

Electronic Supplementary Information

Experimental and Computational Insights into the Conformations of Tunicyclin E, A New Cycloheptapeptide from *Psammosilene tunicoides*

Jun-Mian Tian,^{a,f} Si-Sheng Ou-Yang,^{b,f} Xu Zhang,^c Ying-Tong Di,^d Hua-Liang Jiang,^{b,e}
Hong-Lin Li,^e Wei-Xing Dai,^a Ke-Yu Chen,^a Mai-Li Liu,^c Xiao-Jiang Hao,^d Yun-Heng
Shen,^{*,a} Cheng Luo,^{*,b} Wei-Dong Zhang^{*,a}

^a Department of Phytochemistry, School of Pharmacy, Second Military Medical
University, 325 Guohe Road, Shanghai 200433, PR China.

Fax: +86-21-81871244; Tel: +86-21-81871244;

E-mail: wdzhangy@hotmail.com; shenyunheng9217018@yahoo.com.cn

^b Drug Discovery and Design Center, Shanghai Institute of Materia Medica, Chinese
Academy of Sciences, Shanghai 201203, China,

E-mail: cluo@mail.shcnc.ac.cn

^c State Key Laboratory of Magnetic Resonance and Atomic and Molecular Physics,
Wuhan Institute of Physics and Mathematic, Chinese Academy of Sciences, Wuhan
430071, China,

^d State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming
Institute of Botany, Chinese Academy of Sciences, Kunming 650204, China,

^e School of Pharmacy, East China University of Science and Technology, Shanghai
200237, China

^f These authors contributed equally to this work

Contents	Page number
Title of manuscript and author list	S1
The structure elucidation of tunicyclin E (1)	S3
Identification the absolute configuration of amino acid residues of tunicyclin E (1)	S5
The crystal data of tunicyclin E (1)	S6
Determination of the key interproton distances of 1a and 1b in solution by off-resonance experiment	S8
Determination of the thermodynamics and kinetics parameters of <i>cis/trans</i> isomerization of 1 by inversion-magnetization transfer experiment	S9
ESI-MS spectrum of tunicyclin E (1)	S13
HR-ESI-MS spectrum of tunicyclin E (1)	S14
¹ H NMR spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S15
¹³ C NMR spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S16
DEPT spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S17
COSY spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S18
HSQC spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S19
HMBC spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S20
TOCSY spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S21
ROESY spectrum of tunicyclin E (1) (in C ₅ D ₅ N)	S22

The structure elucidation of tunicyclin E (**1**)

Compound **1** was isolated as colorless crystal ($[\alpha]_{\text{D}}^{20} - 2$, c 0.13, MeOH) with the molecular formula $\text{C}_{35}\text{H}_{50}\text{N}_8\text{O}_9$ as established by negative HR-ESI-MS (m/z $[\text{M}-\text{H}]^-$ 725.3625, calcd 725.3623). Two sets of proton resonances with a ratio of 3:1 in ^1H NMR spectrum implicated the presence of two conformations in solution (**1a** and **1b**). By 1D and 2D NMR experiments, two sets of NMR data (**1a** and **1b**) were unambiguously assigned. In **1a**, the correlation between the amide proton (δ_{H} 10.03, 1 H) and two α -H (δ_{H} 4.84, 1 H and 3.88, 1 H) in ^1H - ^1H COSY and TOCSY spectra indicated that the presence of a Gly residue. The chemical shift of α -C of Gly residue was assigned as 43.80 ppm from the HSQC spectrum. The chemical shift of carbonyl carbon of Gly residue was assigned as 169.36 ppm based on the HMBC correlation of α -H (δ_{H} 4.84 and 3.88) and Gly-CO (δ_{C} 169.36). A spin system comprized with the amide proton (δ_{H} 9.36, 1 H) and other protons of two methyl (δ_{H} 1.05, 3 H and 1.02, 3 H) and two methine (δ_{H} 4.31, 1 H and 2.28, 1 H) in TOCSY spectrum indicated that **1a** also contained a Val residue. The chemical shifts of α -C, β -C, γ -C and γ' -C of Val residue were respectively assigned as 61.18, 30.05, 19.25, 19.44 ppm, based on the HSQC correlations. Furthermore, the carbonyl carbon of Val residue was undoubtedly assigned to δ_{C} 172.74 based on HMBC correlations between the carbonyl carbon and β protons of Val residue (δ_{H} 2.28). The spin system comprized with the protons of α -H (δ_{H} 4.75, 1 H), β -H (δ_{H} 2.07, 1 H), β' -H (δ_{H} 1.92, 1 H), γ -H (δ_{H} 1.70, 1 H), γ' -H (δ_{H} 1.50, 1 H), δ -H (δ_{H} 3.62, 1 H) and δ' -H (δ_{H} 3.40, 1 H) in TOCSY spectrum, corresponding α -C (δ_{C} 61.25), β -C (δ_{C} 28.39), γ -C (δ_{C} 24.92) and δ -C (δ_{C} 47.31) determined by HSQC spectrum, indicated the presence of Pro residue in **1a**. The HMBC correlation between β -H (δ_{H} 2.07, 1 H), β' -H (δ_{H} 1.92, 1 H) of Pro and the carbonyl carbon (δ_{C} 172.74) determined the chemical shift of carbonyl carbon of Pro residue. Based on the TOCSY correlations between the amide proton (δ_{H} 8.15) and the protons of α -H (δ_{H} 4.90, 1 H), β -H (δ_{H} 1.95, 1 H), β' -H (δ_{H} 1.87, 1 H), γ -H (δ_{H} 1.78, 1 H), δ -H (δ_{H} 0.76, 3 H) and δ' -H (δ_{H} 0.73, 3 H), a Leu residue could be deduced in **1a**. The chemical shifts of carbons of Leu residue also were assigned to δ_{C} 53.40 (α -C), 40.64 (β -C), 24.79 (γ -C) 22.93 (δ -C) and 21.52 (δ' -C) from HSQC spectrum. The chemical shift of carbonyl carbon of Leu also was assigned to δ_{C} 174.49 by the

HMBC correlation between β -H (δ_{H} 1.95, 1 H), β' -H (δ_{H} 1.87, 1 H) of Leu and the carbonyl carbon (δ_{C} 174.49). The presence of the four protons (δ_{H} 7.92 d, $J = 7.80$, 1 H; 7.07 dd, $J = 7.80$, 7.14, 1 H; 7.16 dd, $J = 7.98$, 7.14, 1 H; and 7.49 d, $J = 7.98$, 1 H) and eight carbons (δ_{C} 124.50, 111.23, 128.02, 118.93, 119.09, 121.61, 111.83, 137.33) indicated the presence of a 3-substituted indolyl group. The HMBC correlations of proton resonance at δ_{H} 3.87 (2H, d, $J = 7.14$ Hz, H- β of Trp) with carbon resonances at δ_{C} 111.23, 124.50 and 128.02 identified the Trp residue. Furthermore, the carbonyl carbons of Trp was undoubtedly assigned to δ_{C} 172.60 based on HMBC correlations between carbonyl carbons and β protons of Trp residue. The presence of two hydroxymethylene signals in DEPT spectrum suggested that there two Ser residues in **1a** (δ_{C} 171.31, 56.41, and 62.14, and δ_{C} 170.38, 53.49, and 63.80). The two Ser residues could be confirmed by their spin system in TOCSY spectrum based on the correlations between the amide proton (δ_{H} 8.41, 1 H) and the protons of α -H (δ_{H} 4.97, 1 H), and β -H (δ_{H} 4.17, 2 H) and the correlations between the amide proton (δ_{H} 8.63, 1 H) and the protons of α -H (δ_{H} 5.36, 1 H), and β -H (δ_{H} 4.27, 2 H), β' -H (δ_{H} 4.08, 1 H). Thus, all the chemical shifts of the seven amino acid residues of **1a** were assigned (Table 1 in manuscript).

