Solvent free, Ni-nanoparticles catalyzed synthesis and photophysical studies of novel 2*H*-indazolo[2,1-*b*] phthalazine-trione derivatives

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1. Experimental Section

Melting points were determined in open capillaries and are uncorrected. IR spectra were recorded on Spectrum BX FT-IR, Perkin Elmer (v_{max} in cm⁻¹) on KBr disks. ¹H NMR and ¹³C NMR (400, 300 MHz and 100, 75 MHz respectively) spectra were recorded on Bruker Avance II-400 spectrometer in CDCl₃ (chemical shifts in δ with TMS as internal standard). Mass spectra were recorded on Waters ZQ-4000. Transmission Electron Microscope (TEM) was recorded on JEOL JSM 100CX. Scanning electron microscope (SEM) was recorded on JSM-6360 (JEOL). XRD was recorded on Bruker D8 XRD instrument SWAX .CHN were recorded on CHN-OS analyzer (Perkin Elmer 2400, Series II). Silica gel G (E-mark, India) was used for TLC. Hexane refers to the fraction boiling between 60 °C and 80 °C.

The organic solvents (ethylacetate and DMSO) used for the photophysical characterization were of spectroscopic grade (>99.5%) as received from Alfa Aesar. The solvent water was obtained from Elix10 water purification system (Millipore India Pvt. Ltd.). Sample concentration ($0.5 \sim 1 \mu M$) was kept sufficiently dilute to avoid any effect from aggregation and/or inner filter effect.

Steady state absorption and fluorescence emission/excitation spectra were recorded in Lambda25 and LS-55 (both from PerkinElmer Inc.) spectrometers, respectively. Quartz cuvettes of 10 mm optical path length received from PerkinElmer, USA (part no. B0831009) and Hellma, Germany (type 111-QS) were used for measuring absorption and fluorescence spectra, respectively. For fluorescence emission, the sample was excited at

365 nm unless otherwise mentioned. In all the cases, 10 nm band pass was used in the excitation and 20nm for emission side.

Fluorescence quantum yields (Δf) were calculated by comparing the total fluorescence intensity(F) under the whole fluorescence spectral range with that of quinine bisulfate in 0.5 M H₂SO₄ solution ($\phi_f = 0.546$).²⁵

$$\phi_f^i = \phi_f^s \times \frac{F^i}{F^s} \times \frac{(1 - 10^{-A^i})}{(1 - 10^{-A^s})} \times (\frac{n^i}{n^s})^2$$

where, A^{i} and A^{s} are the optical density of the sample and standard ,respectively, and n^{i} is the refractive index of solvent at 293 K. The relative experimental error of the measured quantum yield was estimated within ±10%.

Fluorescence decay measurements were performed by using time correlated single photon counting (TCSPC) technique using LED based nanosecond time-resolved spectrophotometer from Photon Technology International (PTI), USA. The excitation was done at 365 nm. The instrument response function (IRF) was obtained by using a dilute colloidal suspension of dried non-dairy coffee whitener. The details of the data collection, analysis procedure for both the steady state and time-resolved measurements and method of calculation of fluorescence yield can be found elsewhere.²⁶

General procedure

A mixture of, dimedone **1** (2 mmol), phthalhydrazide **2** (2 mmol), aryl aldehyde **3a-n** (2 mmol) and Ni NPs (10 mol %) was heated at 80 °C for the time mentioned in the **table1**. After completion (TLC), the reaction mixture was cooled to room temperature; 10 mL of ethyl acetate was added to dissolve to it and filtered. The filtrate was recovered and removed in high vacuum and product was purified by column chromatography using ethyl acetate: hexane (4:6) to afford the pure products **4a-n**.

Table S1: Spectral parameters of the synthesized phthalazine derivatives (4a-k) in ethyl acetate (EA) and DMSO^a.

Products	Solvents	λ_{abs} /nm	λ_{em} /nm	φ	Decay parameters ^b		$<\tau>^{c}/ns$
					$\tau_{1}(a_{1})$	$\tau_{2}(a_{2})$	
4a	DMSO	368	466-485	0.030	0.7 (99.8)	13.4 (0.2)	0.73
	EA	364	478	0.006	0.3 (99.8)	4.5 (0.2)	0.31
4b	DMSO	369	461-485	0.031	0.6 (99.6)	3.0 (0.4)	0.60
	EA	365	478	0.007	0.6 (99.7)	3.8 (0.3)	0.61
4c	DMSO	369	464-488	0.032	0.5 (99.8)	5.0 (0.2)	0.51
	EA	365	478	0.008	0.6 (99.2)	5.0 (0.8)	0.64
4d	DMSO	368	460-483	0.049	0.9 (99.8)	11.9(0.2)	0.92
	EA	364	470	0.010	0.5(99.8)	5.2(0.2)	0.51
4e	DMSO	368	460-485	0.008	0.9 (99.0)	6.3(1.0)	0.95
	EA	364	480	0.010	0.6(99.8)	6.2 (0.2)	0.61
4f	DMSO	368	462-485	0.041	0.4(99.7)	2.4(0.3)	0.41

