

Supporting information

Synthesis of cationic poly(3-acrylamidopropyl trimethylammonium chloride) by SARA ATRP in ecofriendly solvent mixtures

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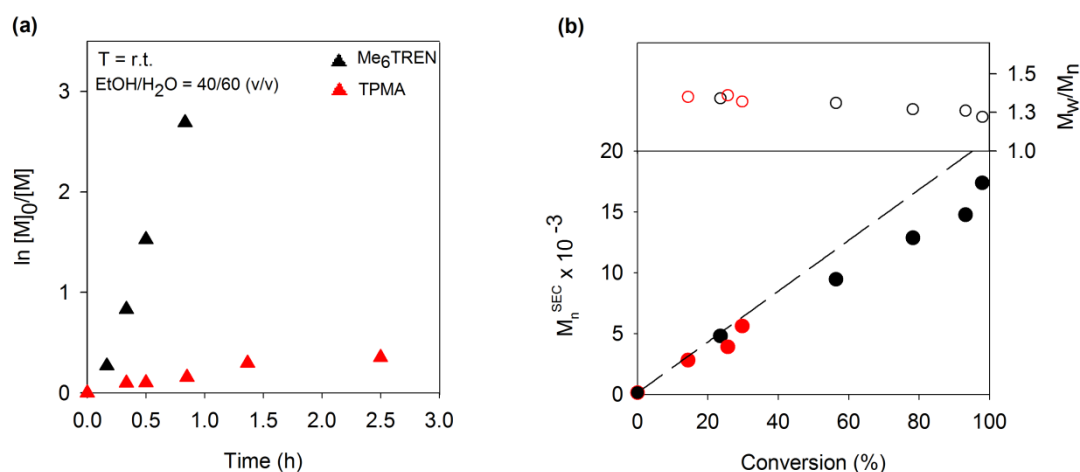


Figure FS1. (a) Kinetic plots of $\ln[M]_0/[M]$ vs. time and (b) plot of number-average molecular weights (M_n^{SEC}) and D (M_w/M_n) vs. monomer conversion (the dashed line represents theoretical molecular weight at a given conversion) for the SARA ATRP of AMPTMA initiated by ECP, using Me₆TREN or TPMA as ligand. Conditions: $[AMPTMA]_0/[ECP]_0/[CuCl_2]_0/[ligand]_0/Cu(0) \text{ wire} = 100/1/0.5/1.0/Cu(0) \text{ wire}$; $[AMPTMA]_0 = 1.45 \text{ M}$; Cu(0) wire: $l = 10 \text{ cm}$; $d = 1 \text{ mm}$; $V_{\text{solvent}} = 5 \text{ mL}$.

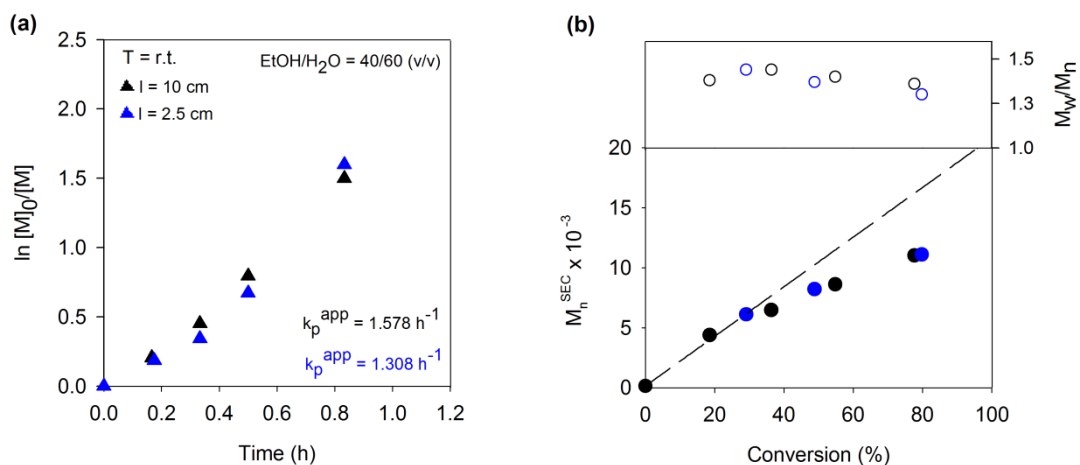


Figure FS2. (a) Kinetic plots of $\ln[M]_0/[M]$ vs. time and (b) plot of number-average molecular weights (M_n^{SEC}) and D (M_w/M_n) vs. monomer conversion (the dashed line represents theoretical molecular weight at a given conversion) for the SARA ATRP of AMPTMA using different lengths of Cu(0) wire. Conditions: $[AMPTMA]_0/[ECP]_0/[CuCl_2]_0/[Me_6TREN]_0/Cu(0) \text{ wire} = 100/1/0.3/0.6/Cu(0) \text{ wire}$; $[AMPTMA]_0 = 1.45 \text{ M}$; Cu(0) wire: $d = 1 \text{ mm}$; $V_{\text{solvent}} = 5 \text{ mL}$.

