

**Electronic Supplementary Information for RSC Advances**

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## **Supporting Information For**

# **Highly selective colorimetric and fluroescent sensors for fluoride anion based on imidazo[4,5-f]-1,10-phenanthroline metal-complexes**

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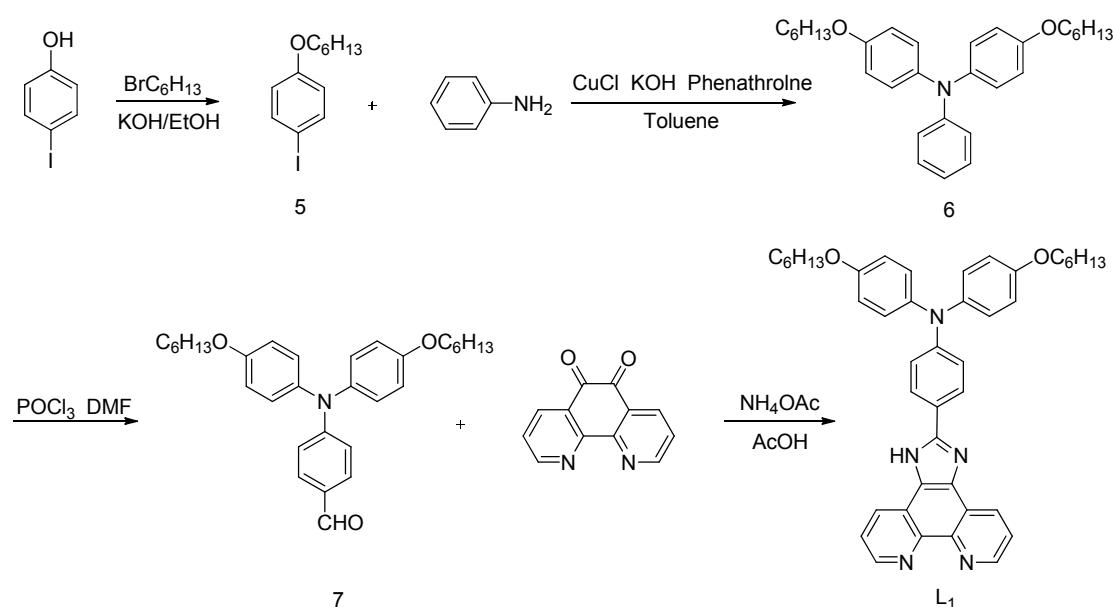
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## Content

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## 1. Synthetic procedure and characterization of ligands L<sub>1</sub> and L<sub>2</sub>



Scheme S1. Synthetic route of ligand L<sub>1</sub>.

**Hexyloxy-4-iodobenzene (5)** To a stirred mixture of 4-iodophenol (26.40g, 0.12 mol), hexylbromide (16.5 g, 0.1 mol), and dimethylformamide (100 mL) under argon was added K<sub>2</sub>CO<sub>3</sub> (27.60 g, 0.20 mol). The reaction mixture was refluxed for 15h. After cooled to room temperature, the resulting mixture was filtered and then extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was washed with a dilute solution of KOH and water and dried with MgSO<sub>4</sub>. After filtration, removal of the solvents, purification with column chromatography, the compound was obtained. Yield, 86%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ = 0.88 (t, J<sub>1</sub> = 7.2 Hz, J<sub>2</sub> = 6.4 Hz 3H, CH<sub>3</sub>), 1.31-1.34 (m, 4H, CH<sub>2</sub>), 1.42-1.45 (m, 2H, CH<sub>2</sub>), 1.72-1.78 (m, 2H, CH<sub>2</sub>), 3.89 (t, J<sub>1</sub> = 6.8 Hz, J<sub>2</sub> = 6.4 Hz, 2H, OCH<sub>2</sub>), 6.66 (d, J = 8.8 Hz, 2H, Ar-H), 7.52 (d, J = 8.8 H, 2H, Ar-H). MS(m/z): 304.03 [M]<sup>+</sup>. Anal. Calcd for C<sub>12</sub>H<sub>17</sub>IO: C, 47.38; H, 5.63. Found: C, 47.42; H, 3.21.

