

Inorganic anion-assisted supramolecular assemblies of bent dipyridines: effects of anionic geometries on the hydrogen-bonding networks

Shi Wang, Xue-Hua Ding, Yong-Hua Li and Wei Huang

Table S1. Hydrogen-bonding Distances and Parameters for the complex of **1–7** (Å, °)

Compound	D–H···A	D–H	H···A	D···A	D–H···A	Symmetry Code
1	N1–H3···O8	0.86	1.82	2.670(3)	170	1+x, 3/2–y, 1/2+z
	N4–H8···O2	0.86	1.88	2.710(3)	161	x, y, 1+z
	N4–H8···O3	0.86	2.41	3.120(3)	140	x, y, 1+z
	O8–H11···O4	0.85	2.21	2.977(3)	149	–x, 2–y, 1–z
	O8–H11···N2	0.85	2.56	3.073(3)	120	1–x, 2–y, 1–z
	O8–H12···N2	0.85	2.52	3.333(3)	161	–1+x, y, –1+z
	O9–H13···O5	0.76	2.10	2.849(3)	168	x, 3/2–y, 1/2+z
	O9–H14···O7	0.85	2.24	2.919(3)	137	
	C1–H1···O9	0.93	2.55	3.451(3)	163	x, 3/2–y, 1/2+z
	C2–H2···O2	0.93	2.34	3.243(4)	163	1–x, –1/2+y, 3/2–z
	C3–H4···O7	0.93	2.34	3.217(4)	157	1+x, y, z
	C4–H5···O4	0.93	2.49	3.412(4)	175	1+x, y, z
	C10–H7···O5	0.93	2.56	3.194(4)	126	x, y, 1+z
	C11–H9···O9	0.93	2.39	3.258(4)	155	1–x, 2–y, 2–z
	C12–H10···O6	0.93	2.40	3.330(3)	174	1–x, 1/2+y, 3/2–z
2	N1–H3···O3	0.86	1.73	2.582(3)	170	
	N4–H8···O4	0.86	1.76	2.607(3)	168	x, 1+y, z
	O2–H11···O5	0.82	1.72	2.514(3)	162	–x, y, 3/2–z
	C2–H2···O3	0.93	2.53	3.452(4)	174	1/2–x, 1/2–y, 2–z
	C4–H5···N2	0.93	2.49	3.384(4)	162	–x, 1–y, 1–z
	C12–H10···O2	0.93	2.44	3.248(4)	145	x, 1–y, –1/2+z
3	O4–H1(O4)···O5	0.88	1.91	2.761(11)	161	
	O4–H2(O4)···I10	0.87	2.86	3.393(8)	121	1–x, 1–y, 1–z
	N1–H3···O5	0.86	1.83	2.684(9)	176	1–x, –y, 1–z
	O5–H1(O5)···I2	0.85	2.90	3.709(6)	160	–1+x, y, z
	O5–H2(O5)···I11	0.85	2.69	3.484(7)	157	–1+x, y, z
	N8–H17···I10	0.86	2.76	3.441(7)	137	x, 1+y, z
	N9–H22···N5	0.86	1.85	2.674(9)	161	2–x, 1–y, –z
	N12–H27···O4	0.86	1.84	2.683(11)	167	–x, 2–y, 1–z
	C10–H7···I10	0.93	2.99	3.677(9)	132	1–x, 1–y, 1–z
	C11–H9···I4	0.93	3.03	3.842(11)	146	1–x, 1–y, 1–z

