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Electronic Supplementary Information

for

New synthesis of a Porous Si / TiO₂ photocatalyst: testing its efficiency and stability under visible light irradiation

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Fig. S1. SEM image of porous silicon. The nanoparticles and mesopore can be seen from the surface of porous silicon.



Fig. S2 Degradation of RhB with PS / TiO_2 (50 mol% PS) in dark. There are almost no variance after 40 minutes which can be seen from the figure.



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Fig. S3. X-ray powder diffraction patterns, (a) PS / TiO_2 (50 mol% PS) (b) self-made TiO_2 (C) PS.



Fig. S4. UV-Vis reflection spectra of different photocatalysts, (a) TiO_2 (b) PS / TiO_2 (50% PS) composite. The adsorption of visible light (from 400 nm to 800 nm) to PS / TiO_2 composite are clearly much more than TiO_2 itself



Fig. S5. The influence of H_2O_2 in photocatalytic degradation of RhB on PS / TiO₂ composite in aqueous solution. (a) without H_2O_2 (b) with 0.2 ml H_2O_2 . When the experiment is conducted in dark, there is almost no degradation can be found from the figure. Meanwhile, no variance of wavelength can be detected. However, a remarkable decrease of concentration of RhB and great wavelength shift can be observed during visible irradiation. Especially, The activity of compostie was obviously increased with only a little of H_2O_2 .



a, without any catalyst under visible light b, PS / TiO₂ (50% PS) composite in dark c, 0.1ml H₂O₂ under visible light d, PS / TiO₂ (50% PS) composite under visible light e, PS / TiO₂ (50% PS) + 0.1ml H₂O₂ under visible light. The solid symbols are the results of wavelength.

Scheme S1. Proposed mechanism for photocatalytic degradation of Rhodamine B in aqueous solution.^{8, 15}



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Fig. S6. Mass spectra recorded in the positive ion mode of RhB aqueous solution after photodegradation.

PS/TiO₂-1-1-400 -120min





m/z

Fig. S7. TOC data of RhB aqueous solution after photodegradation for 2 hour.

