

Supplementary Information

Synthesis of microporous polymers by Friedel-Crafts reaction of 1-bromoadamantane with aromatic compounds and their surface modification

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Measurements

N₂ and CO₂ uptake amounts were measured by using a Belsorp-Max (BEL Japan, Inc.) apparatus. Ultra-high purity grade CO₂ and N₂ were used for all adsorption measurements. Solid state NMR measurements were carried out on a Bruker 400 Solid/Micro-Imaging High Resolution NMR with resonance frequencies of 100 MHz. A 4 mm triple resonance CP/MAS NMR probe with a rotation frequency of 7 kHz was used. Fourier transform infrared (FT-IR) measurements were recorded on a PERKIN ELMER Spectrum GX I using KBr pellets. Elemental analyses were performed by an EA 1110 (CE Instrument) and a Flash 1112 (Thermo Electron corporation).

Synthesis of polymers 1-3

A typical procedure is as follows: To a solution of 1-bromoadamantane (1g, 4.6 mmol) in nitrobenzene (50 ml) were added tert-butyl bromide (1.9 g, 13.8 mmol), aluminum chloride (2.4 g, 18.4 mmol) and an aromatic compound (benzene, biphenyl, or terphenyl, 9.2 mmol). After stirring for 24 h at 150 °C, the precipitated polymers were isolated by filtration, washed with methanol, acetone and DMF. Further purification was done by Soxhlet extraction with

methanol. For polymer **1**: ^{13}C CP/MAS NMR δ 144, 122, 115, 34, 25; IR (cm^{-1} , KBr pellet) 3034, 2930, 2900, 2852, 2210. For polymer **2**: ^{13}C CP/MAS NMR δ 144, 134, 121, 119, 34, 25; IR (cm^{-1} , KBr pellet) 3034, 2930, 2900, 2852. For polymer **3**: ^{13}C CP/MAS NMR δ 144, 134, 121, 119, 34, 25; IR (cm^{-1} , KBr pellet) 3034, 2930, 2900, 2852.

Post-modification of polymers 1-3 with nitrobenzoyl chloride

A typical procedure is as follows: Polymer **1** (0.2 g) was stirred in nitrobenzene (50 ml) for 1 h at 150 °C. To the mixture was added aluminum chloride (0.20 g, 1.50 mmol) and 4-nitrobenzoyl chloride (0.30 g, 1.60 mmol). After stirring for 24 h at 150 °C, the polymer was isolated by filtration, washed with methanol, acetone and DMF. Further purification was done by Soxhlet extraction with methanol. For polymer **1'**: ^{13}C CP/MAS NMR δ 144, 122, 115, 34, 25; IR (cm^{-1} , KBr pellet) 3034, 2930, 2900, 2852, 1671, 1526, 1350. For polymer **2'**: ^{13}C CP/MAS NMR δ 144, 134, 121, 119, 34, 25; IR (cm^{-1} , KBr pellet) 3034, 2930, 2900, 2852, 1671, 1526, 1350. For polymer **3'**: ^{13}C CP/MAS NMR δ 144, 134, 121, 119, 34, 25; IR (cm^{-1} , KBr pellet): 3034, 2930, 2900, 2852, 1671, 1526, 1350.

Post-modification of polymer 3 with benzyl chloride

Polymer **3** (0.20 g) was stirred in nitrobenzene (50 ml) for 1 h at 120 °C. To the mixture was added aluminum chloride (0.70 g, 5.40 mmol) and benzyl chloride (0.60 g, 5.0 mmol). After stirring for 24 h at 120 °C, the polymer was isolated by filtration, washed with methanol, acetone and DMF. Further purification was done by Soxhlet extraction with methanol. ^{13}C CP/MAS NMR δ 144, 134, 121, 119, 34, 25; IR (cm^{-1} , KBr pellet) 3034, 2930, 2900, 2852.

