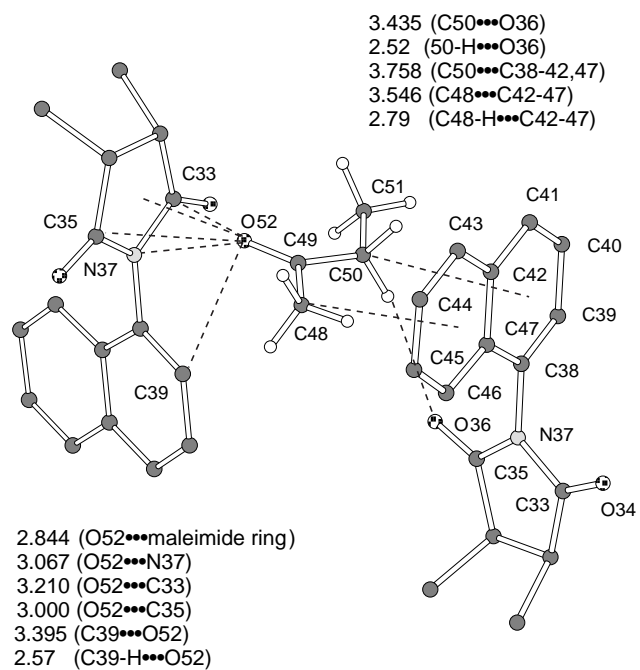


Electronic Supporting Information

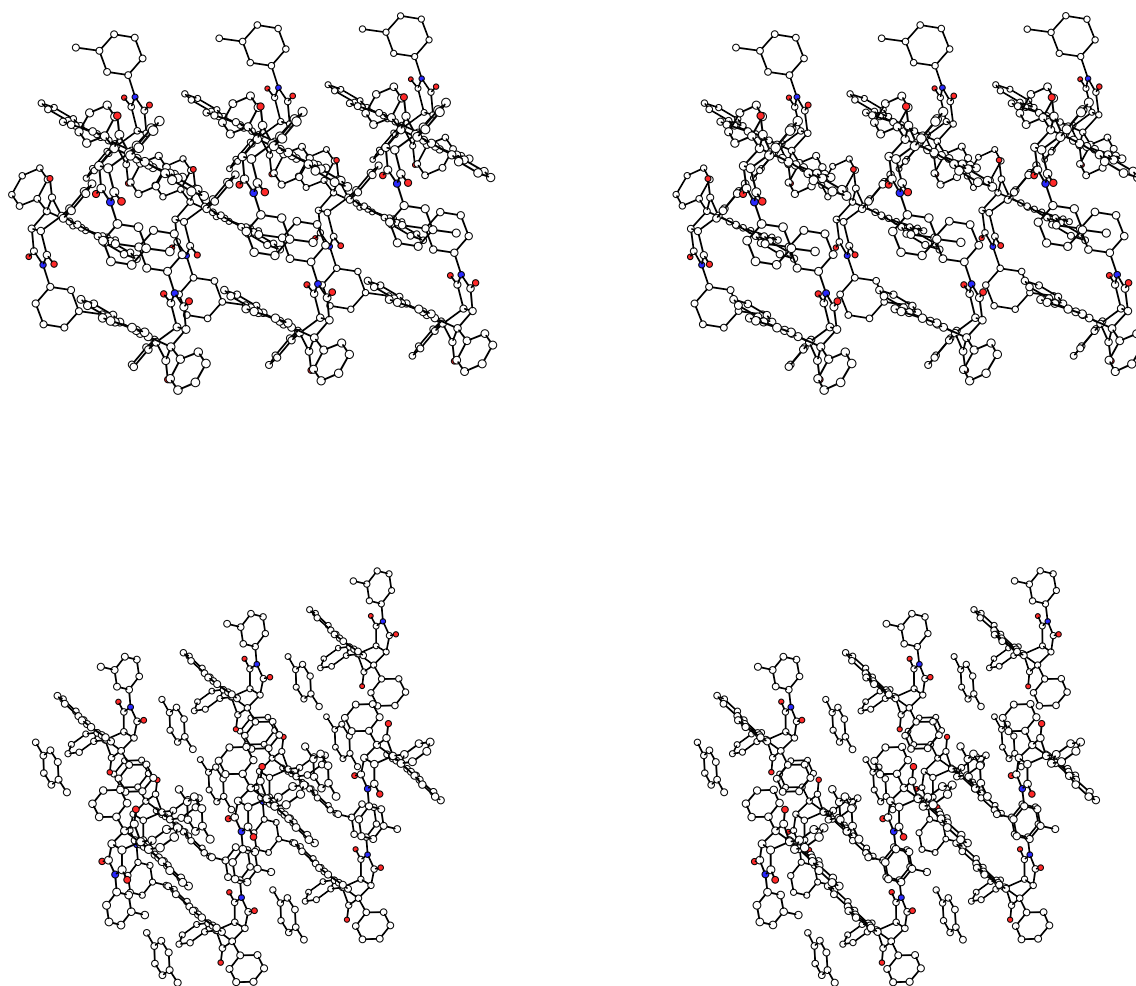
Non-hydroxylic Clathrate Hosts of [4+2] π Cycloadducts of Phencyclone and *N*-Arylmaleimides. Recognition of Aromatic Guests

