Electronic Supporting Information

Cyclophane-size driving the photochemical behaviour of benzophenone.

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Synthesis and Characterization of 1 and 2.

Materials and method

Chemicals were used from commercial sources without any purification. Thin layer chromatography (TLC) was performed on silica gel 60 F 254 (Merck 5719) or on silica gel 60 F 254 (Merck 7747) prepared plates and revealed with a ultra-violet lamp. Column chromatography was carried out on silica gel 60 (40–63 μ m) from Merck (7736) for flash chromatography or silica gel 60 (63–200 μ m) from Merck (7734) for gravity columns.

Absorption spectra were recorded on a Hewlett-Packard 8452A Diode Array Spectrophotometer. Samples for absorption spectra were prepared in spectroscopic grade solvents and adjusted to a linear range response. Emission spectra were recorded in a Jasco FP-750 spectrofluorometer. Samples for emission spectra were prepared in spectroscopic grade solvents and adjusted to a linear range response. No fluorescent contaminants were detected in the wavelength region of experimental interest upon excitation. Fluorescence quantum yields were determined by comparison with 0.1 M quinine sulphate in 0.05 M sulphuric acid as a reference, and corrected for the refractive index of the solvent. The samples were irradiated in an immersion well photo reactor (Pyrex) equipped with a 150 W medium-pressure mercury lamp.

Synthetic scheme and numbering of the molecules



Figure S1. Synthetic scheme for 3,5,8,12-Tetraoxa- 4^4 -benzoyl-1,4,7(1,3),10(1,4)-tetrabenzenacyclododecaphane (1) and numbering of the molecules



Figure S2. Synthetic scheme for 2,5,9-trioxa-1⁴-benzoyl-1,4(1,3),7(1,4)-tribenzenacyclononaphane (2) and numbering of the molecules

Characterization of 1. Assignment of the signals.



Figure S3. Expansion of the ¹H-RMN (CDCl₃, 400 MHz) of 1. In the structure are shown some of nOe effects observed.



Figure S4. ¹H-RMN (400 MHz) spectra of 1 recorded in en C_6D_6 (a), $CDCl_3:C_6D_6$ (9:1) (b) and $CDCl_3$ (c).



Figure S5. NOEDIFF experiments (CDCl₃:C₆D₆ (9:1), 200 MHz) of **1**.

| | ¹ H-RMN (ppm) | ¹³ C-RMN (ppm) | HMBC (J^3) correlations |
|----------------|---------------------------------------|---------------------------|---|
| 1 ² | 6.70 (s, 1H) | 111.4 | C-1 ⁴ , C-1 ⁶ , C-2 |
| 1^4 | 6.96 (dd, 1H, <i>J</i> = 7.8, 2.1 Hz) | 117.7 | $C-1^2$, $C-1^6$ |
| 1^{5} | 7.20 (m, 3H) | 129.5 | $C-1^1$, $C-1^3$ |
| 1^{6} | 6.77 (d, 1H, <i>J</i> = 7.8 Hz) | 119.6 | C-1 ² , C-1 ⁴ , C-2 |
| 4 ² | 5.91 (d, 1H, <i>J</i> = 2.1 Hz) | 102.3 | $C-4^4$, $C-4^6$ |
| 4 ⁵ | 7.30 (m, 3H) | 131.7 | $C-4^1$, $C-4^3$ |
| 4 ⁶ | 6.45 (dd, 1H, <i>J</i> = 7.8, 2.1 Hz) | 103.8 | $C-4^2$, $C-4^4$ |
| 7^{2} | 6.62 (s, 1H) | 110.6 | C-7 ⁴ , C-7 ⁶ , C-6 |
| 7^4 | 6.84 (dd, 1H, <i>J</i> = 7.8, 2.1 Hz) | 116.6 | C-7 ² , C-7 ⁶ |
| 7 ⁵ | 7.10 (t, 1H, <i>J</i> = 7.8 Hz) | 130.0 | C-7 ¹ , C-7 ³ |
| 7^6 | 6.65 (d, 1H, <i>J</i> = 7.8 Hz) | 118.3 | C-7 ² , C-7 ⁴ , C-6 |
| $10^2 - 10^6$ | 7.30 (m, 3H) | 127.3 | C-10 ² ,10 ⁶ , C-10 ⁴ , C-9 |
| $10^3 - 10^5$ | 7.20 (m, 3H) | 125.6 | C-10 ³ ,10 ⁵ , C-10 ¹ , C-11 |
| 2´, 6´ | 7.76 (d, 2H, <i>J</i> = 7.8 Hz) | 129.7 | C-2´, 6´, C-4´, CO |
| 3´, 5´ | 7.40 (t, 2H, <i>J</i> = 7.8 Hz) | 128.1 | C-1´, C-3´, 5´ |
| 4´ | 7.51 (t, 1H, <i>J</i> = 7.8 Hz) | 132.5 | C-2′, 6′ |
| 2 | 4.82 (s, 4H) | 69.5 | $C-1^2$, $C-1^6$, $C-4^1$ |
| 6 | 4.82 (s, 4H) | 70.3 | $C-4^3$, $C-7^2$, $C-7^6$ |
| 9 | 5.16 (s, 2H) | 68.6 | C-10 ² ,10 ⁶ , C-7 ³ |
| 11 | 5.20 (s, 2H) | 69.6 | C-10 ³ ,10 ⁵ , C-1 ³ |