**Bis(4-(2'-hexyloxy)phenyl)amino)benzene (6)** The mixture of 5 (9.73 g, 32 mmol), aniline (1.4 g, 15 mmol), CuCl (0.12 g, 1.2 mmol), phenanthroline (0.26 g, 1.44 mmol), KOH (14.7 g, 0.26 mol), and toluene (35 mL) was refluxed for 24 h and cooled to room temperature. The resulting mixture was poured into plenty of water and then extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic phase was washed several times with water, dried with MgSO<sub>4</sub>. Then filtered, evaporated, and purified with

column chromatography. Yield, 73%.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 0.86 (t,  $J_1$  = 7.2 Hz,  $J_2$  = 7.6 Hz, 6H,  $\text{CH}_3$ ), 1.32-1.35 (m, 8H,  $\text{CH}_2$ ), 1.44-1.47 (m, 4H,  $\text{CH}_2$ ), 1.75-1.78 (m, 4H,  $\text{CH}_2$ ), 3.90 (t,  $J_1$  = 6.8 Hz,  $J_2$  = 6.4 Hz, 4H,  $\text{OCH}_2$ ), 6.79 (d,  $J$  = 9.2 Hz, 4H, Ar-H), 6.92 (d,  $J$  = 8 Hz, 1H, Ar-H), 7.01 (d,  $J$  = 9.2 Hz, 4H, Ar-H), 7.13 (d,  $J$  = 8 Hz, 2H, Ar-H). MS(m/z): 445.30 [M] $^+$ . Anal. Calcd for  $\text{C}_{30}\text{H}_{39}\text{NO}_2$ : C, 80.86; H, 8.82; N, 2.96. Found: C, 80.59; H, 8.76; N, 2.88.

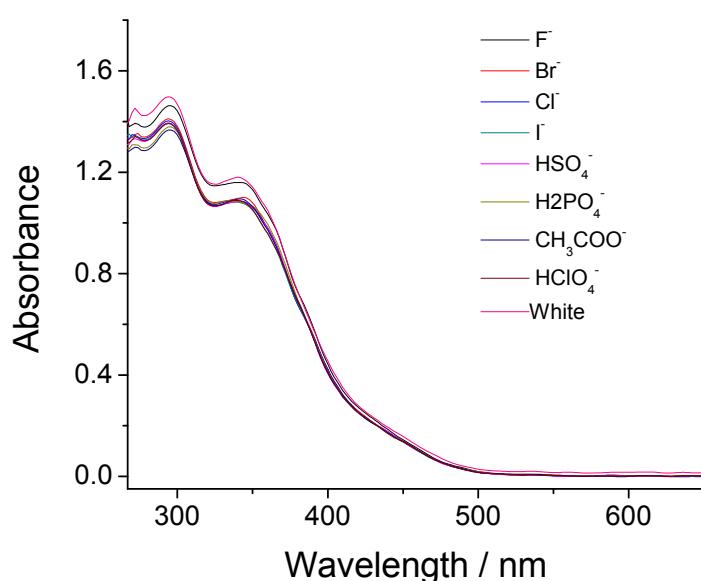
**4-(Bis(4’-(2”-hexyloxy)phenyl)amino)benzaldehyde (7)** Phosphorus oxychloride (2 mL, 21.6 mmol) was added dropwise to DMF (10 mL, 130 mmol) at 0 °C. The mixture was stirred for 1 h at 0 °C and additionally stirred at room temperature for 1h. After the addition of **6** (1.43 g, 3.2 mmol) in 1,2-dichloroethane (10 mL), the mixture was stirred at 90 °C for 2 h. After cooling to room temperature, the solution was poured into plenty of water, the resulting mixture was neutralized to pH7 with ammonia and then extracted with  $\text{CH}_2\text{Cl}_2$ . Then the extract was washed successively with water and brine. The organic extracts were dried with  $\text{MgSO}_4$ , then evaporated and purified with column chromatography. Yield, 66%.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 0.90 (t,  $J_1$  = 4.8 Hz,  $J_2$  = 4.4 Hz, 6H,  $\text{CH}_3$ ), 1.34-1.35 (m, 8H,  $\text{CH}_2$ ), 1.44-1.47 (m, 4H,  $\text{CH}_2$ ), 1.76-1.81 (m, 4H,  $\text{CH}_2$ ), 3.93 (t,  $J_1$  = 4.4 Hz,  $J_2$  = 4.4 Hz, 4H,  $\text{OCH}_2$ ), 6.83 (d,  $J$  = 6 Hz, 2H, Ar-H), 6.90 (d,  $J$  = 6 Hz, 4H, Ar-H), 7.10 (d,  $J$  = 6 Hz, 4H, Ar-H), 7.61 (d,  $J$  = 6 Hz, 2H, Ar-H), 9.75 (s, 1H, CHO). MS(m/z): 473.29 [M] $^+$ . Anal. Calcd for  $\text{C}_{31}\text{H}_{39}\text{NO}_3$ : C, 78.61; H, 8.30; N, 2.96. Found: C, 78.68; H, 8.44; N, 2.74.

