

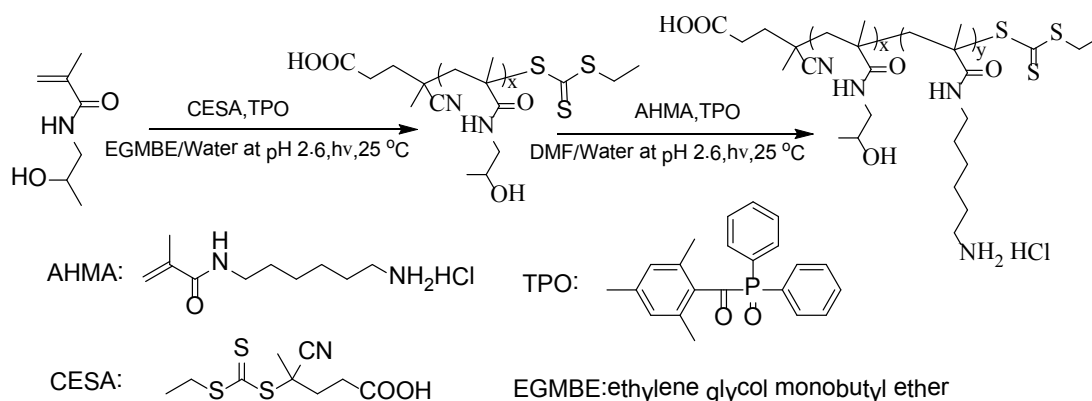
Subcomponent self-assembly of polymer chains based on dynamic and geometrical coordination diversity of the first row transition metal ions

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Electronic Supplementary Information

Synthesis of the targeted PHPMA-*b*-PAHMA copolymers

As shown in **Scheme S1**, poly(*N*-2-hydroxypropyl methacrylamide)-*block*-poly(*N*-6-aminohexyl methacrylamide hydrochloride) (PHPMA₆₉-*b*-PAHMA₇₀) was synthesized *via* aqueous RAFT polymerisation of HPMa monomer, and then the chain extension of AHMA monomer using an above-synthesized PHPMA macromolecular chain transfer agent (macro-CTA) under visible light irradiation at 25 °C.



Scheme S1. Synthesis of PHPMA-*b*-PAHMA copolymers *via* aqueous RAFT polymerization under visible light irradiation at 25 °C.

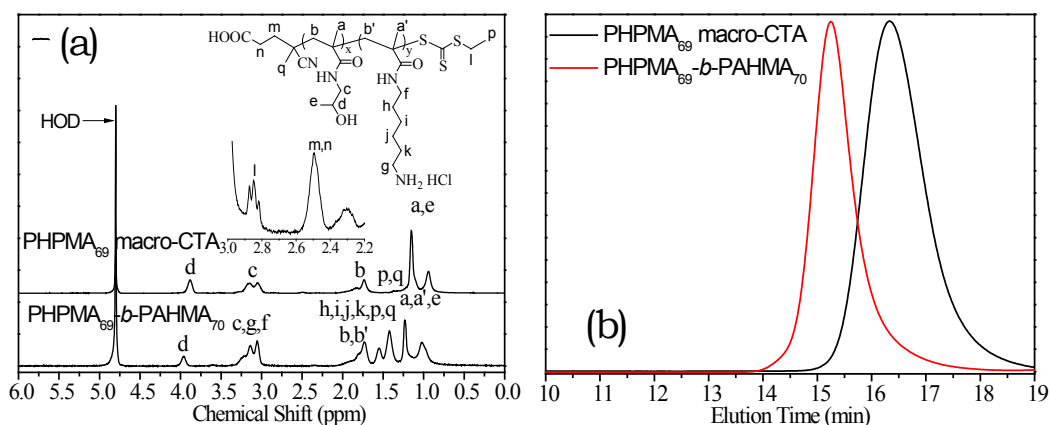


Fig. S1 ¹H NMR spectra (a) and GPC traces (b) of poly(*N*-2-hydroxypropylmethacrylamide) (PHPMA₆₉) and the corresponding block copolymer with poly(6-aminohexylmethacrylamide hydrochloride) (PHPMA₆₉-*b*-PAHMA₇₀).

