

Supplementary Information for “Experimental and theoretical study of the degradation of malonamide extractant molecules under ionizing radiation”

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Figure 1. Infrared spectrum in the 1500-1750 cm⁻¹ wavenumber range of 10⁻¹ mol.dm⁻³ *N*-methyl-*N*,*N*'-dioctylhexylethoxymalonamide MDOHEMA in octane solution.

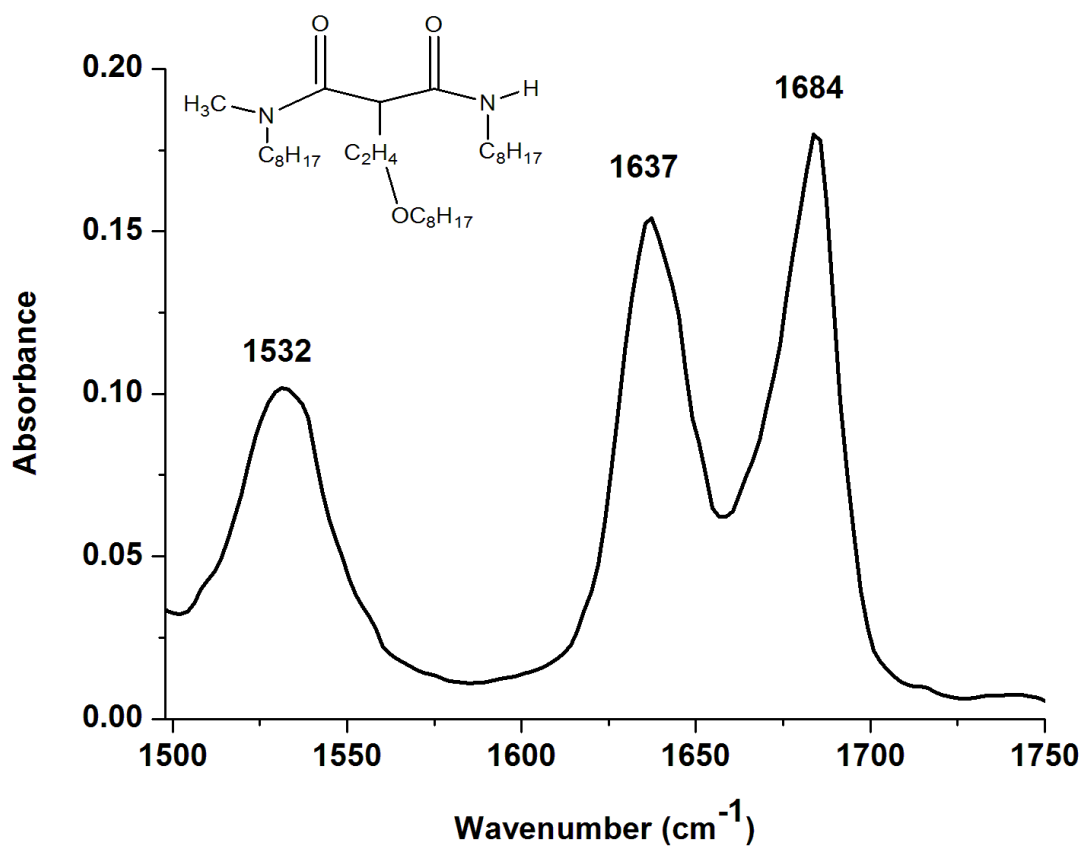


Figure 2. Infrared spectrum in the 1500-1800 cm^{-1} wavenumber range of $10^{-1} \text{ mol.dm}^{-3}$ *N*-methyl-*N*-butylhexadecylamide in octane solution.