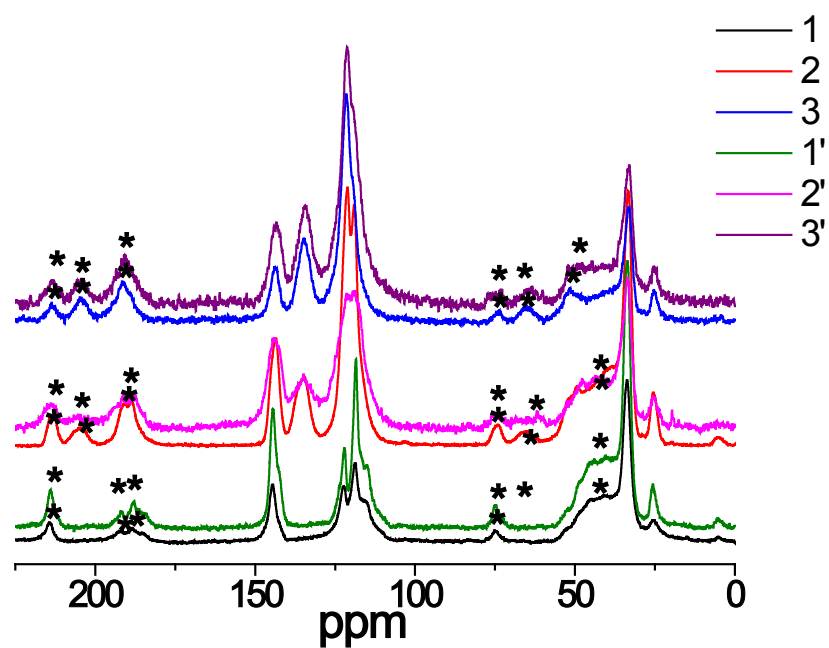


Fig. S1 ^{13}C CP/MAS NMR spectra of polymers 1-3'. The asterisk marks a rotational sideband.

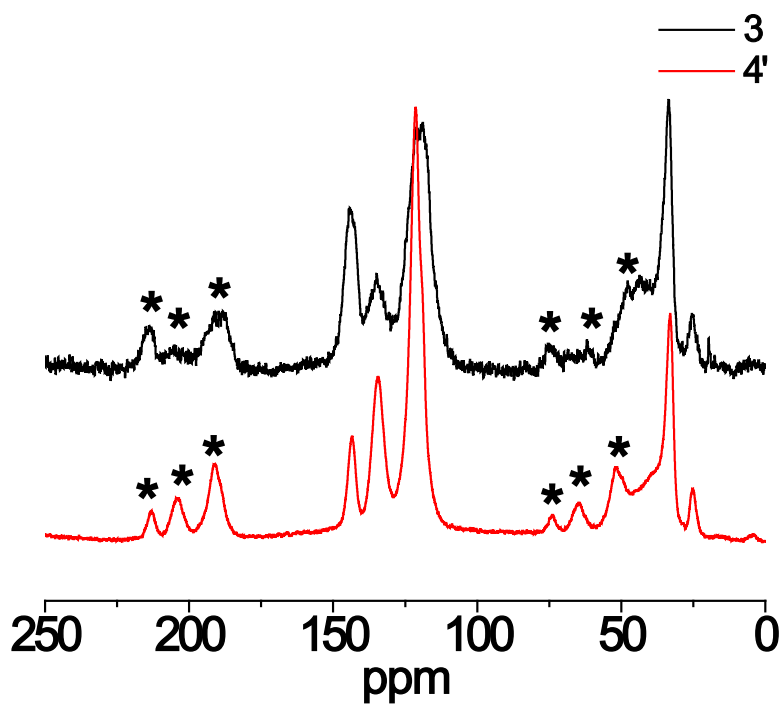


Fig. S2 ^{13}C CP/MAS NMR spectra of polymers 3 and 4'. The asterisk marks a rotational sideband.

