

# ESI–MS Mechanistic Studies of the Wacker Oxidation of Alkenes: Dinuclear Species as Catalytic Active Intermediates

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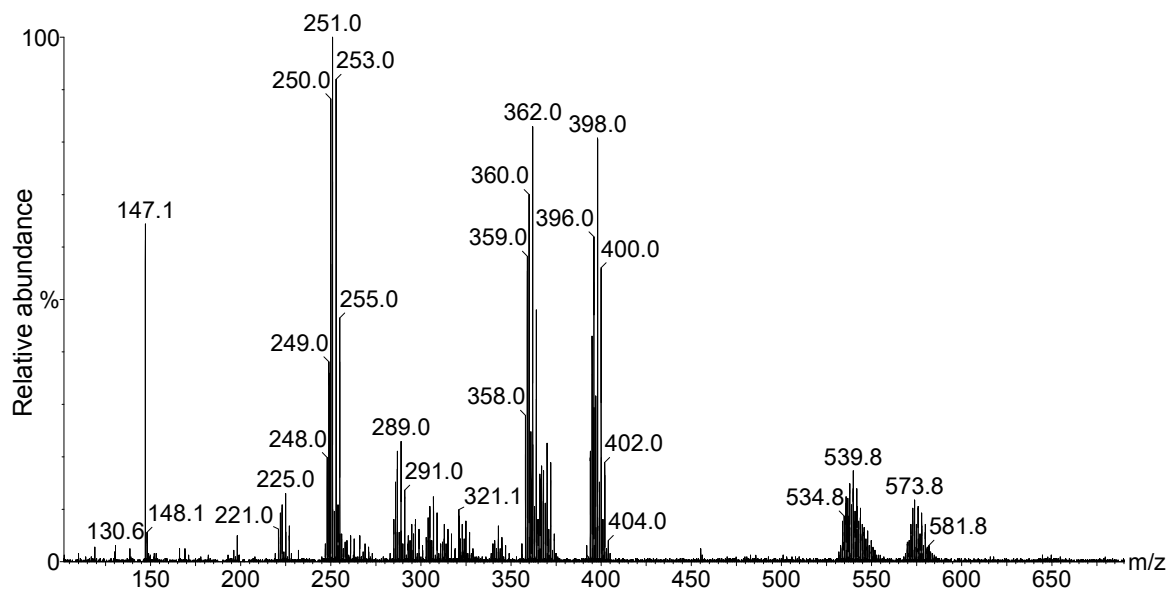
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## Supporting Information

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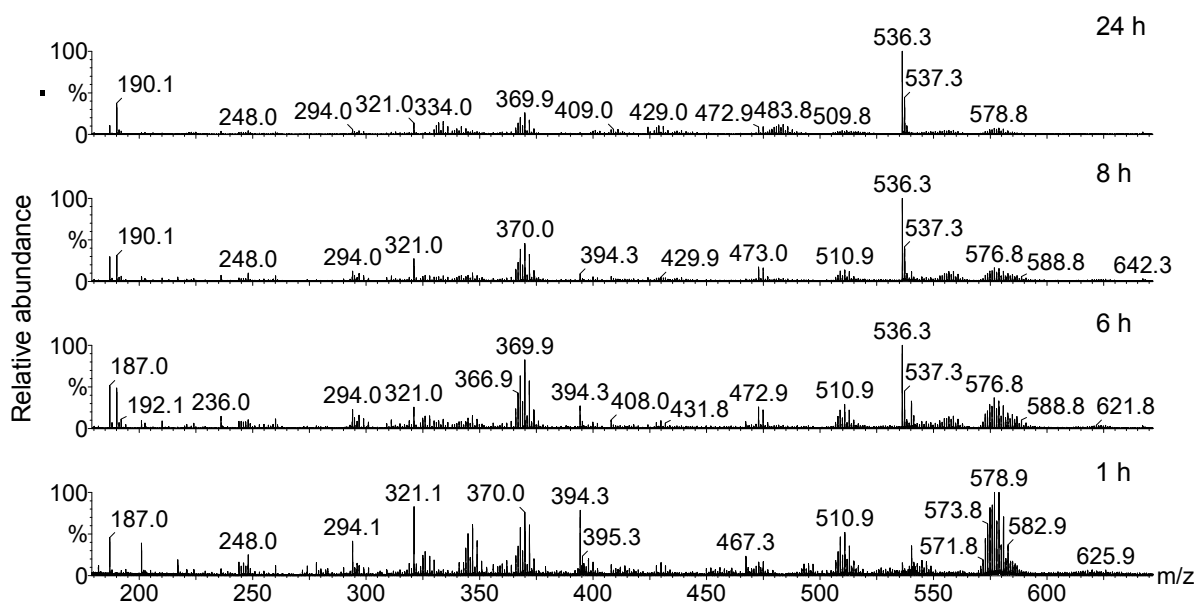
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**Figure S1.** ESI(+)-MS of a mixture of PdCl<sub>2</sub>, BQ, DMF, and H<sub>2</sub>O.<sup>a</sup>