The HMBC correlation between Ser²-NH (δ_{H} 8.41, 1 H) and Pro¹-CO (δ_{C} 171.98) indicated that the amide group of Ser² residue attached to the carbonyl group of Pro¹ residue *via* an amide bond. The linkage of Ser²-Trp³ was also confirmed by the HMBC correlation of Trp³-NH/CO-Ser². Furthermore, based on the HMBC correlation between Leu⁴-NH (δ_{H} 8.15, 1 H) and Trp³-CO (δ_{C} 172.60), The Leu⁴ residue was assigned to attach at the Trp³ residue. The linkages of Val⁵-Leu⁴, Gly⁶-Val⁵ and Ser⁷-Gly⁶ were also determined by the following HMBC crosspeaks: Val⁵-NH/CO-Leu⁴, Gly⁶-NH/CO-Val⁵ and Ser⁷-NH/CO-Gly⁶ (Fig. 3 in manuscript). The plane structure of **1a** was deduced as cyclo(Pro¹-Ser²-Trp³-Leu⁴-Val⁵-Gly⁶-Ser⁷). Furthermore, the ROESY correlations of Pro¹- α H/NH-Ser², Pro¹- δ H/NH-Ser², Ser²-NH/NH-Trp³, Trp³-NH/NH-Leu⁴, Leu⁴- α H/NH-Val⁵, Val⁵- α H/NH-Gly⁶, Gly⁶-NH/NH-Ser⁷, and Ser⁷- α H/ δ H, δ' H-Pro¹ also supported the gross structure (Fig. 4 in manuscript). The presence of strong NOE correlations between α proton of Ser⁷ and both of δ , δ' protons of Pro¹ suggested that the amide bond of Ser⁷-Pro¹ was *trans*. The β and γ carbon chemical shifts of Pro¹ at 28.39

and 24.92 ppm further supported the presence of *trans* peptidyl-prolyl bond. The amino acid sequence of **1a** was established as cyclo(Pro¹-Ser²-Trp³-Leu⁴-Val⁵-Gly⁶-Ser⁷)

In **1b**, the same amino acid residues of Pro, Ser, Trp, Leu, Val, Gly, and Ser as those in **1a** were also deduced by 1D and 2D NMR experiments. The complete assignment for the ¹H and ¹³C NMR data of **1b** (Table 2 in manuscript) was similar that of **1a**. In addition, based on the HMBC correlations of Ser²-NH (δ_{H} 8.83, 1 H) and Pro¹-CO (δ_{C} 172.74), Trp³-NH (δ_{H} 9.73, 1 H) and Ser²-CO (δ_{C} 173.44), Leu⁴-NH (δ_{H} 8.45, 1 H) and Trp³-CO (δ_{C} 172.60), Val⁵-NH (δ_{H} 7.81, 1 H) and Leu⁴-CO (δ_{C} 173.67), Gly⁶-NH (δ_{H} 8.00, 1 H) and Val⁵-CO (δ_{C} 171.60), and Ser⁷-NH (δ_{H} 9.13, 1 H) and Gly⁶-CO (δ_{C} 170.63) (Fig. 3 in manuscript), the amino acid sequence of **1b** was then established to be identical with that of **1a**. The result was also supported by the ROESY correlations of Pro¹- $\delta\text{H}/\text{NH-Ser}^2$, Trp³-NH/NH-Leu⁴, Leu⁴- $\alpha\text{H}/\text{NH-Val}^5$, Val⁵-NH/NH-Gly⁶, Gly⁶-NH/NH-Ser⁷, and Ser⁷- $\alpha\text{H}/\alpha\text{H-Pro}^1$ (Fig. 4 in manuscript). However, the β and γ carbon chemical shifts of Pro¹ in **1b** at δ_{C} 31.59 and 22.50 ppm respectively suggested the presence of *cis* peptidyl-prolyl bond in **1b**.⁵ This was further confirmed by the strong NOE correlation between α proton of Pro¹ and α proton of Ser⁷. The result indicated that **1a** and **1b** were *cis/trans* peptidyl-prolyl bond isomers.

Identification the absolute configuration of amino acid residues of tunicyclin E (1)

50 μL of a 50 mM aqueous solution of L-configurations of Pro, Ser, Leu and Val were added 20 μL of 1 M sodium bicarbonate and then 100 μL of 25 mM L-FDLA (TCI, Japan) in acetone, respectively. The solutions were incubated at 37 °C for 60 min. Reactions were quenched by addition of 20 μL of 1 N HCl, respectively. Samples were diluted with 810 μL of acetonitrile, and 400 μL of the solutions were analyzed by HPLC-ESI-MS, respectively.

100 μL of 1 mg/mL **1** was hydrolyzed at 100 °C for 24 h by adding 200 μL of 10 N HCl. The solution was evaporated to dryness. Then, the residue was dissolved in 50 μL of water. The amino acid solution was added 20 μL of 1 M sodium bicarbonate and 50 μL of 25 mM L-FDLA in acetone. The solution was incubated at 37 °C for 60 min. Reaction

was quenched by addition of 20 μL of 1 N HCl. Sample was diluted with 810 μL of acetonitrile, and 400 μL of the solution was analyzed by HPLC-ESI-MS.

HPLC was performed on an Agilent 1100 system. Separations were carried out on a TSKgel ODS-100V column (150×4.6 mm i.d., 3 μm , TOSOH) maintained at 40 $^{\circ}\text{C}$. Acetonitrile-0.01 M trifluoroacetic acid (TFA) was used as mobile phase under a linear gradient elution mode (acetonitrile, 20-100%, 80 min). The flow rate was 1 mL/min with detection at 340 nm by photodiode array detection and ESI-MS. The mass spectrometer used was a LC/MSD Trap XCT mass spectrometer (Agilent, USA). The auxiliary and sheath gas nitrogen pressure were set at 10 unit and 35 psi, respectively, and the capillary was heated to 350 $^{\circ}\text{C}$. A mass range of m/z 100-1000 was covered with a scan time of 200 μs , and data were collected in negative ion mode.

The absolute configurations of Pro¹, Ser², Leu⁴, Val⁵ and Ser⁷ residues of **1** were all identified as L configurations by comparison the retention times and m/z values of the chiral derivatives of the amino acid residues in acid hydrolysate of **1** with those of corresponding standard L-configuration amino acids (see Table 1).² Taken together with the relative configuration, established by X-ray crystallography, all amino acid residues had the L (S) configuration. Thus, **1** was determined as cyclo(L-Pro¹-L-Ser²-L-Trp³-L-Leu⁴-L-Val⁵-L-Gly⁶-L-Ser⁷).

Table 1 Analysis of L-FDLA derivatives of acid hydrolysate of **1** and those of standard L-configuration amino acids by HPLC-ESI-MS

Amino acid	t_{RL} (min)	t_{R} (min)	$[\text{M-H}]^{-}$ (m/z)	absolute configurations
Pro	24.54	24.54	408	L
Ser	20.34	20.37	398	L
Leu	31.98	31.97	424	L
Val	29.10	28.90	410	L

The crystal data of tunicyclin E (**1**)

Table 2 Crystal data and structure refinement for **1**

molecular formula	$\text{C}_{35} \text{H}_{62} \text{N}_8 \text{O}_{15}$
-------------------	--

molecular weight	834.93
crystal system	Orthorhombic
space group	P 2 ₁ 2 ₁ 2 ₁
Z, molecules / unit cell	4
unit cell dimensions	a = 10.798 (17) Å b = 15.09 (2) Å c = 27.46 (4) Å
volume	4473 (12) Å ³
calculated density	1.240 g/cm ³
wavelength	0.71073 Å
Reflections collected / unique	17448 / 5080 [R(int) = 0.0791]
Final R indices [I>2σ(I)]	R ₁ = 0.0810, wR ₂ = 0.1884
Temperature	293(2) K

Table 3 Final atomic parameters ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1**

atom	x	y	z	U(eq)	atom	x	y	z	U(eq)
N ₁	5657 (1)	184 (1)	4490 (1)	52 (1)	C ₄ ^γ	9178 (2)	4777 (1)	3495 (1)	122 (1)
C ₁ ^α	4523 (1)	142 (1)	4779 (1)	57 (1)	C ₄ ^δ	10570 (2)	4577 (2)	3436 (1)	190 (1)
C ₁ ^β	4848 (2)	-567 (1)	5153(1)	90 (1)	C ₄ ^{δ'}	8893 (4)	5672 (2)	3284 (1)	197 (2)
C ₁ ^γ	6170 (2)	-467 (2)	5229 (1)	136 (1)	C ₄ [']	6479 (2)	3334 (1)	3008 (1)	63 (1)
C ₁ ^δ	6727 (2)	-230 (1)	4742 (1)	77 (1)	O ₄ [']	6252 (1)	2611 (1)	3210 (1)	75 (1)
C ₁ [']	4185 (1)	996 (1)	5036 (1)	47 (1)	N ₅	6272 (1)	3462 (1)	2534 (1)	77 (1)
O ₁ [']	3344 (1)	992 (1)	5335 (1)	65 (1)	C ₅ ^α	5759 (2)	2738 (1)	2238 (1)	72 (1)
N ₂	4829 (1)	1730 (1)	4924 (1)	45 (1)	C ₅ ^β	5581 (2)	3070 (2)	1709 (1)	102 (1)
C ₂ ^α	4603 (1)	2569 (1)	5178 (1)	46 (1)	C ₅ ^γ	4747 (2)	2453 (2)	1434 (1)	145 (1)
C ₂ ^β	5576 (1)	2757 (1)	5567 (1)	57 (1)	C ₄ ^{γ'}	6768 (2)	3217 (2)	1440 (1)	124 (1)
O ₂ ^γ	6736 (1)	2963 (1)	5383 (1)	71 (1)	C ₅ [']	6589 (2)	1926 (1)	2283 (1)	72 (1)
C ₂ [']	4373 (1)	3310 (1)	4817 (1)	45 (1)	O ₅ [']	7740 (1)	1972 (1)	2277 (1)	98 (1)
O ₂ [']	3351 (1)	3425 (1)	4645 (1)	98 (1)	N ₆	6013 (1)	1137 (1)	2315 (1)	79 (1)
N ₃	5319 (1)	3841 (1)	4699 (1)	46 (1)	C ₆ ^α	6647 (2)	314 (1)	2376 (1)	88 (1)
C ₂ ^α	5161 (1)	4633 (1)	4404 (1)	45 (1)	C ₆ [']	6914 (2)	33 (1)	2897 (1)	78 (1)
C ₂ ^β	5920 (1)	5401 (1)	4616 (1)	50 (1)	O ₆ [']	7382 (2)	-684 (1)	2986 (1)	134 (1)
N ₃ ^{1'}	5265 (1)	5832 (1)	5905 (1)	95 (1)	N ₇	6640 (1)	632 (1)	3239 (1)	66 (1)
C ₃ ^{2'}	5879 (2)	5402 (1)	5554 (1)	82 (1)	C ₇ ^α	6875 (1)	493 (1)	3750 (1)	58 (1)
C ₃ ^{3'}	5493 (1)	5705 (1)	5111 (1)	55 (1)	C ₇ ^β	7599 (1)	1266 (1)	3963 (1)	64 (1)
C ₃ ^{3a'}	4583 (1)	6364 (1)	5198 (1)	55 (1)	O ₇ ^γ	6865 (1)	1984 (1)	4113 (1)	74 (1)
C ₃ ^{4'}	3861 (1)	6893 (1)	4908 (1)	66 (1)	C ₇ [']	5624 (1)	412 (1)	4013 (1)	56 (1)
C ₃ ^{5'}	3029 (2)	7483 (1)	5128 (1)	104 (1)	O ₇ [']	4648 (1)	595 (1)	3809 (1)	77 (1)
C ₃ ^{6'}	2928 (2)	7524 (1)	5619 (1)	118 (1)	Solvent				