4	C21–H15...N3	0.93	2.62	3.360(11)	137	1–x, 1–y, 1–z
	C22–H16...I11	0.93	3.05	3.739(9)	132	1–x, 2–y, 1–z
	C33–H25...N11	0.93	2.54	2.855(11)	100	
	C35–H28...I7	0.93	3.02	3.792(10)	142	1–x, 2–y, –z
	N4–H7...N8	0.86	1.90	2.737(6)	165	2–x, 1–y, 1–z
	N5–H12...N1	0.86	1.90	2.707(6)	157	1–x, –y, –z
	O3–H19...F10	0.85	2.30	3.135(15)	168	1–x, 1–y, 1–z
	O3–H20...N3	0.85	2.58	3.393(14)	161	
	C1–H1...F8	0.93	2.51	3.331(15)	147	1–x, –y, 1–z
	C2–H2...O3	0.93	2.25	3.159(14)	167	x, –1+y, z
	C4–H4...N6	0.93	2.54	3.267(6)	135	1–x, 1–y, –z
	C9–H5...F9	0.93	2.42	3.355(7)	177	1–x, –y, 1–z
	C10–H6...F3	0.93	2.51	3.235(8)	135	1–x, –y, 1–z
	C11–H8...F12	0.93	2.39	3.315(7)	171	1–x, 1–y, 1–z
	C13–H10...F6	0.93	2.43	3.35(2)	170	1+x, y, z
5	C21–H15...F6	0.93	2.41	3.329(15)	171	1+x, y, z
	N1–H3...F1	0.86	2.31	2.901(5)	126	2–x, –y, 2–z
	N1–H3...F2	0.86	2.12	2.811(6)	137	1–x, –y, 2–z
	N4–H8...O2	0.86	1.78	2.628(6)	169	–1/2+x, 3/2–y, 1/2+z
	O2–H11...F4	0.96	2.12	2.855(15)	132	1–x, 1–y, 1–z
	O2–H12...F5'	0.96	2.22	2.834(12)	120	
	C2–H2...F3	0.93	2.40	3.242(6)	150	–1+x, y, z
	C4–H5...F6	0.93	2.50	3.384(6)	158	3/2–x, –1/2+y, 3/2–z
	C9–H6...F7'	0.93	2.38	3.303(10)	175	–x, 1–y, 1–z
	C10–H7...F4	0.93	2.49	3.180(16)	131	1/2–x, 1/2+y, 3/2–z
6	C10–H7...F4'	0.93	2.18	2.990(17)	145	1/2–x, 1/2+y, 3/2–z
	C12–H10...N3	0.93	2.40	3.317(6)	169	2–x, 1–y, 2–z
	C2–H2...F4	0.93	2.41	3.282(3)	156	x, y, 1+z
	C3–H3...F4	0.93	2.45	3.343(4)	162	1–x, –y, 1–z
	C9–H5...N3	0.93	2.47	3.239(4)	141	1–x, 1–y, –z
	C10–H6...F2	0.93	2.52	3.347(3)	149	–x, 1–y, –z
	C11–H8...F2	0.93	2.46	3.271(3)	146	x, 1+y, z
	N1–H3...O3	0.86	2.53	3.143(2)	129	x, y, 1+z
	N1–H3...O6	0.86	1.81	2.654(2)	166	x, y, 1+z
	N2–H6...O3	0.86	1.93	2.752(2)	161	x, 1/2–y, 1/2+z
7	N3–H7...O5	0.86	1.91	2.732(2)	159	1+x, 1/2–y, 1/2+z
	N4–H10...O4	0.86	1.73	2.578(2)	170	1+x, y, z
	C3–H1...O6	0.93	2.44	3.309(2)	156	1+x, y, 1+z
	C5–H4...O1	0.93	2.38	3.215(3)	150	–1+x, y, z
	C6–H5...O5	0.93	2.47	3.095(2)	125	x, 1/2–y, 1/2+z
	C9–H8...O5	0.93	2.43	3.264(2)	150	1+x, 1/2–y, 1/2+z
	C10–H9...O2	0.93	2.25	3.123(3)	157	1+x, y, z
	C11–H11...O3	0.93	2.53	3.131(2)	122	

Table S2. Weak Intermolecular Interactions for the Complex of 1–7 (Å)

Compound	Weak Interactions	Shortest Atom-Atom Distance	Anion/Centroid- Centroid Distance
1	N6–O6···Cg1	3.111	3.228
	N6–O5···Cg2	3.216	3.766
2	P1–O3···Cg1	3.165	3.440
	Cg2···Cg3(C2···C11)	3.285	3.741
	Cg1···Cg1(C6···C6)	3.398	3.287
	Cg1···Cg1(C7···C7)	3.295	3.287
3	I3···I9	3.837	
	I1···I6	3.922	
	Cg2···Cg1(C2···C7)	3.235	3.589
	Cg7···Cg4 ⁱ (C19···N11)	3.162	3.400
4	P1–F5···Cg1	2.999	3.312
	P1–F6'···Cg4	3.019	3.175
	P1–F1···Cg2	3.041	3.722
	P2–F12···Cg3	3.122	3.490
	P2–F8···Cg1	3.044	3.051
	Cg4···Cg5(N6···C15)	3.181	3.418
5	B1–F1···Cg3	3.166	3.633
	B1–F3···Cg2	3.136	2.978
	B1–F4···Cg3	2.827	3.287
	B1–F4'···Cg3	2.916	3.112
	B2–F5'···Cg1	3.080	3.133
	Cg3···Cg1(C12···C7)	3.395	3.820
6	B1–F1···Cg1	2.974	2.940
	B1–F3···Cg3	3.156	3.706
	B1–F4···Cg3	3.165	3.963
	N2–H10···Cg2	2.859	3.452
7	S1–O4···Cg1 ⁱ	3.088	3.774
	S1–O5···Cg1 ⁱ	3.095	4.229

Cg1 = centroid of the O1C6N2N3C7 ring, Cg1ⁱ = centroid of the N1C4C3C2C6C5 ring, Cg2 = centroid of the N1C2C1C5C4C3 ring, Cg3 = centroid of the N4C10C9C8C12C11 ring, Cg4 = Cg7 = centroid of the O2C18N6N7C19 ring, Cg4ⁱ = centroid of the O3C30N10N11C31 ring, Cg5 = centroid of the N5C14C13C17C16C15 ring.

