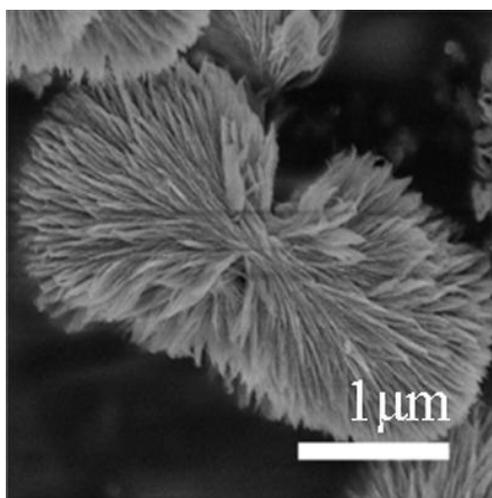


## Electronic Supplementary Information

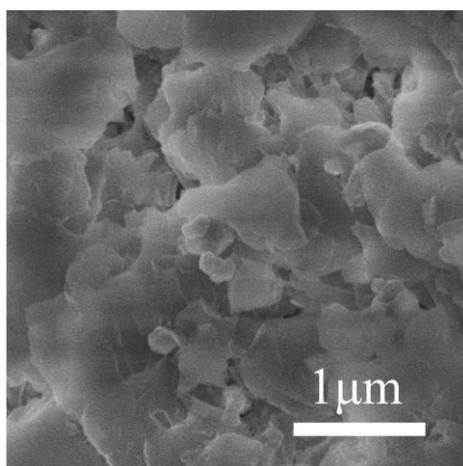
### Room temperature solution synthesis of hierarchical bows-like $\text{Cu}_2\text{O}$ with high visible-light driven photocatalytic activity

Xiangying Meng, Guohui Tian, Yajie Chen, Yang Qu, Juan Zhou, Kai Pan, Wei Zhou,  
Guoliang Zhang and Honggang Fu\*

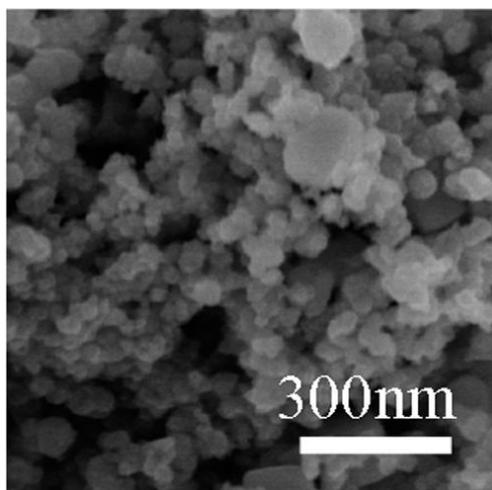
Key Laboratory of Functional Inorganic Material Chemistry, Ministry of Education of  
the People's Republic of China, Heilongjiang University, Harbin 150080 P. R. China



**Fig. S1** High- magnification SEM image of the hierarchical bows-like  $\text{Cu}_2\text{O}$ .



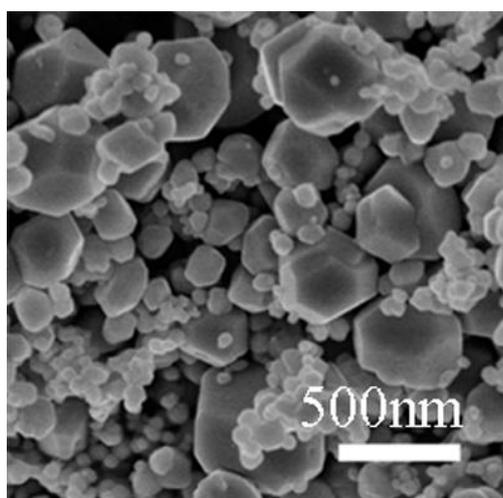
**Fig. S2** SEM images of the  $\text{Cu}_2\text{O}$  crystal in the presence of SDS (0.1 g).



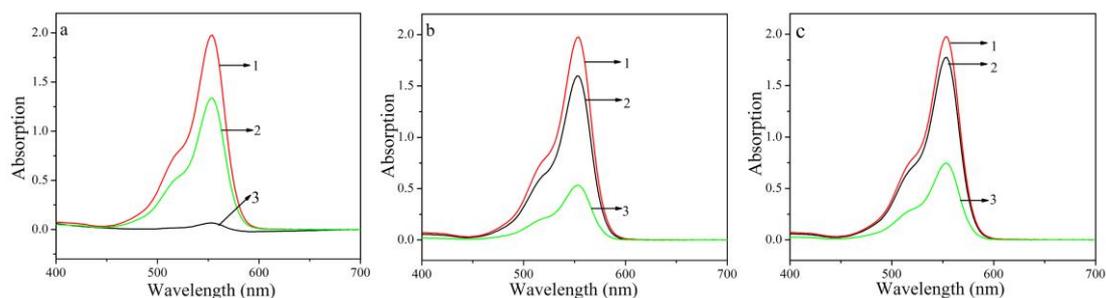
**Fig. S3** SEM images of the  $\text{Cu}_2\text{O}$  crystal in the presence of CTAB (0.1 g)