As shown in **Fig. S1a**, except for HOD signal at $\delta = 4.79$ ppm, no signal of other impurities could be detected. The integral ratio of $I_d : I_c : I_b : I_{a+e}$ of upper spectrum was equal to 1 : 2 : 2 : 6,

which was consistent with the proton ratio of the targeted PHPMA. Thus, the degree of polymerisation (DP_{PHPMA}) was assessed according to **Equation S1**, where I_d is the integral of the signal of CH_3CHOH in HPMa units and I_{m+n} is the signal $\text{CH}_2\text{CH}_2\text{COOH}$ at CESA chain-ends.

$$DP_{PHPMA} = \frac{4 \times I_d}{I_{m+n}} \quad (\text{S1})$$

$$DP_{AHMA} = \frac{I_{c+g+f} - 2I_d}{4I_d} \times DP_{HPMA} \quad (\text{S2})$$

The block copolymer composition was assessed according to **Equation S2**, where I_d is the integral of signal CH_3CHOH in HPMa units, I_{c+g+f} is the total integral of signals CONHCH_2 in HPMa units and $\text{CONHCH}_2(\text{CH}_2)_4\text{CH}_2\text{NH}_2$ in AHMA units.

As shown in **Fig. S1b**, these GPC traces were excellently unimodal and symmetrical, and clearly shifted to higher molecular weight side after the chain extension, suggesting the well-controlled and living character of this polymerisation. Accordingly, the molecular structure parameters are assessed as follows. PHPMA: $DP_{HPMA} = 69$ (^1H NMR); $M_n = 9.20 \text{ kg mol}^{-1}$, $M_w/M_n = 1.21$ (GPC); Block copolymer: $DP_{AHMA} = 70$, $\text{PHPMA}_{69}\text{-}b\text{-PAHMA}_{70}$ (^1H NMR); $M_n = 12.2 \text{ kg mol}^{-1}$, $M_w/M_n = 1.12$ (GPC).

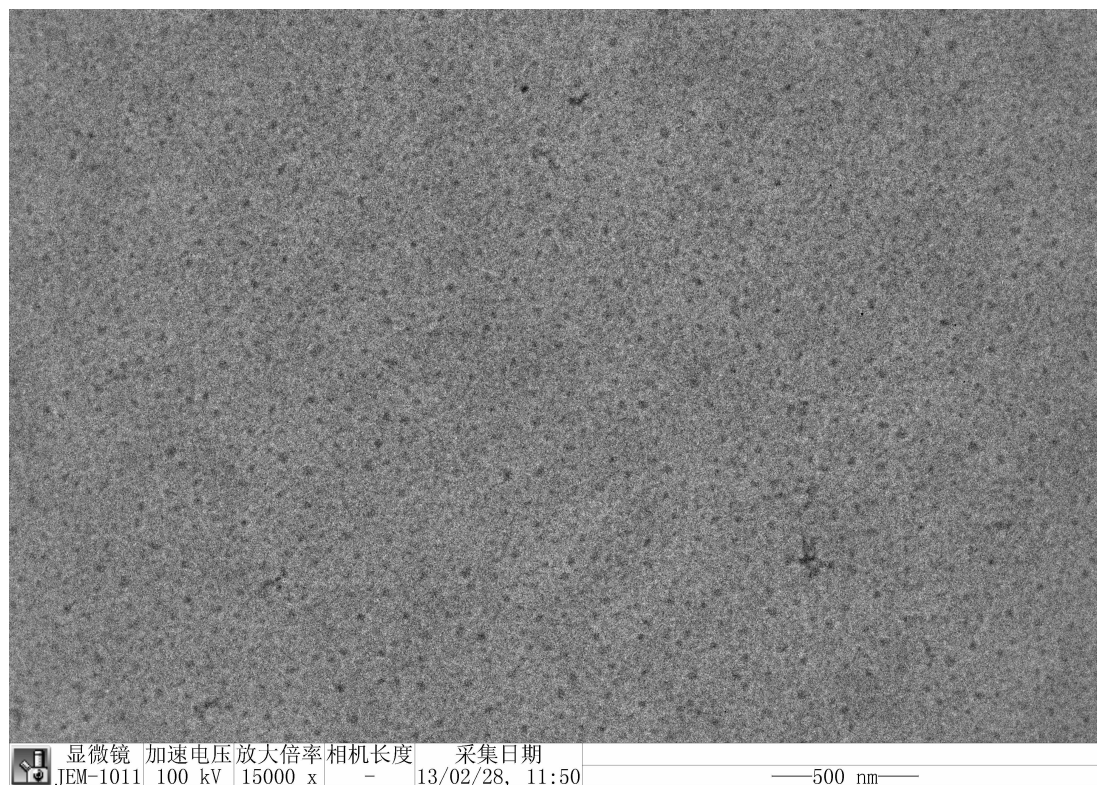


Fig. S2 TEM image of Zn(II)-bound particles formed after subcomponent self-assembly of $\text{PHPMA}_{69}\text{-}b\text{-PAHMA}_{70}$ solution (1.0 mg mL^{-1} in methanol) at a $[\text{AHMA}]_0:[\text{HMBA}]_0:[\text{Zn(II)}]_0 = 1:1:0.5$ at 25°C for 24 h.