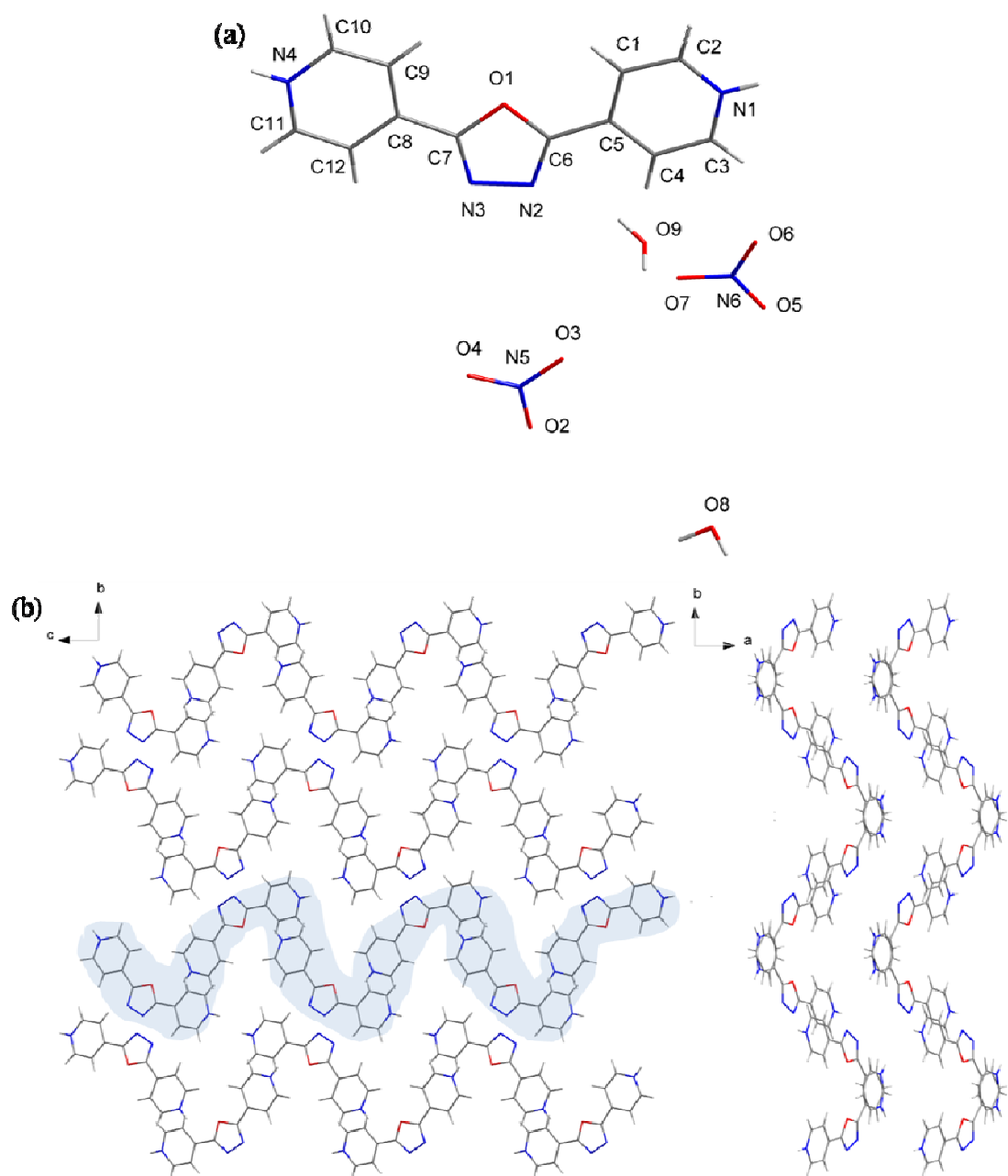


Fig. S1 (a) The asymmetric unit of **1** together with the numbering scheme; (b) Separate molecular structure of 4-H₂bpo²⁺ cations.

