# A cathodic electrogenerated chemiluminescence biosensor based on 

# luminol and hemin-graphene nanosheets for cholesterol detection 

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Materials: Graphene oxide was purchased from Nanjing xianfeng nano Co. (Nanjing, China). Hemin, cholesterol oxidase (ChOx, EC 1.1.3.6, $\geq 50$ units/mg, from Brevibacterium sp.), cholesterol $\left(\mathrm{C}_{27} \mathrm{H}_{46} \mathrm{O}, \mathrm{M}_{\mathrm{r}}: 386.67, \geq 99 \%\right.$ purity, from lanolin $)$, and Triton X-100 $\left(\mathrm{C}_{34} \mathrm{H}_{62} \mathrm{O}_{11}\right.$, MW: 646.85) were obtained from Sigma Chemical Co. (St. Louis, MO, USA). Phosphate buffer solutions (PBS, containing $0.9 \% \mathrm{NaCl}$ ) with pH 7.4 were prepared with $0.05 \mathrm{M} \mathrm{KH}_{2} \mathrm{PO}_{4}$ and $0.05 \mathrm{M} \mathrm{Na}_{2} \mathrm{HPO}_{4}$. The stock solution was prepared by dissolving cholesterol in the mixture of 2-propanol and Triton X-100, and then diluted it only with Triton X-100 solution for preparing standard solutions. Other chemicals used were of analytical grade and were used as received. Double distilled water was used throughout this study.

Apparatus and measurements: Cyclic voltammetry (CV) was performed with a CHI 600D electrochemical work station (Shanghai Chenhua Instruments Co., China). The ECL emission was monitored with a model MPI-A electrochemiluminescence analyzer (Xi'an Remax Electronic Science \& Technology Co. Ltd., China) with the
voltage of the photomultiplier tube (PMT) set at 600 V in the process of detection. All experiments were performed with a conventional three-electrode system. The modified glassy carbon electrode (GCE) as working electrode, a platinum wire as counter electrode and a saturated calomel electrode (SCE) or $\mathrm{Ag} / \mathrm{AgCl}($ sat. KCl ) as reference electrode. The UV-Vis absorption spectra were recorded in the range of 200-800 nm, using a UV-Vis spectrometer (UV-Vis 8500). All the electrochemical experiments were carried out at room temperature.

Preparation of hemin-graphene nanosheets: According to Guo's work ${ }^{1}$, hemin-graphene nanosheets (H-GNs) were synthesized with a simple wet-chemical strategy through the $\pi-\pi$ interactions. First, 20.0 mL of the homogeneous graphene oxide dispersion ( $0.5 \mathrm{mg} / \mathrm{mL}$ ) was mixed with 20.0 mL of $0.5 \mathrm{mg} / \mathrm{mL}$ hemin aqueous solution and $200.0 \mu \mathrm{~L}$ of ammonia solution, followed by the addition of $30 \mu \mathrm{~L}$ of hydrazine solution. After being vigorously stirred for a few minutes, the vial was put in a water bath $\left(60{ }^{\circ} \mathrm{C}\right)$ for 3.5 h . Finally, the product was obtained by filtration and washed several times. The obtained $\mathrm{H}-\mathrm{GNs}$ can be redispersed readily in water by ultrasonication. Additionally, the preparation of pure graphene was similar to $\mathrm{H}-\mathrm{GNs}$ except no addition of hemin.

Construction of the cholesterol biosensor: Glassy carbon electrode (GCE, $\Phi=4$ mm ) was polished with 0.3 and $0.05 \mu \mathrm{~m}$ alumina slurry, and then ultrasonically cleaned in ethanol and water thoroughly. After it was allowed to dry at room temperature, $10 \mu \mathrm{~L}$ H-GNs dispersed solution was dropped on the GCE. Subsequently, $5 \mu \mathrm{LChOx}(1 \mathrm{mg} / \mathrm{mL}$ in $0.1 \mathrm{M} \mathrm{PBS}, \mathrm{pH} 7.0)$ solutions were dropped on the surface of
the electrode to construct a cholesterol biosensor (noted as $\mathrm{ChOx} / \mathrm{H}-\mathrm{GNs} / \mathrm{GCE}$ ). For comparison, $\mathrm{ChOx} / \mathrm{GNs} / \mathrm{GCE}$ was prepared similarly. The modified electrodes were stored at $4^{\circ} \mathrm{C}$ for future use.


Fig. S1 Effect of luminol concentration on the ECL responses to cholesterol at ChOx/H-GNs/GCE in 0.05 M PBS (pH 7.4). Scan rate: $100 \mathrm{mV} / \mathrm{s}$.


Fig. S2 The ECL responses of luminol $(0.15 \mathrm{mM})$ at $\mathrm{ChOx} / \mathrm{H}-\mathrm{GNs} / \mathrm{GCE}$ to (A) 0.38 mM cholesterol, (B) 2 mM ascorbic acid, (C) 2 mM uric acid, (D) 2 mM dopamine, and (E) 2 mM glycine in 0.05 M PBS ( pH 7.4 ). Scan rate: $100 \mathrm{mV} / \mathrm{s}$.

Table S1 Comparison of performance of some cholesterol sensors

| Electrode materials | Determine method | Linear range <br> $(\mu \mathrm{M})$ | Detection <br> limit $(\mu \mathrm{M})$ | Refs. |
| :---: | :---: | :---: | :---: | :---: |
| ChOx/Chi-IL/MWNT(SH)-Au | Chronoampertry | $500-5000$ | - | 2 |
| ChOx-PPy/Pt | Chronoampertry | $25-300$ | 5.7 | 3 |
| Gold electrode polymerized | Molecularly <br> with 2-MBI | $5-30$ | 0.42 | 4 |
| ChOx/p(pyrrole)/p(HEMA) | Chronoampertry | $500-1500$ | 120 | 5 |
| ChOx/H-GNs/GCE | ECL | $0.17-1120$ | 0.06 | This work |

ChOx, cholesterol oxidase. Chi, chitosan. IL, ionic liquid. MWNTs, multiwall carbon nanotubes. PPy, polypyrrole. MBI, maslach burnout inventory. p(HEMA), poly(2-hydroxyethyl methacrylate).

Table S2. Application of the biosensor for determination the recovery of cholesterol.

| Sample | Detected $^{\mathrm{a}}(\mu \mathrm{M})$ | Added $(\mu \mathrm{M})$ | Found $^{\mathrm{a}}(\mu \mathrm{M})$ | Recovery $(\%)$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 10.0 | 10.0 | $20.5 \pm 0.8$ | 102.5 |
| 2 | 10.0 | 15.0 | $25.7 \pm 1.5$ | 102.8 |
| 3 | 15.0 | 25.0 | $44.8 \pm 1.8$ | 112.0 |
| 4 | 75.0 | 75.0 | $150.8 \pm 2.0$ | 100.5 |
| 5 | 150.0 | 200.0 | $345.1 \pm 0.9$ | 98.6 |
| 6 | 300.0 | 400.0 | $699.4 \pm 0.6$ | 99.9 |

All samples were analyzed using standard addition method ( $\mathrm{n}=3$ ).
${ }^{a}$ Mean value $\pm$ standard deviation ( $n=3$ ).

## References

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