Table **S1**. Spectroscopic data of **1** assigned based on the HMQC and HMBC spectra of **1**.





Figure **S6**. Expansion of the ¹H-RMN (CDCl₃, 400 MHz) of **2**. In the structure are shown some of nOe effects observed.

| | ¹ H-RMN (ppm) | ¹³ C-RMN (ppm) | HMBC (J^3) correlations |
|--------------------------------|---------------------------------------|---------------------------|--|
| 1^{2} | 5.64 (d, 1H, <i>J</i> = 2.4 Hz) | 109.9 | $C-1^4$, $C-1^6$ |
| 1^{5} | 7.26 (d, 1H, <i>J</i> = 7.9 Hz) | 131.5 | C-1 ¹ , C-1 ³ , CO |
| 1^{6} | 6.50 (dd, 1H, <i>J</i> = 7.9, 2.4 Hz) | 110.5 | $C-1^2$, $C-1^4$ |
| 4 ² | 6.05 (sa, 1H) | 117.4 | C-4 ⁴ , C-4 ⁶ , C-3 |
| 4^4 | 6.96 (dd, 1H, <i>J</i> = 7.9, 2.4 Hz) | 119.7 | $C-4^2$, $C-4^6$ |
| 4 ⁵ | 7.20 (t, 1H, <i>J</i> = 7.9 Hz) | 129.5 | $C-4^1$, $C-4^3$ |
| 4 ⁶ | 6.82 (d, 1H, <i>J</i> = 7.9 Hz) | 120.7 | C-4 ² , C-4 ⁴ , C-3 |
| 7 ² -7 ⁶ | 6.99 (d, 2H, <i>J</i> = 7.9 Hz) | 128.4 | C-7 ² , 7 ⁶ , C-7 ⁴ , C-6 |
| 7 ³ -7 ⁵ | 7.04 (d, 2H, <i>J</i> = 7.9 Hz) | 129.0 | C-7 ³ , 7 ⁵ , C-7 ¹ , C-8 |
| 2´, 6´ | 7.82 (d, 2H, <i>J</i> = 7.9 Hz) | 129.8 | C-2´, 6´, C-4´, CO |
| 3′, 5′ | 7.44 (t, 2H, <i>J</i> = 7.9 Hz) | 128.1 | C-1´, C-3´, 5´ |
| 4′ | 7.55 (t, 1H, <i>J</i> = 7.9 Hz) | 132.7 | C-2′, 6′ |
| 3 | 4.97 (s, 2H) | 69.1 | $C-1^1$, $C-4^2$, $C-4^6$ |
| 6 | 5.10 (s, 2H) | 73.2 | C-4 ³ , C-7 ² , 7 ⁶ |
| 8 | 4.82 (s, 2H) | 74.7 | C1 ³ , C-7 ³ , 7 ⁵ |
| | | | |

Table S2. Spectroscopic data of 2 assigned based on the HMQC and HMBC spectra of 2.

Photochemical reaction and numbering of the molecules



Figure S7. Photochemical reaction of 1 and numbering of the molecules



Figure S8. Photochemical reaction of 2 and numbering of the molecules



Figure S9. ¹H-RMN (200 MHz) spectra of **6** recorded in en immediately after solution preparation (a), after 15 minutes (b) and after 1 h (c).

NMR-Spectra



Figure S10. Proton, carbon and SEFT NMR spectra (CDCl₃) of 2,4-Bis[3'-(tosyloxy)benzyloxy] benzophenone



Figure S11. Proton, carbon and SEFT NMR spectra (CDCl₃) of 2,4-Bis[3'-(hydroxyl)benzyloxy] benzophenone (4a)



Figure S12. Proton, carbon and SEFT NMR spectra (CDCl $_3$) of 2-Hydroxy-4-[3'-(tosyloxy)benzyloxy] benzophenone



Figure **S13**. Proton, carbon and SEFT NMR spectra (CDCl₃) of 2-Hydroxy-4-[3'-(hydroxy)benzyloxy] benzophenone (**4b**)



Figure S14. Proton, carbon and SEFT NMR spectra (CDCl₃) of 1