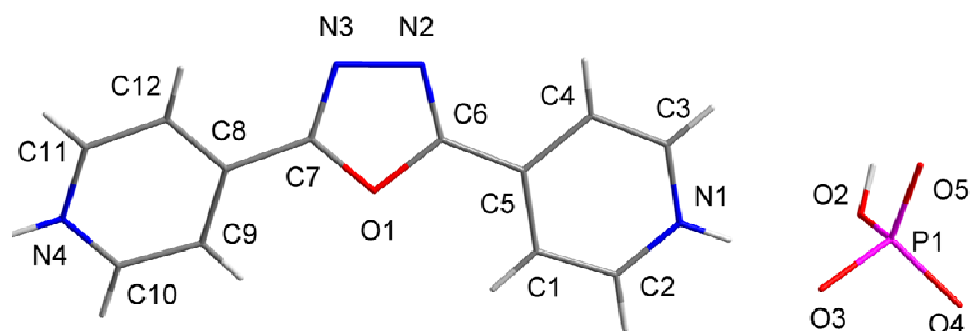
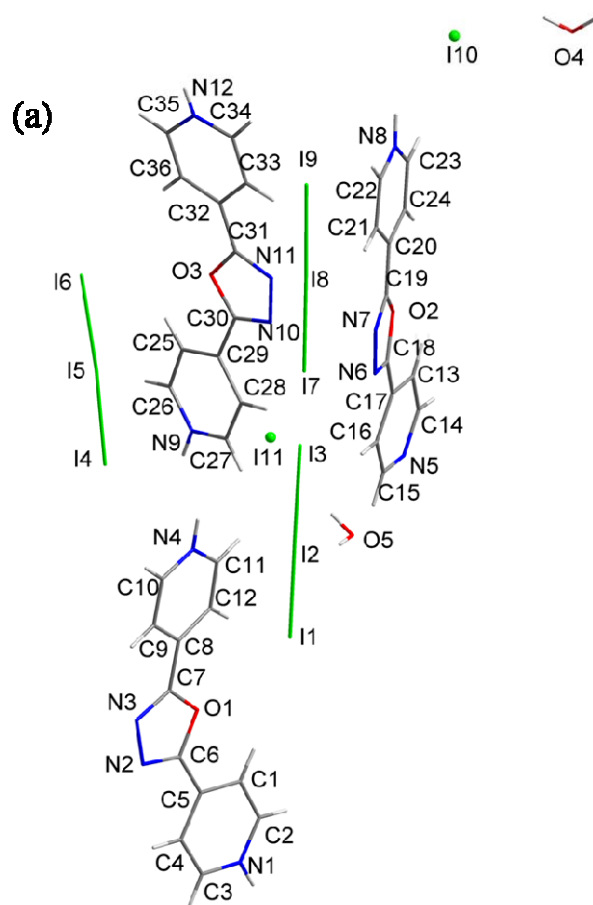


Fig. S2 The asymmetric unit of **2** together with the numbering scheme.



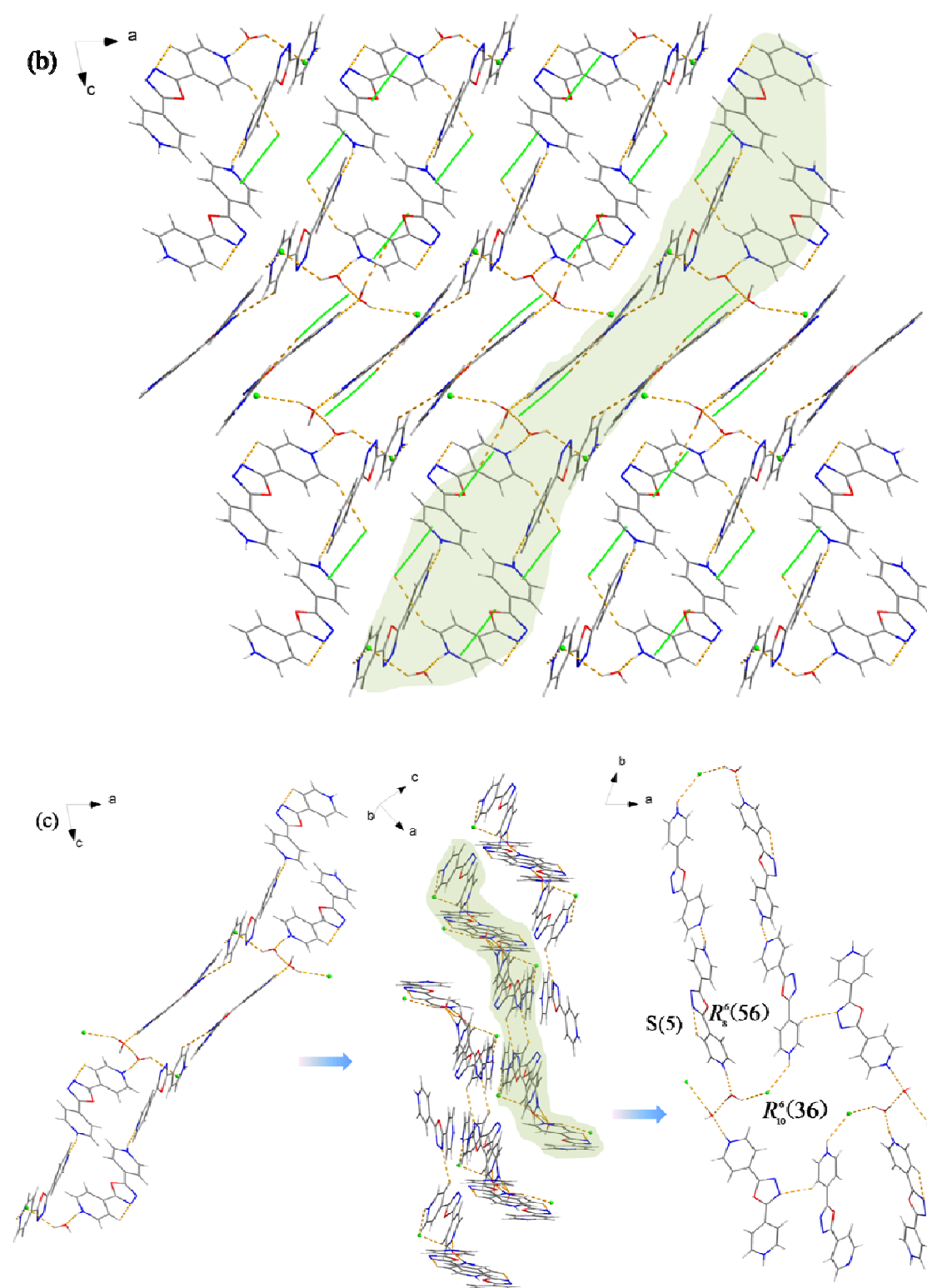


Fig. S3 (a) The asymmetric unit of **3** together with the numbering scheme where ball-and-stick model was used for I^- ions for clarity; (b) The 3D hydrogen-bonding network parallel to the ac -plane; (c) Independent molecular structure with I_3^- anions deleted and the side view as well as a moiety of structure in pale green viewed along

the *c*-axis.

