

## Supporting Information

### Fabrication of well-defined electromagnetic Fe<sub>3</sub>O<sub>4</sub>/polyaniline hollow microspheres and their application on Pb<sup>2+</sup> uptake

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Table S1 Adsorption capacities for Pb<sup>2+</sup> by Fe<sub>3</sub>O<sub>4</sub>/PANI adsorbents prepared with different weight ratio of ANI/PS<sup>a</sup>

Samples	Weight ratio of ANI/PS	Thickness of PANI shell (nm)	Adsorption capacity (mg/g)
1 <sup>b</sup>	1:4		273
2	1:2.5	30	416
3	1:1.7	43	535

<sup>a</sup> The other conditions of preparing Fe<sub>3</sub>O<sub>4</sub>/PANI ( Sample1, 2 and 3) are the same, with the exception of different weight ratio of ANI/PS.

<sup>b</sup> When the weight ratio of ANI/PS is 1:4, the obtained Fe<sub>3</sub>O<sub>4</sub>/PANI (Sample 1) does not maintain the hollow structure.

Table S2 Adsorption capacities for Pb<sup>2+</sup> by Fe<sub>3</sub>O<sub>4</sub>/PANI adsorbents prepared with various amount of citric acid modified Fe<sub>3</sub>O<sub>4</sub> particles<sup>a</sup>

Samples	Citric acid modified Fe <sub>3</sub> O <sub>4</sub> (g)	Adsorption capacity (mg/g)
4	0.05	661
3	0.12	535
5	0.28	481

<sup>a</sup> The other conditions of preparing Fe<sub>3</sub>O<sub>4</sub>/PANI (Sample 3, 4 and 5) are the same, with the exception of various amount of citric acid modified Fe<sub>3</sub>O<sub>4</sub> particles.