b						
	C	H	N	(CH ₂) OP	(CH ₂) OC	(CH) _{Ar}	NH	CH ₂ N	CH ₃	CH ₃
7 C ₂₁ H ₃₁ O ₆ N ₄ P ₃ Cl ₂	<u>42.15</u> 42.07	<u>5.28</u> 5.17	<u>9.28</u> 9.35	4.60-4.40 (m,4H)	4.40-3.45 (3m,12H)	7.88-7.45 (3m,7H)	1.87-1.67 (2m,1H)	3.00-2.80 (m,2H)	0.99-0.77 (m,3H)	1.44-1.19 (2m,2H)
10a C ₇₈ H ₈₀ O ₁₆ N ₈ P ₆	<u>59.73</u> 59.62	<u>5.18</u> 5.10	<u>7.00</u> 7.13	4.60-4.40 (m,4H)	4.40-3.45 (3m,12H)	7.88-7.45 (3m,7H)		2.50-2.10 (m,4H)		
10b C ₇₉ H ₈₂ O ₁₆ N ₈ P ₆	<u>59.99</u> 59.85	<u>5.27</u> 5.18	<u>7.01</u> 7.07	4.34-4.10 (2m,8H)	3.71-3.36 (m,24H)	7.82-7.12 (m,42H)		2.18 (d,4H)		0.67 (s,2H)
10c C ₈₂ H ₈₈ O ₁₆ N ₈ P ₆	<u>60.47</u> 60.52	<u>5.36</u> 5.41	<u>6.92</u> 6.89	4.65-4.20 (2m,8H)	3.83-3.39 (m,24H)	7.89-7.20 (m,42H)		2.45-2.00 (m,4H)		0.60 (m,8H)
10d C ₈₃ H ₉₀ O ₁₆ N ₈ P ₆	<u>60.85</u> 60.73	<u>5.61</u> 5.49	<u>6.72</u> 6.83	4.20-3.78 (2m,8H)	3.67-3.37 (m,24H)	7.89-7.20 (m,42H)		2.31-1.96 (2m,4H)		0.83-0.54 (2m,10H)
10e C ₈₆ H ₉₆ O ₁₆ N ₈ P ₆	<u>61.44</u> 61.36	<u>5.83</u> 5.71	<u>6.60</u> 6.66	4.75-4.22 (2m,8H)	3.52-3.47 (m,24H)	7.80-7.23 (m,42H)		2.40-1.83 (m,4H)		0.93 (m,8H)
10f C ₈₈ H ₁₀₀ O ₁₆ N ₈ P ₆	<u>61.90</u> 61.75	<u>5.93</u> 5.85	<u>6.49</u> 6.55	4.16-3.81 (2m,8H)	3.65-3.34 (m,24H)	7.87-7.18 (m,42H)		2.31-2.33 (m,4H)		0.76-1.26 (m,10H)

^aCalcd./found. ^bIn CDCl₃.