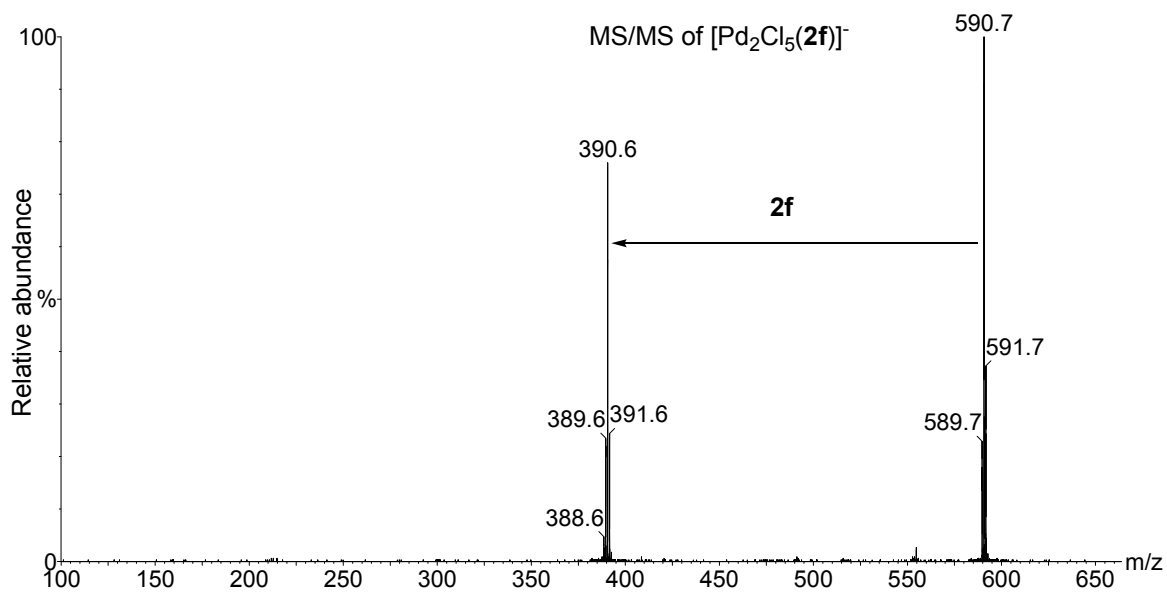
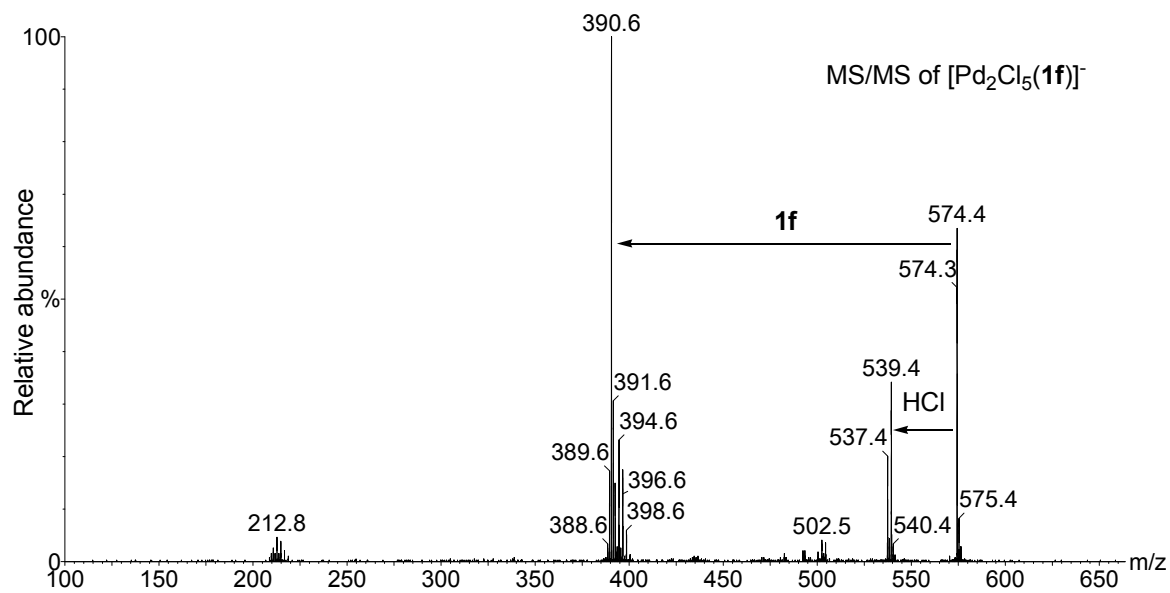
<sup>a</sup> PdCl<sub>2</sub> (0.05 mmol), BQ (1.1 mmol), DMF (1.75 mL), H<sub>2</sub>O (0.25 mL).

