Highly Stable Hybrid Selenophene-3,4-

Ethylenedioxythiophene as Electrically Conducting and Electrochromic Polymers

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Figure S1 ¹H NMR spectrum of 2-bromoselenophene.

-7.00



Figure S2 ¹H NMR spectrum of 2,5-dibromoselenophene.



Figure S3 ¹³C NMR spectrum of 2,5-dibromoselenophene.



Figure S4 ¹H NMR spectrum of 2,5-dibromo-EDOT.



Figure S6 ¹³C NMR spectrum of Se-EDOT.





Figure S8 ¹³C NMR spectrum of Se-EDOT-Se.



Figure S10 ¹³C NMR spectrum of EDOT-Se-EDOT.



155 150 145 140 135 130 125 120 115 110 105 100 95 90 85 80 75 70 65 60 55 50 45 40 ppm

Figure S12 ¹³C NMR spectrum of EDOT-Se-Se-EDOT.



Figure S13 The anodic oxidation curves of 0.01 mol L⁻¹ Se-EDOT (A), Se-EDOT-Se
(B), EDOT-Se-EDOT (C), and EDOT-Se-Se-EDOT (D) on a Pt working electrode in CH₂Cl₂-Bu₄NPF₆ (0.10 mol L⁻¹). Potential scan rate: 50 mV s⁻¹.



Figure S14 Chronoamperometric curves of 0.01 mol L⁻¹ Se-EDOT (A), Se-EDOT-Se (B), EDOT-Se-EDOT (C), and EDOT-Se-Se-EDOT (D) on a Pt working electrode in



 CH_2Cl_2 -Bu₄NPF₆ (0.10 mol L⁻¹) at different applied potentials for 500 s.

Figure S15 Optimized structures of Se-EDOT oligomers.



Figure S16 Main atomic electron density populations for Se-EDOT oligomers.



Table S1 The experimental switching colors of the hybrid polymers from the doped to

Figure S17 Optical activity and stability upon repeated cycling of PSe-EDOT (A), PSe-EDOT-Se (B), PEDOT-Se-EDOT (C), and PEDOT-Se-Se-EDOT (D) at 580, 520,

510, 580 nm, respectively. Switching time: 10 s.