

## Supplementary Information

### F...HO intramolecular hydrogen bond forming five-membered rings hardly appears in monocyclic organofluorine compounds

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#### Table of Contents

Page 2. Selected geometric parameters for compounds **1-Cl** to **10-Cl** (interatomic distances in angstroms and bond angles in degrees).

Page 3. Electronic density and its Laplacian at the H-bond BCP ( $\rho_{\text{HBBCP}}$  and  $\nabla^2\rho_{\text{HBBCP}}$ , respectively) and hydroxyl hydrogen integrated atomic properties for the conformers **A** and **B** of 3-chlorobuta-1,3-dien-2-ol (**1-Cl**), obtained by QTAIM (in au).

Page 4. Geometric representations for the conformer **A** of compounds **1-Cl** to **10-Cl**.

Page 5. Molecular graphs of compounds **1-Cl** to **10-Cl** (conformer **A**). Green points represent bond critical points (BCP) and red points indicate ring critical points (RCP).

Page 6. Selected geometric parameters for compounds **1-Br** to **10-Br** (interatomic distances in angstroms and bond angles in degrees).

Page 7. Electronic density and its Laplacian at the H-bond BCP ( $\rho_{\text{HBBCP}}$  and  $\nabla^2\rho_{\text{HBBCP}}$ , respectively) and hydroxyl hydrogen integrated atomic properties for the conformers **A** and **B** of 3-bromobuta-1,3-dien-2-ol (**1-Br**), obtained by QTAIM (in au).

Page 8. Geometric representations for the conformer **A** of compounds **1-Br** to **10-Br**.

Page 9. Molecular graphs of compounds **1-Br** to **10-Br** (conformer **A**). Green points represent bond critical points (BCP) and red points indicate ring critical points (RCP).

Page 10. Structure and geometrical representation of 2-fluorophenol derivatives obtained from the Cambridge Structural Data.

Page 11. Selected geometric parameters for the reference compound **1** and of 2-fluorophenol derivatives obtained from the Cambridge Structural Database (interatomic distances in angstroms and bond angles in degrees).

Page 12. References of compounds obtained from the Cambridge Structural Database (CSD).