**Figure S2.** ESI(+)-MS of the reaction solution of the oxydation of **1c** over 24h.<sup>a</sup>

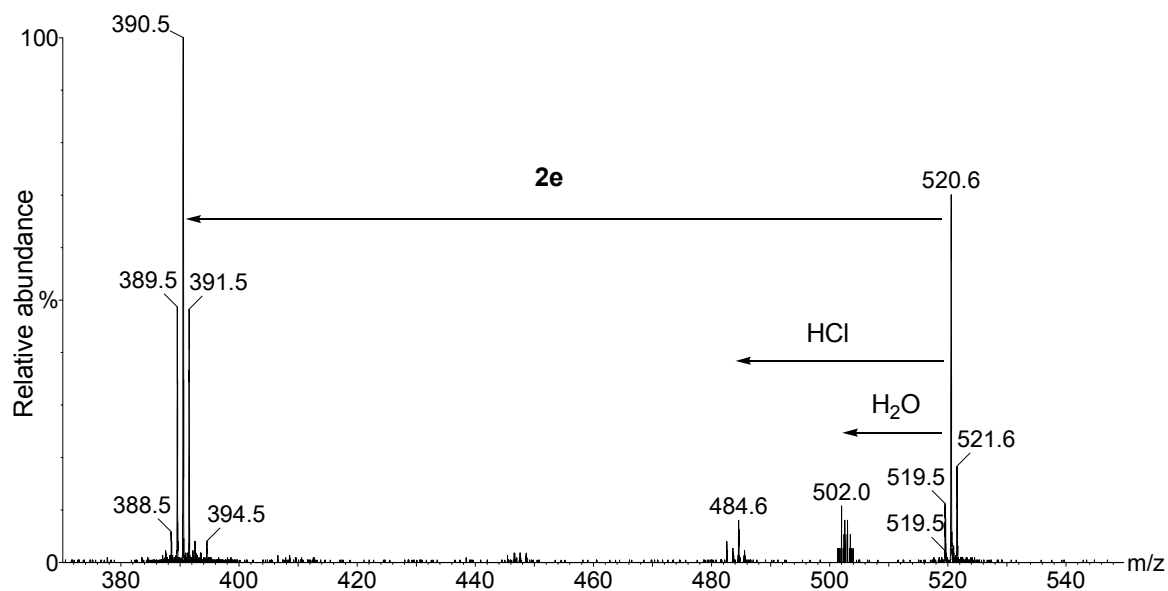
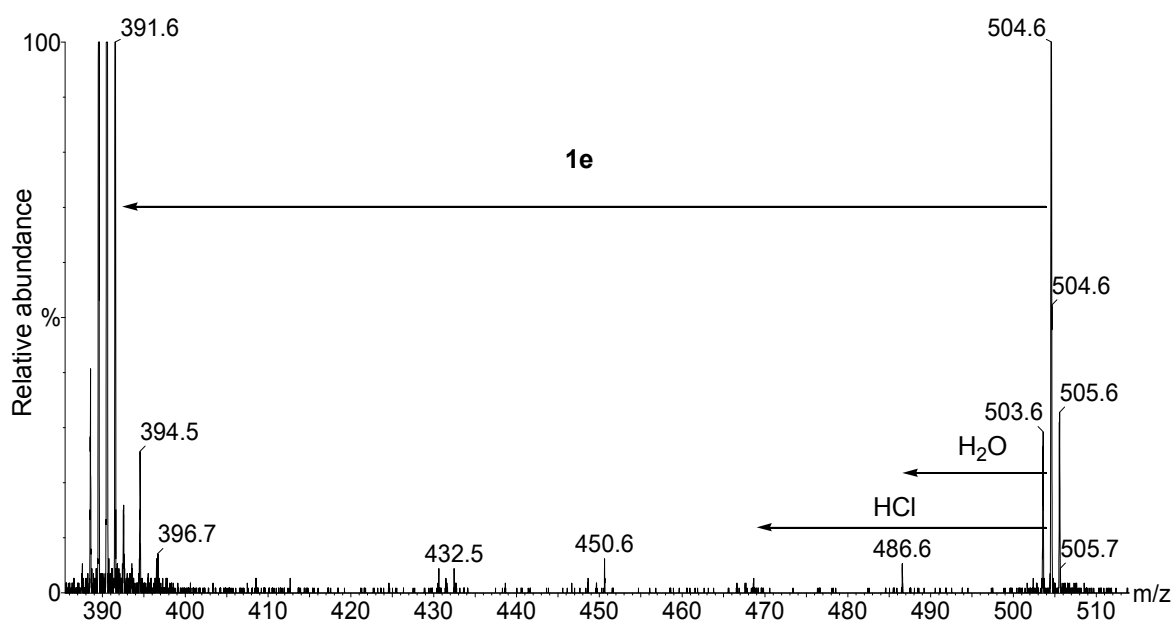


