

Electronic Supporting Information

Controlled stereoselective polymerization of lactide monomers by group 4 metal initiators that contain a (OSSO)-type tetradentate bis(phenolate) ligand

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List of Contents

1. NMR Characterizations of Complexes	
1.1. Complex Zr-1	S3
Fig. S1 and S2 ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of complex Zr-1	
1.2. Complex Zr-2	S4
Fig. S3 and S4 ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of complex Zr-2	
1.3. Complex Zr-3	S5
Fig. S5 and S6 ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of complex Zr-3	
1.4. Complex Zr-4	S6
Fig. S7 and S8 ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of complex Zr-4	
1.5. Complex Ti-4	S7
Fig. S9 and S10 ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of complex Ti-4	
1.6. Complex Ti-4a	S8
Fig. S11 and S12 ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of complex Ti-4a	
2. Microstructure Analysis	S9
2.1. Polymerization of meso-lactide	S9
Fig S13-S16 $^{13}\text{C}\{^1\text{H}\}$ spectra of polymerization of <i>meso</i> -lactide	S9
2.2. Polymerization of rac-lactide	S11
Fig S17-S20 $^1\text{H}\{^1\text{H}\}$ spectra of polymerization of <i>rac</i> -lactide	S11
2.3. Polymerization of L-lactide	S13
Fig S21-S24 $^1\text{H}\{^1\text{H}\}$ spectra of polymerization of L-lactide	S13
3. Polymerization Kinetics	S15
3.1. Kinetic of polymerization of rac-lactide	S15
Fig. S25 Semilogarithmic plots of <i>rac</i> -lactide	S15
3.2. Kinetic of polymerization of L-lactide	S16
Fig. S26 Semilogarithmic plots of L-lactide using Zr-1 and Zr-4	S16
Fig. S27-S28 Semilogarithmic plots of L-lactide using Ti-4 and Ti-4a	S16
4. Polymerization Data	S17
4.1. Polymerization of meso-lactide	S17
Table S1-S6 Tables of polymerizations of <i>meso</i> -lactide	S17
4.2. Polymerization of rac-lactide	S18
Table S7-S10 Tables of polymerizations of <i>rac</i> -lactide	S18
4.3. Polymerization of L-lactide	S19
Table S11-S14 Tables of polymerizations of L-lactide	S19

1 NMR characterisation of complexes

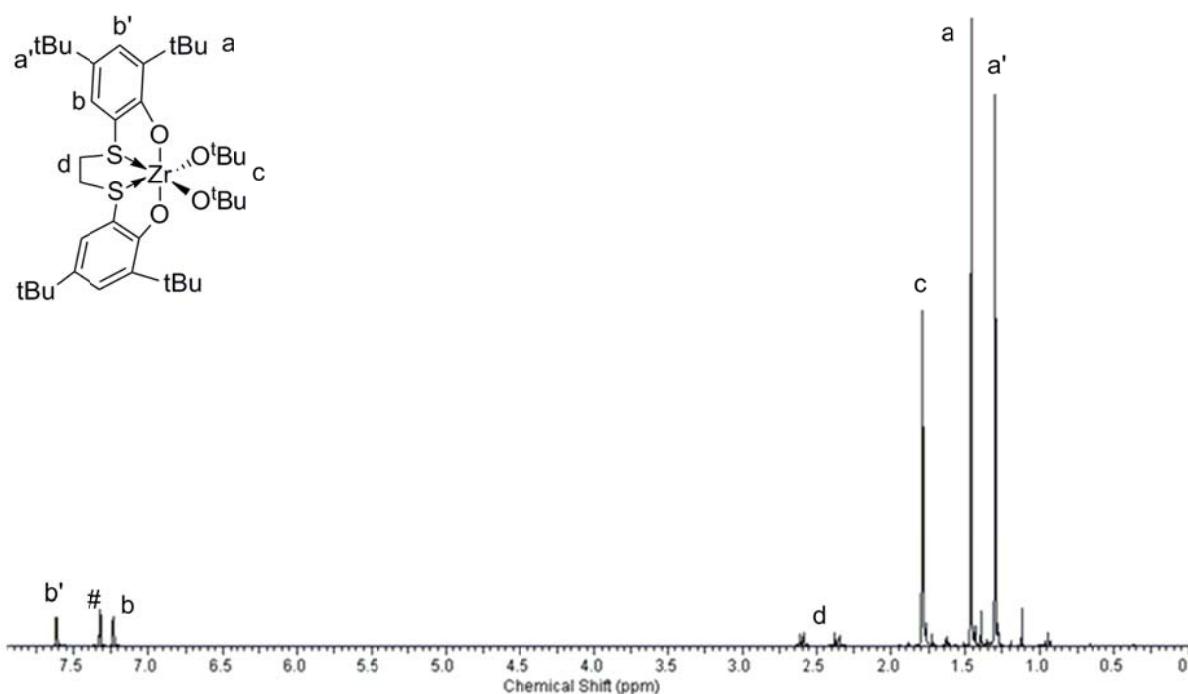


Fig. S1 ^1H -NMR spectrum of complex **Zr-1** (400 MHz, C_6D_6 [#], 25 °C)

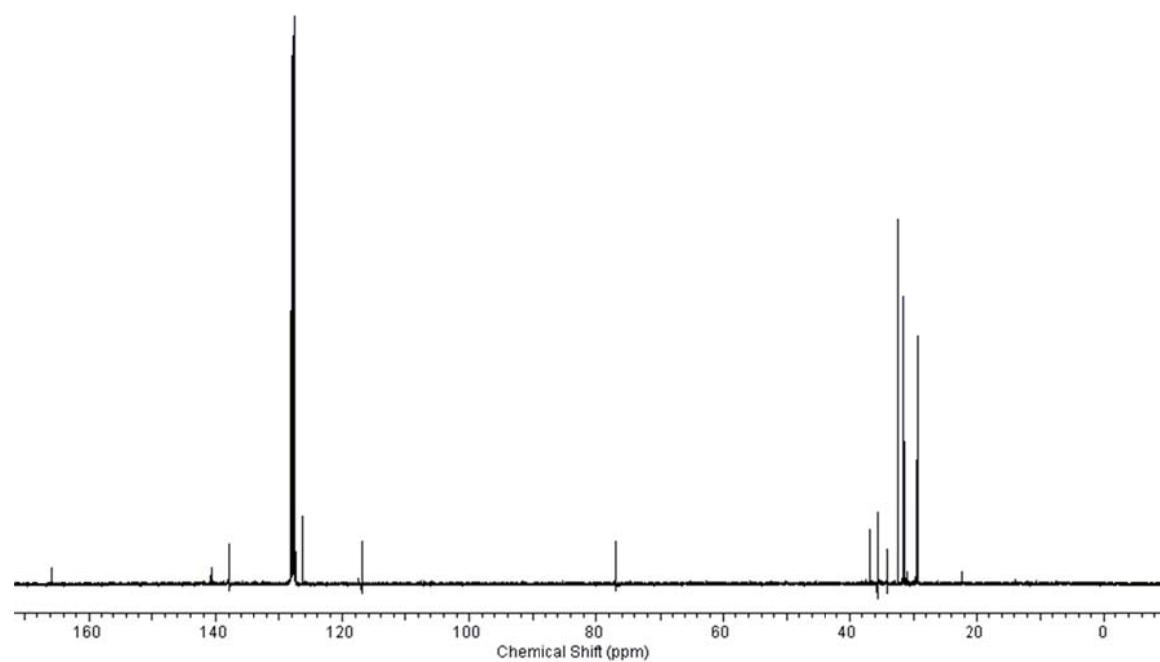


Fig. S2 $^{13}\text{C}\{\text{H}\}$ -NMR spectrum of complex **Zr-1** (100.1 MHz, C_6D_6 [#], 25 °C).