**Ligand (**L**<sub>1</sub>)** A mixture of **7** (0.71 g, 1.5 mmol), 1,10-phenanthroline-5,6-dione (0.315 g, 1.5 mmol), ammonium acetate (2.31 g, 30 mmol) and glacial acetic (30 mL) was refluxed for 2 h, and then cooled to room temperature and diluted with water. The resulting mixture was neutralized to pH=7 with ammonia to get a yellow precipitate, which was collected and washed with water. The crude product in ethanol was purified with column chromatography. Yield, 94%.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 0.85 (t,  $J_1$  = 7.2 Hz,  $J_2$  = 6.4 Hz, 6H,  $\text{CH}_3$ ), 1.27-1.31 (m, 8H,  $\text{CH}_2$ ), 1.36-1.41 (m, 4H,  $\text{CH}_2$ ), 1.67-1.73 (m, 4H,  $\text{CH}_2$ ), 3.81 (t,  $J_1$  = 7.6 Hz,  $J_2$  = 6 Hz, 4H,  $\text{OCH}_2$ ), 6.71 (d,  $J$  = 8.8 Hz, 6H, Ar-H), 6.91 (d,  $J$  = 8.4 Hz, 4H, Ar-H), 7.28 (m, 2H, pyridine-H), 8.03 (d,  $J$  = 8.8 Hz, 2H, Ar-H), 8.745 (m, 4H, pyridine-H). MS(m/z): 663.36 [M] $^+$ . Anal. Calcd for  $\text{C}_{43}\text{H}_{45}\text{N}_5\text{O}_2$ : C, 77.80; H, 6.83; N, 10.56. Found: C, 77.87; H, 6.64; N, 10.34.

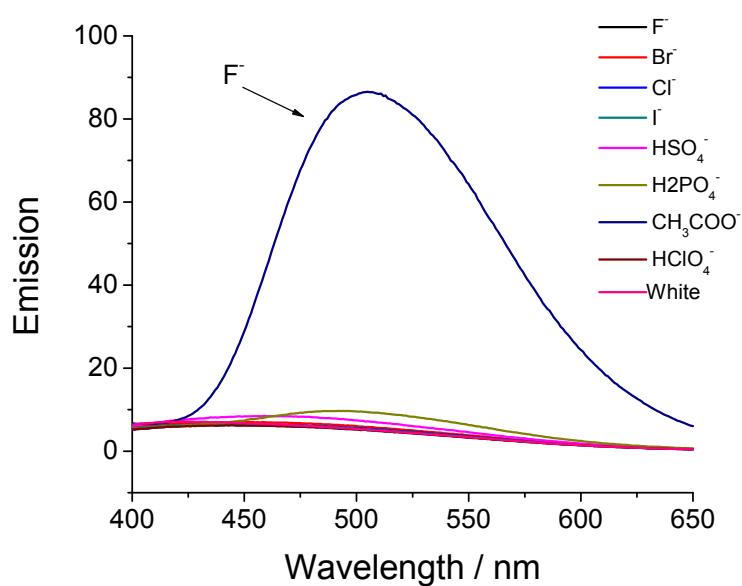
**Ligand (**L**<sub>2</sub>)** A mixture of **L**<sub>1</sub> (0.66 g, 1 mmol), indomethane (0.28 g, 2 mmol) and  $\text{K}_2\text{CO}_3$  (0.28 g, 2 mmol) in anhydrous DMF (15 mL) was stirred over night. The solution was poured into plenty