<sup>a</sup> **1c** (1.0 mmol), PdCl<sub>2</sub> (0.05 mmol), BQ (1.1 mmol), DMF (1.75 mL), H<sub>2</sub>O (0.25 mL).

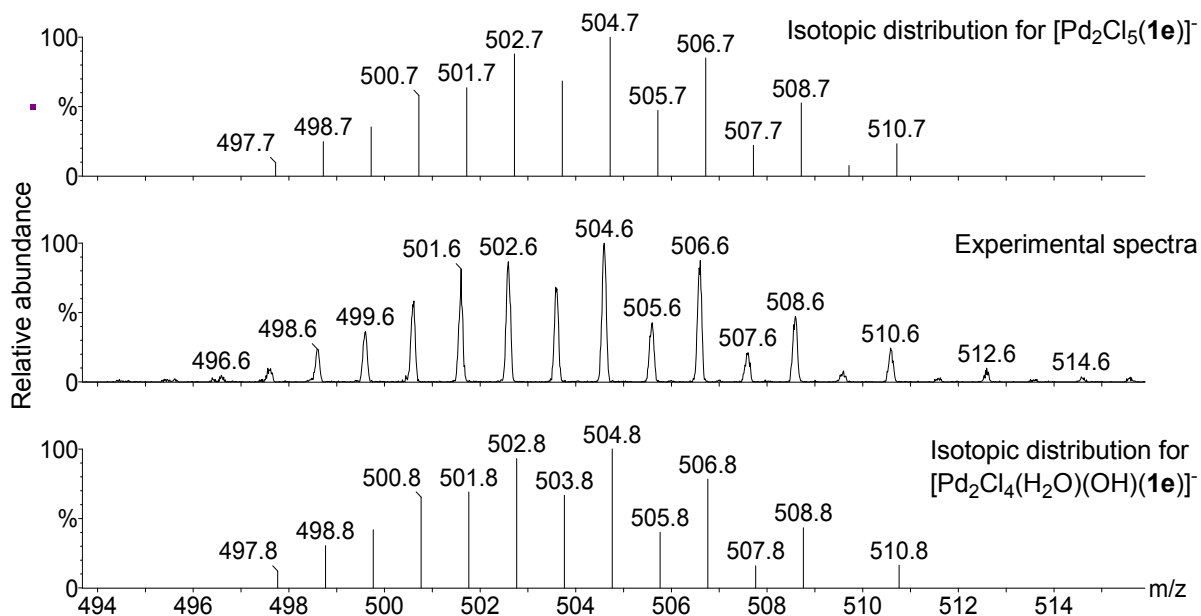
**Figure S3.** ESI(-)-MS/MS (6-8 eV) of  $[\text{Pd}_2\text{Cl}_5(\mathbf{1f})]^-$  and  $[\text{Pd}_2\text{Cl}_5(\mathbf{2f})]^-$ .



