N-alkyl urea peptoid oligomers and its polymer conjugate form fibers in

solutions

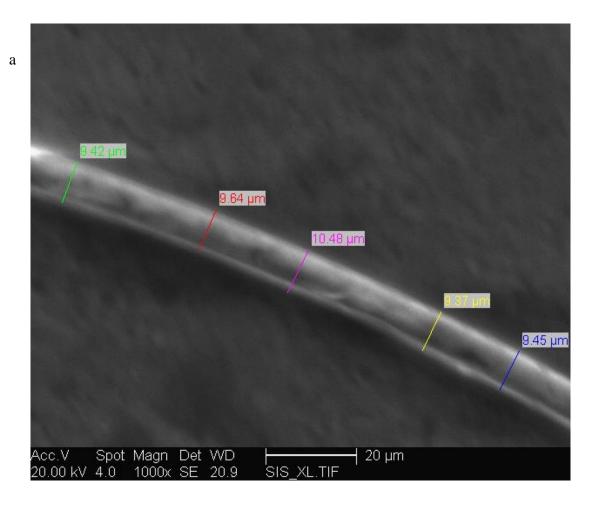
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Supporting Information

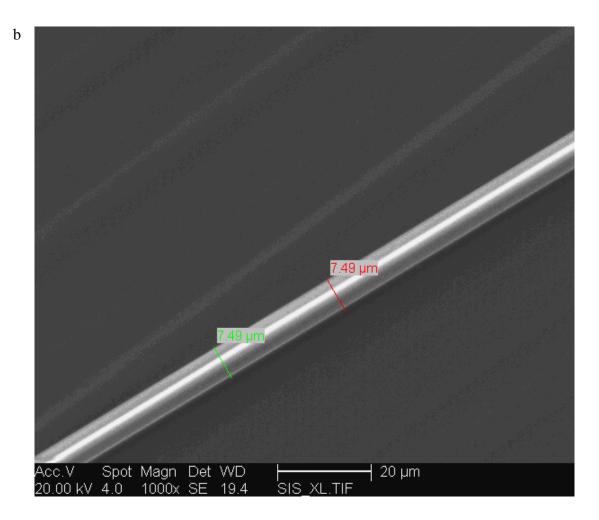
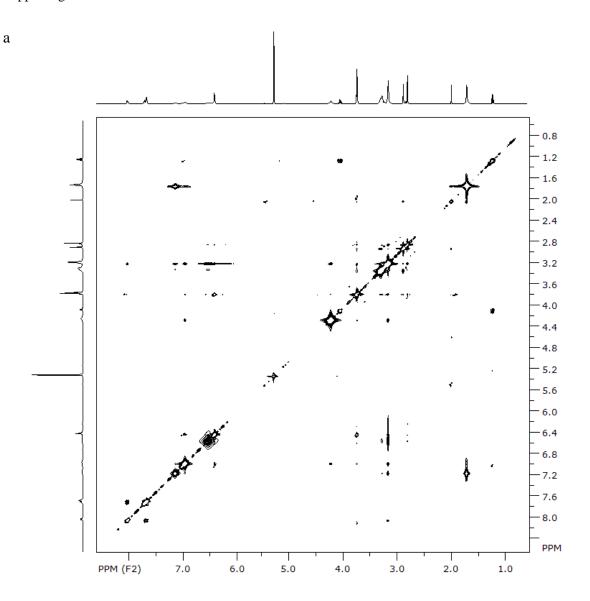
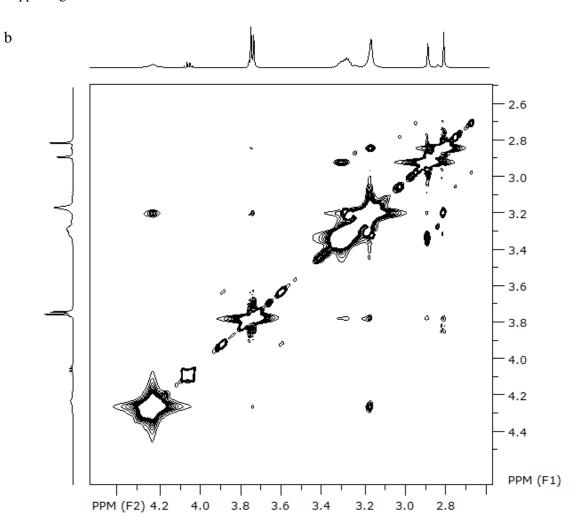