of water, the solid precipitated was filtered and was washed with water and dried to give a yellow solid. The crude product in ethanol was purified with column chromatography. Yield, 88%.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 0.90 (t,  $J_1$  = 6.4 Hz,  $J_2$  = 6.4 Hz, 6H,  $\text{CH}_3$ ), 1.25 (m, 8H,  $\text{CH}_2$ ), 1.35-1.36 (m, 4H,  $\text{CH}_2$ ), 1.76-1.83 (m, 4H,  $\text{CH}_2$ ), 3.94 (t,  $J_1$  = 6.4 Hz,  $J_2$  = 6.4 Hz, 4H,  $\text{OCH}_2$ ), 4.29 (s, 3H,  $\text{CH}_3$ ), 6.87 (d,  $J$  = 8.8 Hz, 4H, Ar-H), 7.05 (d,  $J$  = 8.8 Hz, 2H, Ar-H), 7.12 (d,  $J$  = 8.8 Hz, 4H, Ar-H), 7.56 (d,  $J$  = 8.8 Hz, 2H, Ar-H), 7.66-7.72 (m, 2H, pyridine-H) 8.75 (d,  $J$  = 8.4 Hz 1H, Ar-H), 9.05 (d,  $J$  = 8.4 Hz 1H, Ar-H), 9.03 (d,  $J$  = 8.4 Hz 2H, pyridine-H). MS(m/z): 677.37 [M] $^+$ . Anal. Calcd for  $\text{C}_{44}\text{H}_{47}\text{N}_5\text{O}_2$ : C, 77.96; H, 6.99; N, 10.33. Found: C, 77.79; H, 6.85; N, 10.21.

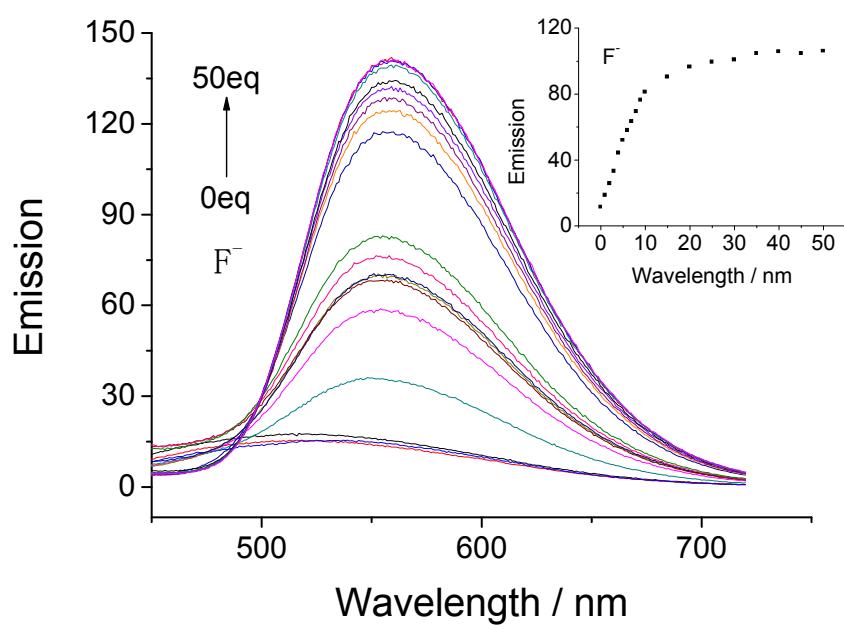
## 2. Supplementary spectra data (Fig. S1 – S6)



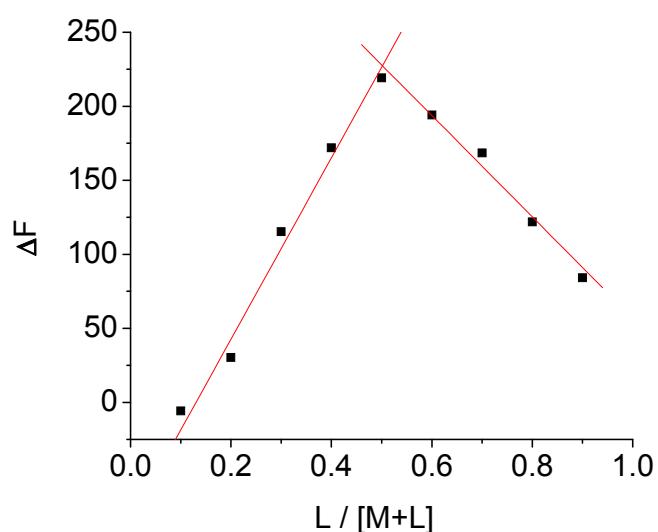
**Fig. S1** Changes in the absorption spectra of complex **2** ( $2 \times 10^{-5}$  mol/L in DMSO) upon addition of 10 equiv of several anions ( $2 \times 10^{-4}$  mol/L in DMSO).