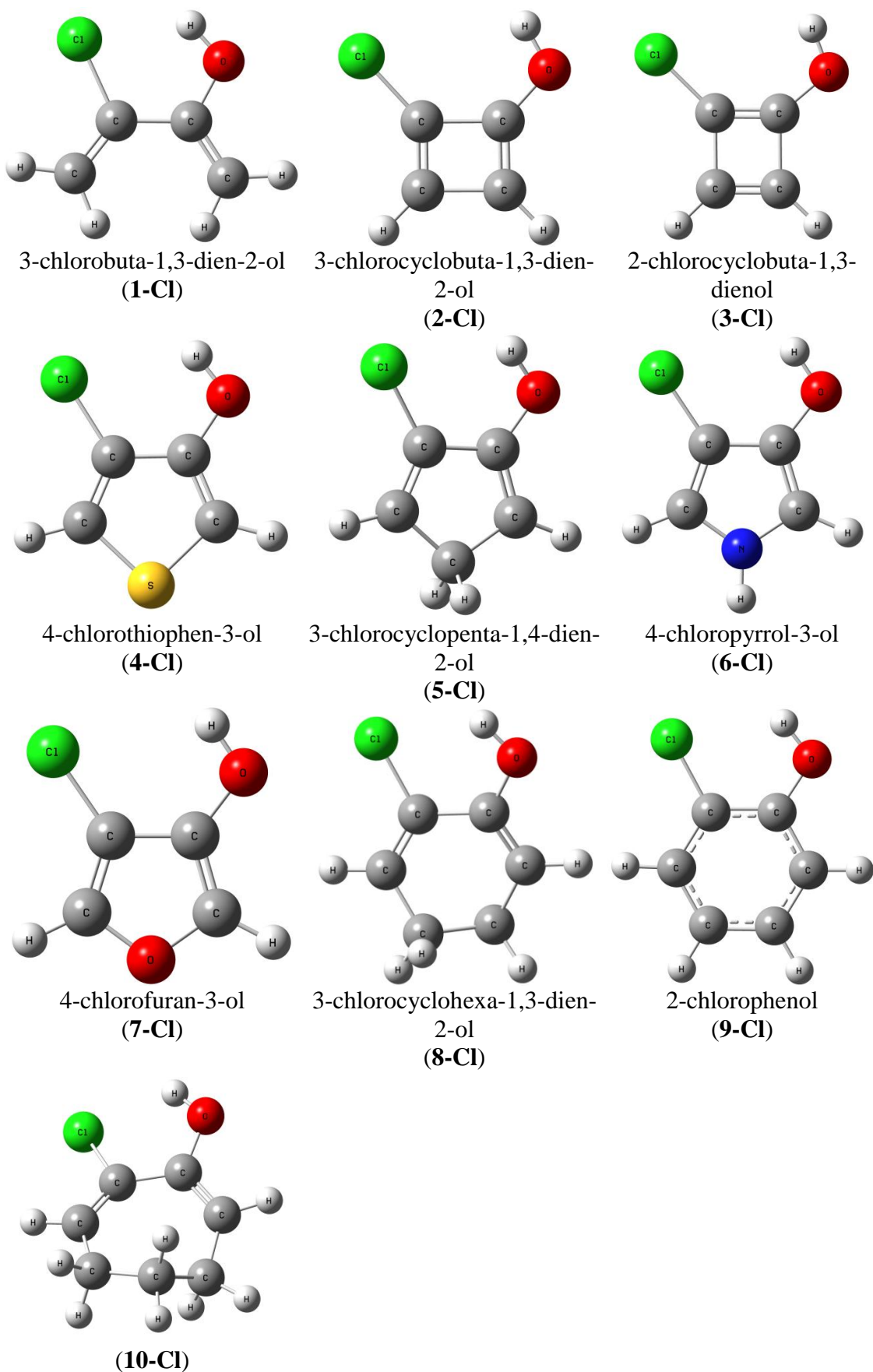
## Chlorine derivatives

**Table S1:** Selected geometric parameters for compounds **1-Cl** to **10-Cl** (interatomic distances in angstroms and bond angles in degrees).

