

## Supporting Information

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### NMR and Crystallographic discussion :

Synthesized compounds were characterized by FTIR,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR and HRMS analysis. Substitution at 2<sup>nd</sup> position, irrespective of its electron withdrawing and electron donating nature, shifts the stereogenic protons Ha towards deshielding region (**1b** > **1** > **2b**  $\Rightarrow$  5.04 > 4.60 > 4.53  $\delta$  ppm) and proton Hb towards shielding region (**1b** < **1** = **2b**  $\Rightarrow$  3.96 < 4.04 = 4.04  $\delta$  ppm) when compared to **1** and respective 4-substituted compounds as shown in Figure S1 (SI). This was further confirmed by replacement of fluorine by bulkier trifluoromethyl group at 2<sup>nd</sup> position, which shifts proton Ha to more deshielding region (**1c** < **1e** = 5.01 < 5.12  $\delta$  ppm) and Hb towards shielding region (**1c** > **1e** = 3.86 > 3.72  $\delta$  ppm) (SI, Figure S2). The vinylic proton (Hc) shifts slightly towards deshielding region when electron withdrawing groups (-I effect) are present on benzylidene ring irrespective of their position whereas that of electron donating groups (+I effect) slightly shifts the proton towards shielding region (**1b** < **1** < **1c** = 6.78 < 6.82 < 6.84  $\delta$  ppm).

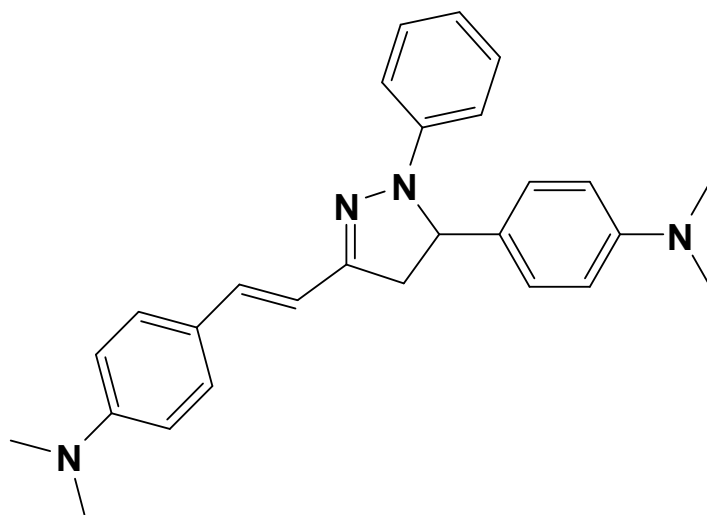
Single crystal structures of some of the representative compounds were determined to examine the conformations of these compounds. Earlier, we have reported structure of **1d**, (7E)-5-Benzyl-7-(2-chlorobenzylidene)-3-(2-chlorophenyl)-2-phenyl-3,3a,4,5,6,7-hexahydro-2H-pyrazolo[4,3-c]pyridine.<sup>1</sup> A search in Cambridge Structural Database (version 5.31) for 2H-pyrazolo[4,3-c]pyridines retrieved none except **1d**. The structures of **1b**, **1c**, **1f** with adopted atomic numbering scheme is shown in Figure 1 a-c. In **1b** and **1c**, asymmetric units comprise of two different molecules with minor conformational differences (RMSD of 0.794 Å and 0.972 Å between different molecules of asymmetric units, respectively for **1b** and **1c**). These compounds are racemic mixtures. Similar to **1d**, the stereogenic centers, C3 and C3A of the reported models of **1b** and **1f** possess (R, R)-configurations (SI, Table S2). Interestingly crystal structure of **1c** reveals coexistence of (S, R) and (R, S) configurations of the stereogenic centers, C3 and C3A. This observation suggests that both configurations are energetically accessible. The coexistence of two configurations within an asymmetric unit has been previously observed for Boc-Leu-Dpg-Val-OMe.<sup>2</sup> The five membered dihydropyrazole ring (N1/N2/C3/C3A/C7A) adopts an envelope conformation with atom C3 at the flap of the envelope and an adjacent 6-membered piperidine ring (C3A/C4/N5/C6/C7/C7A) assumes a chair conformation, but substantially distorted from ideal geometry. Short intra-molecular C—H...halogen and C-H...O contacts were observed in **1b**, **1c** and **1f** leading to modulation of their photophysical properties (vide infra). A dimer is formed in **1f** (similar to **1d**) by

C29—Cl3•••Cg5<sup>i</sup> [symmetry code (i): 1 - x, 1 - y, 1 - z, Cg5 is the centroid of (C21–C26) ring]. The Cl3•••Cg5 distance and C29—Cl3•••Cg5 angle are 3.7407(15) Å and 137.9(1)°, respectively, whereas the minimum atomic distance in Cl3••• Cg5 is 3.366 (4) Å. Cg5 is the centroid of (C21–C26) ring. The C—Halogen••• $\pi$  dimeric interactions [also referred as PHD;  $\pi$ -halogen-dimer interactions] have been shown recently,<sup>3</sup> to play an important role in host-guest chemistry.<sup>4</sup> The notable interactions in the crystal packing are C-H... $\pi$  interactions (SI, Table S3).

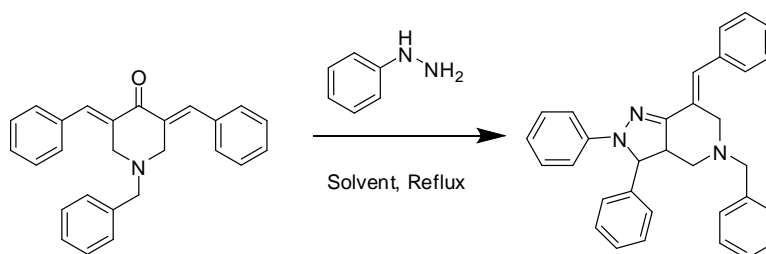
## References

1. N. S. Karthikeyan, B. U. Mahesh, K. Sathiyarayanan, P. Raghavaiah and R. S. Rathore, *Acta Cryst.*, 2010, E66, o1734.
2. S. Prasad, S. Mitra, E. Subramanian, D. Velmurugan, R. B. Rao and P. Balaram, *Biochem. Biophys. Res. Commun.*, 1994, 198, 424.
3. A. N. M. M. Rahman, R. Bishop, D. C. Craig and M. L. Scudder, *Org. Biomol. Chem.*, 2004, 2, 175.
4. B. Nagaraj, T. Narasimhamurthy, H. S. Yathirajan, P. Nagaraja, R. S. Narasegowda and R. S. Rathore, *Acta Cryst. C*, 2005, 61, o177.

Scheme S1



**Table S1** Choice of solvents for the synthesis of **1** under reflux conditions<sup>a</sup>



Entry	Solvent	Time(h)	% Yield <sup>b</sup>
1	Methanol	48	60
2	Ethanol	48	58
3	Iso-Propanol	14	86
4	n-Butanol	18	74
5	Acetic acid	36	68
6	Acetonitrile	36	30
7	Toluene	48	46
8	Ethyl acetate	48	10
9	Chloroform	24	0
10	Tetrahydrofuran	24	0
11	Dichloromethane	24	0
12	Acetone	24	0
13	Diethyl ether	24	0
14	Dimethylformamide	36	70
15	Dimethylsulfoxide	36	74

<sup>a</sup>All the reactions were carried out by employing 0.001mol of curcumin derivatives (3a-3p), 0.001 mol of phenyl hydrazine in 10 ml of given solvent <sup>b</sup>isolated yields

**Table S2** Crystal data of selected compound

compounds	<b>1c</b>	<b>1f</b>	<b>1b</b>
Chemical formula	C <sub>32</sub> H <sub>27</sub> F <sub>2</sub> N <sub>3</sub>	C <sub>32</sub> H <sub>25</sub> Cl <sub>4</sub> N <sub>3</sub>	C <sub>34</sub> H <sub>33</sub> N <sub>3</sub> O <sub>2</sub>
Molecular Weight	491.57	593.35	515.63
Crystal size (mm)	0.6×0.4×0.2	0.4×0.4×0.4	0.4×0.4×0.3
Morphology	block, colorless	block, colorless	block, colorless
Crystal system	Triclinic	Monoclinic	Triclinic
Space group	P1bar	P2 <sub>1</sub> /c	P1bar
Unit cell parameters, a(Å), b(Å), c(c), α(°), β(°), γ(°)	12.8417(9), 14.1308(9), 16.3607(11), 81.235(5), 69.026(6), 69.420(6)	13.7524(4), 15.4132(5), 13.4826(4), 90.0, 101.715(3), 90.0	14.9779(8), 14.9860(8), 15.5420(9), 67.173(5), 88.293(4), 62.667(5)
Volume (Å <sup>3</sup> )	2594.1(3)	2798.36(15)	2809.1(3)
Z/Z'	4/2	4/1	4/2
Cell measuring reflections	6319	12840	7106
θ-range (°)	2.7-29.4	2.6-29.2	2.6-29.2
μ(mm <sup>-1</sup> )absorption correction	0.084, multi-scan	0.451, multi-scan	0.076, multi-scan
F(000)	1032	1224	1096
D <sub>x</sub> (calculated) (g cm <sup>-3</sup> )	0.071	0.224	2.318
<b>Data Collection</b>			
Radiation (Å)	0.71073 (MoKα)	0.71073 (MoKα)	0.71073 (MoKα)
Temperature (°K)	293	293	293
θ Range (°)	2.7-26.0	2.6-26.0	2.6-26.0
Indices	h = -15 → 15 k = -17 → 17 l = -19 → 20	h = -16 → 16 k = -19 → 18 l = -16 → 16	h = -18 → 18 k = -18 → 18 l = -19 → 19
Scan type	ω scans	ω scans	ω scans
Independent reflections	10192	5500	11037
Observed Reflections [I > 2σ(I)]	4826	3932	7218
<b>Refinement</b>			
Final Indices	R = 0.0534, wR = 0.1529	R = 0.0482, wR = 0.1216	R = 0.0529, wR = 0.1232
Goodness of fit (S)	0.997	1.024	1.020
Extinction coefficient	nil	nil	nil
(Δ/σ) <sub>max</sub>	0.0	0.0	0.0
Δρ <sub>max</sub> and Δρ <sub>min</sub> (e Å <sup>-3</sup> )	0.177, -0.198	0.371, -0.406	0.158, -0.207
Data/restraints/ parameter	10192/0/668	5500/0/352	11037/0/707