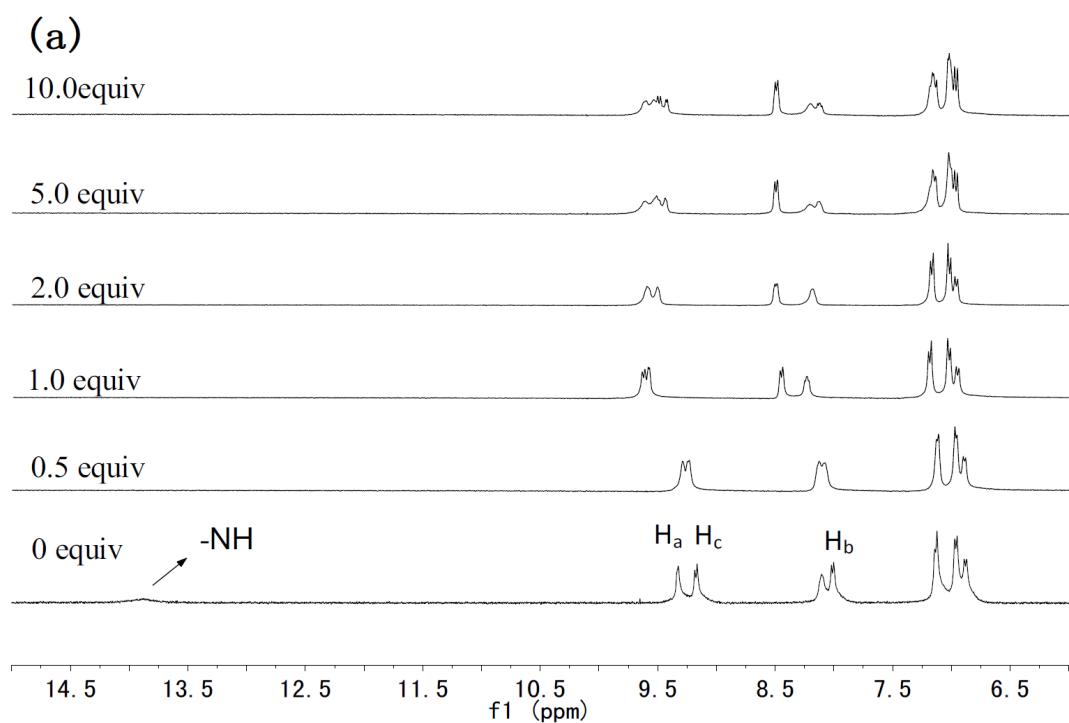
**Fig. S2** Fluorescence spectra of complex **4** ( $2 \times 10^{-5}$  mol/L in DMSO) upon addition of 10 equiv of several anions ( $2 \times 10^{-4}$  mol/L in DMSO) excited at 364nm.

