

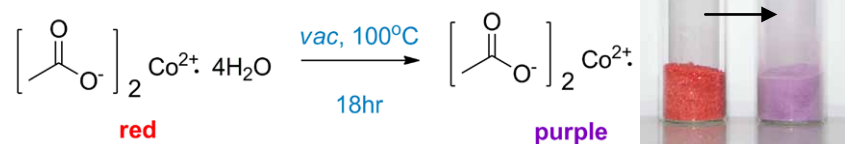
# Biological surface modification by 'thiol-ene' addition of polymers synthesised by catalytic chain transfer polymerisation (CCTP)

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## Supplementary Information

### CoBF Synthesis

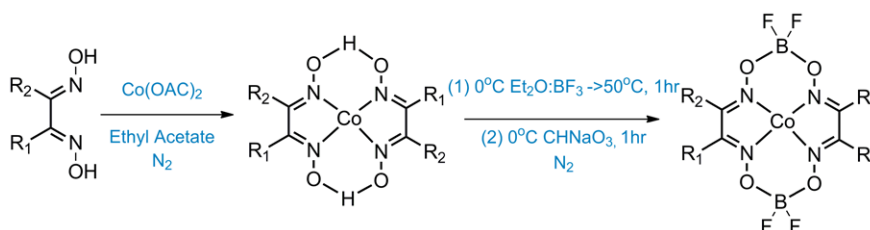
#### Preparation of anhydrous cobalt acetate



#### S1 Reaction scheme for anhydrous cobalt acetate preparation

Cobalt (II) acetate tetrahydrate was dehydrated to keep the water content in the synthesis of CoBF to a minimum. The red/pink powder changed to a purple colour once dehydration was complete.

#### Synthesis of CoBF (dimethanol complex)



#### S2 Reaction scheme for CoBF synthesis

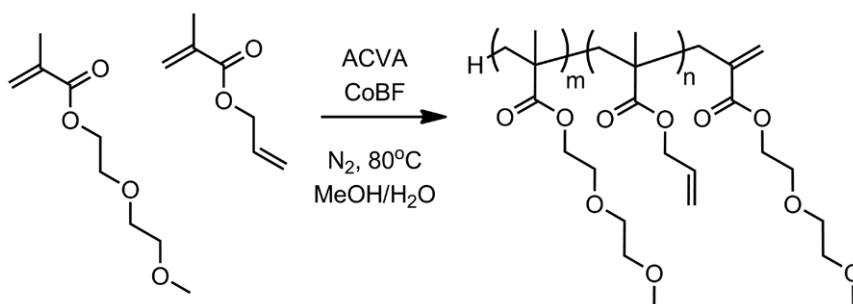
The reaction was kept under a dinitrogen atmosphere throughout preparation and isolation. Purification was not carried out under nitrogen therefore the time that the catalyst remained wet from solvents was kept to a minimum. Solvents were purged with nitrogen prior to use.

Dimethyl glyoxime (2.02 g, 17.3 mmol) and anhydrous cobalt acetate (2.16 g, 8.7 mmol) were placed in a round bottom flask with a magnetic stirrer bar and purged with nitrogen. 100 mL deoxygenated ethyl acetate was cannulated into the round bottom flask and cooled to

0 °C. Boron trifluoride diethyl etherate (5.89 mL, 47.8 mmol) was added slowly using a degassed syringe. The resulting brown solution was heated to 50 °C for 1 hour. Sodium hydrogen carbonate (1.61 g, 19.1 mmol) was added under a flow of nitrogen and the reaction was stirred for 1 hour at 0 °C. The brown solution was then filtered, washed with water (2 x 50 mL), methanol (2 x 15 mL) and dried under vacuum. The filtrate was refrigerated to obtain crystals as the product is readily soluble in water and methanol. Yield= 1.64 g (42 %)

IR  $\nu$  (cm<sup>-1</sup>): 3591 + 3521 O-H (from Ligand), 1623 (C=N), 1172 (N-O), 1094 (N-O), 946 (B-F), 825 (B-O), 730 (C=N-O), 506 (Co-N). Mass spectroscopy (m/z): 407.9 Da (Na<sup>+</sup>).

### Polymer synthesis example: CCTP of di(ethylene glycol) methyl ether methacrylate-co-allyl methacrylate (5% AMA incorporation)



**S3 Polymerisation reaction scheme**

### 10 % incorporation

The same reaction procedure was carried out as explained above with the only changes shown in the table below:

Compound	Mass (g)	Moles (M)	MW (g. mol <sup>-1</sup> )
Allyl methacrylate	2.016	0.016	126

$M_n$  (GPC) = 2400 g. mol<sup>-1</sup>,  $M_w$  (GPC) = 6200 g. mol<sup>-1</sup>, PDI = 2.58, DP (NMR) = 12,  $M_n$  (NMR) = 3800 g. mol<sup>-1</sup>, conversion (NMR) = 81 %

### 20 % incorporation:

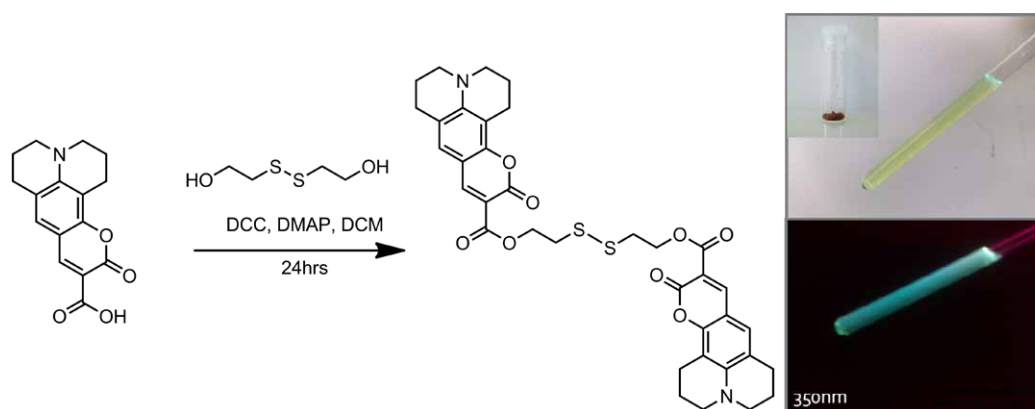
Compound	Mass (g)	Moles (M)	MW (g. mol <sup>-1</sup> )
Allyl methacrylate	4.032	0.032	126

$M_n$  (GPC) = 6100 g. mol<sup>-1</sup>,  $M_w$  (GPC) = 14 500 g. mol<sup>-1</sup>, PDI = 2.36, DP (NMR) = 14,  $M_n$  (NMR) = 4400 g. mol<sup>-1</sup>, conversion (NMR) = 94 %

No LCST as the polymer was not water soluble.

With targeted AMA incorporation of 5, 10 and 20%, the obtained copolymers contained 2, 5 and 14% respectively, as determined *via* <sup>1</sup>H NMR.

### Synthesis of coumarin disulfide dye



**S4 Reaction scheme for coumarin disulfide synthesis**