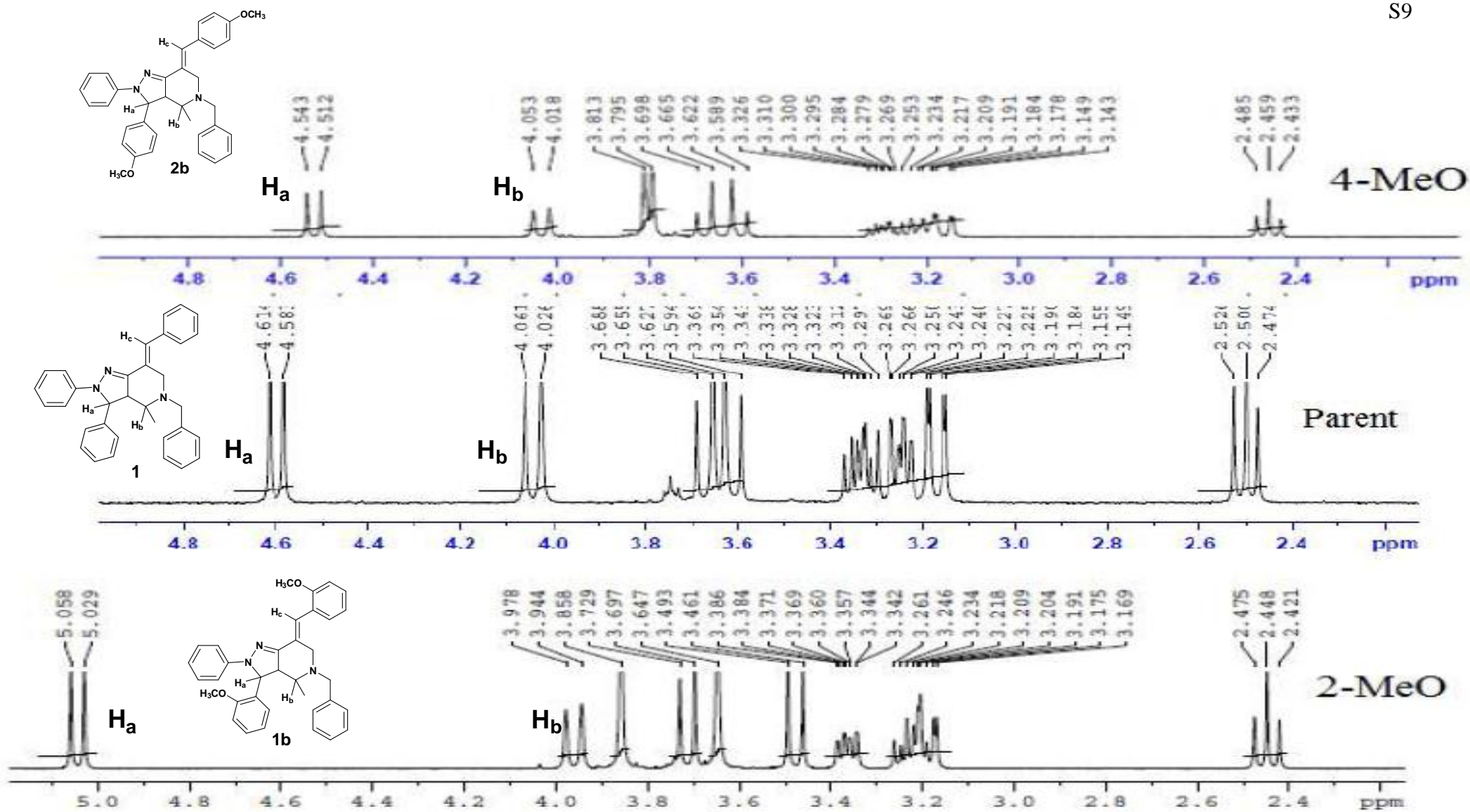
$w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP]$  where  $P = (F_o^2 + 2F_c^2)/3$ , parameters  $a$  and  $b$  are:  
 0.0471, 0.0 (1c), 0.0473, 1.7119 (1f) and 0.0425, 0.4372 (1b), respectively.

**Table S3** Interactions observed in **1b**, **1c**, and **1f**

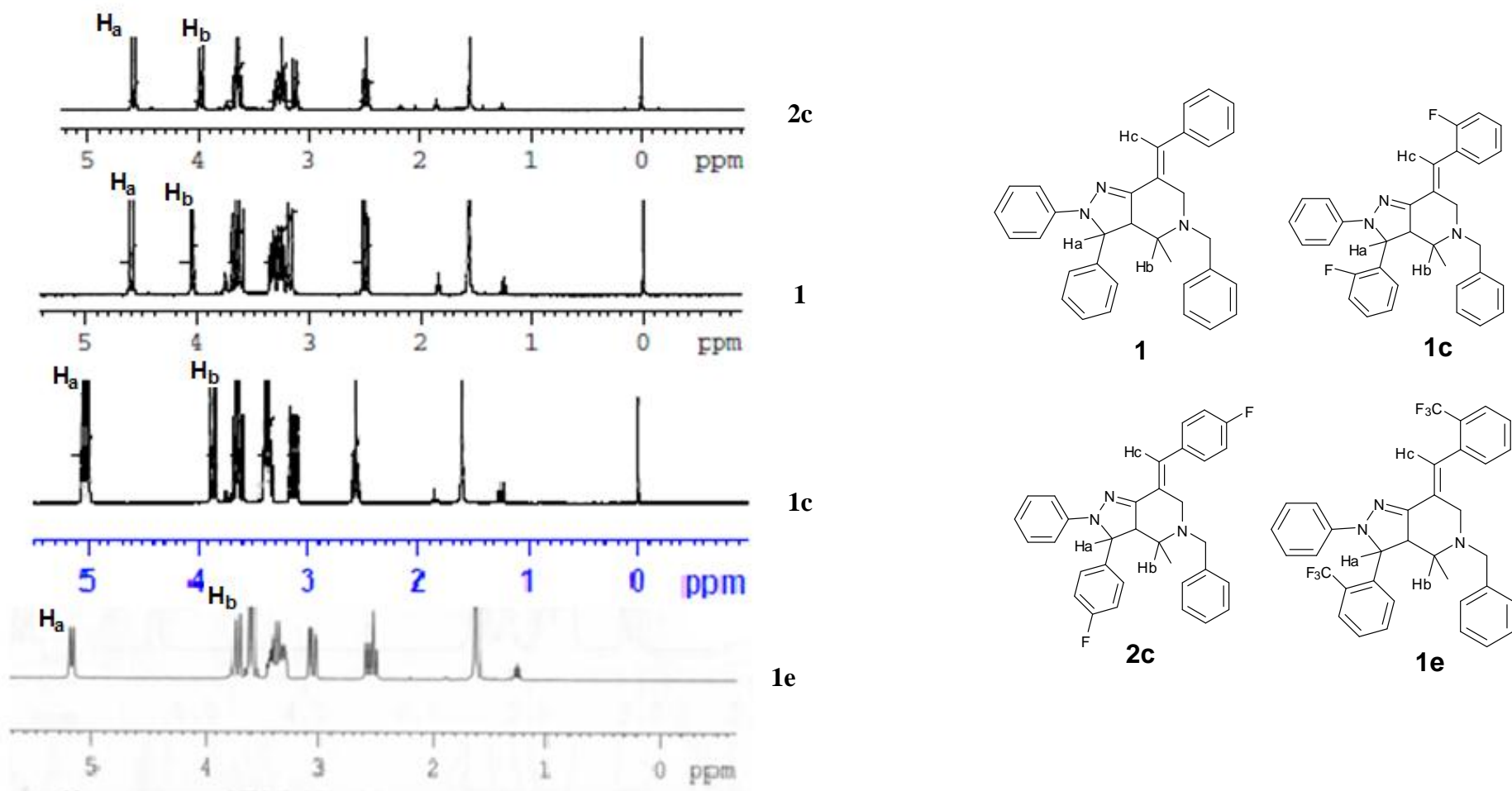
	Interactions	D-H...A	D-H (Å)	H...A (Å)	D...A (Å)	D-H...A (°)
<b>1c</b>	Intra-molecular	C3A-H3A...F1A	0.98	2.39	2.781(4)	103
		C3B-H3B...F1B	0.98	2.44	2.781(4)	100
		C13A-H13A...N1A	0.93	2.40	2.730(3)	100
	Inter-molecular	C32A-H32A...Cg10 <sup>i</sup>	0.93	2.97	3.761(5)	144
<b>1f</b>	Intra-molecular	C3-H3...C11	0.98	2.62	3.135(3)	113
		C27-H27...C13	0.93	2.74	3.034(3)	100
<b>1b</b>	Intra-molecular	C4A-H4A2...O1A	0.97	2.41	3.079(2)	126
		C4B-H4B2...O1B	0.97	2.40	3.072(2)	126
		C19A-H19A...N2A	0.93	2.48	2.844(3)	103
		C19B-H19B...N2B	0.93	2.48	2.842(3)	103
		C20A-H20A...Cg11	0.97	2.96	3.923(3)	172
		C34B-H34F...Cg12	0.96	2.99	3.772(3)	140
	Inter-molecular	C12B-H12B...Cg13 <sup>ii</sup>	0.93	2.95	3.718(3)	141
		C16B-H16B...Cg13 <sup>iii</sup>	0.93	2.93	3.776(3)	152
		C25A-H25A...Cg10 <sup>iv</sup>	0.93	2.86	3.684(3)	149

Symmetry codes (i) 1-X,1-Y,1-Z, (ii), (iii).