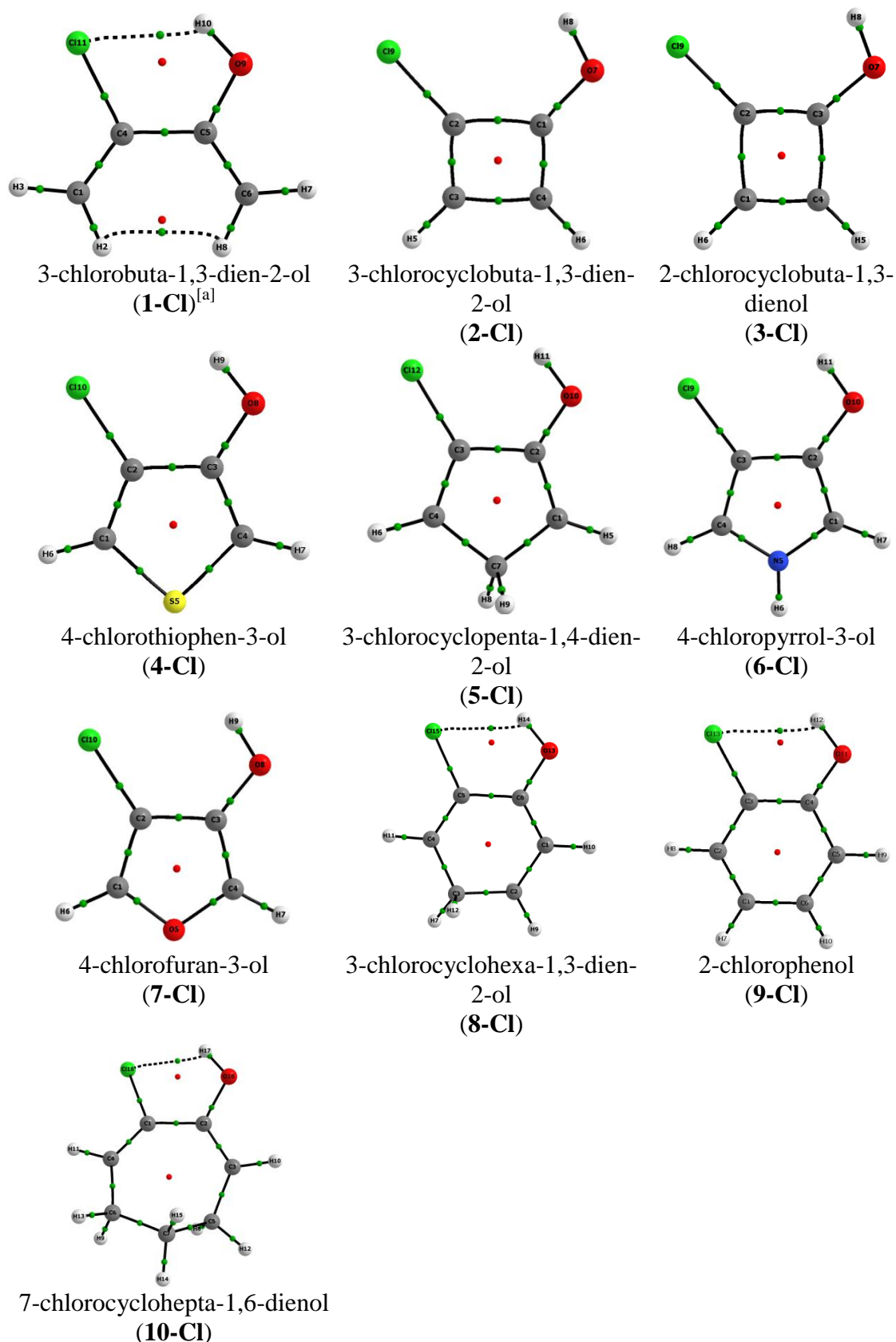
Compound	$r(\text{Cl}\cdots\text{HO})$	$r(\text{Cl}\cdots\text{O})$	$\angle \text{O-C-C(Cl)}$	$\angle \text{Cl-C-C(O)}$	$\angle \text{C-C(Cl)-C(O)}$	$\angle \text{C-C(O)-C(Cl)}$
<b>1-Cl</b>	<b>2.24</b>	<b>2.90</b>	<b>117.9</b>	<b>114.7</b>	<b>127.8</b>	<b>123.1</b>
<b>2-Cl</b>	3.23	3.65	132.2	132.3	89.8	90.8
<b>3-Cl</b>	3.31	3.62	139.7	135.7	89.1	90.7
<b>4-Cl</b>	2.56	3.13	124.4	120.9	114.2	111.8
<b>5-Cl</b>	2.58	3.15	123.8	121.0	110.9	108.5
<b>6-Cl</b>	2.72	3.25	127.4	124.2	108.5	107.0
<b>7-Cl</b>	2.74	3.26	127.4	124.4	107.2	105.8
<b>8-Cl</b>	2.36	2.99	119.7	116.6	122.4	118.9
<b>9-Cl</b>	2.41	3.01	123.1	118.2	121.6	118.4
<b>10-Cl</b>	2.27	2.92	117.6	113.1	131.1	124.3

**Table S2:** Electronic density and its Laplacian at the H-bond BCP ( $\rho_{\text{HBCP}}$  and  $\nabla^2\rho_{\text{HBCP}}$ , respectively) and hydroxyl hydrogen integrated atomic properties for the conformers **A** and **B** of 3-chlorobuta-1,3-dien-2-ol (**1-Cl**), obtained by QTAIM (in au).

<b>Conformer</b>	$\rho_{\text{HBCP}}$	$\nabla^2\rho_{\text{HBCP}}$	$q(\text{H})$	$E(\text{H})$	$M_1(\text{H})$	$V(\text{H})$
<b>A</b>	0.025	+0.075	+0.599	-0.3450	0.159	0.018
<b>B</b>	-	-	+0.591	-0.3524	0.169	0.013



**Figure S1:** Geometric representations for the conformer **A** of compounds **1-Cl** to **10-Cl**.



<sup>[a]</sup> H...H interaction is unstable (ellipticity = 1.370 au) and does not affect considerably the geometrical parameters of this compound.

**Figure S2.** Molecular graphs of compounds **1-Cl** to **10-Cl** (conformers A). Green points represent bond critical points (BCP) and red points indicate ring critical points (RCP).

