

Supporting Information for:

**A Facile Synthesis Method for Ni(OH)₂ Ultrathin Nanosheets and Their
Conversion to Porous NiO Nanosheets Used for Formaldehyde Sensing**

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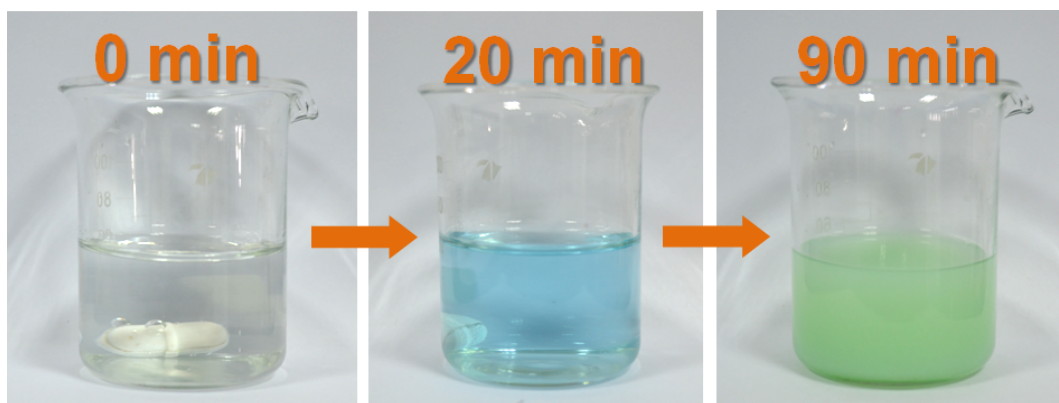


Fig. S1: Photograph of electrolyte (60 mmol NaCl, 10 mmol NH₄Cl, and 7.5 mmol NaOH in 100 ml deionized water) at different reaction time (0, 20, 90 min).

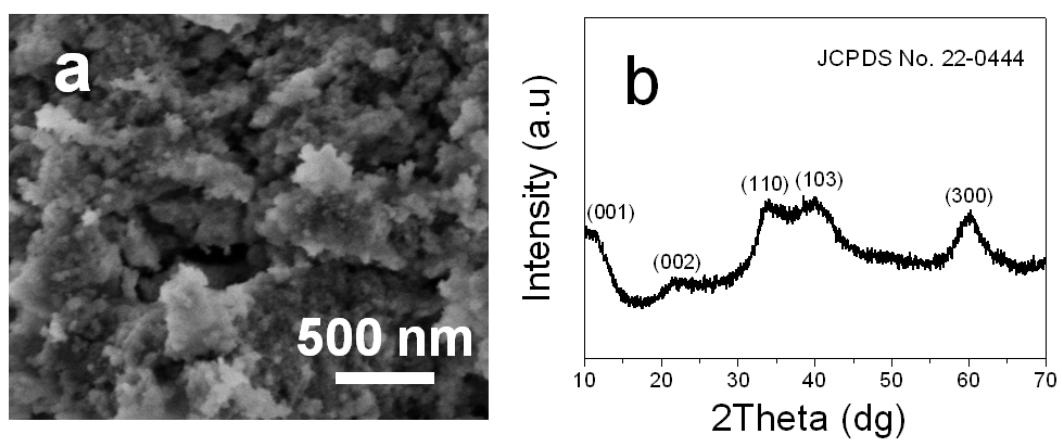


Fig. S2: (a) SEM image and (b) XRD pattern of irregular shaped α -Ni(OH)₂ obtained via electrochemical reaction with 1 M NaCl (without NaOH and NH₄Cl).