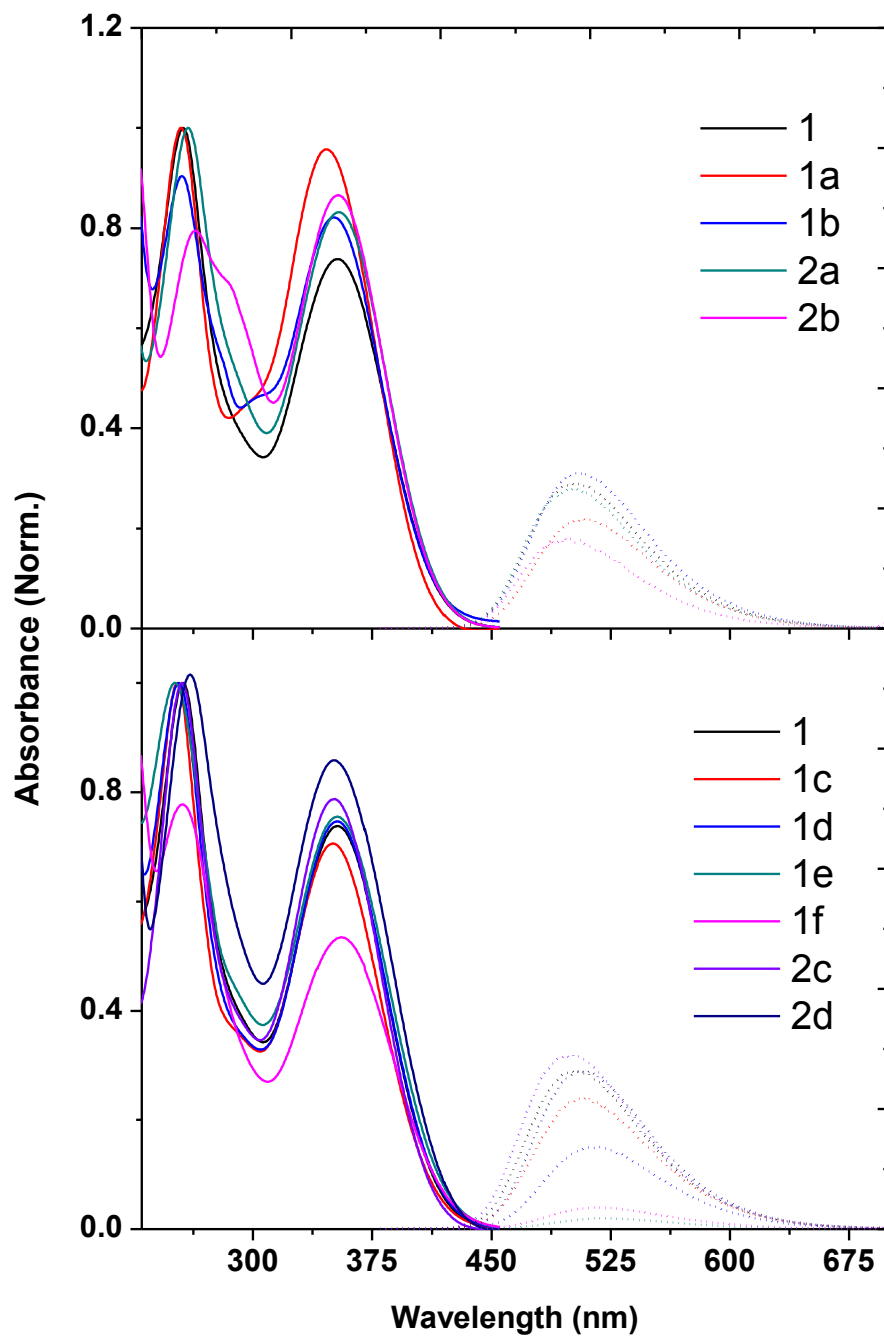




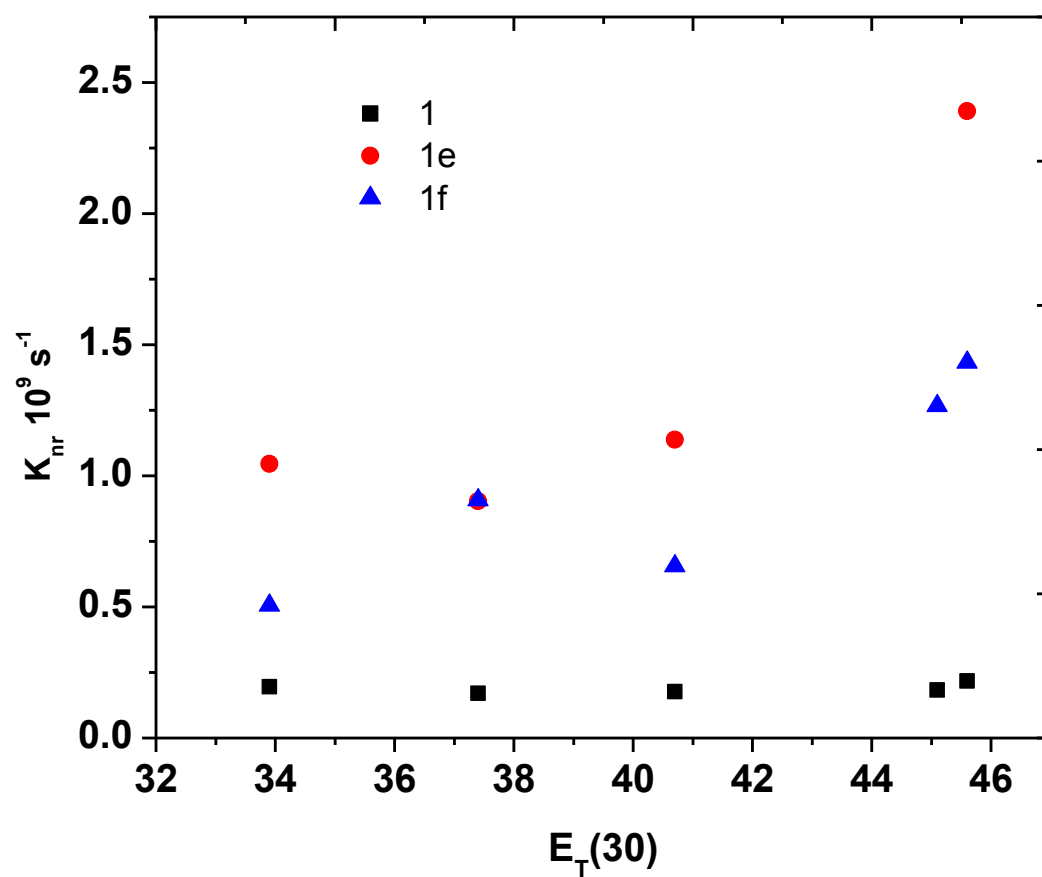
**Figure S1** Deshielding of proton H<sub>a</sub> and shielding of proton H<sub>b</sub> with respect to position



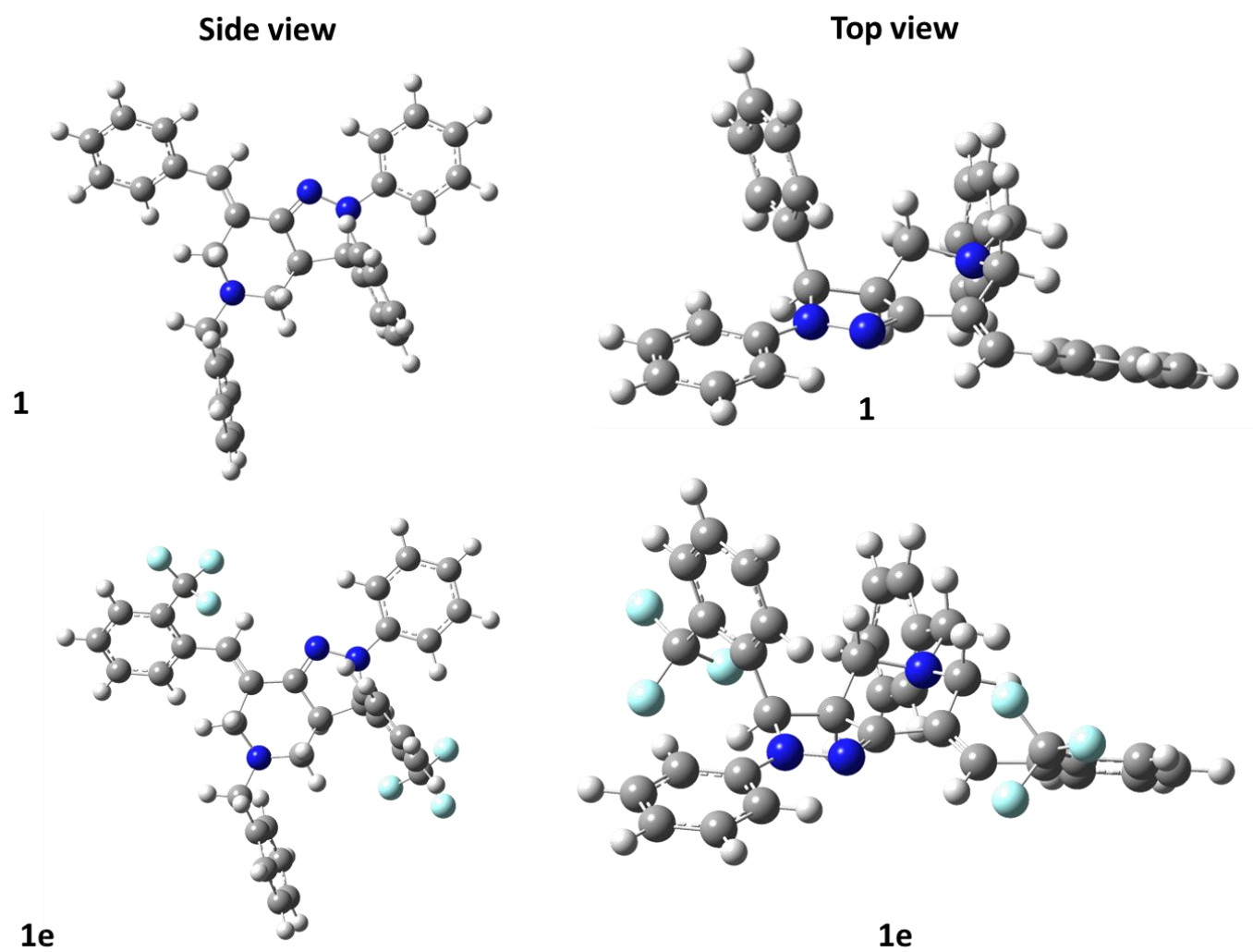
**Figure S2** Deshielding of proton H<sub>a</sub> and Shielding of proton H<sub>b</sub> with respect to position and bulkiness