	EA	364	475	0.007	0.5(99.8)	4.0 (0.2)	0.51
4g	DMSO	368	460-480	0.017	0.5(99.9)	3.2 (0.1)	0.50
	EA	364	472	0.007	0.8 (99.7)	4.7(0.3)	0.81
4h	DMSO	368	462-483	0.017	0.6(99.9)	6.8 (0.1)	0.61
	EA	364	470	0.006	1.0(99.7)	14.0 (0.3)	1.04
4i	DMSO	368	461-483	0.054	0.9(99.8)	9.7 (0.2)	0.92
	EA	364	477	0.010	0.4 (99.9)	4.5(0.1)	0.40
4j	DMSO	369	458-485	0.024	0.5 (99.8)	3.7 (0.2)	0.51
	EA	366	506	0.005	0.7(99.7)	5.0(0.3)	0.71
4k	DMSO	368	458-485	0.017	0.5 (99.5)	5.3(0.5)	0.52
	EA	363	472	0.020	0.6(99.8)	6.3(0.2)	0.61

 $^{a}\lambda_{abs}$ = absorbance maxima, λ_{abs} = emission maxima, ϕ = quantum yield, ^bthe measured values are within ± 0.1 ns; ^ccalculated from equation (1b)

Spectral Data

1. Compound 4a



Yellow solid. IR (KBr): 2965, 2375, 1666 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) δ = 8.28-8.17 (m, 2H), 7.79-7.75 (m, 2H), 7.34-7.19 (m, 5H), 6.37 (s, 1H), 3.36-3.13 (AB system, J = 18.4 Hz, 2H), 2.26 (s, 2H), 1.13 (s, 6H). ESI- MS: m/z 373 [M + H]⁺. Anal. Calcd for C₂₃H₂₀N₂O₃: C, 74.18; H, 5.41; N, 7.52. Found: C, 74.07; H, 5.35; N, 7.35.

2. Compound 4b



Yellow solid. IR (KBr): cm⁻¹.¹H NMR (CDCl₃, 400 MHz) δ =8.28-8.18 (m, 2H), 7.78-7.76 (m, 2H), 7.23 7.07 (d, *J* = 7.6 Hz, 2H), (d, *J* = 7.6 Hz, 2H), 6.34 (s, 1H), 3.36-3.13 (AB system, *J* = 18.2 Hz, 2H), 2.26 (s, 2H), 2.22 (s, 3H), 1.13 (s, 6H). ESI- MS: *m*/*z* 387 [M + H]⁺. Anal. Calcd for C₂₄H₂₂N₂O₃: C, 74.59; H, 5.74; N, 7.25. Found: C, 74.51; H, 5.80; N, 7.08.

3. Compound 4c



Yellow solid. IR (KBr): 2963, 2376, 1660 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) δ = 8.28-8.18 (m, 2H), 7.78-7.75 (m, 2H), 7.28 (d, *J* = 8.8 Hz, 2H), 6.79 (d, *J* = 8.4 Hz, 2H), 6.34 (s, 1H), 3.69 (s, 3H), 3.37-3.13 (AB system, *J* = 19.2 Hz, 2H), 2.27 (s, 2H), 1.15 (s, 3H), 1.13 (s, 3H). ESI- MS: *m*/*z* 403 [M + H]⁺. Anal. Calcd for C₂₄H₂₂N₂O₄: C, 71.63; H, 5.51; N, 6.96. Found: C, 71.83; H, 5.65; N, 6.87.

4. Compound 4d



Yellow solid. IR (KBr): 2966, 2375, 1665 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) δ = 8.31-8.17 (m, 2H), 7.82-7.79 (m, 2H), 7.58 (d, *J* = 8.4 Hz, 2H), 7.48 (d, *J* = 8.4 Hz, 2H), 6.38 (s, 1H), 3.34-3.15 (AB system, *J* = 19.2 Hz, 2H), 2.26 (s, 2H), 1.14 (s, 3H), 1.11 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz) δ = 191.0, 154.9, 153.5, 150.5, 140.5, 133.7, 132.9, 131.5, 127.9, 127.6, 127.1, 126.8, 126.7, 117.4, 116.3, 111.5, 63.3, 49.7, 37.0, 33.7, 27.6, 27.3. ESI- MS: *m*/*z* 398 [M + H]⁺. Anal. Calcd for C₂₄H₁₉N₃O₃: C, 72.53; H, 4.82; N, 10.57. Found: C, 72.71; H, 4.70; N, 10.73.

