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## **Supporting Information**

## *"Turn-On"* fluorescent chemosensor for Zinc (II) dipodal ratiometric receptor: Application in live cell imaging.

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Figure S1. IR spectra of receptor 2







Figure S3. <sup>13</sup>C-NMR spectra of receptor 2



Figure S4. LC-MS spectra of receptor 2 (M+H<sup>+</sup>)



**Figure S5a.** TGA of receptor **2** (heating rate 10 °C per minute under nitrogen environment).



**Figure S5b.** TGA of receptor  $2.Zn^{2+}$  (heating rate 10 °C per minute under nitrogen environment).



Figure S6a. DSC of receptor 2.



Figure S6b. DSC of receptor 2.Zn<sup>2+</sup>.



**Figure S7a**. Fluorescence titration spectra of receptor **2** (0.1 mM) in the presence of different concentrations of  $Zn^{2+}$  (1 mM) ( $\lambda ex = 278$  nm,  $\lambda em = 341$  nm, excitation and emission slit 5 nm) up to 10 µl (0.05 equiv.).



**Figure S7b**. Fluorescence titration spectra of receptor **2** (0.1 mM) in the presence of different concentrations of  $Zn^{2+}$  (1 mM) ( $\lambda ex = 278$  nm,  $\lambda em = 341$  nm, excitation and emission slit 5 nm) up to 400  $\mu$ L (2 equiv.).



from 2:1 to 1:1.



Figure S8. Fluorescence spectrometric response ( $\Delta F = F - F_0$ ) of receptor 2 (0.1 mM) upon addition of 100  $\mu$ L of respective cation salts (1 mM) in CH<sub>3</sub>CN/H<sub>2</sub>O.



**Figure S9.** A fluorescence sensing of  $Zn^{2+}$  ion (1 mM, 1equiv) by receptor **2** (0.1 mM) in the presence of other competing cations (1mM, 2 equiv.).



**Figure S10a.** Job's plot representing the stoichiometry of complex **2**.Zn<sup>2+</sup> (host : guest; 1:1).



**Figure S10b.** Normalized plot obtained from fluorescence spectroscopy ( $\lambda em = 341$  nm).



**Figure S10c.** Fluorescence intensity at 341nm of receptor **2** (0.1 mM) versus increasing concentration of  $\text{Log}[\text{Zn}^{2+}]$ . The fluorescence response fits to a Hill coefficient of 1.1446, which is in concordance with the 1:1 binding stoichiometry for the receptor **2**.Zn<sup>2+</sup> complex.

<u>Spectrum Plot - 8/8/2013 10:54 AM</u> 1 A Scan 37 from c:\iit\2013-2014\lems-45\S-4-Zn-371-msms.xms



Figure S11. LC-MS spectra of 2.Zn<sup>2+</sup> complex (M+H<sup>+</sup>)



**Figure S12a.** A Benesi-Hildebrand methodology for receptor **2**,  $(1/\Delta F)$  vs 1/[G],  $K_a = 1.29 \times 10^6$  M<sup>-1</sup>.



**Figure S12b.** A Scatchard methodology for receptor **2**,  $\Delta F/[G]$  vs  $\Delta F$ ,  $K_a = 1.00 \times 10^6$  M<sup>-1</sup>.



**Figure S12c.** Connor's fitting method for receptor **2**,  $(1-F/F_0)/[G]$  vs  $F/F_0$ ,  $K_a = 1.94 \times 10^6 \,\mathrm{M}^{-1}$ 



Figure S13. <sup>1</sup>H-NMR spectra (A) only receptor 2 (B) complex 3.



**Figure S14:** Change in fluorescence intensity of receptor **2** at 341 nm upon varying the pH of the solution.



**Figure S15:** Change in fluorescence intensity of receptor **2** at 341 nm at various temperature.



**Figure S16:** Fluorescence intensity of receptor **2** at 341 nm upon addition of  $Zn^{2+}$  ion (1 mM) over time.

| References                | Detection Limit        | Solvent                             |
|---------------------------|------------------------|-------------------------------------|
|                           |                        |                                     |
| Zhou et. al., (2012)      | 2.2×10 <sup>-7</sup> M | CH <sub>3</sub> CN-H <sub>2</sub> O |
| Dong et. al., (2014)      | 1×10-7 M               | CH <sub>3</sub> CN-Tris-HCl         |
| Li et. al., (2014)        | 4.9×10 <sup>-8</sup> M | CH <sub>3</sub> CN-H <sub>2</sub> O |
| Kaur et. al., (2014)      | 1×10 <sup>-6</sup> M   | DMF-H <sub>2</sub> O                |
| Sivaraman et. al., (2012) | 1.5×10 <sup>-7</sup> M | Phosphate buffer                    |
| Present work              | 6.5×10 <sup>-7</sup> M | CH <sub>3</sub> CN-H <sub>2</sub> O |
|                           |                        |                                     |

 Table S1: Comparison of literature reports with present work.

| Parameter           | Receptor 2 | 3        |
|---------------------|------------|----------|
| Dihedral angles (°) |            |          |
| N25-C29-C30-N26     | 61.09      | 55.65    |
| C29-C30-N26-C24     | -135.57    | -158.35  |
| C5-C4-C23-N25       | 33.34      | 115.51   |
| O21-C3-C4-C23       | -0.50      | 6.07     |
| C27-C23-N25-C29     | -66.67     | -63.07   |
| C3-C4-C23-N25       | -147.43    | -68.10   |
| O22-C12-C11-C24     | -2.34      | 6.08     |
| C11-C24-N26-C30     | 157.50     | 169.33   |
| Bond angles (°)     |            |          |
| C23-N25-C29         | 116.22     | 113.26   |
| C5-C4-C23           | 121.38     | 123.03   |
| C30-N26-C24         | 120.29     | 113.26   |
| N26-C24-C28         | 113.49     | 110.98   |
| N25-C29-C30         | 109.24     | 111.36   |
| Bond Length (Å)     |            |          |
| N25-C29             | 1.46       | 1.52     |
| C30-N26             | 1.46       | 1.52     |
| C24-N26             | 1.47       | 1.54     |
| C23-N25             | 1.47       | 1.54     |
| C12-O22             | 1.39       | 1.46     |
| C3-O21              | 1.40       | 1.46     |
| O22-H45             | 0.97       | 0.97     |
| O21-H44             | 0.97       | 0.97     |
| N25-H42             | 1.01       | 1.02     |
| N26-H43             | 1.01       | 1.02     |
| C23-C27             | 1.54       | 1.53     |
| Energy (a.u.)       | -960.28    | -1025.33 |

 Table S2: An optimized bond angles, dihedral angles, bond length and energy

 calculated at B3LYP/ LANL2DZ level.