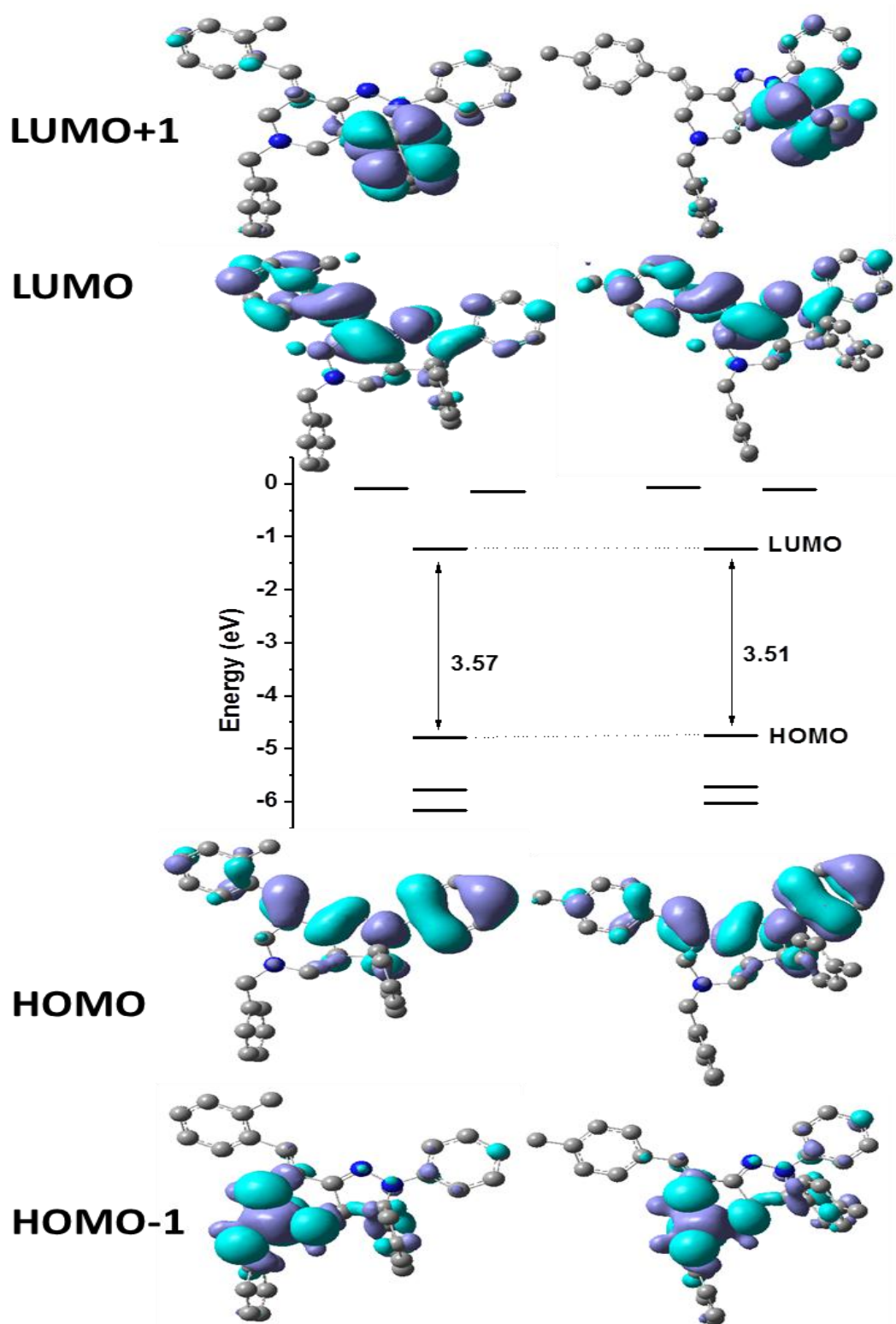
**Figure S3** UV-visible and fluorescence spectra of pyrazolo pyridines dissolved in acetonitrile.



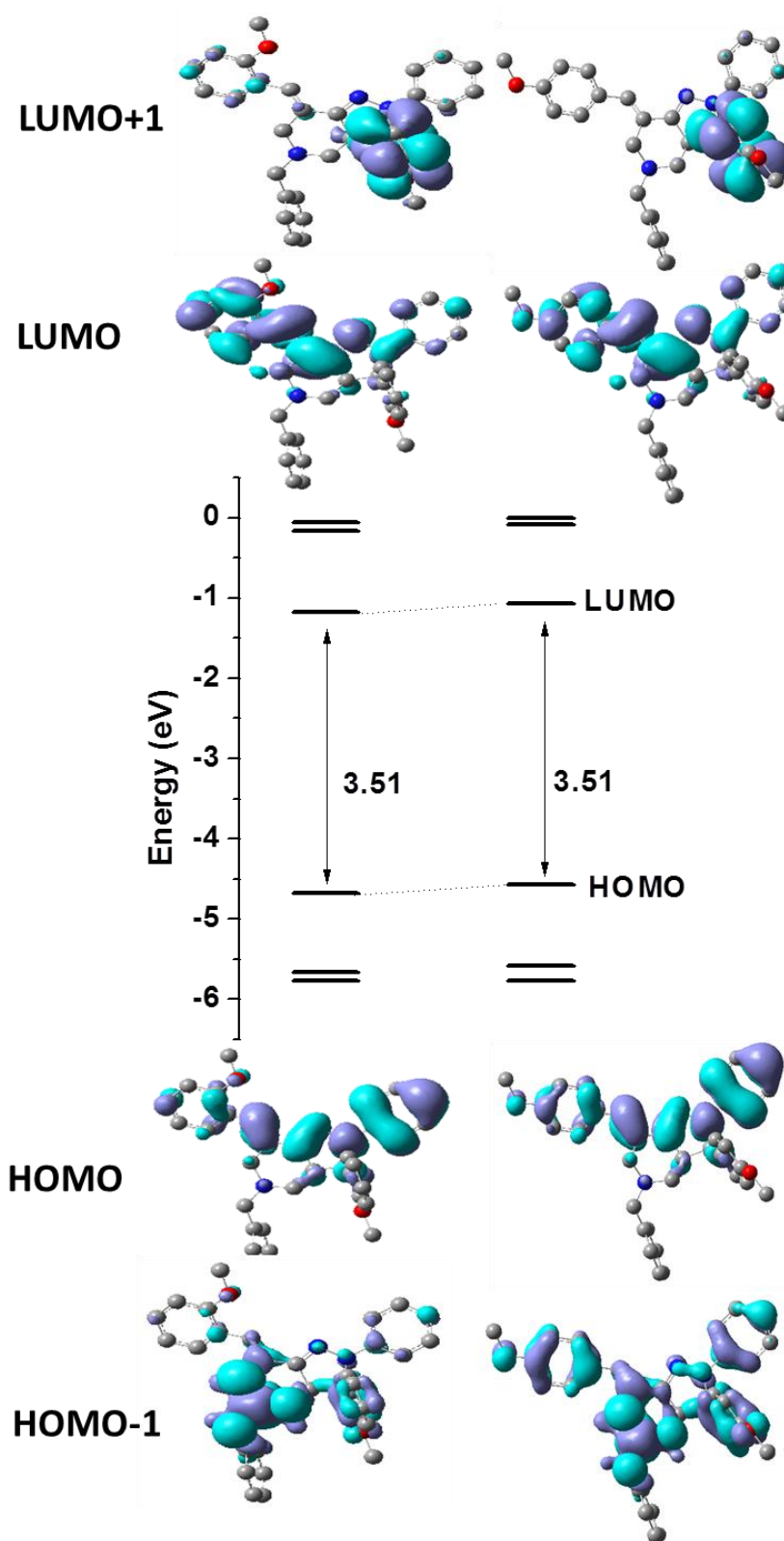
**Figure S4** Plot of solvent polarity parameter  $E_T(30)$  versus non-radiative rate constant ( $K_{nr}$ ) of **1**, **1e**, and **1f**