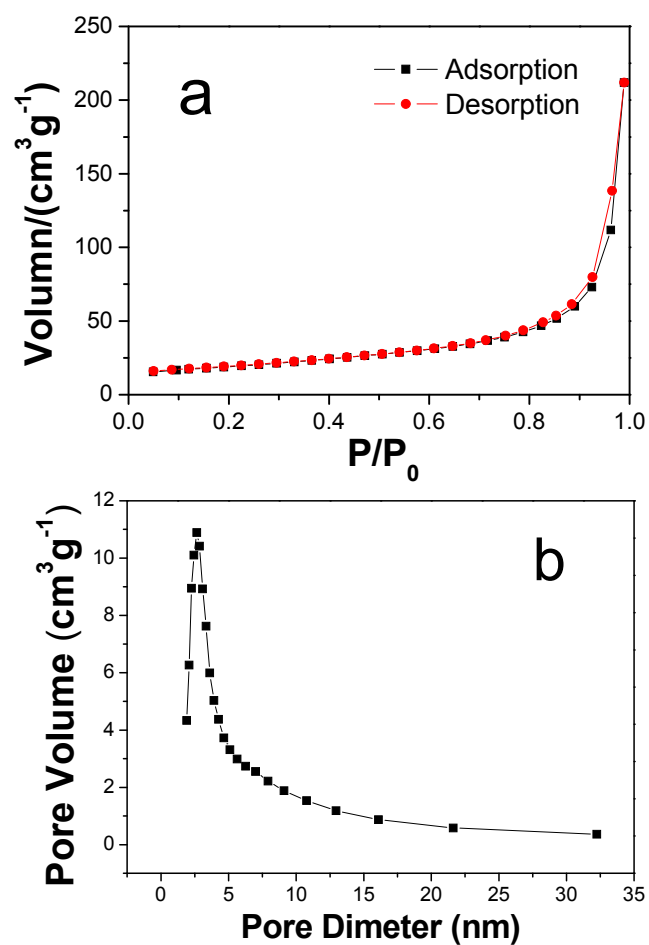


Fig. S3 (a) Typical N_2 adsorption-desorption isotherm, and (b) pore-size distribution of the NiO nanosheets.

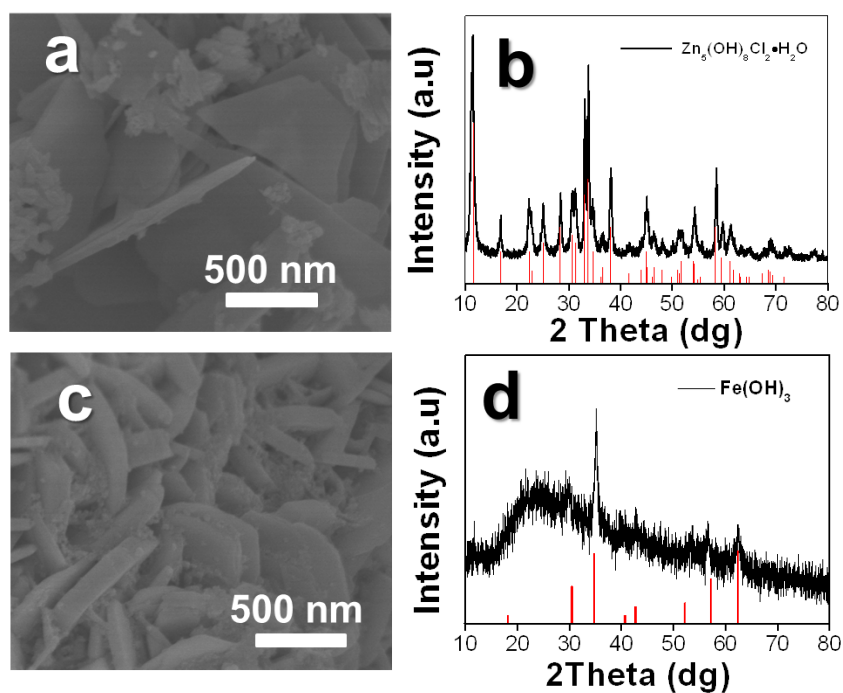


Fig. S4 (a, c) SEM images and (b, d) XRD patterns of obtained Zn₅(OH)₈Cl₂·H₂O and Fe(OH)₃ nanosheets, which were synthesized by electrochemical reactions of Zn and Fe electrodes with electrolyte composed of 0.2 M NaCl, 0.1 M NH₄Cl and 0.075 M NaOH under potentiostatic (5 V) control for 90 min.