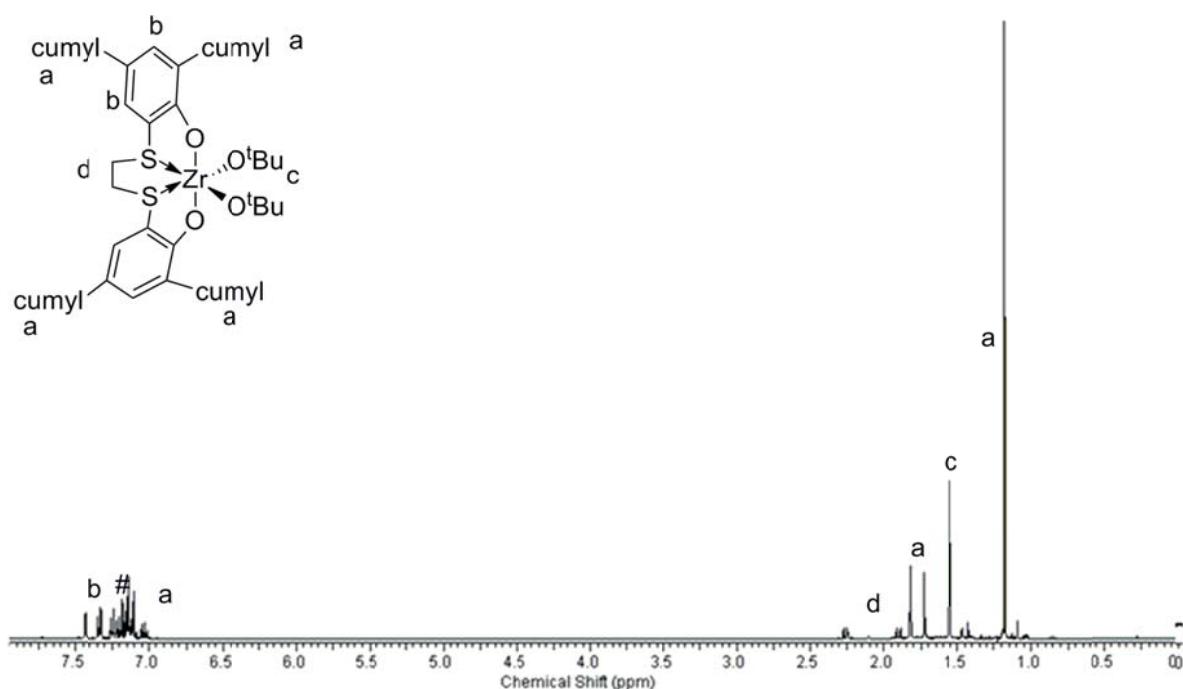


Fig. S3 ^1H -NMR spectrum of complex **Zr-2** (400 MHz, C_6D_6 [#], 25 °C)

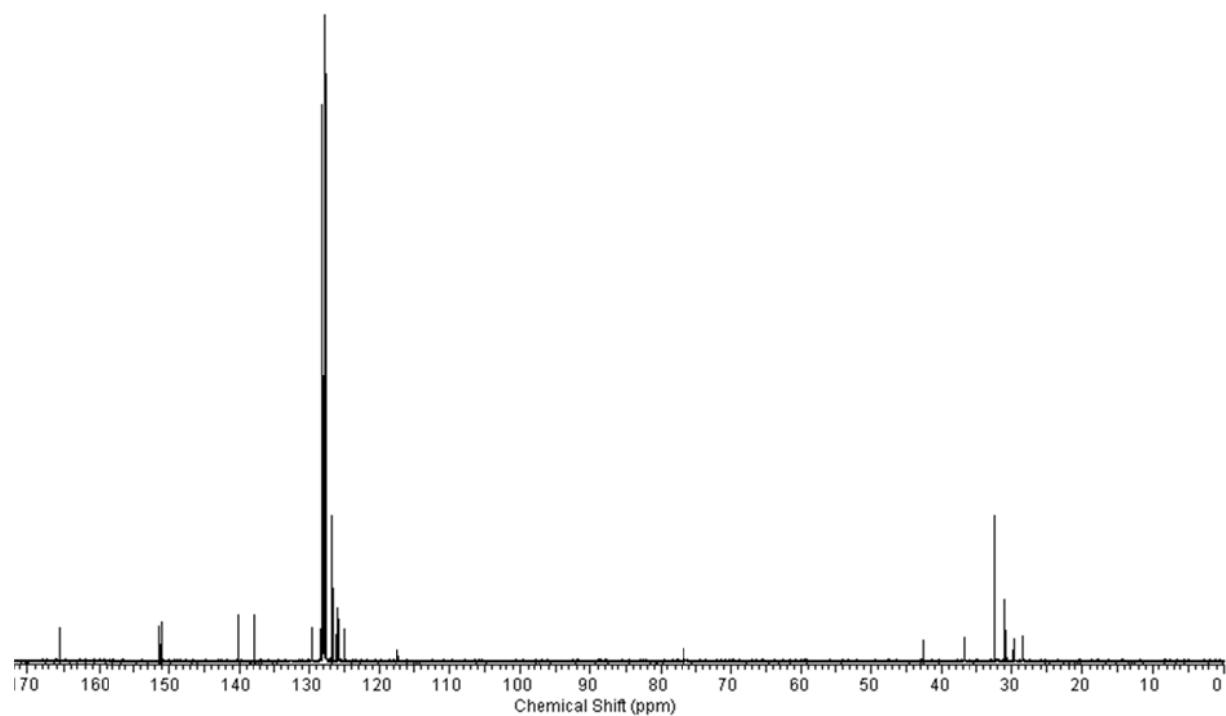


Fig. S4 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **Zr-2** (100.1 MHz, C_6D_6 [#], 25 °C).