C_3^T	3642 (2)	7018 (1)	5920 (1)	105 (1)	O_W^1	3548 (2)	285 (1)	2882 (1)	143 (1)
$C_3^{7a'}$	4455 (2)	6443 (1)	5703 (1)	73 (1)	O_W^2	9667 (2)	3281 (1)	2083 (1)	206 (1)
C_3'	5493 (2)	4497 (1)	3874 (1)	63 (1)	O_W^3	5889 (2)	4962 (2)	6902 (1)	238 (1)
O_3'	4795 (1)	4739 (1)	3559 (1)	127 (1)	O_W^4	8043 (2)	-2141 (1)	2476 (1)	222 (1)
N_4	6616 (1)	4155 (1)	3774 (1)	59 (1)	O_W^5	9762 (2)	1623 (2)	2935 (1)	217 (1)
C_4^α	7081 (2)	4107 (1)	3273 (1)	67 (1)	O_W^6	3148 (3)	2666 (2)	3493 (1)	220 (2)
C_4^β	8483 (2)	4021 (1)	3268 (1)	84 (1)	O_W^7	3468 (6)	1386 (2)	2320 (2)	281 (2)

^a $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor

Determination of the key interproton distances of 1a and 1b in solution by off-resonance experiment

For determination of proton-proton distance, an off-resonance ROESY experiment was performed by employing ROESYPHPR.2 pulse sequence to suppress the TOCSY type magnetization transfer: $\pi = 200 \mu\text{s}$, $-\pi = 200 \mu\text{s}$, total mixing time 200 ms, zero filling in F2 to 8 K and in F1 to 2 K.

The quantitative measured the interproton-distance using the equation:

$$r_{ij} = r_{\text{ref}} \left(\frac{a_{\text{ref}} c_{\text{ref}}}{a_{ij} c_{ij}} \right)^{1/6}$$

The correction factor, c_{ij} , was given by

$$c_{ij} = \frac{1}{(\sin^2 \theta_i \sin^2 \theta_j)}$$

with

$$\tan \theta_i = \frac{\gamma B_1}{(\omega_i - \omega_0)}.$$

a_{ij} was the volume intergration of ROESY cross peak of nucleus i and j , B_1 was the field strength of locking field ($\gamma B_1 = 2500 \text{ Hz}$), and $(\omega_i - \omega_0)$ was the resonance offset of nucleus i . The interprotpon-distance were calibrated with respect to the distance of the

indole NH proton to the 7'H proton ($r_{\text{ref}} = 282 \text{ pm}$). All the unambiguous interproton-distance of **1a**, **1b** were obtained and shown in Table 4.

Table 4 The key interproton distances (in Å) obtained from the ROESY spectrum of **1a**, **1b** and those from crystal structure of **1**

1a		1b		Crystal	
Involved protons	distance	Involved protons	distance	Involved protons	distance
Pro ¹ H α – Ser ² NH	2.81	Pro ¹ H α – Ser ⁷ H α	2.20	Pro ¹ H α – Ser ² NH	3.10
Pro ¹ H δ – Ser ² NH	3.05	Pro ¹ H δ – Ser ² NH	3.03	Pro ¹ H δ – Ser ² NH	3.29
Pro ¹ H δ – Ser ⁷ H α	2.47	Ser ² H α – Trp ³ NH	2.27	Pro ¹ H δ – Ser ⁷ H α	2.72
Pro ¹ H δ – Ser ⁷ H β	2.15	Ser ² H β – Val ⁵ NH	3.12	Pro ¹ H δ – Ser ⁷ H β	2.10
Pro ¹ H δ' – Ser ⁷ H α	2.08	Ser ² H β – Gly ⁶ NH	3.14	Pro ¹ H δ' – Ser ⁷ H α	2.40
Pro ¹ H δ' – Ser ⁷ H β	3.00	Trp ³ NH – Leu ⁴ NH	2.39	Pro ¹ H δ' – Ser ⁷ H β	3.09
Ser ² NH – Trp ³ NH	2.99	Trp ³ H α – Leu ⁴ NH	2.82	Ser ² NH – Trp ³ NH	3.24
Ser ² NH – Ser ⁷ H β	3.30	Trp ³ H α – Gly ⁶ NH	3.47	Ser ² NH – Ser ⁷ H β	3.39
Ser ² H α – Trp ³ NH	2.73	Trp ³ H β' – Leu ⁴ NH	3.07	Ser ² H α – Trp ³ NH	3.40
Trp ³ NH – Leu ⁴ NH	2.69	Leu ⁴ NH – Val ⁵ NH	2.32	Trp ³ NH – Leu ⁴ NH	2.56
Trp ³ H α – Leu ⁴ NH	3.14	Leu ⁴ H α – Val ⁵ NH	3.03	Trp ³ H α – Leu ⁴ NH	3.49
Leu ⁴ H α – Val ⁵ NH	2.56	Val ⁵ NH – Gly ⁶ NH	2.34	Leu ⁴ H α – Val ⁵ NH	2.29
Val ⁵ H α – Gly ⁶ NH	2.23	Gly ⁶ NH – Val ⁵ H α	2.93	Val ⁵ H α – Gly ⁶ NH	2.41
Val ⁵ H α – Ser ⁷ NH	3.73	Gly ⁶ NH – Ser ⁷ NH	3.19	Val ⁵ H α – Ser ⁷ NH	3.48
Gly ⁶ NH – Ser ⁷ NH	2.28			Gly ⁶ NH – Ser ⁷ NH	2.49
Ser ⁷ H α – Ser ⁷ H β	2.28			Ser ⁷ H α – Ser ⁷ H β	2.22

Determination of the thermodynamics and kinetics parameters of *cis/trans* isomerization of 1 by inversion-magnetization transfer experiment

To clearly determine the thermodynamics and kinetics parameters, the pair of *cis/trans* proton resonance of Ser⁷-NH was subjected. The proton resonance of Ser⁷-NH in **1a** (*trans*) was selectively inverted with the pulse sequence: $90^\circ_x - \tau - 90^\circ_x - t - 90^\circ_{\pm x, \pm y}$ – acquisition, where the τ is a fixed delay equal to $1/(2|v_{\text{cis}} - v_{\text{trans}}|)$ and t is a variable delay during which transfer of magnetization takes place by interchange between the *cis* and *trans* isomers. The intensity of proton resonance of Ser⁷-NH in **1b** was measured as a function of t . t values were varied from 0.001 to 24 s. Analysis of the time-depended

intensity of the *cis/trans* pair of Ser⁷-NH to obtain rate constants was similar to that developed by Alger et al. (J. R. Alger, J. H. Prestegard, *J. Magn. Reson.* **1977**, 27, 137).