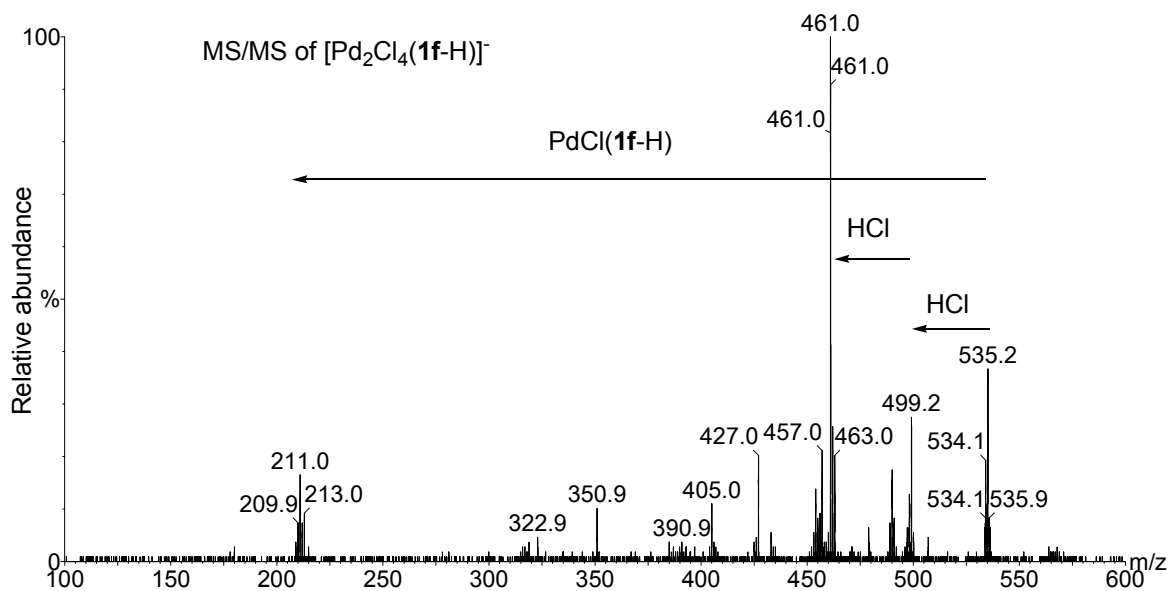
**Figure S4.** ESI(-)-MS/MS (6-8 eV) of  $[\text{Pd}_2\text{Cl}_5(\mathbf{1e})]^-$  and  $[\text{Pd}_2\text{Cl}_5(\mathbf{2e})]^-$  showing the loss of  $\text{H}_2\text{O}$  and  $\text{HCl}$ .