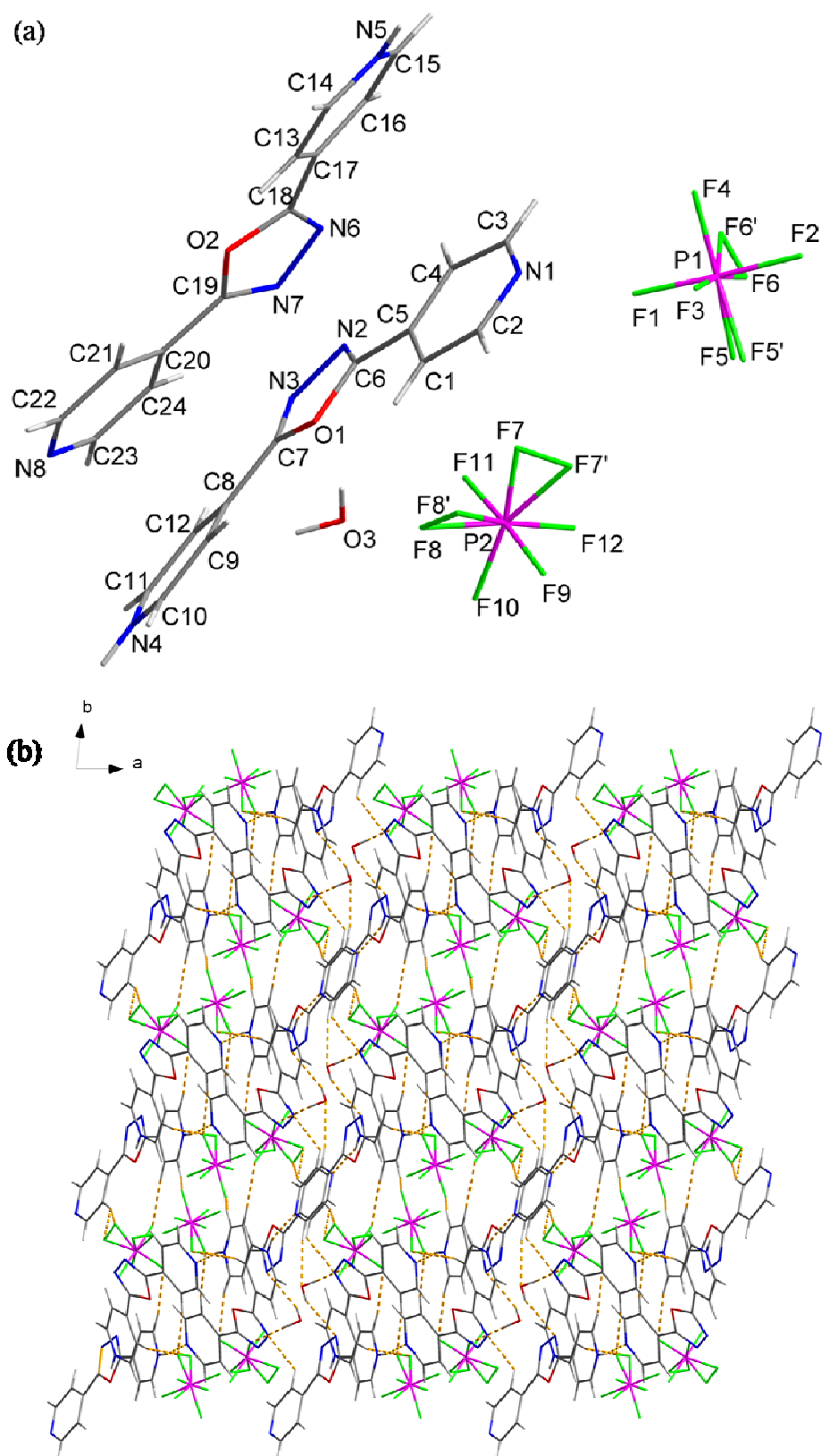


Fig. S4 (a) The asymmetric unit of **4** together with the numbering scheme; (b) The 3D

hydrogen-bonding network parallel to the *ab*-plane.