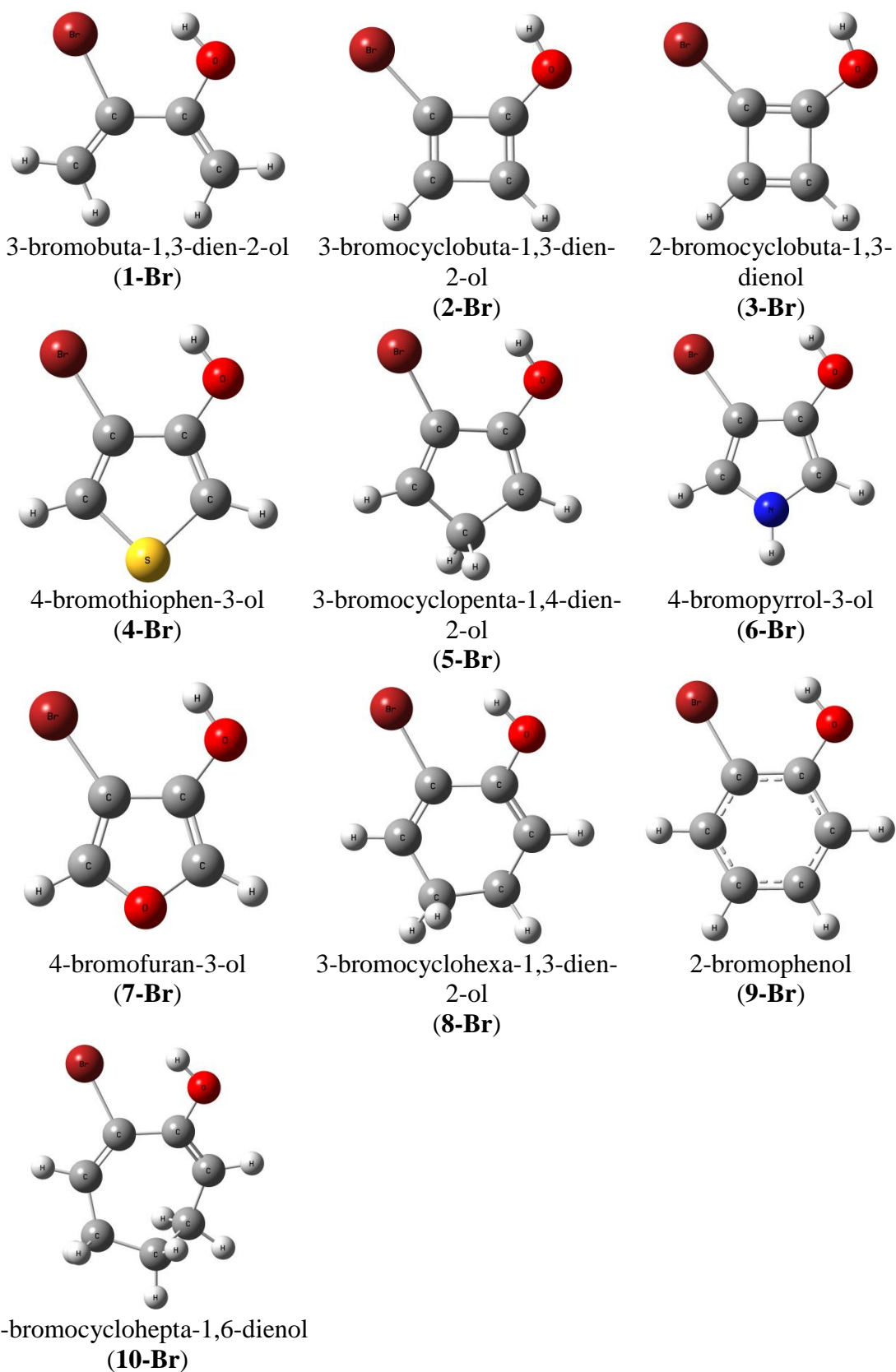
## Bromine derivatives

**Table S3:** Selected geometric parameters for compounds **1-Br** to **10-Br** (interatomic distances in angstroms and bond angles in degrees).

Compound	$r(\text{Br}\cdots\text{HO})$	$r(\text{Br}\cdots\text{O})$	$\angle \text{O-C-C}(\text{Br})$	$\angle \text{Br-C-C}(\text{O})$	$\angle \text{C-C}(\text{Br})-\text{C}(\text{O})$	$\angle \text{C-C}(\text{O})-\text{C}(\text{Br})$
<b>1-Br</b>	<b>2.34</b>	<b>3.03</b>	<b>118.5</b>	<b>115.4</b>	<b>127.6</b>	<b>122.9</b>
<b>2-Br</b>	3.31	3.76	132.1	132.6	89.5	90.9
<b>3-Br</b>	3.38	3.72	139.5	135.4	88.8	90.9
<b>4-Br</b>	2.64	3.24	124.8	121.2	114.1	111.8
<b>5-Br</b>	2.66	3.26	124.1	121.3	110.8	108.5
<b>6-Br</b>	2.80	3.36	127.6	124.3	108.4	107.0
<b>7-Br</b>	2.82	3.37	127.5	124.6	107.0	105.8
<b>8-Br</b>	2.46	3.11	120.2	117.1	122.3	118.8
<b>9-Br</b>	2.50	3.13	123.6	118.6	121.5	118.4
<b>10-Br</b>	2.38	3.05	118.2	113.5	131.0	123.8