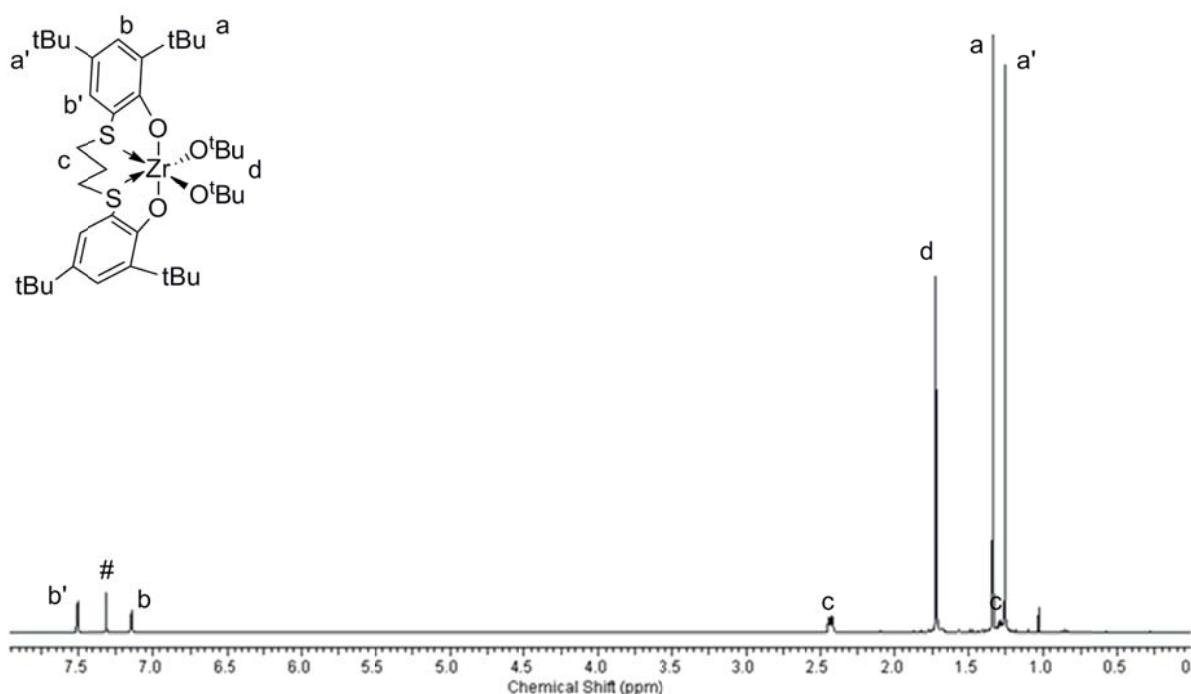


Fig. S5 ¹H-NMR spectrum of complex **Zr-3** (400 MHz, C₆D₆ [#], 25 °C)

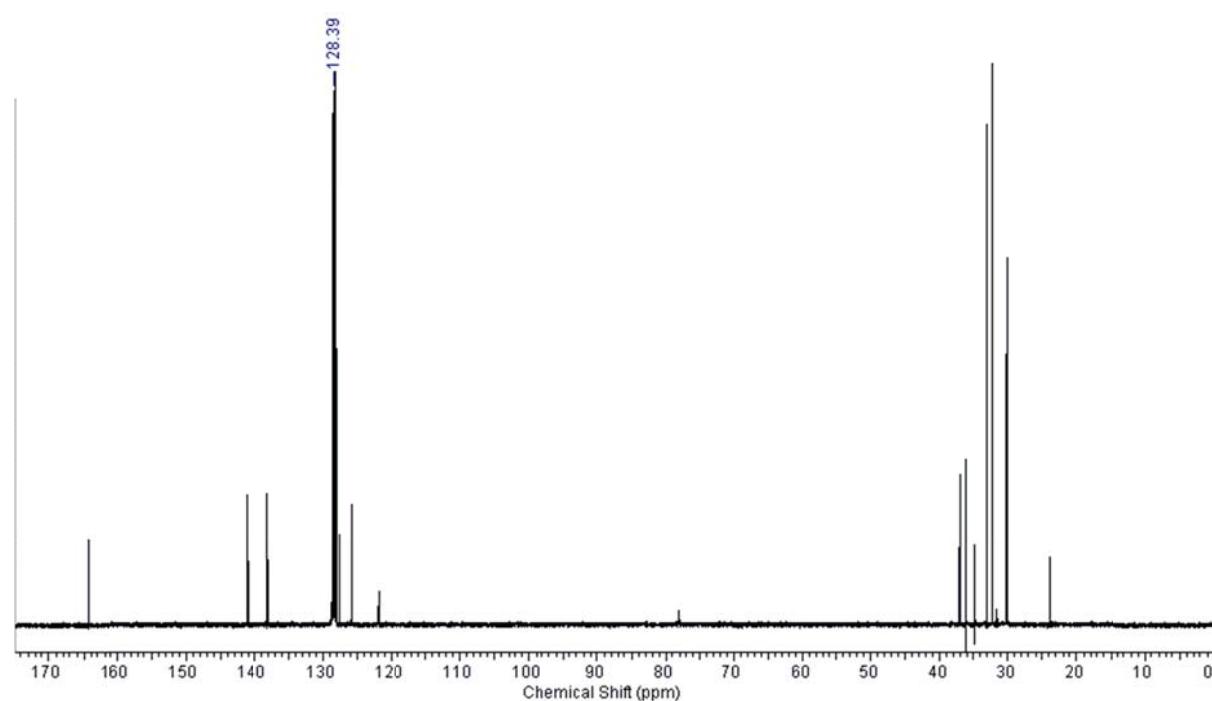


Fig. S6 ¹³C{¹H}-NMR spectrum of complex **Zr-3** (100.1 MHz, C₆D₆ [#], 25 °C).