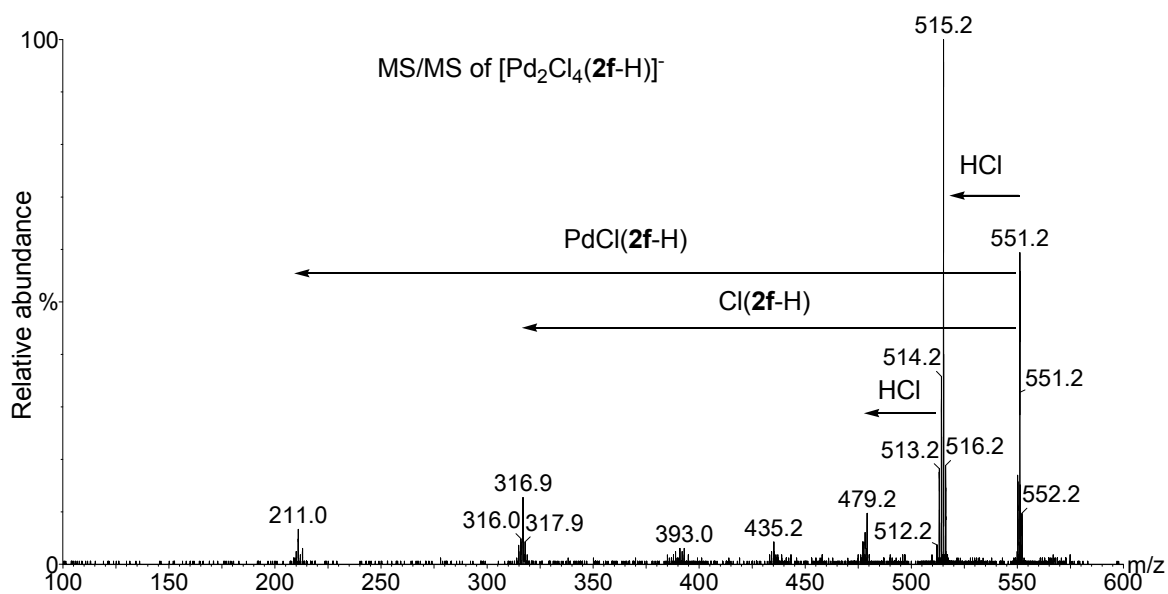


**Figure S5.** Isotopic distribution of  $[\text{Pd}_2\text{Cl}_5(\mathbf{1e})]^-$ ,  $[\text{Pd}_2\text{Cl}_4(\text{H}_2\text{O})(\text{OH})(\mathbf{1e})]^-$  and experimental spectra.