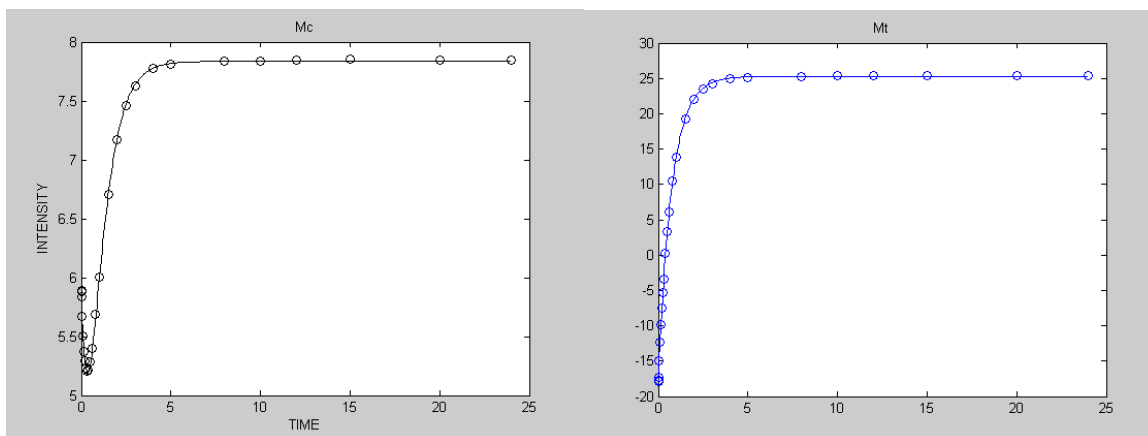


Fig. 1 Inversion-magnetization transfer of **1** at 300 K

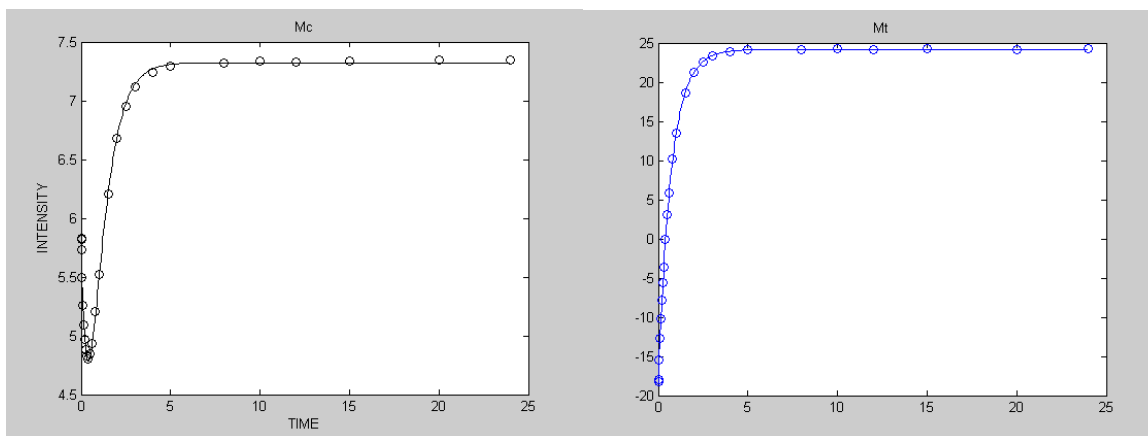


Fig. 2 Inversion-magnetization transfer of **1** at 305 K

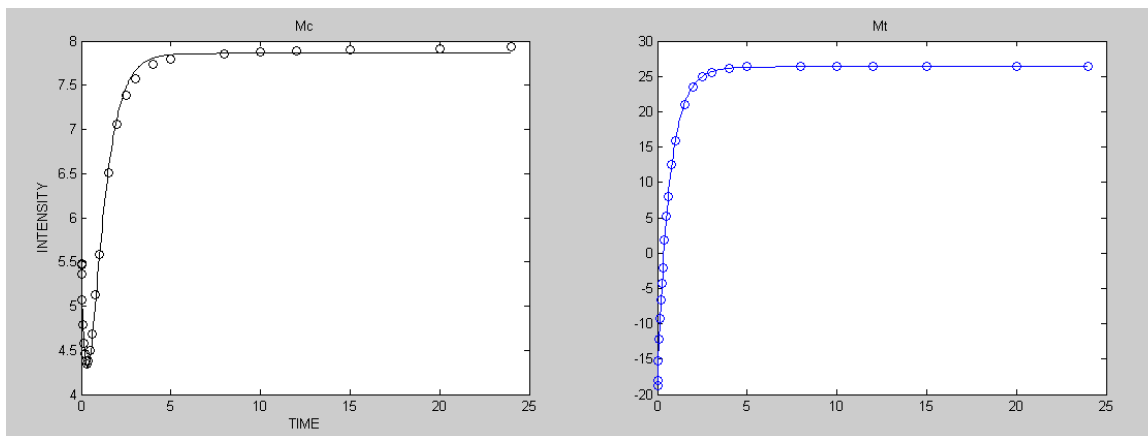


Fig. 3 Inversion-magnetization transfer of **1** at 310 K

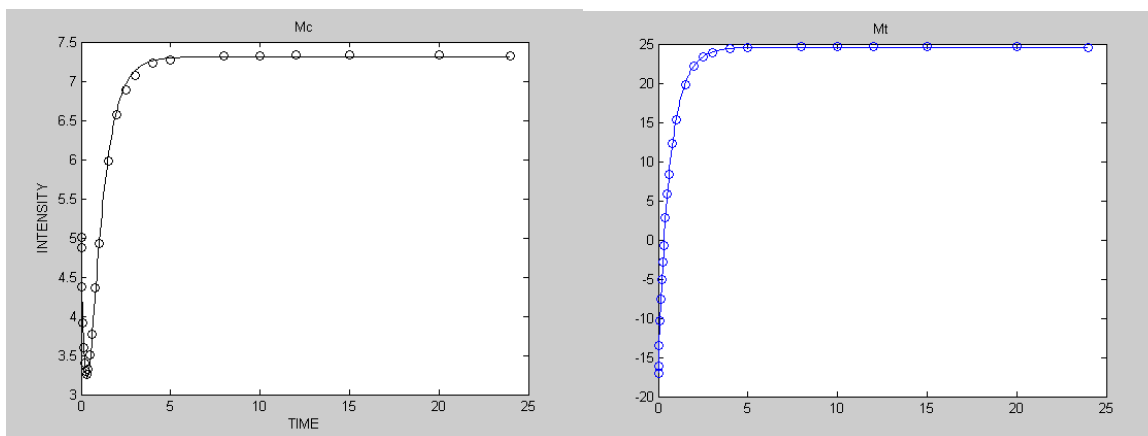


Fig. 4 Inversion-magnetization transfer of **1** at 315 K

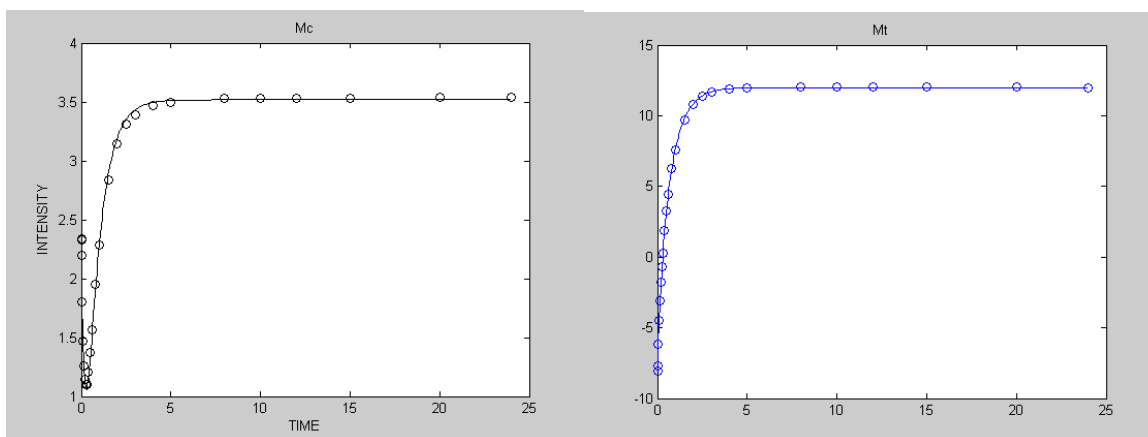


Fig. 5 Inversion-magnetization transfer of **1** at 320 K

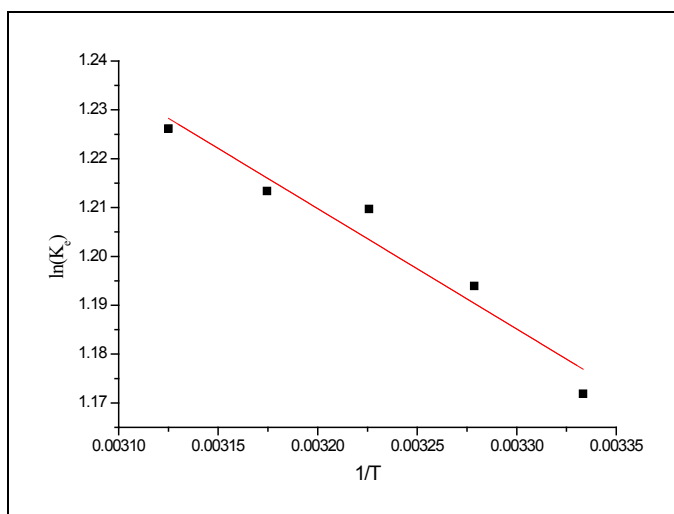


Fig. 6 Van't Hoff plot for the *cis/trans* isomerization of **1**

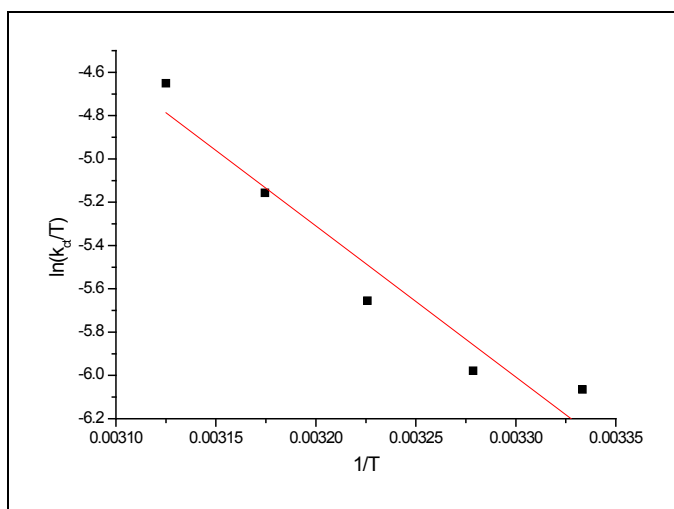


Fig. 7 Eyring plot for *cis*-to-*trans* isomerization of **1**

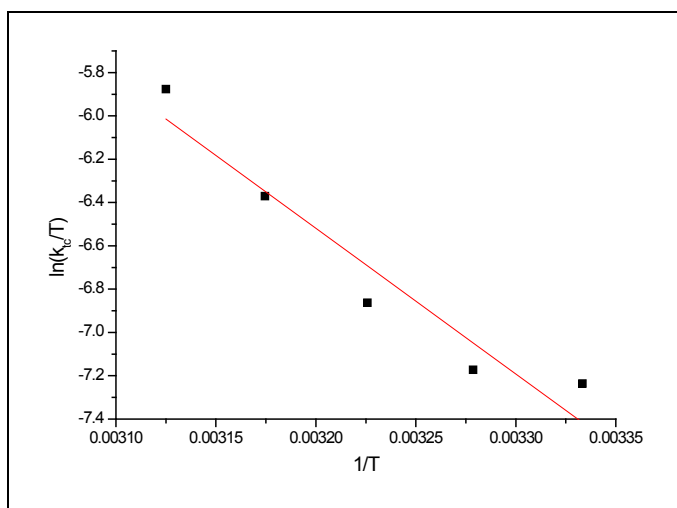
