Combining coordination and supramolecular chemistry to explore uranyl assembly in the solid state
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## Supporting Info Section

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## II. Powder X-ray Diffraction data

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## I. Additional Figures



Figure S1: Complex 3 shown down the [010] direction highlighting the hydrogen bonding interactions that decorate the 1D chain of uranyl monomers.


Figure S2: Complex 7 viewed down approximately the [110] direction highlighting the localized $\mathrm{Br}-\pi$ interactions that assemble neighboring uranyl monomers.


Figure S3: Complex 10 viewed down approximately the [011] direction highlighting the Type I Br-Cl interaction that links neighboring uranyl monomers.


Figure S4: Complex 12 viewed along approximately the [100] direction highlighting the localized $\mathrm{Cl}-\pi$ interaction that links adjacent uranyl dimers.

## II. Powder X-ray diffraction data

For the following PXRD spectra it is important to note that calculated patterns are from low temperature ( 100 K ) (complexes $\mathbf{1 , 3 , 7} 7$ and 11) data collections while observed patterns were collected at room temperature ( 298 K ). This difference may result in slight shifts in two-theta values. For complexes 2, 4-6, 8-10 and $\mathbf{1 2}$ calculated and observed patterns were both collected a room temperature (298K).


Figure S5: The observed PXRD pattern of structure 1 with calculated pattern overlaid in blue.


Figure S6: The observed PXRD pattern of structure 2 with calculated pattern overlaid in blue. We acknowledge a minor impurity as indicated with an asterisk and we have identified this impurity as excess 1,10-phenanthroline (calculated CIF overlaid in red).


Figure S7: The observed PXRD pattern of structure 3 with calculated pattern overlaid in blue. We acknowledge several impurities in the bulk product. Some of these impurities have been identified as 1,10-phenanthroline (calculated CIF overlaid in red) and these are indicated with purple asterisks. Impurities that could not be identified are indicated with green asterisks.


Figure S8: The observed PXRD pattern of structure 4 with calculated pattern overlaid in blue. We acknowledge several impurities as indicated with asterisks.


Figure S9: The observed PXRD pattern of structure 5 with calculated pattern overlaid in blue. We acknowledge a minor impurity as indicated with an asterisk.


Figure S10: The observed PXRD pattern of structure $\mathbf{6}$ with calculated pattern overlaid in blue. We acknowledge two minor impurities as indicated with asterisks.


Figure S11: The observed PXRD pattern of structure 7 with calculated pattern overlaid in blue. We acknowledge several impurities in the bulk product. Some of these impurities
have been identified as $2,2^{\prime}: 6^{\prime}, 2^{\prime \prime}$-terpyridine (calculated CIF overlaid in red) and these are indicated with purple asterisks. Impurities that could not be identified are indicated with green asterisks.


Figure S12: The observed PXRD pattern of structure $\mathbf{8}$ with calculated pattern overlaid in blue.


Figure S13: The observed PXRD pattern of structure 9 with calculated pattern overlaid in blue. We acknowledge two minor impurities as indicated with asterisks.


Figure S14: The observed PXRD pattern of structure 10 with calculated pattern overlaid in blue. We acknowledge a number of minor impurities as indicated with asterisks.


Figure S15: The observed PXRD pattern of structure 11 with calculated pattern overlaid in blue. We acknowledge a couple of minor impurities in the bulk product. One of these impurities was identified as $4^{\prime}$-chloro- $2,2^{\prime}: 6^{\prime}, 2^{\prime \prime}$-terpyridine (calculated CIF overlaid in red) and it is indicated with a purple asterisk. Impurities that could not be identified are indicated with green asterisks.


Figure S16: The observed PXRD pattern of structure $\mathbf{1 2}$ with calculated pattern overlaid in blue.

## III. Thermal Ellipsoid Plots



Figure S17: ORTEP illustration of complex 1. Ellipsoids are shown at 50\% probability level.


Figure S18: ORTEP illustration of structure 2. Ellipsoids are shown at $50 \%$ probability level.


Figure S19: ORTEP illustration of structure 3. Ellipsoids are shown at $50 \%$ probability level. Atoms labelled with an "i"" are reproduced through symmetry.


Figure S20: ORTEP illustration of structure 4. Ellipsoids are shown at $50 \%$ probability level. Atoms labelled with an "i'" are reproduced through symmetry.


Figure S21: ORTEP illustration of complex 5. Ellipsoids are shown at $50 \%$ probability level.


Figure S22: ORTEP illustration of structure 6. Ellipsoids are shown at 50\% probability level.


Figure S23: ORTEP illustration of structure 7. Ellipsoids are shown at 50\% probability level.


Figure S24: ORTEP illustration of structure 8. Ellipsoids are shown at 50\% probability level.


Figure S25: ORTEP illustration of structure 9. Ellipsoids are shown at 50\% probability level.