**Figure S5** Optimized geometries of **1** and **1e** using Gaussian 03 at B3LYP/6-31G level

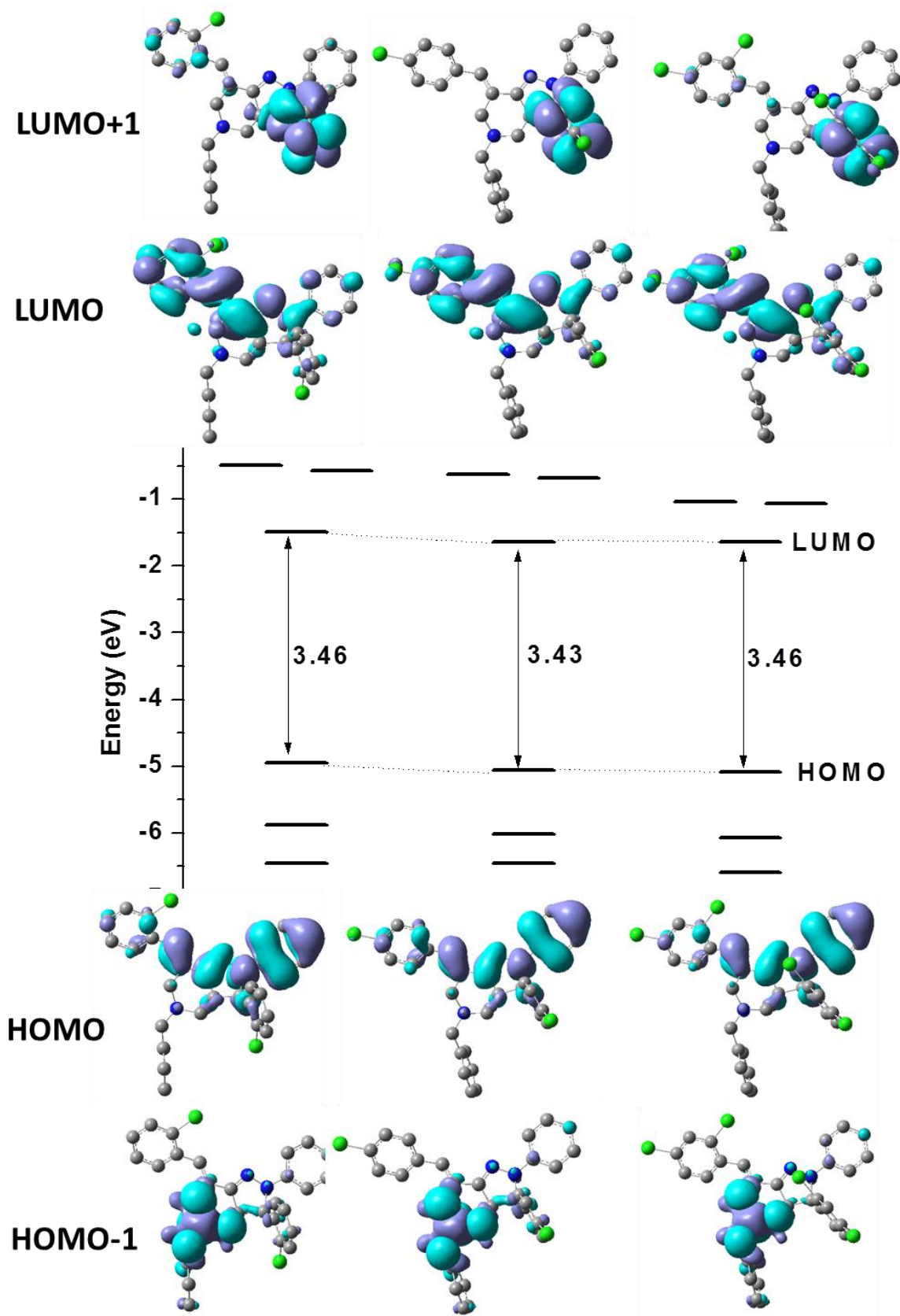


**Figure S6** Molecular orbital diagrams of **1a** and **2a** calculated using Gaussian 03 at B3LYP/6-31G level of theory. Hydrogen atoms are omitted for clarity.



**Figure S7** Molecular orbital diagrams of **1b** and **2b** calculated using Gaussian 03 at B3LYP/6-31G level of theory. Hydrogen atoms are omitted for clarity.





**Figure S8** Molecular orbital diagrams of **1d**, **2d**, and **1f** calculated using Gaussian 03 at B3LYP/6-31G level of theory. Hydrogen atoms are omitted for clarity.