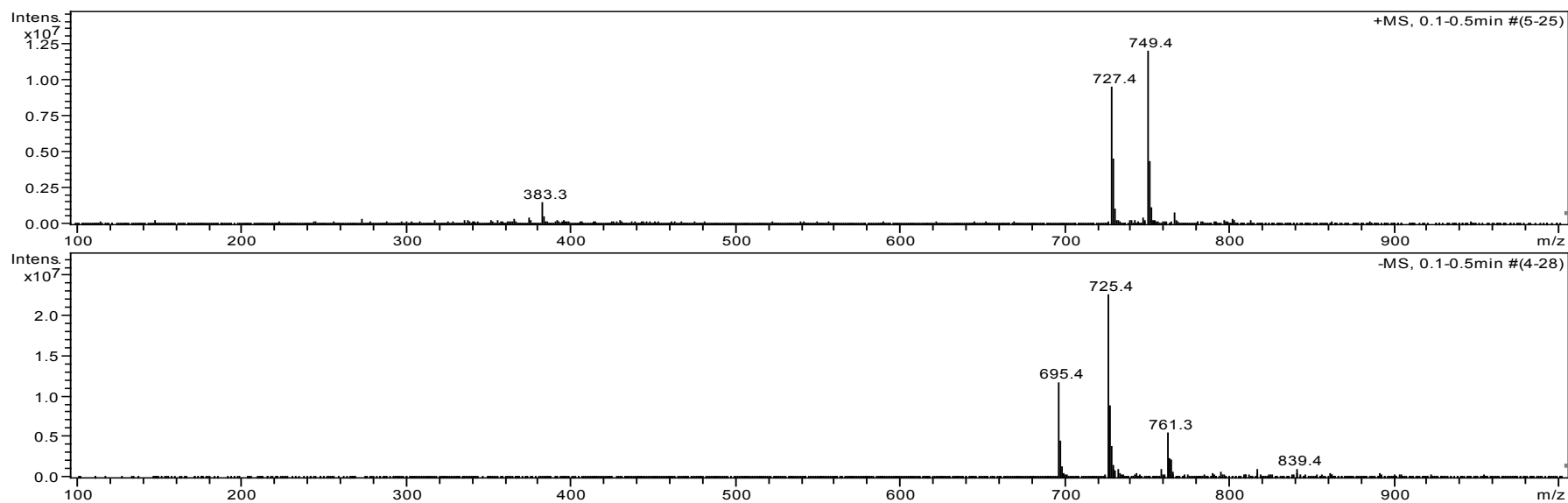


Fig. S8 Eyring plot for *trans*-to-*cis* isomerization of **1**

CC(C)[C@H](NC(=O)[C@H](C(C)C)NC(=O)NCC(=O)N[C@@H](CO)C(=O)N1CCCC1)C(=O)N[C@@H](Cc2c[nH]c3ccccc23)C(=O)N[C@@H](CO)C(=O)N[C@@H](C(C)C)C(=O)N[C@@H](C(C)C)C(=O)N

HR-ESI-MS spectrum of tunicyclin E (1)

Elemental Composition Report

Page 1

Single Mass Analysis

Tolerance = 50.0 PPM / DBE: min = -1.5, max = 50.0

Selected filters: None

Monoisotopic Mass, Even Electron Ions

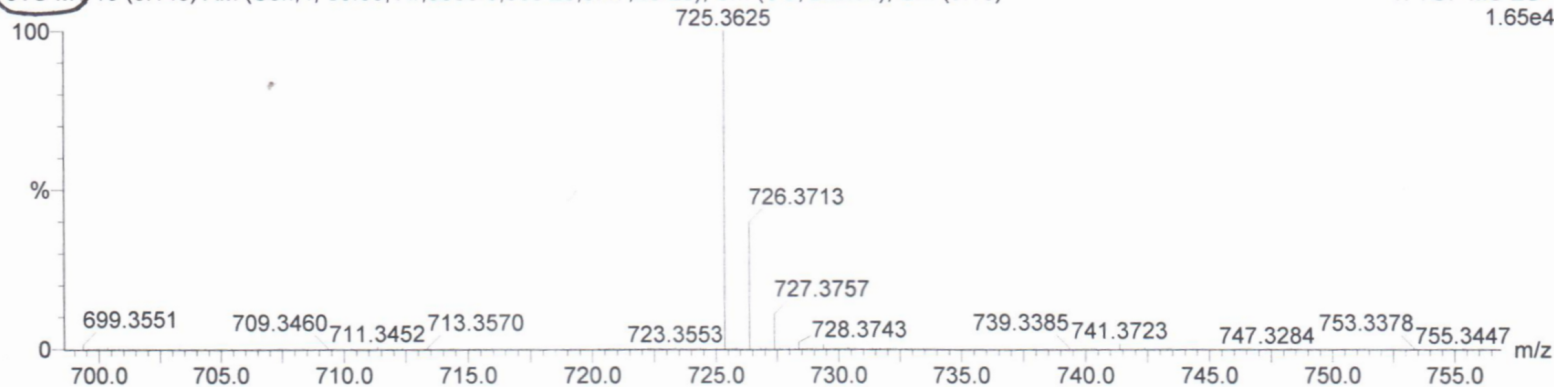
2 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

Elements Used:

C: 34-35 H: 47-50 N: 7-8 O: 9-10 Na: 0-1

JTS-M1 13 (0.143) AM (Cen,4, 80.00, Ar,5000.0,568.28,0.47,LS 20); Sm (SG, 2x3.00); Cm (6:16)