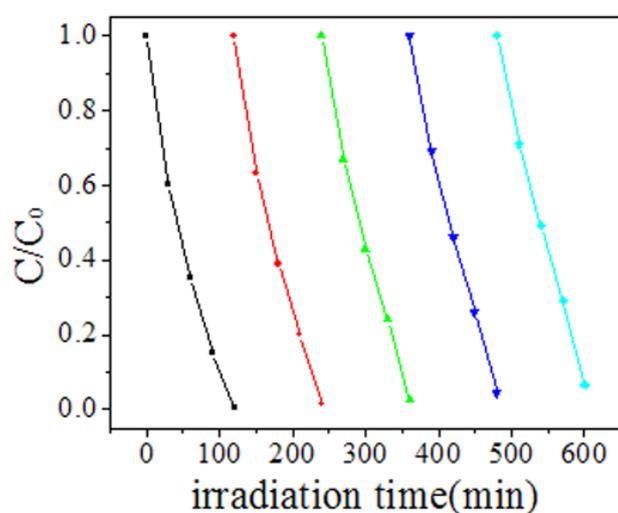
**Fig. S4** The color change of the solutions obtained at different reaction time during the formation process of hierarchical bows-like  $\text{Cu}_2\text{O}$ : (a) 0 min, (b) 90 min, (c) the supernatant after centrifugation of the solution obtained at 90 min, (d) 200 min, (e) the supernatant after centrifugation of the solution obtained at 200 min.



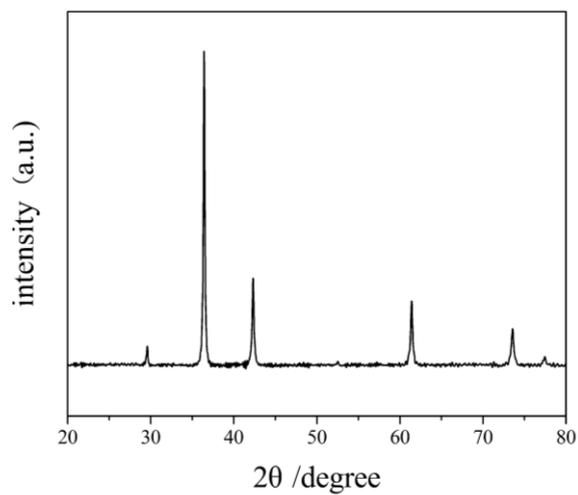
**Fig. S5** SEM image of  $\text{Cu}_2\text{O}$  crystal formed under the condition of nitrogen.



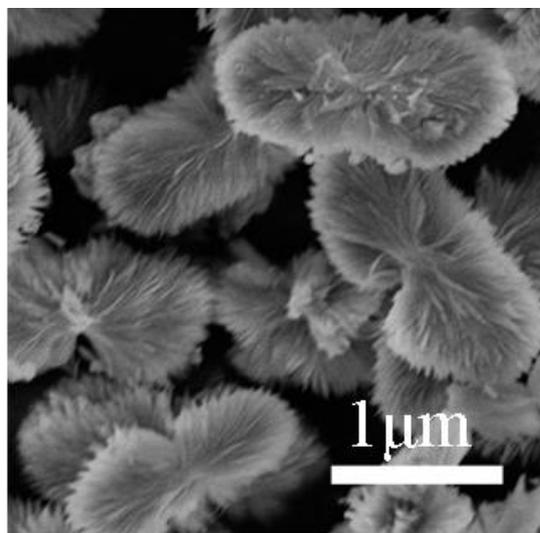
**Fig. S6** The UV-vis absorption spectra of the RhB solutions after dark adsorption and visible-light photocatalytic degradation in the presence of the different catalysts. a: bows-like  $\text{Cu}_2\text{O}$ , b: nitrogen-doped anatase  $\text{TiO}_2$ , c: nitrogen-doped Degussa P25. (1: Initial RhB solution, 2: After dark adsorption (30 min), 3: After 120 min photocatalytic test).



**Fig. S7** Cycling runs in the photodegradation of RhB in the presence of hierarchical bows-like  $\text{Cu}_2\text{O}$  under visible light irradiation.



**Fig. S8** XRD patterns of the  $\text{Cu}_2\text{O}$  crystal sample prepared from the solid after photocatalytic test.



**Fig. S9** SEM image of the  $\text{Cu}_2\text{O}$  crystal sample prepared from the solid after photocatalytic test.

**Table S1** The copper ions concentrations of the solutions obtained after different reaction time determined by the flame atomic absorption spectrometry (AAS, Thermo Elemental SOLAAR-M, limit of identification: 5 m g/L).

Reaction time (min)	0	90	200	420
Copper ions concentration (mol/L)	0.017	0.003	0.007	0.015

**Table S2** The RhB concentrations after dark adsorption and visible-light photocatalytic degradation in the presence of the different catalysts.

RhB Concentration (mg/L) Sample name	Initial concentration of RhB	After dark adsorption (30 min)	After photocatalytic reaction (120 min)
bows-like Cu <sub>2</sub> O	10	6.8	0.05
nitrogen-doped anatase TiO <sub>2</sub>	10	8.75	2.75
nitrogen-doped Degussa P25	10	8	3.70