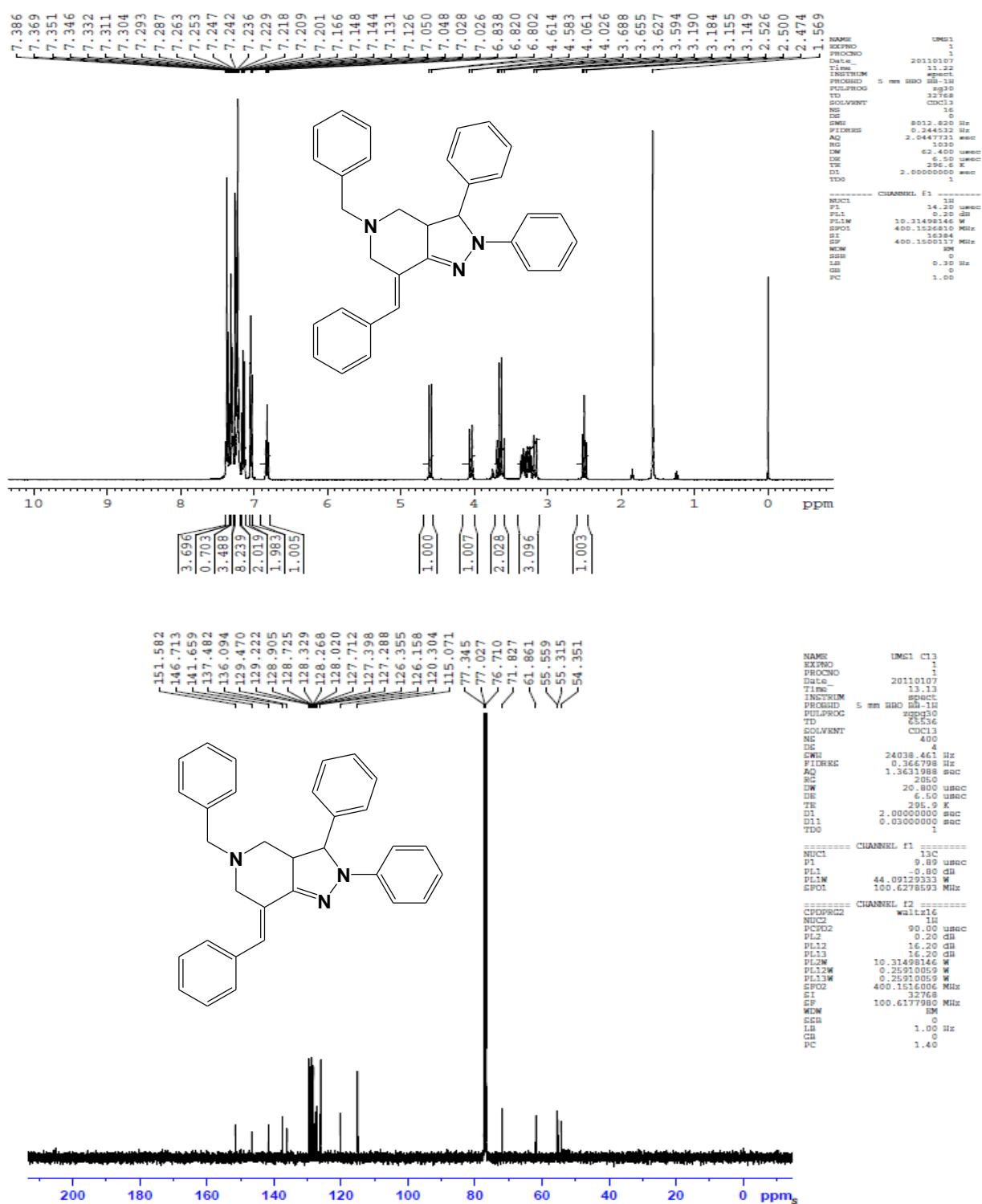


Figure S9 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 1



Figure S10 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 1a

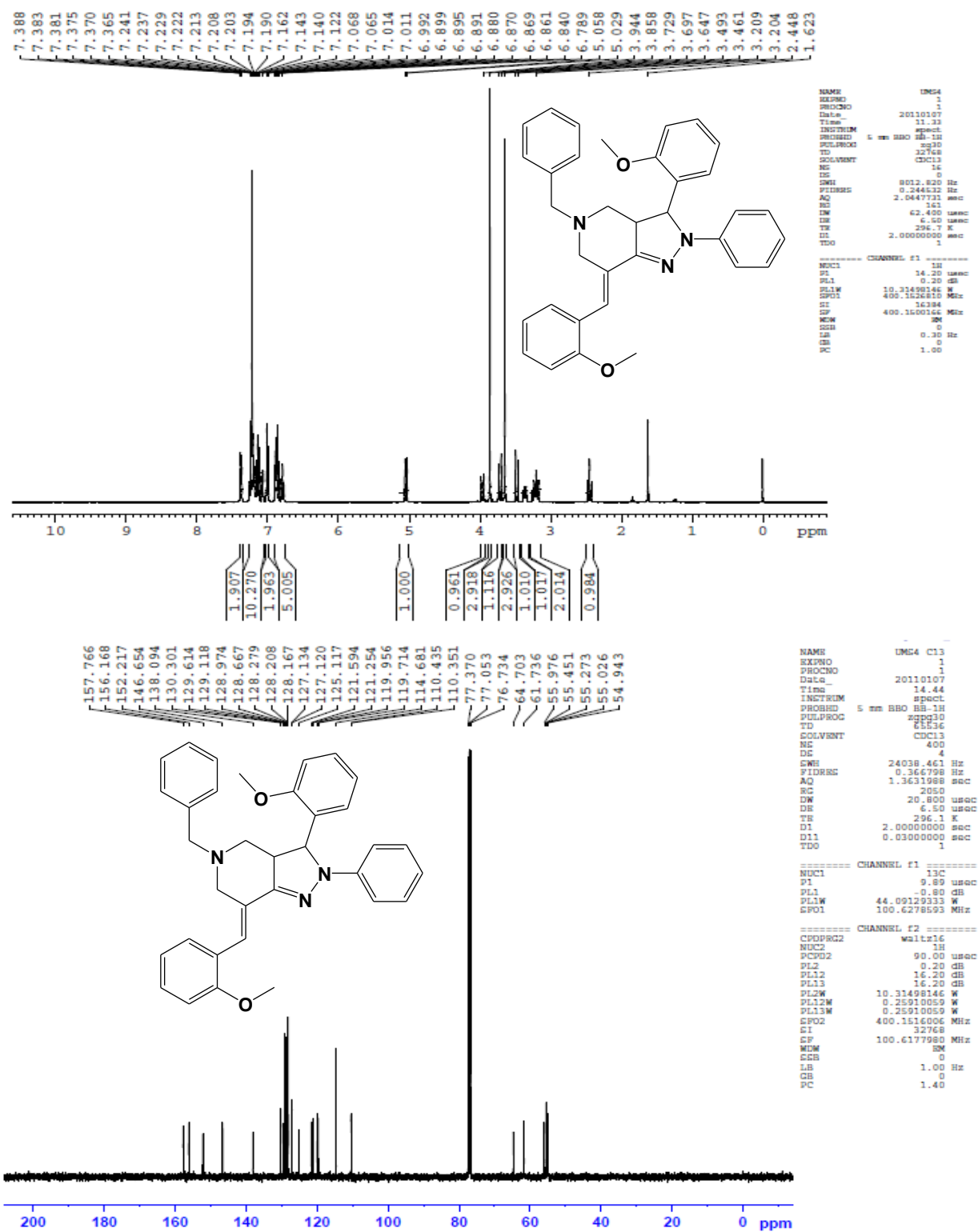


Figure S11 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 1b

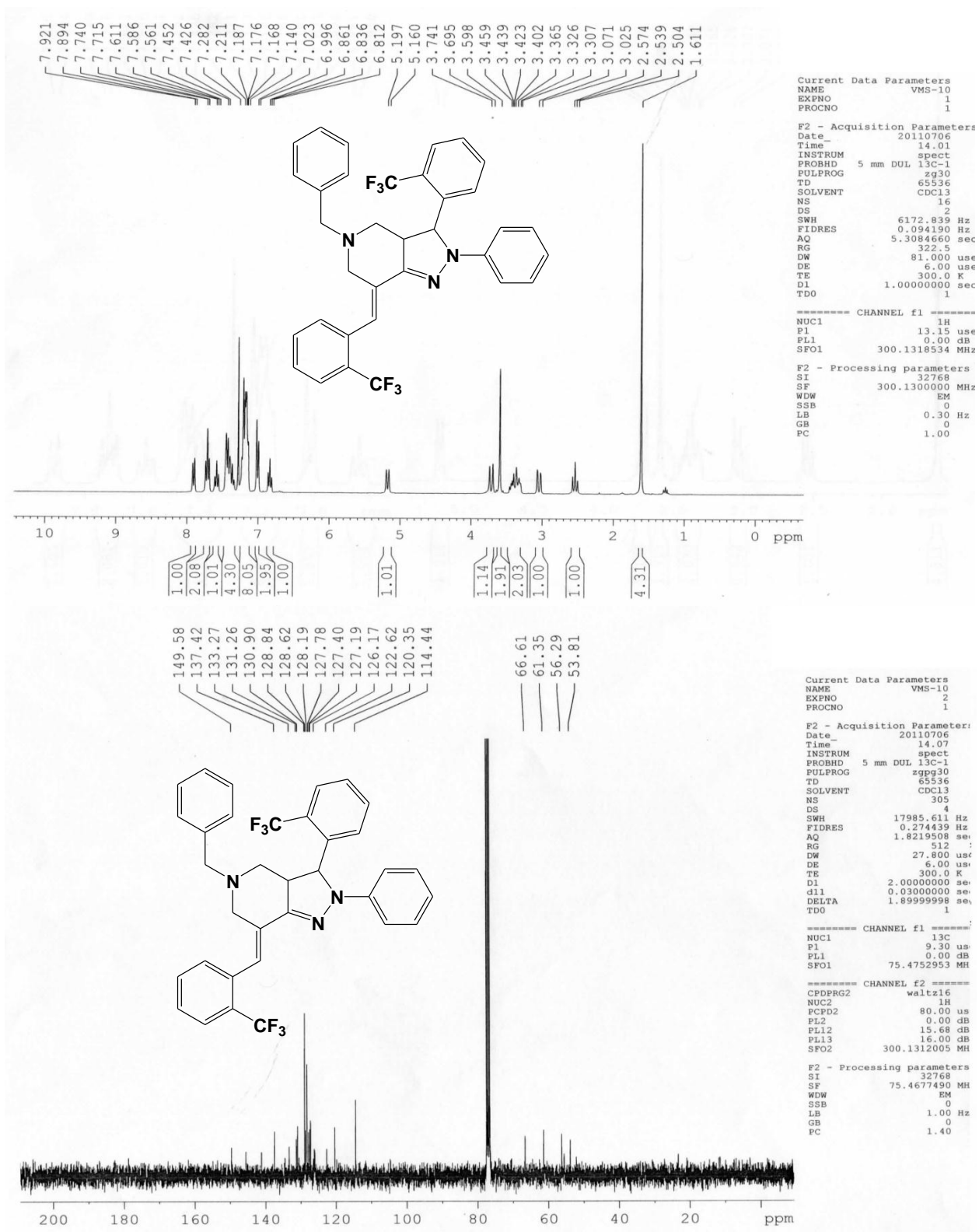


Figure S12 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 1e

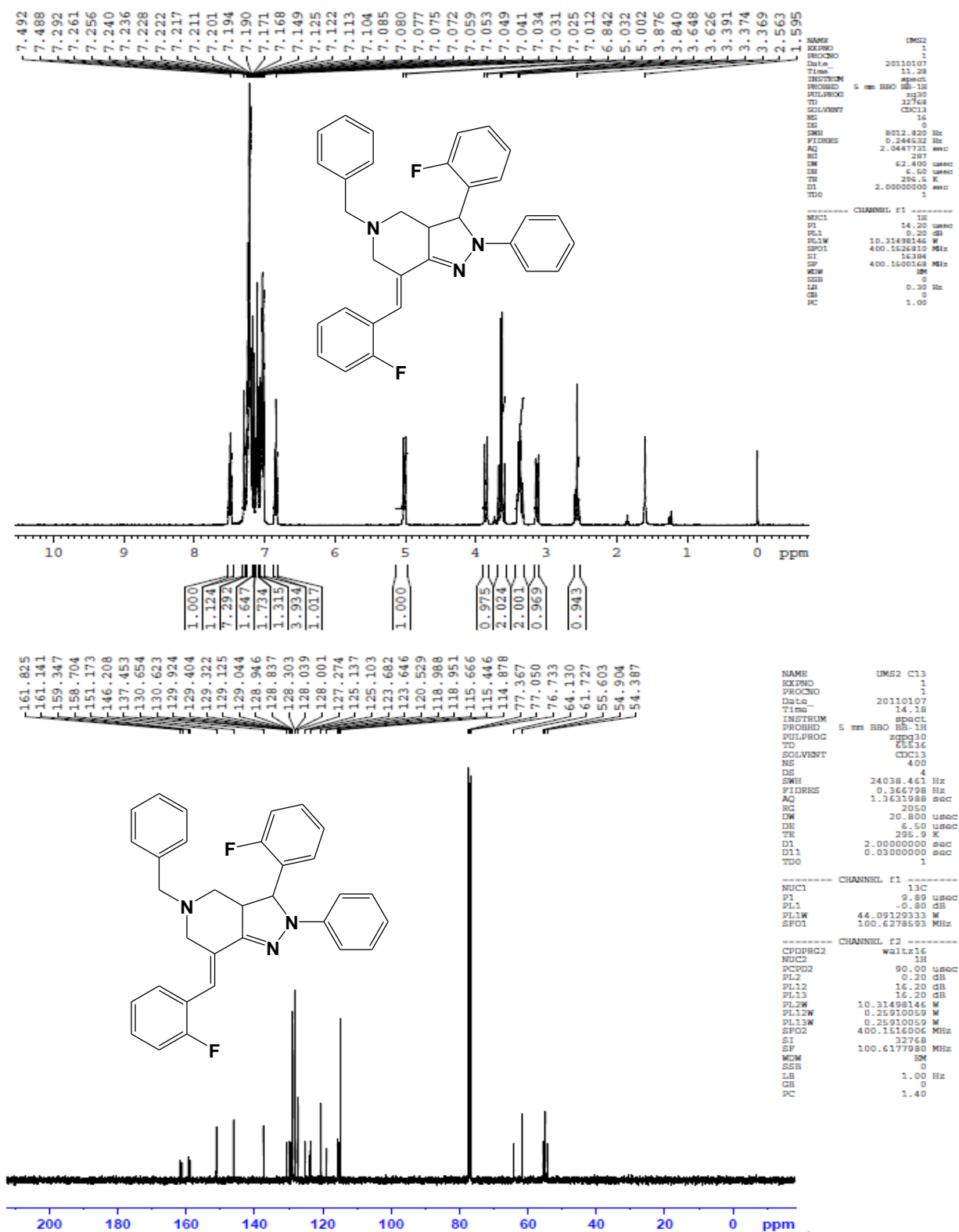


Figure S13 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 1c

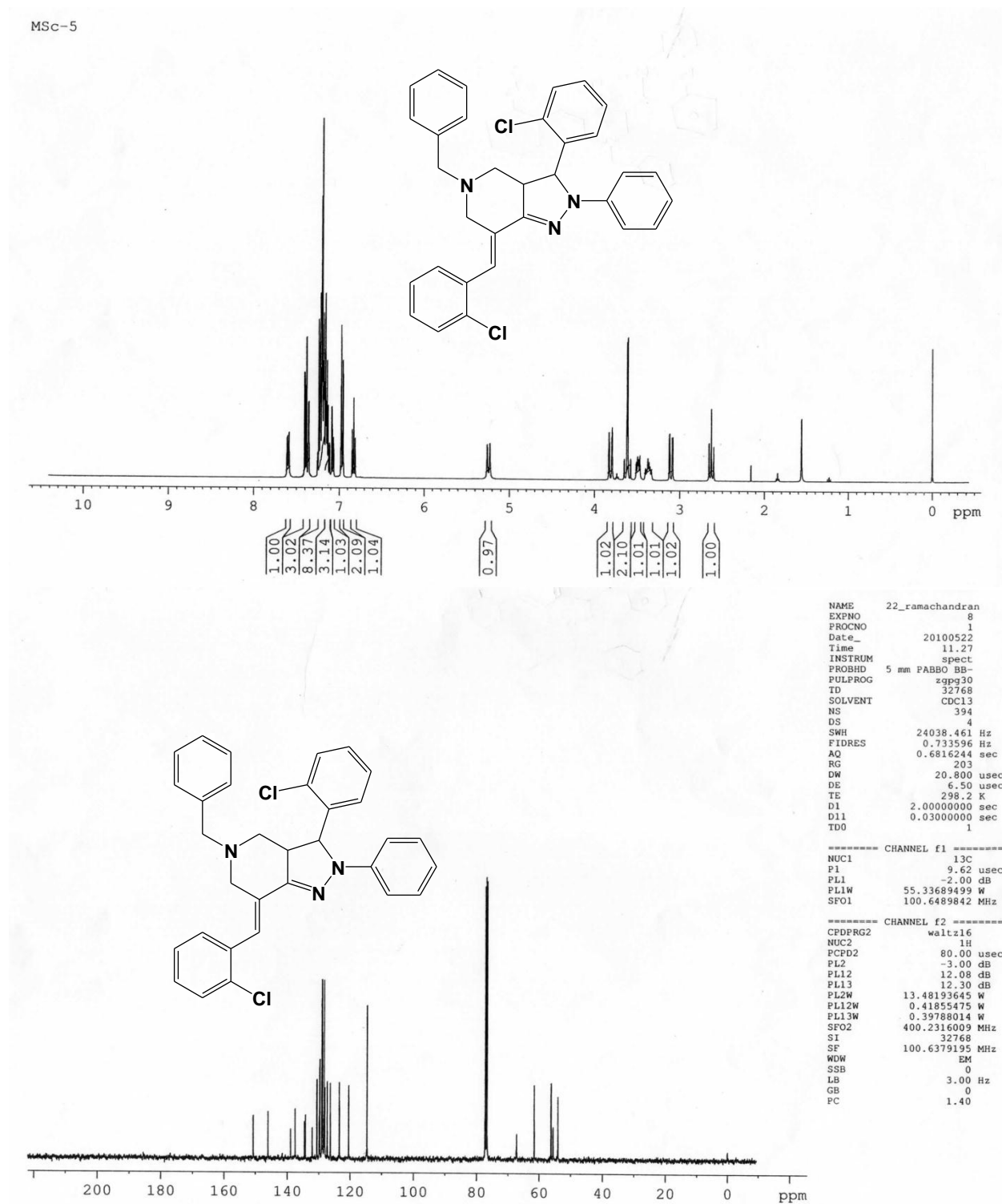


Figure S14  $^1\text{H}$  (top) and  $^{13}\text{C}$  (bottom)-NMR spectra of compound **1d**



