## **Supporting information**

## Effect of Nanocavity on the Torsional dynamics of Thioflavin T in Various Non-aqueous Reverse Micelles

Aninda Chatterjee and Debabrata Seth\*

Department of Chemistry, Indian Institute of Technology, Patna 800013, Bihar, INDIA

E-mail : debabrata@iitp.ac.in; Fax : 91-612-2277383

Table S-1:

The absorption, emission peak and emission quantum yields of ThT in glycerolmethanol mixtures.

Sr no	System	$\lambda_{\max}^{abs}$ (nm)	$\lambda_{\max}^{emi}$ (nm)	$\Phi_{ m f}$	Viscosity (cP)
1.	ThT+20% glycerol	417	478	$5.741 \times 10^{-4}$	4.8
2.	ThT+25% glycerol	417	479	$6.15 \times 10^{-4}$	6.0
3.	ThT+30% glycerol	417	481	$7.23 \times 10^{-4}$	7.7
4.	ThT+35% glycerol	419	482	8.36x10 <sup>-4</sup>	10
5.	ThT+40% glycerol	418	484	$1.22 \times 10^{-3}$	13
6.	ThT+45% glycerol	419	484	$1.32 \times 10^{-3}$	18
7.	ThT+50% glycerol	419	483	$1.703 \times 10^{-3}$	28
8.	ThT+55% glycerol	419	485	$2.05 \times 10^{-3}$	36
9.	ThT+60% glycerol	420	485	$2.6 \times 10^{-3}$	58
10.	ThT+65% glycerol	420	486	$3.6 \times 10^{-3}$	70
11.	ThT+70% glycerol	420	486	$5.8 \times 10^{-3}$	130
12.	ThT+75% glycerol	421	487	$7.5 \times 10^{-3}$	170
13.	ThT+80% glycerol	420	487	0.0106	250
14.	ThT+85% glycerol	422	487	0.013	350
15.	ThT+90% glycerol	422	487	0.018	630

## Table S-2:

The absorption and emission property of ThT molecule in different dioxane-water

mixtures

Sr	System	$\lambda^{abs}_{ m max}$	Absorption	$\lambda^{emi}_{\max}$	Emission	$\Phi_{ m f}$	E <sub>T</sub>
no		(nm)	energy <sup>#</sup>	(nm)	energy <sup>#</sup>		(30)
		()	$E_a$ (Kcal mol <sup>-1</sup> )	()	$E_{f}$ (Kcal mol <sup>-1</sup> )		
1	ThT+dioxane	411	69.56	475	60.19	$2.08 \times 10^{-3}$	36.6
2	ThT+99.1% dioxane	411	69.56	478	59.81	$1.46 \times 10^{-3}$	38.5
3	ThT+97.2% dioxane	414	69.06	480	59.56	$1.28 \times 10^{-3}$	42.0
4	ThT+95.3% dioxane	414	69.06	481	59.44	$1.18 \times 10^{-3}$	43.2
5	ThT+92.5% dioxane	418	68.40	483	59.19	$9.95 \times 10^{-4}$	45.6
6	ThT+81.3% dioxane	421	67.91	484	59.07	$7.26 \times 10^{-4}$	48.7
7	ThT+71.9% dioxane	421	67.91	485	58.95	$6.97 \times 10^{-4}$	50.5
8	ThT+57.8% dioxane	421	67.91	486	58.83	$6.58 \times 10^{-4}$	52.6
9	ThT+43.2% dioxane	421	67.91	486	58.83	$5.51 \times 10^{-4}$	54.9
10	ThT+34.4% dioxane	420	68.07	486	58.83	$4.55 \times 10^{-4}$	56.5
11	ThT+25.3% dioxane	418	68.40	485	58.95	$4.13 \times 10^{-4}$	57.9
12	ThT+11% dioxane	416	68.73	484	59.07	$2.79 \times 10^{-4}$	60.9
13	ThT+6.2% dioxane	414	69.06	483	59.19	$2.64 \times 10^{-4}$	62.8
14	ThT+water	412	69.39	481	59.44	$2.04 \times 10^{-4}$	63.6

<sup>#</sup>E (Kcalmol<sup>-1</sup>) = 28590/ ( $\lambda_{max}/nm$ ).



Fig. S 1: The  $log\phi_f$  vs logn plot for ThT in glycerol-methanol mixtures.



**Fig. S 2:** Residual for the fitted fluorescence lifetime decay ThT in (a) glycerol/AOT/isooctane reverse micelle at w = 3 (b) ethylene glycol/AOT/isooctane reverse micelle at w = 1 and (c) DMF/AOT/isooctane reverse micelle at w = 0.5.