5. Compound 4e



Yellow solid. IR (KBr): 2965, 2375, 1666 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) $\delta = 8.29$ -8.08 (m, 4H), 7.80-7.79 (m, 2H), 7.53 (d, J = 8.4 Hz, 2H), 6.41 (s, 1H), 3.35-3.15 (AB system, J = 19.2 Hz, 2H), 2.25 (s, 2H), 1.13 (s, 3H), 1.10 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz) $\delta = 192.1$, 155.9, 154.5, 151.7, 147.8, 143.4, 134.8, 133.9, 128.9, 128.5, 128.2, 128.0, 127.7, 124.0, 117.2, 64.1, 50.7, 37.9, 34.7, 28.6, 28.3. ESI- MS: m/z 418 [M + H]⁺. Anal. Calcd for C₂₃H₁₉N₃O₅: C, 66.18; H, 4.59; N, 10.07. Found: C, 65.90; H, 4.45; N, 10.23.

6. Compound 4f



Yellow solid. IR (KBr): 2939, 2229, 1666 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) δ = 8.29-8.18 (m, 2H), 7.80-7.78 (m, 2H), 7.30 (d, *J* = 8.8 Hz, 2H), 7.24 (d, *J* = 8.4 Hz, 2H), 6.34 (s, 1H), 3.35-3.13 (AB system, *J* = 18.2 Hz, 2H), 2.26 (s, 2H), 1.13 (s, 6H). ¹³C NMR (CDCl₃, 100 MHz) δ = 192.1, 156.0, 154.3, 151.1, 134.9, 134.6, 134.5, 133.6, 128.97, 128.93, 128.91, 128.5, 128.0, 127.7, 118.0, 64.3, 50.8, 38.0, 34.6, 28.6, 28.4. ESI-MS:

m/z 407, 409 [M + H]⁺. Anal. Calcd for C₂₃H₁₉ ClN₂O₃: C, 67.90; H, 4.71; N, 6.89. Found: C, 68.17; H, 4.87; N, 6.98.

7. Compound 4g



Yellow solid. IR (KBr): cm⁻¹.¹H NMR (CDCl₃, 400 MHz) $\delta = 8.29$ -8.18 (m, 2H), 7.80-7.78 (m, 2H), 7.40 (d, J = 8.4 Hz, 2H), 7.23 (d, J = 8.4 Hz, 2H), 6.32 (s, 1H), 3.35-3.13 (AB system, J = 18.2 Hz, 2H), 2.26 (s, 2H), 1.13 (s, 3H), 1.30 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz) $\delta = 192.1$, 156.0, 154.3, 151.1, 135.4, 134.6, 133.7, 131.9, 128.9, 128.89, 128.82, 128.0, 127.7, 122.7, 118.0, 64.4, 50.8, 38.0, 34.6, 28.6, 28.4. ESI- MS: m/z 451, 453 [M + H]⁺. Anal. Calcd for C₂₃H₁₉ BrN₂O₃: C, 61.21; H, 4.24; N, 6.21. Found: C, 61.40; H, 4.12; N, 6.48.

8. Compound 4h



Yellow solid. IR (KBr): 2965, 2369, 1666 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) δ = 8.28-8.18 (m, 2H), 7.80-7.78 (m, 2H), 7.34-7.31 (m, 2H), 6.97 (t, *J* = 8.2 Hz, 2H), 6.36 (s, 1H), 3.36-3.14 (AB system, *J* = 18.8 Hz, 2H), 2.27 (s, 2H), 1.14 (s, 6H). ESI- MS: *m*/*z* 391 [M + H]⁺. Anal. Calcd for C₂₃H₁₉ FN₂O₃: C, 70.76; H, 4.91; N, 7.18. Found: C, 70.65; H, 4.88; N, 7.30.

9. Compound 4i



Yellow solid. IR (KBr): 2959, 2362, 1666 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) $\delta = 8.30$ -8.19 (m, 2H), 7.81-7.78 (m, 2H), 7.30-7.19 (m, 4H), 6.33 (s, 1H), 3.35-3.15 (AB system, J = 18.6 Hz, 2H), 2.27 (s, 2H), 1.14 (s, 6H). Anal. Calcd for C₂₃H₁₉ ClN₂O₃: C, 67.90; H, 4.71; N, 6.89. Found: C, 67.61; H, 4.56; N, 7.15.