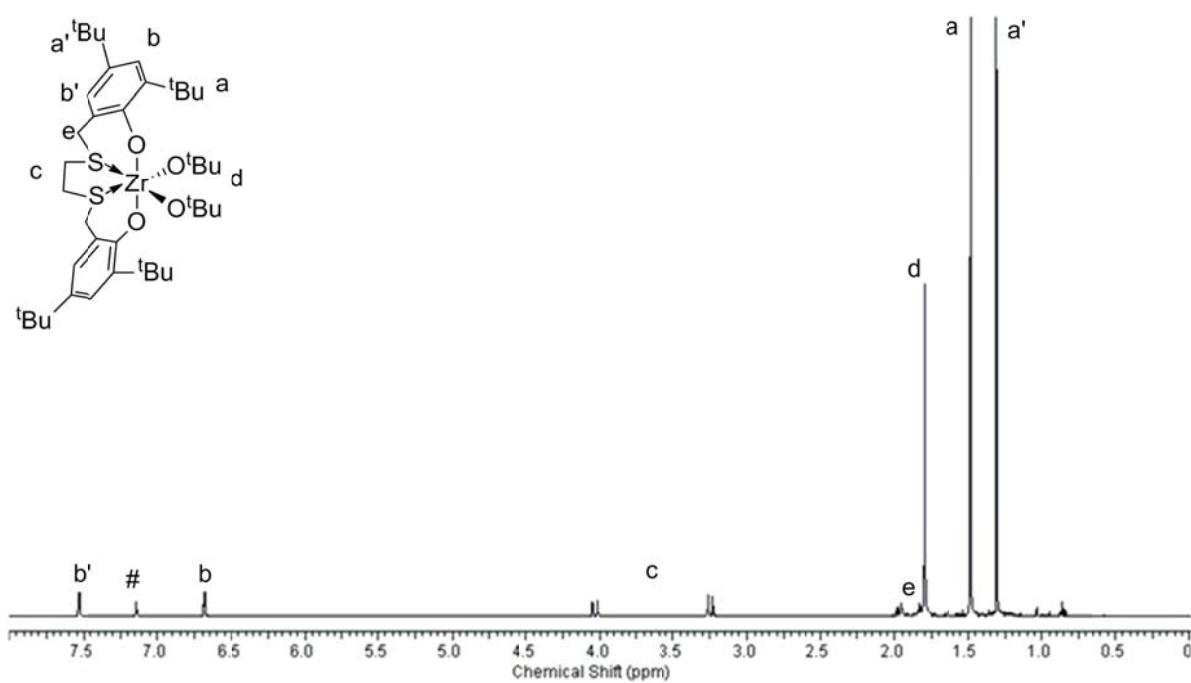


Fig. S7 ^1H -NMR spectrum of complex **Zr-4** (400 MHz, C_6D_6 [#], 25 °C)

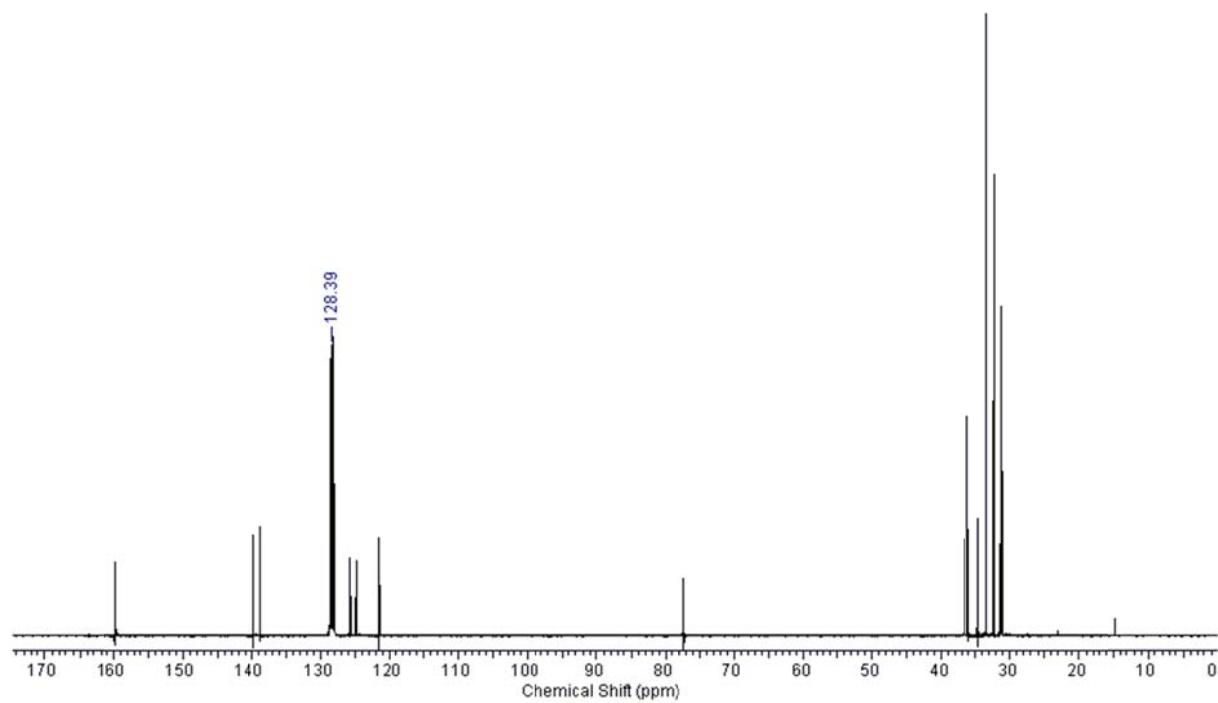


Fig. S8 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **Zr-4** (100.1 MHz, C_6D_6 [#], 25 °C).

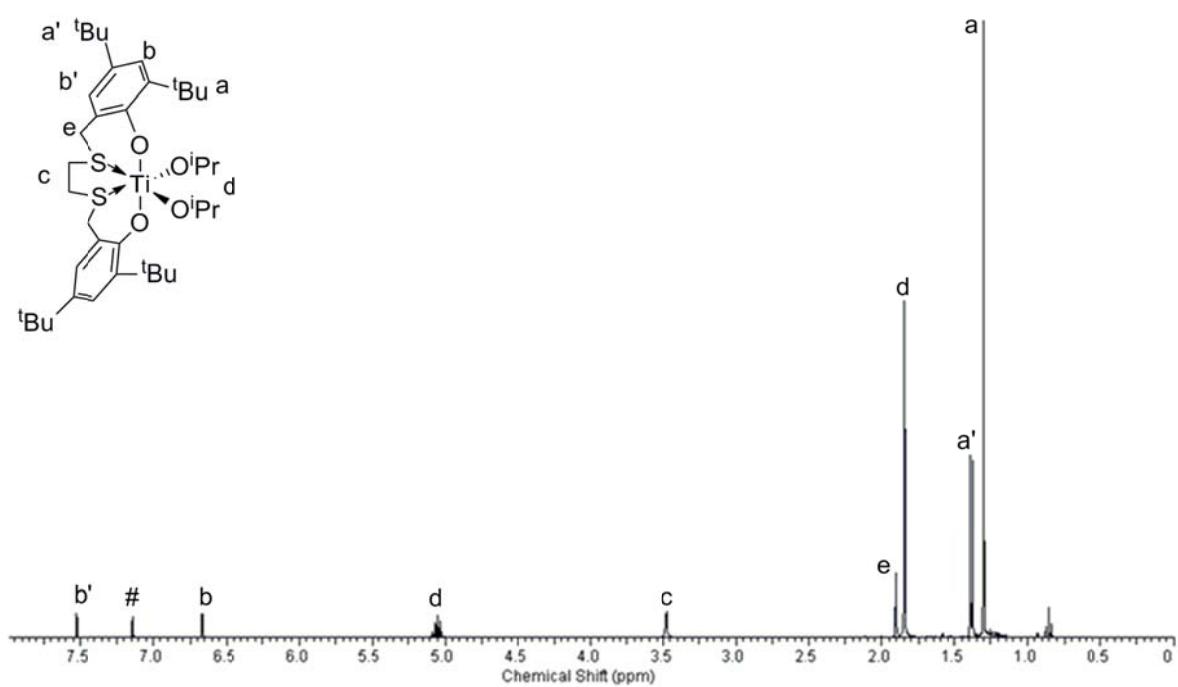


Fig. S9 ${}^1\text{H}$ -NMR spectrum of complex **Ti-4** (400 MHz, C_6D_6 [#], 25 °C)

