## **Supporting Information**

## Nickel nanocrystals: fast synthesis of cubes, pyramids and tetrapods

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Figure S1: Differential Scanning Calorimetry under nitrogen with heating slope of 10°C/min





Figure S3: <sup>1</sup>H NMR of the supernatant after filtration of the Ni NPs





Figure S4: XRD of Ni NPs (sample 1d) casted on a glass slide (a); Transmission Electron Microscopy images of Ni nanoparticles synthesized in different ligands with a constant ratio precursor/ligand: 1/2 in a microwave at T =  $170 \degree C$  for 30 minutes b) benzylamine, c) triphenylphosphine



Figure S5: Transmission Electron Microscopy images of Ni nanoparticles synthesized in the microwave (see table 1 below)

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This journal is © Th	<sup>e</sup> Sampfe <sup>oc</sup>	>eppDfA/Nemistry(20)2		t (min)	Morphology	Size (nm)					
	2a	10	80-120°C	30	sphere	5 ± 3					
	2b	10	150-160 °C	45	rod, sphere	5*15; 7 ± 3					
	2c	10	170°C	2	triangle, cube and sphere	$5\pm 2$					
	2d	10	150-170 °C	240	Sphere	4 ± 3					



Figure S6: Low magnification TEM of pyramidal and cubic NPs (Microwave, Hexadecylamine 2 equivalent,  $T = 170^{\circ}$ C, 20 min reaction)

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Figure S7: TEM of pyramidal and cubic NPs (Microwave, Hexadecylamine 2 equivalent,  $T = 170^{\circ}C$ , 30 min (a) and after post-synthesis annealing ( $T = 150^{\circ}C$ , 60 min) (b)



Figure S8: Transmission Electron Microscopy images of Ni nanoparticles synthesized in a microfluidic cell (see table 2 below)

Sam-	Injection rate	t (min)	T (°C)	Morphology	Size (nm)
ple	$(\mu L.min^{-1})$				
4a	100	2.5	170	Cube, triangle, sphere	$7 \pm 2$
<b>4b</b>	25	10	170	Cube, triangle, sphere	8 ± 4
<b>4</b> c	25	10	150	Cube, triangle, sphere,	$10 \pm 4$
				tetrapod	
<b>4d</b>	8	30	170	Cube, triangle, sphere,	8 ± 4 (pod
				tetrapod	diameter)

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