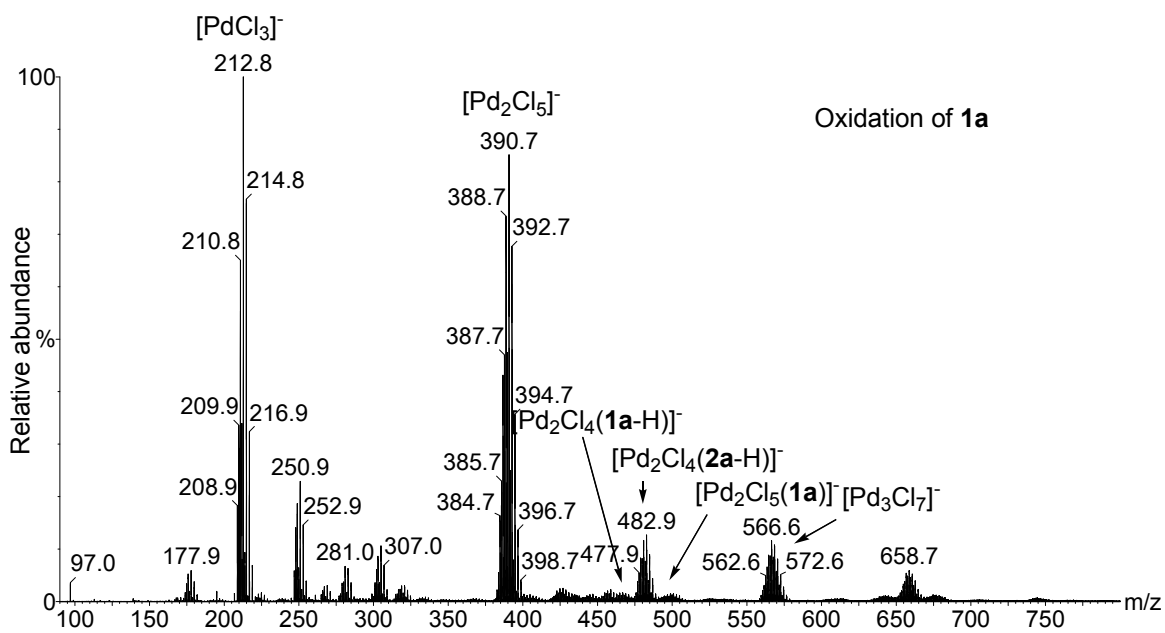


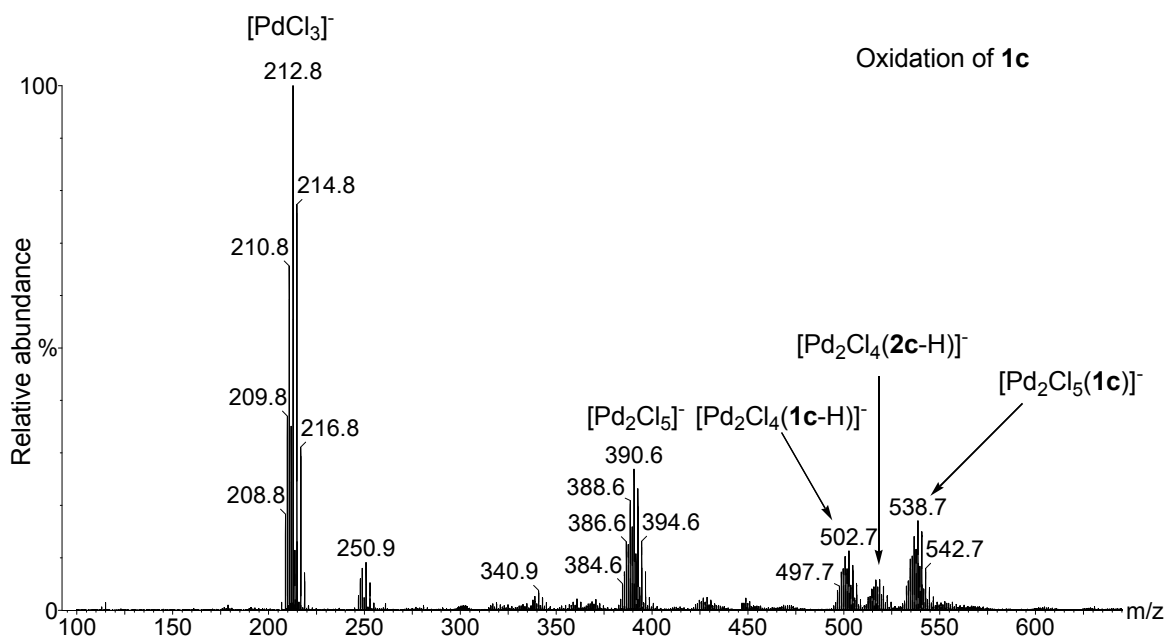
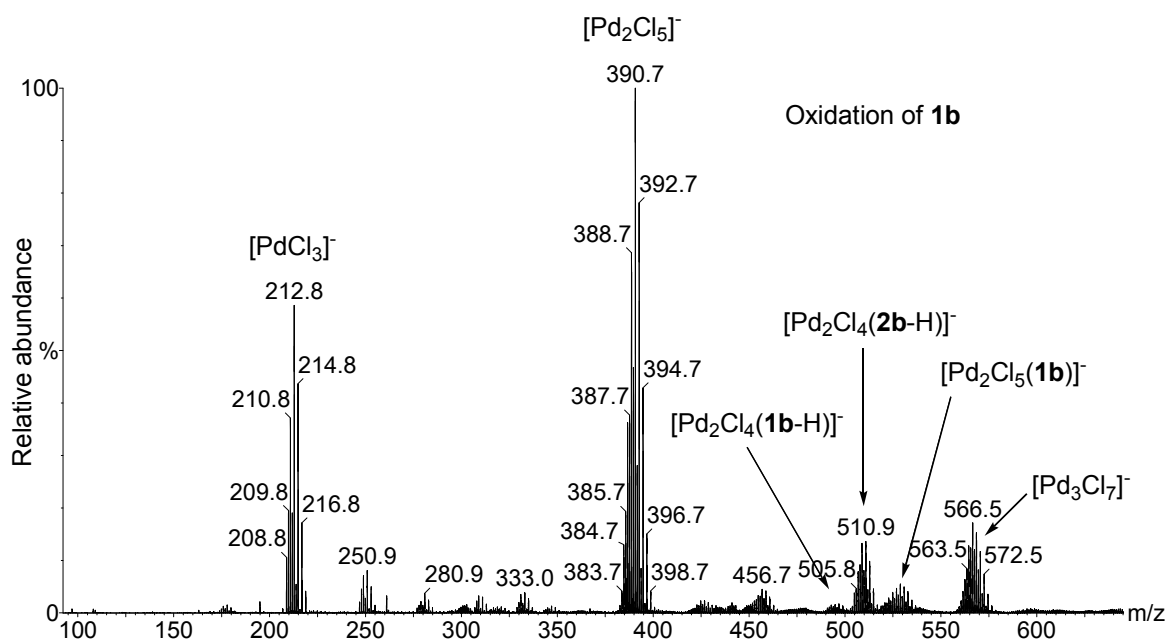
**Figure S6.** ESI(-)-MS/MS (15-20 eV) of  $[\text{Pd}_2\text{Cl}_4(\mathbf{1f-H})]^-$  and  $[\text{Pd}_2\text{Cl}_4(\mathbf{2f-H})]^-$ .