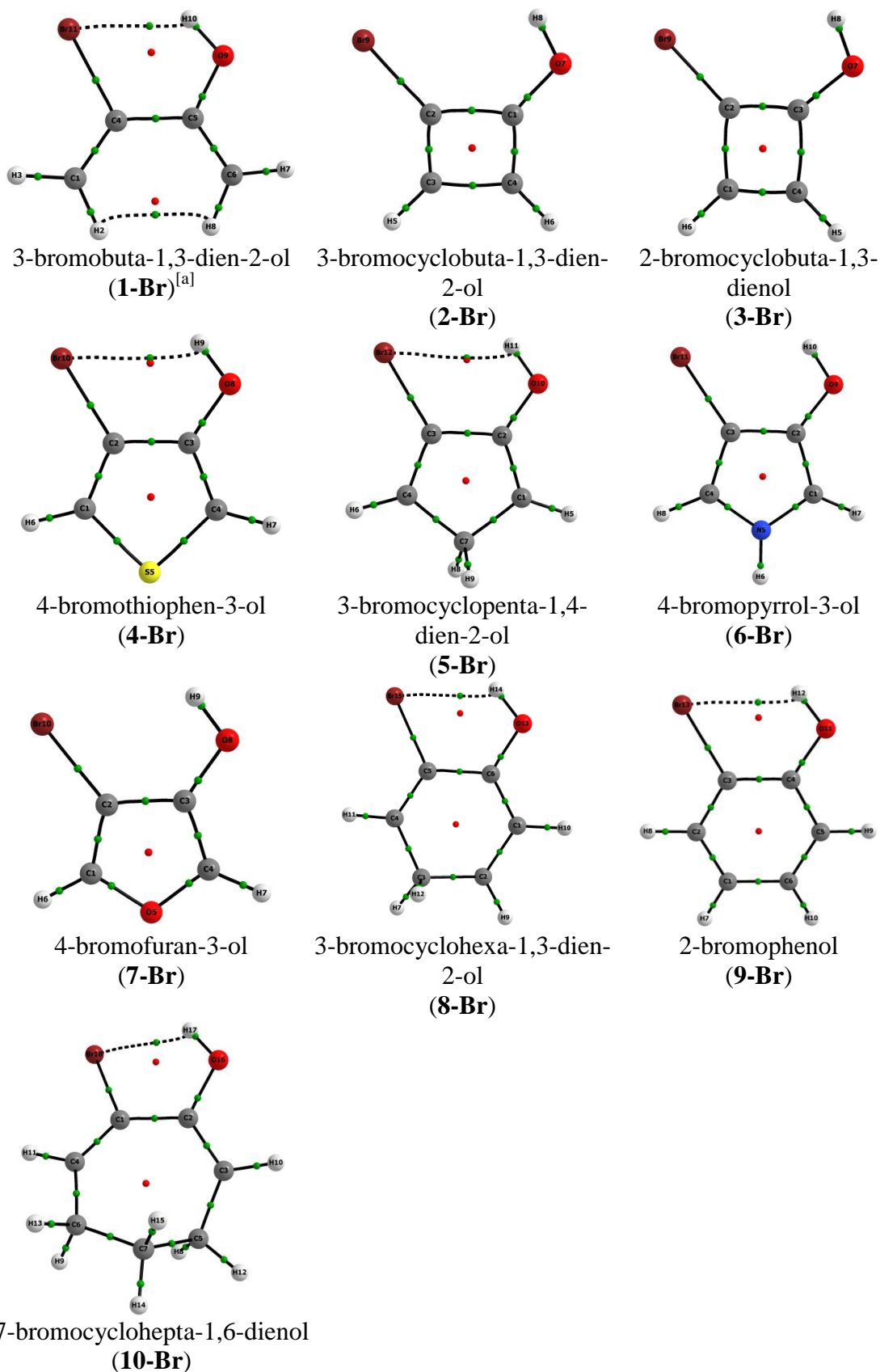
**Table S4:** Electronic density and its Laplacian at the H-bond BCP ( $\rho_{\text{HBCP}}$  and  $\nabla^2\rho_{\text{HBCP}}$ , respectively) and hydroxyl hydrogen integrated atomic properties for the conformers **A** and **B** of 3-bromobuta-1,3-dien-2-ol (**1-Br**), obtained by QTAIM (in au).