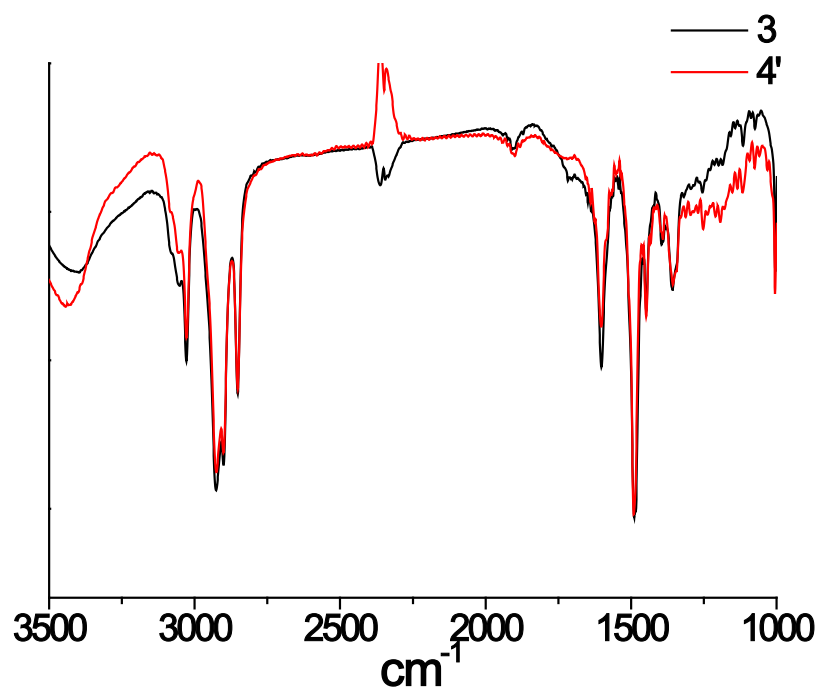


Fig. S3 FT-IR spectra of polymers **3** and **4'**.

Table S1 Elemental analysis results of polymers 1-3'

Polymers	N (%)	C (%)	H (%)
1	-	89.28	7.18
1'	0.70	86.95	6.71
2	-	88.08	6.52
2'	2.73	76.04	4.51
3	-	80.63	5.52
3'	1.82	79.41	4.80

Table S2 Pore volumes of the polymers

Polymer	^a $V_{\text{total}}/\text{cm}^3 \text{ g}^{-1}$	^b $V_{\text{micro}}/\text{cm}^3 \text{ g}^{-1}$	$V_{\text{micro}}/V_{\text{total}}$
1	2.74	0.64	0.23
1'	2.49	0.62	0.25
2	2.00	0.46	0.23
2'	1.94	0.43	0.22
3	0.64	0.28	0.43
3'	1.08	0.40	0.37
4'	0.76	0.31	0.40

^aTotal pore volume determined from the N_2 isotherm at $P/P_0 = 0.995$. ^bMicropore volume determined from the N_2 isotherm at $P/P_0 = 0.050$.

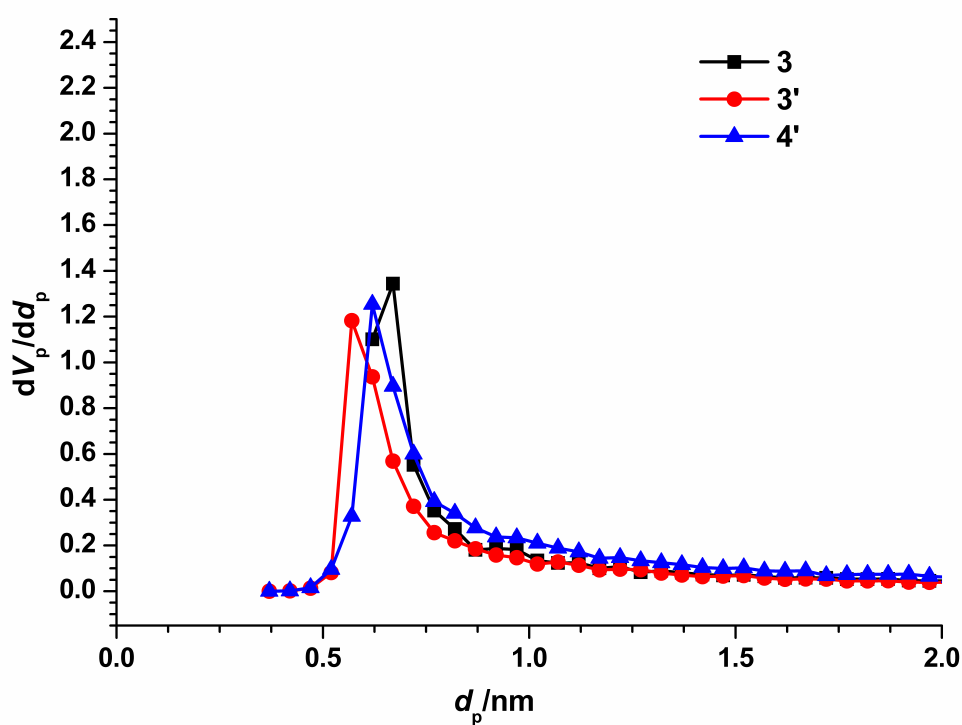


Fig. S4 Micropore size distribution of polymers 3, 3' and 4' calculated by the Horvath-Kawazoe method.