Figure S26: ORTEP illustration of structure 10. Ellipsoids are shown at $50 \%$ probability level.


Figure S27: ORTEP illustration of structure 11. Ellipsoids are shown at $50 \%$ probability level.


Figure S28: ORTEP illustration of structure 12. Ellipsoids are shown at $50 \%$ probability level.

## IV. Tables of Bond Distances

Table S1: U-O Axial Bond Lengths in $\mathrm{UO}_{2}{ }^{2+}$ complexes (1-12).

| Complex | $\mathrm{d}_{\mathrm{U} 1-\mathrm{O} 1}$ <br> $[\AA]$ | $\mathrm{d}_{\mathrm{U} 11-\mathrm{O} 2}$ <br> $[\AA]$ | $\mathrm{d}_{\mathrm{U} 2-\mathrm{O} 3}$ <br> $[\AA]$ | $\mathrm{d}_{\mathrm{U} 2-\mathrm{O} 4}$ <br> $[\AA]$ |
| :--- | :---: | :---: | :---: | :---: |
| 1 | $1.780(3)$ | $1.769(2)$ |  |  |
| 2 | $1.764(3)$ | $1.755(3)$ |  |  |
| 3 | $1.773(3)$ |  |  | $1.794(3)$ |
| 4 | $1.776(3)$ | $1.799(3)$ | $1.794(3)$ | $1.758(4)$ |
| 5 | $1.765(3)$ | $1.758(3)$ | $1.760(4)$ |  |
| 6 | $1.760(3)$ | $1.773(3)$ |  |  |
| 7 | $1.776(3)$ | $1.772(3)$ |  | $1.766(7)$ |
| 8 | $1.774(7)$ | $1.759(7)$ | $1.749(7)$ | $1.755(4)$ |
| 9 | $1.763(3)$ | $1.757(4)$ | $1.764(4)$ |  |
| 10 | $1.755(4)$ | $1.754(4)$ |  | $1.769(5)$ |
| 11 | $1.758(4)$ | $1.760(4)$ | $1.778(5)$ | $1.765(4)$ |
| 12 | $1.766(4)$ | $1.756(4)$ | $1.767(5)$ |  |

Table S2: U-O Equatorial Bond Lengths in $\mathrm{UO}_{2}{ }^{2+}$ complexes (1-12).

| Complex | $\mathrm{d}_{\mathrm{Ul}-\mathrm{O} 3}$ $[\AA]$ | $\mathrm{d}_{\mathrm{U} 1-\mathrm{O} 4}$ [Å] | $\mathrm{d}_{\mathrm{Ul}-\mathrm{O}}$ [ $\AA$ ] | $\mathrm{d}_{\mathrm{U} 1-\mathrm{O} 6}$ <br> [ $\AA$ ] | $\begin{gathered} \mathrm{d}_{\mathrm{U} 2-\mathrm{O}} \\ {[\AA]} \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{d}_{\mathrm{U} 2-\mathrm{O6}} \\ {[\AA \bar{\AA}]} \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{d}_{\mathrm{U} 2-07} \\ {[\AA]} \\ \hline \end{gathered}$ | $\mathrm{d}_{\mathrm{U} 2-\mathrm{O} 8}$ <br> [ $\AA$ ] | $\begin{gathered} \mathrm{d}_{\mathrm{UL}-\mathrm{O9}} \\ {[\AA]} \\ \hline \end{gathered}$ | $\mathrm{d}_{\mathrm{U2}-\mathrm{O} 10}$ <br> [ $\AA$ ] |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 2.405(2) | 2.442(2) | 2.225(3) |  |  |  |  |  |  |  |
| 2 | 2.404(3) | 2.533(3) | 2.488(3) | 2.439(3) |  |  |  |  |  |  |
| 3 | 2.438(4) | 2.438(4) | 2.235(4) |  |  |  |  |  |  |  |
| 4 |  |  | 2.320(3) | 2.233(3) | 2.355(3) | 2.322(3) |  | 2.376(3) |  |  |
| 5 |  |  | 2.255(3) | 2.398(3) | 2.374(3) |  | 2.314(4) | 2.453(4) | 2.548(3) | 2.276(3) |
| 6 | 2.230(3) |  | 2.285(3) |  |  |  |  |  |  |  |
| 7 | 2.265(3) |  | 2.268(3) |  |  |  |  |  |  |  |
| 8 |  |  | 2.241 (6) | 2.388(6) | 2.361(6) |  | 2.383(6) | 2.435(7) | 2.451(7) | 2.263(7) |
| 9 |  |  | 2.260(3) | 2.409(3) | 2.406(3) |  | 2.346(3) | $2.438(4)$ | 2.436(3) | 2.310(4) |
| 10 | 2.290(4) |  | 2.240(4) |  |  |  |  |  |  |  |
| 11 |  |  | 2.264(4) | 2.402(5) | 2.390(4) |  | 2.336(4) | 2.451(4) | 2.428(4) | 2.293(4) |
| 12 |  |  | 2.226(4) | 2.398(4) | 2.390(4) |  | 2.391(4) | 2.440(4) | $2.445(5)$ | 2.244(5) |