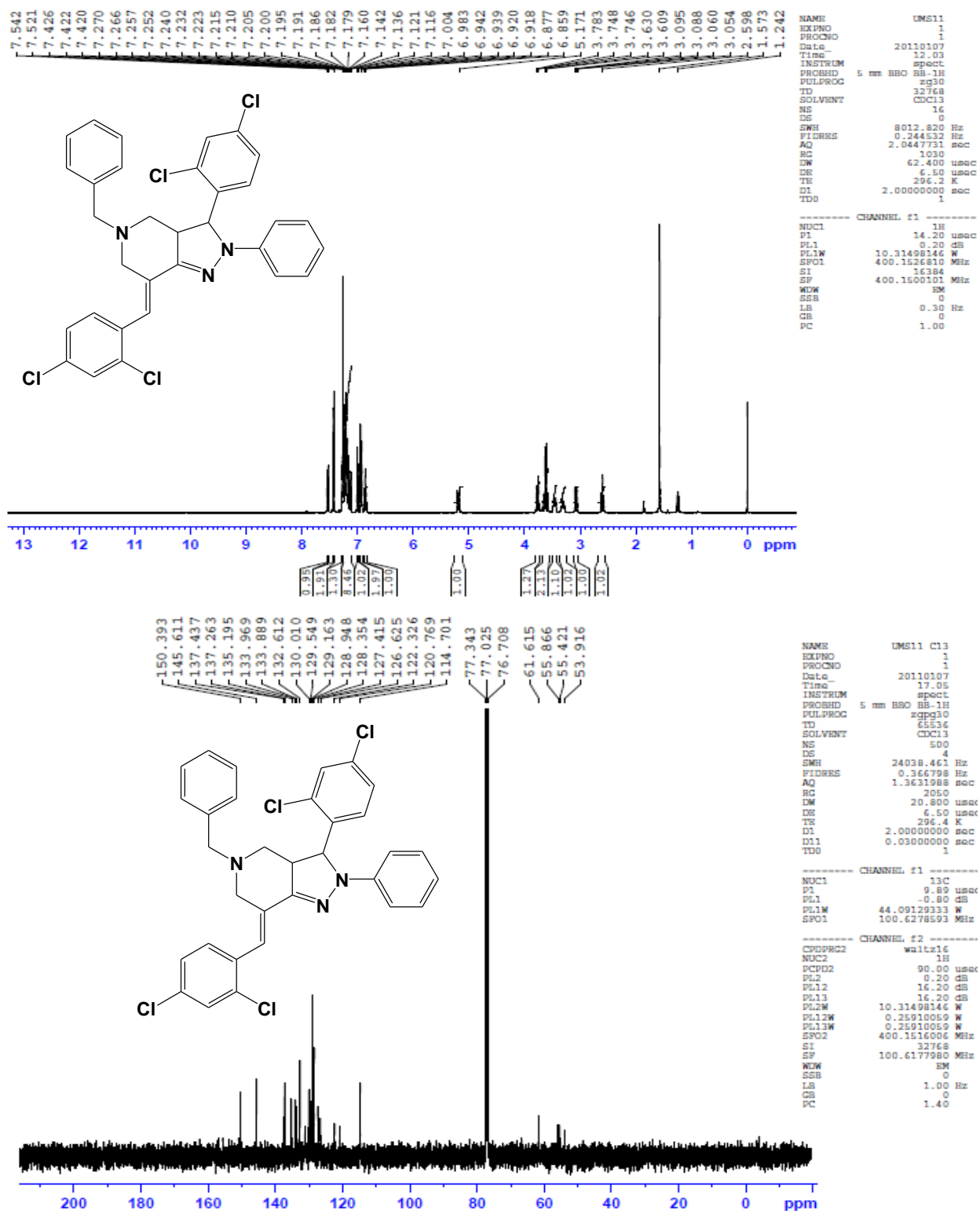


Figure S15 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound **1f**

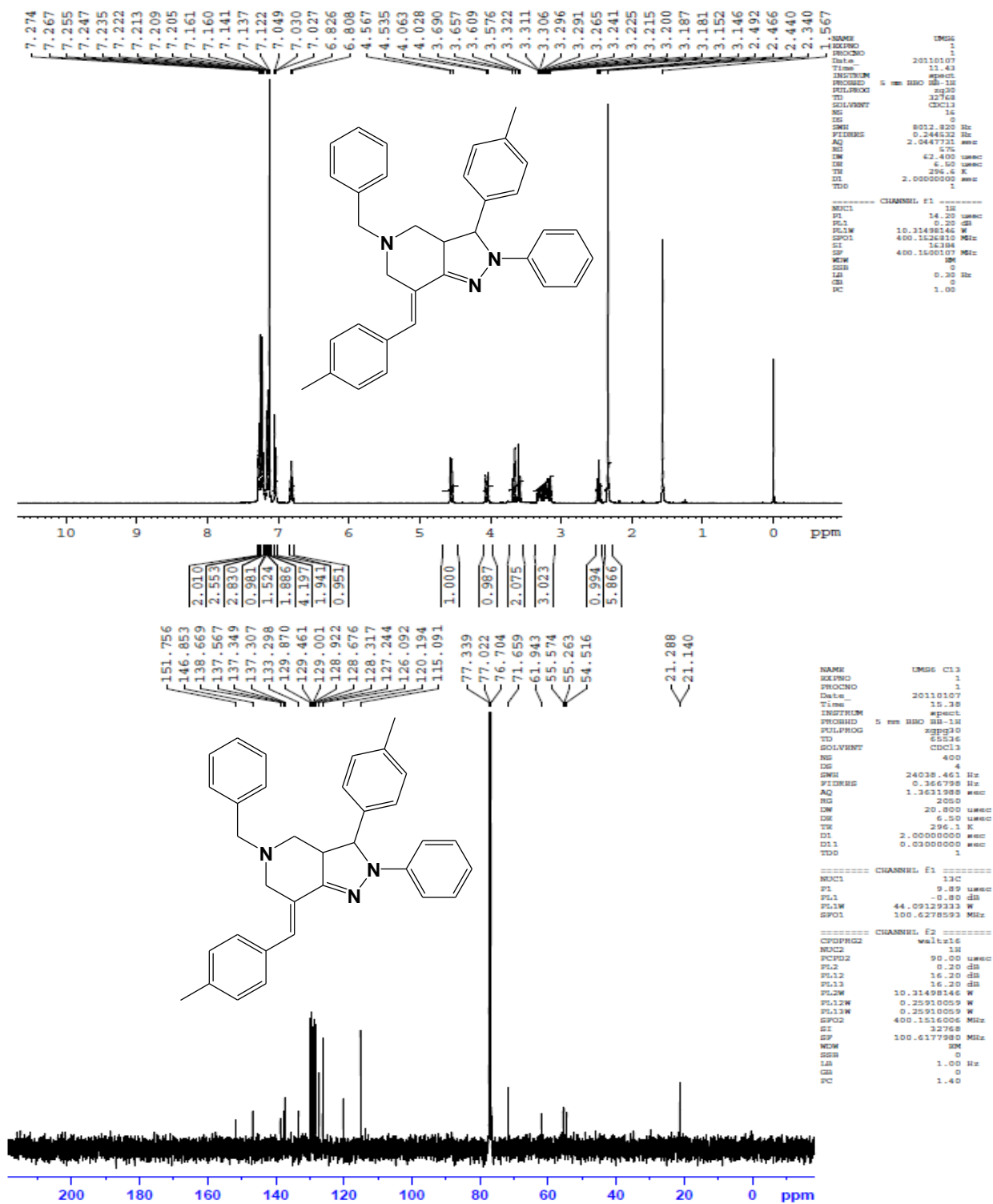
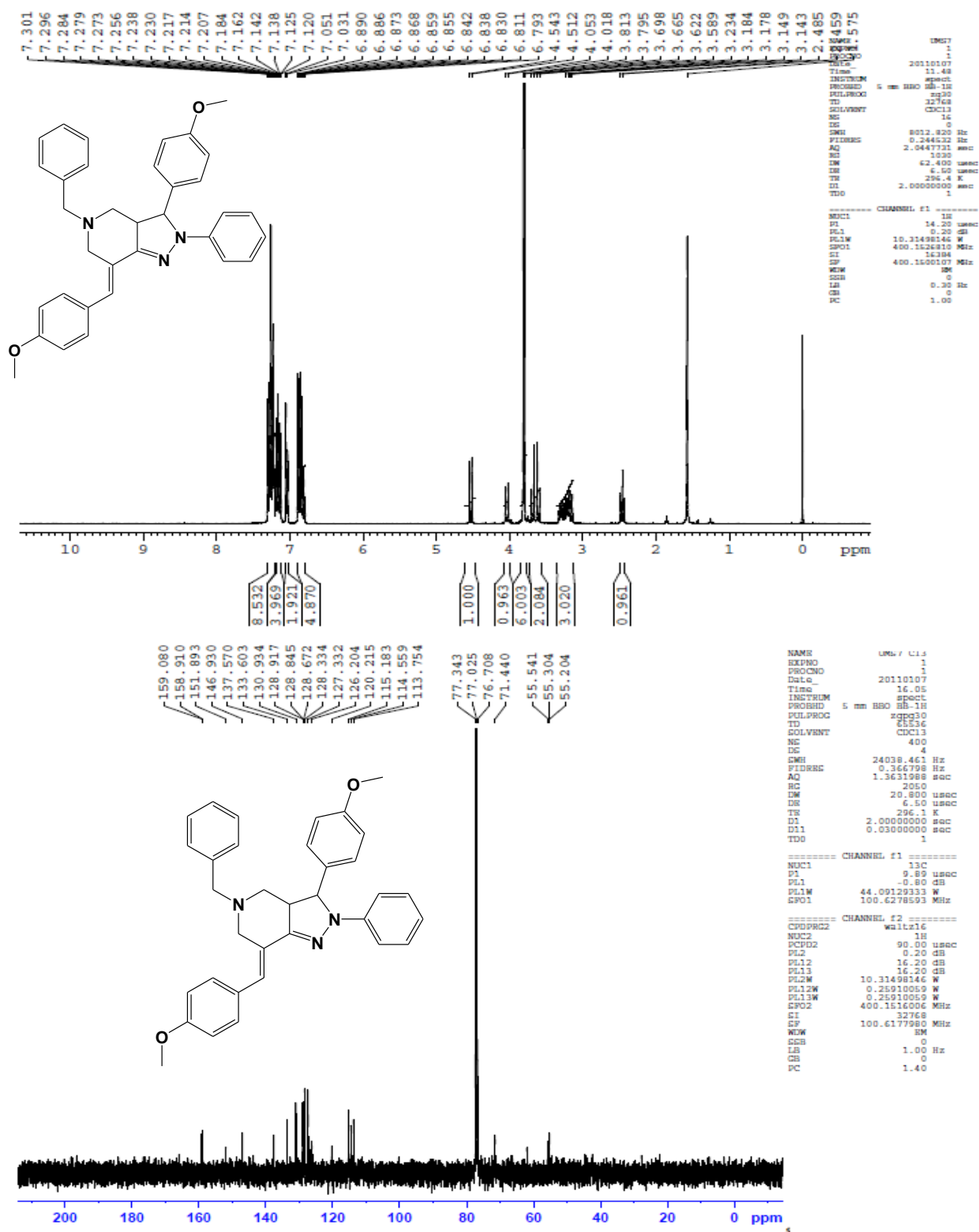