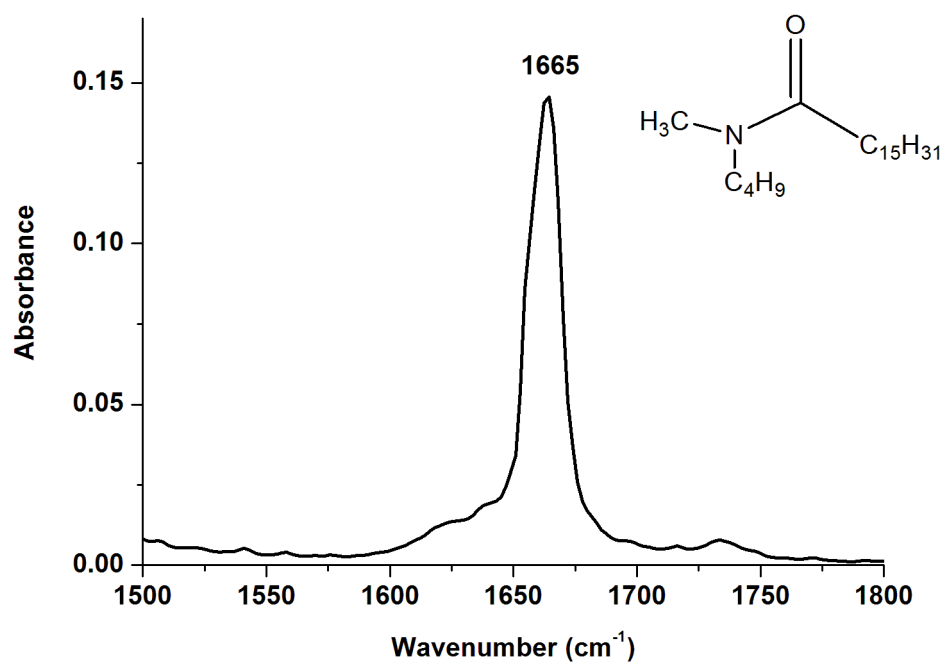


Figure 3. Infrared spectrum in the 1500-1800 cm^{-1} wavenumber range of $10^{-1} \text{ mol.dm}^{-3}$ *N,N*-dibutylformamide in octane solution.

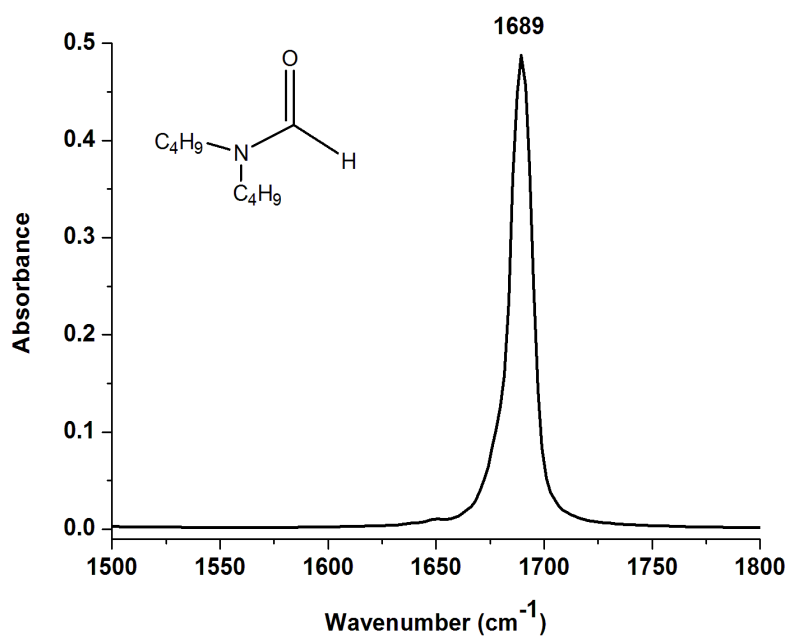


Figure 4. Infrared spectrum in the 1500-1800 cm^{-1} wavenumber range of $10^{-1} \text{ mol.dm}^{-3}$ *N,N'*-dimethyl-*N,N'*-dibutyltetradecylmalonamide (DMDBTDMA) in octane solution.

