

Supporting Information

A Facile Synthesis of Clickable and Acid-Cleavable PEO for Acid-Degradable Block Copolymers

Kotaro Satoh,^{†,§} Justin E. Poelma,[†] Luis M. Campos,^{†,‡} Brian Stahl and Craig J. Hawker^{*†,‡}

[†]Materials Research Laboratory and Materials Department, University of California, Santa Barbara, California 93106, USA

[‡]Department of Chemistry and Biochemistry, University of California, Santa Barbara, California 93106, USA

[§]Department of Applied Chemistry, Graduate School of Engineering, Nagoya University, Furo-cho, Chikusa-ku, Nagoya 464-8603, Japan

*To whom correspondence should be addressed: E-mail: hawker@mrl.ucsb.edu

Contents:

Figure S1. GPC curves during CuAAc click reaction between PEO-acetal-N₃ and and PEO-alkyne, followed by intentional cleavage of acetal junction.

Figure S2. ¹H NMR spectra of PEOs obtained in the same experiments as for Figure S1.

Figure S3. GPC curves for removal PEO segment from PEO-acetal-PS diblock copolymer, followed by Cu-catalyzed ATRP of MMA.

Figure S4. ¹H NMR spectra of PS-OH, PS-Br, and PS-*b*-MMA block copolymer obtained in the same experiments as for Figure S3.

Figure S5. AFM images for morphologies of solvent-annealed films of the PEO-acetal-PS diblock copolymer (A), after exposure to TFA vapor for 4 h (B), nanoporous film after immersion in methanol for 4h (C).