**Yasuyuki Yoshitake,^a Junichi Misaka,^a Koji Setoguchi,^a Masaki Abe,^a
Tomohiro Kawaji,^a Masashi Eto^b and Kazunobu Harano^{a*}**

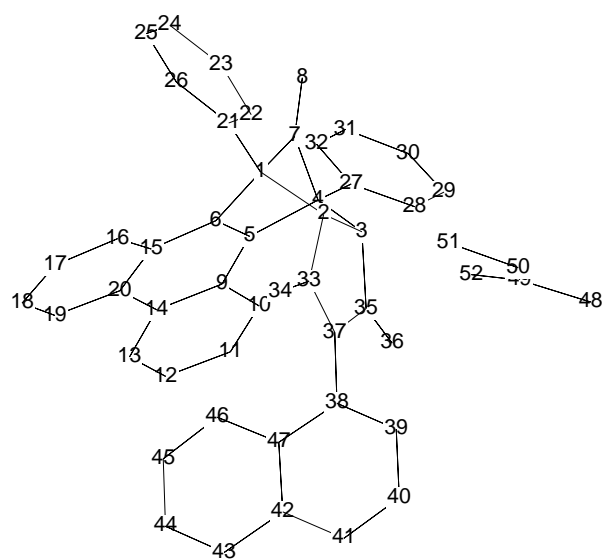
^aFaculty of Pharmaceutical Sciences, Kumamoto University, 5-1 Oe-hon-machi, Kumamoto 862-0973, Japan. School of Agriculture, ^bKyushu Tokai University, 5435 Kawayo, Choyoson, Asogun, Kumamoto 869-1404, Japan
(E-mail: harano@gpo.kumamoto-u.ac.jp)



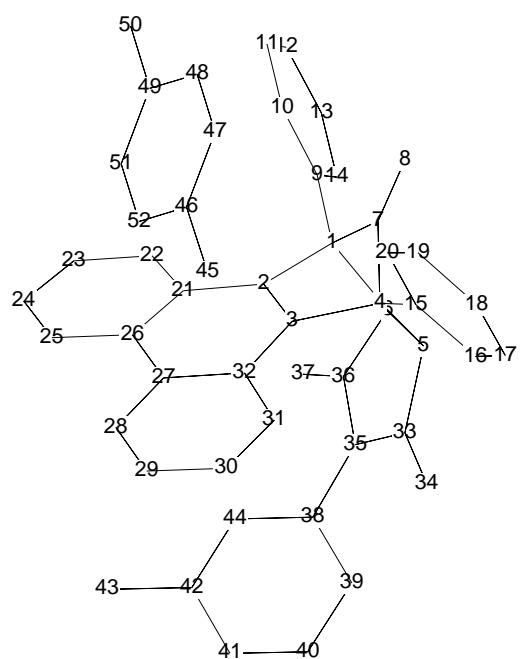
ESI-1. Intermolecular atom-plane and atom-atom distances (Å) of the **3c•**butanone complex