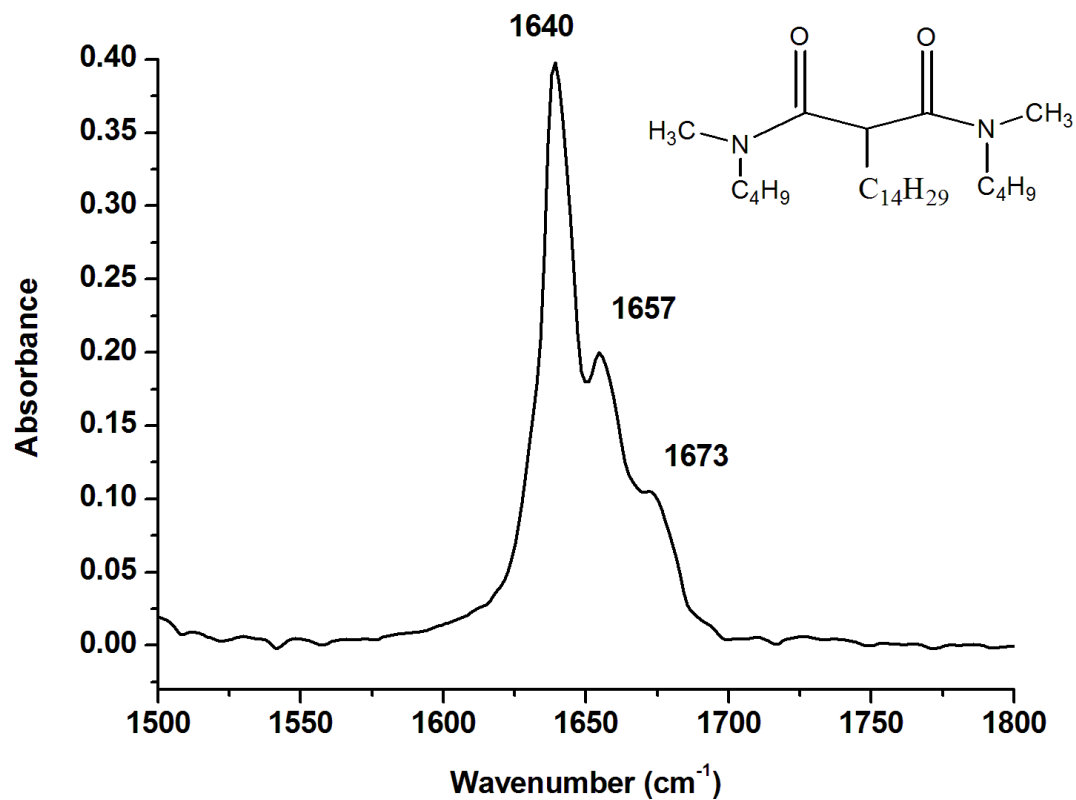


Figure 5. Infrared spectrum in the 1500-1800 cm^{-1} wavenumber range of $10^{-1} \text{ mol.dm}^{-3}$ *N,N'*-dimethyl-*N,N'*-dioctylhexylethoxymalonamide (DMDOHEMA) in octane solution.

