Ring-Opening Polymerization of Lactones using Supramolecular Organocatalysts, Under Simple Conditions

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1/ Optimization of the H-bond Acceptor Catalyst

To define the appropriate amine that could activate together with the phenol catalyst, preliminary ROP were achieved with δ -VL (2M in CH₂Cl₂), with a 5 mol % catalytic loading, in the presence of BPM as an initiator (5 mol %), in 6 h (except Sp in 24 h) (Table S1).



Table S1. Screening of the catalytic system (H-bond Donor + Acceptor) on the ROP of δ -VL, based on the percentage of conversion after 6 h.^{*a,b*}

Run	H-Bond Donor	Sp ^c	BTMG	DBN	DBU	MTBD
1	None, no BPM	12	26	23	24	10
2	None	15	30	26	34	24
3	TU	42	66	80	100	75
4	TUS	28	41	35	59	30
5	1d	12	43	17	73	50
6	1f	18	37	21	73	45
7	1g	12	52	21	54	48
8	2a	30	31	29	73	43
9	2b	30	40	39	78	53
10	2c	45	38	19	70	42
11	3a	26	48	22	84	40
12	3 b	26	50	29	84	45

^{*a*} Conditions of ROP: δ-Valerolactone (2 M in dichloromethane), H-bond donor catalyst (5 mol %), amine/amidine/guanidine cocatalyst (5 mol %), 4-biphenylmethanol as an initiator (5 mol %), 4 Å molecular sieves, 20 °C, 6 h. ^{*b*} Determined by ¹H NMR (CDCl₃). ^{*c*} Reaction during 24 h.

Note: pK_{A CH3CN} values: Sp (21.7), BTMG (23.6), DBN (23.9), DBU (24.3), MTBD (25.4) and phenols (26-28).

2/ Mass Spectra of representative polylactones obtained from the general procedure (Tables 1 and 2)

a) Figure S1: PVL (Run 31) using VL (4M in CH₂Cl₂), DBU/5-methoxyresorcinol/BPM 5:5:5



b) Figure S2: PVL (Run 22) using VL (4M in CH₂Cl₂), DBU/Catechol/BPM 5:5:1



Enlargement Figure S2



c) Figure S3: PCL using CL (4M in CH₂Cl₂), DBU/4-tert-butylcatechol/BPM 5:5:5



3/ ¹³C NMR and ¹H NMR (CDCl₃)

of representative polymers PVL and PCL (Tables 1 and 2)

Figure S4: ¹³C NMR of PVL (**Run 31**) from VL (4M, CH₂Cl₂), DBU/5-methoxyresorcinol/BPM 5:5:5







Figure S6: ¹³C NMR of PCL from CL (4M in CH₂Cl₂), DBU/4-tert-butylcatechol/BPM 5:5:5





Figure S7: ¹H NMR of PCL from CL (4M in CH₂Cl₂), DBU/4-tert-butylcatechol/BPM 5:5:5

4/ ¹³C NMR and ¹H NMR (CDCl₃) of representative copolymers (PVL-b-PLA) and (PCL-b-PLA) (Table 3)

Figure S8: ¹³C NMR of copolymer (PVL-b-PLA), from ROP in the presence of propan-1-ol /2-(trifluoromethyl)phenol/DBU 5:5:5, with ZOOM



Figure S9: ¹H NMR of copolymer (PVL-b-PLA), from ROP in the presence of propan-1-ol/2-(trifluoromethyl)phenol/DBU 5:5:5



Figure S10: ¹H NMR of copolymer (PVL-b-PLA), from ROP in the presence of propan-1-ol/4-*tert*-butylcatechol /DBU 5:5:5













5/ DSC Analyses of representative copolymers (PVL-b-PLA) and (PCL-b-PLA)



b) Figure S14: (PCL-b-PLA) Copolymer

1) Hold for 1.0 min at -90.00°C



2) Heat from -90.00°C to 200.00°C at 10.00°C/min

200

6 / SEC Analyses of representative polymers PVL, PCL, (PVL-*b*-PLA) and (PCL-*b*-PLA)

Figure S15: SEC (RI) of PVL from ROP in the presence of BPM /5-methoxyresorcinol/DBU 5:5:5



Peak No	Мр	Mn	Mw	Mz	Mz+1	Μv	PD
1	1444	1618	1792	2013	2291	1762	1.10754

Figure S16: SEC (UV) of PVL from ROP in the presence of BPM /5-methoxyresorcinol/DBU 5:5:5



MW Averages										
Peak No	Мр	Mn	Mw	Mz	Mz+1	Mv	PD			
1	1442	1401	1572	1712	1855	1551	1.12206			





MW Averages										
Peak No	Мр	Mn	Mw	Mz	Mz+1	Mv	PD			
1	2877	2680	3099	3685	4578	3027	1.15634			

Figure S18: SEC (RI) of copolymer (PVL-b-PLA) from ROP in the presence of propan-1-ol/2-(trifluoromethyl)phenol/DBU 5:5:5



MW	Averages	
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Peak No	Мр	Mn	Mw	Mz	Mz+1	Mv	PD
1	6684	5550	7358	9741	12336	7043	1.32577

Figure S19: SEC (UV) of copolymer (PVL-b-PLA) from ROP in the presence of propan-1-ol/2-(trifluoromethyl)phenol/DBU 5:5:5





Figure S20: SEC (RI) of first block PCL from ROP in the presence of propan-1-ol/4-*tert*-butylcatechol/DBU 5:5:5

Figure S21: SEC (RI) of copolymer (PCL-*b*-PLA) from ROP in the presence of propan-1-ol/4-tert-butylcatechol/DBU 5:5:5



MW Averages										
Peak No	Мр	Mn	Mw	Mz	Mz+1	Μv	PD			
1	6407	6051	7434	9296	11577	7193	1.22856			
2	0	0	0	0	0	0	0			
3	8	7	7	8	8	7	1			



