Electronic Supplementary Material (ESI) for Polymer Chemistry

Supporting Information

Effects of surface charge on the stability of PEG-b-PCL micelles: simulate the interactions between charged micelles and plasma components

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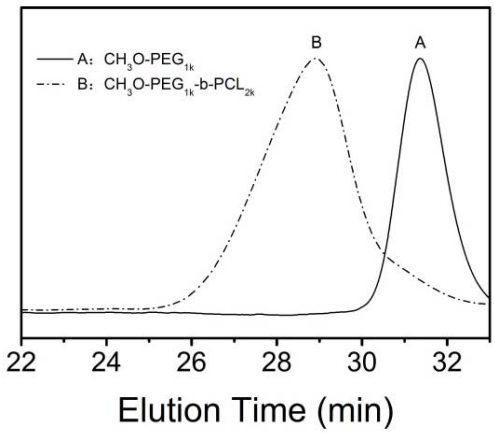

Fig. S1 GPC traces of CH_3O-PEG-OH (**A**: M_W=1000, PDI=1.05), CH_3O -PEG-b-PCL (**B**: M_W=3100, PDI=1.15) in DMF

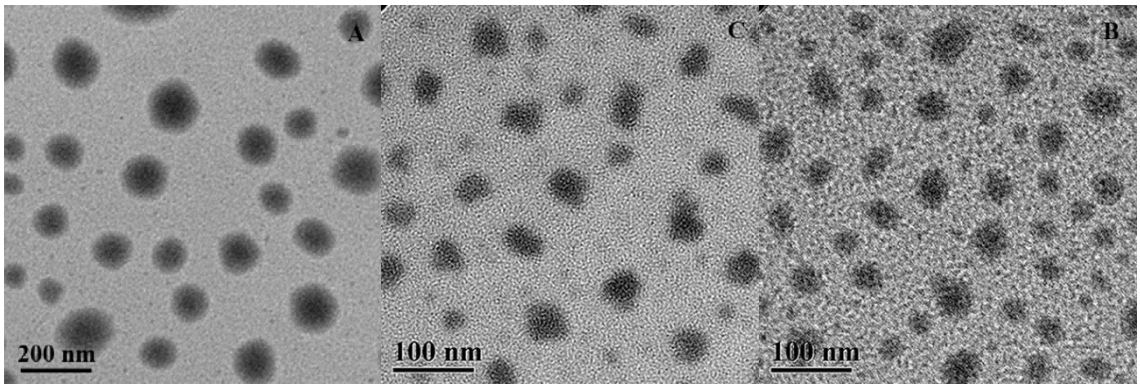

Fig. S2 TEM images of CH_3O micelles (**A**), NH_2 /COOH(1:1) mixed micelles (**B**) and COOH micelles(**C**)

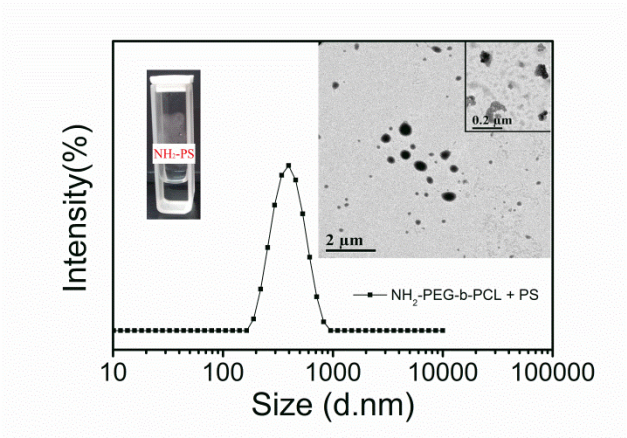

Fig. S3 DLS and TEM result of NH_2-PEG-PCL micelles after degradation at 25 ^oC (pH 7.0)

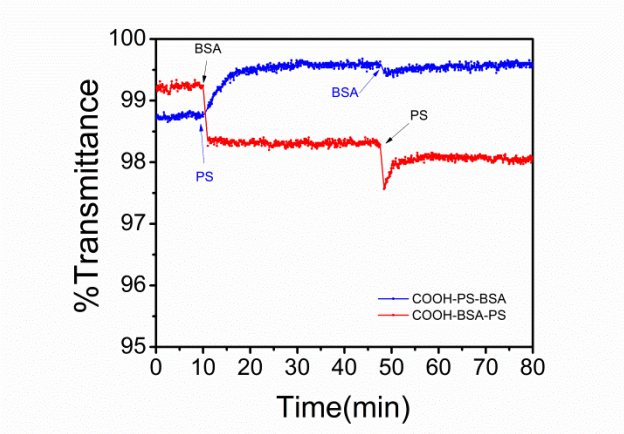

Fig. S4 Time dependence of UV transparency at 500 nm recorded for 1 mg/mL aqueous solution of COOH micelles prepared by COOH-PEG-b-PCL diblock copolymer upon treating with 1 mg/mL lipase PS and 1mg/mL BSA at 25 ^{o}C (pH 7.0)

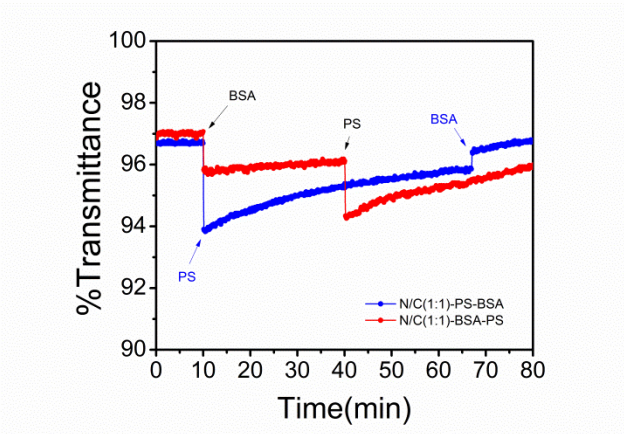

Fig. S5 Time dependence of UV transparency at 500 nm recorded for 1 mg/mL aqueous solution of NH_2/COOH (1:1) micelles prepared by NH_2-PEG-b-PCL and COOH-PEG-b-PCL diblock copolymers upon treating with 1mg/mL lipase PS and 1mg/mL BSA at 25 ^{o}C (pH 7.0)

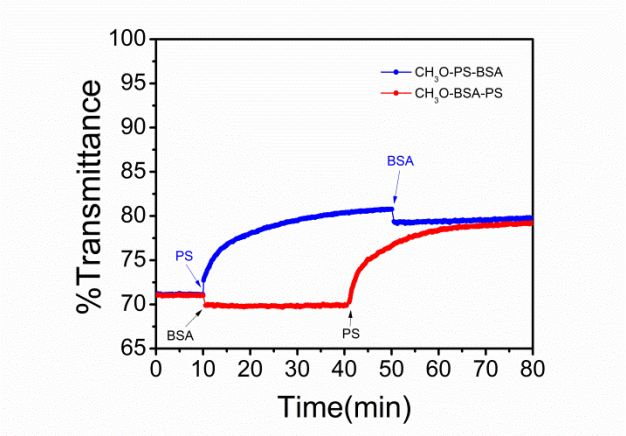

Fig. S6 Time dependence of UV transparency at 500 nm recorded for 1 mg/mL aqueous solution of CH_3O micelles prepared by CH_3O -PEG-b-PCL diblock copolymer upon treating with 1 mg/mL lipase PS and 1mg/mL BSA at 25 ^{o}C (pH 7.0)