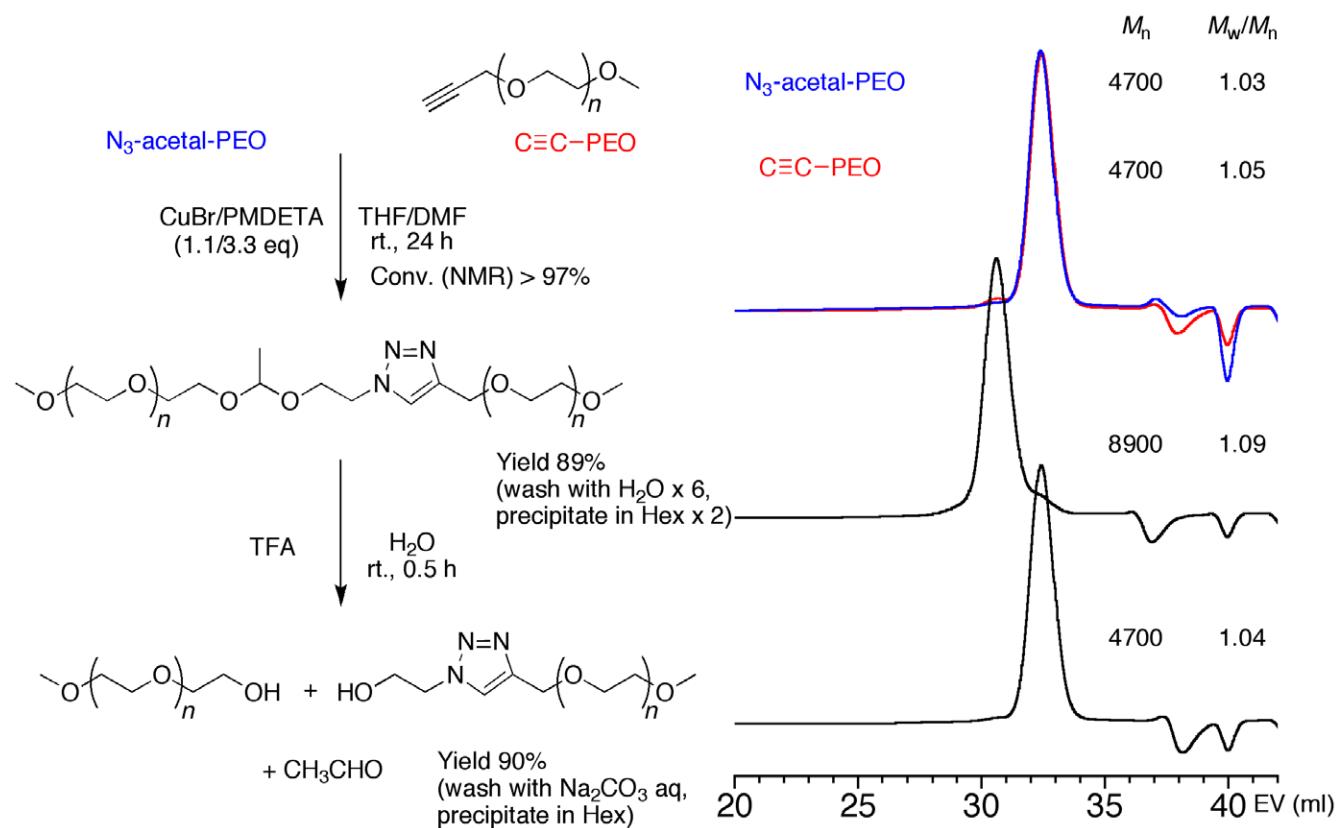


Figure S1. GPC curves during CuAAC click reaction between PEO-acetal-N₃ and and PEO-alkyne:
[PEO-acetal-N₃]₀ = [PEO-alkyne]₀ = 0.21 mmol (1.1 g), [CuBr]₀ = 0.23 mmol, [PMDETA]₀ = 0.69 mmol in 30 mL of THF/DMF (1/1 v/v) at r.t. for 24 h, and