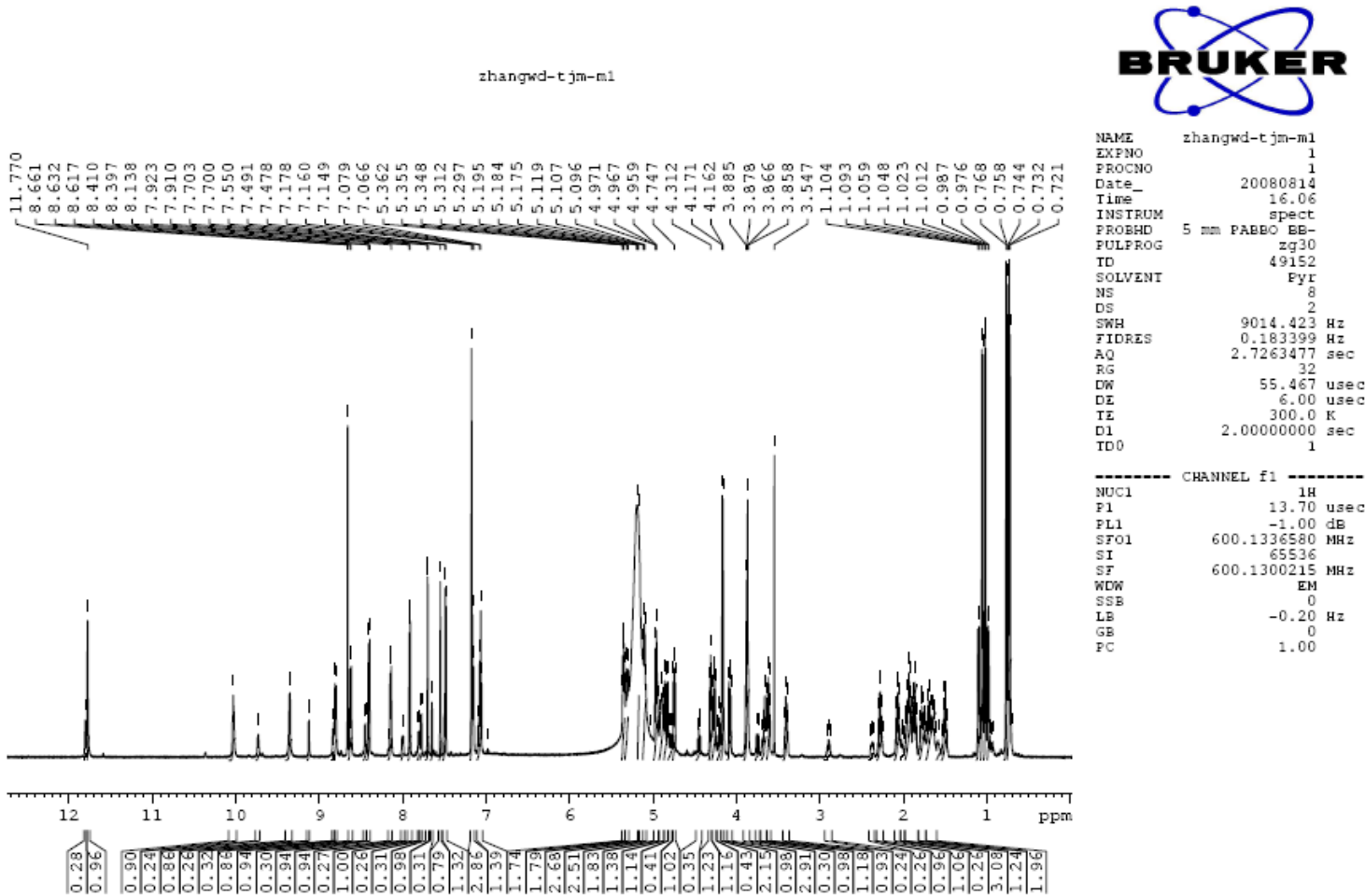
1: TOF MS ES-
1.65e4



Minimum: -1.5
Maximum: 5.0 50.0 50.0

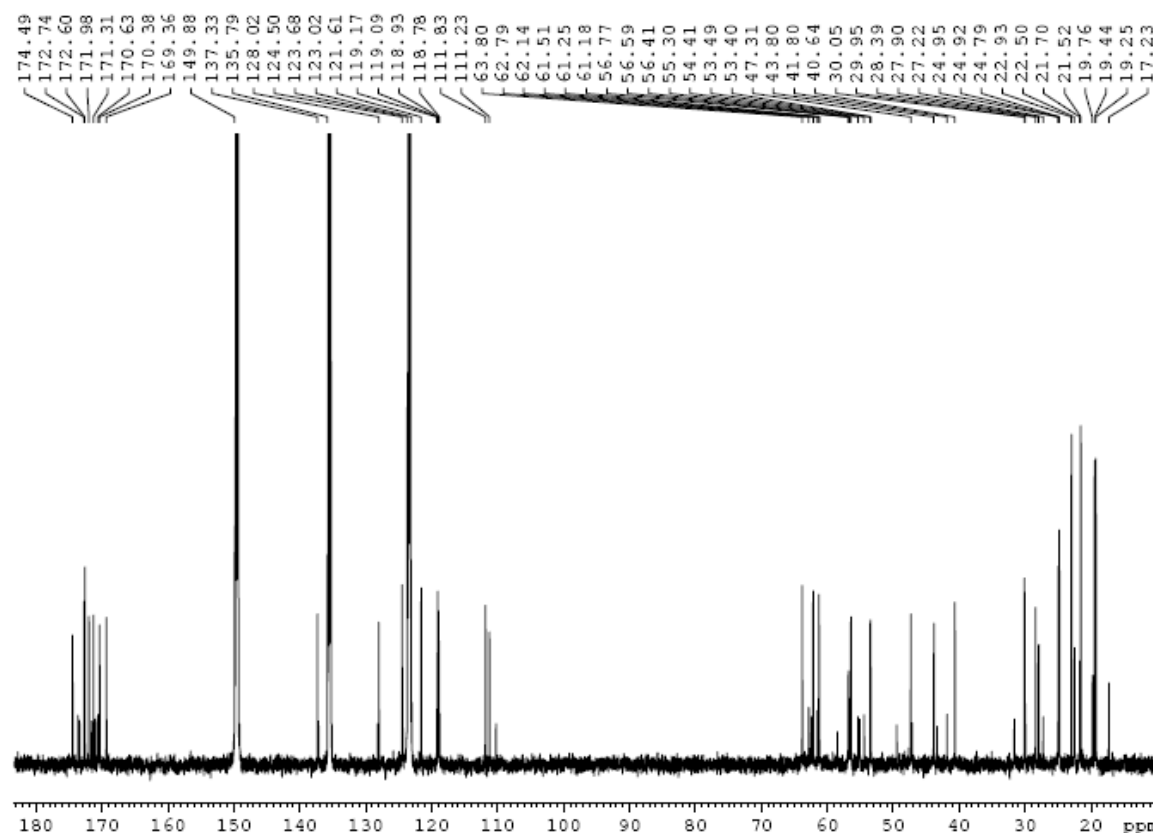
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
725.3625	725.3623	0.2	0.3	15.5	16.8	C35 H49 N8 O9

¹H NMR spectrum of tunicyclin E (1) (in C₅D₅N)



^{13}C NMR spectrum of tunicyclin E (1) (in $\text{C}_5\text{D}_5\text{N}$)

zhangwd-tjm-m1

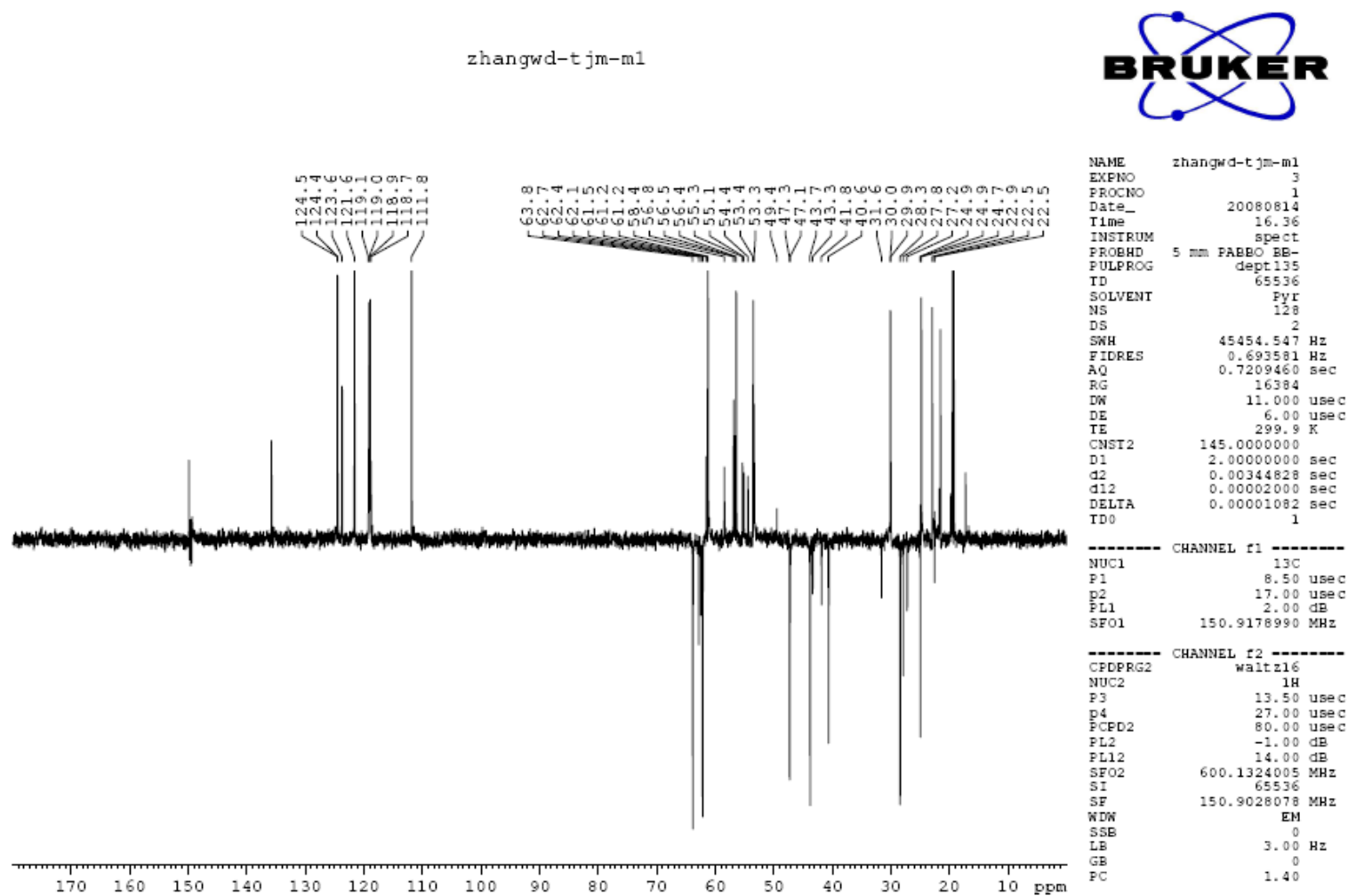


```
NAME      zhangwd-tjm-m1
EXPNO     2
PROCNO    1
Date_     20080814
Time      16.29
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   zgpg30
TD        65536
SOLVENT   Pyr
NS        446
DS        0
SWH       45454.547 Hz
FIDRES    0.693581 Hz
AQ        0.7209460 sec
RG        32768
DW        11.000 usec
DE        6.00 usec
TE        300.1 K
D1        2.00000000 sec
d11       0.03000000 sec
DELTA     1.89999998 sec
TD0       1
```

