Supporting Information



Fig. S1. (a)UV-vis absorption spectra of ethyl 2-aminobenzoate 25μM dissolved in (red)water; (yellow)α-CD 0.030M; (green)β-CD 7.0 mM; (blue·)SDS 0.025 M and (black)TTABr 0.030 M; (b)Fluorescence emission of **2** dissolved in water, α-CD 0.030M; β-CD 7.0 mM; SDS 0.025 M and TTABr 0.030 M; λ_{ex} 335 nm; slits ex/em 4/2.



Fig. S2. (a)UV-vis absorption spectra of ethyl 4-aminobenzoate 12 μ M dissolved in (red)water; (yellow) α -CD 0.030M; (green) β -CD 7.0 mM; (cyan)SDS 0.025 M and (blue)TTABr 0.030 M; (b)Fluorescence emission of **4** dissolved in (red)water; (cyan)SDS 0.025 M; (yellow) α -CD 0.030M; (green) β -CD 7.0 mM, and (blue)TTABr 0.030 M; λ_{ex} 300 nm; slits ex/em 4/2.

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Fig.S3. (a)UV-vis absorption spectra of ethyl 3-aminobenzoate 65 μ M dissolved in (red)water; (blue) α -CD 0.050 M, and (green) γ -CD 0.060 M (the absorbance increase is the consequence of the light-scattering by the solvent); (b)Excitation en emission spectra of **3** dissolved in (---)water; (—) γ -CD 0.060 M, and (o) α -CD 0.050 M; λ_{ex} 325 nm, slits ex/em 4/4.



Fig. S4. (a)Spectra of (-)2 in aqueous SDS micelles below (at [SDS]= 0 and 7.47 mM) and above the cmc (at [SDS]= 9.33, 14, 28 mM and 0.24 M following increase emission intensity maxima) and (\circ)in TTABr micelles of 0.24 M; (b)spectra of **3** in water (negligible intensity) and in aqueous-dioxane mixtures of %v/v dioxane equal to 16.7; 33.3; 42; 50; 63, and 83.3% following the increase of emission maxima and blue shift. Data in Table 2.

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5.



 λ/nm Fig. S6 (a)Reaction spectra for basic hydrolysis of 2; (b)reaction spectra for the basic hydrolysis of 4. Experimental conditions in Table 6.



 λ/nm Fig. S7. Reaction spectra for the basic hydrolysis of 3. Experimental conditions in Table 6