followed by intentional cleavage of acetal junction: [PEO-acetal-PEO]₀ = 0.02 mmol (0.2 g) and TFA (1.0 mL) in H₂O at r.t. for 0.5 h.

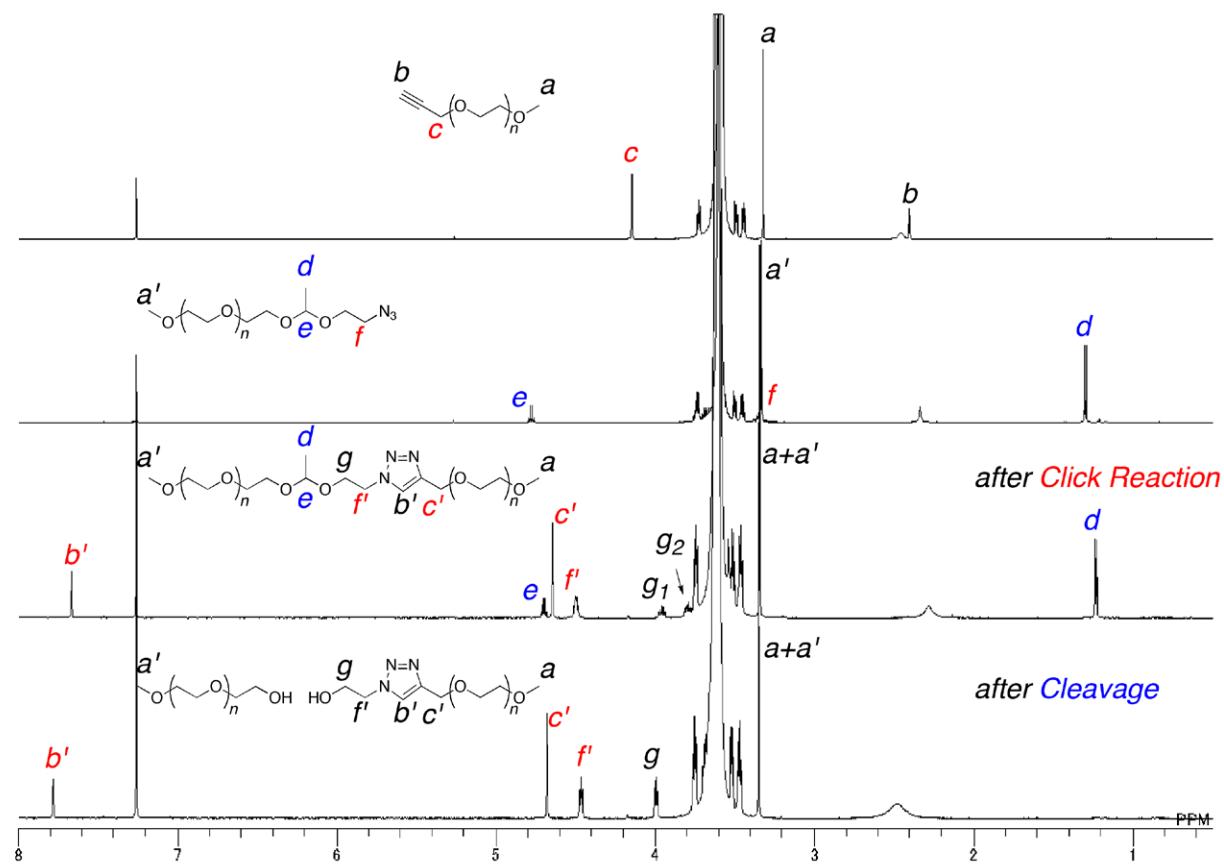


Figure S2. ¹H NMR spectra (CDCl₃, r.t.) of PEOs obtained in the same experiments as for Figure S1.