10. Compound 4j



Yellow solid. IR (KBr): cm⁻¹.¹H NMR (CDCl₃, 400 MHz) $\delta = 8.30-8.16$ (m, 2H), 7.79-7.77 (m, 2H), 7.41-7.14 (m, 4H), 6.60 (s, 1H), 3.35-3.14 (AB system, J = 18.0 Hz, 2H), 2.25 (s, 2H), 1.14 (s, 6H). ESI- MS: m/z 407, 409 [M + H]⁺. Anal. Calcd for C₂₃H₁₉ ClN₂O₃: C, 67.90; H, 4.71; N, 6.89. Found: C, 67.93; H, 4.77; N, 6.71.

11. Compound 4k



Yellow solid. IR (KBr): 2960, 2382, 1666 cm⁻¹.¹H NMR (CDCl₃, 400 MHz) $\delta = 8.32$ -8.08 (m, 4H), 7.83-7.82 (brs, 3H), 7.51-7.47 (m, 1H), 6.45 (s, 1H), 3.38-3.18 (AB system, J = 18.7 Hz, 2H), 2.28 (s, 2H), 1.15 (s, 6H). ESI- MS: m/z 418 [M + H]⁺. Anal. Calcd for C₂₃H₁₉N₃O₅: C, 66.18; H, 4.59; N, 10.07. Found: C, 66.37; H, 4.47; N, 10.34.

12. Compound 4l



Yellow solid. IR (KBr): 2965, 2372, 1666 cm⁻¹.¹H NMR (CDCl₃, 300 MHz) $\delta = 8.37$ -8.24 (m, 4H), 7.87-7.13 (m, 4H), 6.71 (s, 1H), 3.43-3.21 (AB system, J = 19.2 Hz, 2H), 2.32 (s, 2H), 1.25 (s, 3H), 1.21 (s, 3H). ESI- MS: m/z 451, 453 [M + H]⁺. Anal. Calcd for C₂₃H₁₉ BrN₂O₃: C, 61.21; H, 4.24; N, 6.21. Found: C, 61.10; H, 4.10; N, 6.24.

13. Compound 4m



Yellow solid. IR (KBr): 2965, 2375, 1666 cm⁻¹.¹H NMR (CDCl₃, 300 MHz) $\delta = 8.27$ -8.24 (m, 4H), 7.88-6.98 (m, 4H), 6.53 (s, 1H), 3.45-3.18 (AB system, J = 18.0 Hz, 2H), 2.33 (s, 2H), 1.25 (s, 3H), 1.92 (s, 3H). ESI- MS: m/z 391 [M + H]⁺. Anal. Calcd for C₂₃H₁₉FN₂O₃: C, 70.76; H, 4.91; N, 7.18. Found: C, 70.71; H, 5.14; N, 7.06.

14. Compound 4n



Yellow solid. IR (KBr): 2966, 2375, 1665 cm⁻¹.¹H NMR (CDCl₃, 300 MHz) $\delta = 8.35$ -8.25 (m, 4H), 7.90-7.84 (m, 4H), 6.39 (s, 1H), 3.44-3.20 (AB system, J = 17.5 Hz, 2H), 2.34 (s, 2H), 1.21 (s, 6H). ESI- MS: m/z 398 [M + H]⁺. Anal. Calcd for C₂₃H₁₉ BrN₂O₃: C, 61.21; H, 4.24; N, 6.21. Found: C, 60.93; H, 4.12; N, 6.30.

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3. ¹H and ¹³ C NMR spectra

1. Compound 4a



¹H NMR Spectra of Compound 4a

2. Compound 4b



¹H NMR Spectra of Compound 4b

3. Compound 4c



¹H NMR Spectra of Compound 4c

4. Compound 4d



¹H NMR Spectra of Compound 4d

5. Compound 4d



¹³C NMR Spectra of Compound 4d

6. Compound 4e



¹H NMR Spectra of Compound 4e

7. Compound 4e



¹³C NMR Spectra of Compound 4e

8. Compound 4f



¹H NMR Spectra of Compound 4f

9. Compound 4f



¹³C NMR Spectra of Compound 4f

10. Compound 4g



¹H NMR Spectra of Compound 4g

11.Compound 4g



¹³C NMR Spectra of Compound 4g

12.Compound 4h



¹H NMR Spectra of Compound 4h

13.Compound 4i



¹H NMR Spectra of Compound 4i

14.Compound 4j



¹H NMR Spectra of Compound 4j

15. Compound 4k



¹H NMR Spectra of Compound 4k

16. Compound 4l



¹H NMR Spectra of Compound 41



¹H NMR Spectra of Compound 4m

18. Compound 4n



¹H NMR Spectra of Compound 4n