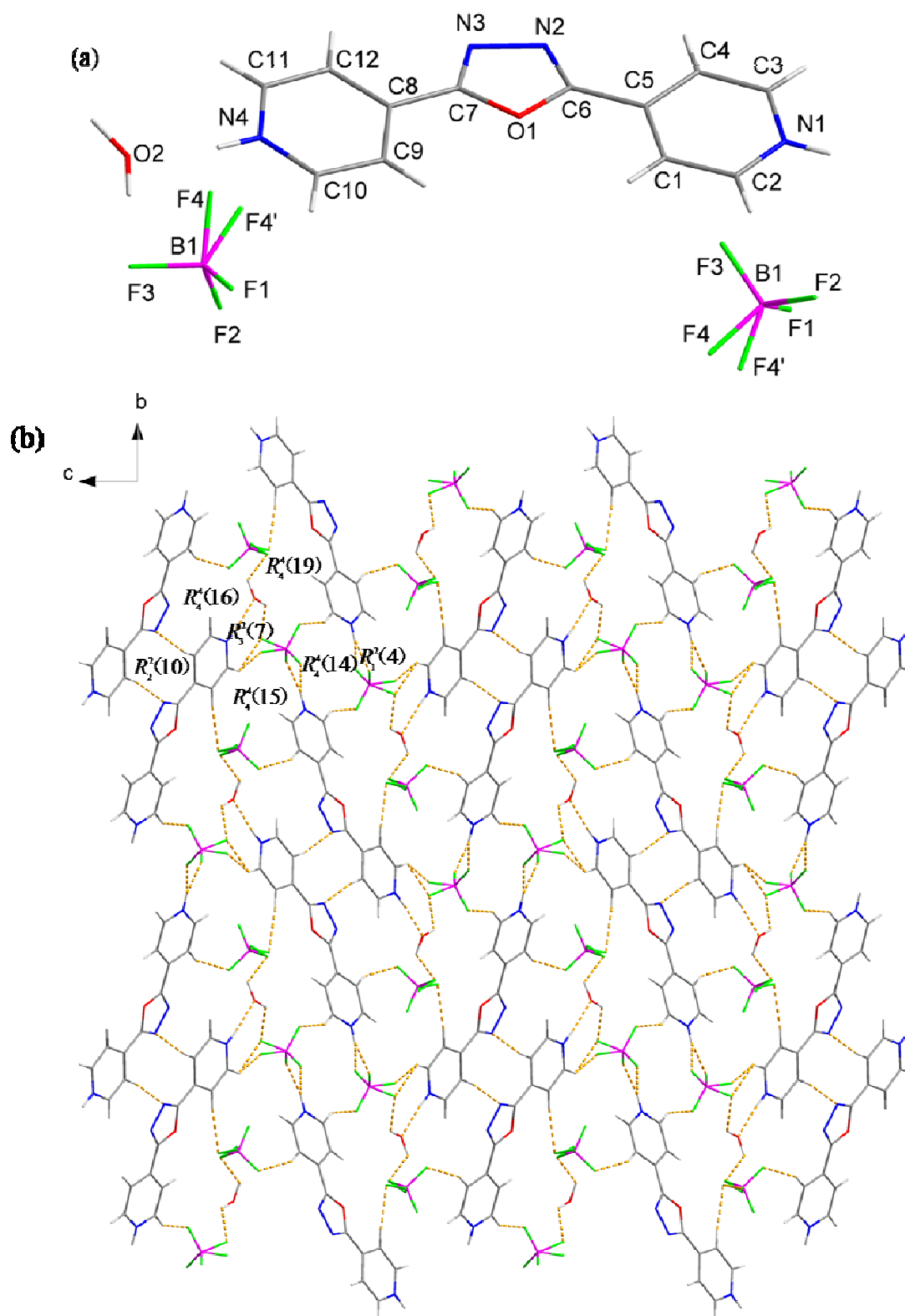


Fig. S5 (a) The asymmetric unit of **5** together with the numbering scheme; (b) The 3D hydrogen-bonding network parallel to the *bc*-plane.