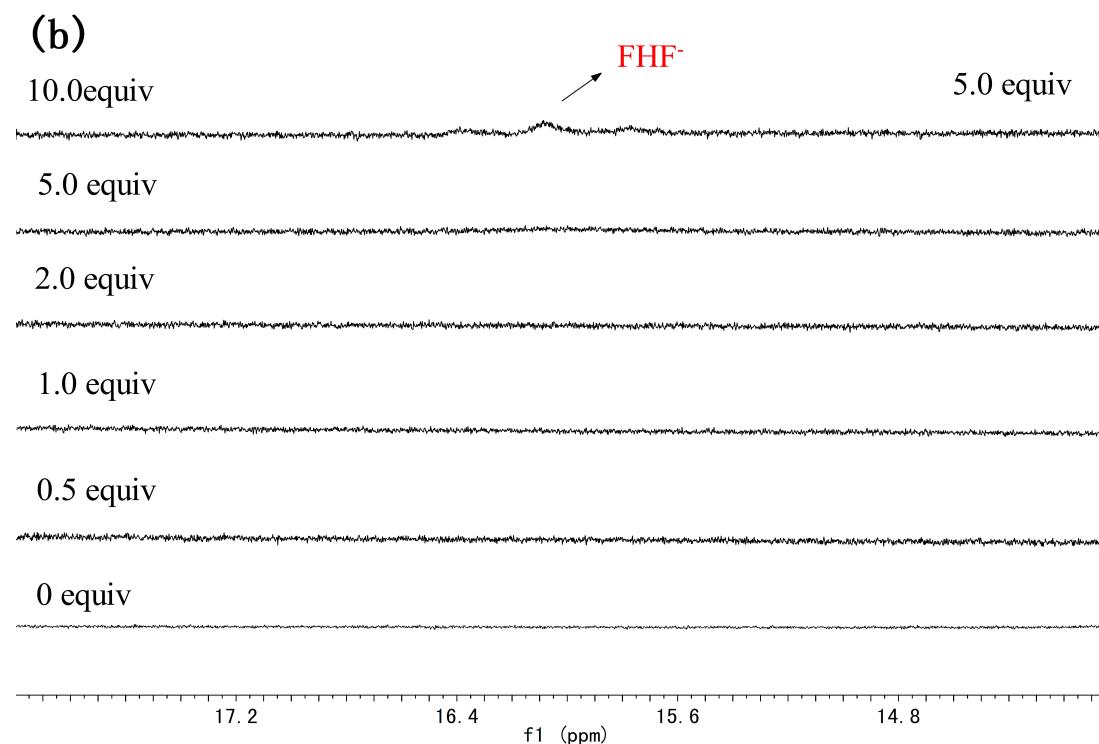


**Fig. S3** Changes in the fluorescence emission of complex **4** ( $2 \times 10^{-5}$  mol/L in DMSO) upon addition of F<sup>-</sup> (equiv = 0, 1, 2, 3, 4, 5, 10, 15, 20, 25, 30, 35, 40, 45, 50.) excited at 364nm