<b>Conformer</b>	$\rho_{\text{HBCP}}$	$\nabla^2\rho_{\text{HBCP}}$	$q(\text{H})$	$E(\text{H})$	$M_1(\text{H})$	$V(\text{H})$
<b>A</b>	0.024	+0.063	+0.591	-0.3470	0.165	0.018
<b>B</b>	-	-	+0.590	-0.3524	0.169	0.013



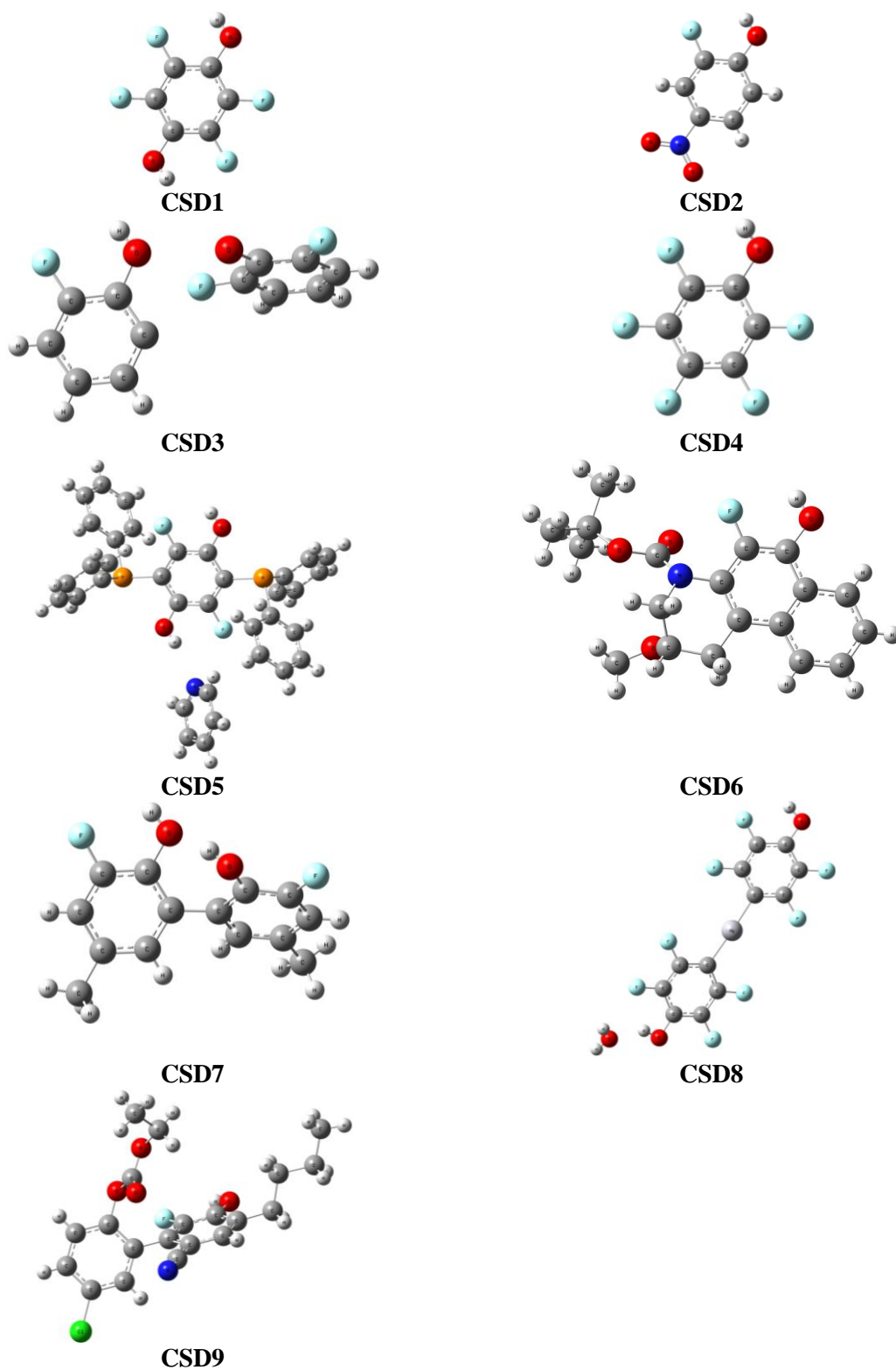
**Figure S3:** Geometric representations for the conformer A of compounds **1-Br** to **10-Br**.