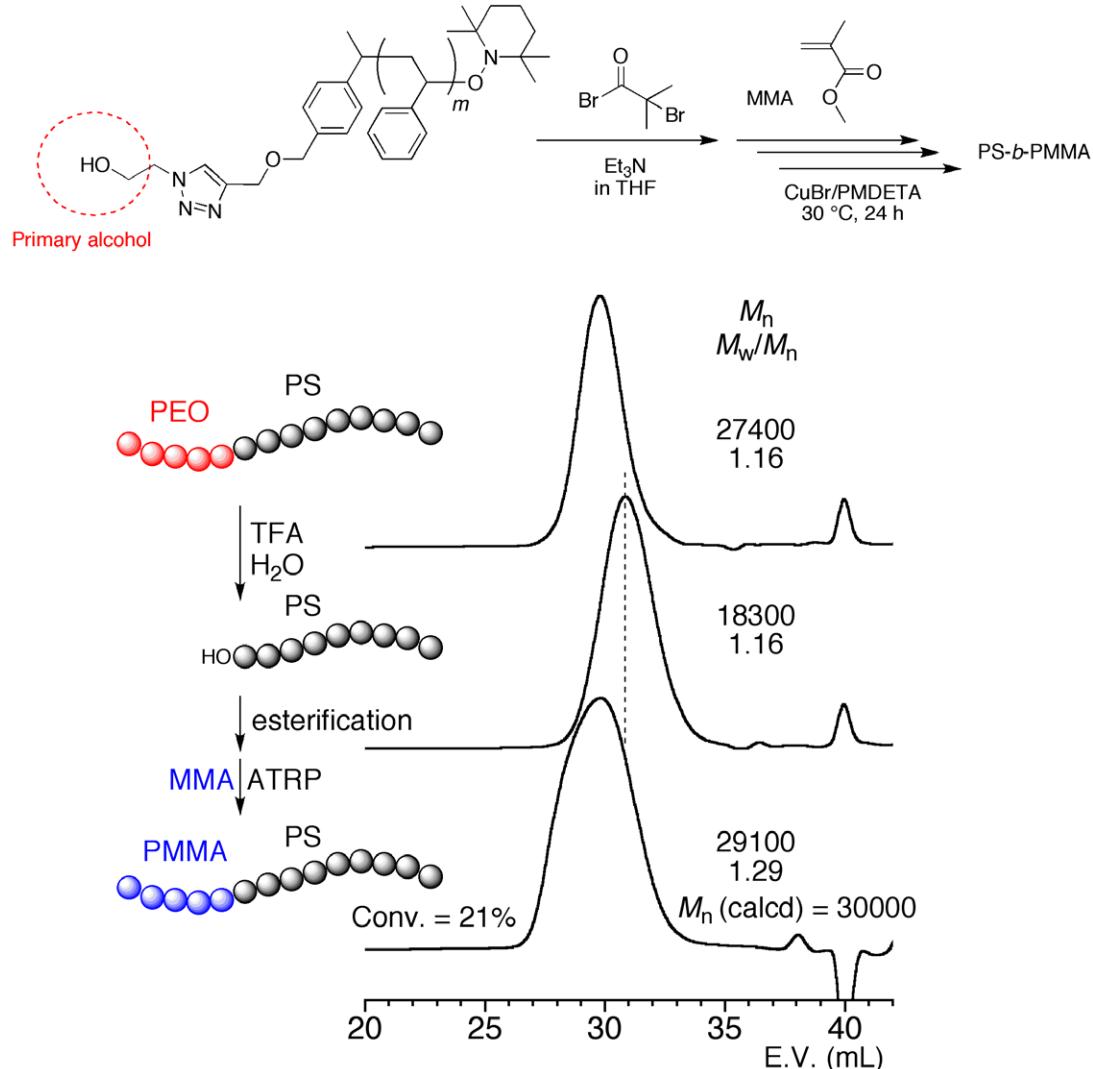


Figure S3. GPC curves for removal PEO segment from PEO-acetal-PS diblock copolymer, followed by Cu-catalyzed ATRP of MMA. For esterification: $[\text{PS-OH}]_0 = 9.0 \mu\text{mol}$ (177 mg), $[2\text{-bromoisobutyryl bromide}]_0 = 0.81 \text{ mmol}$, $[\text{Et}_3\text{N}]_0 = 7.2 \text{ mmol}$ in THF (5mL) at 0 °C for 15 h. For ATRP: $[\text{PS-Br}]_0 = 4.0 \mu\text{mol}$ (79 mg), $[\text{MMA}]_0 = 2.0 \text{ mmol}$, $[\text{CuBr}]_0 = 0.04 \text{ mmol}$, $[\text{PMDETA}]_0 = 0.08 \text{ mmol}$ in toluene (1.8 mL) at 30 °C for 24 h (MMA conversion = 21%).