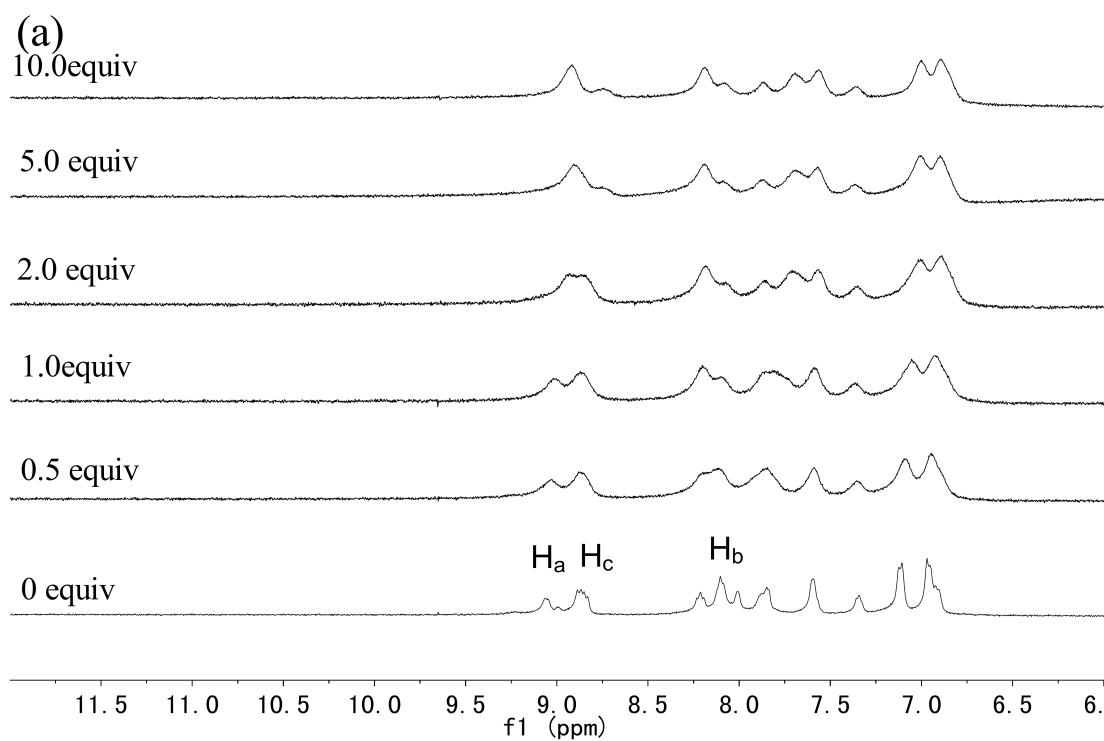


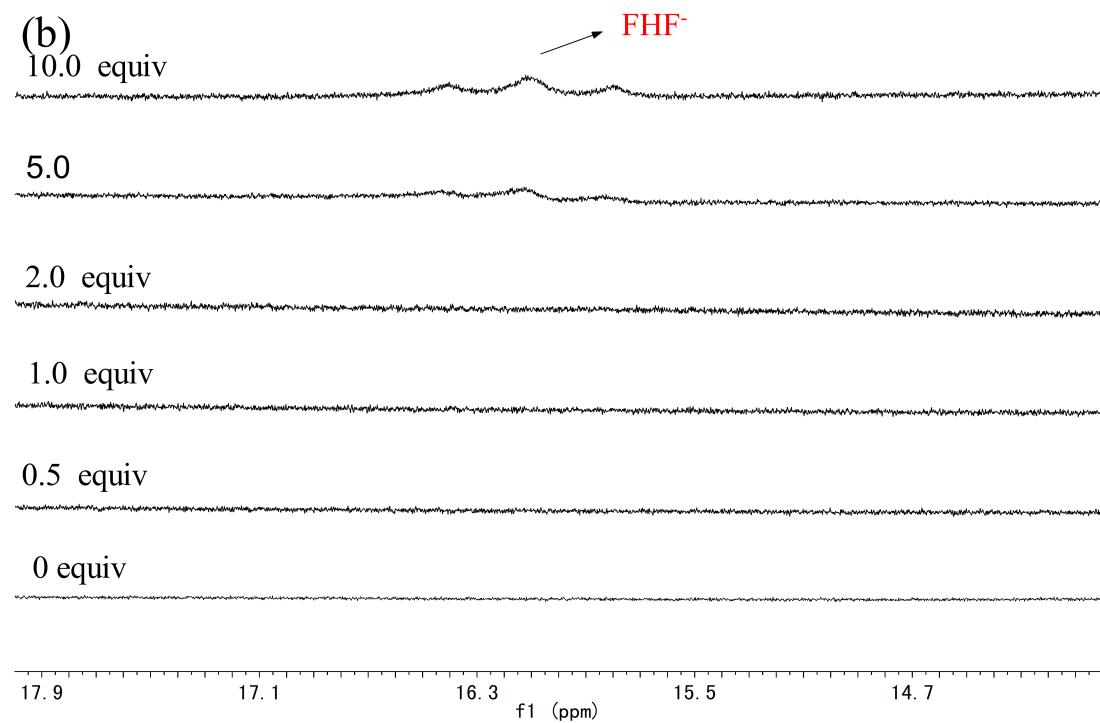
**Fig. S4** Job's plots for the reactions of complex **4** with  $F^-$  in DMSO