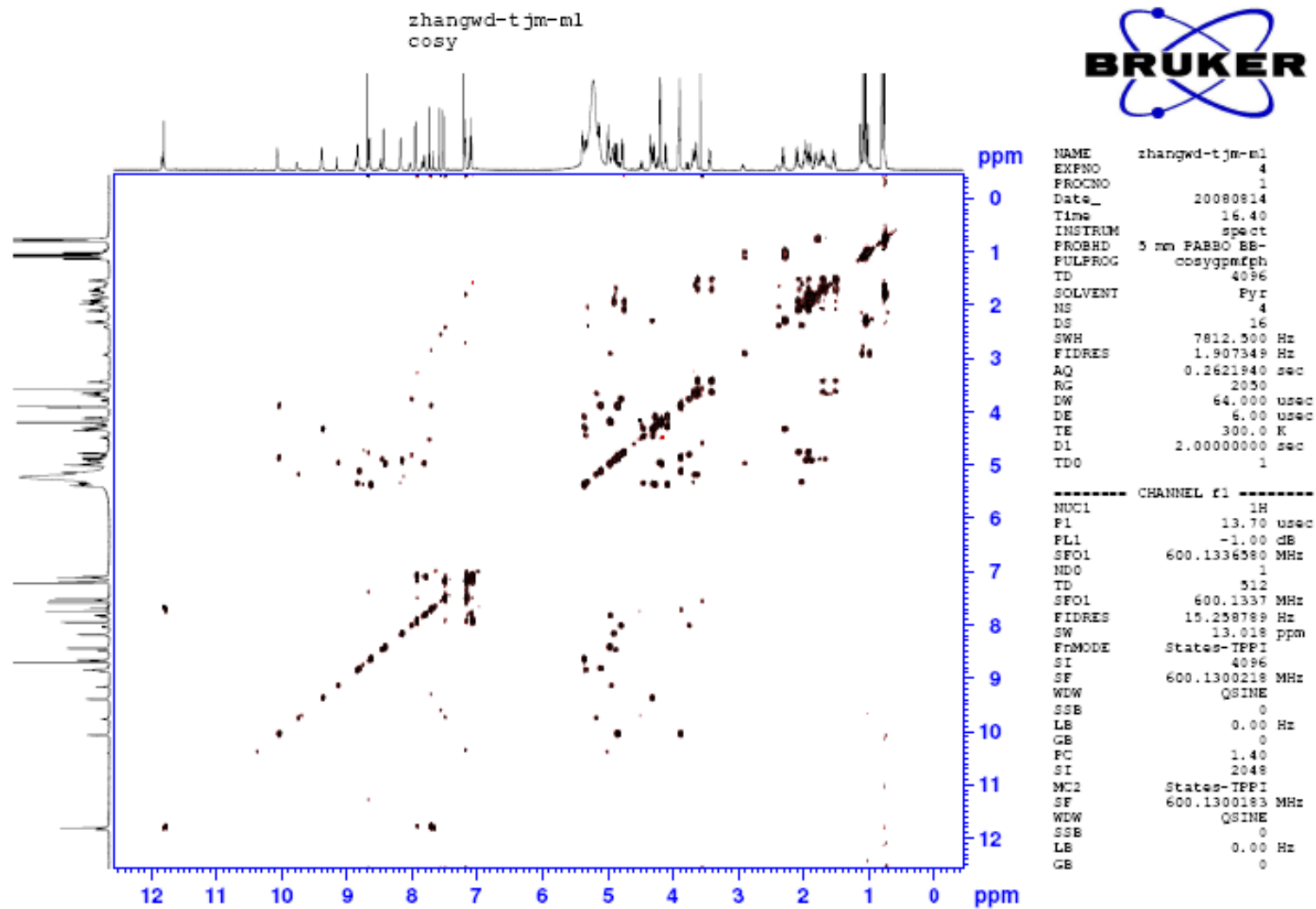
```
----- CHANNEL f1 -----
NUC1      13C
P1        8.50 usec
PL1       2.00 dB
SFO1      150.9178990 MHz
```

```
----- CHANNEL f2 -----
CPDPRG2   waltz16
NUC2      1H
PCPD2     80.00 usec
PL2       -1.00 dB
PL12      14.00 dB
PL13      14.00 dB
SFO2      600.1324005 MHz
SI        65536
SF        150.9028078 MHz
WDW       EM
SSB       0
LB        2.00 Hz
GB        0
PC        1.40
```

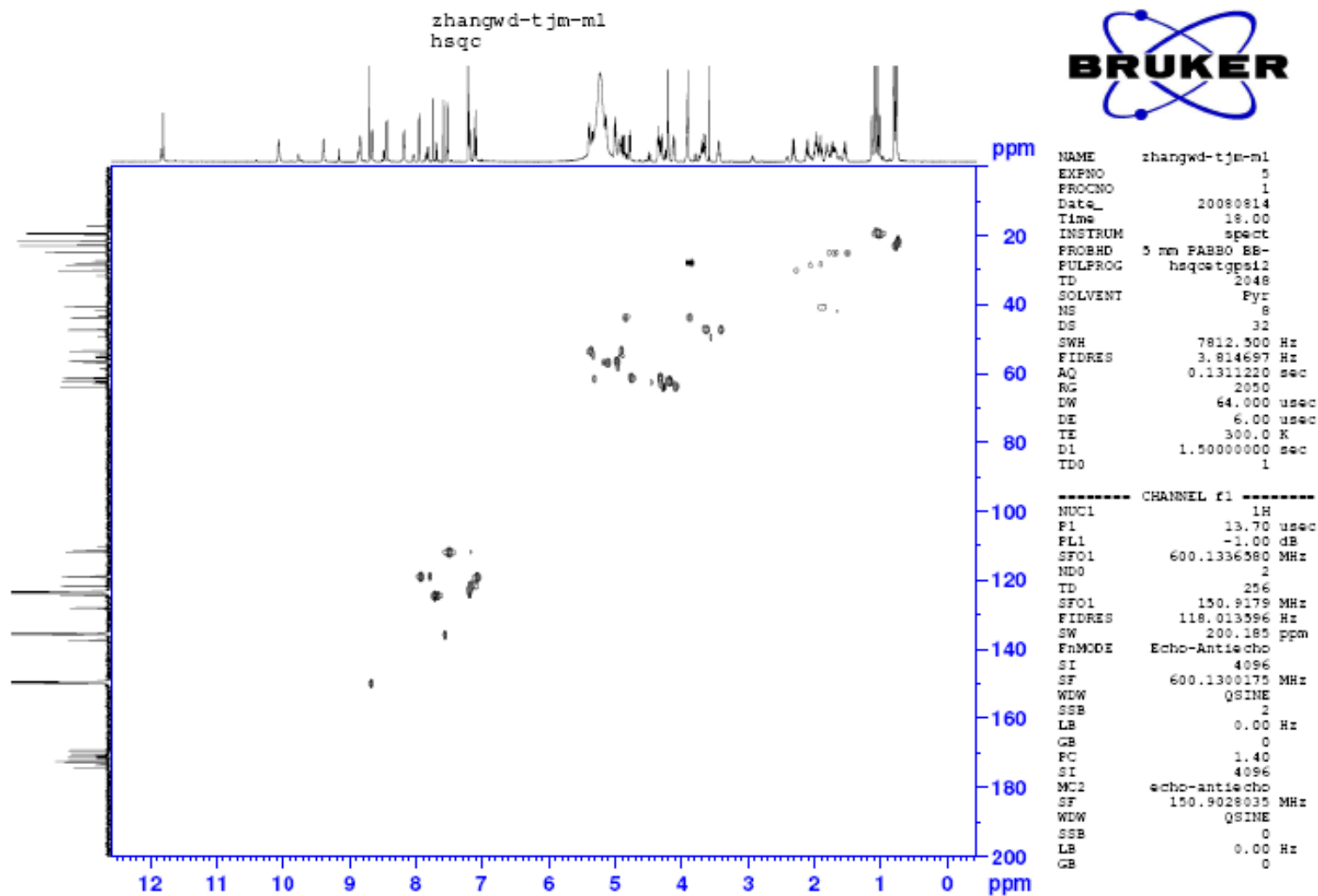
DEPT spectrum of tunicyclin E (**1**) (in C₅D₅N)