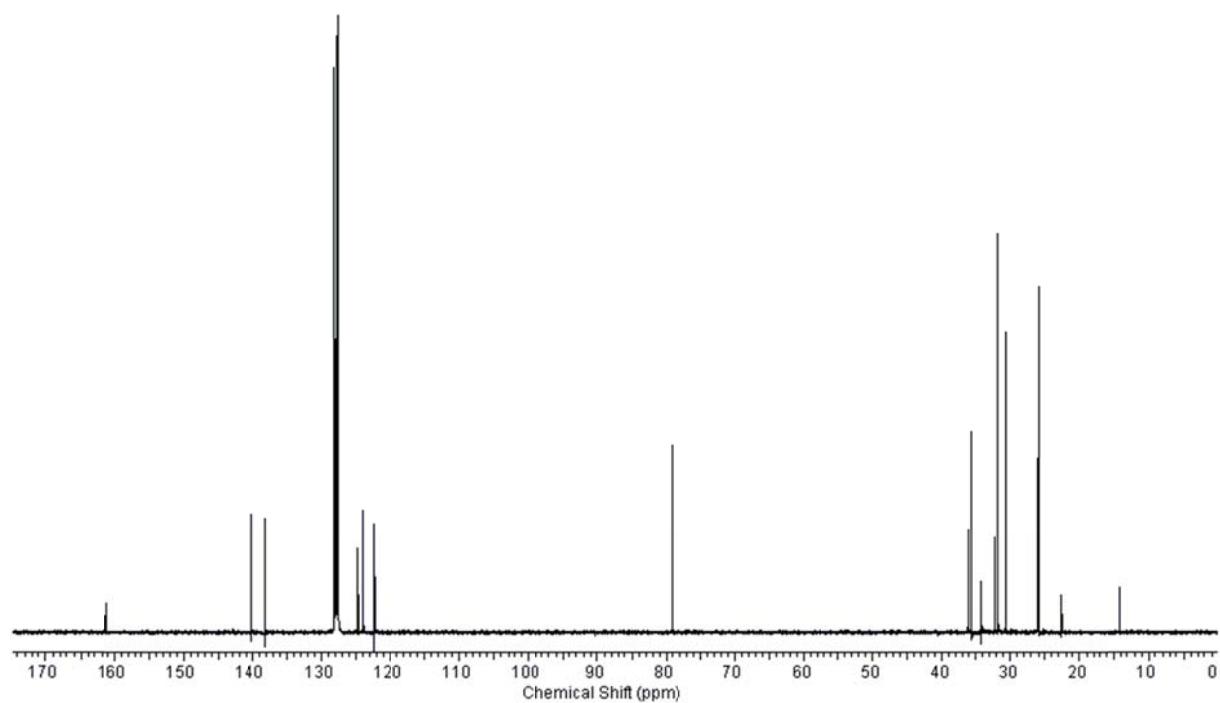


Fig. S10: ${}^{13}\text{C}\{{}^1\text{H}\}$ -NMR spectrum of complex **Ti-4** (100.1 MHz, C_6D_6 [#], 25 °C).

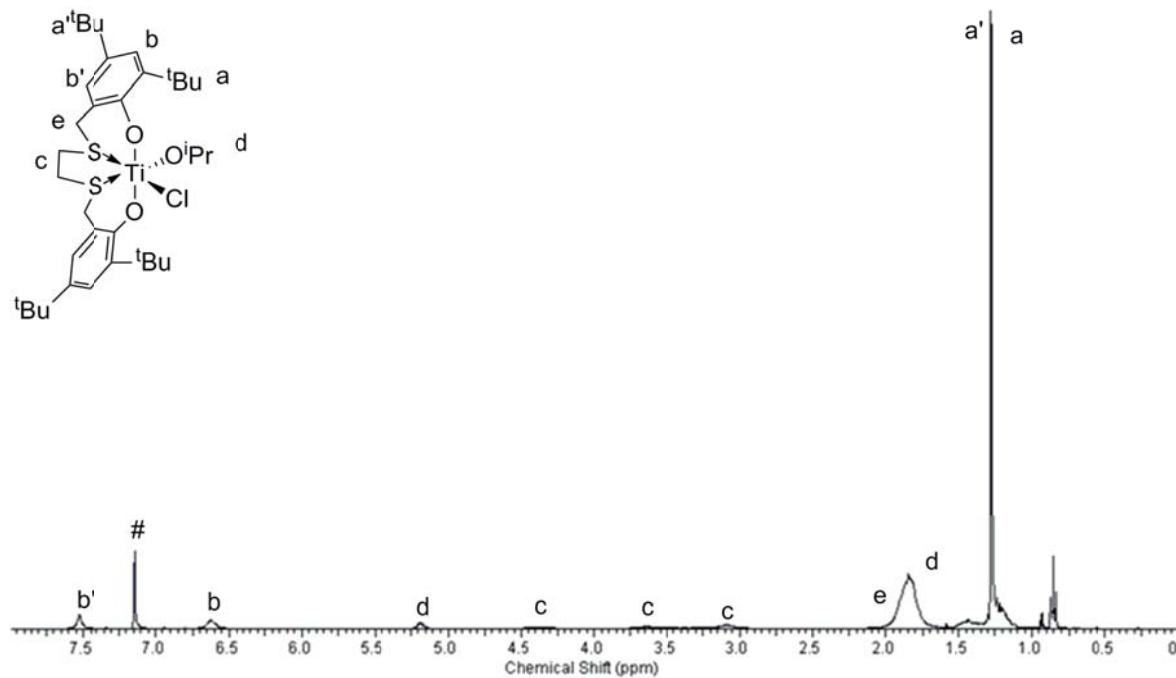


Fig. S11 ^1H -NMR spectrum of complex **Ti-4a** (400 MHz, C_6D_6 [#], 25 °C)