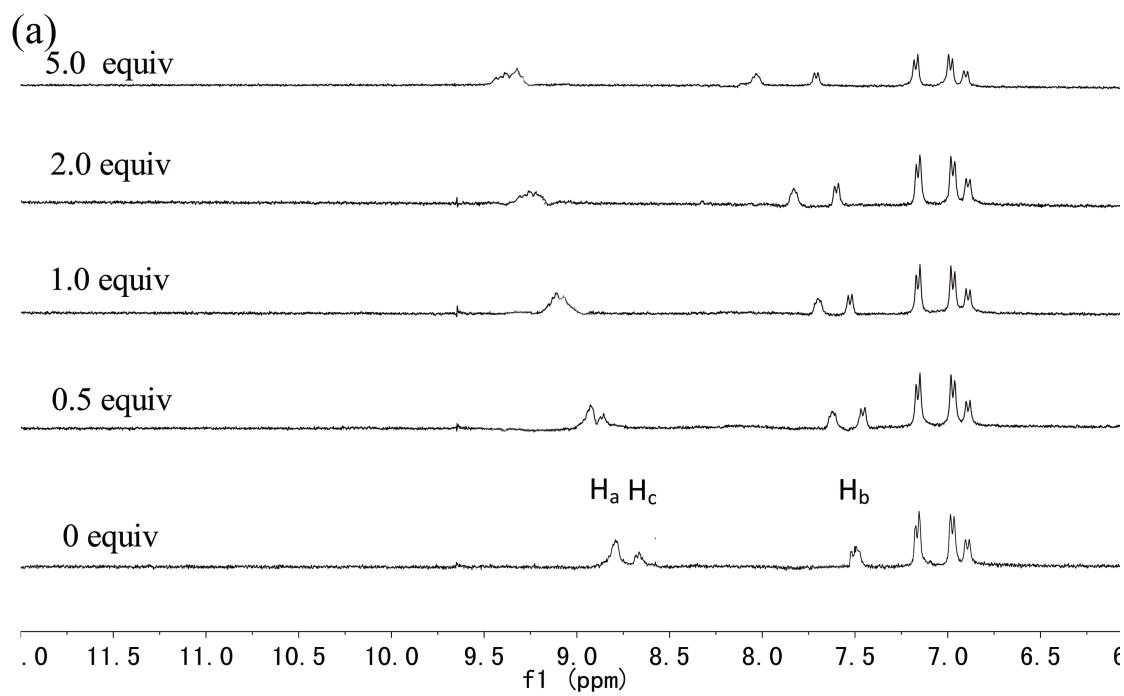


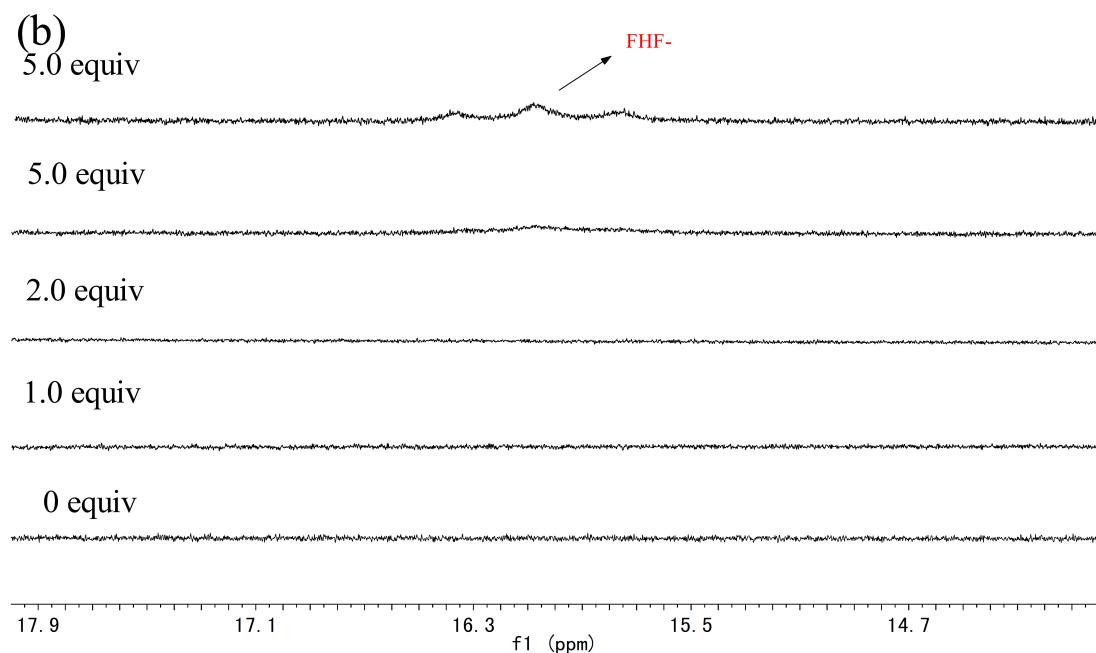
**Fig. S5** Partial  $^1\text{H}$  NMR spectra (400 MHz in  $d^6$ -DMSO at 298 K) of **1** upon addition of 0-10 equiv of TBAF.





**Fig. S6** Partial <sup>1</sup>H NMR spectra (400 MHz in <sup>6</sup>-DMSO at 298 K) of **3** upon addition of 0-10 equiv of TBAF.





**Fig. S7** Partial  $^1\text{H}$  NMR spectra (400 MHz in  $d^6$ -DMSO at 298 K) of **4** upon addition of 0-10 equiv of TBAF.