Figure S16 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 2a





**Figure S17** <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound **2b**



Figure S18 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 2c

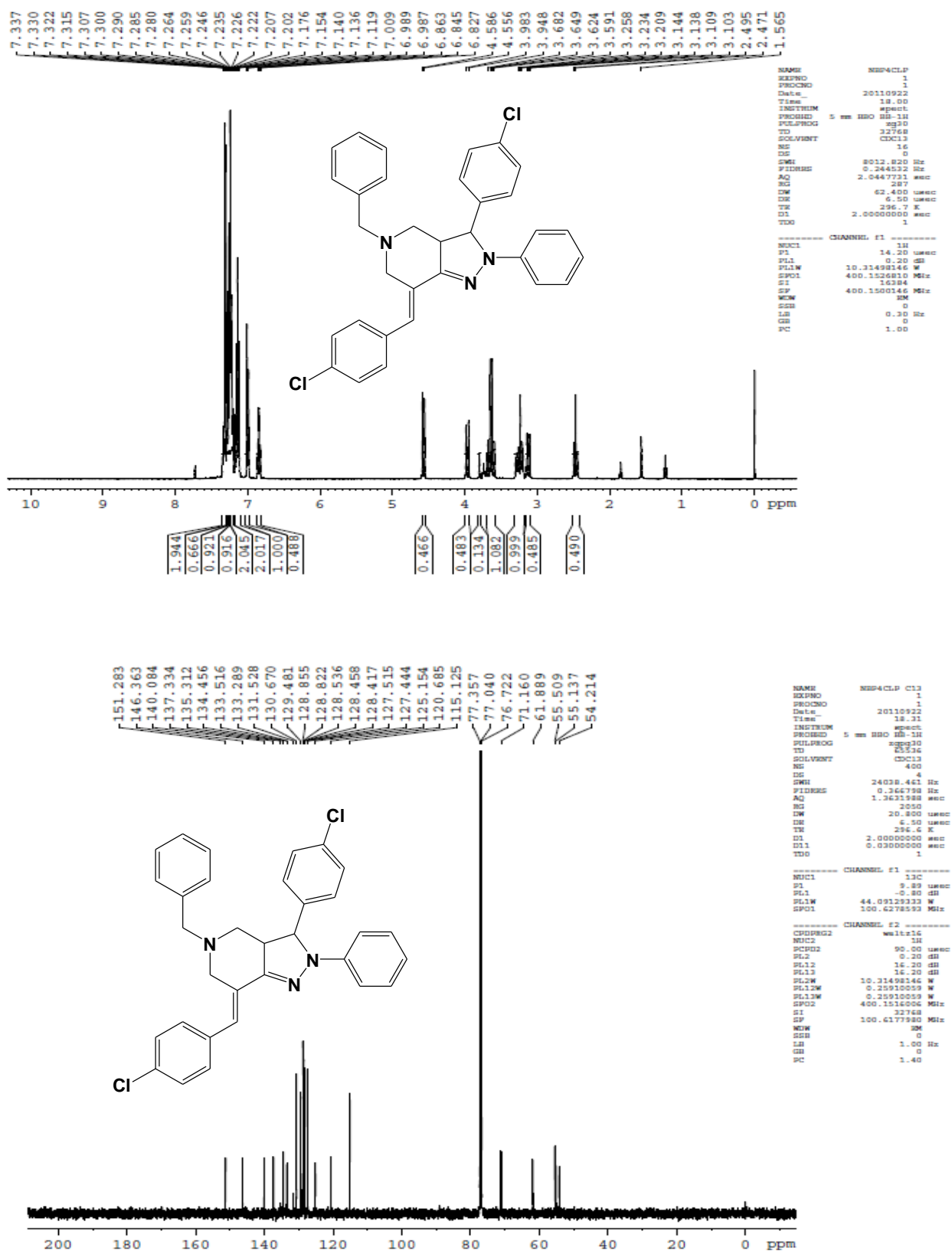


Figure S19 <sup>1</sup>H (top) and <sup>13</sup>C (bottom)-NMR spectra of compound 2d

### Cartesian coordinates of compound 1

C	0.34732600	-0.54418000	-0.61765000
C	0.19108200	0.93678100	-0.30921800
C	-1.11129200	1.50370800	0.02384100
C	-2.05575500	0.51004600	0.69442200
C	-0.55746700	-1.39637400	0.27741400
C	1.90101300	-0.72583000	-0.56008800
H	-1.82877400	0.51996900	1.78367500
H	-0.20660400	-1.37125900	1.32519800
C	-1.36811500	2.82394600	-0.14990500
C	-2.60864700	3.57451900	0.12699500
C	-2.51973200	4.83402500	0.75821500
C	-3.88505400	3.12271300	-0.27118600
C	-3.66357800	5.59202900	1.02006800
H	-1.54346100	5.20936900	1.05109300
C	-5.02866100	3.88406600	-0.01346600
H	-3.97459500	2.19184400	-0.82128100
C	-4.92514400	5.11764500	0.64039800
H	-3.57049300	6.55341700	1.51581900
H	-5.99944500	3.52017600	-0.33604900
H	-5.81411500	5.70780000	0.83882200
H	-0.53727300	3.42654400	-0.51487400
H	-0.55991000	-2.43988700	-0.04794700
H	-3.09333900	0.82391400	0.59064200
H	0.01097100	-0.70246300	-1.65272000
C	3.63990100	1.14440400	-0.93542600
C	3.91772200	2.52721600	-0.98841800
C	4.68651100	0.22485200	-1.15374000
C	5.21364600	2.96762400	-1.25376800
H	3.11179600	3.22955900	-0.82325900
C	5.97839100	0.68650300	-1.42080800
H	4.50108400	-0.84098700	-1.10089800
C	6.25555900	2.05618800	-1.47401000
H	5.41005500	4.03504500	-1.29141700
H	6.77215800	-0.03613300	-1.58470300
H	7.26084100	2.40733500	-1.68136100
C	2.43489000	-1.40370300	0.69767400
C	2.67992800	-0.68036400	1.87616600
C	2.65742500	-2.79050600	0.69132000
C	3.13143100	-1.33517000	3.02670500
H	2.53072300	0.39418200	1.88296700
C	3.10685100	-3.44645500	1.84267300
H	2.48295400	-3.35852200	-0.21933500
C	3.34371000	-2.71940500	3.01471500
H	3.32216200	-0.76421500	3.92997900
H	3.27834600	-4.51818100	1.82168000
H	3.69777500	-3.22465100	3.90767400
N	2.33886700	0.69413300	-0.68875800
N	1.31356300	1.59630300	-0.40565300
N	-1.93111800	-0.86118100	0.15364500
H	2.23860000	-1.28958900	-1.43559700
C	-2.96901000	-1.77627100	0.66443600
H	-3.91672300	-1.22112600	0.64733500
C	-3.11531000	-3.03698400	-0.17004200

C	-3.14443800	-2.96287500	-1.57343500
C	-3.26764100	-4.28867900	0.44546000
C	-3.32775700	-4.11644400	-2.34152400
H	-3.00868200	-1.99656200	-2.04769400
C	-3.45624600	-5.44481900	-0.32200600
H	-3.23753300	-4.35829400	1.52974400
C	-3.48714600	-5.36141200	-1.71805300
H	-3.34821800	-4.04572700	-3.42490500
H	-3.57247800	-6.40604100	0.16922700
H	-3.63058700	-6.25648500	-2.31532000
H	-2.79077500	-2.05457200	1.72233100