Table S3: U-N Bond Lengths in $\mathrm{UO}_{2}{ }^{2+}$ complexes (1-12).

| Complex | $\mathrm{d}_{\mathrm{U}-\mathrm{N} 1}$ <br> $[\AA]$ | $\mathrm{d}_{\mathrm{U}-\mathrm{N} 2}$ <br> $[\AA]$ | $\mathrm{d}_{\mathrm{U}-\mathrm{N} 3}$ <br> $[\AA]$ |
| :--- | :---: | :---: | :---: |
| 1 | $2.552(3)$ | $2.600(3)$ |  |
| 2 | $2.664(4)$ | $2.705(4)$ |  |
| 3 | $2.561(5)$ | $2.540(5)$ |  |


| 4 | $2.673(4)$ | $2.663(4)$ |  |
| :--- | :--- | :--- | :--- |
| 5 | $2.592(4)$ | $2.590(4)$ | $2.580(4)$ |
| 6 | $2.558(4)$ | $2.600(3)$ | $2.587(3)$ |
| 7 | $2.576(3)$ | $2.595(4)$ | $2.560(3)$ |
| 8 | $2.561(8)$ | $2.583(8)$ | $2.575(8)$ |
| 9 | $2.581(5)$ | $2.603(4)$ | $2.574(4)$ |
| 10 | $2.568(5)$ | $2.604(4)$ | $2.589(5)$ |
| 11 | $2.568(6)$ | $2.587(5)$ | $2.560(5)$ |
| 12 | $2.580(5)$ | $2.585(5)$ | $2.564(5)$ |

## V. Bond Valence Summations

Table S4: Bond Valence Summations for oxygen atoms in Compound 4

| O5 | Distance $(\AA)$ | Bond Valence <br> Sum | O6 | Distance $(\AA)$ | Bond Valence <br> Sum |
| :--- | :---: | :---: | :--- | :---: | :---: |
| Bound atoms |  |  | Bound atoms |  |  |
| U1 | 2.320 | 0.5886 | U1 | 2.233 | 0.6961 |
| U2 | 2.355 | 0.5502 | U2 | 2.322 | 0.5864 |
|  | Sum | 1.138 | U2 | 2.236 | 0.6921 |
|  |  |  |  | Sum | 1.974 |
| OW1 |  |  |  |  |  |
| Bound atoms |  |  |  |  |  |
| U2 | 2.595 | 0.3465 |  |  |  |
|  | Sum | 0.3465 |  |  |  |

Bond valence summations for selected oxygen atoms in 4. The values indicated that O 5 is a hydroxyl group while O6 is an oxide group. ${ }^{1,2}$

Table S5: Bond Valence Summations for hydroxide oxygen atom in Compound 5

| O5 | Distance $(\AA)$ | Bond Valence Sum |
| :--- | :---: | :---: |
| Bound atoms |  |  |
| U1 | 2.225 | 0.7069 |
| U2 | 2.374 | 0.5305 |
|  | Sum | 1.237 |

Table S6: Bond Valence Summations for hydroxide oxygen atom in Compound $\mathbf{8}$

| O5 | Distance $(\AA)$ | Bond Valence Sum |
| :--- | :---: | :---: |
| Bound atoms |  |  |
| U1 | 2.241 | 0.6854 |
| U2 | 2.361 | 0.5439 |
|  | Sum | 1.229 |

Table S7: Bond Valence Summations for hydroxide oxygen atom in Compound 9

| O5 | Distance $(\AA)$ | Bond Valence Sum |
| :--- | :---: | :---: |
| Bound atoms |  |  |
| U1 | 2.260 | 0.6608 |
| U2 | 2.406 | 0.4987 |
|  | Sum | 1.159 |

Table S8: Bond Valence Summations for oxygen atoms in Compound 11

| O5 | Distance $(\AA)$ ) | Bond Valence Sum |
| :--- | :---: | :---: |
| Bound atoms | 2.264 |  |
| U1 | 2.390 | 0.6557 |
| U2 | Sum | 0.5144 |
|  | 1.170 |  |

Table S9: Bond Valence Summations for oxygen atoms in Compound 12

| O5 | Distance $(\AA)$ | Bond Valence Sum |
| :--- | :---: | :---: |
| Bound atoms |  |  |
| U1 | 2.264 | 0.7055 |
| U2 | 2.390 | 0.5144 |
|  | Sum | 1.219 |

## VI. References

1. N. E. Brese and M. O'Keeffe, Acta Crystallographica Section B, 1991, 47, 192197.
2. P. C. Burns, R. C. Ewing and F. C. Hawthorne, The Canadian Mineralogist, 1997, 35, 1551-1570.
