## **Supporting Information**

Synthesis of Donor-Acceptor Conjugated Copolymers Based on Benzo[1,2-*b*:4,5-*b'*]dithiophene and 2,1,3-Benzothiadiazole via Direct Arylation Polycondensation : Toward Efficient C-H Activation in Nonpolar Solvents

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**Figure S1.** GPC curves of **PBDTBTs** polymerized in different catalyst systems. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h, at the condition of **line 1**: Pd(OAc)<sub>2</sub> (5 mol %)/ PCy<sub>3</sub> HBF<sub>4</sub> (10 mol %), K<sub>2</sub>CO<sub>3</sub> (0.6 mmol), PivOH (0.06 mmol) in DMAc (1 mL); **line 2**: Herrmann Catalyst (5 mol %)/ (*o*-MeOPh)<sub>3</sub>P (10 mol %), Cs<sub>2</sub>CO<sub>3</sub> (0.6 mmol), PivOH (0.06 mmol) in THF (1 mL); **line 3**: Pd<sub>2</sub>dba<sub>3</sub> (5 mol %)/(*o*-MeOPh)<sub>3</sub>P (10 mol %), Cs<sub>2</sub>CO<sub>3</sub> (0.6 mmol), Cs<sub>2</sub>CO<sub>3</sub> (0.6 mmol), PivOH (0.06 mmol), PivOH (0.06 mmol) in THF (1 mL); **line 3**: Pd<sub>2</sub>dba<sub>3</sub> (5 mol %)/(*o*-MeOPh)<sub>3</sub>P (10 mol %), Cs<sub>2</sub>CO<sub>3</sub> (0.6 mmol), PivOH (0.06 mmol), PivOH (0.0



**Figure S2.** GPC curves of **PBDTBTs** polymerized in different solvents. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100  $^{\circ}$ C for 24 h in solvent (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), Cs<sub>2</sub>CO<sub>3</sub> (0.6 mmol) and PivOH (0.06 mmol).



**Figure S3.** UV-vis absorption spectra of chloroform solutions of **PBDTBTs** polymerized in different solvents. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in solvent (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %),  $(o-\text{MeOPh})_3\text{P}$  (10 mol %), Cs<sub>2</sub>CO<sub>3</sub> (0.6 mmol) and PivOH (0.06 mmol).



**Figure S4.** GPC curves of **PBDTBTs** polymerized with different base. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100  $^{\circ}$ C for 24 h in ODMB (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), base (0.6 mmol) and PivOH (0.06 mmol).



**Figure S5.** UV-vis absorption spectra of chloroform solutions of **PBDTBTs** polymerized with different base. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), base (0.6 mmol) and PivOH (0.06 mmol).



**Figure S6.** GPC curves of **PBDTBTs** polymerized with different equivalent of  $K_2CO_3$ . Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %),  $K_2CO_3$  and PivOH (0.06 mmol).



**Figure S7.** UV-vis absorption spectra of chloroform solutions of **PBDTBTs** polymerized with different equivalent of  $K_2CO_3$ . Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), K<sub>2</sub>CO<sub>3</sub> and PivOH (0.06 mmol).



**Figure S8.** GPC curves of **PBDTBTs** polymerized with different equivalent of PivOH. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), K<sub>2</sub>CO<sub>3</sub> (0.6 mmol) and PivOH.



**Figure S9.** UV-vis absorption spectra of chloroform solutions of **PBDTBTs** polymerized with different equivalent of PivOH. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100  $^{\circ}$ C for 24 h in ODMB (1 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), K<sub>2</sub>CO<sub>3</sub> (0.6 mmol) and PivOH.



**Figure S10.** GPC curves of **PBDTBTs** polymerized at different concentration. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (1, 2 or 4 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %),  $K_2CO_3$  (1 mmol) and PivOH (0.06 mmol).



**Figure S11.** UV-vis absorption spectra of chloroform solutions of **PBDTBTs** polymerized at different concentration. Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (1, 2 or 4 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), K<sub>2</sub>CO<sub>3</sub> (1 mmol) and PivOH (0.06 mmol).



**Figure S12.** GPC curves of **PBDTBTs** polymerized with different phase transfer agents (PTA). Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (2 mL), with Pd<sub>2</sub>dba<sub>3</sub> (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), K<sub>2</sub>CO<sub>3</sub> (1 mmol), PivOH (0.06 mmol) and PTA (0.06 mmol).



**Figure S13.** UV-vis absorption spectra of chloroform solutions of **PBDTBTs** polymerized with different phase transfer agents (PTA). Polymerizations of **HDBDT** (0.2 mmol) and **BrBT** (0.2 mmol) were run at 100 °C for 24 h in ODMB (2 mL), with  $Pd_2dba_3$  (5 mol %), (*o*-MeOPh)<sub>3</sub>P (10 mol %), K<sub>2</sub>CO<sub>3</sub> (1 mmol), PivOH (0.06 mmol) and PTA (0.06 mmol).



**Figure S14**. UV-vis absorption spectra of chloroform solutions of **PBDTBTs** with different molecular weight synthesized in ODMB under different conditions: lines corresponding to the polymers with  $M_n/M_w$  of 3.4/7.7, 5.4/8.5, 9.2/15.7, 13.2/23.7, 16.8/35.9 and 24.5/60.1 (kg/mol), respectively.