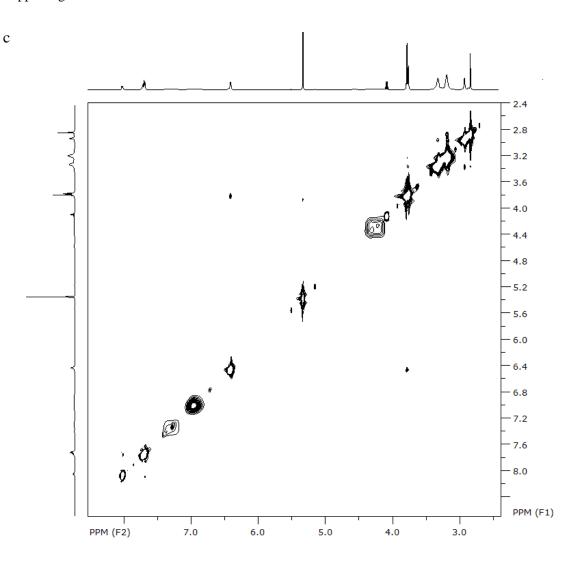
Figure S1 SEM images of (a) measurement of the diameter of a single fiber of 1 in DMF and (b) measurement of the diameter of a single fiber formed from compound 3 in H_2O .

NOESY Spectra

Compound **1** was dissolved in CD_2Cl_2 at a concentration of 10 mg/ml (7.7 μ M) in a sealed NMR tube. An equimolar amount of ethyl acetate to **1** was added to the NMR tube purposely to test the stability of the hydrogen bonding between molecules. No effect was seen. The full NOESY spectrum of compound **1** was shown in Figure S2a.







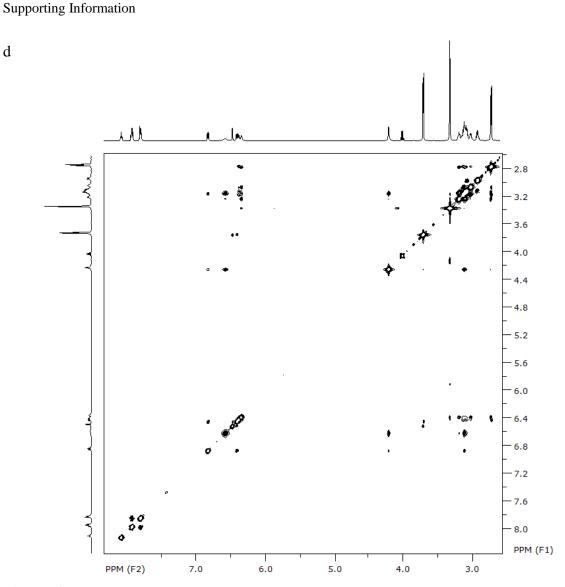


Figure S2. (a) NOESY spectrum of compound **1** in CD_2Cl_2 ; (b) magnification of the coupling of -N-C<u>H</u>₂- and -N-C<u>H</u>₃ groups on urea-peptoid backbone; (c) NOESY spectrum of compound **1** in CD_2Cl_2 at concentration of 4 mg/ml; (d) NOESY spectrum of compound **1** in d_6 -DMSO

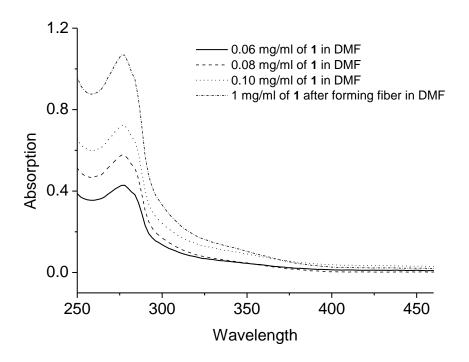


Figure S3. UV-Vis absorption spectrum of urea-peptoid **1** at various concentrations. The absorption of the same solution decreased after forming fiber because of the concentration decreased since **1** formed fibers and fibers were out of solution.