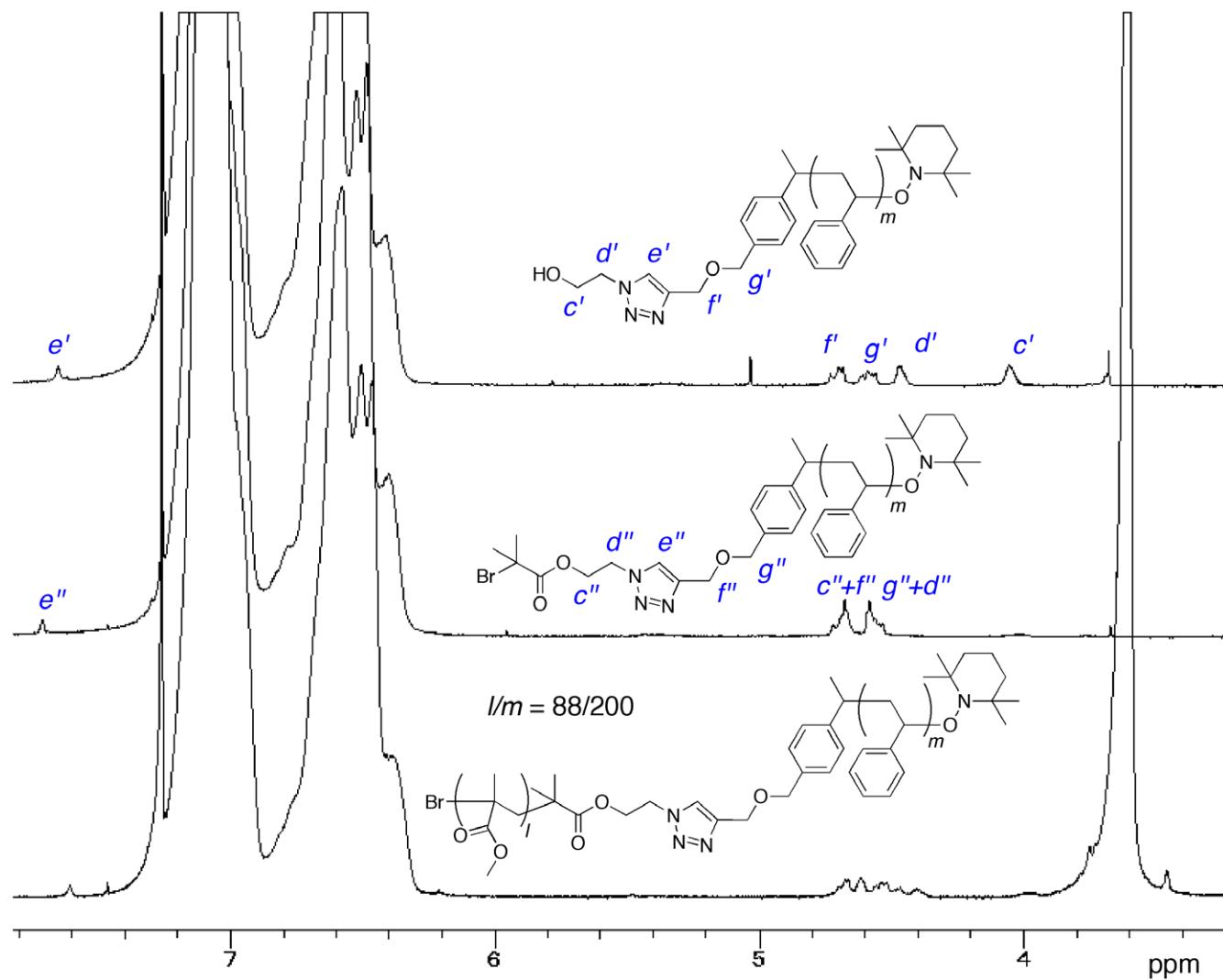


Figure S4. ¹H NMR spectra (CDCl₃, r.t.) of PS-OH, PS-Br, and PS-*b*-MMA block copolymer obtained in the same experiments as for Figure S3.

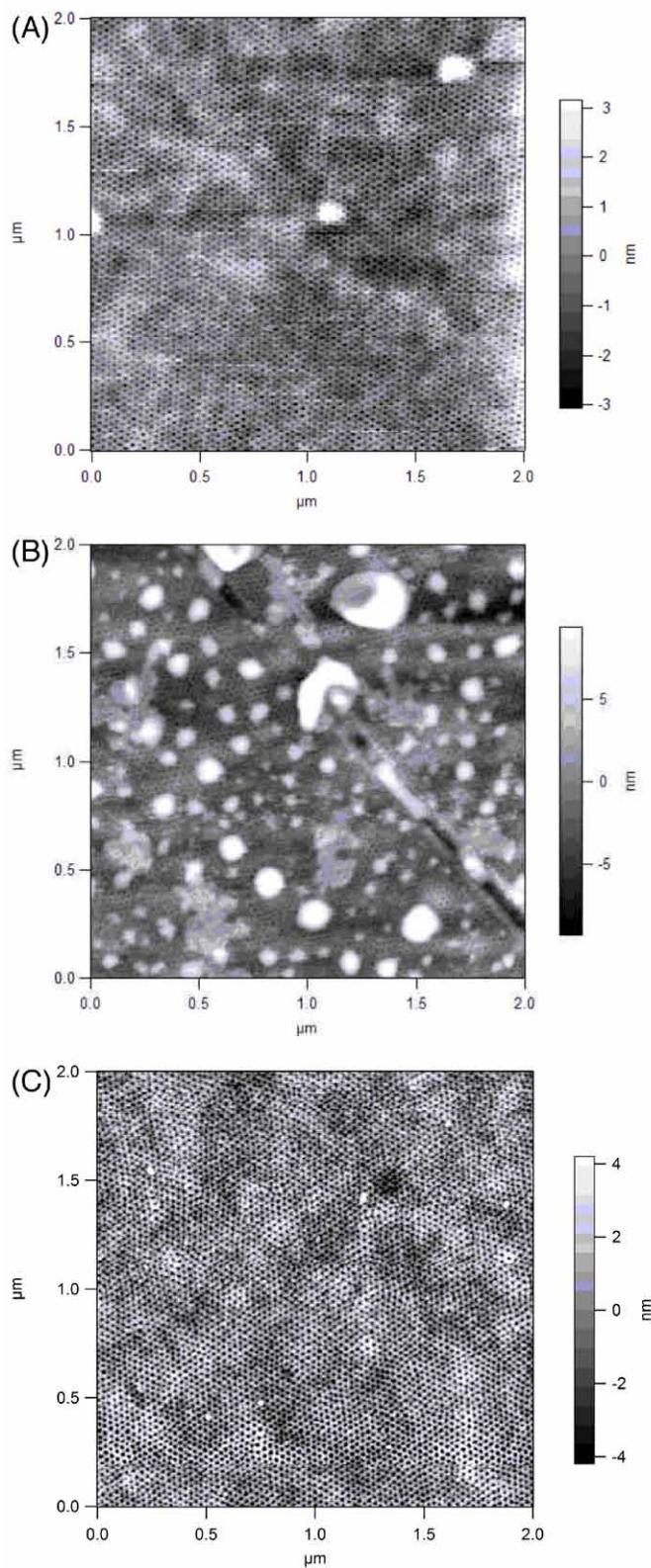


Figure S5. AFM images for morphologies of solvent-annealed films of the PEO-acetal-PS diblock copolymer (A), after exposure to TFA vapor for 4 h (B), and nanoporous film after immersion in methanol for 4h (C).