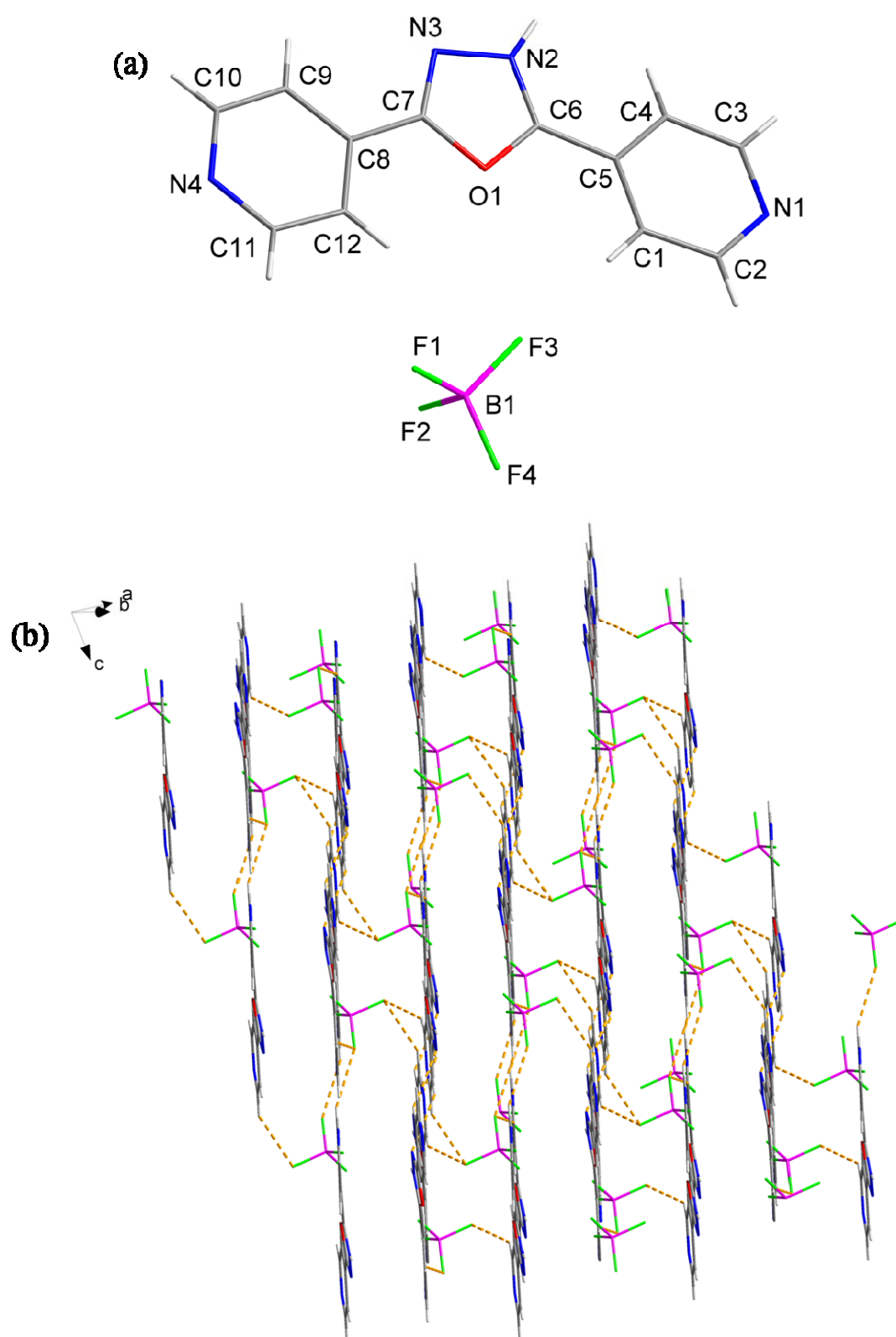


Fig. S6 (a) The asymmetric unit of **6** together with the numbering scheme; (b) A side view of the 3D crystal packing.

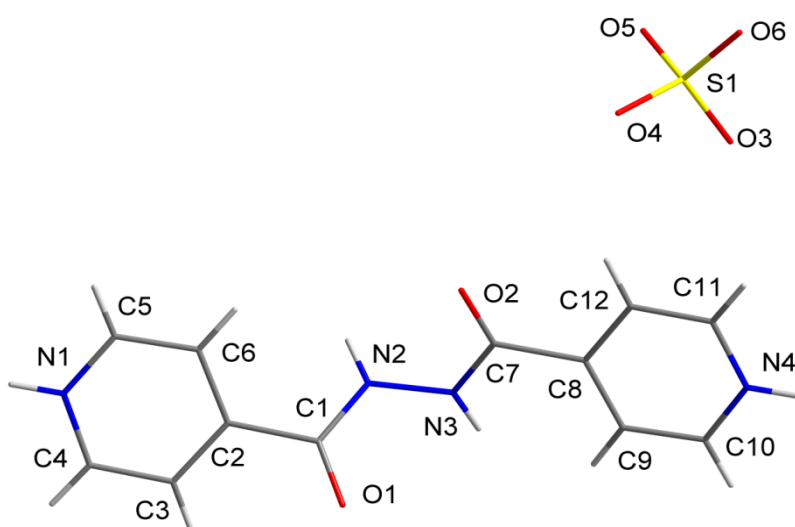


Fig. S7 The asymmetric unit of **7** together with the numbering scheme.