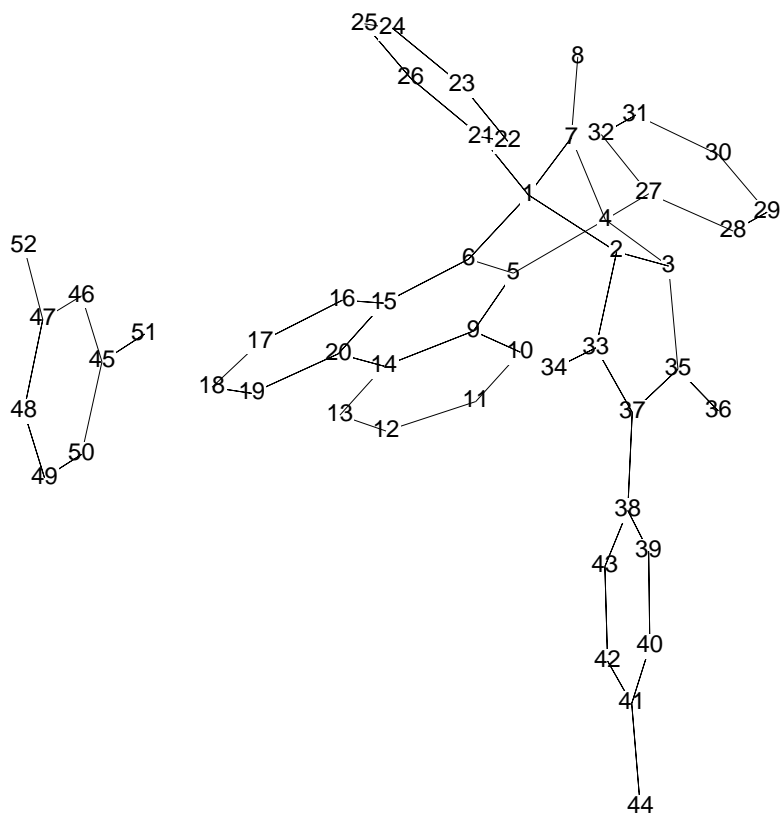
ESI-2. Stereo drawings of the crystal structure of the free host (**3g**) and the inclusion complex of **3g•p-xylene**.



ESI-3. Numering sequence of **3c**•butanone

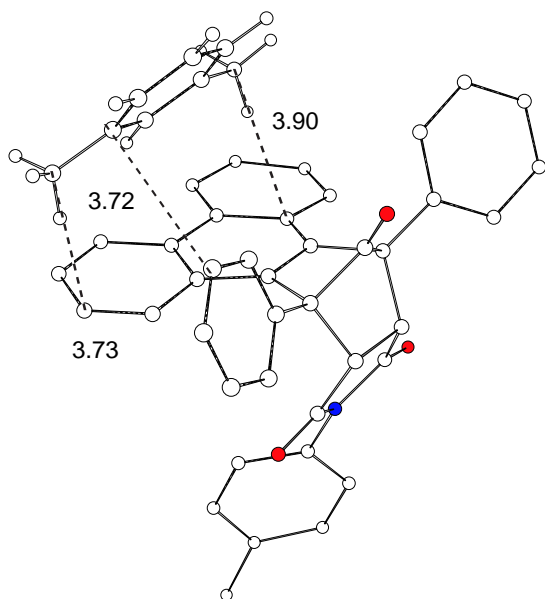


ESI-4. Numering sequence of of the free host (**3g**) and the inclusion complex of **3g**•*p*-xylene