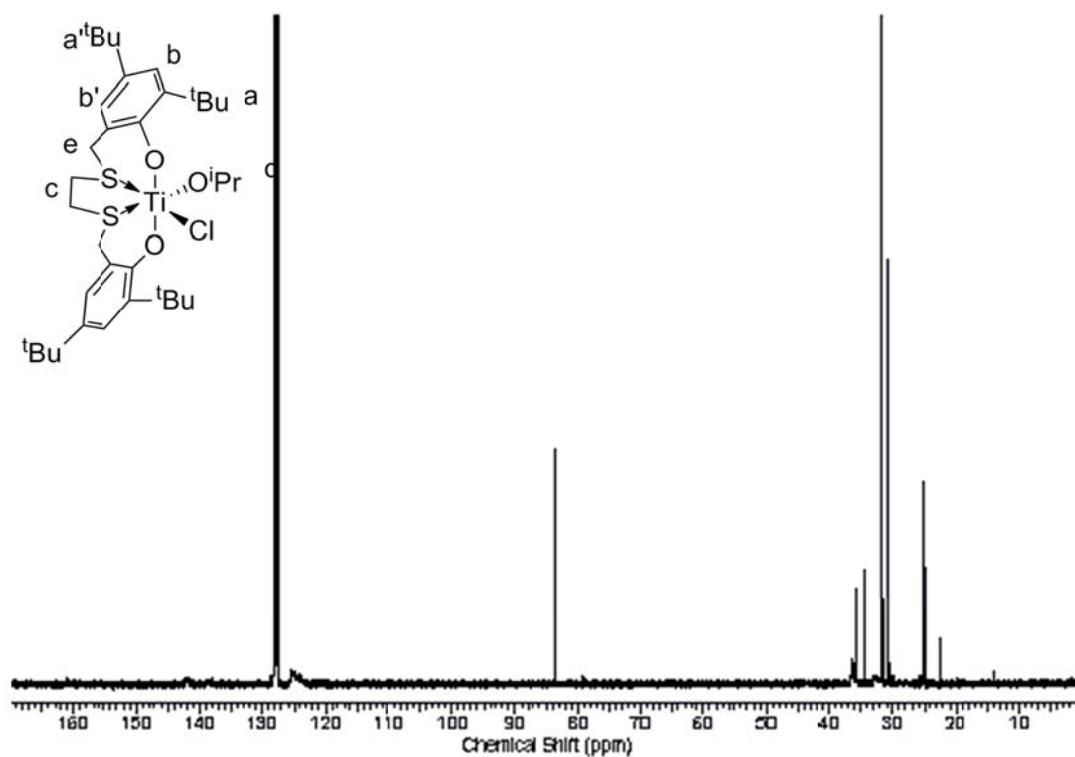


Fig. S12: $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of complex **Ti-4a** (100.1 MHz, C_6D_6 [#], 25 °C).

2 Polymerization of lactide monomers

2.1 Polymerization of *meso*-lactide

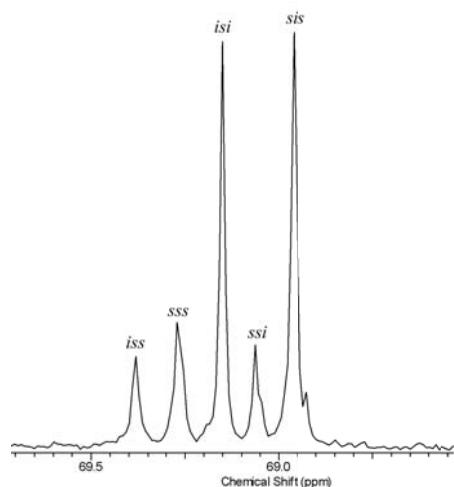


Fig. S13 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of the polymerization of *meso*-lactide using **Zr-2**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

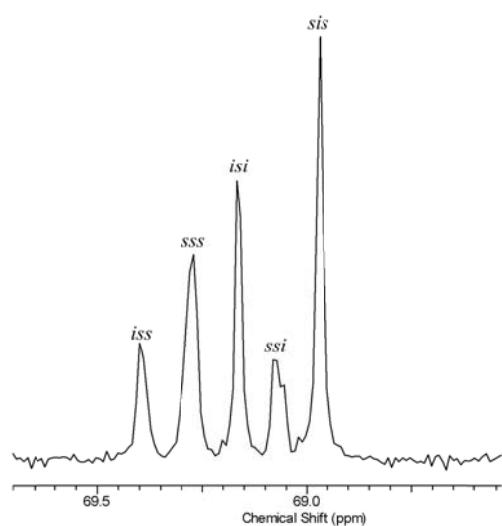


Fig. S14 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of the polymerization of *meso*-lactide using **Zr-3**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

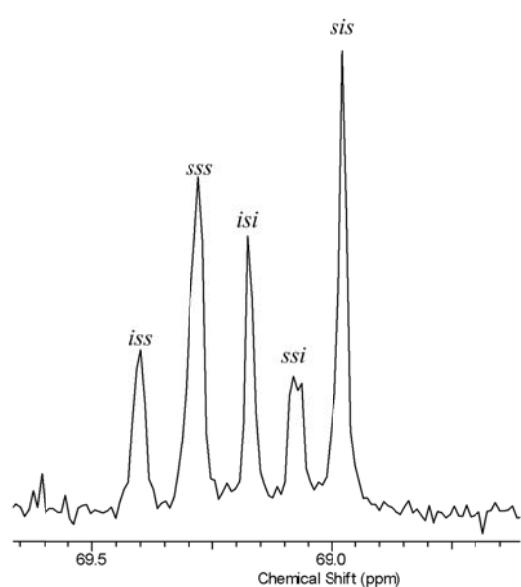


Fig. S15 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of the polymerization of *meso*-lactide using **Ti-4**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

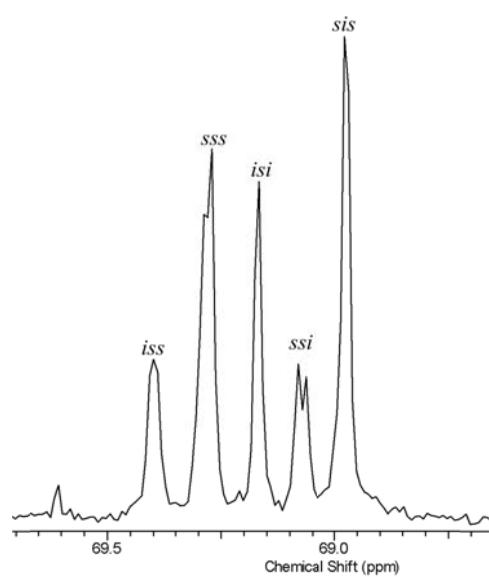


Fig. S16 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of the polymerization of *meso*-lactide using **Ti-4a**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

2.2 Polymerization of *rac*-lactide

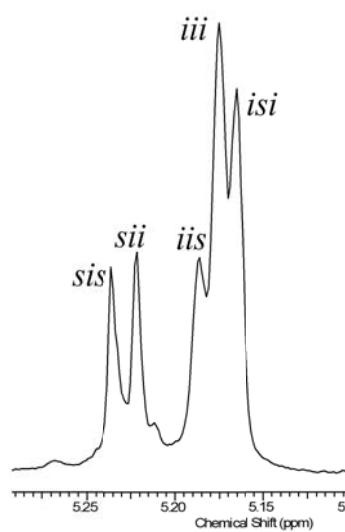


Fig. S17 $^1\text{H}\{^1\text{H}\}$ -NMR spectrum of the polymerization of *rac*-lactide using **Zr-1**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