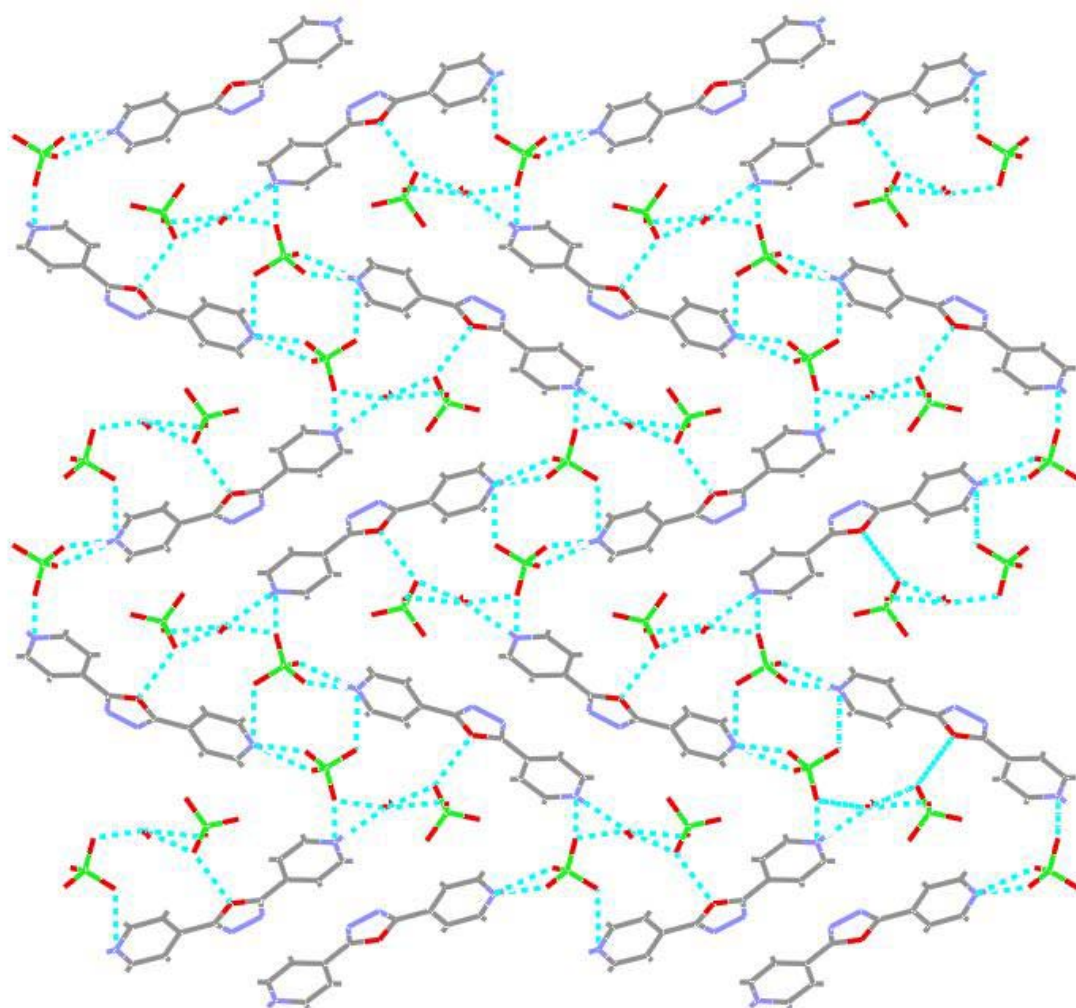


Fig. S8 The crystal packing of perchlorate in ref. 44.

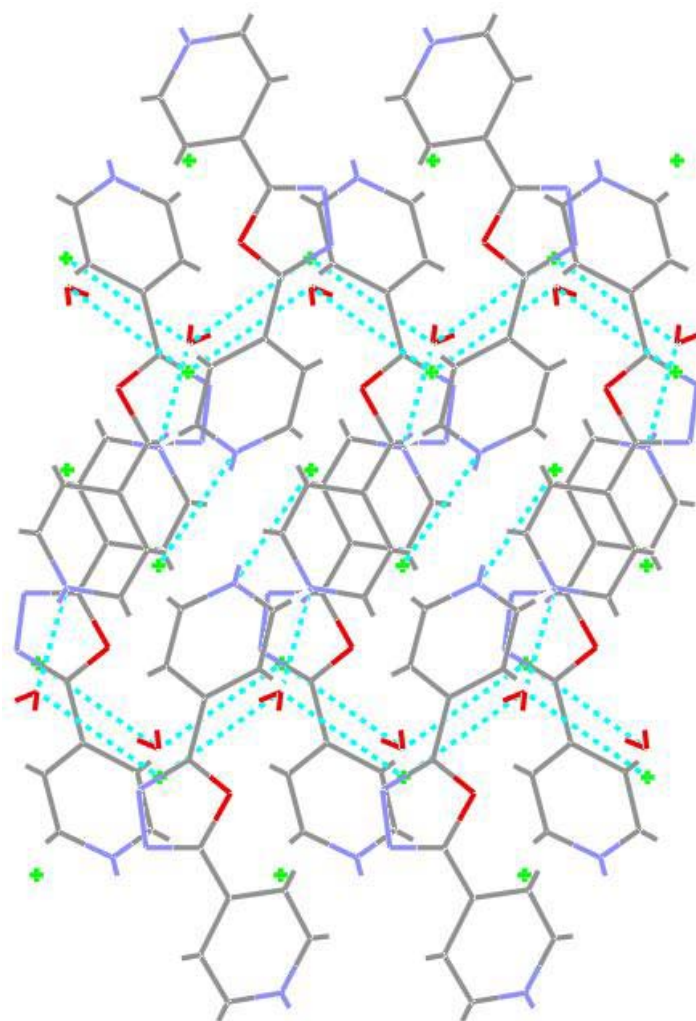


Fig. S9 The crystal packing of bromide in ref. 45.