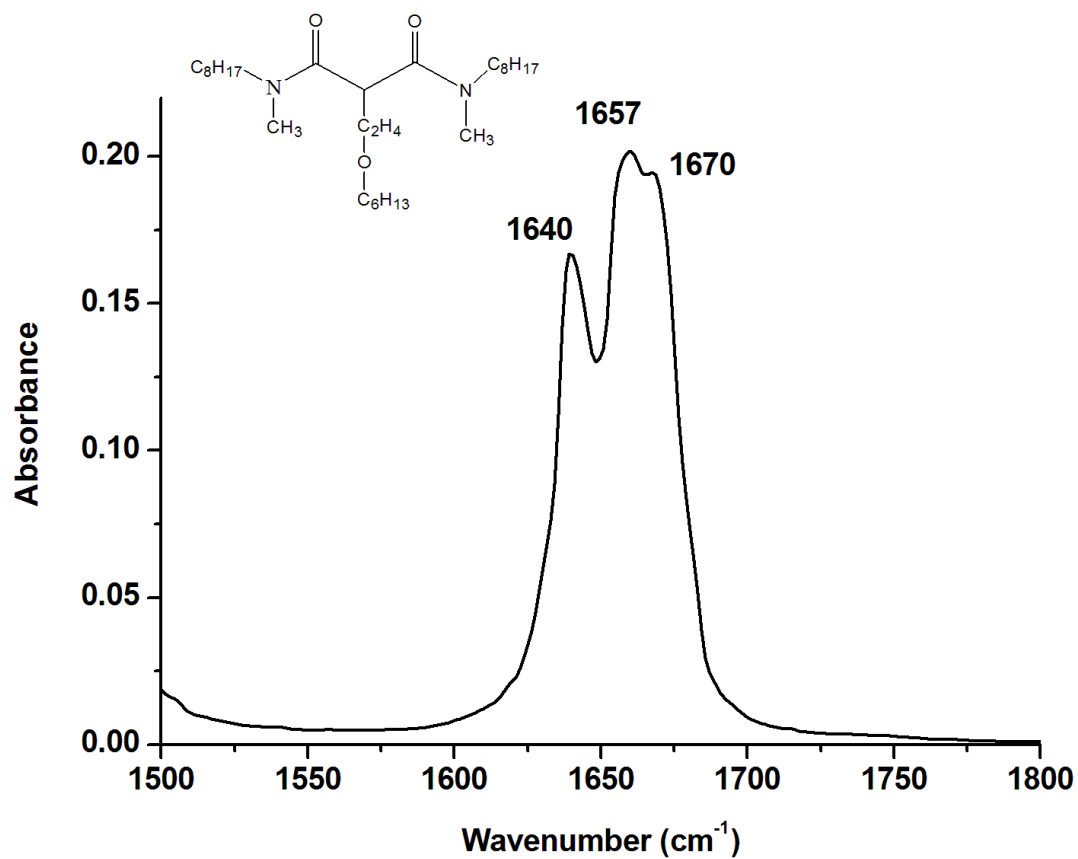


Figure 6. ESI-MS² spectrum of $m/z = 327.4$ of a 10^{-1} mol.dm⁻³ TEMA solution in *n*-octane after radiolysis. The inset is the formula proposed from the spectrum.

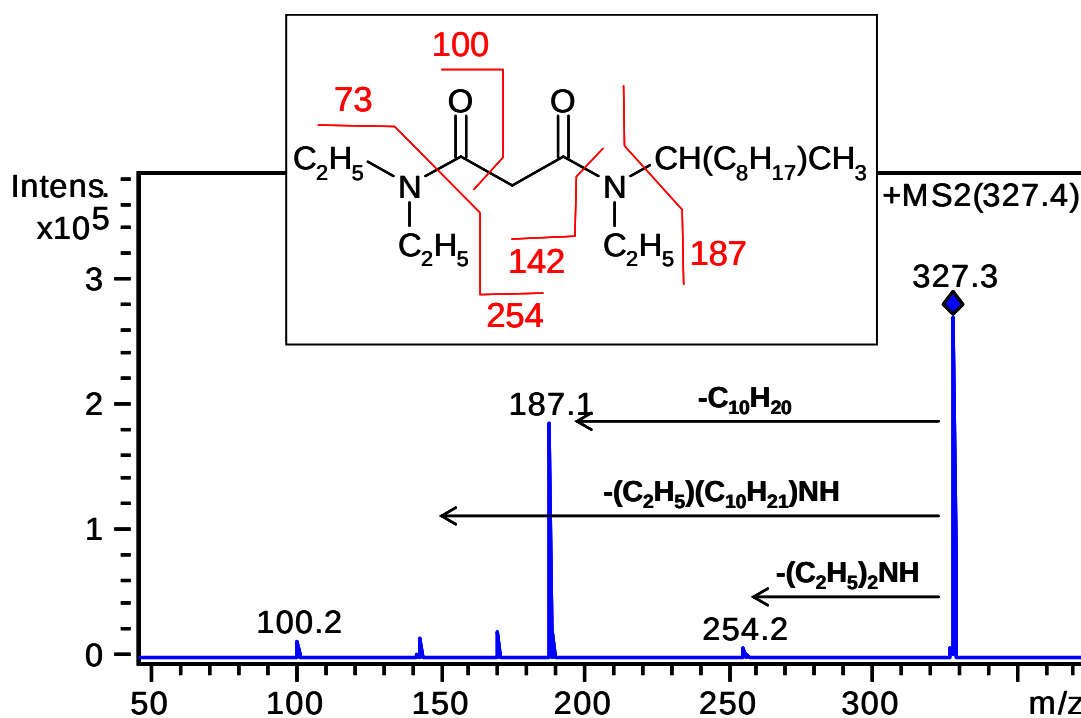


Figure 7. Evolution of the differential absorbance (after/before) irradiation of a $10^{-1} \text{ mol.dm}^{-3}$ *N,N,N',N'*-tetrabutylmalonamide ($((\text{C}_4\text{H}_9)_2\text{NCO})_2\text{CH}_2$) solution in the $1560\text{--}1760 \text{ cm}^{-1}$ spectral region after a 20 kGy irradiation. The spectrum is recorded just after irradiation. The bands formed after irradiation are at 1673 and 1698 cm^{-1} .