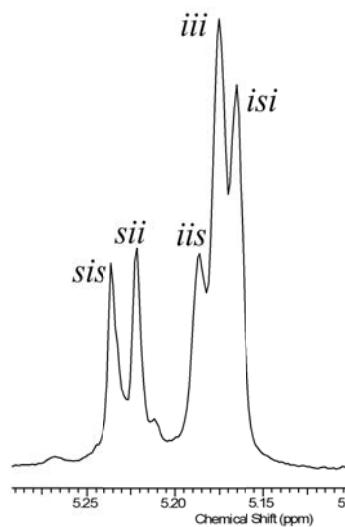


Fig. S18 $^1\text{H}\{^1\text{H}\}$ -NMR spectrum of the polymerization of *rac*-lactide using **Zr-4**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

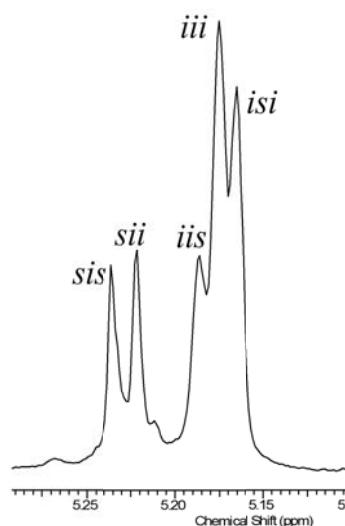


Fig. S19 ¹H{¹H}-NMR spectrum of the polymerization of *rac*-lactide using **Ti-4**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, C_6D_6 .

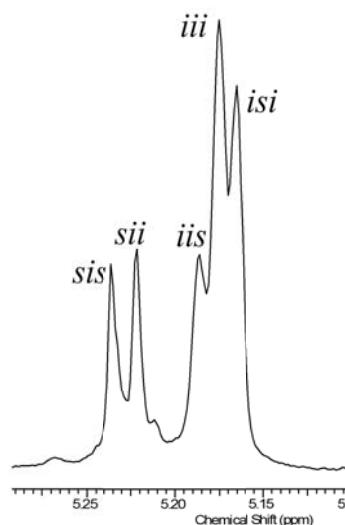


Fig. S20 ¹H{¹H}-NMR spectrum of the polymerization of *rac*-lactide using **Ti-4a**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, C_6D_6 .

2.3 Polymerization of L-lactide

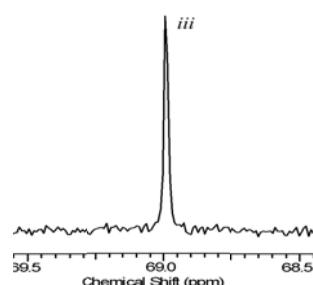


Fig. S21 $^{13}\text{C}\{\text{H}\}$ -NMR spectrum of the polymerization of L-lactide using **Zr-1**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

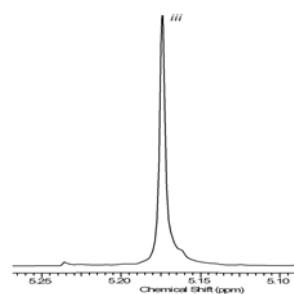


Fig. S22 $^1\text{H}\{\text{H}\}$ -NMR spectrum of the polymerization of L-lactide using **Zr-4**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6 .

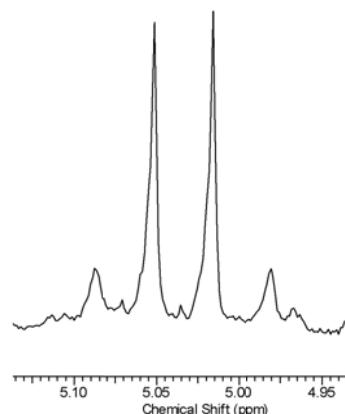


Fig. S23 ^1H NMR spectrum of the polymerization of L-lactide using **Ti-4**, $[\text{LA}]_0/[\text{init}]_0 = 100$, $[\text{LA}]_0 = 0.520 \text{ M}$, $T = 100^\circ\text{C}$, C_6D_6

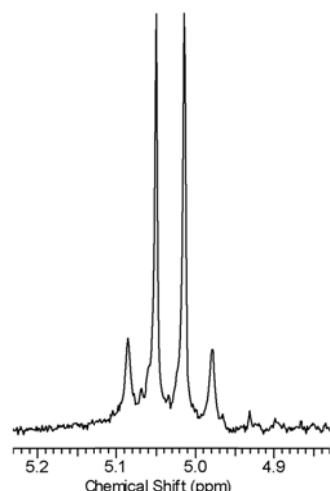


Fig. S24 ¹H NMR spectrum of the polymerization of L-lactide using **Ti-4a**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, C_6D_6 .

3 Kinetic of polymerization

3.1 Kinetic of polymerization of *rac*-lactide

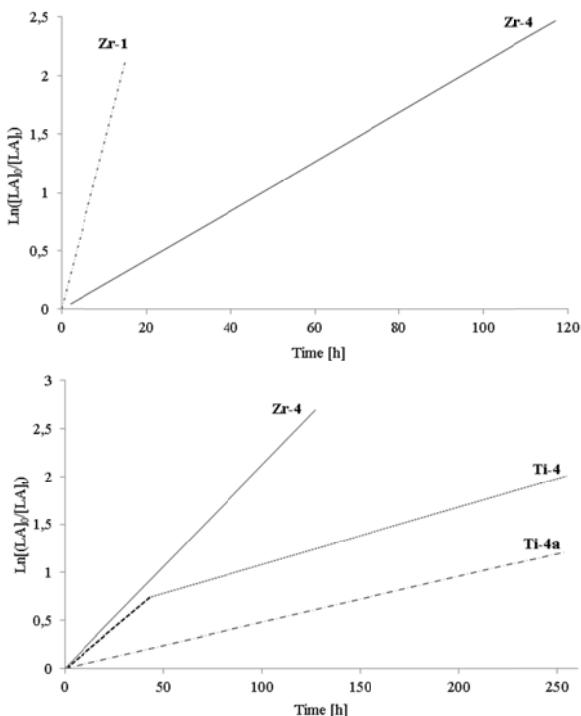


Fig. S25 Semilogarithmic plots of *rac*-lactide conversion vs. time, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL): a) using zirconium complexes: **Zr-1** ($k_{\text{obs}} = 142 \times 10^{-3}\text{ h}^{-1}$), and **Zr-4** ($k_{\text{obs}} = 21 \times 10^{-3}\text{ h}^{-1}$), and b) using the flexible ligand **4**: **Ti-4** ($k_{\text{obs}} = 14 \times 10^{-3}\text{ h}^{-1}$) and ($k_{\text{obs}} = 5 \times 10^{-3}\text{ h}^{-1}$), and **Ti-4a** ($k_{\text{obs}} = 5 \times 10^{-3}\text{ h}^{-1}$).