COSY spectrum of tunicyclin E (**1**) (in C₅D₅N)

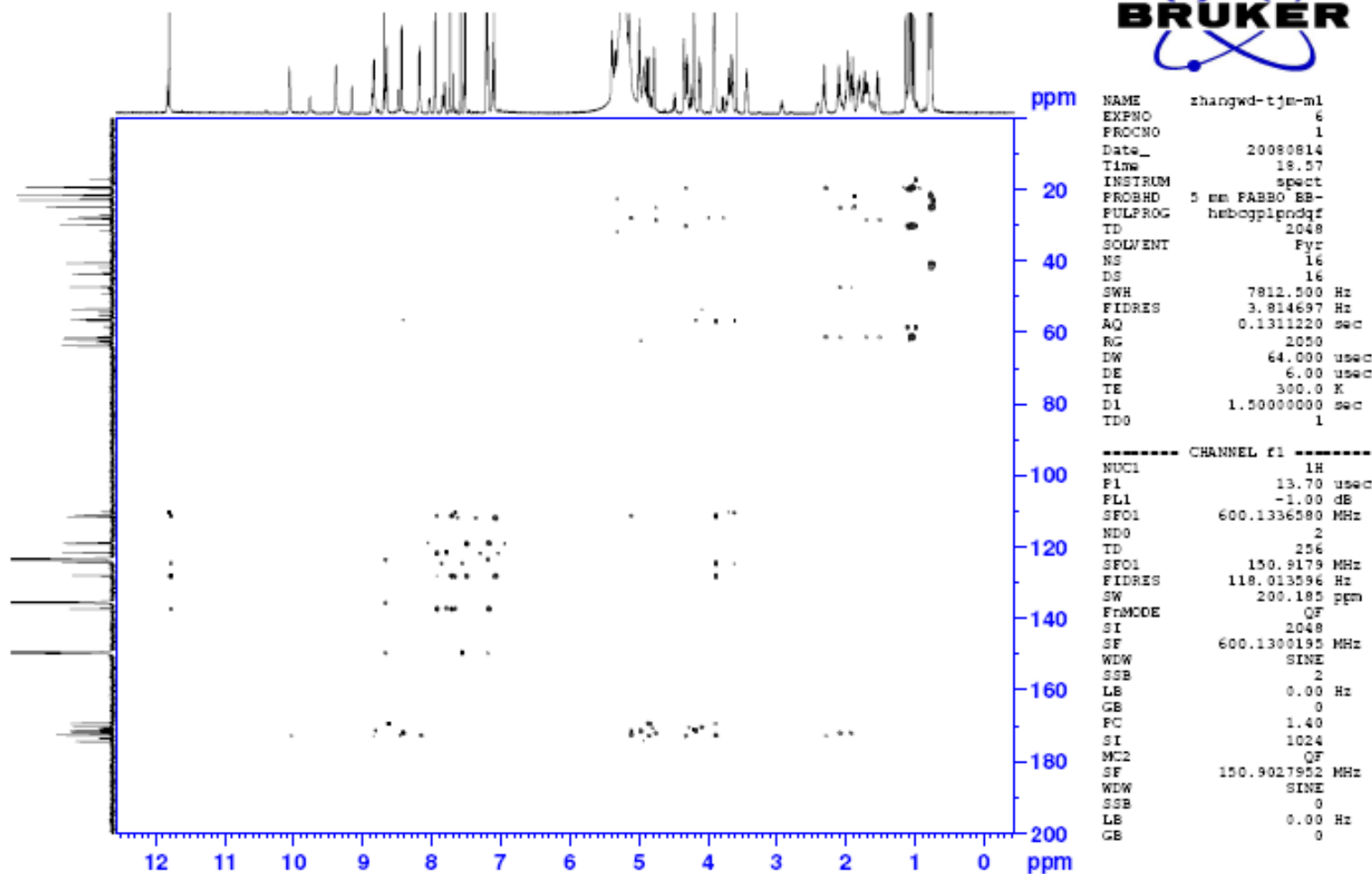


HSQC spectrum of tunicyclin E (1) (in C₅D₅N)

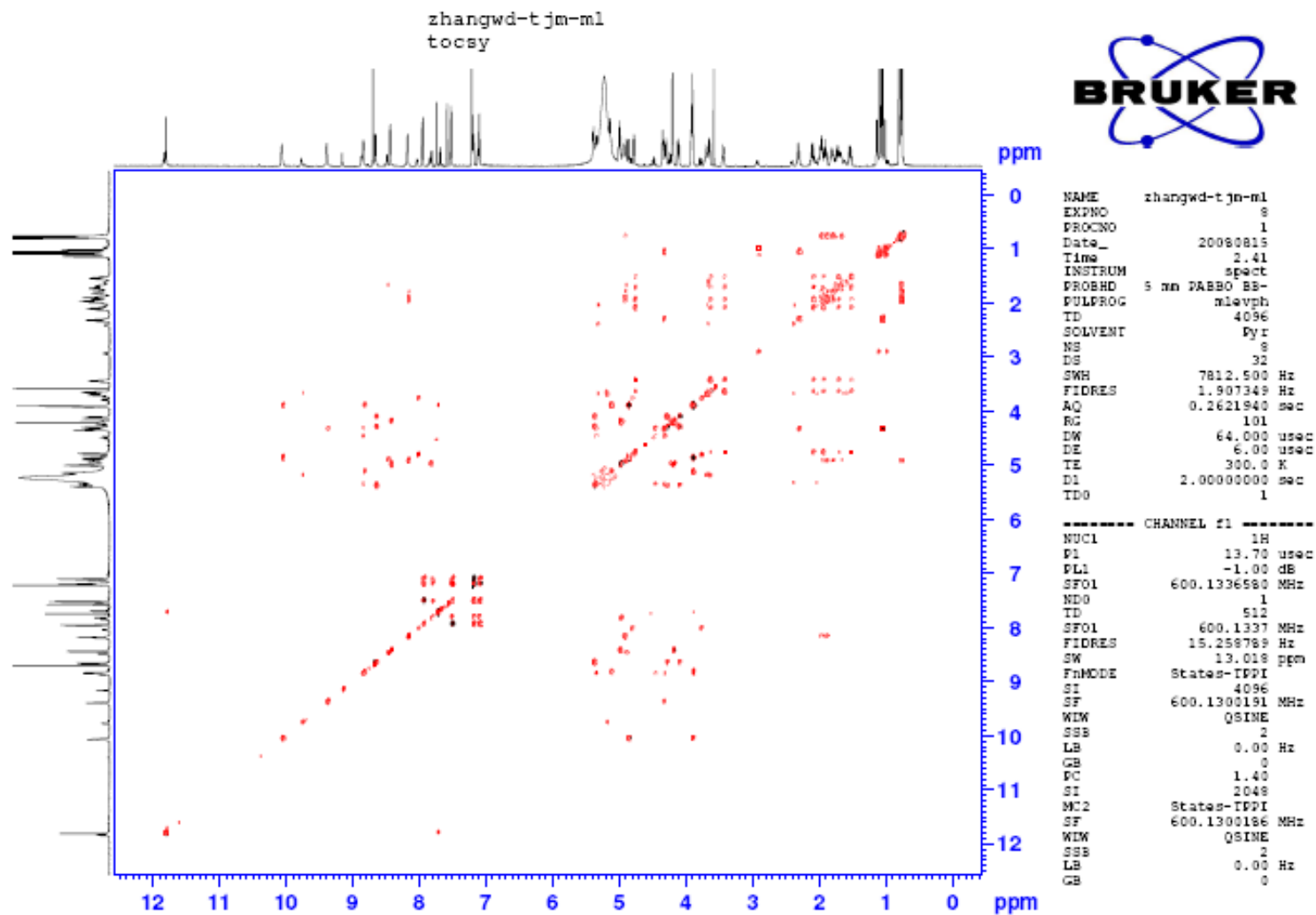


HMBC spectrum of tunicyclin E (1) (in C₅D₅N)

zhangwd-tjm-m1
hmbc



TOCSY spectrum of tunicyclin E (**1**) (in C₅D₅N)



ROESY spectrum of tunicyclin E (**1**) (in C₅D₅N)