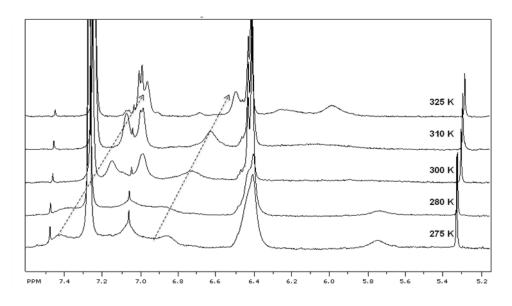


Figure S4. Variable temperature ¹H NMR spectra of oligomer **8** in CDCl₃ at 275 (bottom spectrum), 280, 300, 310, and 325 K (top spectrum). Dashed arrows have been added to guide the eye.

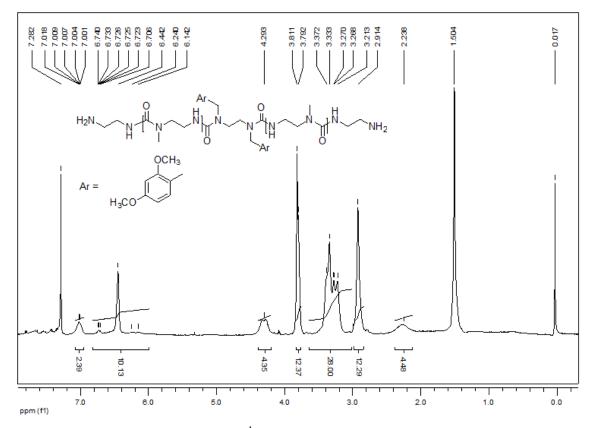
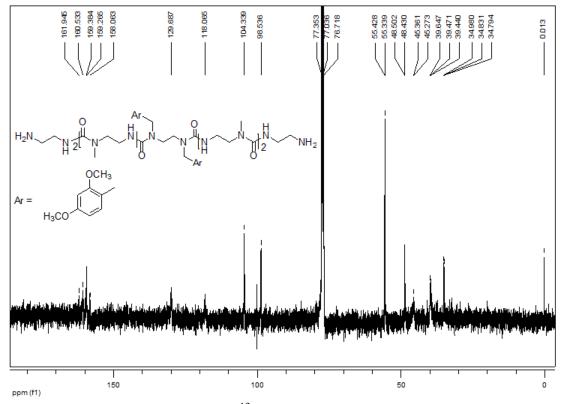