3.2 Kinetic of polymerization of L-lactide

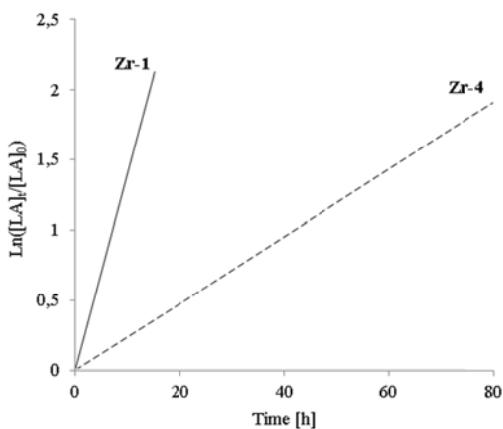


Fig. S26 Semilogarithmic plots of L-lactide conversion vs. time, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL): **Zr-1** ($k_{\text{obs}} = 164 \times 10^{-3}\text{ h}^{-1}$), and **Zr-4** ($k_{\text{obs}} = 24 \times 10^{-3}\text{ h}^{-1}$).

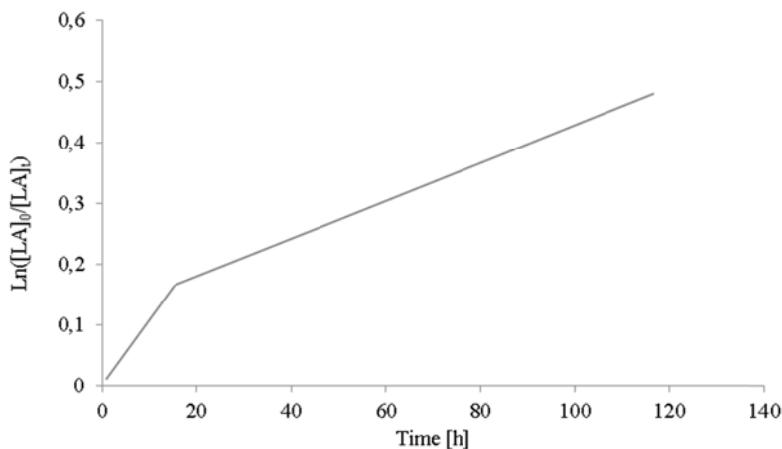


Fig. S27 Semilogarithmic plots of L-lactide conversion vs. time, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL) using **Ti-4** ($k_{\text{obs}} = 3 \times 10^{-3}\text{ h}^{-1}$).

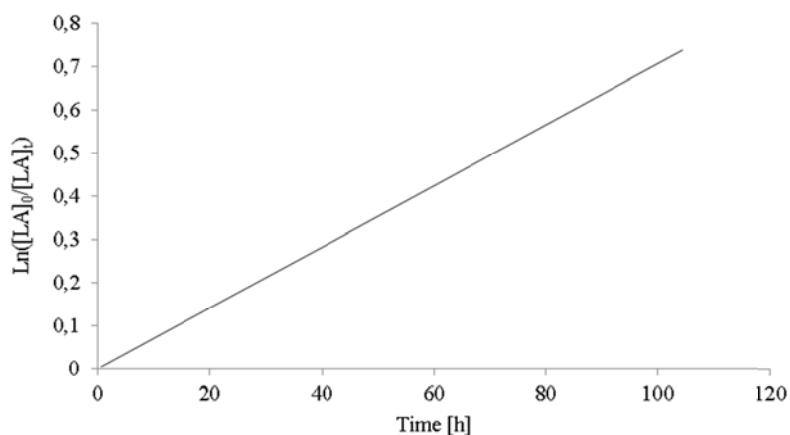


Fig. S28 Semilogarithmic plots of L-lactide conversion vs. time, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL) using **Ti-4a** ($k_{\text{obs}} = 7 \times 10^{-3}\text{ h}^{-1}$).

4 Polymerization Data

4.1 Tables of polymerization of *meso*-lactide

Table S1 Polymerization of *meso*-lactide using **Zr-1**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	1	12
2	2	22
3	4	47
4	8	77
5	13	88
6	15	91

Table S2 Polymerization of *meso*-lactide using **Zr-2**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	0.5	8
2	1	20
3	2	41
4	4	67

Table S3 Polymerization of *meso*-lactide using **Zr-3**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	0.5	10
2	1	21
3	2	53
4	4	89
5	4.5	92
6	5	93

Table S4 Polymerization of *meso*-lactide using **Zr-4**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	0.5	1
2	2.67	5
3	4	9
4	16	44
5	32	73
6	48	84

Table S5 Polymerization of *meso*-lactide using **Ti-4**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time/ hours	Conversion/ %
1	4	7
2	8	14
3	72	67
4	88	78
5	103	83
6	121	85
7	137	85

Table S6 Polymerization of *meso*-lactide using **Ti-4a**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	4	2
2	8	5
3	72	48
4	88	56
5	103	63
6	121	70
7	137	74

4.2 Tables of polymerization of *rac*-lactide

Table S7 Polymerization of *rac*-lactide using **Zr-1**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	1	10
2	2	21
3	4	48
4	8	72
5	15	87

Table S8 Polymerization of *rac*-lactide using **Zr-4**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	2	7
2	8	18
3	16	33
4	42.5	61
5	69	78
6	93	86
7	117	91
8	141	91

Table S9 Polymerization of *rac*-lactide using **Ti-4**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	2	3
2	8	14
3	16	24
4	42.5	42
5	93	67
6	117	73
7	141	75
8	209.5	82
9	233.5	85
10	254.5	86

Table S10 Polymerization of *rac*-lactide using **Ti-4a**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	2	1
2	8	6
3	16	7
4	42.5	21
5	69	30
6	93	40
7	117	47
8	141	54
9	209	63
10	233.5	65
11	254.5	69

4.3 Tables of polymerization of L-lactide

Table S11 Polymerization of L-lactide using **Zr-1**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520\text{ M}$, $T = 100\text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	1	12
2	2	22
3	4	47
4	8	77
5	13	88
6	15	91

Table S12 Polymerization of L-lactide using **Zr-4**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	4	9
2	12	16
3	32	50
4	56	76
5	80	85
6	104.5	88

Table S13 Polymerization of L-lactide using **Ti-4**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	0	0
2	4	0,05
3	8	0,08
4	16	0,15
5	24	0,18
6	91	0,37

Table S14 Polymerization of L-lactide using **Ti-4a**, $[LA]_0/[init]_0 = 100$, $[LA]_0 = 0.520 \text{ M}$, $T = 100 \text{ }^\circ\text{C}$, benzene- d_6 (0.5 mL)

Entry	Time [h]	Conversion [%]
1	4	3
2	12	10
3	32	21
4	56	29
5	80	44
6	104.5	53