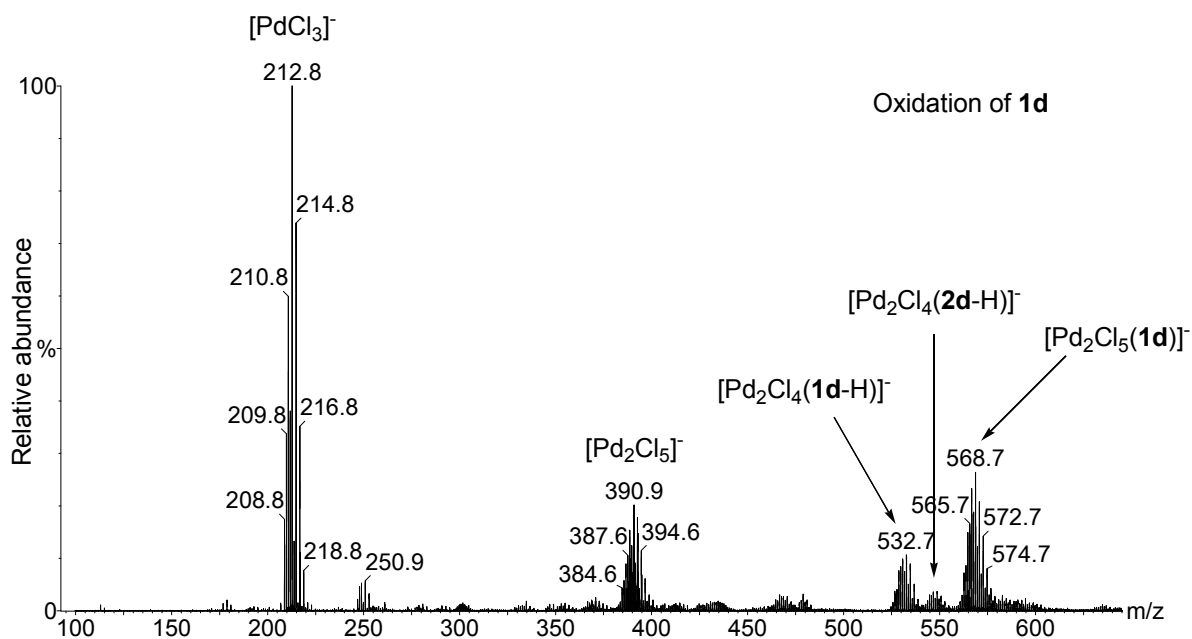




**Figure S7.** ESI(-)-MS of the reaction solution of the oxidation of **1a-1d** after 10 min.







**Figure S8.** ESI(-)-MS/MS (6-8 eV) of  $[\text{Pd}_2\text{Cl}_5(\mathbf{1b})]^-$ ,  $[\text{Pd}_2\text{Cl}_5(\mathbf{1c})]^-$ , and  $[\text{Pd}_2\text{Cl}_5(\mathbf{1d})]^-$ .