Figure S5. ¹H NMR of Compound 2





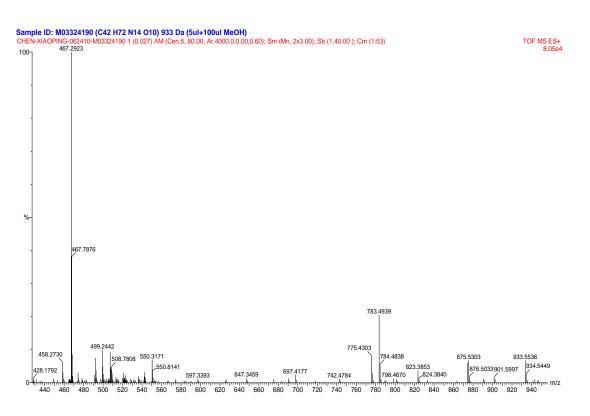


Figure S7. Mass Spectrum of Compound 2

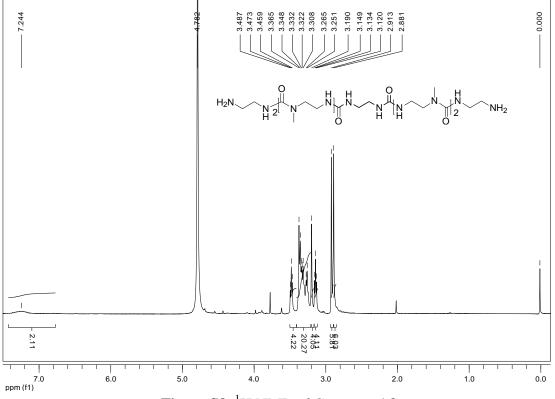
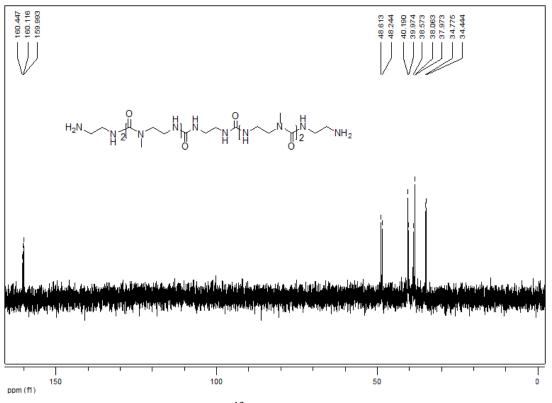
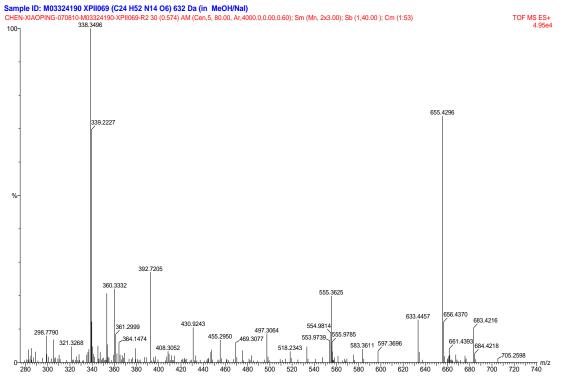
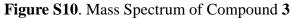


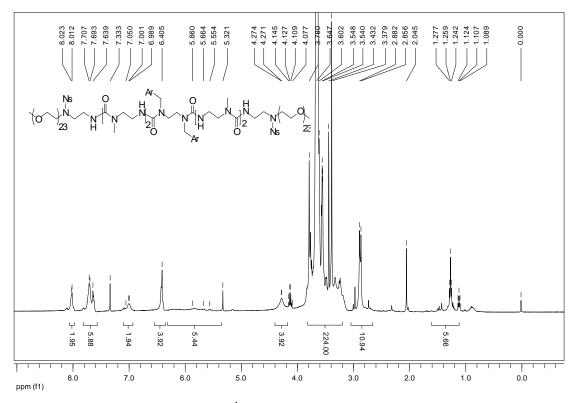
Figure S8. ¹H NMR of Compound 3













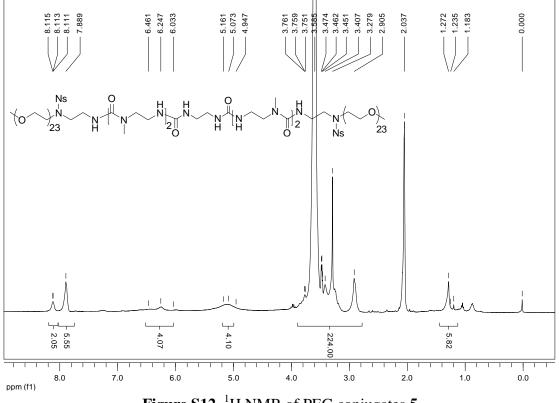






Figure S13. SEM image of fibers resulting from the self-assembly of conjugate **5** in water at lower magnification.

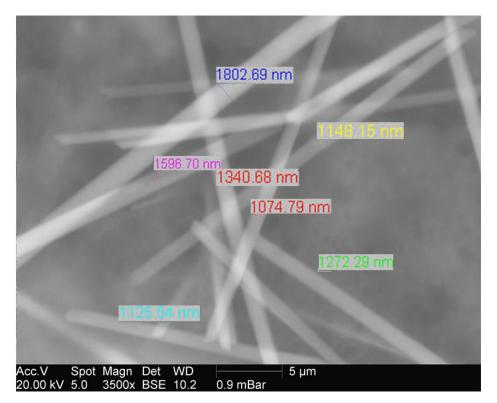


Figure S14. Diameters of fibers resulting from the self-assembly of conjugate 5 in

water.