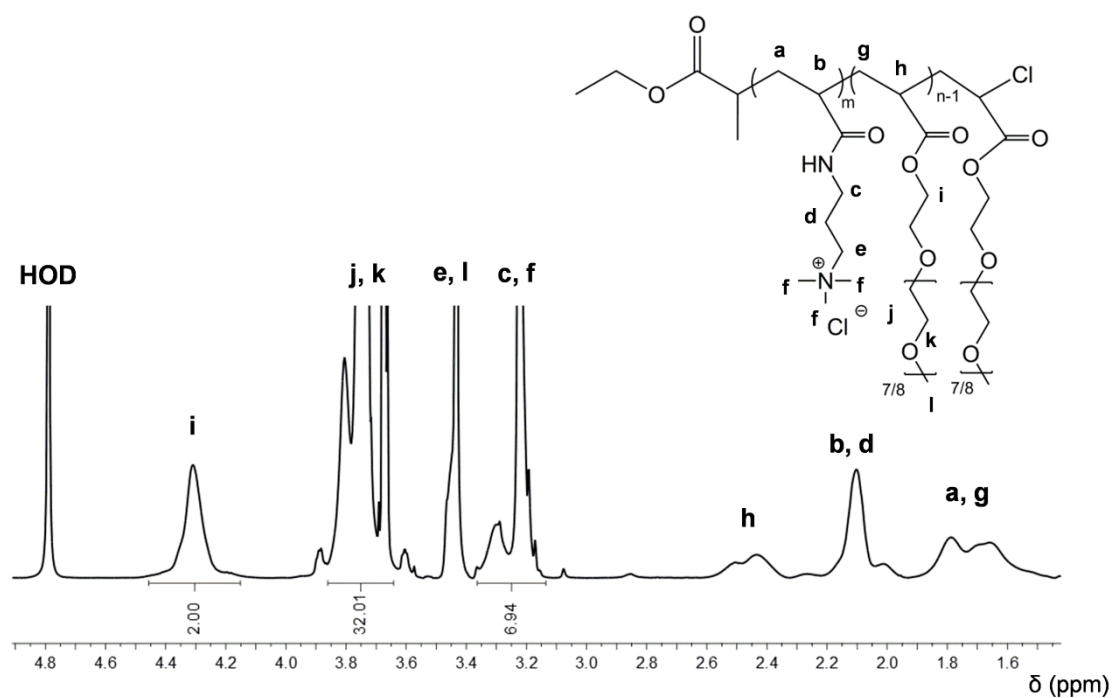


Figure FS3. 500 MHz ^1H NMR spectrum of the PAMPTMA₂₄-*b*-POEOA₃₉, in D₂O, obtained by “one-pot” SARA ATRP at room temperature. Reaction conditions: first block - [AMPTMA]₀/[ECP]₀/[CuCl₂]₀/[Me₆TREN]₀/Cu(0) wire = 25/1/0.5/1.0; Cu(0) wire: $l = 5$ cm; $d = 1$ mm; [AMPTMA]₀ = 1.45 M; conversion = 97 %; $V_{\text{solvent}} = 1.25$ mL; $T = 25$ °C; second block - [OEOA₄₈₀]₀ = 0.4 M; [OEOA]₀/[ECP]₀ = 50; conversion = 90 %.

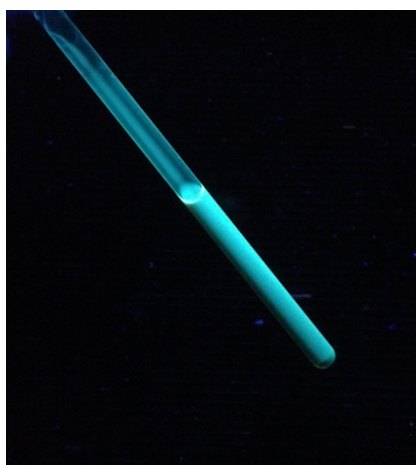


Figure FS4. UV light ($\lambda = 366$ nm) irradiated NMR tube containing a solution of fluorescent coumarin-functionalized PAMPTMA in CD₃OD, obtained by a “click” reaction.