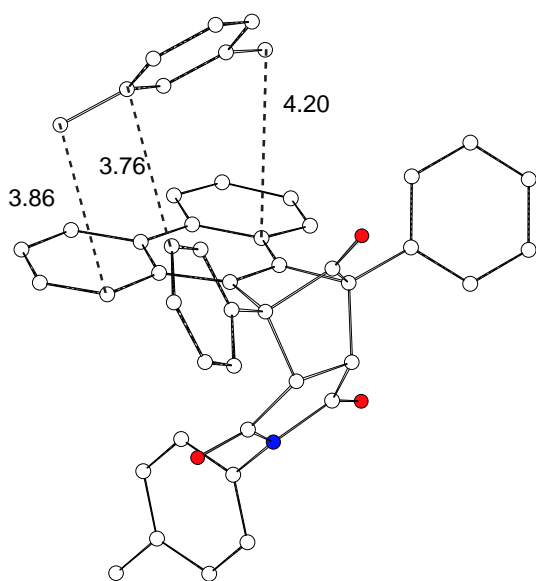


ESI-5. Numbering sequence of **2h•m**-xylene

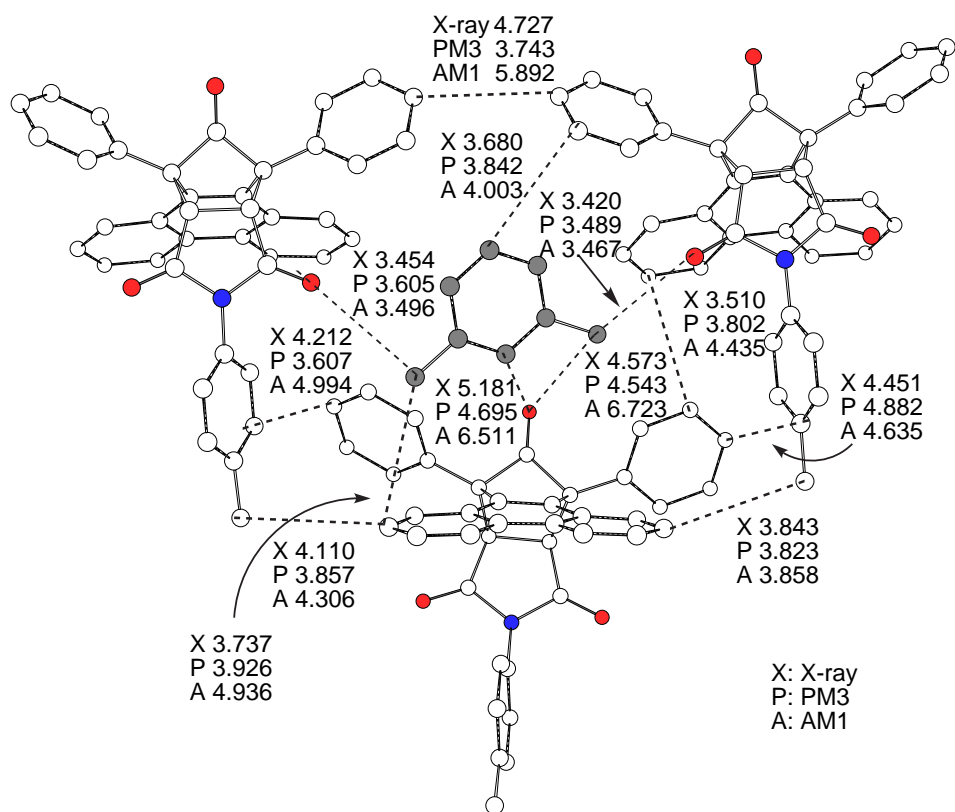
X-Ray



PM3



ESI-6. X-Ray and PM3 structures of the 1:1 complex model of m-xylene and **3h**



ESI-7. PM3-Calculated Geometry of 3:1 Complex extracted from the inclusion compound **3h** ÅEm -xylene and Comparison of the Interatomic Distance Data between the PM3- and AM1-Calculated Data and X-Ray Data

ESI-8. Heats of formation^a for the complex formation of **3h**•*m*-xylene calculated by PM3 method

compound	ΔH_f^b	$\Delta\Delta H_f^b$
<i>m</i> -xylene	4.786 (6.851)	
3h	60.082 (100.221)	
3h (1 mol) + <i>m</i> -xylene	64.868 (107.072)	0.0
3h (1 mol) • <i>m</i> -xylene	61.232 (106.999)	-3.636 (-0.073)

a) In kcal/mol. b) AM1 values in parentheses.