<sup>[a]</sup> H...H interaction is unstable (ellipticity c.a. 1.000 au) and does not affect considerably the geometrical parameters in this compound.

**Figure S4.** Molecular graphs of compounds **1-Br** to **10-Br** (conformer A). Green points represent bond critical points (BCP) and red points indicate ring critical points (RCP).



**Figure S5:** Structure and geometrical representation of 2-fluorophenol derivatives obtained from the Cambridge Structural Database.

**Table S5:** Selected geometric parameters for the reference compound **1** and of 2-fluorophenol derivatives obtained from the Cambridge Structural Database (interatomic distances in angstroms and bond angles in degrees).

<b>Compound</b>	<b><math>r(\text{F}\cdots\text{HO})</math></b>	<b><math>r(\text{F}\cdots\text{O})</math></b>	<b><math>\angle\text{O-C-C(F)}</math></b>	<b><math>\angle\text{F-C-C(O)}</math></b>	<b><math>\angle\text{C-C(F)-C(O)}</math></b>	<b><math>\angle\text{C-C(O)-C(F)}</math></b>
<b>1</b>	<b>1.990</b>	<b>2.575</b>	<b>115.7</b>	<b>111.2</b>	<b>130.0</b>	<b>124.3</b>
<b>CSD1</b>	2.374	2.774	123.3	118.9	121.4	117.4
<b>CSD2</b>	2.319	2.738	122.6	116.8	123.1	118.2
<b>CSD3</b>	2.345	2.690	119.9	117.9	122.2	118.5
<b>CSD4</b>	2.447	2.774	123.1	119.0	121.7	117.6
<b>CSD5</b>	2.277	2.704	120.4	117.8	122.8	120.9
<b>CSD6</b>	2.335	2.725	122.8	118.1	122.8	118.9
<b>CSD7</b>	2.291	2.719	121.7	116.6	123.1	118.9
<b>CSD8</b>	2.278	2.798	124.5	117.9	121.4	117.2
<b>CSD9</b>	2.300	2.739	122.1	117.6	123.9	119.7

## References of compounds obtained from the Cambridge Structural Database (CSD):

**CSD1:** Thalladi, V. R.; Weiss, H. C.; Boese, R.; Nangia, A.; Desiraju, G. R. *Acta Cryst.* **1999**, *B55*, 1005.

**CSD2:** Chambers, R. D.; Fox, M. A.; Sandford, G.; Trmcic, G.; Goeta, A., *J. Fluor. Chem.*, **2007**, *128*, 29.

**CSD3:** Oswald, I. D. H.; Allan, D. R.; Motherwell, W. D. S.; Parsons, S., *Struct. Sci.*, **2005**, *61*, 69.

**CSD4:** Gdaniec, M., *Crist. Eng. Comm.*, **2007**, *9*, 286.

**CSD5:** Pignotti, L. R.; Kongprakaiwoot, N.; Brennessel, W. W.; Baltrusaitis, J.; Luck, R. L.; Urnezius, E., *J. Organomet. Chem.*, **2008**, *693*, 3263.

**CSD6:** Boger, D. L.; Brunette, S. R.; Garbaccio, R. M., *J. Org. Chem.*, **2001**, *66*, 5163.

**CSD7:** Kirste, A.; Nieger, M.; Malkowsky, I. M.; Stecker, F.; Fischer, A.; Waldvogel, S. R., *Chem. Eur. J.*, **2009**, *15*, 2273.

**CSD8:** Deacon, G. b.; Felder, P. W.; Junk, P. C.; Buschbaum, K. M.; Ness, T. J.; Quitmann, C. C., *Inorg. Chem. Acta*, **2005**, *358*, 4389.

**CSD9:** Ibad, M. F.; Abid, O. R.; Adeel, M.; Nawaz, M.; Wolf, V.; Villinger, A.; Langer, P., *J. Org. Chem.*, **2010**, *75*, 8315.